

Effect of Oxidation on Mechanical Properties of Fibrous Monolith Si₃N₄/BN at Elevated Temperatures in Air

Young-Hag Koh,*,† Hae-Won Kim,* and Hyoun-Ee Kim*

School of Materials Science and Engineering, Seoul National University, Seoul, 151-742, Korea

John W. Halloran*

Materials Science and Engineering Department, University of Michigan, Ann Arbor, Michigan 48109-2136

The oxidation behavior and its effect on the mechanical properties of fibrous monolith $\mathrm{Si_3N_4/BN}$ after exposure to air at temperatures ranging from 1000° to $1400^\circ\mathrm{C}$ for up to 20 h were investigated. After exposure at $1000^\circ\mathrm{C}$, only the BN cell boundary was oxidized, forming a $\mathrm{B_2O_3}$ liquid phase. With increasing exposure temperature, the $\mathrm{Si_3N_4}$ cells began to oxidize, forming crystalline $\mathrm{Y_2Si_2O_7}$, $\mathrm{SiO_2}$, and silicate glass. However, in this case, a weight loss was observed due to extensive vaporization of the $\mathrm{B_2O_3}$ liquid. After exposure at $1400^\circ\mathrm{C}$, large $\mathrm{Y_2Si_2O_7}$ crystals with a glassy phase formed near the BN cell boundaries. The oxidation behavior significantly affected the mechanical properties of the fibrous monolith. The flexural strength and work-of-fracture decreased with increasing exposure temperature, while the noncatastrophic failure was maintained.

I. Introduction

 $S^{\rm ILICON\ NITRIDE\ }(Si_3N_4)$ based composites have been regarded as one of the most promising materials for high-temperature structural applications. $^{1-4}$ Among these, fibrous monolith Si_3N_4 boron nitride (BN), consisting of a hexagonal arrangement of strong Si_3N_4 cells surrounded by weak BN cell boundaries, has been found to exhibit noncatastrophic failure with high strength both at room temperature and at high temperatures due to extensive crack interactions (crack delaminations and crack deflections) through the BN cell boundaries. $^{4-7}$

As this material was intended for use at high temperatures, the oxidation behavior is one of the more important criteria that need to be clearly understood before it can be applied. The oxidation resistance of $\rm Si_3N_4$ containing sintering aids is strongly influenced by the composition and crystalline state of the secondary phase. $^{8-12}$ However, when BN is exposed to an oxidizing atmosphere above 1000°C , it reacts with oxygen, forming a $\rm B_2O_3$ liquid or a gas phase. 13 Such oxidation behavior strongly affects the mechanical properties of the materials by forming an oxide scale on the surface. 12

In this paper, the oxidation behavior of a fibrous monolith in air at temperatures between 1000° and 1400°C was investigated. The oxidation behavior was measured by monitoring the weight changes of the specimens. The oxidation products that

J. L. Smialek—contributing editor

formed on the surface were identified by X-ray diffraction (XRD), scanning electron microscopy (SEM), and energy dispersive spectroscopy (EDS). In addition, the mechanical properties, such as the strength and work-of-fracture (WOF), were measured using four-point bending tests and related to crack propagations.

II. Experimental Procedure

Fibrous monolith $\mathrm{Si_3N_4/BN}$ was fabricated by hot pressing using a method described elsewhere. Sintered billets were cut into dimensions of 2 mm \times 4 mm \times 10 mm to measure the weight changes, and were machined with a 600-grit diamond wheel, and subsequently polished down to 3 μ m with a resin-bonded diamond wheel. Specimens with dimensions of 3 mm \times 4 mm \times 50 mm were prepared for the mechanical tests using a similar methodology. The tensile surface was polished down to 3 μ m and slightly chamfered to remove the existing defects. In addition, the side surfaces of each specimen were polished down to 30 μ m.

Before oxidation, the surfaces were ultrasonically cleaned in acetone and ethanol. The oxidation test was conducted in a vertical alumina tube-furnace at temperatures ranging from 1000° to 1400°C for 20 h in laboratory air. The furnace was heated at a heating rate of 10°C/min and maintained at the exposure temperatures. The polished specimens, suspended at the end of a platinum wire, were inserted into the hot zone from the top. Such a rapid heating process was selected to minimize oxidation during the heating stage, while a furnace cooling process was used to minimize the thermal stress. The weights of each sample were measured both before and after exposure using a digital balance with an accuracy of 0.1 mg. The oxidized surfaces were examined by SEM, XRD, and EDS. The flexural strength and apparent work-of-fracture were measured using a four-point flexural configuration with inner and outer spans of 20 and 40 mm, respectively, at a crosshead speed of 0.5 mm/min. In addition, crack propagation after the bending test was examined by optical microscopy.

III. Results and Discussion

The oxidation behavior was analyzed by monitoring the weight change of the specimens exposed to air between 1000° and 1400° C for up to 20 h, as shown in Fig. 1. When the specimens were exposed at 1000° C, the weight changes were found to be a function of the exposure time. At an early stage of oxidation, a weight gain was observed, implying that $B_2O_3(l)$ was formed on the cell boundary. However, as the exposure time was increased, a weight loss was observed as a result of $B_2O_3(l)$ vaporization. As the exposure temperature was increased to 1200° C, the $B_2O_3(l)$ vaporization became more

Manuscript No. 187162. Received February 5, 2002; approved September 23, 2002.

^{*}Member, American Ceramic Society.

[†]Now with the University of Michigan.

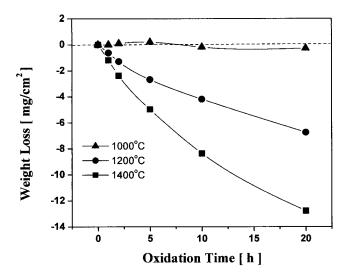
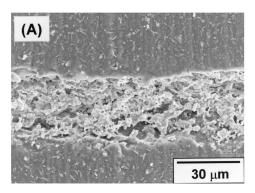


Fig. 1. Weight changes of the samples exposed to air at temperatures between 1000° and 1400° C for up to 20 h.

vigorous, leading to an extensive weight loss. However, at this temperature, the $\mathrm{Si_3N_4}$ cells began to oxidize, forming crystal-line $\mathrm{Y_2Si_2O_7}$, $\mathrm{SiO_2}$ (cristobalite), and a glassy phase. At higher exposure temperatures (1400°C), the weight loss became more pronounced even though the $\mathrm{Si_3N_4}$ was oxidizing to a greater extent. Therefore, the observed weight loss was mainly caused by BN oxidation, indicating that the kinetic constant for BN oxidation is much higher than that of $\mathrm{Si_3N_4}$.

Oxidized surfaces after the different exposure temperatures are shown in Figs. 2(A) and (B). Hot-pressed fibrous monolith consists of $\sim\!250~\mu m$ cells of composition $\mathrm{Si}_3\mathrm{N}_4$ separated by 15–25 $\,\mu m$ boron nitride cell boundaries. After exposure at 1000°C for 20 h, the surface morphology had changed relatively little. That is, the cell boundary was slightly damaged, revealing a $\mathrm{B}_2\mathrm{O}_3$ liquid phase with platelike BN. As the exposure



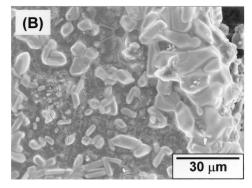


Fig. 2. SEM micrographs of the samples exposed to air for 20 h at (A) 1200° and (B) 1400° C.

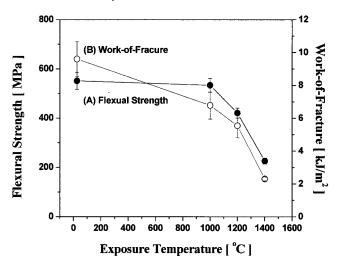
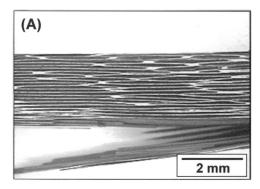


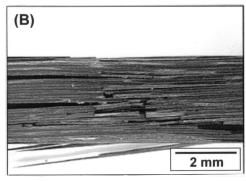
Fig. 3. (A) Flexural strength and (B) work-of-fracture of the sample as a function of the exposure temperature.

temperature was increased to 1200°C, the BN cell boundary was severely damaged because of extensive oxidation, leaving a large amount of a glassy phase without the platelike BN, as shown in Fig. 2(A). In addition, needlelike $Y_2Si_2O_7$ crystals, confirmed by EDS, which were embedded in the glassy phase, formed in the Si_3N_4 cell region. At 1400°C, the Si_3N_4 cells were also severely damaged, revealing an oxide layer composed of large $Y_2Si_2O_7$ crystals and a glassy phase, as shown in Fig. 2(B). The surface cracks were due to crystallization of a glassy phase on cooling. However, the BN cell boundary layers were completely covered by oxidation products.

Before oxidation, the fibrous monolith exhibited noncatastrophic failure due to extensive crack interactions through weak BN cell boundaries. This leads to a high WOF $(9.6 \pm 1.1 \text{ kJ/m}^2)$ and a high strength (551 \pm 34 MPa). Noncatastrophic failure was observed after oxidation, regardless of the exposure temperature. However, with increasing exposure temperature, the maximum apparent strength and crack interactions decreased, which was apparently due to degradation of the cell boundary layers. The flexural strength and WOF of the fibrous monolith after oxidation at the different temperatures are shown in Fig. 3. When the sample was exposed to air at 1000°C for 20 h, the reduction in flexural strength was negligible. However, the WOF decreased because of the reduction in the crack interactions with the cell boundaries. At higher exposure temperatures, both the flexural strength and the WOF decreased. However, even after exposure to air at 1400°C for 20 h, the sample maintained 41% (\sim 226 MPa) and 21% (\sim 2.3 kJ/m²) of its initial strength and WOF, respectively.

The changes in the mechanical properties are related to the interactions of cracks with the weak BN cell boundaries, as shown in Figs. 4(A–C). When the sample was exposed to air at 1000°C, the sample showed extensive crack interactions with the weak BN cell boundaries, which suggest that the surface layer was not severely damaged (Fig. 4 (A)). With further increases in the exposure temperature to 1200°C, the BN cell boundaries were severely damaged because of B₂O₃(l) vaporization. However, extensive crack interactions were also observed (Fig. 4(B)). Therefore, the reduced WOF was mainly due to a reduction in strength retention before the surface fracture. This implies that a high WOF can be achieved when the sample retains a high strength before the first failure with significant crack interactions. At the highest exposure temperature (1400°C), the surface layer was surrounded by a thick oxide layer as a result of Si₃N₄ and BN oxidation, leading to a lower number of crack interactions, as shown in Fig. 4(C). However, significant crack interactions were still observed in the load versus the deflection curve during the bending test because the inner part was not damaged extensively by the exposure.





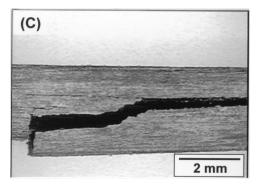


Fig. 4. Optical micrographs of the crack propagations in the sample during the bending tests after exposure to air for 20 h at (A) 1000° , (B) 1200° , and (C) 1400° C.

IV. Summary and Conclusions

The oxidation behavior of fibrous monolith $\mathrm{Si_3N_4/BN}$ was investigated and correlated with the changes in the mechanical properties after exposure to air at temperatures ranging from 1000° to $1400^\circ\mathrm{C}$ for up to 20 h. As the exposure temperature was increased, the $\mathrm{B_2O_3}$ liquid vaporized rapidly, leading to an overall weight loss. However, $\mathrm{Si_3N_4}$ was oxidized at above $1200^\circ\mathrm{C}$, leaving $\mathrm{Y_2Si_2O_7}$ crystals surrounded by a glassy phase. These oxidized surfaces strongly affected the mechanical properties of the fibrous monolith. As the exposure temperature was further increased, the flexural strength and WOF decreased, while the noncatastrophic failure was maintained.

References

- ¹A. G. Evans, "Perspective on the Development of High-Toughness Ceramics," J. Am. Ceram. Soc., **73** [2] 187–206 (1990).
- ²W. J. Clegg, K. Kendall, N. McN. Alford, T. W. Button, and J. D. Birchall, "A Simple Way to Make Tough Ceramics," *Nature (London)*, **357**, 455–57 (1990).
- ³D. Kovar, M. D. Thouless, and J. W. Halloran, "Crack Deflection and Propagation in Layered Silicon Nitride–Boron Nitride Ceramics," *J. Am. Ceram. Soc.*, **81** [4] 1004–12 (1998).
- ⁴D. Kovar, B. H. King, R. W. Trice, and J. W. Halloran, "Fibrous Monolithic Ceramics," *J. Am. Ceram. Soc.*, **80** [10] 2471–87 (1997).
- ⁵B. H. King, "Influence of Architecture on the Mechanical Properties of Fibrous Monolithic Ceramics"; Ph.D. Thesis. University of Michigan, Ann Arbor, MI, 1997.
- ⁶R. W. Trice, "The Elevated Temperature Mechanical Properties of Silicon Nitride/Boron Nitride Fibrous Monoliths"; Ph.D. Thesis. University of Michigan, Ann Arbor, MI, 1998.
- ⁷Y.-H. Koh, H.-W. Kim, and H.-E. Kim, "Mechanical Properties of Fibrous Monolithic Si₃N₄/BN Ceramics with Different Cell Boundary Thicknesses," *J. Mater. Res.*, submitted.
- ⁸D. S. Fox, "Oxidation Behavior of Chemically-Vapor-Deposed Silicon Carbide and Silicon Nitride from 1200° to 1600°C," *J. Am. Ceram. Soc.*, **81** [4] 945–50 (1998).
- 9 K. Komeya, Y. Haruma, and T. Meguro, "Oxidation Behavior of the Sintered Si_3N_4 – Y_2O_3 – Al_2O_3 System," *J. Mater. Sci.*, **27**, 5727–34 (1992).
- ¹⁰D. M. Mieskowski and W. A. Sanders, "Oxidation of Silicon Nitride with Rare-Earth Oxide Additions," *J. Am. Ceram. Soc.*, 68 [7] C-161–C-163 (1985).
- ¹¹M. K. Cinibulk and G. Thomas, "Oxidation Behavior of Rare-Earth Disilicate—Si₃N₄ Ceramics," J. Am. Ceram. Soc., 75 [8] 1044–49 (1992).
- ¹²H.-W. Kim, Y.-H. Koh, and H.-E. Kim, "Oxidation and Effect of Oxidation on the Strength of Si₃N₄-SiC Nanocomposite," *J. Mater. Res.*, **15** [7] 1478-82 (2000)
- ¹³N. S. Jacobson, S. Farmer, A. Moore, and H. Sayir, "High-Temperature Oxidation Behavior of Boron Nitride: I, Monolithic Boron Nitride," *J. Am. Ceram. Soc.*, **82** [2] 393–98 (1999). □