

# CHEMISTRY

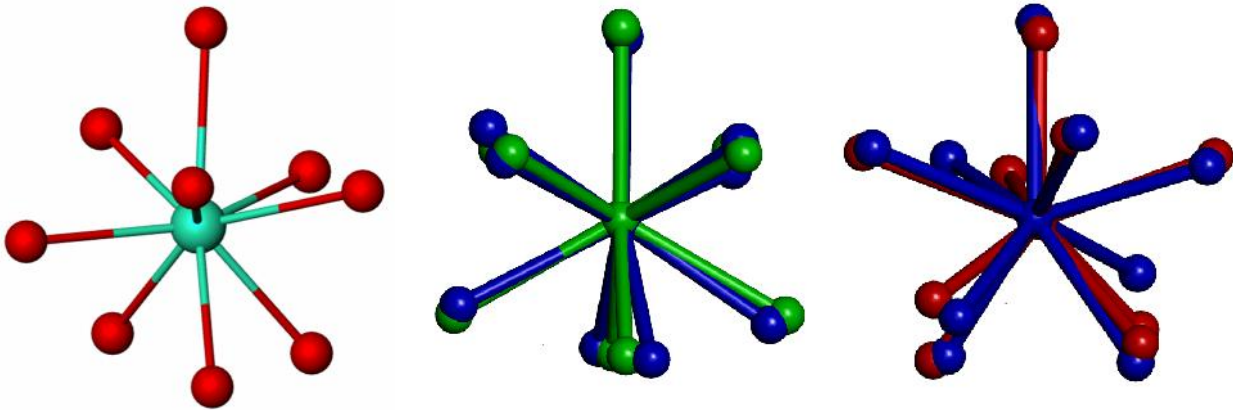
## A **European** Journal

### Supporting Information

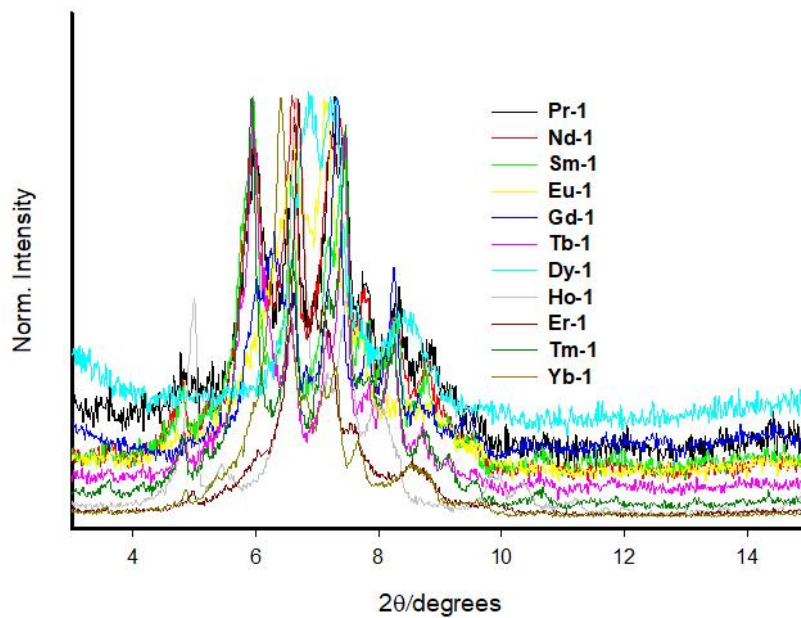
#### **A Unique Ln<sup>III</sup>{[3.3.1]Ga<sup>III</sup> Metallacryptate} Series That Possesses Properties of Slow Magnetic Relaxation and Visible/Near-Infrared Luminescence**

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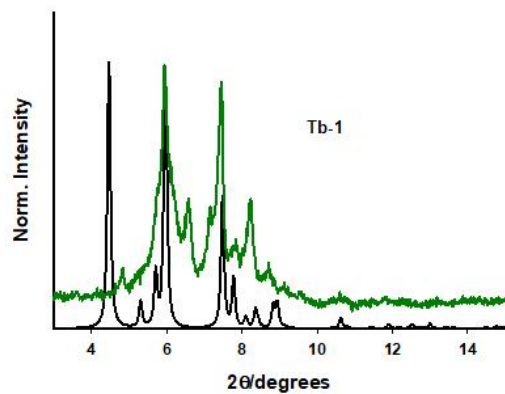
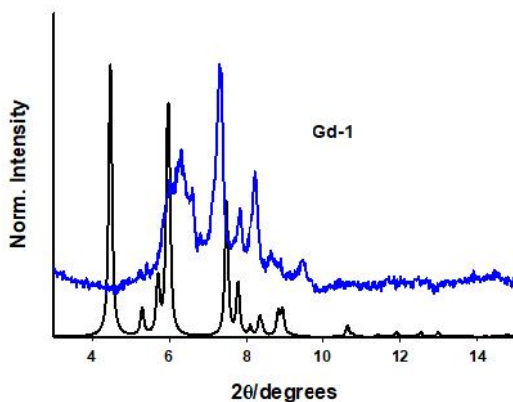
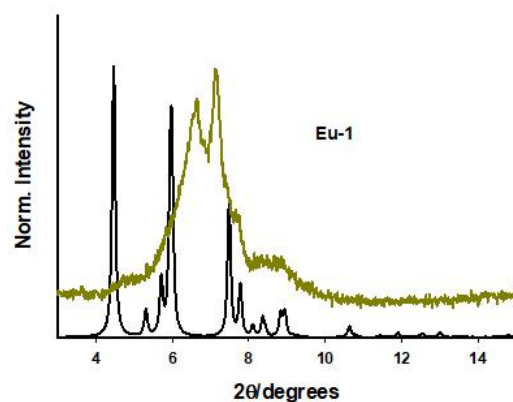
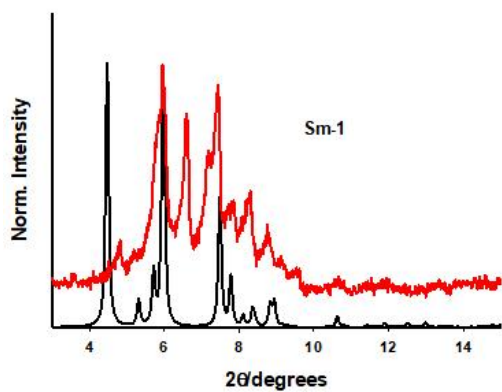
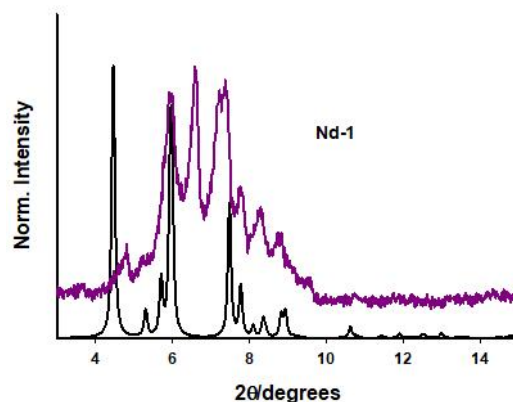
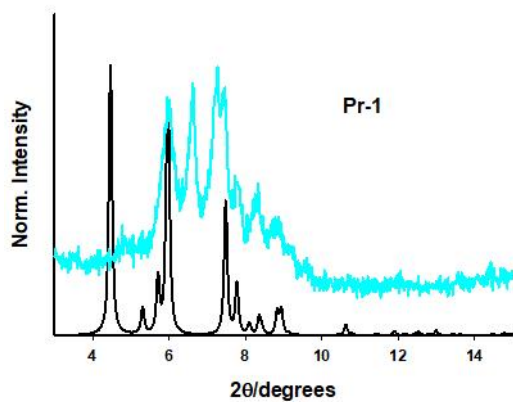
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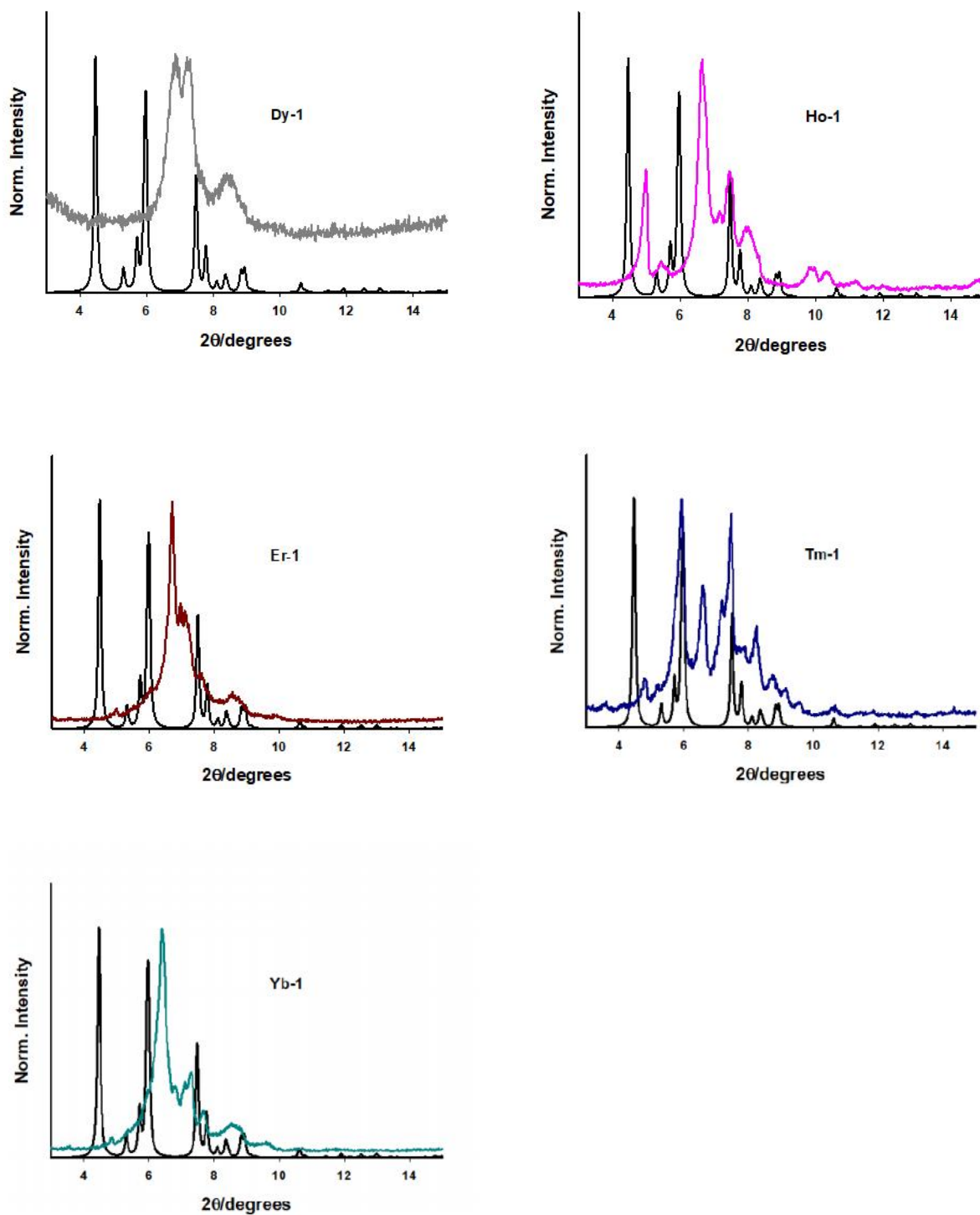


**Figure S1.** First coordination sphere of Tb1 in **Tb-1**, b) **Tb-1** (blue) overlaid with an ideal tricapped trigonal prism (green), c) **Tb-1** (blue) overlaid with an ideal monocapped square antiprism (red).



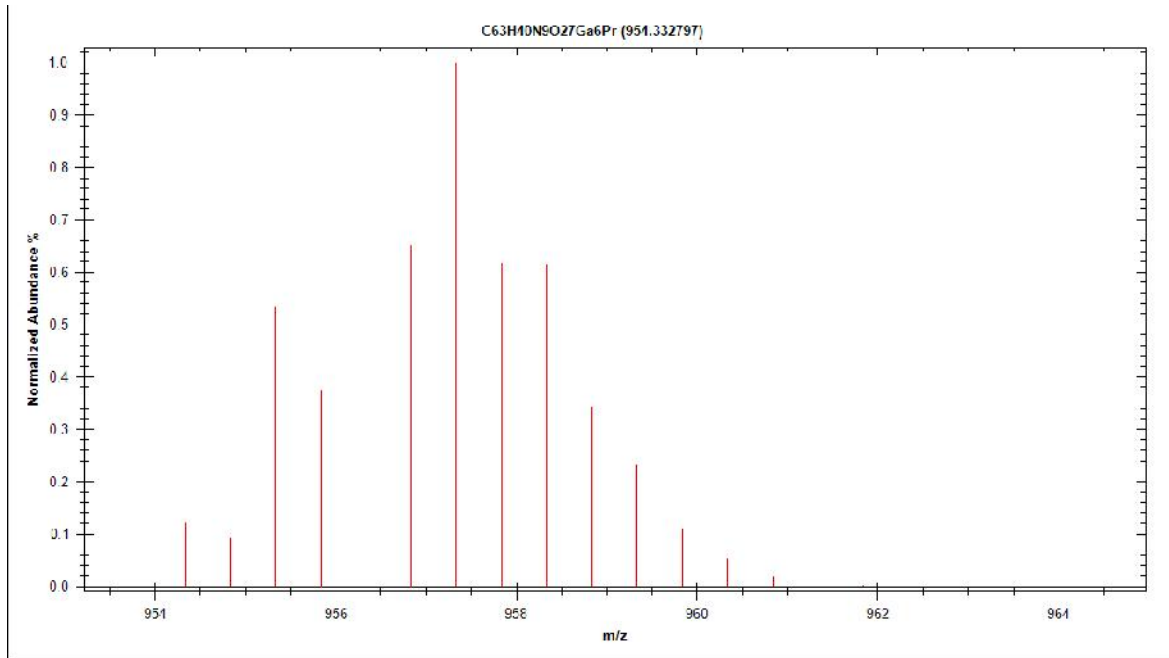
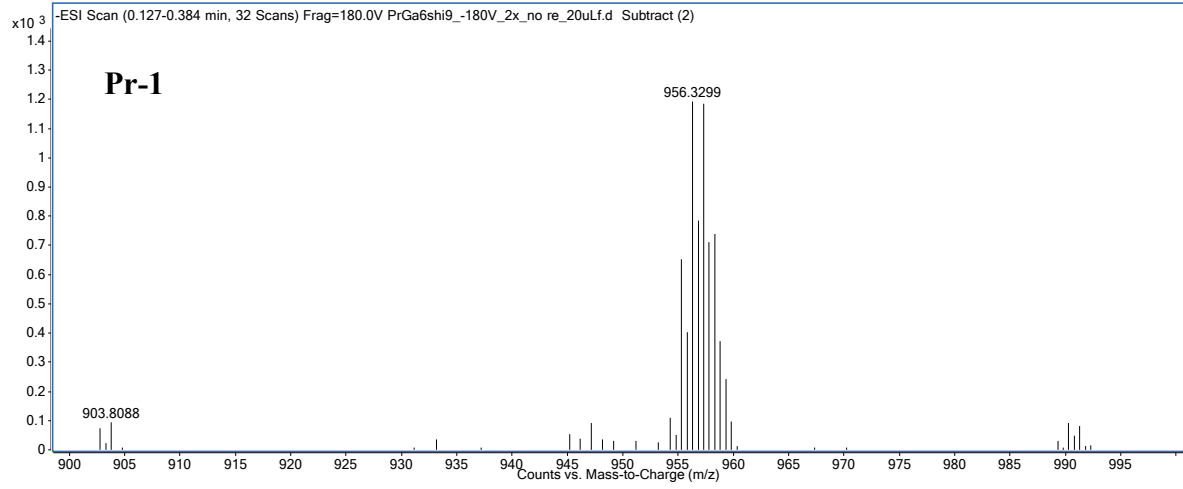
**Figure S2a.** Overlay of **Ln-1** PXRD spectra show nearly consistent composition.

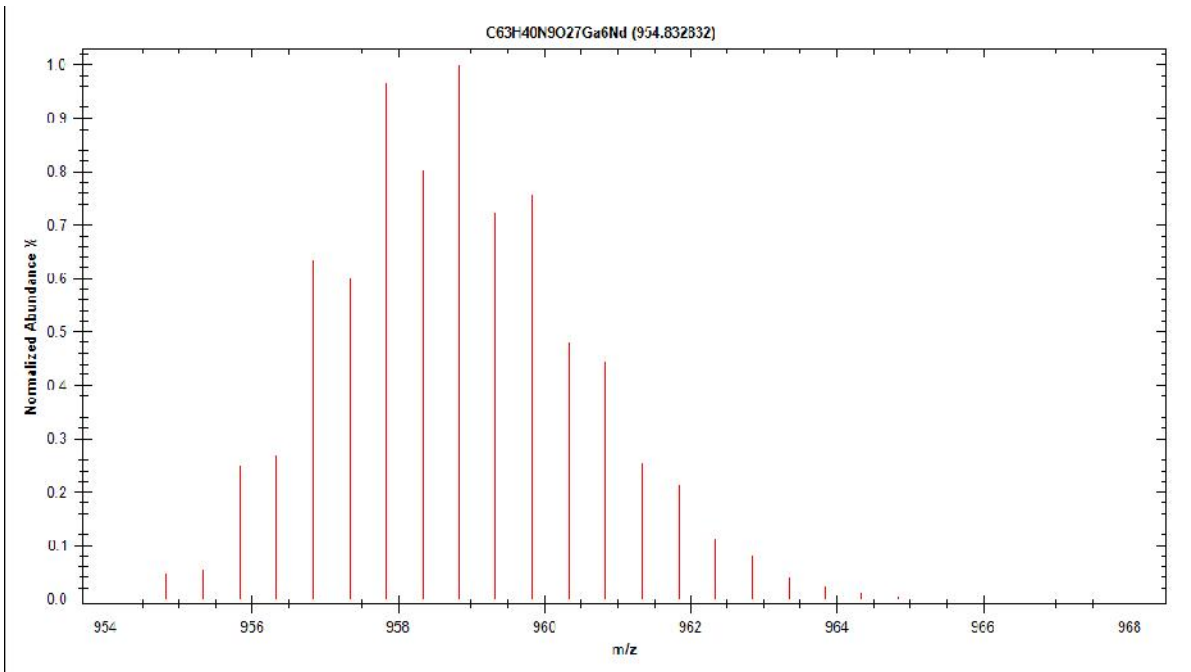
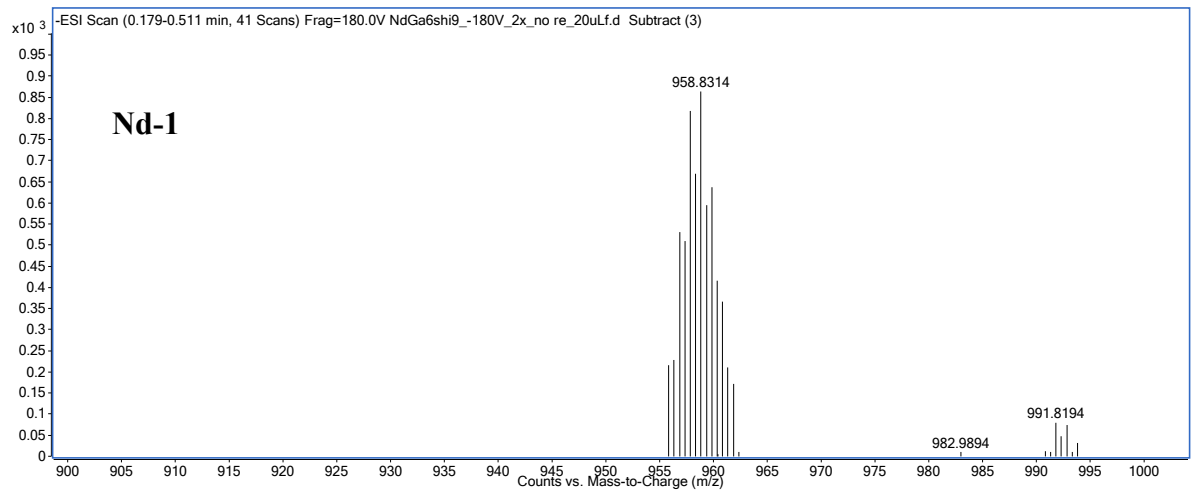


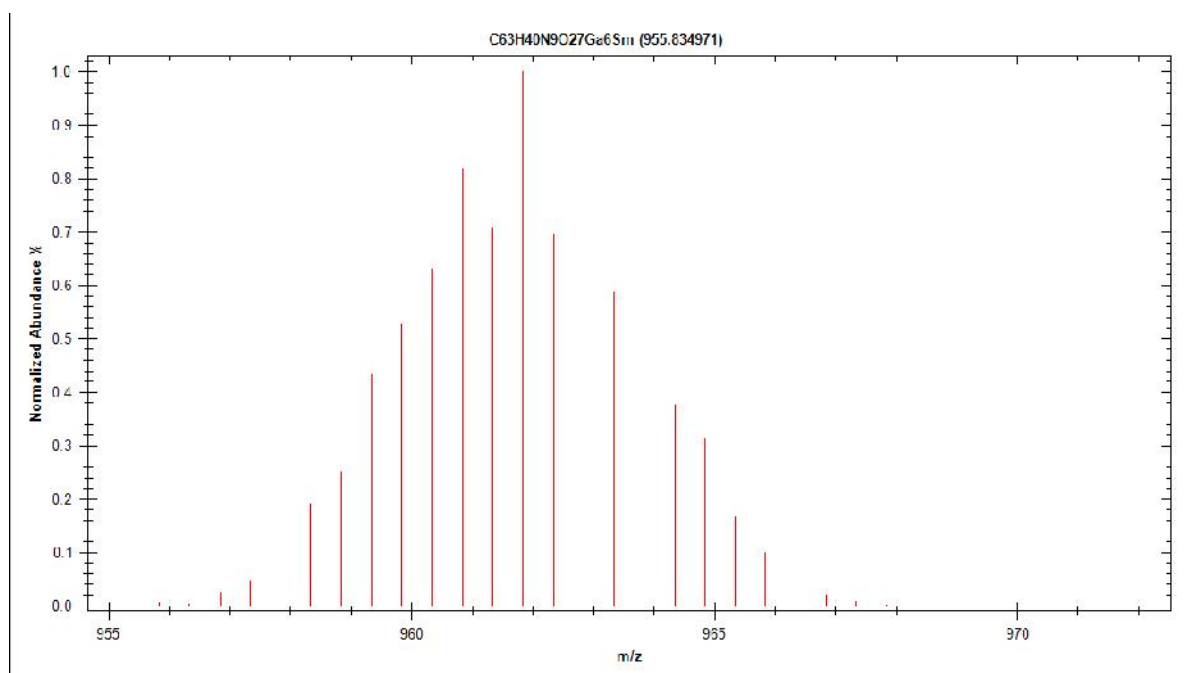
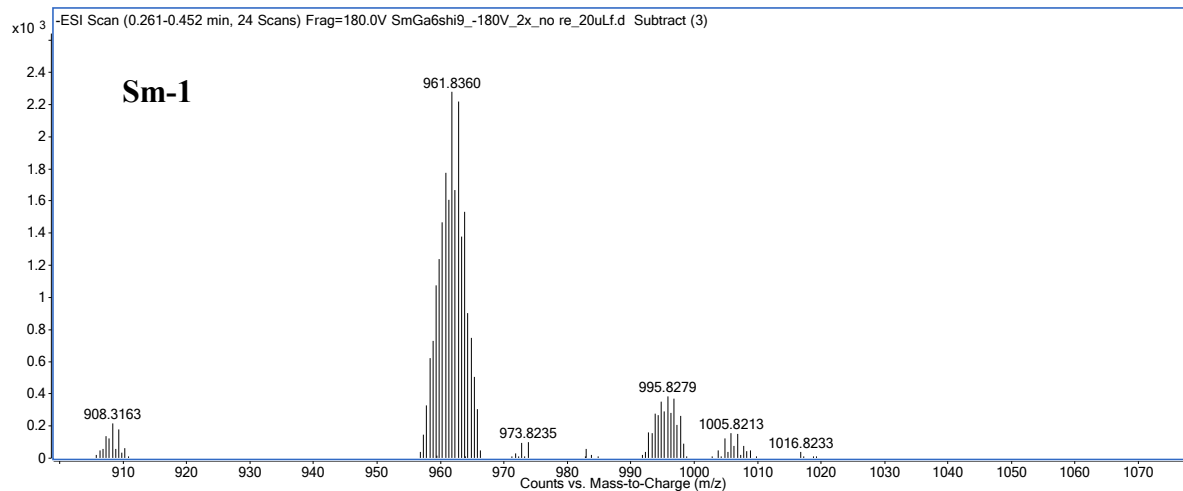


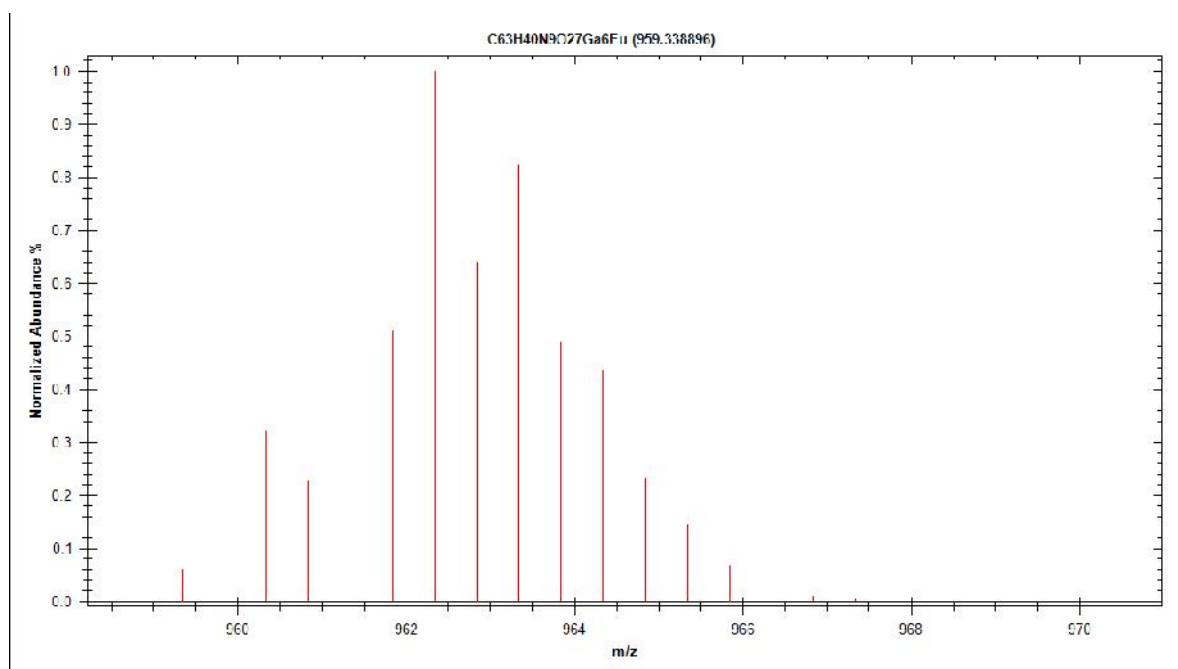
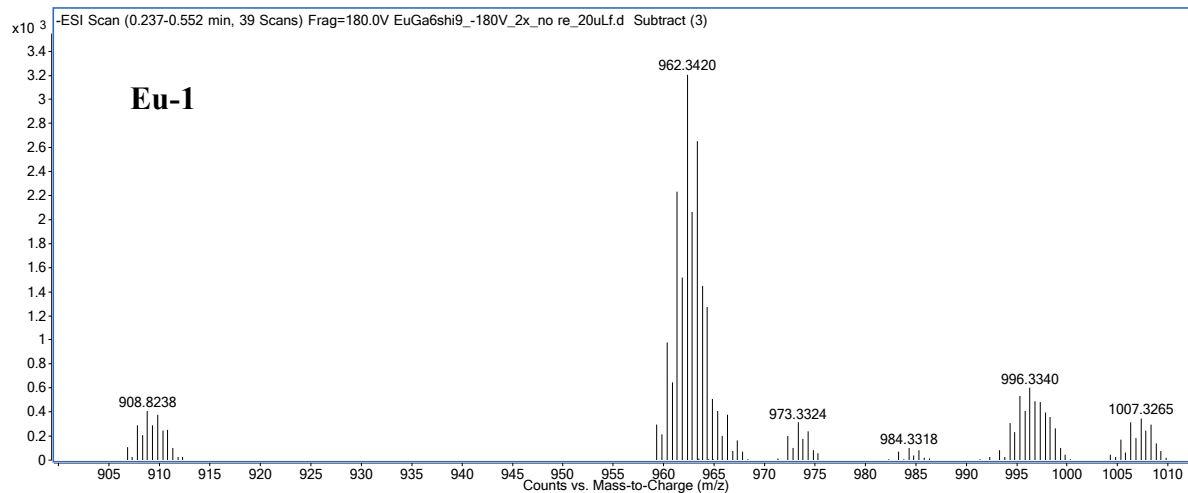
**Figure S2b.** Powder X-ray diffraction of **Ln-1** metallacryptates compared to a simulated pattern from the **Tb-1** single crystal x-ray data. There is significant deviation due to lattice solvent loss, giving altered reflections and less crystallinity.

# ESI-MS

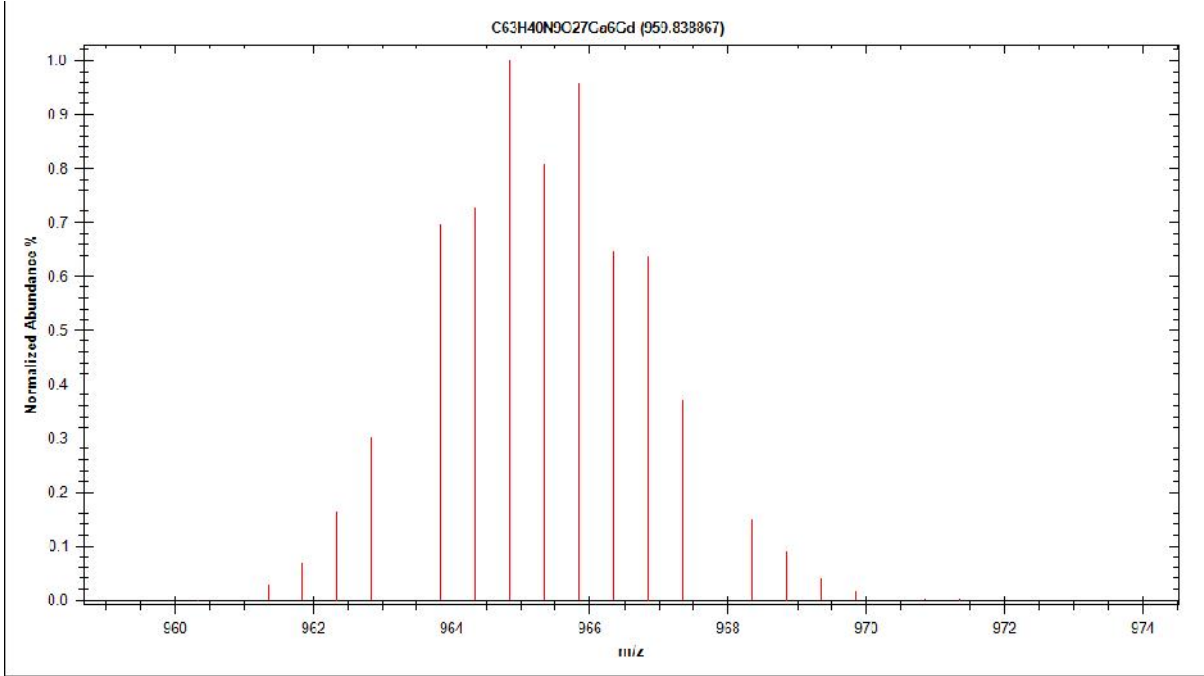
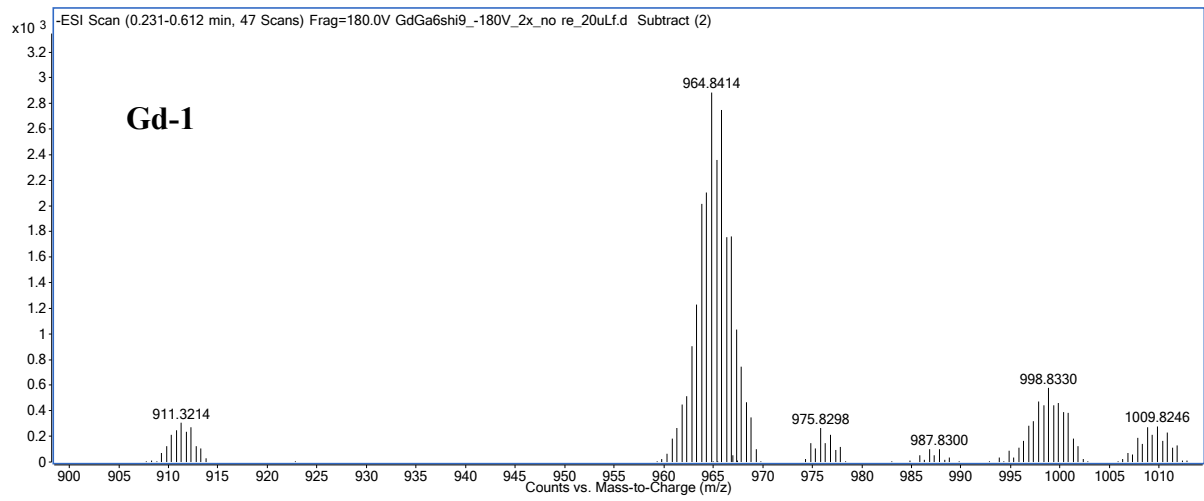


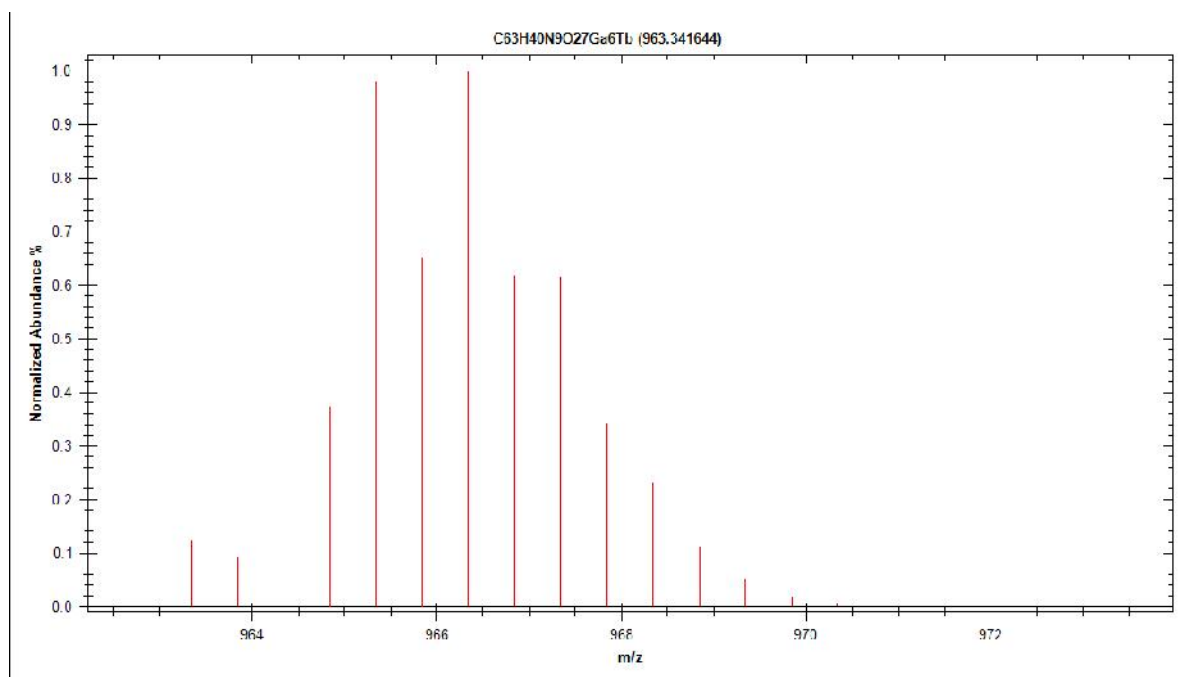
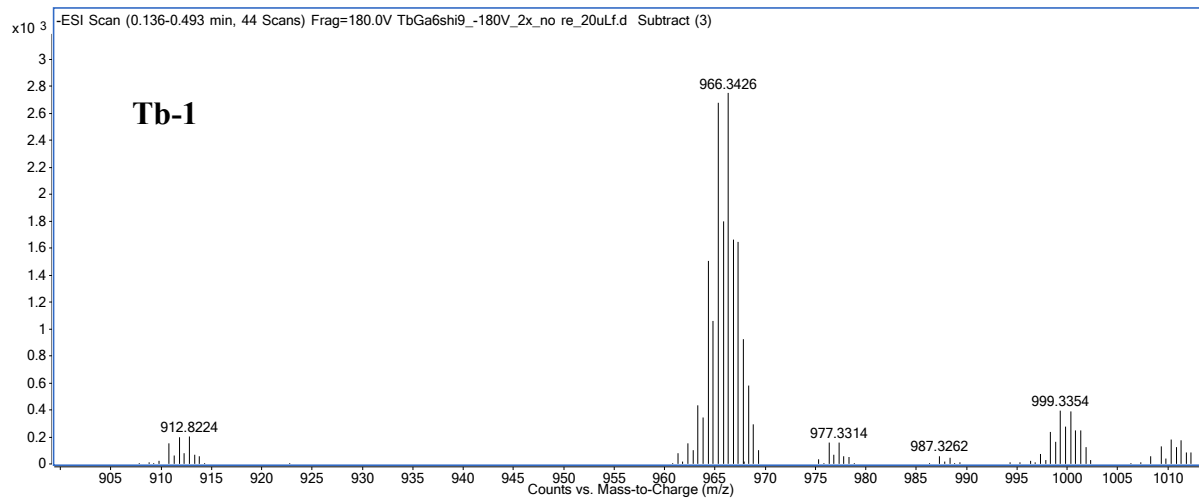


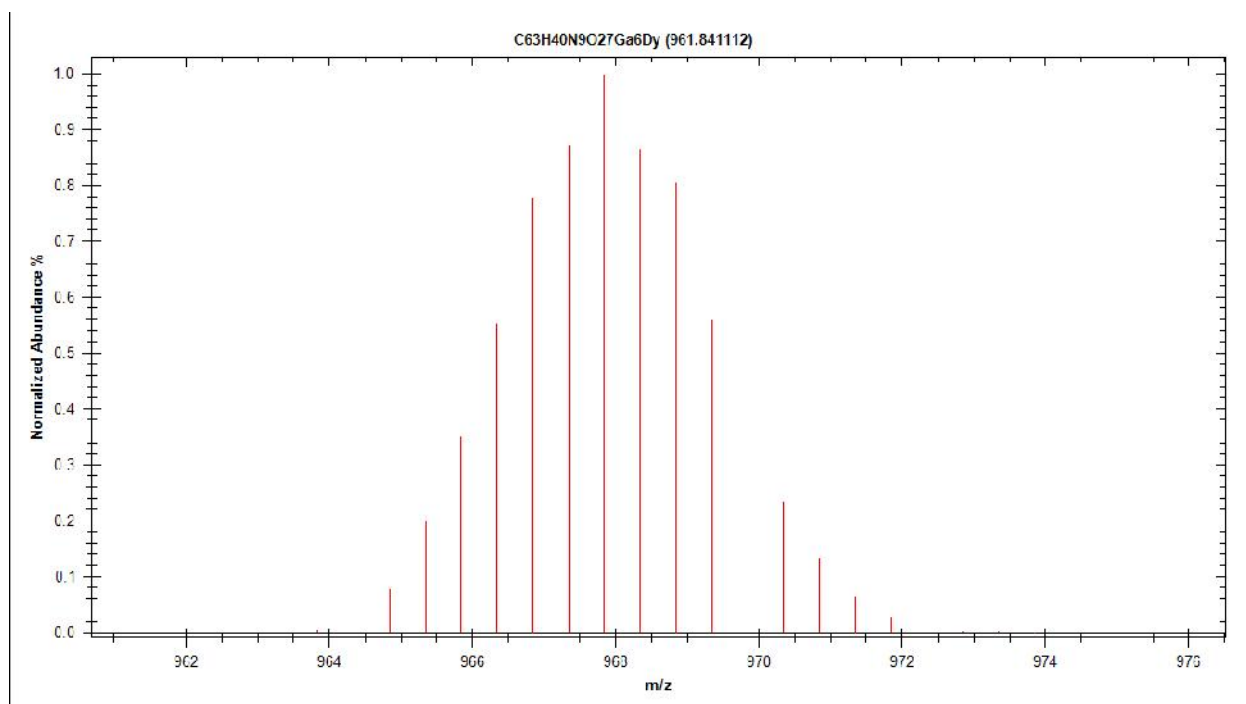
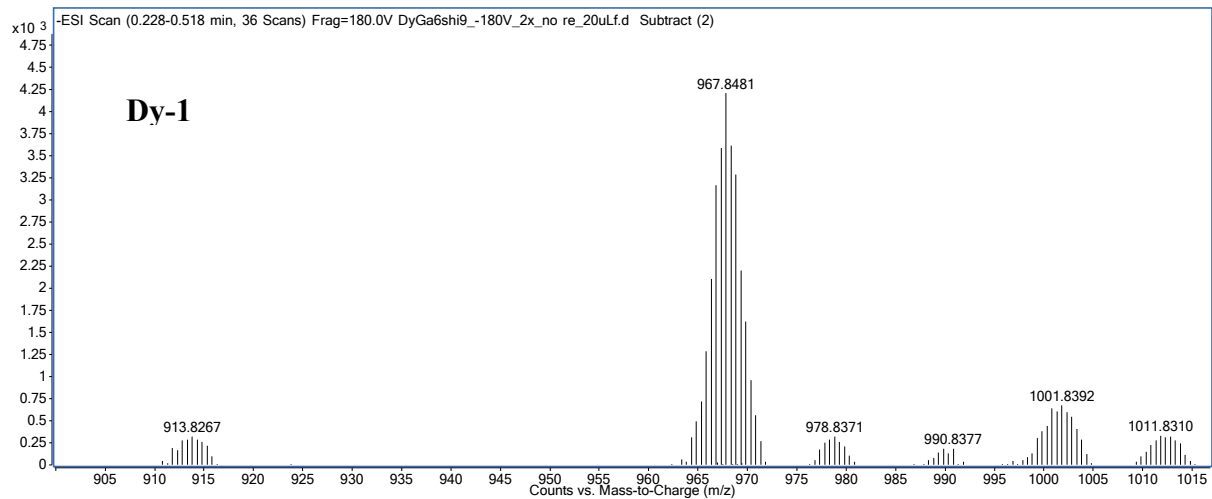


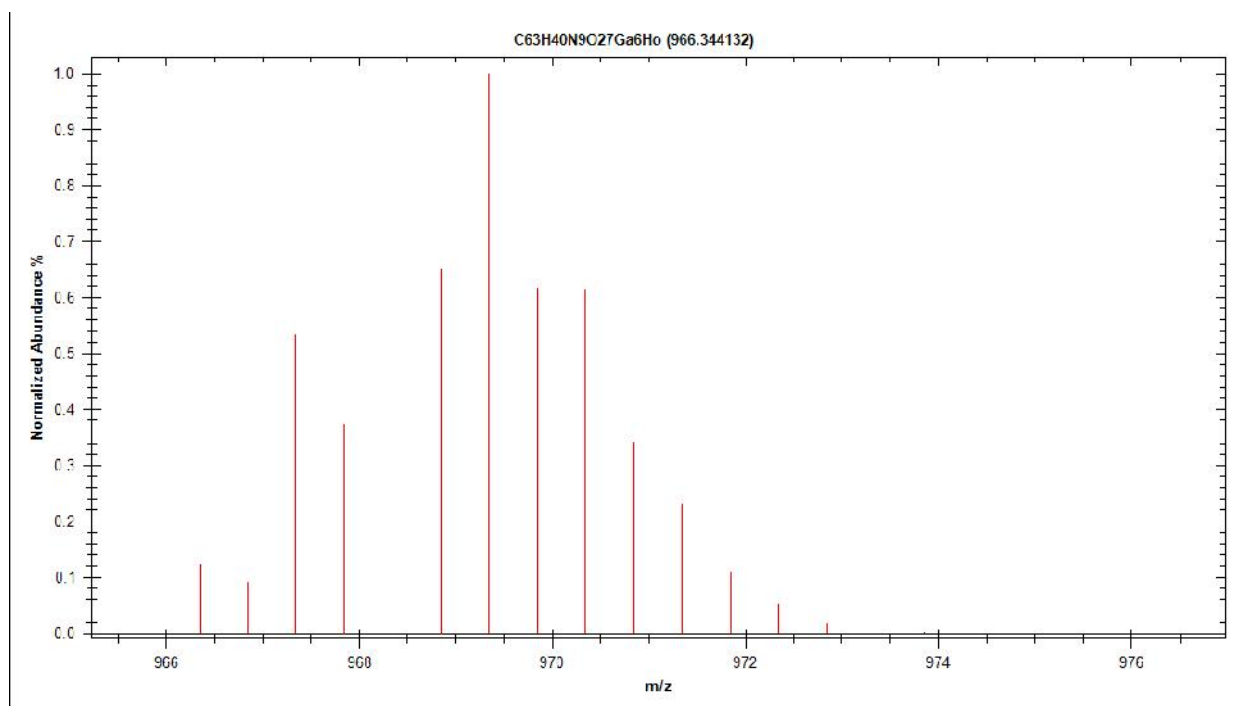
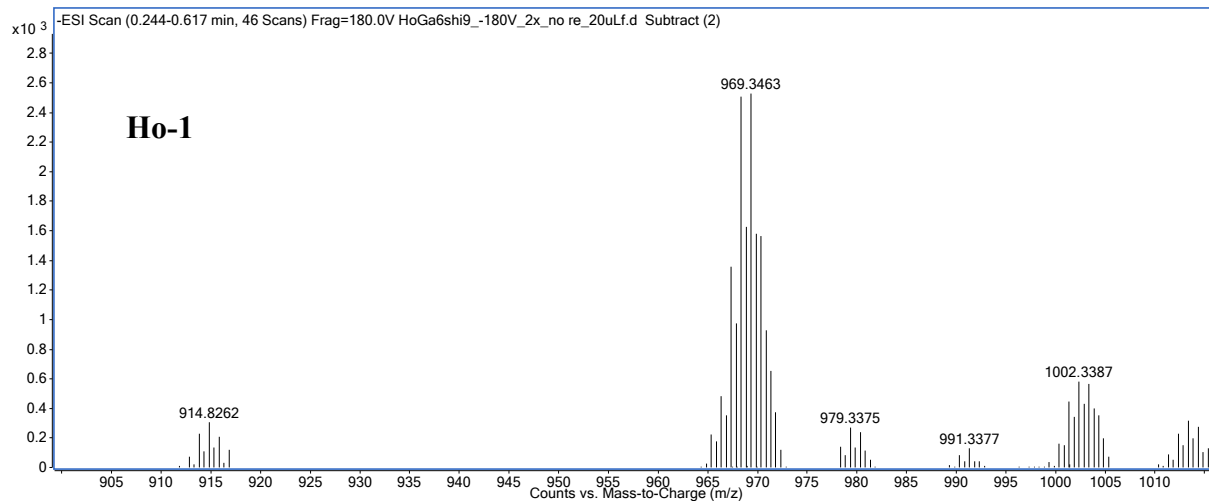


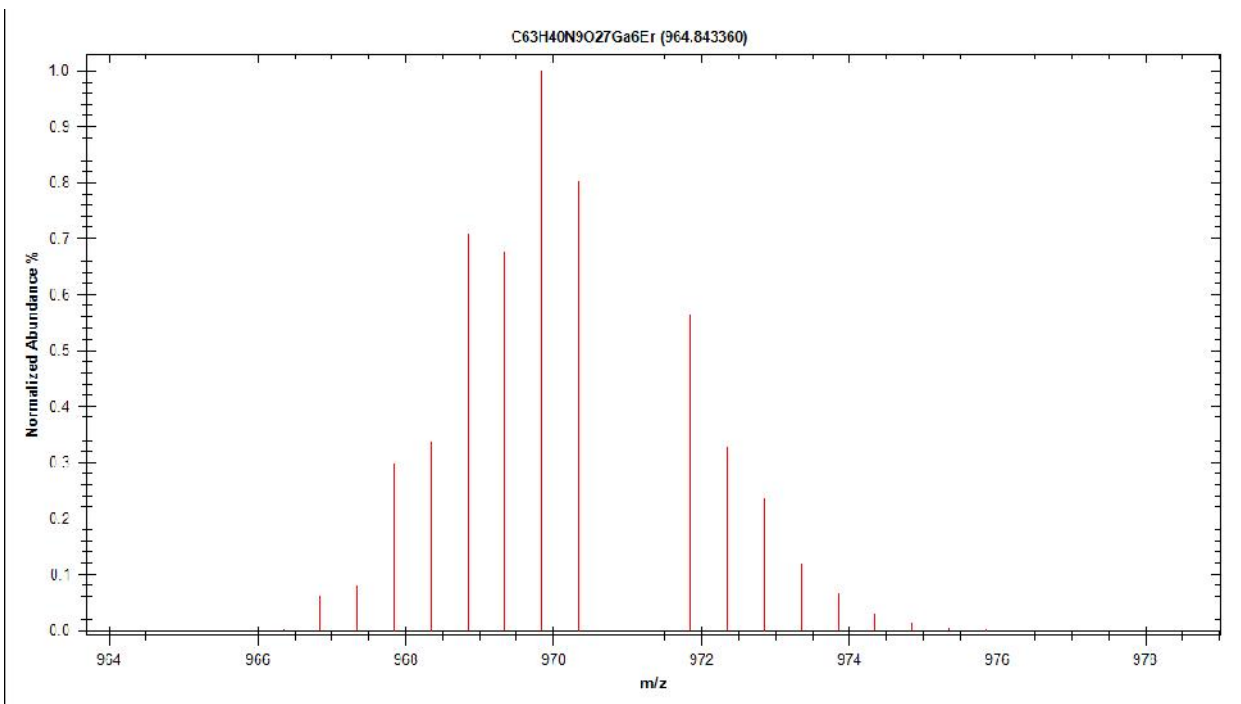
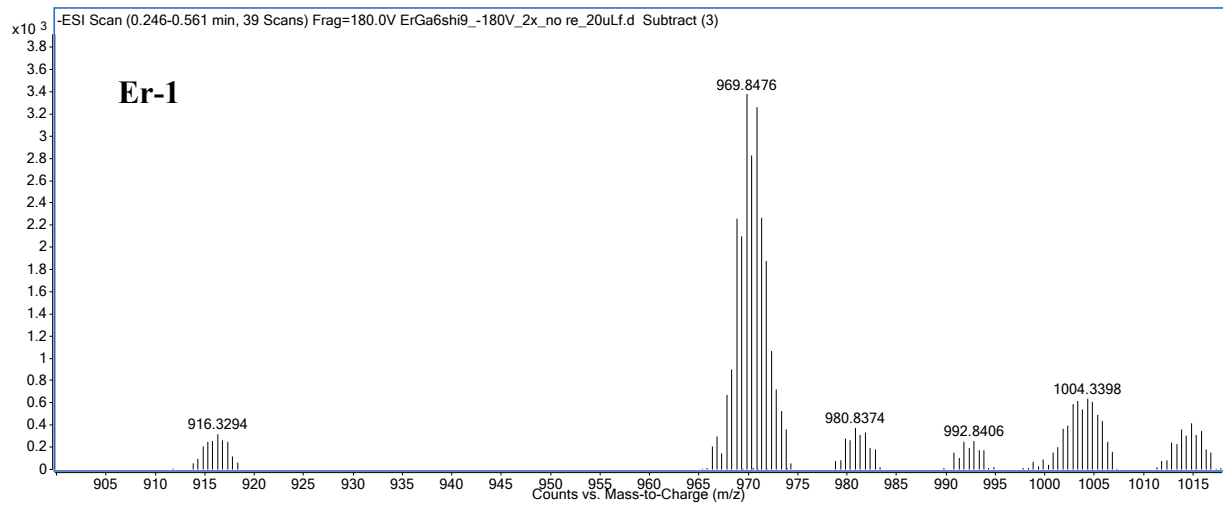


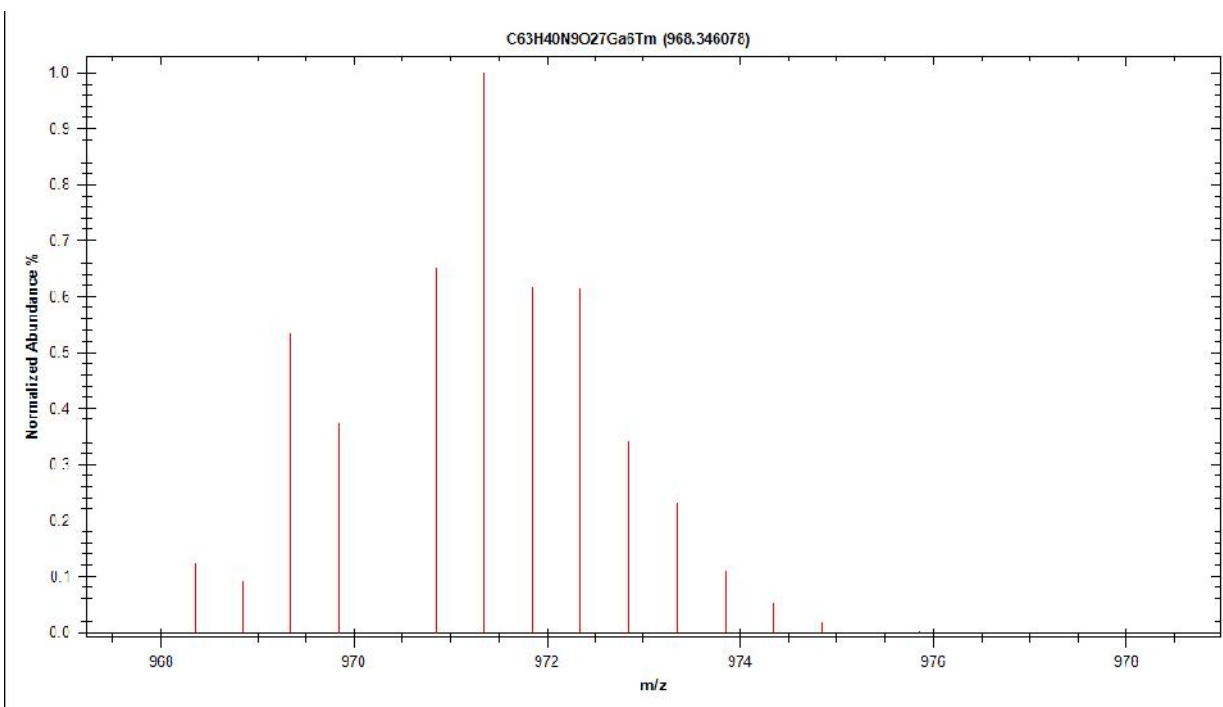
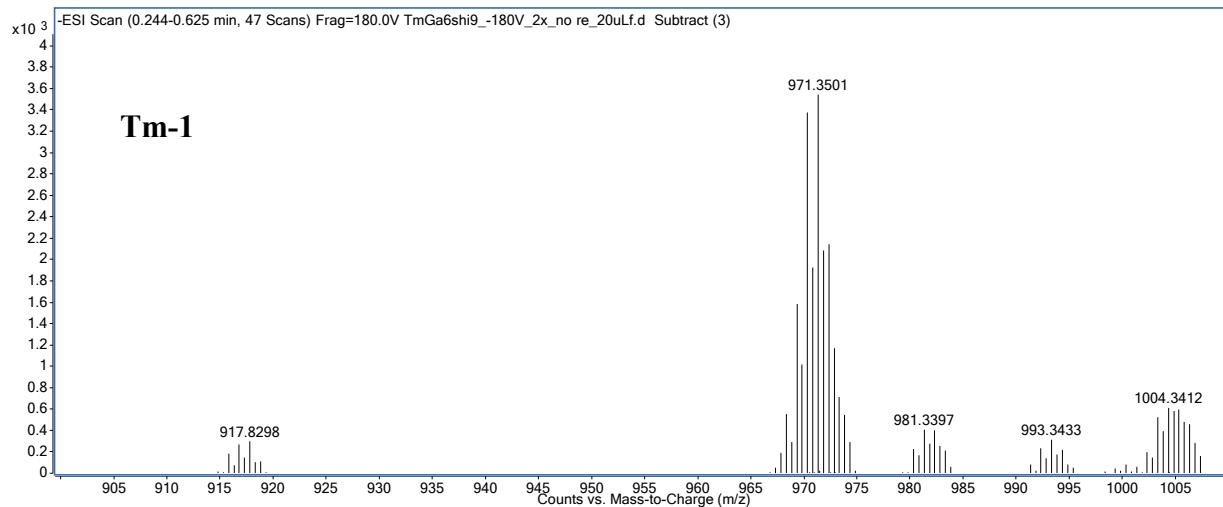


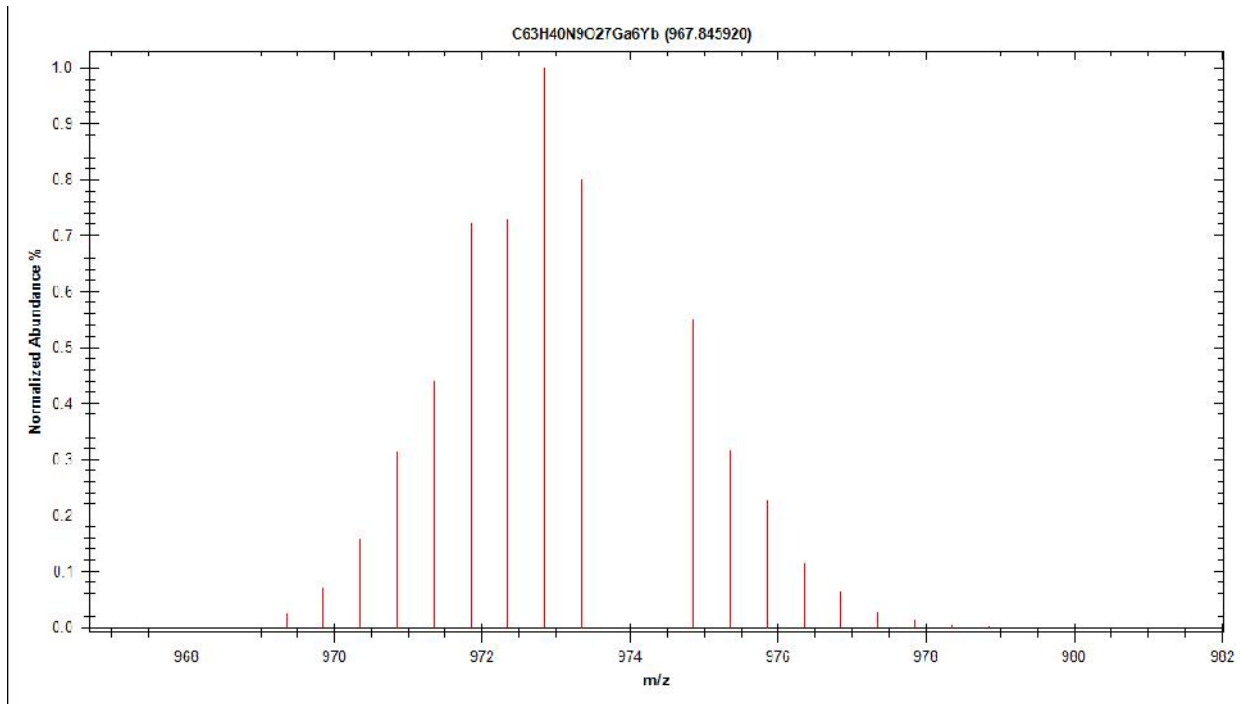
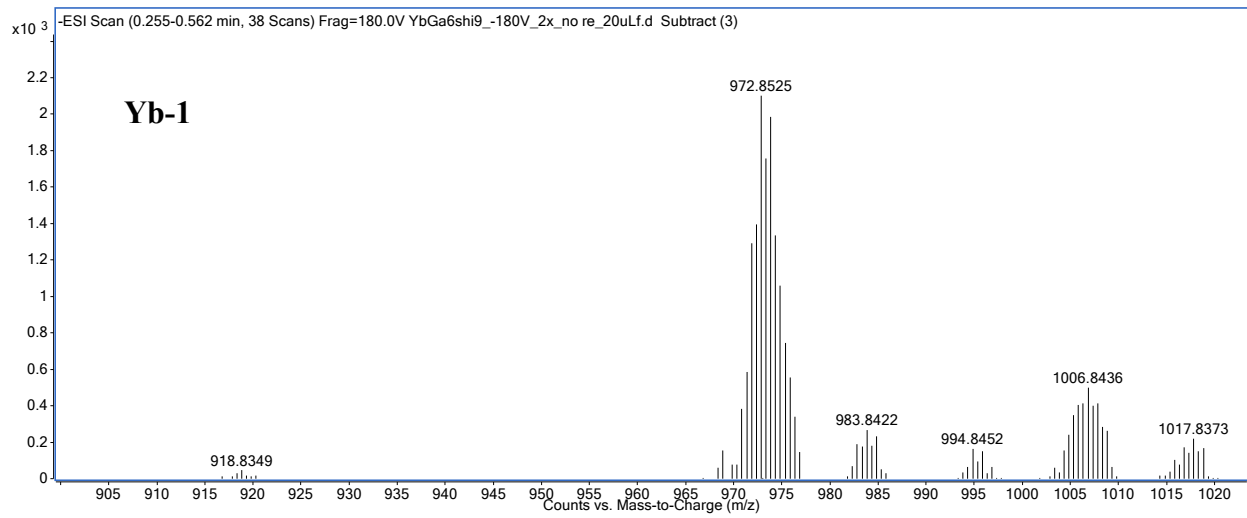


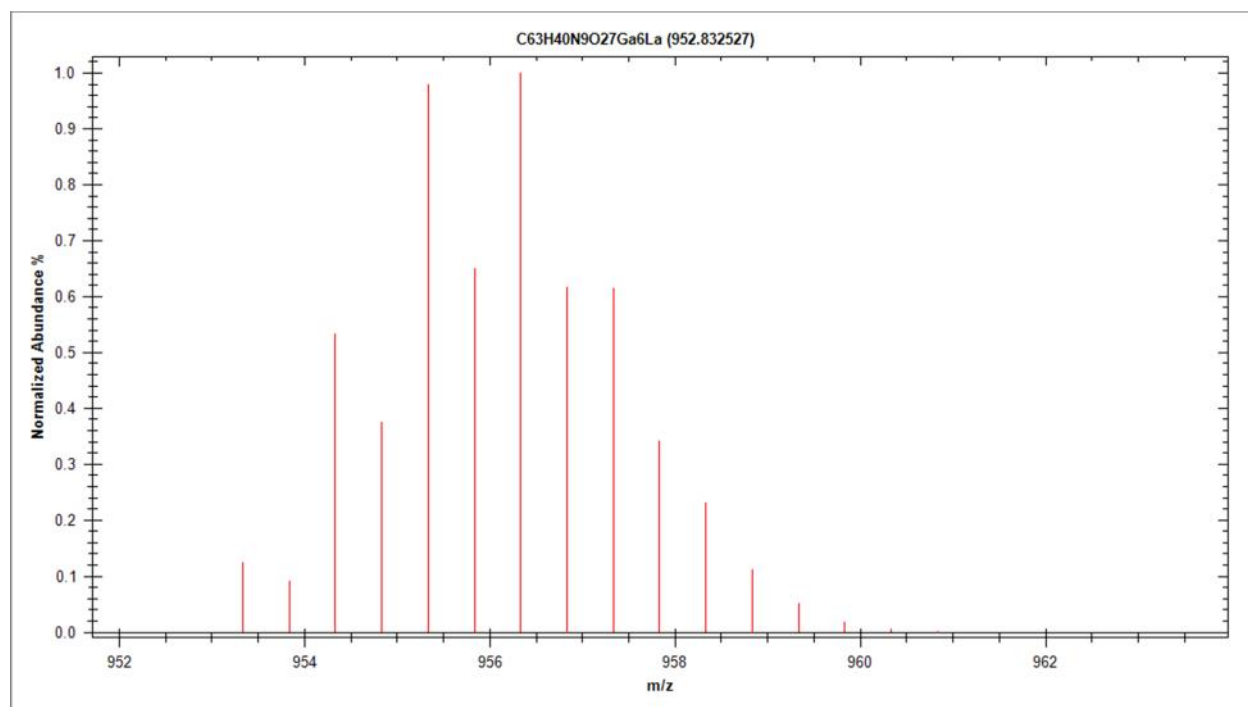
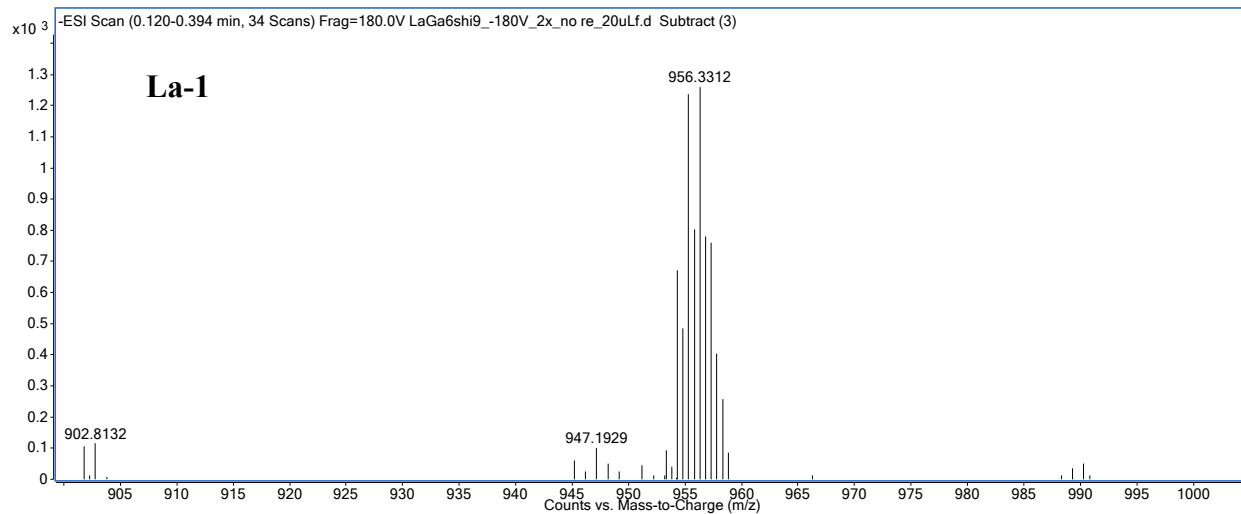




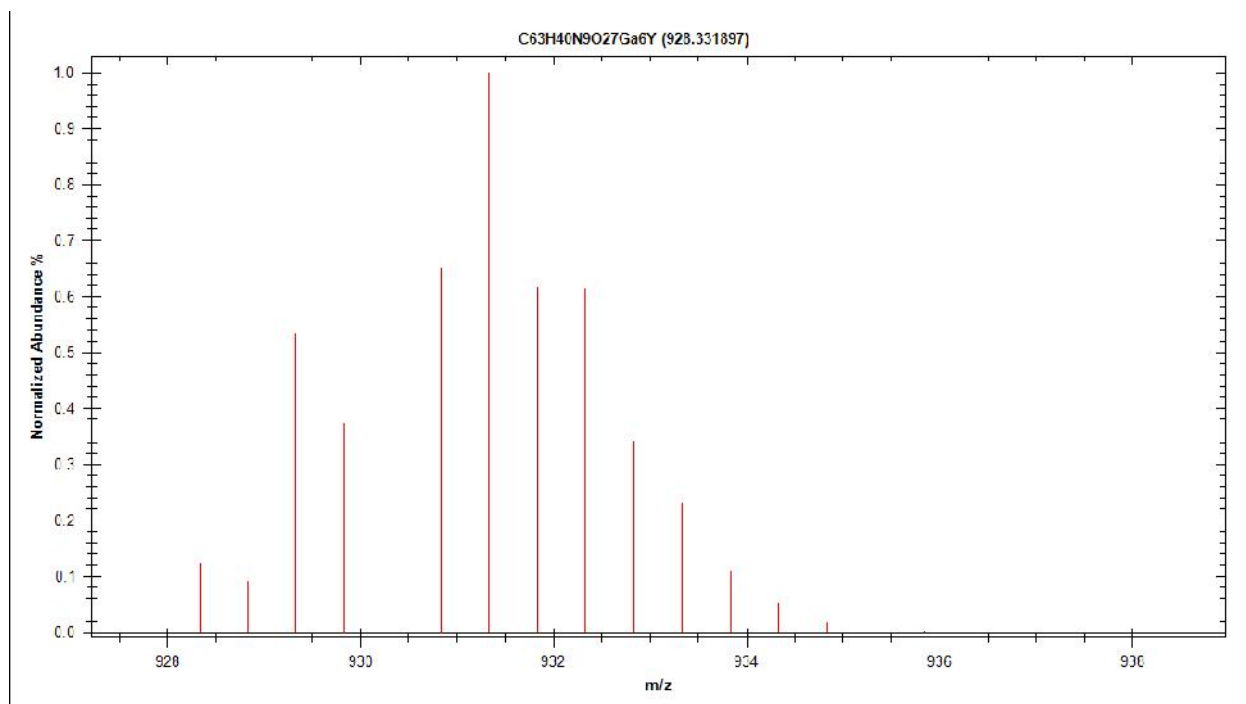
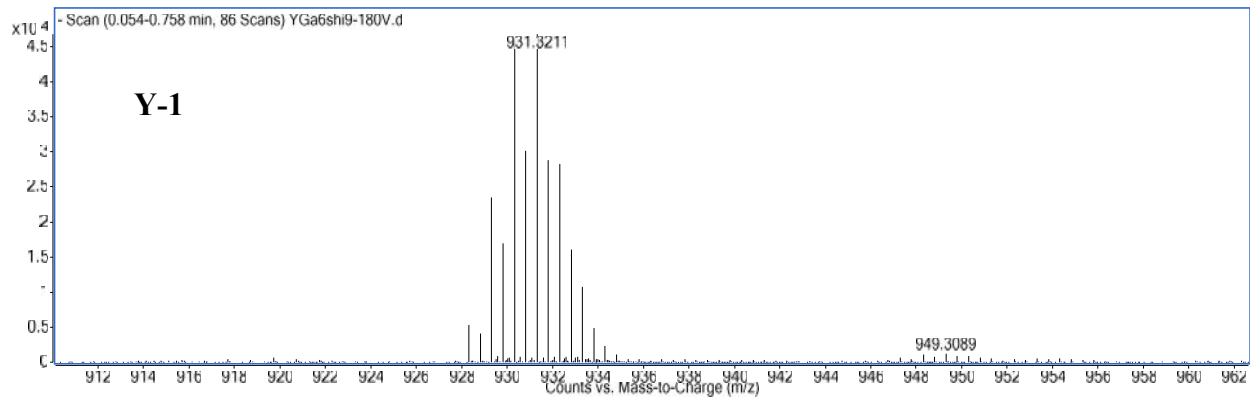


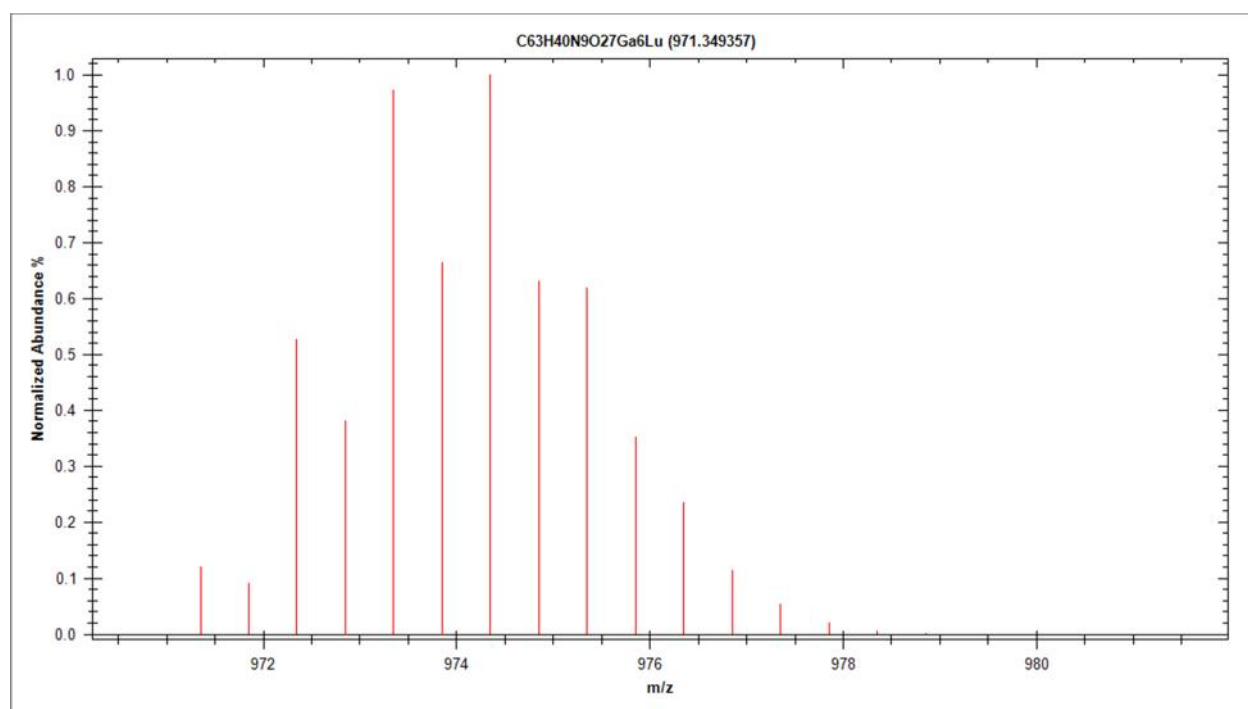
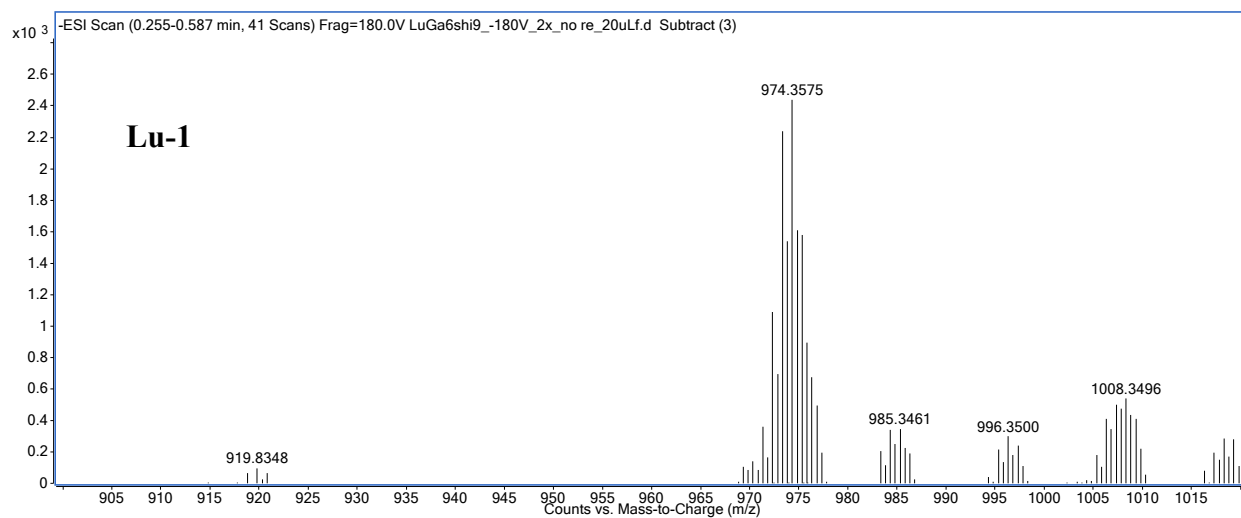




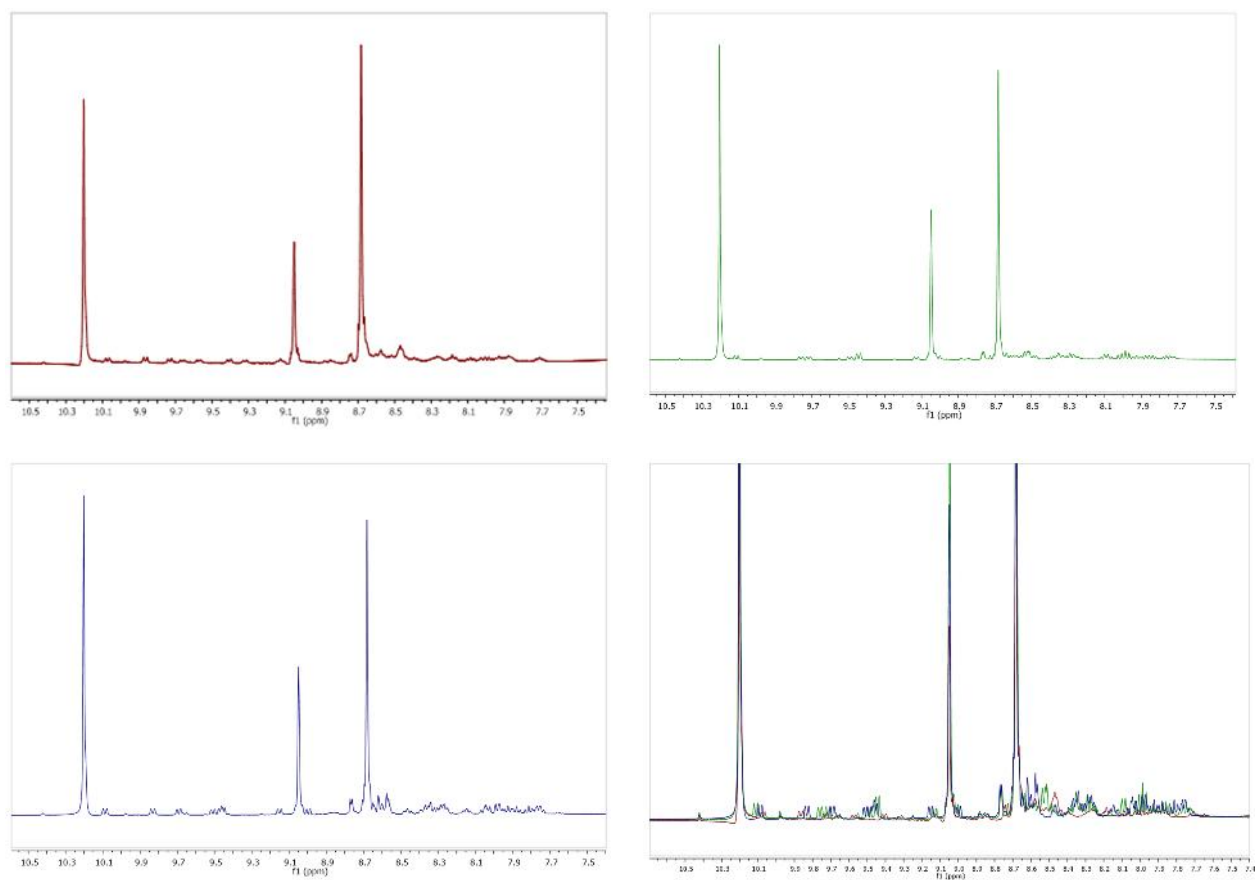




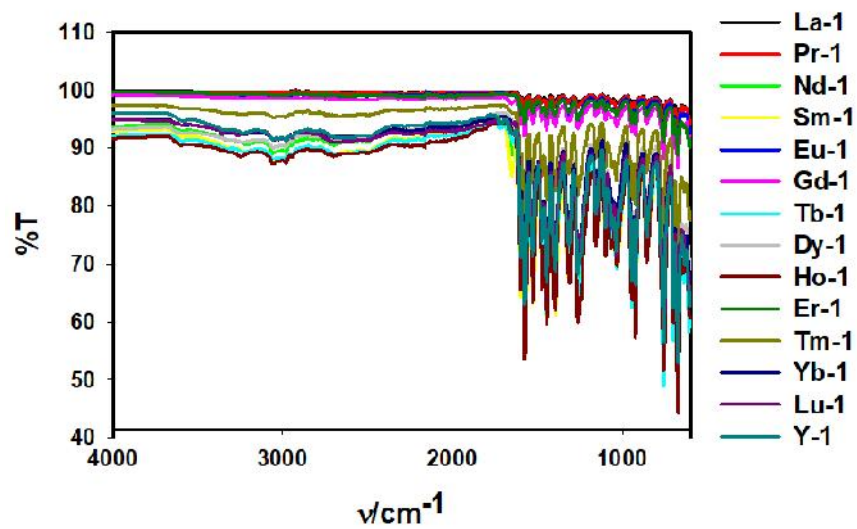




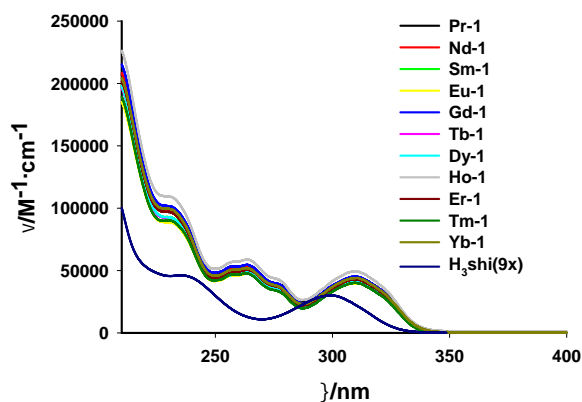
**Figure S3.** ESI-QTOF mass-spectra of **Ln-1** complexes and simulated spectra of each analog. Spectra collected in negative ion mode with fragmentation voltage of 180 V. Background spectra were subtracted three times.



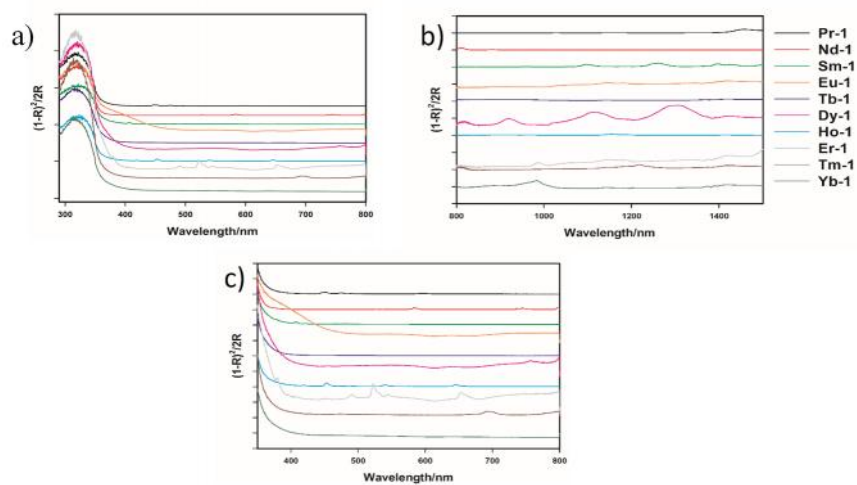
**Figure S4.**  $^1\text{H-NMR}$  of a) **La-1**, b) **Y-1**, c) **Lu-1** and d) overlay of all three in  $d_5$ -pyridine at RT.



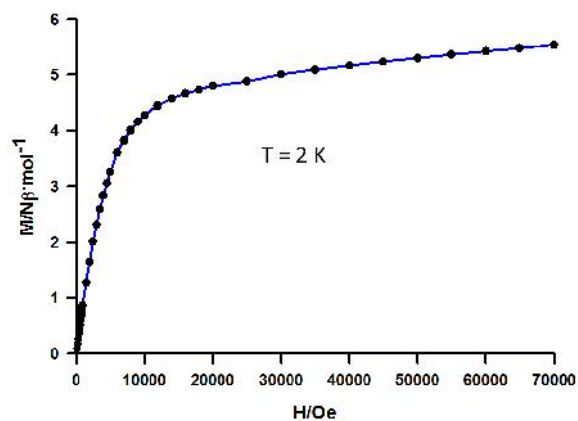
**Figure S5.** FT-IR of **Ln-1** complexes.



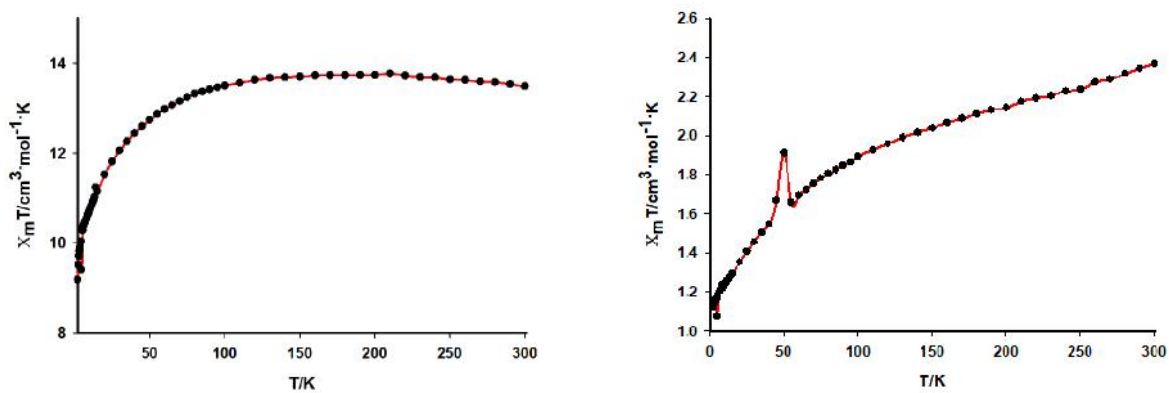
**Figure S6.** UV-vis absorption spectra of 1-10  $\mu\text{M}$  **Ln-1** metallacryptates and  $\text{H}_3\text{shi}$  ligand (multiplied by 9 to match the number of ligands present in the complex) in methanol solution at room temperature.



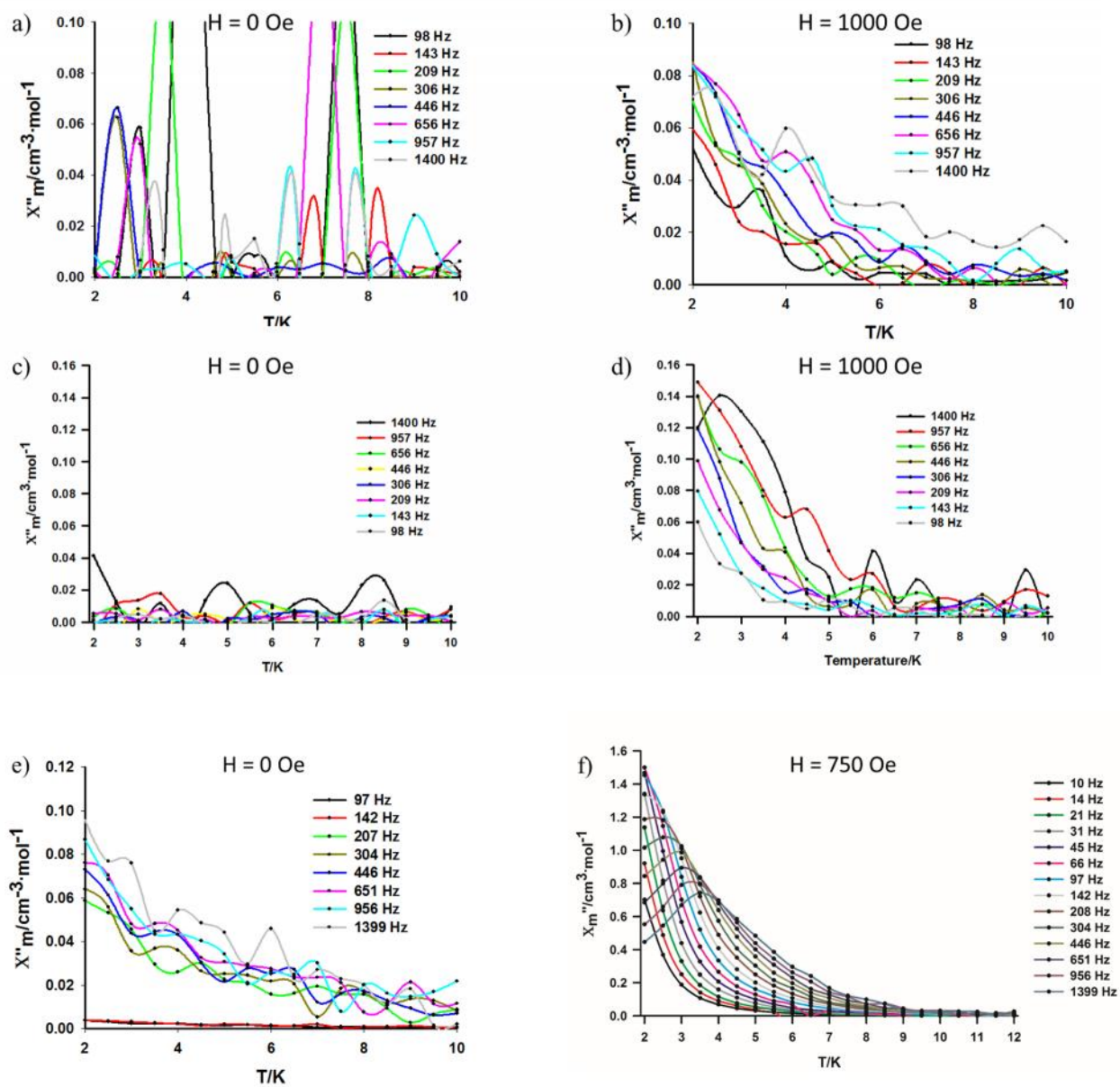
**Figure S7.** Solid state diffuse reflectance spectra of **Ln-1** complexes; a) UV-vis region; b) near-infrared region; c) zoom in on f-f transitions in the visible. Trace colors are as follows: black = Pr, red = Nd, green = Sm, orange = Eu, blue = Tb, pink = Dy, aqua = Ho, gray = Ho, brown = Tm, dark green = Yb.



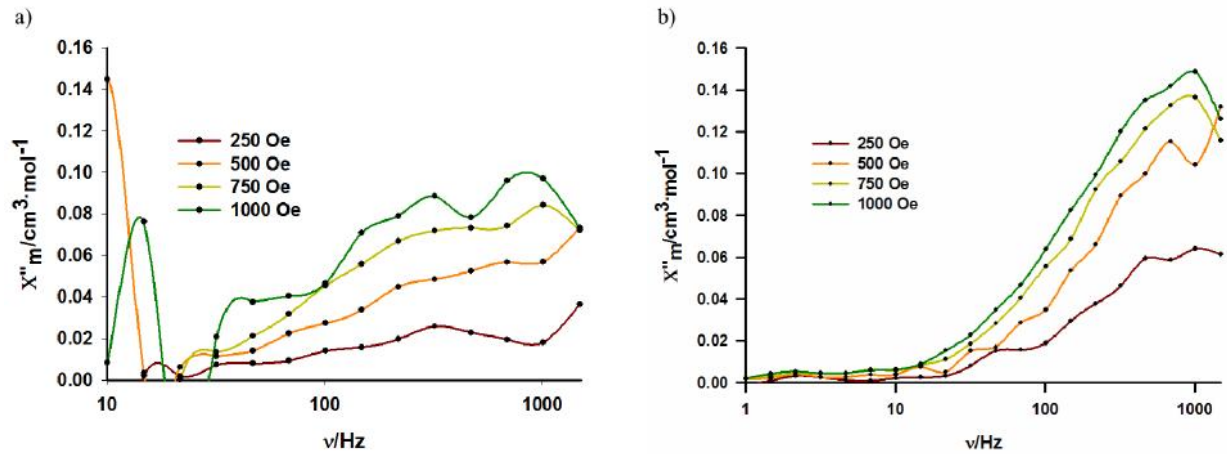
**Figure S8.** Isothermal magnetization of **Dy-1** at 2K. The blue line is a guide for the eye.



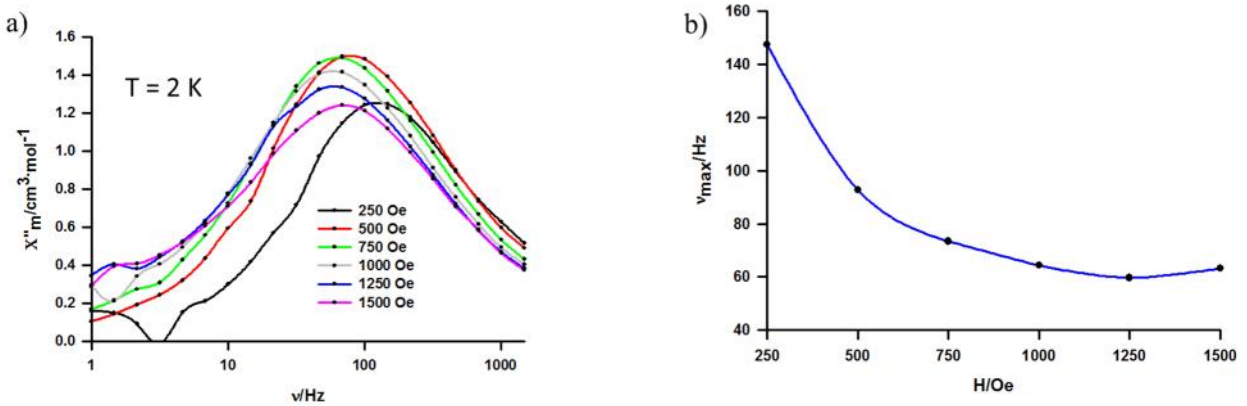
**Figure S9.** DC magnetic susceptibility of **Dy-1** (top left), **Nd-1** (top right), and **Yb-1** (bottom) in a 2000 Oe applied field. Red lines are a guide for the eye.



**Figure S11.** AC out of phase susceptibility measurements of **Ln-1** using a 3 Oe drive field. a) **Nd-1** with zero applied field and b) applied field of 1000 Oe; c) **Yb-1** with zero applied field and d) applied field of 1000 Oe; e) **Dy-1** with zero applied field and f) applied field of 750 Oe.



**Figure S12.** Variable frequency AC out of phase behavior of a) **Nd-1** and b) **Yb-1** in various applied DC fields at 2K.



**Figure S13.** a) Variable frequency AC out of phase behavior of **Dy-1** in various applied fields at 2 K; b) plot of  $\nu_{\max}$  minimization as a function of applied field  $H$ .  $\nu_{\max}$  was determined from fitting a logarithmic peak function  $\chi''_m = a \cdot e^{(-0.5 \cdot (\ln(\nu/\nu_{\max})/2)^2)}$ . 750 Oe is chosen since this field showed a low  $\nu_{\max}$  and still had a high  $\chi''_m$  signal.

**Table S1.** Comparison of photophysical parameters of Gd<sup>3+</sup>/Ln<sup>3+</sup> MCs([Ln[12-MC-4)])<sup>a</sup> and metallacryptates **Ln-1** (Ln[3.3.1])

Lanthanide Complex	$\epsilon/\text{M}^{-1}\cdot\text{cm}^{-1b}$	$\phi_{\text{em}}^{\text{vis}}/\%$	$\tau_{\text{obs}} / \mu\text{s}$
Sm[12-MC-4]	21910	2.91(8) <sup>c</sup>	148(1)
Sm[3.3.1]	42184	1.70(9) <sup>c</sup>	70(1)
Tb[12-MC-4]	22517	34.7(1)	1080(10)
Tb[3.3.1]	40733	0.189(3)	20.7(5) : 71% 4.54(6) : 29%
Ho[12-MC-4]	23246	$2.0(2)\cdot 10^{-3}$	0.029(1)
Ho[3.3.1]	49267	$1.1(2)\cdot 10^{-3}$	0.037(1)
Er[12-MC-4]	20133	0.044(1)	6.75(3)
Er[3.3.1]	42879	$7.1(2)\cdot 10^{-3}$	0.905(8)
Yb[12-MC-4]	21934	5.88(2)	55.7(3)
Yb[3.3.1]	43975	0.65(3)	7.26(2)

<sup>a</sup> Taken from Ref.<sup>1</sup>

<sup>b</sup> Molar absorption coefficients are given at 310nm for both Ln[12-MC-4] and Ln[3.3.1].

<sup>c</sup> Total quantum yield of visible and NIR emissions.

**Table S2.** Cole-Cole fitting for the parameter  $\alpha$ .

T/K	$\alpha (X_m')$	$\alpha (X_m'')$
2.0	0.2247	0.2375
2.5	0.2041	0.2316
3.0	0.2265	0.2393
3.5	0.2661	0.2634
4.0	0.2790	0.2823



**Table S3. Crystallographic Data for Tb[3.3.1]**

Chemical Formula	TbGa <sub>6</sub> C <sub>93.5</sub> H <sub>99.5</sub> N <sub>14.5</sub> O <sub>27</sub>
Formula Weight	2435.61 g/mol
Crystal System, Space Group	Triclinic, $P\bar{1}$ (No. 2)
T	85(2) K
a	16.4342(3) Å
b	17.2269(3) Å
c	21.6881(4) Å
$\alpha$	75.6312(16) <sup>o</sup>
	70.7189(18) <sup>o</sup>
	89.4616(14) <sup>o</sup>
Volume	5596.7(2) Å <sup>3</sup>
$\lambda$	1.54184 Å
calc	1.445 mg/m <sup>3</sup>
Z	2
$\mu$	5.254 mm <sup>-1</sup>
F(000)	2458
range	2.235 <sup>o</sup> to 73.947 <sup>o</sup>
Limiting Indices	-20 < h < 20
	-21 < k < 21
	-26 < l < 24
Reflections collected/unique	179289/22055
Completeness to	98.7%
No. of Data/Restraints/Params	22055/636/1493
GooF on F <sup>2</sup>	1.051
<sup>a</sup> R <sub>1</sub>	0.0472 [I>2 (I)]; 0.0485 (all data)
<sup>b</sup> wR <sub>2</sub>	0.1406 [I>2 (I)]; 0.1426 (all data)
Largest Diff. Peak, Hole	1.422 e/Å <sup>3</sup> ; -1.687 e/Å <sup>3</sup>
<sup>a</sup> $R_1 = \frac{(\sum  F_o  - \sum  F_c )}{\sum  F_o }$ ; <sup>b</sup> $wR_2 = \frac{[\sum (F_o^2 - F_c^2)^2]}{[\sum (F_o^2)]^{1/2}}$ ; $w = 1/[\sum (F_o^2) + (mp)^2 + np]$ ; $p = [\max(F_o^2, 0) + 2F_c^2]/3$ ( $m$ and $n$ are constants); $\sigma = \frac{[\sum (F_o^2 - F_c^2)^2 / (n - p)]^{1/2}}$	

## REFERENCES

- (1) Chow, C. Y.; Eliseeva, S. V; Trivedi, E. R.; Nguyen, T. N.; Kampf, J. W.; Petoud, S.; Pecoraro, V. L. *J. Am. Chem. Soc.* **2016**, *138* (15), 5100–5109.