## Context dependent lotic macroinvertebrate responses in Michigan's Upper and Lower

Peninsulas to bioavailable sediment copper

By Laura Y. Podzikowski

A project submitted in partial fulfillment of the requirements for the degree of Master of Science (Natural Resources and Environment) at the University of Michigan December 2013

Thesis Committee: Professor G. Allen Burton, Chair Professor Mike Wiley

# Table of Contents:

Acknowledgementsiii
Abstractiv
1.0 Introduction1
2.0 Materials and Methods
2.1 Sediment selection and Cu amendment
2.2 Field deployment site selection
2.3 Sediment deployment
2.4 Invertebrate and geochemical sampling
2.5 Chemical analysis
2.6 Statistical analysis
3.0 Results
3.1 Oxic sediment depth
3.2 Cu partitioning in spiked sediments
3.3 Macroinvertebrate colonization of Cu spiked chambers
3.4 Macroinvertebrate colonization of field contaminated sediments14
4.0 Discussion
5.0 Conclusions
Literature Cited
Supporting Information
Appendix

### **Acknowledgements:**

I would like to take this opportunity to extend my gratitude to the many people who made this research possible. First, I would like to thank my advisor, Dr. Allen Burton, for his advice, guidance, and mentoring patience. To my co-committee member, Dr. Mike Wiley, I greatly appreciated his comments, invertebrate expertise, and his general enthusiasm for the natural world. I would like to thank Dr. Dave Costello, without whom this research would not have been possible. His efforts planning and implementing the project and his mentorship were invaluable. The laboratory and field work in this project would not have been possible without the help, guidance, and support of the entire lab. For their training and support I would like to thank Anna Harrison and Kyle Fetters. Others I would like to thank include Raissa Mendonça, Maggie Grundler, Lauren Eastes, Olivia Rath, and Corrine Cramer for their field and lab assistance. I would like to extend my gratitude to the School of Natural Resources and the Environmental for their educational guidance and resources. For the use of their beautiful field station I would like to thank the Huron Mountain Club. Finally, I would like to thank my family and friends for their patience, understanding, and unyielding support throughout my education.

#### Abstract:

This research investigated context dependent responses of two macroinvertebrate communities to the same Cu treatments to see how community responses differed and changed with sediment aging and oxidation. Sites were located in Michigan's Upper (UP) and Lower Peninsulas (LP) that experience relatively low and high anthropogenic disturbance. We spiked clean sediments with Cu to establish five treatments (0-2100 mg/kg) and placed those sediments in two watersheds. Sediments were aged in situ for 12 weeks in the Pine (UP) and Little Molasses (LP) Rivers, then sampled at 1, 4, and 12 weeks for invertebrate colonization and geochemical composition. We found macroinvertebrate responses to Cu were context dependent and varied with site and season. We observed a 30% reduction in acid volatile sulfides (AVS) after 12 weeks due to oxidation. In turn Cu bound to FeO<sub>x</sub>+MnO<sub>x</sub> significantly increased after 12 weeks aging, which potentially decreased Cu bioavailability. This was supported by the significance of invertebrate metrics responding to Cu bound to Fe and Fe fractions in multiple regression analyses. We observed increased sediment oxidation after 12 weeks, which was likely the result of burrowing invertebrates at the Pine and sandy sedimentation at Little Molasses. Since we observed varied responses with only two sites, this suggests context dependency could play an important role in ecotoxicology and further research is needed addressing confounding issues of natural variation in ecotoxicology. We stress the need to incorporate FeO<sub>x</sub>+MnO<sub>x</sub> fractions in bioavailability models for oxic sediments in order to improve predictions of toxicity, as using sulfide and organic carbon solely to predict invertebrate responses can lead to the overestimation of toxicity when FeO<sub>x</sub>+MnO<sub>x</sub> fractions are present.

*Keywords:* copper partitioning, macroinvertebrate community ecology, ecotoxicology, redox environment

#### 1. Introduction:

Freshwater sediments contaminated with metals can impair benthic communities causing degradation of aquatic ecosystems. Divalent metals (e.g. Cu, Zn, Pb) tend to adsorb to particulate matter and then settle out of solution and accumulate in sediments, where they are exposed to biogeochemically dynamic conditions (Lee *et al.* 2000; Eggleton & Thomas 2004; Kelderman & Osman 2007; Cantwell, Burgess & King 2008). Many divalent metals, such as copper, bind to a variety of chemical pools within sediments including sulfides, organic carbon (OC), and iron and manganese oxides (FeOx+MnOx; Allen, Fu & Deng 1993; Calmano, Hong & Forstner 1993; Kostka & Luther III 1994; Perin et al. 1997; Simpson, Apte & Batley 1998; Lee et al. 2000; Yu et al. 2001; Burton et al. 2005; Cantwell et al. 2008; Teuchies et al. 2010; Costello et al. 2011; De Jonge et al. 2012a). The bioavailable fraction of metal is the metal available for biological uptake and this fraction is often composed of metals dissolved in porewater (Eggleton & Thomas 2004; Burton 2010). Under equilibrium conditions it is predicted there will be a direct relationship between the metal concentrations in sediments, pore water, and benthic organisms (Burton 2010). This relationship has been defined procedurally by the formation of sulfide metal complexes. In molar excess, metals will displace Fe and Mn to form a stable precipitate with sulfides, rendering metals unavailable for biological uptake. Current methods estimate the bioavailable fractions in sediments by sulfide and metal concentrations, which are determined procedurally using a 1 N HCl acid extraction; the simultaneously extracted metal (SEM) in molar excess of acid volatile sulfides (AVS;  $\Sigma$ SEM<sub>Me</sub> – AVS; Allen et al. 1993). Since metals bind readily to OC (Calmano et al. 1993; Perin et al. 1997; Cantwell et al. 2008), bioavailability models can be normalized by the fraction organic carbon in sediments ( $(SEM_{Me} - AVS)/f_{OC}$ ) to improve estimations of bioavailable metals

(Burton 2010). These estimations are dependent on size of chemical pools (ligands available) and the partitioning coefficients, but also assume homogeneous, anoxic conditions, and a stable redox environment with sulfide binding dominating (Perin *et al.* 1997; Lee *et al.* 2000; Yu *et al.* 2001).

However, we know that the concentration of ligands (e.g. sulfide, OC, and  $FeO_x+MnO_x$ ) are not stable and can vary spatially with redox potential (Perin et al. 1997; Eggleton & Thomas 2004) and temporally (van Griethuysen et al. 2006; Burton & Johnston 2010). In sediments an oxidized layer (oxic zones), defined by the penetration depth of O<sub>2</sub>, lies just above the suboxic zone, in which oxidized species are present, but  $O_2$  is not. Typically suboxic zones are defined by the presence of Fe oxides. These exist on top of reducing environments (anoxic zones), defined by the reduction of sulfate (Kristensen 2000). Oxic, suboxic, and anoxic zones create a dynamic redox environment affecting the partitioning of metals, such as Cu. Under anoxic conditions, sulfides form from microbially-mediated reduction of sulfate. In the presence of sulfides, CuS complexes form a highly insoluble precipitate (Allen et al. 1993; Simpson et al. 1998; DeJonge et al. 2012a). When anoxic sediments are exposed to O<sub>2</sub>, sulfide oxidation occurs, which can release free Cu ions (Simpson et al. 1998; Lee et al. 2000; Teuchies et al. 2010; DeJonge *et al.* 2012a). In oxic and suboxic zones FeO<sub>x</sub>+MnO<sub>x</sub> are present (Kostka & Luther III 1994; Kristensen 2000), which can also bind metals thereby decreasing bioavailability in surface layers (Kelderman & Osman 2007; Costello et al. 2011; DeJonge et al. 2012a). The sizes of chemical pools are dynamic, shifting with changes in redox potential. Microbial metabolism accounts for the rapid declines in  $O_2$  in surficial sediments, but is dependent on temperature, organic matter supply, current, and light (Kristensen 2000). Bioturbation (a process in which benthic organisms burrow, physically displacing anoxic sediment and increasing

surface water transport) increases substrates for microbial degradation from secretions through mixing (Kristensen 2000; Meysman, Middelburg & Heip 2006). Bioturbation can significantly increase oxidation depths, but depend strongly on the invertebrate community composition and even season (e.g. presence of burrowers; Charbonneau & Hare 1998; Kristensen 2000).

Similarly, community responses to metals depend on context, in that the effects can vary along spatial, temporal, and environmental gradients (Clements, Hickey & Kidd 2012). Some taxa are known to be less tolerant of metal pollution than others (Burton & Johnston 2010). Ephemeroptera (mayfly), Plecoptera (stonefly), and Trichoptera (caddisfly) larvae (EPT) are sensitive to many stressors and these insects are commonly used as indicators of a host of environmental disturbances, including metal contamination, when they are present (Clements et al. 1989; Clements 1994; Clements & Kiffney 1995; Kiffney & Clements 1996; Costello et al. 2011). The amphipod Hyalella azteca is used for a wide range of toxicity testing and is sensitive to metals (Burton 1991; Burton et al. 2005; Costello et al. 2011). Another amphipod, Gammarus spp., has also shown sensitivity to sediment Zn in previous studies (Costello et al. 2011). Other organisms (e.g. dipterans and oligochaetes) are readily abundant in the environment and are generally considered more tolerant of metal pollution (Clements et al. 1989; Clements & Kiffney 1995; DeJonge, Blust & Bervoets 2010; Costello et al. 2011). Communities with a high percentage of sensitive taxa might respond more readily to metal contamination than those largely composed of tolerant taxa; thus environmental context could inform predictions of toxicity (Clements et al. 1989, 2012; Clements & Kiffney 1995; DeJonge et al. 2012b).

Michigan's Upper and Lower Peninsulas offer a platform for comparing how different communities respond to Cu spiked sediments. These assemblages vary not only regionally, but also in disturbance regime. The Upper Peninsula site, Pine River (Marquette Co.), was located

on privately owned land that is relatively undisturbed by human activities. The Lower Peninsula site, Little Molasses River (Gladwin Co.), is located in a state forest, which is well trafficked and prone to higher amounts of disturbances. The purpose of this study was firstly, to explore the responses of invertebrate communities that vary in ecological context to the same Cu treatments. Secondly, to monitor the invertebrate communities as sediments incubated *in situ* to see how responses changed. We further monitored physicochemical sediment characteristics in order to explore changes in Cu partitioning in sediments and explain why changes occurred with *in situ* aging. We expected to see context dependent responses to Cu and a greater magnitude of effect at the Pine, because of the low disturbance regime and high prevalence of EPT taxa. Secondly, we expect invertebrate responses to Cu to diminish with aging as Cu complexes with different pools of ligands and sediment oxidation increases, but for responses to continue to vary between sites.

## 2. Methods:

## 2.1. Sediment selection and Cu amendment:

Non-contaminated Raisin River depositional sediment was selected for Cu amendment because of its high AVS content (11.65  $\mu$ mol g<sup>-1</sup> dw). Surficial sediments were collected with shovels and stored under N<sub>2</sub> atmosphere at room temperature. Sediments were spiked with CuCl<sub>2</sub>· 2H<sub>2</sub>O using the indirect-spiking method (Simpson, Angel & Jolley 2004; Hutchins *et al.* 2009; Brumbaugh *et al.* 2013). Briefly, a small volume of sediment was spiked with a high concentration of Cu, pH buffered, equilibrated for 2 weeks under N<sub>2</sub> atmosphere, and diluted to desired treatments (0, 380, 750, 1200, and 2100 mg Cu kg<sup>-1</sup> dw). Both the superspike and dilutions were buffered with NaOH to maintain the pH within 0.5 units of initial sediment pH. Final sediment treatments were allowed to equilibrate under N<sub>2</sub> atmosphere for 14 days prior to deployment. All treatments were mixed twice a week (rolled for >1 hr.) to homogenize throughout the equilibration period.

Field contaminated sediments were collected in December 2011 from the Ocoee River floodplain in Polk Co., Tennessee. Historical mining activities from the mid 1800s until the late1980s caused metal contamination (predominantly Cu, secondarily Pb and Zn) of sediments (Carr & Zeller 2006). Sediments were chosen to represent a gradient of Cu contamination ranging from 170 – 1600 mg Cu kg<sup>-1</sup> dw. Three of the sediment types were clay or clay/silt, while the lowest Cu concentration had a sandy texture. AVS concentrations also varied among the sediment types ranging from concentrations below detection limits to 10.39  $\mu$ mol g<sup>-1</sup> dw (Table 1).

## 2.2. Field deployment site description:

Two stream locations were selected for deployment of sediments for their variations in watershed and physicochemical characteristics, as well as differences in macroinvertebrate assemblages. The Pine River (P) in the Upper Peninsula (Marquette Co., MI) had a relatively undisturbed watershed, heterogeneous substrate, soft water (53  $\pm$  4.16 mg L<sup>-1</sup> CaCO<sub>3</sub>; Briggs & Ficke 1977), and high abundance of sensitive EPT taxa (Yanoviak & McCafferty 1996). Little Molasses River (LM) in the Lower Peninsula (Gladwin Co., MI) had moderately hard to hard water (116  $\pm$  23.4 mg L<sup>-1</sup> CaCO<sub>3</sub>; Briggs & Ficke 1977), homogeneous fine substrate (sand as opposed to cobble), and an abundance of sensitive amphipods (Costello *et al.* 2011: Honick 2013). Most importantly, macroinvertebrate community composition differed between sites. Background samples of depositional sediments from the Pine River had fewer taxa, higher diversity (Simpson's D), and greater evenness (Pielou's J). Little Molasses had more taxa than the Pine, but were dominated by fly larvae and Gammaridae leading to lower diversity and lower evenness (Pielou's J) when compared to the Pine (Table 2). Dissolved oxygen, temperature, conductivity, pH, and turbidity were measured hourly *in situ* throughout most of the 13 week experiment with datasondes (YSI 6920 V2). Surface water grab samples were collected 4 times—at deployment and each of the subsequent sampling days—to determine water hardness and alkalinity (Table 2).

## 2.3. Sediment deployment:

Sediments were deployed in July 2012 in the Little Molasses and Pine Rivers using *in situ* chambers (Burton *et al.* 2005; Costello *et al.* 2011; Honick 2013). Plastic chambers were made with plastic baskets (25.4 x 7.7 x 5.7 cm in dimension) lined with 1.5 mm mesh to prevent

sediment loss (Fig. 1a). Chambers (n = 3 per treatment) were placed flush with the sediment surface and secured to the stream using steel frames and rebar (Fig. 1 a, b). Cu treatments were placed with lowest concentrations upstream of higher concentrations to minimize contamination due to sediment transport. Chambers were deployed in nylon mesh bags with openings of 5 mm (sufficient size for most macroinvertebrate colonization) to minimize sediment loss in times of high discharge (Fig. 1b).

#### 2.4. Invertebrate and geochemical sampling:

Chambers were destructively collected to measure invertebrate colonization and geochemical composition after 1, 4, and 12 week aging periods in July, August, and October 2012 respectively. Each chamber was subdivided with two-thirds of the chamber reserved for invertebrate colonization and one third for geochemical sampling. The presumed biologically active layer of sediment (top 2 cm), was collected, sieved (45 µm), and preserved in 70% ethanol for identification (family level) and enumeration of macroinvertebrates (Hilsenhoff 1995; Merritt & Cummins 1996; Bouchard Jr. 2004). Invertebrate data was used to calculate community composition metrics including abundance, richness, EPT richness-defined as the number of EPT taxa (Lenat 1988; Merritt & Cummins 1996)—EPT abundance, relative EPT abundance, chironomid abundance, relative chironomid abundance, amphipod abundance, relative amphipod abundance, and Simpson's diversity (1 - D; SI Table 1). The remaining sediment was submerged in a plastic container with site water and analyzed on site for sediment oxygen content at depth immediately after sampling using a microelectrode and motorized profiler set up within 100 feet of the sample site (Unisense OX 100). Dissolved oxygen (DO) concentrations were recorded every 500 µm until concentrations reached zero. The depth of the oxic layer was

defined as the distance between the surface-water interface (i.e. the height just before DO concentrations began to decline) and the height at which DO concentrations first reached zero in the sediments (SI Fig. 1). The sediment reserved for geochemical analysis was divided into surface (top 2 cm) and deep (below 2 cm) fractions and stored frozen in plastic 50 mL centrifuge tubes for chemical analyses.

## 2.5. Chemical analyses:

All sediment samples were analyzed for AVS and SEM, total metals, organic carbon content (%OC), sediment water content (% dry weight), and FeO<sub>x</sub>+MnO<sub>x</sub>. AVS was extracted from wet sediment using a 1 N HCl acid volatilization process under anoxic conditions (Allen *et al.* 1993). Briefly, sulfides are volatilized, then trapped in NaOH, and measured colormetrically. SEM metals are those dissolved metals (0.45  $\mu$ m filter) liberated during the acid volatilization of AVS. Total metals were extracted using concentrated acids (3:1 HNO<sub>3</sub>: HCl) and microwave assisted digestion (U.S. EPA 2007). OC content was determined through measurement of organic matter content by loss on ignition (LOI) at 450°C and conversion to %OC via the Redfield ratio (0.36; Costello *et al.* 2011). Amorphous FeO<sub>x</sub>+MnO<sub>x</sub> were determined by incubating 0.1 – 0.5 g wet sediment in an ascorbate solution at room temperature on a shaker table for 24 hours and filtered (0.45  $\mu$ m; Kostka & Luther III 1994). Cu-bound to FeO<sub>x</sub>+MnO<sub>x</sub> (Cu<sub>ascorbate</sub>) were those metals liberated during the ascorbate extractions. All metal extractions were stored at room temperature and analyzed for Cu, Fe, and Mn on the Inductively Coupled Plasma-Optical Emission Spectrometry (ICP-OES).

## 2.6. Statistical analyses:

All statistical analyses were performed using R 2.14.1 (R Development Core Team 2011). Changes in sediment oxic layer depths were analyzed using a two-way analysis of variance (ANOVA) to determine how depths changed among treatments and with aging. Models were tested for normality using a Shapiro-Wilk test and if a significant model was observed ( $\alpha =$ 0.05) a TukeyHSD multiple comparisons post-hoc test (MCT) was performed to analyze the differences between factors. Changes to Cu partitioning during in situ aging were analyzed using a multi-way analysis of covariance (ANCOVA) to determine how relationships between pools of Cu and binding fractions (Cutotal, AVS, CuSEM, FeSEM, MnSEM, FeOx+MnOx, and Cu<sub>ascorbate</sub>) changed though time with location as a block. Dependent variables were ln or square root transformed to meet the assumption of normality as determined by a Shapiro-Wilk test. A storm event caused deposition of sandy sediment on some replicates of Little Molasses Cu spiked chambers on week 4 (380, 750, 1200, and 2100 mg kg<sup>-1</sup> surface sediments) and week 12 (750 and 1200 mg kg<sup>-1</sup> surface and 1200 mg kg<sup>-1</sup> deep sediments). This additional sand altered the sediment chemistry as evidenced by reductions in Fetotal, Mntotal, and Cutotal concentrations. For analyses of Cu partitioning during aging, those replicates were excluded from our statistical analysis to avoid the confounded factor of sediment burial (bear in mind patterns in week 4 surficial sediments reflect the Pine data only).

Multiple regression analysis with a forward stepping procedure was used to determine the geochemical or metal fractions that best predicted invertebrate metrics. Invertebrate colonization after week 1 was insufficient to characterize the communities (< 15 individuals for reference and spiked replicates) and was not included in multiple regression analyses. We did, however, include week 4 and 12 samples altered by sedimentation at Little Molasses (both response and

predictor variables) in multiple regression analyses, because the disturbance did not appear to affect invertebrate colonization. All predictor variables were ln transformed prior to model stepping procedure except (CuseM-AVS)/foc and CuseM-AVS, for which negative values (which are predicted non-toxic; Burton 2010) were converted to zero and data were ln(x + 1)transformed. To account for differences in background community composition and potential variation in dose-response between sites, invertebrate metrics were analyzed with location as a potential factor or interaction term. When an interaction term was selected during the stepping procedure, main effects were only included in the final models if stepwise procedures selected them. Variables with the lowest significant p-value were added until there were no further significant variables (criteria  $\alpha = 0.05$  for parameter inclusion). Given the high degree of correlation between deterministic variables, at each step all remaining variables were regressed against one another to account for multicollinearity. If the variables already in the model explained 50% or more of the variation in the new parameter, it was excluded from the model selection procedure as the variation inflation factor (VIF)  $\geq 2$  (Graham 2003). A single final best model was selected from the suite of stepwise models when all significant parameters were added to the model (Thompson 1978). Ocoee data were analyzed separately from Raisin sediments and qualitatively.

#### 3. <u>Results:</u>

#### 3.1. Oxic sediment depth

At the Pine River, the depth of the oxic layer differed significantly among sample weeks  $(F_{[0.05, 2, 31]} = 42.583, p < 0.0001)$ , while at Little Molasses, once excluding samples altered by sedimentation, aging did not affect depth of the oxic layer. At the Pine River, depth of the oxic layer was similar between weeks 1 and 4 for all treatments. After 12 weeks *in situ* aging oxic layers were significantly deeper in lower treatments (0, 380, and 750 mg kg<sup>-1</sup>; MCT, p < 0.05), while mean depths remained similar to earlier sampling periods in sediments treated with higher Cu concentrations (Fig. 2a). At Little Molasses River oxic layers were deeper in samples affected by sedimentation, while depths were not significantly different between other treatments and days (Fig. 2b).

## 3.2. Cu partitioning in spiked sediments

We observed statistically significant changes in redox sensitive sulfide species and simultaneously extracted Cu in surface sediments with sediment aging. As expected (Allen *et al.* 1993; Simpson *et al.* 1998), AVS was negatively correlated to Cu<sub>total</sub> in surface ( $t_{[0.05, 17]} = -10.053$ , p < 0.001, Fig. 3a) and deep ( $t_{[0.05, 22]} = -8.888$ , p < 0.001; Fig. 3d) sediments. After 12 weeks *in situ* aging, we observed a 30% reduction in surficial AVS compared to previous weeks ( $t_{[0.05, 17]} = -2.823$ , p = 0.01) and a decrease in the magnitude of correlation between AVS and Cu<sub>total</sub> ( $t_{[0.05, 17]} = -2.392$ , p = 0.03; Fig. 3a). Note that this is a conservative estimate of oxidation as oxic layers only include a fraction (10-50%) of total surface samples run for geochemical analysis. Cu<sub>SEM</sub> was positively related to Cu<sub>total</sub> in both surface ( $t_{[0.05, 17]} = 10.017$ , p < 0.001) and deep ( $t_{[0.05, 22]} = 7.653$ , p < 0.001) sediments. Week 1, Cu<sub>SEM</sub> accounted for 55 ± 19% of Cu<sub>total</sub>

and decreased to  $43 \pm 16\%$  week 4 resulting in a significant decrease in the slope of the Cu<sub>SEM</sub> and Cu<sub>total</sub> relationship (t<sub>[0.05, 17]</sub> = -2.7, p = 0.02; t<sub>[0.05, 22]</sub> = -2.291, p = 0.035). Proportions returned to  $49 \pm 24\%$ , similar to week 1 concentrations in week 12 (Fig. 3b,e).

Pools of FeO<sub>x</sub>+MnO<sub>x</sub> and associated Cu in surface sediments also changed as sediments aged in situ. FeO<sub>x</sub>+MnO<sub>x</sub> in deep sediments were not related to Cu<sub>total</sub> and were homogenous between sample weeks and locations. Surface  $FeO_x+MnO_x$  concentrations were highest at week 1. We observed a 32% reduction in surface concentrations after 4 weeks ( $t_{[0.05, 17]} = -2.9$ , p = 0.01) and, compared to week 1, a 23% reduction after 12 weeks ( $t_{[0.05, 17]} = -2.2$ , p = 0.04). Concentrations of FeO<sub>x</sub>+MnO<sub>x</sub> in surface sediments at Little Molasses were 18% lower than concentrations at the Pine in surface sediments ( $t_{[0.05, 17]} = -2.8$ , p = 0.01). At Little Molasses concentrations of Cuascorbate in surface sediments were significantly lower than at the Pine ( $t_{[0.05, 17]} = -2.9$ , p = 0.009). We observed a positive relationship between Cu<sub>ascorbate</sub> and Cu<sub>total</sub> in both surface ( $t_{0.05}$ ,  $_{171} = 5.03$ , p = 0.0001; Fig. 3c) and deep ( $t_{10.05, 221} = 9.3$ , p < 0.0001; Fig. 3f) sediments; the slope of that relationship significantly increased after week 12 in surficial sediments only  $(t_{10.05, 17}) =$ 2.8, p = 0.01; Fig. 3c). The moles of FeO<sub>x</sub>+MnO<sub>x</sub> required for binding one mole of Cu also decreased with time. The mean molar ratio of FeO<sub>x</sub>+MnO<sub>x</sub> to Cu<sub>ascorbate</sub> for all treatments at both sites was greatest on week 1 (738.94 $\pm$  316.2) and decreased weeks 4 (612.1  $\pm$  322.4) and 12  $(583.9 \pm 340.8)$ .

#### 3.3. Macroinvertebrate colonization of Cu spiked chambers

Macroinvertebrates metrics strongly responded to sediment Cu after 4 weeks of aging but these effects diminished by week. After 4 weeks, 8 out of 10 macroinvertebrate metrics, including abundances, richness, and relative abundance, responded to at least one measure of sediment Cu pools (Table 3). No single Cu pool predicted all invertebrate responses as different metrics selected for Cusem, Cuascorbate, and (Cusem-AVS)/foc (Table 3). Total abundance, chironomid abundance, and amphipod abundance negatively responded to Cusem. At the Pine, EPT abundance negatively responded to Cuascorbate (Fig. 4e). Richness negatively responded to (Cusem-AVS)/foc at Little Molasses (Fig. 5c). Relative amphipod abundance negatively responded to Cu<sub>ascorbate</sub>, while relative chironomid abundance did not respond to sediment Cu, but rather was negatively related to AVS (Fig. 6c, a). Gammaridae at Little Molasses drove the negative relationships between amphipod densities, relative amphipod abundance and Cu, because Gammaridae occur in higher densities than Hyalellidae. After 12 weeks *in situ* aging, we observed weakened relationships between the macroinvertebrate community and sediment Cu with only 3 out of 10 benthic metrics responding negatively to sediment Cu pools (Table 3). Abundance negatively responded to Cutotal (Fig. 4b) and diversity negatively responded to CUSEM-AVS as a primary predictor and as a secondary predictor Cutotal at the Pine (Fig. 5b). Richness negatively responded to (Cusem-AVS)/foc (Fig. 5d) and relative chironomid abundance positively responded to Cusem-AVS (Fig. 6b).

Responses of benthic macroinvertebrates to Cu were location dependent after 4 weeks of aging. Of the 10 metrics analyzed, 6 included location as a factor or interaction term with Cu (Table 3). In the Pine, abundance of EPT (Fig. 4e) and EPT richness (Fig. 5e) responded negatively to Cu<sub>ascorbate</sub>, while no significant relationship was observed at Little Molasses. Location alone explained greater diversity and relative EPT abundance at the Pine compared to Little Molasses (Fig. 5a; Fig. 6e), but neither responded to Cu. At Little Molasses relative amphipods, relative chironomids, and overall richness responded to Cu (Fig. 6c, a, 5c), while at the Pine no relationship was observed. We observed a decrease in EPT taxa at the Pine with

increasing ( $Cu_{SEM}$ -AVS)/ $f_{OC}$ , while no relationship between EPT taxa and Cu was observed at Little Molasses (Fig. 5e). After 12 weeks of *in situ* aging, only 3 out of the 10 metrics included location as a factor or interaction term and none of the interactions were with measures of Cu. As a secondary variable diversity negatively responded to Cu<sub>total</sub> at the Pine, while no relationship was observed Little Molasses (Table 3).

Redox sensitive species, FeO<sub>x</sub>+MnO<sub>x</sub>, Fe<sub>SEM</sub>, and AVS, fractions explained additional variation as primary and secondary parameters in chironomid, amphipod, and abundance models on weeks 4 and 12. After 4 weeks *in situ* aging AVS primarily and Fe<sub>SEM</sub> secondarily explained variation in relative chironomids, while Fe<sub>SEM</sub> (secondary to Cu<sub>SEM</sub>) explained additional variation in chironomid abundance (Table 3). After 12 weeks FeO<sub>x</sub>+MnO<sub>x</sub> and Fe<sub>SEM</sub> were primary predictors explaining variation in amphipod abundance and relative amphipods (which no longer responded to Cu; Fig. 4h, 6c). As a secondary predictor, FeO<sub>x</sub>+MnO<sub>x</sub> fractions explained variation in total abundance at both locations (Table 3).

### 3.4. Macroinvertebrate colonization of field contaminated sediments

Colonization of field contaminated sediments was driven by physicochemical characteristics and did not respond to Cu. Physicochemically, the field sediments include a wide range of textures and pH values. Ocoee1 was sandy in texture with a mean pH of  $4.3 \pm 1.2$  and Ocoee 4 was clay with a mean pH of  $5.4 \pm 0.8$ . Ocoee 2 and 3 were also clay sediment textures with mean pH values of  $4.2 \pm 0.7$  and  $6.2 \pm 0.8$ , respectively. Invertebrate colonization did not fit Cu bioavailability models, with high colonization of sediments at and above toxic thresholds  $(100 - 150 \mu mol g_{oc}^{-1}; Burton 2010)$  and low abundance below thresholds (Fig 7). Most of the

invertebrates colonizing Ocoee chambers were epibenthic taxa (Gammaridae, Elmidae) or large predators (Gomphidae, Cambaridae).

### 4. Discussion:

Since the Pine and Little Molasses Rivers differed in disturbance regime, substrate heterogeneity, and macroinvertebrate community composition, we expected to see context dependent responses to Cu (Clements et al. 2012) and for the magnitude of response to be greater at the Pine. Background community composition did indeed drive invertebrate responses to Cu after 4 weeks in situ aging (sensitive EPT taxa at the Pine and Gammarus abundance at Little Molasses) and the magnitude of responses varied with sites (Table 3; Fig. 4, 5, 6). Rohr & Crumrine (2005) also observed that initial community composition affected freshwater community responses to pesticides in mesocosm experiments. Longitudinal variation is perhaps one of the most well documented examples of context dependent variation in stream ecology (Vannote et al. 1980), a concept which has been applied ecotoxicology. A previous mesocosm study found effects of metals were 12 - 85% greater at small headwater high altitude streams, suggesting responses to metals vary along longitudinal stream gradients (Kiffney & Clements 1996). However field collections suggest that natural changes in community composition that occur with elevation can interfere with determining if changes to community composition were directly effected by metals (Clements & Kiffney 1995). In this study variations in community composition alone (e.g. densities and relative sensitivity of EPT and amphipods) add a great deal of variability. Overlooking context dependent patterns in community composition could result in both over- or under-estimations of toxicity and choosing the correct metrics for the community is critical in assessing ecological effects. At Little Molasses, for instance, if only EPT metrics were analyzed, the assessor might over-estimate toxicological effects and would miss important shifts from relative Gammarus to chironomid dominance after 4 week exposures in response to Cu. Since invertebrate colonization is highly dependent on aquatic habitat and distribution of taxa

tend to be conserved between similar substrate in lentic and lotic systems (Merritt & Cummins 1996), a perhaps more appropriate way to assess toxicity at local spatial scales would be to develop substrate specific indices for invertebrate responses in order to account for context dependent responses to metals.

We expected context dependent responses to persist over time; however invertebrate responses to Cu after 12 weeks aging were not location dependent (Table 3). This was surprising, but seasonal changes in community composition could account for these patterns. After 12 weeks, EPT and Gammarus densities decreased overall, suggesting community composition changed temporally. At Little Molasses we observed over a 50% reduction in reference densities between week 4 and 12 as a result of lower Gammarus densities. At the Pine, a reduction in EPT abundances was observed in both reference and spiked sediments after 12 weeks. Without accurate reference samples, one might conclude that reductions in EPT densities suggest effects of Cu persisted with aging. This could result in an over-estimate of toxicity, when seasonal variation could explain observed patterns. Temporal changes in community structures are important in assessing ecological risk; if this is neglected if could alter how we interpret field data. Broad community metrics (e.g. abundance, chironomid abundance, and relative chironomids) responded to Cu both days without location as a factor or interaction term, suggesting that at regional scales less resolute metrics may more accurately predict adverse responses to metals. Marchant, Barmuta & Chessman (1995) suggest that at large spatial scales family level taxonomic resolution may suffice, while at within-stream levels finer taxonomic resolution may be appropriate. A similar approach could be used in ecological risk assessment with community metrics to address some issues of ecological variation. Colonization of field contaminated sediments appear to respond to physicochemical characteristics rather than metal

contamination, which highlights that factors other than contamination can lead to biological degradation (e.g. habitat destruction). Risk assessors should carefully consider these potential confounding factors so resources aren't misallocated for clean-up efforts when, for instance, habitat rehabilitation would suffice.

Oxidation of AVS after 12 weeks *in situ* aging occurred, while Cu binding to surficial FeO<sub>x</sub>+MnO<sub>x</sub> species increased in turn; potentially decreasing the fraction adversely affecting invertebrates (Fig. 3). Similar to previous studies we found Cu has a strong affinity for sulfides and OC, accounting for 43 - 55% of Cu partitioning in spiked sediments (Allen et al. 1993; Perin et al. 1997; Lee et al. 2000; Simpson et al. 2004; DeJonge et al. 2012a). Loss of AVS is similar to those observed in previous studies, where 8 - 12 hours after resusupension events or 18 - 54days after incubation in oxic settings researchers observed 65 – 95% reductions in surficial AVS (Simpson et al. 1998; Lee et al. 2000; Teuchies et al. 2010; DeJonge et al. 2012a). However, we did not see an increase in adverse affects to invertebrates after 12 weeks, possibly due to increased concentrations of Cu scavenged by FeO<sub>x</sub>+MnO<sub>x</sub>. The invertebrate data supports that claim. Community metrics responded to Cu-bound FeO<sub>x</sub>+MnO<sub>x</sub> and Fe fractions even though this fraction accounted for 1 - 2% of Cu partitioning in sediments. Previous research suggests that higher proportions of Zn are found pooled in FeO<sub>x</sub>+MnO<sub>x</sub> fractions, while Cu tends to reside in OC and sulfide fractions (Kelderman & Osman 2007; DeJonge et al. 2012a). It is surprising that forward stepping procedures selected FeOx+MnOx and Cuascorbate fractions to explain invertebrate colonization even though Cu has such a strong affinity for sulfides and OC. This suggests firstly, that redox sensitive Fe and Mn species are important fractions for predicting toxicity in oxic sediments and secondly, that perhaps Cu and OC or sulfide complexes are so stable that smaller fractions are driving community responses. Current lab based spiking

methods use AVS to predict toxicity, yet two issues with these models arose in this study. Firstly, they fail to account for Cu bound to Fe fractions in surface sediments, which were important predictors of invertebrate colonization at the Pine and Little Molasses. Secondly, in this work multiple regression analysis more often selected Cu<sub>SEM</sub> or Cu<sub>ascorbate</sub>, suggesting that perhaps a weak acid and ascorbate extraction might be a cheaper and easier alternative for setting sediment quality guidelines in oxic sediments. These data suggest that short-term beaker tests on spiked anoxic sediments are not realistic comparisons to field conditions. In this work, Cu partitioning and oxic layer depths were significantly changing after 12 weeks in surficial sediments, suggesting that commonly used 14-day assays are insufficient to characterize long term community responses, especially under dynamic redox conditions.

Increases in oxic sediment depths at the Pine River were likely caused by invertebrate colonization, further altering the sediment redox environment. At the Pine, we observed increased oxic layer depth through time but the increase was not homogeneous across all treatments. Thus, water column oxygen concentrations in the stream (which were high and increased through time (data not shown)) were likely not driving the changes in redox stratification. Benthic organisms can increase oxic layer depths through bioturbation, by mixing anoxic, suboxic, and oxic layers and increasing sediment porosity (Kristensen 1984, 2000; Meysman *et al.* 2006). At the Pine, we found the burrowing mayfly Ephemeridae (*Hexagenia sp.*), and its abundance was greater in lower Cu sediments. Charbonneau and Hare (1998) found that *Hexagenia limbata* burrowed deeper in littoral sediments than all other burrowing taxa and estimate they alone could account for 98% of sediment displacement. Additional burrowing taxa (i.e., some chironomids, oligochaetes) were also observed in greater abundance we have (1998)

observed benthic organisms burrowed deeper in the autumn. It is likely that differences in the colonizing invertebrate community and seasonality may have contributed to changes in oxic layer depth through time at the Pine. There appears to be a positive feedback occurring at the Pine; deeper oxygen penetration causes oxidation of Fe species, further decreasing bioavailable Cu and potentially decreasing adverse affects on benthic taxa that further mix sediments. In contrast, the dominant taxa in lower treatments at Little Molasses were non-burrowing epibenthic organisms (Gammaridae), while benthic organisms (Chironomidae) were observed at lower densities than at the Pine. We observed increased oxic layer depths for only treatments affected by sand deposition, which is likely a result of differences in grain size. Increased grain size is directly related to greater porosity (Shepherd 1989), which leads to deeper oxygen penetration in sediments. This suggests that when epibenthic organisms dominate community composition, physical characteristics of the sediments might be determining sediment redox stratification. This is further evidence suggesting, firstly, short term toxicity tests are not sufficient to characterize Cu partitioning in sediments as oxic layer depths changed after 12 weeks aging and could be related to seasonal changes in invertebrate behaviors. Our data suggest that where burrowing organisms are present, determining the fraction of metals bound to redox sensitive fractions may be more important in assessing ecological risk. Toxicologists could use these patterns to their advantage. Sediments without benthic organisms might be at risk of more dramatic changes in partitioning as a result of disturbances, such as resusupension.

#### 5. <u>Conclusions:</u>

This research investigated context dependent responses of macroinvertebrate communities to Cu and how community responses shifted with sediment aging and oxidation. We found macroinvertebrate responses were context dependent, varying with site and season. AVS oxidized after 12 weeks in situ, while Cu binding to surficial FeO<sub>x</sub>+MnO<sub>x</sub> increased. This potentially decreased the fraction adversely affecting invertebrates, which was supported by the prevalence of invertebrate metrics responding to Cu bound to Fe and Fe fractions in multiple regression analyses. Oxic layers deepened with aging, which might be the result of burrowing fauna at the Pine and increased grain size from sandy sedimentation at Little Molasses. We stress the need to incorporate FeO<sub>x</sub>+MnO<sub>x</sub> fractions in bioavailability models for oxic sediments in order to improve toxicity models. Current bioavailability models rely solely on sulfide and OC chemistry, but are missing important metal binding fractions, which could result in overestimations of toxicity. Cu partitioning and oxic layer depths were significantly changing after 12 weeks in surface sediments, suggesting that commonly used short term tests (e.g. 14 day assays) are insufficient to characterize chronic responses, especially under dynamic redox conditions. This study was limited by a relatively narrow range of physicochemical characteristics (highlighted by the field contaminated sediments), as a single sediment type was used. Further research needs to be done with a wider range of sediment types to fully incorporate FeO<sub>x</sub>+MnO<sub>x</sub> into widely applicable models. Furthermore, we used only two sites to explore context dependent responses. Future research should examine context dependency with more sites to make broader suggestions for dealing with environmental context in ecotoxicology.

Sediment	AVS	OC	pН	Cu Trtmt	Tot. Cu	Tot. Pb	Tot. Zn	Tot. Fe	Tot. Mn
	$(\mu mol g^{-1} dw)$ (%) (mg kg <sup>-1</sup> dw)								
Raisin ref	10.3	3.2	7.1	ref	b.d.				
Raisin 380	3.5	3.2	6.9	low	395				
Raisin 750	1.2	3.2	6.9	med.	761				
Raisin 1200	0.3	3.2	7.0	high	949				
Raisin 2100	0.07	3.2	7.0	very high	1960				
Ocoee 1	b.d.	0.5	3.2	ref	170	82	230	34	177
Ocoee 4	1.66	1.4	4.7	low	599	200	1098	55	655
Ocoee 2	0.02	1.6	3.4	medium	1199	439	1099	107	332
Ocoee 3	10.39	1.7	6.1	high	1600	210	1800	117	874

 Table 1: Background sediment physicochemical properties. Reference sample total Cu was below the detection limits (b.d.) of the ICP-OES.

Table 2: Physical and biological community properties of the Pine (UP) and Little Molasses (LP) Rivers throughout sample aging. Data are presented as means and standard deviations. Temperature, dissolved oxygen, conductivity, pH, and turbidity were sampled hourly by datasondes throughout most of the 3 month aging period. Hardness and alkalinity were sampled 4 times at each deployment and sampling period. Biological communities were qualitatively assessed from depositional samples taken at deployment using a D-net.

	Pine R., UP	Little Molasses R., LP
Physical Properties		
latitude	N 46° 52'	N 43° 56'
longitude	W 87° 52'	W 84° 12'
temp (°C)	20.1 (3.8) <sup>a</sup>	$16.7 (2.8)^{c}$
DO (% sat.)	97.8 (5.9) <sup>b</sup>	83.7 (7.2) <sup>c</sup>
conductivity (mS cm <sup>-1</sup> )	$34.4(19)^{a}$	92.1 $(27.3)^d$
рН	7.7 (0.15) <sup>a</sup>	$7.7(0.23)^{\circ}$
turbidity (NTU)	$0.82(37.8)^{a}$	$3.2(10.4)^{\circ}$
hardness (mg CaCO <sub>3</sub> L <sup>-1</sup> )	53 (4.16)	116 (23.4)
alkalinity (mg CaCO <sub>3</sub> L <sup>-1</sup> )	54 (5.89)	101 (20.3)
<b>Biological Properties</b>		
richness (No.)	8	10
Shannon (H)	0.98	0.76
Simpson's D (1-D)	0.72	0.33
Pielou (J)	1.52	0.76
<u>Amphipods</u>		
Hyalellidae (%)	30.2	0
Gammaridae (%)	0	9.8
Isopods (%)	27.9	0
<u>Odonata</u>		
Calopterygidae (%)	9.3	1.0
Aeshnidae (%)	13.9	0
Gomphidae (%)	7.0	0
Ephemeroptera		
Tricorythidae (%)	4.6	0
Baetidae_(%)	0	<1
Heptageniidae (%)	0	<1
Diptera		
Chironomidae (%)	<1	81.1
Tabanidae (%) <sup>a</sup> sampled hourly from 18 July to 4	0	3.8

<sup>a</sup> sampled hourly from 18 July to 4 October 2012

<sup>b</sup> sampled hourly from 24 July to 4 October 2012

<sup>c</sup> sampled hourly from 27 July to 18 September 2012

<sup>d</sup> sampled hourly from 11 August to 18 September 2012

<sup>e</sup> background samples taken at deployment of depositional areas



Figure 1: Diagram of sediment deployment set-up and chambers *in situ*. A) Chambers were deployed with 3 replicates for each treatment attached to a steel frame. B) Treatments were secured with rebar in the stream flush with the surface sediments in mesh nylon bags from low to high Cu concentration (upstream to downstream).

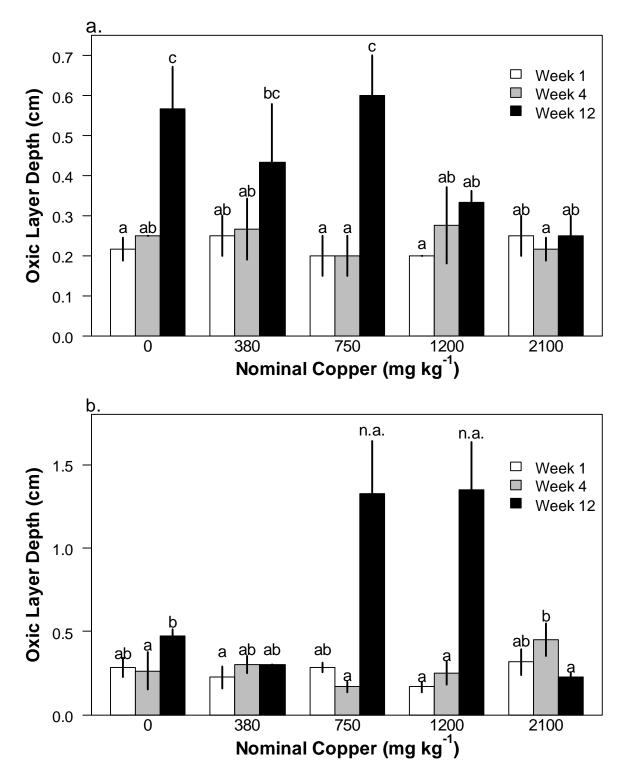


Figure 2: Oxic layer depth for copper spiked sediments aged in the Pine (a) and Little Molasses (b) Rivers. Samples were collected independently 1, 4, and 12 weeks after they were deployed in both locations designated by white, grey, and black representations respectively. Groups with the same letter designation are statistically homogeneous as determined by a MCT TukeyHSD test ( $\alpha = 0.05$ ). Little Molasses samples affected by sedimentation were removed from the statistical analyses, but are represented on the graph by n.a. designation. n = 46 (Pine), 40 (Little Molasses).

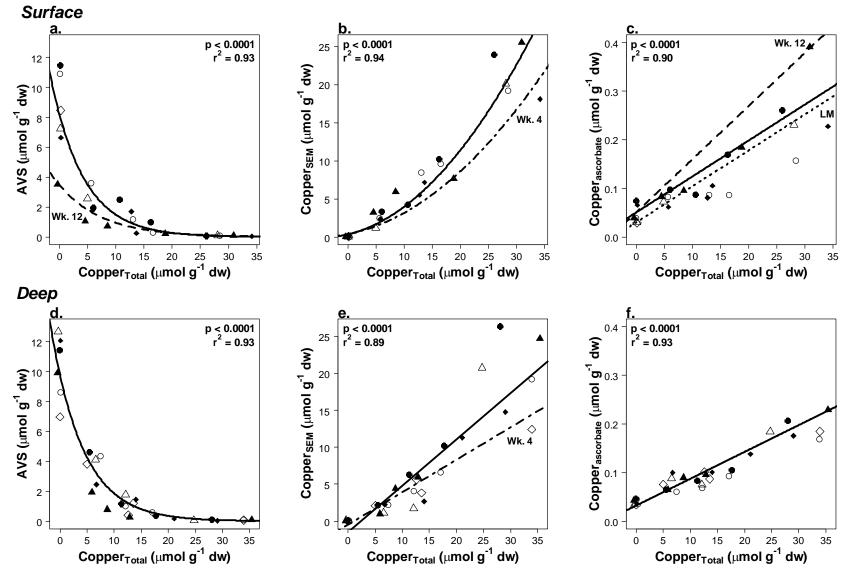
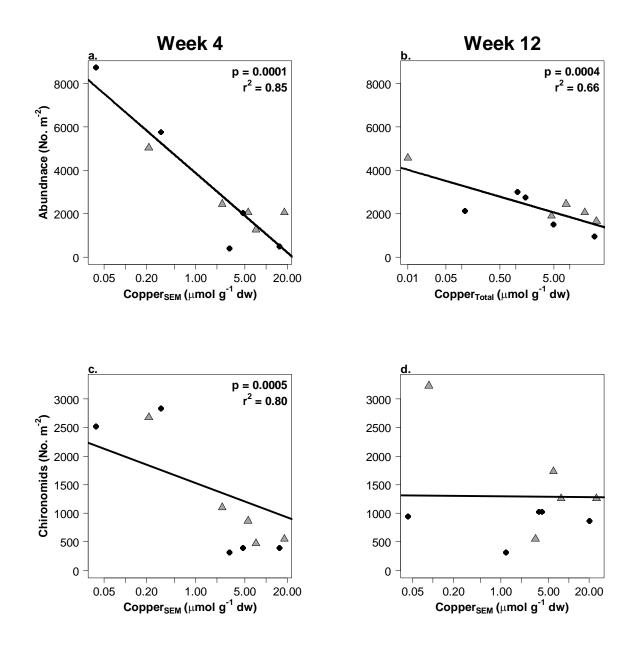


Figure 3: The relationship between total copper and AVS (a, d), CusEM (b, e), and Cuascborbate (c, f) fractions in surface and deep layers. Significant slopes as determined by ANCOVA are indicated as bold lines, while significant interactions and/or factors are indicated with dotted (factor LM), dot-dash (interaction week 4), and dashed (interaction or/and factor week 12) lines. Symbols represent the location and sampling period (Week 1 P [●], Week 1 LM[○], Week 4 P [♦], Week 4 LM [◊], Week 12 P [▲], Week 12 LM [△]). Models were run excluding samples altered by sedimentation at LM (surface n = 54; deep n = 59).

Table 3: Stepwise multiple regression (forward stepping) between benthic macroinvertebrate community metrics, copper, and sediment physicochemical factors. Columns indicate response variables and predictors are in 1° or 2° categories indicating the order in which factors were added to the model. Italicized benthic metrics identify terms with location interactions. Layer is indicated in parentheses (surface or deep) and directional effect are represented [-, +] for each metric including location in superscript (Little Molasses [LM], Pine [P]) where necessary. n = 10.

Response	1° Predictor	2° Predictor	r <sup>2</sup>	p-value	
Week 4					
abundance	Cu <sub>SEM</sub> (surface) [-]		0.98	< 0.0001	
chiron. ab.	Cu <sub>SEM</sub> (surface) [-]	Fesem (deep) [+]	0.92	0.0001	
EPT ab.	Cuascorbate (deep) $[-]^P$		0.86	0.001	
amphipod ab.	Cusem (deep) [-]		0.68	0.003	
diversity (1-D)	Stream		0.70	0.002	
taxa	$(Cu_{SEM}-AVS)/f_{OC}$ (surface) [-] <sup>LM</sup>	Stream	0.87	0.004	
EPT taxa	Cuascorbate (deep) [-] <sup>P</sup>		0.69	0.02	
Rel. chirono. ab.	AVS (surface) [-] <sup>LM</sup>	Fesem (deep) [-] <sup>LM,P</sup>	0.95	0.001	
Rel. amphipod ab	Cuascorbate (deep) [-] <sup>LM</sup>		0.72	0.01	
Rel. EPT ab.	Stream		0.77	0.0008	
Week 12					
abundance chiron. ab. EPT ab.	Cutotal (surface) [-]	FeO <sub>x</sub> +MnO <sub>x</sub> (deep) [-] <sup>LM,P</sup>	0.64	0.03	
amphipod ab.	$FeO_x+MnO_x$ (deep) [-] <sup>LM,P</sup>		0.83	0.002	
diversity (1-D)	(Cu <sub>SEM</sub> -AVS) (deep) [-]	Cu <sub>total</sub> (deep) [-] <sup>P</sup>	0.64	0.006	
taxa	$(Cu_{SEM}-AVS)/f_{OC}$ (deep) [-]		0.45	0.03	
EPT taxa	Stream		0.41	0.05	
Rel. chirono. ab.	(Cusem-AVS) (deep) [+]		0.54	0.02	
Rel. amphipod ab Rel. EPT ab.	$Fe_{SEM}$ (surface)[-] <sup>LM,P</sup> ]		0.80	0.004	



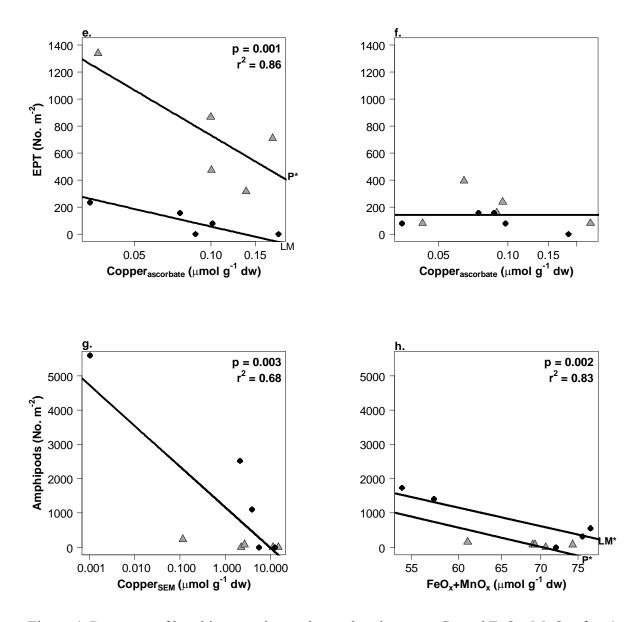
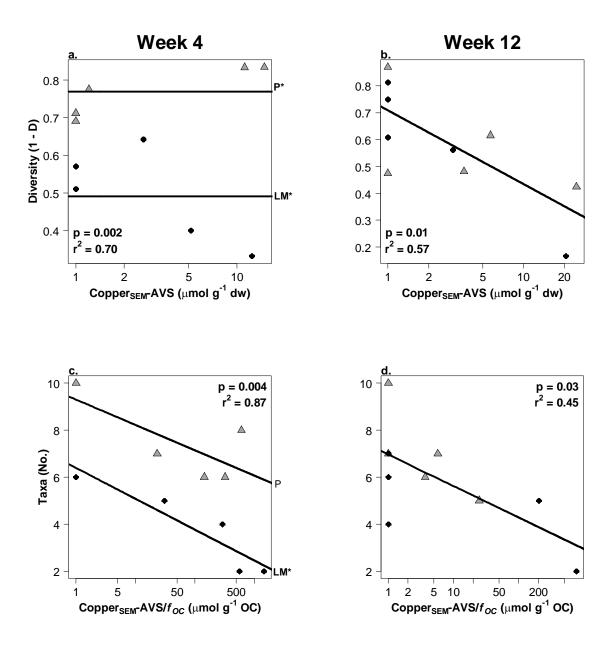


Figure 4: Response of benthic macroinvertebrate abundances to Cu and FeO<sub>x</sub>+MnO<sub>x</sub> after 4 and 12 weeks *in situ* aging. Abundance (a., b.), chironomids (c., d.), EPT (e., f.), and amphipods (g., h) responded to Cu<sub>total</sub>, Cu<sub>SEM</sub>, Cu<sub>ascorbate</sub>, or FeO<sub>x</sub>+MnO<sub>x</sub> indicated on the x-axis (note the ln scale). Bold lines represent slopes predicted from forward stepping models. For models incorporating interactions Pine (P) and Little Molasses (LM) responses are included with asterisks to indicate significant slopes. n = 10.



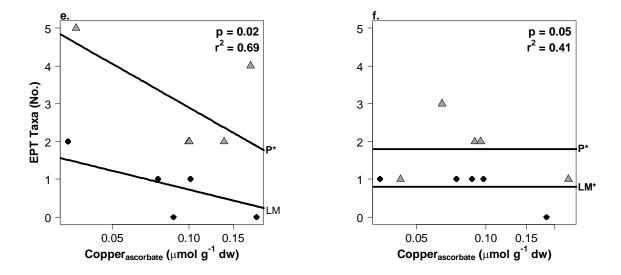
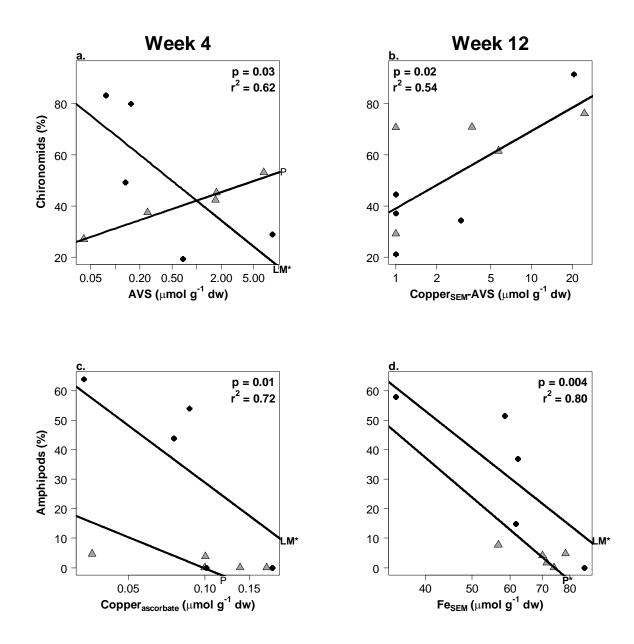


Figure 5: Response of benthic macroinvertebrate diversity and richness to Cu after 4 and 12 weeks *in situ* aging. Diversity (a., b.), taxa (c., d.), and EPT taxa (e., f) responded to Cu<sub>SEM</sub>-AVS, Cu<sub>SEM</sub>-AVS/ $f_{OC, or}$  Cu<sub>ascorbate</sub> indicated on the x-axis (note the ln scale). Bold lines represent slopes predicted from forward stepping models. For models incorporating interactions Pine (P) and Little Molasses (LM) responses are included with asterisks to indicate significant slopes. In models containing location solely as factor, significant intercepts are indicated by asterisks next to the location designation. n = 10.



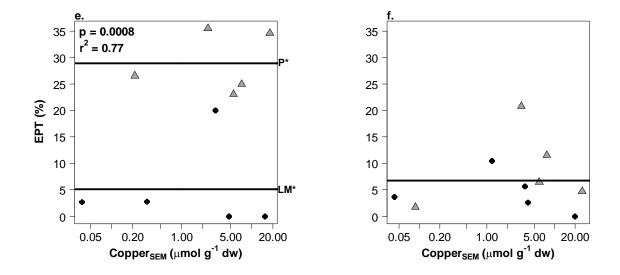


Figure 6: Response of benthic macroinvertebrate percent community composition to Cu, AVS, and Fe after 4 and 12 weeks *in situ* aging. Relative chironomid abundance (a., b.), relative amphipod abundance (c., d.), and relative EPT abundance (e., f) responded to Cusem-AVS, Cusem, Cuascorbate, AVS, or Fesem indicated on the x-axis (note the ln scale. Bold lines represent slopes predicted from forward stepping models. For models incorporating interactions Pine (P) and Little Molasses (LM) responses are included with asterisks to indicate significant slopes. In models containing location solely as factor, significant intercepts are indicated by asterisks next to the location designation. n = 10.

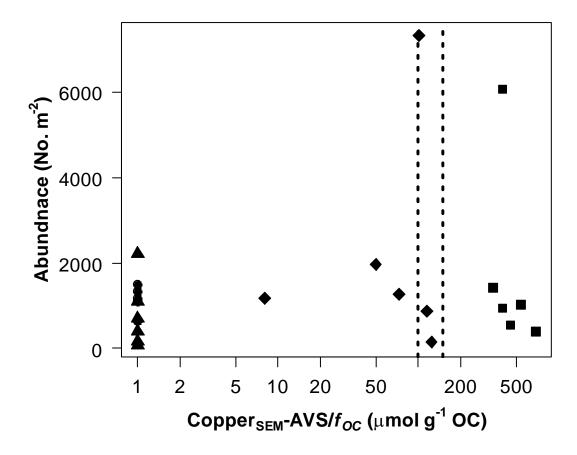


Figure 7: Invertebrate colonization of field contaminated sediments in relation to bioavailable Cu ((Cu<sub>SEM</sub> – AVS)/foc). Different symbols represent the each sediment (Ocoee 1 [♦], Ocoee 4 [▲], Ocoee 2 [■], Ocoee 3 [●]). Note the ln scale on the x-axis. Dotted lines represent the toxic threshold for chronic responses (100 – 150 µmol/g<sub>oc</sub>).

## **Literature Cited:**

- Allen H.E., Fu C. & Deng B. (1993) Analysis of acid-volatile sulfide (AVS) and simultaneously extracted metals (SEM) for the estimation of potential toxicity in aquatic sediments. *Environmental Toxicology and Chemistry* **12**, 1441–1453.
- Bouchard Jr. R.W. (2004) Guide to Aquatic Invertebrates of the Upper Midwest: identifacation manual for students, citizen monitors, and aquatic resource professionals. (Eds L.C. Ferrington Jr. & M.L. Karius), University of Minnesota.
- Briggs J.C. & Ficke J.F. (1977) *Quality of rivers of the United States, 1975 water year--based on the national stream quality accounting network (NASQAN).* United States Department of the Interior Geological Survey, Reston, Virginia.
- Brumbaugh W.G., Besser J.M., Ingersoll C.G., May T.W., Ivey C.D., Schlekat C.E., *et al.* (2013) Preparation and characterization of nickel-spiked freshwater sediments for toxicity tests: Toward more environmentally realistic nickel partitioning. *Environmental toxicology and chemistry* 32, 2482–94.
- Burton G.A. (1991) Assessing the toxicity of freshwater sediments. *Environmental Toxicology and Chemistry* **10**, 1585–1627.
- Burton G.A. (2010) Metal bioavailability and toxicity in sediments. *Critical Reviews in Environmental Science and Technology* **40**, 852–907.
- Burton G.A. & Johnston E.L. (2010) Assessing contaminated sediments in the context of multiple stressors. *Environmental toxicology and chemistry* **29**, 2625–43.
- Burton G.A., Nguyen L.T.H., Janssen C., Baudo R., McWilliam R., Bossuyt B., *et al.* (2005) Field validation of sediment zinc toxicity. *Environmental Toxicology and Chemistry* **24**, 541–553.
- Calmano W., Hong J. & Forstner U. (1993) Binding and mobilization of heavy metals in contaminated sediments affected by pH and redox potential. *Water Science and Technology* 28, 223–235.
- Cantwell M.G., Burgess R.M. & King J.W. (2008) Resuspension of contaminated field and formulated reference sediments Part I: Evaluation of metal release under controlled laboratory conditions. *Chemosphere* **73**, 1824–31.
- Carr L. & Zeller C. (2006) Copper basin mine technology case study: constructed passive wetlans system at McPherson branch. United States Environmental Protection Agency.
- Charbonneau P. & Hare L. (1998) Burrowing behavior and biogenic structures of mud-dwelling insects. *Journal of the North American Benthological Society* **17**, 239–249.

- Clements W.H. (1994) Benthic invertebrate community responses to heavy metals in the Upper Arkansas River Basin, Colorado. *Journal of the North American Benthological Society* **13**, 30–44.
- Clements W.H., Farris J.L., Cherry D.S. & Cairns J. (1989) The influence of water quality on macroinvertebrate community responses to copper in outdoor experimental streams. *Aquatic Toxicology* **14**, 249–262.
- Clements W.H., Hickey C.W. & Kidd K.A. (2012) How do aquatic communities respond to contaminants? It depends on the ecological context. *Environmental toxicology and chemistry / SETAC* **31**, 1932–40.
- Clements W.H. & Kiffney P.M. (1995) The influence of elevation on benthic community responses to heavy metals in Rocky Mountain streams. *Canadian Journal of Fisheries and Aquatic Sciences* **52**, 1966–1977.
- Costello D.M., Burton G.A., Hammerschmidt C.R., Rogevich E.C. & Schlekat C.E. (2011) Nickel phase partitioning and toxicity in field-deployed sediments. *Environmental science* & technology 45, 5798–5805.
- DeJonge M., Blust R. & Bervoets L. (2010) The relation between Acid Volatile Sulfides (AVS) and metal accumulation in aquatic invertebrates: implications of feeding behavior and ecology. *Environmental pollution* **158**, 1381–91.
- DeJonge M., Teuchies J., Meire P., Blust R. & Bervoets L. (2012a) The impact of increased oxygen conditions on metal-contaminated sediments part I: Effects on redox status, sediment geochemistry and metal bioavailability. *Water Research* **46**, 2205–2214.
- DeJonge M., Teuchies J., Meire P., Blust R. & Bervoets L. (2012b) The impact of increased oxygen conditions on metal-contaminated sediments part II: Effects on metal accumulation and toxicity in aquatic invertebrates. *Water Research* **46**, 3387–3397.
- Eggleton J. & Thomas K. V (2004) A review of factors affecting the release and bioavailability of contaminants during sediment disturbance events. *Environment international* **30**, 973–80.
- Graham M.H. (2003) Confronting multicollinearity in ecological multiple regressions. *Ecology* **84**, 2809–2815.
- Van Griethuysen C., de Lange H.J., van den Heuij M., de Bies S.C., Gillissen F. & Koelmans A.A. (2006) Temporal dynamics of AVS and SEM in sediment of shallow freshwater floodplain lakes. *Applied Geochemistry* 21, 632–642.
- Hilsenhoff W.L. (1995) Aquatic Insects of Wisconsin: keys to Wisconsin genera and notes on biology, habitat, distribution and species, 3rd edn. Natural History Museums Council University of Wisconsin-Madison.

- Honick A.S. (2013) Field Deployed Ni-Amended Sediments Shows Varying Effects In Two Central Michigan Streams. University of Michigan.
- Hutchins C.M., Teasdale P.R., Lee S.Y. & Stuart L. (2009) The effect of sediment type and pHadjustment on the porewater chemistry of copper- and zinc-spiked sediments. *Soil and Sediment Contamination* **18**, 37–41.
- Johnson J.B. & Omland K.S. (2004) Model selection in ecology and evolution. *Trends in ecology & evolution* **19**, 101–8.
- Kelderman P. & Osman A.A. (2007) Effect of redox potential on heavy metal binding forms in polluted canal sediments in Delft (The Netherlands). *Water research* **41**, 4251–61.
- Kiffney P.M. & Clements W.H. (1996) Effects of Metals on Stream Macroinvertebrate Assemblages from Different Altitudes. *Ecological Applications* **6**, 472–481.
- Klemm D.J., Lewis P.A., Fulk F. & Lazorchak J.M. (1990) *Macroinvertebrate field and laboratory methods for evaluating the biological integrity of surface waters*. United States Enivironmental Protection Agency, Cincinnati, OH.
- Kostka J.E. & Luther III G.W. (1994) Partitioning and speciation of solid phase iron in saltmarsh sediments. *Geochimica et Cosmochimica Acta* **58**, 1701–1710.
- Kristensen E. (1984) Effects of natural concentrations on nutrient exchange between a polychaete burrow in estuarine sediment and the overlying water. *Journal of Experimental Marine Biology and Ecology* **75**, 171–190.
- Kristensen E. (2000) Organic matter diagenesis at the oxic/anoxic interface in coastal marine sediments, with emphasis on the role of burrowing animals. *Hydrobiologia* **426**, 1–24.
- Lee J.-S., Lee B.-G., Luoma S.N., Choi H.J., Koh C.-H. & Brown C.L. (2000) Influence of Acid Volatile Sulfides and Metal Concentrations on Metal Partitioning in Contaminated Sediments. *Environmental Science & Technology* **34**, 4511–4516.
- Lenat D.R. (1988) Macroinvertebrates water quality assessment of streams using a qualitative collection method for benthic macroinvertebrates. *Journal of the North American Benthological Society* 7, 222–233.
- Magurran A.E. (2004) Measuring Biological Diversity, 1st edn. Blackwell Publishing.
- Marchant R., Barmuta L.A. & Chessman B.C. (1995) Influence of sample quantification and taxonomic resolution on the ordination of macroinvertebrate communities from tunning waters in Victoria, Australia. *Marine Freshwater Research* **46**, 501–506.

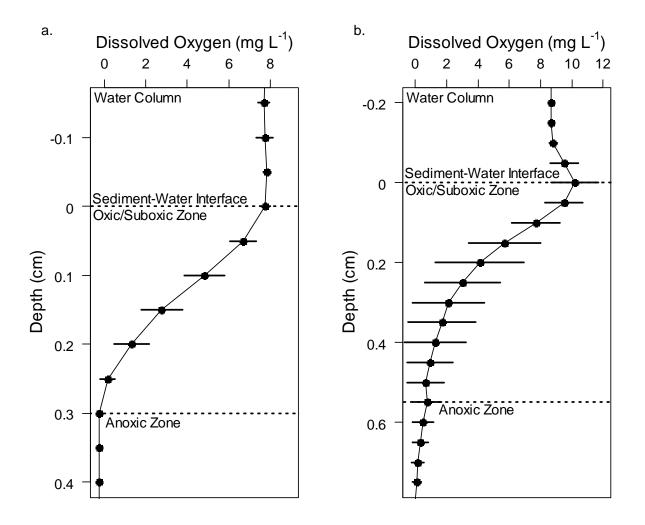
- Merritt R.W. & Cummins K.W. (1996) *Ecology and distribution of aquatic insects." An introduction to the aquatic insects of North America*, 3rd edn. Kendall/Hunt Publishing Co., Dubuque, IA.
- Meysman F.J.R., Middelburg J.J. & Heip C.H.R. (2006) Bioturbation: a fresh look at Darwin's last idea. *Trends in ecology & evolution* **21**, 688–95.
- Perin G., Fabris R., Manente S., Rebello Wagner A., Hamacher C. & Scotto S. (1997) A fiveyear study on the heavy metal pollution of Guanabara Bay sediments (Rio de Janeiro, Brazil) and evaluation of the metal bioavailability by means of geochemical speciation. *Water Research* 31, 515–522.
- Plafkin J.L., Barbour M.T., Porter K.D., Gross S.K. & Hughes R.M. (1989) *Rapid bioassesment* protocols for use in streams and rivers. Washington, D.C.
- R Development Core Team (2011) R: A language and environment for statistical computing.
- Rohr J.R. & Crumrine P.W. (2005) Effects of an herbicide and an insecticide on pond community structure and processes. *Ecological Applications* **15**, 1135–1147.
- Shepherd R.G. (1989) Correlations of permeability and grain size. Groundwater 27, 633-638.
- Simpson E.H. (1949) Measurement of diversity. *Nature* 163, 688.
- Simpson S.L., Angel B.M. & Jolley D.F. (2004) Metal equilibration in laboratory-contaminated (spiked) sediments used for the development of whole-sediment toxicity tests. *Chemosphere* **54**, 597–609.
- Simpson S.L., Apte S.C. & Batley G.E. (1998) Effect of Short-Term Resuspension Events on Trace Metal Speciation in Polluted Anoxic Sediments. *Environmental Science & Technology* 32, 620–625.
- Teuchies J., Bervoets L., Cox T.J.S., Meire P. & Deckere E. (2010) The effect of waste water treatment on river metal concentrations: removal or enrichment? *Journal of Soils and Sediments* **11**, 364–372.
- Thompson M.L. (1978) Selection of Variables in Multiple Regression: Part I. A Review and Evaluation. *International Statisical Review* **46**, 1–19.
- U.S. EPA (2007) *Method 3015A: Microwave assisted acid digestion og aqueous samples and extracts.* Washington, D.C.
- Vannote R.L., Minshall G.W., Cummins K.W., Sedell J.R. & Cushing C.E. (1980) The river coninuum concept. *Canadian Journal of Fisheries and Aquatic Sciences* 37, 130–137.

- Yanoviak S.P. & McCafferty W.P. (1996) Comparison of macroinvertebrate assemblages inhabiting pristine streams in the Huron Mountains of Michigan, USA. *Hydrobiologia* **330**, 195–211.
- Yu K.C., Tsai L.J., Chen S.H. & Ho S.T. (2001) Chemical binding of heavy metals in anoxic river sediments. *Water research* **35**, 4086–94.

## Supporting Information:

SI Table 1: Description of benthic macroinvertebrate indices.

Benthic Metric	Metric Description	citation
<b>abundance</b> (No. m <sup>-2</sup> )	The total number of individuals divided by the area of the chambers sampled $(0.0127 \text{ m}^2)$ .	Klemm <i>et al.</i> 1990; Cummins & Merritt. 1996; Plafkin <i>et al.</i> 1989
<b>chironomid abundance (</b> No. m <sup>-2</sup> )	The total number of Chironomidae individuals divided by the area of the chambers sampled (0.0127 $m^2$ ).	Klemm <i>et al.</i> 1990; Cummins & Merritt. 1996; Plafkin <i>et al.</i> 1989
<b>EPT abundance</b> (No. m <sup>-2</sup> )	The total number of Ephemeroptera, Plecoptera, and Trichoptera individuals divided by the area of the chambers sampled $(0.0127 \text{ m}^2)$ .	Klemm <i>et al.</i> 1990; Cummins & Merritt. 1996; Plafkin <i>et al.</i> 1989
amphipod abundance (No. m <sup>-2</sup> )	The total number of Amphipoda individuals divided by the area of the chambers sampled $(0.0127 \text{ m}^2)$ .	Klemm <i>et al.</i> 1990; Cummins & Merritt. 1996; Plafkin <i>et al.</i> 1989
diversity (1-D)	Simpson's D (presented as $1 - D$ ):	Simpson 1949; Magurran 2004; Klemm <i>et al.</i> 1990
	$D = \sum [n(n-1)] / (N(N-1))$	Kienini et ul. 1750
	where n is the number of individuals in each family and N is the total number of individuals in the sample. Simpson index is only applicable if $2 \le N$ .	
taxa (No.)	Number of separate taxa key to the family level when possible.	Lenat 1988; Klemm et al. 1990
EPT taxa (No.)	Number of separate EPT taxa key to the family level when possible.	Lenat 1988; Cummins & Merritt. 1996
relative chironomids (%)	Percent individuals in the total sample that belong to the Chironomidae family	Klemm et al. 1990
relative amphipod (%)	Percent individuals in the total sample that belong to the Amphipoda order	Klemm et al. 1990
relative EPT (%)	Percent individuals in the total sample that belong to the orders Ephemeroptera, Plecoptera, and Trichoptera.	Klemm et al. 1990



SI Figure 1: Depth profiles of sediment dissolved oxygen (DO) concentrations from week 4 Raisin reference (a) and Ocoee 1 (b). Zero represents the sediment-water interface, determined as the point at which DO decreased (represented by the top dotted line at 0). The increase DO in panel b is due to primary producers growing on the surface sediments. Points are represented as the mean and standard deviation of all replicates at that depth (a. n = 3; b. n = 5). The depth of the oxic layer was calculated for each replicate separately, from which a mean for each treatment was determined (represented by the dotted lines at 0.3 and 0.54 respectively).

## Appendix:

Table A1: Summary of 2-way ANOVA statistics of Raisin mean oxic layer depths. Samples affected by sedimentation at Little Molasses were not included in the analyses (Pine: n = 46, LM: n = 40).

	df	Sum Sq.	Mean Sq.	F value	p-value	
Pine						
DAY	2	0.4190	0.20948	42.583	< 0.0001	***
CU	4	0.0697	0.01743	3.543	0.017	*
DAY:CU	8	0.2188	0.02735	5.560	0.0002	***
Residuals	31	0.1525	0.00492			
Little Molas	ses					
DAY	2	0.01785	0.008925	2.264	0.12	
CU	4	0.06658	0.016644	4.221	0.009	**
DAY:CU	6	0.18886	0.031477	7.983	< 0.0001	***
Residuals	27	0.10646	0.003943			

Table A2: Summary of Tukey post-hoc multiple comparison test from ANOVA statistics of Raisin mean oxic layer depths. Samples affected by sedimentation at Little Molasses were not included in the analyses and represented by NA (Pine: n = 46, LM: n = 40).

	Pine	Little Molasses
Comparison	p adj	p adj
Day28:Ref - Day6:Ref	0.9999988	1
Day88:Ref - Day6:Ref	0.0000775	0.1115762
Day6:380 - Day6:Ref	0.9999988	0.9943185
Day28:380 - Day6:Ref	0.9998387	1
Day88:380 - Day6:Ref	0.0387505	1
Day6:750 - Day6:Ref	1	1
Day28:750 - Day6:Ref	1	0.6143701
Day88:750 - Day6:Ref	0.0000156	NA
Day6:1200 - Day6:Ref	1	0.6143701
Day28:1200 - Day6:Ref	0.9982195	0.9999987
Day88:1200 - Day6:Ref	0.762505	NA
Day6:2100 - Day6:Ref	0.9999988	0.9999947
Day28:2100 - Day6:Ref	1	0.1344645
Day88:2100 - Day6:Ref	0.9999988	0.9943185
Day88:Ref - Day28:Ref	0.0003885	0.0325249
Day6:380 - Day28:Ref	1	0.9998779
Day28:380 - Day28:Ref	1	0.9999503
Day88:380 - Day28:Ref	0.141076	0.9999503
Day6:750 - Day28:Ref	0.9998387	1

Day28:750 - Day28:Ref	0.9998387	0.7827535
Day88:750 - Day28:Ref	0.0000775	NA
Day6:1200 - Day28:Ref	0.9998387	0.7827535
Day28:1200 - Day28:Ref	0.9999999	1
Day88:1200 - Day28:Ref	0.9740397	NA
Day6:2100 - Day28:Ref	1	0.9972226
Day28:2100 - Day28:Ref	0.9999988	0.03238
Day88:2100 - Day28:Ref	1	0.9998779
Day6:380 - Day88:Ref	0.0003885	0.0061658
Day28:380 - Day88:Ref	0.0008663	0.1961057
Day88:380 - Day88:Ref	0.5791849	0.1961057
Day6:750 - Day88:Ref	0.0000347	0.1115762
Day28:750 - Day88:Ref	0.0000347	0.0008511
Day88:750 - Day88:Ref	0.9999988	NA
Day6:1200 - Day88:Ref	0.0000347	0.0008511
Day28:1200 - Day88:Ref	0.0004911	0.0673723
Day88:1200 - Day88:Ref	0.0190069	NA
Day6:2100 - Day88:Ref	0.0003885	0.3226029
Day28:2100 - Day88:Ref	0.0000775	1
Day88:2100 - Day88:Ref	0.0003885	0.0061658
Day28:380 - Day6:380	1	0.952804
Day88:380 - Day6:380	0.141076	0.952804
Day6:750 - Day6:380	0.9998387	0.9943185
Day28:750 - Day6:380	0.9998387	0.9943185
Day88:750 - Day6:380	0.0000775	NA
Day6:1200 - Day6:380	0.9998387	0.9943185
Day28:1200 - Day6:380	0.9999999	0.9999999
Day88:1200 - Day6:380	0.9740397	NA
Day6:2100 - Day6:380	1	0.8280993
Day28:2100 - Day6:380	0.9999988	0.0048729
Day88:2100 - Day6:380	1	1
Day88:380 - Day28:380	0.2458419	1
Day6:750 - Day28:380	0.9965195	1
Day28:750 - Day28:380	0.9965195	0.4111306
Day88:750 - Day28:380	0.0001736	NA
Day6:1200 - Day28:380	0.9965195	0.4111306
Day28:1200 - Day28:380	1	0.9998236
Day88:1200 - Day28:380	0.9965195	NA
Day6:2100 - Day28:380	1	1
Day28:2100 - Day28:380	0.9998387	0.2460511
Day88:2100 - Day28:380	1	0.952804
Day6:750 - Day88:380	0.0190069	1
Day28:750 - Day88:380	0.0190069	0.4111306

Day88:750 - Day88:380	0.2458419	NA
Day6:1200 - Day88:380	0.0190069	0.4111306
Day28:1200 - Day88:380	0.2265443	0.9998236
Day88:1200 - Day88:380	0.9015432	NA
Day6:2100 - Day88:380	0.141076	1
Day28:2100 - Day88:380	0.0387505	0.2460511
Day88:2100 - Day88:380	0.141076	0.952804
Day28:750 - Day6:750	1	0.6143701
Day88:750 - Day6:750	0.000007	NA
Day6:1200 - Day6:750	1	0.6143701
Day28:1200 - Day6:750	0.9811439	0.9999987
Day88:1200 - Day6:750	0.5791849	NA
Day6:2100 - Day6:750	0.9998387	0.9999947
Day28:2100 - Day6:750	1	0.1344645
Day88:2100 - Day6:750	0.9998387	0.9943185
Day88:750 - Day28:750	0.000007	NA
Day6:1200 - Day28:750	1	1
Day28:1200 - Day28:750	0.9811439	0.9730778
Day88:1200 - Day28:750	0.5791849	NA
Day6:2100 - Day28:750	0.9998387	0.2460511
Day28:2100 - Day28:750	1	0.0005836
Day88:2100 - Day28:750	0.9998387	0.9943185
Day6:1200 - Day88:750	0.000007	NA
Day28:1200 - Day88:750	0.0000877	NA
Day88:1200 - Day88:750	0.0041989	NA
Day6:2100 - Day88:750	0.0000775	NA
Day28:2100 - Day88:750	0.0000156	NA
Day88:2100 - Day88:750	0.0000775	NA
Day28:1200 - Day6:1200	0.9811439	0.9730778
Day88:1200 - Day6:1200	0.5791849	NA
Day6:2100 - Day6:1200	0.9998387	0.2460511
Day28:2100 - Day6:1200	1	0.0005836
Day88:2100 - Day6:1200	0.9998387	0.9943185
Day88:1200 - Day28:1200	0.9982195	NA
Day6:2100 - Day28:1200	0.9999999	0.9962976
Day28:2100 - Day28:1200	0.9982195	0.0824583
Day88:2100 - Day28:1200	0.9999999	0.9999999
Day6:2100 - Day88:1200	0.9740397	NA
Day28:2100 - Day88:1200	0.762505	NA
Day88:2100 - Day88:1200	0.9740397	NA
Day28:2100 - Day6:2100	0.9999988	0.4111306
Day88:2100 - Day6:2100	1	0.8280993
Day88:2100 - Day28:2100	0.9999988	0.0048729

Model					df	<b>F-Value</b>	R <sup>2</sup>	p-valu
Surface								•
ln(AV	VS) ~ Cu <sub>total</sub> * Time + Lo	cation			6 and 17	37.24	0.93	< 0.000
		Coeffic.	t- value	p - value				
	Alpha	2.3	7.281	1.28e <sup>-6</sup>	***			
	Cutot	-0.19	-10.053	1.43e <sup>-8</sup>	***			
	Day 28	-0.57	-1.326	0.2025				
	Day 88	-1.1	-2.823	0.0117	*			
	LM	0.26	1.050	0.3083				
	Cutot:Day 28	0.04	1.468	0.1604				
	Cutot:Day 88	0.06	2.392	0.0286	*			
sqrt(C	$Cu_{SEM}$ ) ~ $Cu_{total}$ * Time +	Location			6 and 17	43.78	0.94	< 0.000
		Coeffic.	t- value	p - value				
	Alpha	-1.14	-0.88	0.391				
	Cutot	0.78	10.017	1.51e <sup>-8</sup>	***			
	Day 28	1.01	0.571	0.576				
	Day 88	0.41	0.243	0.811				
	LM	-0.64	-0.621	0.543				
	Cu <sub>TOT</sub> :Day 28	-0.26	-2.291	0.035	*			
	Cutot:Day 88	-0.03	-0.321	0.752				
Cu <sub>asco</sub>	$_{rbate} \sim Cu_{total} * Time + Loc$	cation			6 and 17	26.84	0.90	< 0.000
		Coeffic.	t- value	p - value				
	Alpha	0.07	3.891	0.0011	**			
	Cu <sub>TOT</sub>	0.005	5.028	0.0001	***			
	Day 28	-0.19	-0.807	0.43				
	Day 88	-0.023	-1.027	0.31				
	LM	-0.04	-2.940	0.009	**			
	Cu <sub>TOT</sub> :Day 28	-0.0006	-0.376	0.711				
	Cu <sub>TOT</sub> :Day 88	0.004	2.781	0.012	*			
Fesem	$\sim Cu_{total} * Time + Locat$	tion			6 and 17	1.338	032	0.29
Mn <sub>SEN</sub>	$M \sim Cu_{total} * Time + Location + Locatii + Location + Location + Location$	ation			6 and 17	1.998	0.41	0.122
FeO <sub>x</sub> -	$+MnO_x \sim Cu_{total} * Time +$	- Location			6 and 17	3.47	0.55	0.02
		Coeffic.	t- value	p - value				
	alpha	98.884	15.35	<0.0001	***			
	Cutot	-0.66	-1.382	0.18				
	Time 28	-31.62	-2.9	0.010	*			
	Time 88	-22.69	-2.2	0.04	*			
	Location 2	-17.88	-2.8	0.01	*			
	Time 28: Location 2	0.18	0.26	0.79				
	Time 88: Location 2	0.87	1.35	0.19				

Table A3: Description of ANCOVA results modeling Cu fractions and sediment physicochemical fractions as a function of total Cu and Time with Location as a block. Transformations are noted in model equations and were determined by a Shapiro-Wilk test for normality.

## Deep

$ln(AVS) \sim Cu_{total} * Time + Lo$		_		6 and 22	37.24	0.93	< 0.0001
Alpha	Coeffic. <b>2.18</b>	t- value 6.634	p - value 1.14e <sup>-6</sup>	***			
Ситот	-0.16	-8.888	9.85e <sup>-9</sup>	***			
Day 28	-0.15	-0.362	0.721				
Day 88	-0.43	-1.006	0.326				
LM	0.17	0.768	0.450				
Cu <sub>TOT</sub> :Day 28	-0.006	-0.245	0.809				
Cu <sub>TOT</sub> :Day 88	0.013	0.506	0.618				
$Cu_{SEM} \sim Cu_{total} * Time + Locat$	tion			6 and 22	29.67	0.89	< 0.0001
	Coeffic.	t- value	p - value				
Alpha	-1.02	-0.664	0.5135				
Cutot	0.72	8.33	3.02e-8	***			
Day 28	1.44	0.69	0.49				
Day 88	-0.500	-0.25	0.81				
LM	-1.66	-1.54	0.14				
Cu <sub>TOT</sub> :Day 28	-0.28	-2.35	0.03	*			
Cu <sub>TOT</sub> :Day 88	0.04	0.35	0.73				
$Cu_{ascorbate} \sim Cu_{total} * Time + Lo$	cation			6 and 22	50.04	0.93	< 0.0001
	Coeffic.	t- value	p - value				
Alpha	0.04	4.570	0.00015	***			
Cu <sub>tot</sub>	0.004	9.347	4.07e-9	***			
Day 28	0.01	0.954	0.35				
Day 88	0.003	0.230	0.82				
LM	-0.01	-1.763	0.09	•			
Cu <sub>TOT</sub> :Day 28	-0.0002	-0.301	0.766				
Cutot:Day 88	0.0009	1.298	0.208				
$Fe_{SEM} \sim Cu_{total} * Time + Locat$	ion			6 and 22	1.72	0.32	0.16
$Mn_{SEM} \sim Cu_{total} * Time + Loca$	tion			6 and 22	1.013	0.21	0.44
FeO <sub>x</sub> +MnO <sub>x</sub> ~ Cu <sub>total</sub> * Time +	Location			6 and 22	0.67	0.15	0.67

Table A4: Multiple regression analysis (forward stepping) of macroinvertebrate colonization in response to copper spiked sediments 4 weeks *in situ* aging. Models were selected from forward stepping procedure once all significant parameters were added. Final models are labeled in bold. Where location was included an interaction term they are designated by colon and as factor by River. Parameters are listed in the order which they were added to the models. n=10

		df	F-value	r <sup>2</sup>	p-value
Density					
	α + Cu <sub>SEM</sub> (surface)	1 and 8	47.34	0.98	< 0.001
	α				
Taxa					
	α (Cusem -AVS)/f <sub>oc</sub> (surface):River + River + (Cusem -AVS)/f <sub>oc</sub> (surface)	<b>3</b> and <b>6</b>	13.65	0.87	0.0043
	$\alpha + (Cu_{SEM} - AVS)/f_{oc}$ (surface):River + River	3 and 6	13.65	0.87	0.0043
	$\alpha + (Cu_{SEM}-AVS)/f_{oc}$ (surface):River	2 and 7	12.41	0.78	0.005
	α				
Chirono	mid Density				
	α + Cu <sub>SEM</sub> (surface) + Fe <sub>SEM</sub> (deep)	2 and 7	43.11	0.92	0.0001
	$\alpha + Cu_{SEM}$ (surface)	1 and 8	32.37	0.80	0.0005
	α				
% Chire	onomids				
	α + AVS (surface):River + Fesem (deep):River	4 and 5	24.05	0.95	0.001
	$\alpha + Fe_{SEM}$ (deep):River	2 and 7	5.653	0.62	0.03
	α				
EPT De	nsity				
	α + Cuascorbate (deep):River	2 and 7	21.35	086	0.001
	α				
% EPT					
	$\alpha$ + River	1 and 8	27.37	0.77	0.0008
	α				
EPT Ta	xa				
	$\alpha + Cu_{ascorbate}$ (deep):River	2 and 7	7.752	0.69	0.02
	α				
Simpsor	1's D (1 – D)				
	$\alpha$ + River	1 and 8	19.07	0.70	0.002
	α				
Amphin	od Density				
	$\alpha + Cu_{SEM}$ (deep)	1 and 8	17.4	0.68	0.003
	α				
% Amp	hipods				
P	$\alpha$ + Cu <sub>ascorbate</sub> (deep):River + Cu <sub>ascorbate</sub> (deep)	2 and 7	8.947	0.72	0.01
	$\alpha + Cu_{ascorbate}$ (deep):River	2 and 7	8.947		0.01
	α				

Table A5: Multiple regression analysis (forward stepping) of macroinvertebrate colonization in response to copper spiked sediments after 12 weeks *in situ* aging. Models were selected from forward stepping procedure once all significant parameters were added. Final models are labeled in bold. Where location was included an interaction term they are designated by colon and as factor by River. Parameters are listed in the order which they were added to the models. n=10

		df	F-value	r <sup>2</sup>	p-value
Density					<u>^</u>
	α + Cu <sub>total</sub> (surface) + FeO <sub>x</sub> +MnO <sub>x</sub> (deep):River + FeO <sub>x</sub> +MnO <sub>x</sub> (deep)	<b>3</b> and <b>6</b>	31.65	0.64	0.03
	$\alpha$ + Cu <sub>total</sub> (surface) + FeO <sub>x</sub> +MnO <sub>x</sub> (deep):River	3 and 6	31.65	0.64	0.03
	$\alpha + Cu_{total}$ (surface)	1 and 8	15.56		
	α				
Taxa			< 101	o 4 <b>-</b>	0.02
	$\alpha$ + (Cusem -AVS)/ $f_{oc}$ (deep) $\alpha$	1 and 8	6.481	0.45	0.03
% Chironom	id				
	$\alpha$ + Cu <sub>SEM</sub> –AVS (deep) $\alpha$	1 and 8	9.306	0.54	0.02
Chironomid	Density α				
EPT Density					
	α				
% EPT					
	α				
EPT Taxa					
	$\alpha$ + River	1 and 8	5.556	0.41	0.05
	α				
Simpson's D					
	$\alpha$ + CU <sub>SEM</sub> – AVS (deep) + Cu <sub>total</sub> (deep):RIVER	3 and 6	14.23	0.64	0.006
	$\alpha + CU_{SEM} - AVS (deep)$ $\alpha$	1 and 8	10.67	0.57	0.01
Amphipod D	ensity				
-	α + FeO <sub>x</sub> +MnO <sub>x</sub> (deep):River + FeO <sub>x</sub> +MnO <sub>x</sub> (deep)	2 and 7	17.08	0.83	0.002
	$\alpha$ + FeO <sub>x</sub> +MnO <sub>x</sub> (deep):River $\alpha$	2 and 7	17.08	0.83	0.002
% Amphipod	ls				
	α + Fe <sub>SEM</sub> (surface):River + Fe <sub>SEM</sub> (surface)	2 and 7	13.68	0.80	0.004
	$\alpha$ + Fe <sub>SEM</sub> (surface):River $\alpha$	2 and 7	13.68	0.80	0.004

			_	Abundance	(No. m <sup>-2</sup> )		Simpson	Richne	ss (No.)	Rela	tive abundance	e (%)
	Wk.	Cu	Ab.	amphipod	Chiron.	EPT	(1 <b>-</b> D)	Taxa	EPT	EPT	Amphipods	Chiron.
		0	1024	79	472	394	0.78	5	2	46.15	7.69	38.46
		380	1181	0	709	472	0.76	4	3	60.00	0.00	40.00
	1	750	394	0	315	79	0.70	3	2	80.00	0.00	20.00
		1200	1496	0	1024	394	0.84	8	6	68.42	0.00	26.32
		2100	866	0	394	236	0.87	6	3	45.45	0.00	27.27
		0	5039	236	1339	2677	0.69	10	5	26.56	4.69	53.13
e		380	2441	0	866	1102	0.71	6	2	35.48	0.00	45.16
Pine	4	750	2047	79	472	866	0.78	7	2	23.08	3.85	42.31
		1200	1260	0	315	472	0.83	6	2	25.00	0.00	37.50
		2100	2047	0	709	551	0.83	8	4	34.62	0.00	26.92
		0	4567	79	79	3228	0.47	7	1	1.72	1.72	70.69
		380	1890	79	394	551	0.87	10	3	20.83	4.17	29.17
	12	750	2441	0	157	1732	0.48	6	2	6.45	0.00	70.97
		1200	2047	157	236	1260	0.62	7	2	11.54	7.69	61.54
		2100	1654	79	79	1260	0.42	5	1	4.76	4.76	76.19
		0	2362	315	236	1811	0.40	3	1	10.00	13.33	76.67
		380	2205	394	157	1654	0.42	3	1	7.14	17.86	75.00
	1	750	1890	315	394	1102	0.61	4	1	20.83	16.67	58.33
		1200	2126	236	79	1732	0.33	4	1	3.70	11.11	81.48
		2100	945	0	79	866	0.17	2	1	8.33	0.00	91.67
es		0	8740	5591	236	2520	0.51	6	2	2.70	63.96	28.83
Little Molasses		380	5748	2520	157	2835	0.57	5	1	2.74	43.84	49.32
Mo	4	750	2047	1102	0	394	0.64	4	0	0.00	53.85	19.23
tle		1200	394	0	79	315	0.40	2	1	20.00	0.00	80.00
Lit		2100	472	0	0	394	0.33	2	0	0.00	0.00	83.33
		0	2126	315	79	945	0.75	7	1	3.70	14.81	44.44
		380	1496	551	157	315	0.81	6	1	10.53	36.84	21.05
	12	750	2756	1417	157	1024	0.61	4	1	5.71	51.43	37.14
	- 4	1200	2750	1732	79	1024	0.56	5	1	2.63	57.89	34.21
		2100	945	0	0	866	0.17	2	0	0.00	0.00	91.67

Table A6: Macroinvertebrate colonization of copper spiked sediments (Raisin) in Pine and LittleMolasses Rivers. Cu represents nominal copper values.

				Abundance (No. m <sup>-2</sup> )			Simpson	Richne	ess (No.)	Rela	tive abundance	e (%)
	Wk.	Cu	Ab.	amphipod	Chiron.	EPT	(1-D)	Taxa	EPT	EPT	Amphipods	Chiron.
		170	157	0.00	0	157	0	1	100.00	0.00	0.00	1
	1	600	709	0.64	0	315	394	3	44.44	0.00	55.56	2
	1	1250	551	0.86	0	157	79	4	28.57	0.00	14.29	1
		1600	630	0.82	0	394	157	4	62.50	0.00	25.00	2
		170	1969	0.89	79	472	472	10	24.00	4.00	24.00	3
Pine	4	600	157	1.00	0	79	0	2	50.00	0.00	0.00	1
Pi	-	1250	394	1.00	0	79	79	5	20.00	0.00	20.00	1
		1600	1339	0.81	0	630	236	7	47.06	0.00	17.65	2
		170	1260	0.84	0	157	157	7	12.50	0.00	12.50	1
	12	600	79	1.00	0	0	79	1	0.00	0.00	100.00	0
	12	1250	1024	0.60	0	0	394	3	0.00	0.00	38.46	0
		1600	1102	0.75	0	236	551	6	21.43	0.00	50.00	2
		170	866	0.56	236	0	551	3	0.00	27.27	63.64	0
	1	600	1102	0.14	79	0	1024	2	0.00	7.14	92.86	0
	1	1250	945	0.45	79	79	709	4	8.33	8.33	75.00	1
		1600	1181	0.47	157	79	866	4	6.67	13.33	73.33	1
Little Molasses		170	7323	0.59	4252	157	1890	6	2.15	58.06	25.81	2
lola	4	600	2205	0.42	1654	0	394	3	0.00	75.00	17.86	0
e M	4	1250	6063	0.54	2126	79	3543	5	1.30	35.06	58.44	1
Littl		1600	1496	0.62	630	157	709	3	10.53	42.11	47.37	1
Π		170	1181	0.54	787	79	79	4	6.67	66.67	6.67	1
	10	600	394	0.80	0	0	157	3	0.00	0.00	40.00	0
	12	1250	1417	0.77	630	0	236	6	0.00	44.44	16.67	0
		1600	1102	0.79	472	236	236	6	21.43	42.86	21.43	2

Table A7: Macroinvertebrate colonization of field contaminated sediments (Ocoee) in Pine andLittle Molasses Rivers. Cu represents nominal copper values.

	Wk	Cu	OC	AVS	Cutotal	Cusem	Cuasebt.	Cusem-AVS	FeO <sub>x</sub> +MnO <sub>x</sub>	Cusem-AVS /foc
			(%)				(µmol g <sup>-1</sup>	dw)		$(\mu mol g_{oc}^{-1})$
		0	0.03	11.47	0.05	0.07	0.07	-11.41	78.16	-356.14
		380	0.03	1.95	6.01	3.33	0.10	1.38	73.41	45.58
	1	750	0.03	2.49	10.63	4.23	0.09	1.74	74.19	62.39
		1200	0.03	0.99	16.20	10.25	0.17	9.26	71.46	286.19
		2100	0.03	0.05	26.02	23.92	0.26	23.87	76.12	840.16
		0	0.03	6.65	0.22	0.22	0.07	-6.54	64.60	-223.66
()		380	0.03	1.76	5.76	2.37	0.06	3.77	107.35	144.20
Pine	4	750	0.03	1.70	12.72	5.52	0.08	0.67	68.78	23.20
щ		1200	0.02	0.25	13.61	7.18	0.11	6.92	75.64	323.71
		2100	0.03	0.04	34.14	18.10	0.23	18.05	64.19	608.96
		0	0.03	3.52	0.01	0.09	0.04	-5.08	68.94	-157.69
		380	0.03	1.04	4.52	3.21	0.08	2.17	74.26	86.95
	12	750	0.03	0.73	8.48	5.94	0.10	5.20	70.59	170.78
		1200	0.03	0.20	18.78	7.67	0.18	7.47	61.08	273.33
		2100	0.03	0.10	30.89	25.52	0.39	25.42	69.24	811.27
		0	0.03	10.90	0.10	0.15	0.04	-10.75	71.80	-344.51
		380	0.03	3.56	5.67	2.49	0.08	-1.06	60.58	-36.18
	1	750	0.03	1.17	13.09	8.47	0.09	7.30	64.07	250.48
		1200	0.03	0.27	16.58	9.60	0.09	9.33	58.61	328.32
		2100	0.03	0.07	28.50	19.15	0.16	19.09	75.44	678.93
Little Molasses		0	0.01	8.47	0.15	0.04	0.03	-8.43	62.95	-1005.51
las		380	0.01	0.13	0.36	0.32	0.04	0.19	78.86	31.27
Mc	4	750	0.01	0.68	5.83	4.68	0.08	4.00	64.45	286.83
le		1200	0.01	0.16	2.53	3.01	0.09	2.86	59.80	567.70
Litt		2100	0.01	0.08	9.63	15.51	0.13	15.43	63.82	1447.64
_ <b>_</b> .		0	0.01	7.24	0.11	0.04	0.03	-7.20	75.61	-674.12
		380	0.02	2.57	4.92	1.19	0.07	-1.38	76.78	-58.60
	12	750	0.01	0.73	1.48	3.62	0.05	2.89	57.36	506.61
		1200	0.004	0.18	1.06	4.03	0.05	3.85	54.04	863.69
		2100	0.02	0.08	28.10	20.12	0.23	20.03	71.95	805.11

Table A8: Sediment geochemical characteristics of Cu spiked surface sediments (Raisin) in Pine and Little Molasses Rivers.

	Wk	Cu	OC	AVS	Cutotal	Cusem	Cuascbt.	CUSEM-AVS	FeO <sub>x</sub> +MnO <sub>x</sub>	Cusem-AVS /foc
			(%)				(µmol g <sup>-1</sup>	dw)		$(\mu mol g_{oc}^{-1})$
	1	0	0.03	11.44	0.01	0.03	0.05	-11.40	78.16	-367.04
		380	0.03	4.62	5.52	2.20	0.06	-2.42	73.41	-79.38
		750	0.03	1.15	11.32	6.32	0.08	-1.02	74.19	-33.87
		1200	0.03	0.38	17.66	10.22	0.10	9.84	71.46	317.68
		2100	0.03	0.12	28.05	26.28	0.21	26.16	76.12	856.99
		0	0.03	12.08	0.11	0.12	0.04	-11.96	64.60	-383.64
Ð		380	0.03	2.44	6.75	2.25	0.10	-0.19	107.35	-6.97
Pine	4	750	0.03	1.44	14.03	2.65	0.10	1.21	68.78	43.98
щ		1200	0.03	0.17	21.08	11.30	0.14	11.13	75.64	383.27
		2100	0.03	0.03	29.04	14.77	0.18	14.74	64.19	490.72
		0	0.04	9.91	0.01	0.03	0.04	-9.88	68.94	-279.11
		380	0.03	1.94	5.90	0.99	0.06	-0.95	74.26	-32.53
	12	750	0.03	0.78	8.72	4.43	0.09	3.65	70.59	145.17
		1200	0.03	0.24	12.89	5.97	0.10	5.73	61.08	193.05
		2100	0.03	0.08	35.40	24.72	0.23	24.64	69.24	870.68
	1	0	0.03	8.59	0.18	0.07	0.04	-8.52	71.80	-279.37
		380	0.03	4.33	7.51	2.18	0.06	-2.15	60.58	-77.05
		750	0.03	0.98	12.19	3.99	0.07	3.01	64.07	100.20
		1200	0.03	0.58	17.15	6.56	0.09	5.97	58.61	210.55
<b>10</b>		2100	0.03	0.04	33.91	19.10	0.17	19.05	75.44	689.62
Little Molasses	4	0	0.03	6.97	0.00	0.00	0.03	-6.97	62.95	-248.02
olas		380	0.02	3.81	5.02	2.10	0.08	-1.72	78.86	-77.06
Ŭ		750	0.03	1.19	13.57	3.82	0.09	2.62	64.45	101.73
tle		1200	0.02	0.40	12.60	5.60	0.10	5.20	59.80	283.74
Lit		2100	0.02	0.06	33.90	12.41	0.18	12.35	63.82	536.45
	12	0	0.03	12.67	0.00	0.07	0.03	-12.60	75.61	-412.99
		380	0.03	4.12	6.57	1.09	0.09	-3.03	76.78	-100.35
		750	0.03	1.75	12.12	1.73	0.07	-0.02	57.36	-0.69
		1200	0.01	0.34	8.24	3.36	0.10	3.02	54.04	203.01
		2100	0.03	0.07	24.74	20.71	0.18	20.64	71.95	753.77

Table A9: Sediment geochemical characteristics of Cu spiked deep sediments (Raisin) in Pine and Little Molasses Rivers.

	Wk.	Cu	OC	AVS	Cutotal	Cusem	Cuasebt.	Cusem-AVS	FeO <sub>x</sub> +MnO <sub>x</sub>	Cusem-AVS /foc
			%			$(\mu mol g_{oc}^{-1})$				
	1	170	0.01	0.01	4.05	0.68	0.33	-0.18	26.25	-34.15
		600	0.02	3.89	9.65	1.48	0.04	-2.89	41.62	-178.70
		1250	0.02	0.00	27.29	7.91	0.22	4.43	54.90	257.24
		1600	0.01	7.98	29.53	2.83	0.03	-3.46	50.03	-248.89
	4	170	0.01	0.35	4.76	0.82	0.08	-0.36	22.87	-38.78
Pine		600	0.02	3.83	9.53	1.57	0.01	-2.70	40.87	-128.63
Pi		1250	0.02	0.08	29.68	12.82	0.16	4.24	54.42	230.55
		1600	0.02	7.50	30.60	2.74	0.01	-3.49	66.62	-208.51
	12	170	0.01	0.02	3.30	0.49	0.21	-0.53	24.65	-81.37
		600	0.02	3.20	8.16	1.24	0.01	-2.34	34.05	-128.23
		1250	0.02	0.01	29.99	9.15	0.20	4.08	41.23	241.62
		1600	0.01	6.49	30.92	3.10	0.01	-2.83	46.77	-219.25
	1	170	0.01	0.00	4.54	0.74	0.32	-0.10	25.54	-16.19
		600	0.02	2.88	10.24	0.90	0.03	-2.65	33.16	-159.55
		1250	0.02	0.01	29.15	7.20	0.24	4.60	47.08	255.31
es		1600	0.01	9.21	29.99	2.86	0.42	-6.47	68.90	-437.92
Little Molasses	4	170	0.01	0.02	2.84	0.55	0.15	-0.45	21.29	-84.95
lol		600	0.02	3.35	7.45	0.90	0.01	-2.95	34.26	-195.86
Z		1250	0.02	0.01	22.53	7.12	0.18	3.61	51.45	203.07
ittle		1600	0.19	14.53	20.99	3.49	0.01	-11.45	60.34	-59.17
Ē	12	170	0.01	0.37	3.33	0.44	0.06	-1.06	13.95	-121.47
		600	0.02	2.17	7.09	1.05	0.01	-1.63	32.85	-87.39
		1250	0.02	0.01	19.34	5.56	0.19	2.08	40.13	128.21
		1600	0.01	6.42	23.08	6.37	0.03	-2.23	30.33	-153.43

Table A10: Sediment geochemical characteristics of field contaminated surface sediments (Ocoee) in Pine and Little Molasses Rivers.

	Wk.	Cu	OC	AVS	Cutotal	Cusem	Cuasebt.	CUSEM-AVS	FeO <sub>x</sub> +MnO <sub>x</sub>	Cusem-AVS /foc
			%			$(\mu mol goc^{-1})$				
	1	170	0.01	0.00	4.48	0.61	0.20	0.80	26.25	114.23
		600	0.02	2.42	9.88	2.24	0.04	0.67	41.62	-10.79
		1250	0.02	0.01	26.04	4.94	0.22	6.54	54.90	284.38
		1600	0.01	9.01	29.98	3.33	0.04	-2.86	50.03	-401.37
	4	170	0.01	0.02	4.52	0.42	0.13	0.57	22.87	67.05
Pine		600	0.02	1.84	8.81	1.38	0.01	0.09	40.87	-21.51
Pi		1250	0.02	0.02	27.83	6.03	0.21	7.77	54.42	285.65
		1600	0.02	8.74	28.02	2.19	0.01	-3.59	66.62	-401.18
		170	0.01	0.02	3.14	0.46	0.12	0.63	24.65	81.88
	12	600	0.02	2.16	7.75	1.30	0.01	-0.27	34.05	-45.12
		1250	0.02	0.01	24.86	6.96	0.20	8.73	41.23	397.87
		1600	0.01	8.51	29.26	3.67	0.01	-2.10	46.77	-369.37
	1	170	0.01	0.01	4.03	0.64	0.18	0.84	25.54	108.37
		600	0.02	4.11	9.35	0.77	0.03	-2.87	33.16	-198.46
		1250	0.02	0.01	25.97	7.25	0.03	9.23	47.08	389.23
es		1600	0.02	9.92	31.38	2.47	0.03	-6.00	68.90	-491.81
ass	4	170	0.01	0.02	2.87	0.48	0.13	0.67	21.29	76.75
Little Molasses		600	0.01	3.91	7.73	0.85	0.01	-2.54	34.26	-209.88
		1250	0.02	0.01	24.19	8.34	0.21	10.21	51.45	479.13
		1600	0.01	7.38	25.60	2.75	0.01	-2.33	60.34	-333.50
Ē	12	170	0.01	0.01	3.02	0.35	0.08	0.48	13.95	58.74
		600	0.02	2.65	6.57	0.76	0.01	-0.33	32.85	-112.90
		1250	0.02	0.02	25.58	6.56	0.01	8.66	40.13	420.40
		1600	0.02	7.79	21.66	2.39	0.07	-3.82	30.33	-306.07

Table A11: Sediment geochemical characteristics of field contaminated deep sediments (Ocoee) in Pine and Little Molasses Rivers.