STRUCTURE OF THE DEXTRAN OF Leuconostoc mesenteroides B-1355*

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(Received February 3rd, 1978; accepted for publication in revised form, January 15th, 1980)

ABSTRACT

The structure of dextran B-1355-S (soluble fraction) has been investigated. Acid hydrolysis of the methylated dextran yielded 2,3,4,6-tetra-, 2,3,4-tri-, 2,4,6-tri-, and 2,4-di-O-methyl-D-glucose, in the molar ratios of 1.0:4.6:3.6:1.0, indicating that the dextran has a branched structure containing (1 \rightarrow 6)- and (1 \rightarrow 3)- α -D-glucosidic linkages with an average repeating-unit of ten sugar residues. The dextran-polyalcohol, derived by successive periodate oxidation and borohydride reduction, gave, on complete hydrolysis with acid, glycerol and glucose (molar ratio, 1.0:0.83); mild hydrolysis (0.1 α acid at 25°) gave glycerol and 1- α -D-glucosylglycerol (molar ratio, 1:1.45).

Exo- G_2 -dextranase of Arthrobacter globiformis T6 acted on the dextran to release isomaltose and O- α -D-glucopyranosyl- $(1\rightarrow 3)$ -O- α -D-glucopyranosyl- $(1\rightarrow 6)$ -D-glucose in the molar ratio of 5.7:1.0, leaving a degraded dextran (limit dextran). Methylation analysis indicated the limit-dextran to be highly branched, with a repeating unit of six glucose residues, consisting mainly of alternate $(1\rightarrow 6)$ - and $(1\rightarrow 3)$ - α -D-glucosidic linkages.

These findings confirm the overall arrangement of alternating $(1\rightarrow 6)$ - and $(1\rightarrow 3)$ -linked α -D-glucose residues, and the absence of consecutive α - $(1\rightarrow 3)$ -linked sugar residues; the latter factor may be related to the water insolubility of α -D-glucans of *Streptococcus mutans*.

^{*}A part of this work was presented at the Annual Meeting of the Japanese Biochemical Society, Fukuoka, October 13–16, 1975, and at the 8th International Symposium on Carbohydrate Chemistry, Kyoto, August 16–18, 1976.

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INTRODUCTION

Most dextrans elaborated by Leuconostoc mesenteroides and related microorganisms contain a substantial number of $(1\rightarrow6)-\alpha$ -D-glucosidic bonds, together with various proportions of $(1\rightarrow2)$ -, $(1\rightarrow3)$ -, and $(1\rightarrow4)-\alpha$ -D-glucosidic linkages (see the extensive, recent review by Sidebotham¹). These non- $(1\rightarrow6)$ linkages usually occur as the origin of branch points, as in the case of dextran² B-512 and dextran^{3,4} B-1397. However, certain dextrans, such as the difficulty water-soluble fractions of dextran B-1299, have been postulated to contain consecutive $(1\rightarrow3)-\alpha$ -D-glucosidic linkages^{5,6}. Considerable attention has recently focused on the physical properties of water-insoluble α -D-glucans of cariogenic oral streptococci. Our recent study established that the insoluble glucan-fraction of the α -D-glucan of Streptococcus mutans OMZ 176 consists of an α - $(1\rightarrow3)$ -linked backbone chain, carrying short $(1\rightarrow6)$ -linked side-chains⁷.

During investigations on the relationship between the structure and the physical and immunochemical properties of dextrans and related α -D-glucans, we became interested in the water-soluble fraction of dextran B-1355 (dextran B-1355-S), which was previously reported to contain a high proportion of $(1\rightarrow 3)$ -linked D-glucose residues, namely, 35% as shown by a periodate-oxidation study⁸, and 47% by Smith degradation⁹. In contrast to the insoluble glucans of *Streptococcus mutans*, however, dextran B-1355-S is readily soluble in water. From the acetolysis product of this dextran, Goldstein and Whelan isolated nigerose (20%) and 3-O-isomaltosyl-D-glucose (2%), in addition to glucose and isomaltose, and suggested the presence of alternate α -(1 \rightarrow 3) and (1 \rightarrow 6) linkages in the same molecule¹⁰.

This paper reports the detailed structural features of dextran B-1355-S with special reference to the sequences of $(1\rightarrow 3)$ - and $(1\rightarrow 6)$ -linked α -D-glucose residues, as revealed by chemical investigations and enzymic-degradation studies.

A recent paper by Seymour *et al.*¹¹ described the methylation data for dextran B-1355-S as well as several other dextrans.

RESULTS AND DISCUSSION

Dextran B-1355-S ($[\alpha]_D$ +220°) was methylated twice by the method of Hakomori¹². The permethylated dextran was hydrolyzed with acid, and the methylated-sugar fragments were analyzed by g.l.c.-m.s. as their alditol acetates, and also as their methyl glycosides. The identities and molar ratios of the cleavage fragments are listed in Table I, showing the presence of 2,3,4,6-tetra-, 2,3,4-tri-, 2,4,6-tri-, and 2,4-di-O-methyl-D-glucose, in the molar ratios of 1.0:4.6:3.6:1.0. Although these data are very close to those obtained by Seymour *et al.*¹¹, our results are based on a more-completely methylated dextran, compared with the previous data which appeared to involve a slightly under-methylated product. Our results indicate that dextran B-1355-S possesses a branched structure having an average repeating-unit of approximately ten glucose residues, of which 3 or 4 glucose residues are joined by

TABLE I MOLAR RATIO OF METHYLATED D-GLUCOSE FRAGMENTS FROM HYDROLYZATES OF METHYLATED DEXTRAN B-1355-S, DEXTRAN-POLYALCOHOL, AND G_2 -DEXTRANASE LIMIT-DEXTRIN

O-Methyl-D-glucose	Native dextran	Dextran– polyalcohol	G2-Dextranase limit-dextran
2,3,4,6-Tetra-	1.0	(trace)	1.0
2,3,4-Tri-	4.6		2.4
2,4,6-Tri-	3.6	3.64	1.47
2,4-Di-	1.0	1.0	1.0

 α -(1 \rightarrow 3), and 4 or 5 other residues are joined by α -(1 \rightarrow 6) linkages, respectively; branching occurs by substitution at O-3 and O-6 of the D-glucosyl residues.

The methylation data suggest that the gross structure of dextran B-1355-S resembles that of the insoluble glucan of *Streptococcus mutans* OMZ 176, which also has a branched structure (average repeating-unit, seven sugar residues) consisting of $(1\rightarrow 3)$ - and $(1\rightarrow 6)$ - α -D-glucosidic linkages in the ratio ⁷ of 5:2.

In order to obtain further information on the structure of dextran B-1355-S, the polysaccharide was oxidized with periodate at 20°. After complete oxidation (7 days; periodate consumption 1.15 mol, and formic acid production 0.61 mol per glucose residue), the oxidized dextran was reduced with sodium borohydride to afford the dextran-polyalcohol. Methylation of the dextran-polyalcohol followed by acid hydrolysis yielded 2,4,6-tri- and 2,4-di-O-methylglucose in the molar ratio of 3.64:1.0, and a trace of 2,3,4,6-tetra-O-methylglucose (Table I). This result confirms that all (1→6)-linked D-glucose residues and non-reducing terminal groups were completely cleaved by periodate. When the dextran-polyalcohol was hydrolyzed by heating it with 0.25M sulfuric acid (complete Smith-degradation)¹³, glycerol and glucose were formed in the molar ratio of 1.0:0.83, as estimated by g.l.c. of the acetylated polyalcohols. This result is in good agreement with the methylation data.

Unambiguous evidence for alternating $(1\rightarrow 3)$ - and $(1\rightarrow 6)$ - α -D-glucosidic linkages was provided by examination of the products from mild acid hydrolysis of the dextran-polyalcohol (mild Smith-degradation)¹⁴. When the polyalcohol was hydrolyzed with 0.05M sulfuric acid at 25°, glycerol and 1-O- α -D-glucosylglycerol ($[\alpha]_D + 134.8^\circ$) were produced, the latter glucoside being identified as the crystalline acetate, and also by g.l.c. as its O-trimethylsilyl derivative. The molar ratio of 1-O- α -D-glucosylglycerol to glycerol was estimated to be 1.45:1.0. As the glucosylglycerol was derived from the sequence $\rightarrow 6$)- α -D-Glcp- $(1\rightarrow 3)$ - α -D-Glcp- $(1\rightarrow 6)$ -D-Glcp- $(1\rightarrow$

TABLE II
effects of α -d-glucan hydrolases on dextran B-1355-S and OMZ 176 mutan

Enzyme	pH	Hydrolysis Percentagea	
		Dextran B-1355-S	OMZ 176 Mutan
Penicillum sp. dextranase	6	7.3 ^b	11.05
G ₂ -Dextranase	4.5	22.2°	5.0°
Exo-(1→6)-D-glucosidase	5.5	3.5 ^b	
(1→3)-α-D-Glucan hydrolase	6.3	0.5 ^b	11.3 ^b

^aDetermined by reducing value after incubation for 24 h at 37°. ^bExpressed as glucose. ^cExpressed as isomaltose.

B-1355-S, unlike the insoluble glucan of Streptococcus mutans OMZ176, consists mainly of alternating $(1\rightarrow 3)$ - and $(1\rightarrow 6)$ - α -D-glucosidic linkages.

The action of various α-D-glucan hydrolases on dextran B-1355-S and the glucan of Streptococcus mutans OMZ 176 (abbreviated as OMZ 176 mutan) was compared. As shown in Table II, a dextranase preparation of *Penicillum* sp. (commercial product) hydrolyzed both polysaccharides to similar extents, with release of glucose, isomaltose, and other oligosaccharides. $(1\rightarrow 3)-\alpha$ -D-Glucan hydrolase (EK-14 enzyme), purified from the culture fluid of Flavobacterium sp. 15, hydrolyzed OMZ 176 mutan, with the release of nigerose and several nigerosaccharides, as reported previously¹⁵, but it failed to attack dextran B-1355-S. On the other hand, the isomalto(G₂)-dextranase of Arthrobacter globiformis T6 (ref. 16), which hydrolyzes $(1\rightarrow 6)-\alpha$ -D-glucosidic linkages in dextrans from the nonreducing termini with the release of isomaltose groups¹⁷, scarcely acted on OMZ 176 mutan, whereas it hydrolyzed dextran B-1355-S to a marked extent (apparent hydrolysis, 38.2%). However, when dextran B-1355-S was incubated with an exo- $(1\rightarrow 6)$ - α -D-glucosidase of Arthrobacter globiformis¹⁸, only 3.5% of glucose was liberated. These results clearly indicate an essential difference between dextran B-1355-S and OMZ 176 mutan in the arrangements of D-glucosidic linkages.

For characterization of the products obtained from the G_2 -dextranase digestion of dextran B-1355-S, the enzyme digest was subjected to gel filtration on a column of Sephadex G-25, whereby two distinct fractions, one of high molecular weight (F-1) and one of low molecular weight (F-2) were obtained (Fig. 1). Paper chromatography (6:1:3, 1-propanol-acetic acid-water) revealed that Fraction 2 contained only isomaltose and a trisaccharide having $R_{isomaltose}$ 0.83. The trisaccharide, which showed $[\alpha]_D + 130^\circ$, was separated from isomaltose on a thick filter-paper sheet. Methylation and acid-hydrolysis of the purified trisaccharide gave equal proportions of 2,3,4,6-tetra-, 2,4,6-tri-, and 2,3,4-tri-O-methyl-D-glucose, as identified by g.l.c. of their alditol acetates. Incubation of the trisaccharide with yeast α -D-glucosidase gave glucose and isomaltose, as detected by paper chromatography. From the

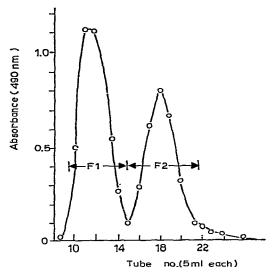


Fig. 1. Gel-filtration profile of the G2-dextranase digest of dextran B-1355-S on Sephadex G-25.

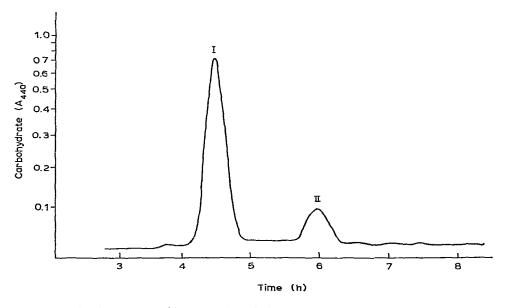


Fig. 2. Liquid-chromatographic separation of oligosaccharides obtained by G_2 -dextranase digestion of dextran B-1355-S. I, Isomaltose; II, 3-O- α -D-glucosylisomaltose. An arrow indicates the position of 6-O- α -D-glucosylnigerose.

methylation and enzymic-degradation data, the trisaccharide was identified as $O-\alpha-D$ -glucosyl- $(1\rightarrow 3)-O-\alpha-D$ -glucosyl- $(1\rightarrow 6)-D$ -glucose (or $3-O-\alpha-D$ -glucosylisomaltose). The isomaltose and the aforementioned trisaccharide in Fraction 2 were also analyzed by automated liquid-chromatography with a column of anion-exchange



Fig. 3. Ramified structural model of dextran 1355-S, and the hydrolytic cleavages of external chains by G_2 -dextranase. Glc: α -D-glucopyranose residue: cleavage point by the action of enzyme.

resin (Jeol LCR-3) and stepwise elution with borate buffers¹⁹. The first peak (Fig. 2) corresponds to isomaltose and the second to 3-O- α -D-glucosylisomaltose, and not to 6-O- α -D-glucosylingerose. From their peak areas, the molar ratio of isomaltose to 3-O- α -D-glucosylisomaltose was estimated to be 5.6:1.

The formation of isomaltose and 3-O- α -D-glucosylisomaltose is consistent with the sequences of $(1\rightarrow 3)$ - and $(1\rightarrow 6)$ -linked D-glucose residues in the dextran molecule. As reported previously¹⁶, in addition to the stepwise liberation of isomaltose from consecutive α - $(1\rightarrow 6)$ -linked D-glucose residues, the G_2 -dextranase also hydrolyzes the tetrasaccharide having the sequence α -D-Glc- $(1\rightarrow 3)$ - α -D-Glc- $(1\rightarrow 6)$ -D-Glc, to give glucose and the trisaccharide α -D-Glc- $(1\rightarrow 3)$ - α -D-Glc- $(1\rightarrow 6)$ -D-Glc. If it is assumed that most D-glucose residues in dextran B-1355-S are joined by alternating α - $(1\rightarrow 3)$ and α - $(1\rightarrow 6)$ linkages, 3-O- α -D-glucosylisomaltose could arise only from the presence of non-reducing, terminal α - $(1\rightarrow 6)$ linkages, and also from the terminal α - $(1\rightarrow 6)$ linkages created by the exo-action of G_2 -dextranase (see Fig. 3) on the dextran.

The liberation of isomaltose may proceed until the enzymic cleavage ceases near the branch points, probably leaving two or three glucose residues. Thus, the formation of isomaltose and $3-O-\alpha-D$ -glucosylisomaltose from alternate $(1\rightarrow 3)$ - and $(1\rightarrow 6)$ -linked α -D-glucose residues may be explained by assuming that the enzyme acts at the reducing ends of alternate 6-substituted α -D-glucose groups, in a fashion similar to the hydrolysis of lichenan by cellulase from certain *Streptomyces* sp.²⁰.

The high-molecular-weight fraction (F-1 in Fig. 1), which may be regarded as the limit-dextran, was subjected to methylation analysis, and the molar ratio of the methylated glucose components was compared with that from the native dextran (see Table I). The methylation data indicate that the limit-dextran has a highly branched structure, with an average repeating-unit of six glucose-residues, consisting of non-terminal $(1\rightarrow6)$ - (2.4 parts) and $(1\rightarrow3)$ -linked α -D-glucose residues (1.47 parts), together with a non-reducing terminal and branch-point residue. This result confirms that the alternating $(1\rightarrow6)$ - and $(1\rightarrow3)$ -linked sugar residues are distributed uniformly in the entire dextran molecule.

On the basis of methylation and acetolysis data, Seymour et al.¹¹ recently proposed two types of repeating-unit structures for dextran B-1355-S: one consisting

of an α -(1 \rightarrow 6)-linked backbone chain with side chains of alternating α -(1 \rightarrow 3) and (1 \rightarrow 6) linkages, and the second consisting of a backbone chain of alternating α -(1 \rightarrow 3)- and (1 \rightarrow 6)-linked units with single α -D-glucosyl stubs attached by (1 \rightarrow 6)-bonds. Our experimental data do not support either of these structures.

The results of our methylation studies indicate that the ratio of internal $(1 \rightarrow 6)$ to $(1 \rightarrow 3)$ linkages is 1.3:1 in the native dextran, and 1.6:1 in the G_2 -dextranase limit-dextran. These data suggest that both $(1 \rightarrow 6)$ and $(1 \rightarrow 3)$ linkages are uniformly distributed.

As regards the action of G_2 -dextranase on the 1355-S dextran, the methylation data strongly suggest that the enzyme probably does not attack the branch points or bypass them, but rather results in the stepwise hydrolysis of $(1\rightarrow 6)$ linkages, which ceases at a certain distance from the branch points¹⁶. The action of the G_2 -dextranase results in the release of an average of four out of 10 glucose residues of the repeatingunit chain.

If the dextran molecule consists of a regular sequence of $(1\rightarrow 6)$ - and $(1\rightarrow 3)$ - α -D-glucosidic linkages, and if the dextran is assumed to possess a ramified, multiply-branched structure, our experimental findings lead to the structural model shown in Fig. 3.

This structure is supported by our recent studies on the isolation of oligo-saccharides (by acetolysis and partial acid hydrolysis) from dextran 1355-S (*Torii et al.*, unpublished). Among the acetolysis products isolated were tri- and tetra-saccharides having the following sequence: α -Glcp- $(1\rightarrow 3)$ - α -Glcp- $(1\rightarrow 6)$ -Glc, α -Glcp- $(1\rightarrow 6)$ -Glc, α -Glcp- $(1\rightarrow 6)$ -Glc, and α -Glcp- $(1\rightarrow 6)$ - α -Glcp- $(1\rightarrow 6)$ - α -Glcp- $(1\rightarrow 3)$ -Glc. A small proportion of nigerotriose was also obtained. Upon partial acid hydrolysis, the dextran gave, in addition to isomaltose, small amounts of isomaltotetraose. These oligosaccharides could have arisen from the sequences of alternate α - $(1\rightarrow 6)$ - and α - $(1\rightarrow 3)$ -linked residues and the sequence of the residues around the branch points in the structure of dextran 1355-S (See Fig. 3).

In view of the physical properties of dextran-like α -D-glucans that contain relatively high proportions of α -(1 \rightarrow 3)-D-glucosidic linkages, it is apparent that the arrangements of (1 \rightarrow 3)- and (1 \rightarrow 6)-linked sugar residues must play an important role in their water solubility. Thus, as already established, the water insolubility of OMZ 176 mutan may be attributed to the long stretches of consecutive α -(1 \rightarrow 3)-linked D-glucose residues in the backbone chain, while short α -(1 \rightarrow 6)-linked sidechains may be related to the adhesive properties 7. The present experimental findings strongly suggest that the water solubility of dextran B-1355-S, which contains a high proportion of α -(1 \rightarrow 3) linkages, may be attributed to the alternating (1 \rightarrow 3)-and (1 \rightarrow 6)-D-glucosidic linkages, which may interfere with hydrogen bonding between adjacent chains. In this connection, it is noteworthy that the water-soluble glucan fraction synthesized by *Streptococcus mutans* OMZ 176 contains mainly (1 \rightarrow 6)-linked α -D-glucose residues similar to those found in most dextrans (Misaki *et al.*, unpublished results).

The interaction of concanavalin A with dextrans has been studied extensively,

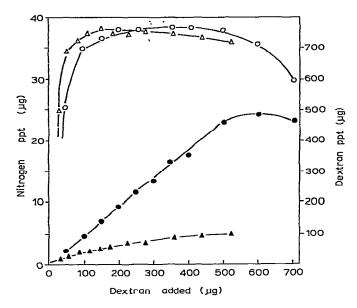


Fig. 4. Quantitative precipitation curves of dextran B-1355-S and G_2 -dextranase limit-dextrin with concanavalin A. $\bigcirc\bigcirc$ —, protein precipitated with native dextran: $\bigcirc\triangle$ —, protein precipitated with G_2 -dextranase limit-dextran; — \bullet —, dextran in the precipitates of concanavalin A-dextran; — \bullet —, G_2 -dextranase limit-dextran in the precipitates of concanavalin A- G_2 -dextranase limit-dextran.

and dextran B-1355-S has been used as the standard polysaccharide for the precipitation assay because of its great reactivity and the low solubility of the dextran-lectin complex. Inasmuch as interaction with concanavalin A principally involves terminal α -D-glucosyl groups of dextran molecules, the G_2 -dextranase limit-dextran was compared with the native dextran in its reactivity with concanavalin A. As may be seen in Fig. 4, maximum precipitation of the protein with the limit-dextran occurs at one-half to one-third the amount of that of the native dextran.

This result may be due to the result of shortening of the chains by the action of G_2 -dextranase, so that the number of the non-reducing terminal-ends per unit weight increases. It is also noteworthy that the amount of the limit-dextran reacting with concanavalin A was found to be much less, approximately one-fourth that of the native dextran.

With regard to the immunochemical properties of dextran B-1355-S, the structural model proposed here can explain the previous observations that dextran B-1355-S reacts with human antidextran sera having specificity for both $(1\rightarrow 3)$ -and $(1\rightarrow 6)-\alpha$ -D-glucosidic linkages. When rabbit antiserum against *L. mesenteroides* B-1355 cells was fractionated on a column of Sephadex G-75, two immunochemically active fractions, one passing through, and the second binding to Sephadex, were obtained: the inhibition tests in both antibody-systems indicated that nigerose was a more potent inhibitor than isomaltose and other isomaltosaccharides²³. In the precipitation system with the Sephadex-binding fraction, the inhibitory powers of

isomaltosaccharides appeared to increase with their degree of polymerization, up to isomalto-tetraose or -pentaose, whereas in the system with the Sephadex non-binding fraction, all isomaltosaccharides showed the same inhibitory activity. The inhibitory activities of isomaltosaccharides, in the system with the Sephadex-binding fraction, may be related to the presence of small proportions of contiguous $(1\rightarrow 6)$ -linked α-D-glucose residues in the dextran, as suggested by the detection of small amounts of isomalto-triose and -tetraose among the products of mild acid hydrolysis of the dextran. It should be pointed out, with regard to the complexity of the antigenic determinants of dextran B-1355-S, that among eight different oligosaccharides, including a series of isomaltosaccharides and those containing both α -(1 \rightarrow 6) and α -(1 \rightarrow 3) linkages, 6-O- α -D-glucosylnigerose was shown to be the best inhibitor in the precipitation system with the Sephadex-binding fraction; on the other hand, 3-Oα-D-glucosylisomaltose was the best inhibitor in the system with the fraction that passed through the Sephadex column (Torii et al., unpublished results). Further studies on the relationship between immunochemical specificities and the structure of dextran B-1335-S are in progress.

EXPERIMENTAL

Materials. — Dextran (soluble fraction) of Leuconostoc mesenteroides NRRL dextran B-1355-S, used in this study, was generously donated by Dr. A. Jeanes, Northern Regional Research Laboratory, Peoria, IL, U.S.A. Exoisomalto- (G_2) -dextranase¹⁶ and exo- $(1\rightarrow6)$ - α -D-glucosidase¹⁸ were purified as reported previously. An endo-dextranase preparation from Penicillum sp. was purchased from Sigma Chem. Co. $(1\rightarrow3)$ - α -D-Glucan hydrolase of Flavobacterium was prepared by the method reported in a previous paper¹⁵.

General methods. — Paper chromatography was performed on Whatman No. 1 or Toyoroshi No. 51A paper by the descending method using the following solvent systems (v/v): (A) 4:1:5 1-butanol-acetic acid-water, (B) 6:4:3 1-butanol-pyridinewater, and (C) 6:1:3 1-propanol-ethanol-water. Preparative paper chromatography was performed with Whatman 3MM paper. Reducing sugars were detected by spraying the chromatograms with p-anisidine hydrochloride in water-saturated butanol, and alditols and glycosides of glucose were detected with Tollens' alkaline silver nitrate reagent. G.l.c. was conducted with a Shimazdu Gas Chromatograph Model CG-6AM equipped with a flame-ionization detector, using a glass column (0.4 cm i.d.). Methylated sugars were converted into their corresponding additol acetates, and separated at 180° on a column packed with 3% of ECNSS-M on Gas-Chrom Q. G.l.c.-m.s. was performed on alditol acetates with a Shimadzu-LKB gas chromatograph-mass spectrometer. For g.l.c. separation of the products from complete Smith-degradation, the hydrolysis products were treated with sodium borohydride and then acetylated. The products from the mild Smith-degradation were analyzed as their trimethylsilylated derivatives by g.l.c. on a column of OV-17, programmed from 80 to 320° (6° per min).

Automated liquid-chromatography was performed with a Jeol liquid chromatograph Model JLC-3BC (Japan Electronic Optical Laboratory, Tokyo) using a column of anion-exchange resin (Jeol LCR-3). The oligosaccharides were separated¹⁹ by stepwise elution with 0.13m (pH 7.5), 0.25m (pH 9.0), and 0.35m (pH 9.6) borate buffer, at 65°.

Total carbohydrate content was usually determined colorimetrically by the phenol-sulfuric acid method²⁴.

Methylation analysis. — Polysaccharides were methylated according to the method of Hakomori¹². Dry samples (20–30 mg each) of the dextran and its limit-dextrin were dissolved in dimethyl sulfoxide (2 mL) by ultrasonication (20 KHz, 5 min) at 25° in a nitrogen atmosphere, and methylated by treatment with methyl-sulfinyl carbanion (1 mL), followed by methyl iodide (2 mL). The mixture was dialyzed, concentrated to low volume, and the methylated product extracted with chloroform. If methylation was incomplete, as judged by i.r. spectroscopy, the methylation procedure was repeated. The methylated polysaccharide (15 mg) was hydrolyzed by heating it with formic acid (0.5 mL) in a sealed tube for 10 h at 100°, the formic acid was distilled off, and the residue was heated further with 0.5M sulfuric acid (0.3 mL) for 3 h at 100°. The hydrolyzate was made neutral with barium carbonate, filtered, and the filtrate evaporated to a syrup. A portion of the hydrolyzate was examined by paper chromatography (solvent A), and the remainder was reduced with sodium borohydride, converted into the acetates by heating with 1:1 pyridine-acetic anhydride (0.2 mL) for 1 h at 100°, and analyzed by g.l.c.

Periodate oxidation and Smith degradation of dextran. — The native dextran (1.5 g) was oxidized with 0.05m sodium periodate (1500 mL) at 5° in the dark; at suitable intervals, the periodate consumption and formic acid production were assayed. After oxidation for 7 days, the excess of periodate was decomposed by the addition of ethylene glycol, and the solution was dialyzed for 24 h at 5°. To the non-dialyzable solution was added sodium borohydride (500 mg in 100 mL of water), and the mixture kept overnight. Excess of borohydride was decomposed by careful addition of acetic acid to pH 7.2. The solution containing dextran-polyalcohol was dialyzed for 2 days at 5°, and evaporated (below 37°) to low volume, from which the dextran-polyalcohol was precipitated by pouring into acetone. The dextran-polyalcohol was collected by centrifugation, washed successively with abs. alcohol and ether, and dried in vacuo (yield, 1.07 g).

Methylation of the foregoing dextran-polyalcohol (50 mg) was achieved by two Hakomori methylations. The methylated dextran-polyalcohol was hydrolyzed by heating with 2M trifluoroacetic acid for 12 h at 100°. After repeated evaporations in the presence of water to remove the acid, the methylated sugar components were analyzed by g.l.c. as their alditol acetates, yielding 2,4,6-tri- and 2,4-di-O-methyl-D-glucose, together with a trace of 2,3,4,6-tetra-O-methyl-D-glucose (see Table I).

The dextran-polyalcohol (10 mg) was hydrolyzed with M sulfuric acid (0.3 mL) for 2 h at 100°, as stated previously²⁵. The hydrolyzate was made neutral with barium carbonate, and the neutral solution was treated with sodium borohydride in order

to convert glucose in the hydrolyzate into glucitol. After removing inorganic salts, the reduced product was acetylated, and analyzed by g.l.c. on a column of 3% ECNSS-M at 65–190°, programmed at the rate of 5° per min. The hydrolysis products were identified as glycerol and glucose, in the molar ratio of 1.0:0.83.

In another experiment, the dextran-polyalcohol (50 mg) was subjected to mild acid hydrolysis with 0.05M sulfuric acid (2 mL) for 18 h at 25°. The hydrolyzate was made neutral with barium carbonate, and the products were examined by paper chromatography (solvent B) which revealed the presence of glycerol and 1-O- α -D-glucosylglycerol ($R_{\rm Gle}$ 0.89). These compounds were separated on a filter-paper sheet with solvent B, and each component was extracted with water. The glucosylglycerol showed $[\alpha]_{\rm D}$ +134.8° (c 1.4, water), and gave, on acid hydrolysis, equal proportions of glycerol and glucose. The glycerol and 1-O- α -D-glucosylglycerol were extracted from the paper and determined colorimetrically by the chromotropic acid²⁶ and phenol-sulfuric acid methods, respectively; their molar ratio was 1:1.45. These components were also identified by g.l.c. on a column of OV-17 after conversion into their trimethylsilylated derivatives.

Degradation of the dextran by exo- G_2 -dextranase and characterization of the products. — The dextran (100 mg) was incubated with exo- G_2 -dextranase (45 units) at pH 4.5 and 37° in a total volume of 14 mL (0.05M acetate buffer) in the presence of a drop of toluene. At suitable intervals, the increase in reducing groups was measured by the method of Nelson and Somogyi (after 2 days, 32.37%; 4 days, 37.8%; and 5 days, 38.2%). After 5 days, the enzyme was inactivated by heating for 3 min at 85°, and the enzyme digest was subjected to gel-filtration on a column (1.5 × 60 cm) of Sephadex G-25, which afforded two carbohydrate-components; Fraction 1 (65 mg) emerging in the void volume, and Fraction 2 (32 mg) corresponding to oligosaccharides.

Paper chromatography (solvent C) showed that Fraction 2 consisted of a mixture of isomaltose (R_{Glc} 0.78) and a trisaccharide (R_{Glc} 0.66). They were separated from each other by preparative paper chromatography with Whatman 3MM paper, giving isomaltose (23 mg) and a trisaccharide (8 mg), which showed $[\alpha]_D + 130^\circ$ (c 1.0, water). A portion (3 mg) of the trisaccharide was dissolved in dimethyl sulfoxide (0.5 mL) and methylated by reaction with methylsulfinyl carbanion (0.5 mL) followed by methyl iodide (1 mL). The methylated trisaccharide in the mixture was extracted with chloroform. After hydrolysis with 2M trifluoroacetic acid for 8 h at 100°, the methylated sugar-fragments from the methylated trisaccharide were examined by g.l.c. as their alditol acetates. The presence of 2,3,4,6-tetra-, 2,3,4-tri-, and 2,4,6-tri-O-methyl-D-glucose, in the molar ratios of 1.0:1.0:0.95, was established. A portion (1 mg) of the trisaccharide was incubated with yeast α -D-glucosidase at pH 5.0 for 15 h. Isolation by paper chromatography revealed the release of glucose and isomaltose. Identification and quantitation of isomaltose and trisaccharide were performed by means of automated liquid-chromatography (see Fig. 2).

The degraded dextran (Fraction 1, 20 mg) was methylated by the method of Hakomori, and the methylated product was hydrolyzed with acid, and analyzed by

g.l.c. as the alditol acetates, giving 2,3,4,6-, tetra-O-, and 2,3,4- and 2,4,6-tri-O-methyl-D-glucose (see Table I).

Quantitative precipitation of the native dextran and its G_2 -dextranase limit dextran with concanavalin A. — Concanavalin A (Calbiochem, 300 μ g) was precipitated with increasing amounts of dextran B-1355-S, and the G_2 -dextranase limit-dextran (Fraction 1) in saline (1 mL) containing 0.018m phosphate buffer, pH 7.2, according to the method of So and Goldstein²⁷. After washing with saline, the precipitates were dissolved in 0.5m acetic acid (3.0 mL) and appropriate aliquots were subjected to nitrogen determination according to Schiffman et al.²⁸. Aliquots were also analyzed for polysaccharide by the phenol-sulfuric acid method.

ACKNOWLEDGMENTS

We are grateful to Dr. A. Jeanes for providing a sample of dextran B-1355-S. A part of this work was supported by N.I.H. Grant AM 10171.

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