

---

1     **Propeller-like Nanorod-Upconversion Nanoparticle Assemblies**  
2     **with Intense Chiroptical Activity and Luminescence Enhancement**  
3                     **in Aqueous Phase**

4     Xiaoling Wu<sup>1,2#</sup>, Liguang Xu<sup>1,2#</sup>, Wei Ma<sup>1,2</sup>, Liqiang Liu<sup>1,2</sup>, Hua Kuang<sup>1,2\*</sup>, Nicholas A. Kotov<sup>3</sup>,  
5                     Chuanlai Xu<sup>1,2\*</sup>

6     <sup>1</sup> State Key Lab of Food Science and Technology, Jiangnan University, Wuxi, Jiangsu, 214122, PRC;

7     <sup>2</sup> International Joint Research Laboratory for Biointerface and Biodetection, and School of Food Science and Technology,  
8     Jiangnan University, Wuxi, Jiangsu, 214122, PRC;

9     <sup>3</sup>Department of Chemical Engineering, University of Michigan, Ann Arbor, MI, 48109-2136, USA

10  
11     #These authors contributed to this paper equally.

12     \*Corresponding Authors: kuangh@jiangnan.edu.cn; xcl@jiangnan.edu.cn

13  
14     **ABSTRACT:** In this study, we demonstrate that propeller-like nanoscale assemblies can also  
15     be produced with exceptionally intense chiroptical activity and strong luminescence, using  
16     gold nanorods and upconversion nanoparticles via DNA self-assembly. By adjusting the sizes  
17     of building blocks, the circular dichroism intensity of the tetramer reached 80.9 mdeg, and  
18     g-factor value was  $2.1 \times 10^{-2}$ . And the enhancement factor of upconversion luminescence was

This is the author manuscript accepted for publication and has undergone full peer review but has not been through the copyediting, typesetting, pagination and proofreading process, which may lead to differences between this version and the [Version of Record](#). Please cite this article as [doi: 10.1002/adma.201601261](https://doi.org/10.1002/adma.201601261).

This article is protected by copyright. All rights reserved.

---

1 as high as 21.3 in aqueous phase, which was mainly attributed to the localized electric field  
2 enhancement of NRs in the tetramers. Moreover, the optically active tetramers can be used  
3 the bioanalysis of oligonucleotide targets with limits of detection of 13.2 aM. The developed  
4 attomolar biosensor may play a significant role in cancer biomarker analysis for early disease  
5 diagnosis and environmental monitoring.

---

1 Chiral assemblies are currently one of the most dynamic research fields<sup>[1]</sup> due to their wide  
2 application in biosensing,<sup>[2]</sup> and chiral catalytic,<sup>[3]</sup> as well as photonic devices.<sup>[4]</sup> The  
3 conformation organization of the building blocks determines the functionalities of the  
4 assemblies, especially the optical properties.<sup>[5]</sup> There are now chiroptical nanoassemblies  
5 helices,<sup>[6]</sup> pyramids,<sup>[7]</sup> scissor-like nanoparticle (NP) pairs,<sup>[2a, 8]</sup> and other superstructures,<sup>[9]</sup>  
6 which are comprised of supramolecular polymers,<sup>[10]</sup> metal NPs,<sup>[1b, 2a, 11]</sup> semiconductor  
7 NPs,<sup>[12]</sup> carbonaceous nanomaterials,<sup>[13]</sup> and nanocomposites.<sup>[9c, 14]</sup> Some studies have also  
8 reported propeller-like chiral structures at the molecular level for organic complexes.<sup>[5, 15]</sup>  
9 However, to date, there are no reports on propeller-like chiral assemblies at the nanoscale.  
10 to their scale and strong electron polarizability of inorganic compounds,<sup>[6c, 11]</sup> metal  
11 nanomaterials-mediated assemblies can give rise to intense circular dichroism (CD) at visible  
12 or near infrared wavelengths. The majority of the current research is focused on the  
13 chiral nanomaterials.<sup>[1b, 2a, 6c, 11, 16]</sup> Other studies have been performed on the plasmon-exciton  
14 coupling between metal nanomaterials and semiconductor NPs or chiral molecules.<sup>[9a, 12]</sup> To  
15 date, there are few reports on the chiral assembly of metal-upconversion nanostructures and  
16 their chiral mechanism is not clear, although multiparticle assemblies reporting upconversion  
17 nanoparticles (UCNPs) have been reported.<sup>[7c, 14, 17]</sup>

18 Lanthanide-doped luminescence upconversion nanomaterials can emit higher energy  
19 than the excitation source.<sup>[2c]</sup> Compared with the traditional down-conversion luminescence  
20 materials, they have a narrower band emission, longer fluorescence lifetime, and are  
21 which has made them promising luminescent materials in the biomedical field,<sup>[2d, 18]</sup> in areas  
22 such as biological imaging,<sup>[19]</sup> and in cancer diagnosis and treatment.<sup>[20]</sup> However, the  
23 conventional luminescence efficiency of lanthanide-doped luminescence upconversion  
24 nanomaterials does not exceed 1%, which restricts their widespread use in the fields of  
25 biomedical science and photonic devices.<sup>[2c, 20a, 21]</sup> It is known that the plasmonic resonance

---

1 coupling of metal nanomaterials can enhance the luminescence of fluorescent dyes<sup>[1d, 22]</sup> and  
2 NPs.<sup>[23]</sup> Enhancement of UCNP light emission upon assembly with plasmonic NPs or  
3 is also possible.<sup>[2c, 21, 24]</sup> However, the plasmon-enhanced upconversion luminescence has  
4 studied for rigid solid metal supports. While this is interesting for electro-optical devices,  
5 biological imaging and diagnostics<sup>[2c, 24d, 24e]</sup> require UCNP-plasmon systems to be  
6 in biological fluids, which has not been realized so far. An additional consideration regarding  
7 the significance of such systems is that optical properties of the deformable colloids  
8 plasmonic NPs can be strongly dependent on their geometry which can be utilized for  
9 biomedical purposes.<sup>[1b, 2a, 16, 25]</sup> As we and other scientists in this field learned recently, even  
10 seemingly minor changes in the geometry of such assemblies are expected to strongly affect  
11 rotatory optical activity.<sup>[1c, 26]</sup>

12 In this study, we fabricated gold nanorod (NR)-UCNP tetramers in solution by adopting  
13 the DNA-driven self-assembly strategy to overcome the above-mentioned obstacles. By  
14 choosing specific building blocks—i.e. lengths of DNA sequences, aspect ratios of NRs, and  
15 sizes of UCNPs—the nanoscale engineered tetramers possessed strong and tailorable chiral  
16 activity in the visible region. Tuning the distance between the UCNP and NRs allowed us to  
17 realize the enhancement of upconversion luminescence with a 21.3-fold increase in aqueous  
18 solution. We also observed that these tetramers have propeller-like geometry that gave rise to  
19 strong chiroptical activity. In addition to having fundamental importance as an analog of  
20 propeller-like chiral molecules known from coordination chemistry, the chiral geometry of  
21 such systems makes possible the attomolar DNA detection with two-model capabilities.

22 Chiral tetramers were assembled from NRs and UCNPs functionalized with  
23 complementary DNA (see Experimental section). Preferential binding of thiol-terminated  
24 DNA to the end facets of NRs allowed for end site blocking of the pre-synthesized NRs with  
25 a molar ratio of helper DNA to NRs of approximately 80.<sup>[2a, 27]</sup> The subsequent addition of

---

1 thiolated DNA (DNA1) led to attachment to the side facets of NRs, with a molar ratio of 1:1  
2 for DNA1 to NRs. To protect the NRs from excessive DNA modifications and make them  
3 stable for the hybridization system, thiolated polyethylene glycol (PEG) was added to the  
4 solution at a ratio of 200 (PEG to NR-DNA1).<sup>[2a]</sup> Simultaneously, the maleimide-PEGylated  
5 UCNPs (Yb/Er-doped NaGdF<sub>4</sub> NPs) were modified with the complementary thiolated DNA  
6 sequence (DNA2) due to the classic thiol-maleimide “click” reaction,<sup>[28]</sup> and the molar ratio  
7 of DNA2 strands to UCNPs was approximately 3, denoted as UCNP-DNA2. To obtain the  
8 tetramers, a 33% excess of the DNA-modified NRs was intentionally maintained; the added  
9 molar ratio was four-fold that of the conjugated UCNPs. The variation in building blocks  
10 brought about the finely controlled construction of the expected NR-UCNP tetramer  
11 assemblies, denoted as AUT (**Figure 1a**).

12 To study the chiroplasmonic properties of the tetramers,  $20 \pm 2.7$  nm UCNPs (UCNP<sub>20</sub>)  
13 and a longitudinal plasmon band ( $\lambda_L$ ) of 750 nm for NRs (average length  $50.2 \pm 2.1$  nm and  
14 an aspect ratio of 3.3, denoted as NR<sub>750</sub>) were used as building blocks for the nanoassemblies,  
15 with 30 base pairs (bp, about 10 nm) of DNA sequence (Figure S1 and Table S1). DNA  
16 hybridization led to the formation of tetramers (AUT<sub>750</sub>), and the corresponding NR-UCNP  
17 assemblies were identified by transmission electron microscopy (TEM, Figure 1c and Figure  
18 S2) and cryo-TEM tomography (Figure 1d and Figure S3). Their formation was also  
19 confirmed by dynamic light scattering measurements (DLS, Figure S4a). The average  
20 hydrodynamic diameter ( $D_h$ ) of the NR-UCNP tetramers was  $86 \pm 3.2$  nm, without larger  
21 aggregations. As a control, the DLS curve of the mixture of NRs and UCNPs was also  
22 measured, and showed a very broad peak in the range of 15–100 nm. The  $\lambda_L$  band of the  
23 tetramers experienced a small blue shift of about 6 nm after the assembly, whereas the  
24 transverse peak changed very little, probably due to the large distance between the NRs,  
25 which led to weak coupling of transverse plasmons (**Figure 2b**).<sup>[29]</sup>

---

1 As seen from the CD spectra in Figure 2a, the tetramers revealed CD bands in the  
2 ultraviolet (UV) section of the spectrum (200–250 nm) attributed to the DNA bridges.<sup>[6a]</sup>  
3 Concurrent with the formation of tetramers, a strong increase in chiroptical activity in the  
4 plasmonic region (300–800 nm) of the spectrum was observed (Figure 2a). In relation to  
5 other NR assemblies, the CD signal of the tetramers displayed a bisignate shape in the  
6 spectrum between the  $\lambda_L$  from 600 to 800 nm.<sup>[2a, 6c]</sup> The spectral “wave” for 600–750 nm was  
7 negative (–32.7 mdeg,  $\lambda_L$  713 nm), and was positive for 750–800 nm (32.9 mdeg,  $\lambda_L$  776  
8 nm).<sup>[30]</sup> A new CD peak in the 280–380 nm spectral window appeared with the assembly of  
9 tetramers, and was assigned to the absorption of UCNPs, consistent with their UV-vis spectra  
10 (Figure 2b). The chiral anisotropy factor,  $g$ , of the assembled tetramers reached  $1.06 \times 10^{-2}$ .  
11 As a comparison, the DNA-conjugated NRs and UCNPs showed strong chiroptical activity in  
12 the UV region (200–250 nm). However, all the tested permutations of modified and  
13 unmodified starting NPs (NRs and UCNPs), exhibited weak CD, if any, bands at 250–800  
14 nm, in agreement with previous reports (Figure 2a).<sup>[8, 31]</sup>

15 In order to better understand the mechanism of rotatory optical activity of the NR-UCNP  
16 assemblies, tetramers with their various geometries and various constituent building blocks  
17 were assembled. Taking into consideration that plasmonic coupling strongly depends on the  
18 distance between the plasmonic particles,<sup>[32]</sup> we first studied the chiral optical activity of the  
19 tetramers assembled with varying lengths of DNA sequence (11, 18, 24, 30, 36, 45 bp). From  
20 the spectral measurements in Figure 2c, all the CD curves possessed similar spectral waves,  
21 with a bisignate signal for plasmonic bands at 600–800 nm. As the DNA length decreased  
22 from 45 to 11 bp, the  $\lambda_L$  of the CD bands significantly shifted to blue by 23 and 22 nm for the  
23 positive and negative part, respectively. This blue-shift of the CD spectra was in good  
24 agreement with the UV-vis spectra, although the latter showed a smaller blue shift (about 18  
25 nm, Figure 2d). Synchronously, CD signals showed a progressive increase in amplitude from

---

1 + 9.3 to + 80.9 mdeg along with a shorter DNA sequence. The  $g$ -factor value reached  $2.1 \times 10^{-2}$ ,  
2 <sup>2</sup>, which was much higher than the molecular system ( $10^{-3} - 10^{-5}$ ) and comparable to the  
3 highest obtained for NP assemblies reported so far.<sup>[8, 33]</sup>

4 In addition to alterations in the length of the DNA strands, NRs with different  $\lambda_L$   
5 absorption were also investigated. Another two sets of NRs, with  $\lambda_L$  absorption of 700 nm  
6 (aspect ratio of 2.6) and 800 nm (aspect ratio of 3.7), were also chosen for tetramer assembly.  
7 These tetramers were denoted as AUT<sub>700</sub> and AUT<sub>800</sub>, respectively (Figures 1 e–f and S5–  
8 S8). The average  $D_h$  value of the tetramers increased from  $75.3 \pm 2.1$  to  $101 \pm 6.7$  nm as the  
9  $\lambda_L$  absorption increased from 700 to 800 nm (Figure S4b). This change in  $D_h$  correlated with  
10 the statistical size measurements from the TEM images (Figures S5 and S6). A strong  
11 chiroplasmonic response was observed with varying  $\lambda_L$  absorption of starting NRs. The  
12 plasmonic CD spectra displayed a marked red shift (89 nm, at the negative peak),  
13 corresponding to that of the absorption changes (Figure 2 e–f). Increased  $\lambda_L$  of starting NRs  
14 (from 700 to 800 nm) led to stronger CD intensity, with an absolute CD value from 21.2 to  
15 38.7 mdeg. These results demonstrate that chiroptical tetramers can be constructed and  
16 engineered by adjustment of the DNA sequence and plasmonic absorption of NRs, and the  
17 largest  $g$ -factor value occurred at tetramer with 11 bp of the DNA sequence,  $\lambda_L$  absorption of  
18 750 nm of the NRs, and 20 nm UCNPs.

19 The three-dimensional (3D) reconstruction of TEM tomography revealed that the three  
20 constitutive NRs in the tetramer were not parallel to each other, but had propeller-like  
21 conformations, which was likely due to the balancing of electrostatic attraction and repulsion  
22 forces, similar to many other nanoscale assemblies with biopolymers.<sup>[5, 8, 15d, 34]</sup> As expected,  
23 such geometry of the tetramers was not superimposable with their mirror images (Figure 1 b  
24 and d–h; Figures S3 and S7–S9), which resulted in strong chiroptical activity.<sup>[15a, 15c]</sup> Similarly  
25 to the case of other chiral NPs or NRs assemblies triggered by biomolecules,<sup>[2a, 8, 26a]</sup> the

---

1 strong chiroptical activity of propeller-like NR-UCNP tetramers was a difference of two  
2 types of enantiomers. The preference for one enantiomer as opposed to another was related to  
3 connecting DNA bridges and the general preference of charged nanoscale NRs as the  
4 conformation with minimal energy.<sup>[2a, 26a]</sup>

5 To further understand the origin of the tetramer's chirality, chiroptical simulations were  
6 carried out based on the Frequency Domain Finite Integral (FDFI) method, with 20 nm  
7 UCNPs and a 10 nm gap between the NRs and UCNPs. The length of the NRs was set to 50.2  
8 nm, and the aspect ratio was 2.6, 3.3, and 3.7, respectively (Figure 2 g–h). The simulated  
9 UV-vis spectra of the tetramers showed obvious blue-shift (15 nm, 24 nm, and 27 nm)  
10 compared with the original NRs (NR<sub>700</sub>, NR<sub>750</sub>, NR<sub>800</sub>; Figure 2h), which was consistent with  
11 the experimental spectra (Figure 2f). The calculated CD curve exhibited a predicted bisignate  
12 signature in the 600–800 nm region (Figure 2g), which correlated with the  $\lambda_L$  chiral  
13 absorption of NRs as observed experimentally (Figure 2e).

14 Based on the above test results, the chiral optical activity of NR-UCNP tetramers was  
15 attributed to the unique propeller-like geometry of the assemblies (Figure 1 b and d–h;  
16 Figures S3 and S7–S9), and was confirmed by theoretical simulation (Figure 2g). In addition,  
17 it must be pointed out that other factors should be considered including plasmon-plasmon  
18 coupling of the NRs.<sup>[6a, 35]</sup> The induced plasmonic CD originating from the electronic  
19 “imprint” of the DNA helix on electronic oscillations is likely to play a minor role based on  
20 previous studies.<sup>[26a, 36]</sup> It should also be pointed out that the propeller-like geometry creates a  
21 strong plasmonic field, resulting in the appearance of chiroptical bands in UCNPs (Figure 2 a,  
22 c, and e), which is unusual for these NPs and certainly worthy of further study.

23 In addition to the chiroptical activity of the NR-UCNP tetramers, the luminescence of  
24 the assemblies was investigated. The AUT<sub>750</sub> tetramer made from UCNP<sub>20</sub>, NR<sub>750</sub> and 30 bp  
25 of DNA, was used as the initial model. Under 980 nm laser excitation, up-conversion



---

1 luminescence (UCL) of Yb/Er-doped UCNPs was observed with the typical emission peaks  
2 located at 529, 546, and 662 nm, which were attributed to the electronic transitions in  $\text{Er}^{3+}$   
3 from  $^2\text{H}_{11/2}$  and  $^4\text{S}_{3/2}$  states to the  $^4\text{I}_{15/2}$  state, and from the  $^4\text{F}_{9/2}$  state to the  $^4\text{I}_{15/2}$  state.<sup>[28]</sup> The  
4 intensity of UCL gradually increased in the first 75 min as tetramers assembled matching the  
5 typical timing of DNA hybridization (**Figure 3**). In the following 60 min of the experiment,  
6 the intensity was unchanged.

7 The luminescence enhancement due to the formation of tetramers can be shown as  $I/I_0$ ,  
8 where  $I_0$  and  $I$  represent the UCL intensity observed for original UCNPs and assembled  
9 tetramers, respectively. The UCL enhancement factors of the tetramers were 3.8, 5.3, and 6.6  
10 for the emission bands at 529, 546 nm, and 662 nm, respectively (Figure 3g). In a control  
11 experiment, which indicated the significance of the assembly for UCL enhancement, we  
12 determined the change in UCL intensity after DNA modification when mixed with the NR  
13 dispersion. Without DNA bridges and tetramer formation, UCL peaks showed a slight  
14 decrease (Figure 3b). We also considered the possibility of UCL fluctuation due to the **buffer**  
15 **conditions and pH value of the assembly system. As seen in Figure S10, there was no obvious**  
16 **variation in UCL in the tested buffer conditions and pH range 7.0 to 8.5, consistent with the**  
17 **related  $D_n$  and electrokinetic potentials measurements (Figure S4c).**

18 To evaluate the influence of distance from the surface of NRs to UCNPs on the UCL  
19 spectra,  $\text{AUT}_{750}$  tetramers were assembled with different DNA strand lengths (11, 18, 24, 30,  
20 36 and 45 bp). The UCL signal was strongly quenched when the DNA sequence was shorter  
21 than 18 bp, especially with a short DNA sequence of 11 bp. The UCL quenching was  
22 ascribed to the non-radiative energy transfer (NRET) to plasmonic states in gold.<sup>[17, 23]</sup> When  
23 the distance between the NRs and NPs increased, i.e. DNA length increased to 24 bp and 36  
24 bp, UCL enhancement compared with that of starting UCNPs was observed (Figure 3c). The  
25 maximum enhancement of UCL was obtained at 30 bp, after which the UCL intensity

---

1 decreased to that of the original UCNPs due to low electric field intensity induced by the  
2 excess distance.<sup>[2c]</sup>

3 When NR<sub>750</sub> were replaced with NR<sub>900</sub> (Figure 3d), the strongest UCL was also obtained  
4 for 30 bp DNA bridges. The maximum enhancement factor of the assembled AUT<sub>900</sub>  
5 tetramers was up to 8.5 and 12.4 at the bands of 529 and 546 nm, respectively (Figure 3g).  
6 For the band at 662 nm, the UCL of AUT<sub>900</sub> tetramers was 21.3-fold enhanced compared  
7 with the original UCNPs, which was approximately 3.3 times higher than the UCL of AUT<sub>750</sub>  
8 tetramers at the same band (Figure 3g).

9 Further experiments were carried out to reveal the relationship between the UCL signal  
10 and the  $\lambda_L$  absorption (700, 750, 800, 850, and 900 nm) of tetramers. The UCL intensity  
11 progressively increased with increasing  $\lambda_L$  absorption (700–900 nm), with a corresponding  
12 enhancement factor of 2.9, 3.8, 5.3, 6.7, and 8.5 at 529 nm, 3.1, 5.3, 7.3, 9.5, and 12.4 at 546  
13 nm, and 3.7, 6.6, 10.7, 15.0, and 21.3 at 662 nm, respectively (Figure 3 e and h). In these  
14 experiments, optimized enhancement was with AUT<sub>900</sub>, probably due to the closer proximity  
15 to the excitation wavelength of UCNPs (980 nm) for  $\lambda_L$  absorption of AUT<sub>900</sub> compared to  
16 the others.<sup>[17]</sup>

17 Dependence of the UCL intensity on plasmonic resonance was further confirmed by the  
18 tetramers assembled with larger UCNPs with a diameter of  $35 \pm 2.2$  nm (UCNP<sub>35</sub>). The  
19 influence of varied  $\lambda_L$  absorption of NRs (700, 750, 800, 850, and 900 nm) on UCL was  
20 investigated based on the AUT tetramers established by UCNP<sub>35</sub> and a DNA sequence of 30  
21 bp. As shown in Figure 3 (f and h), the UCL enhancement factor was 2.2, 3.7, 4.7, 7.2, and  
22 9.6 at 529 nm, and 2.1, 3.4, 5.0, 6.9, and 9.1 at 546 nm; whereas at 662 nm, the enhancement  
23 factor was 2.7, 5.2, 8.1, 11.5, and 16.0, respectively. These were all smaller in the case of  
24 tetramers comprising UCNP<sub>20</sub>, indicating that the plasmonic-UCL enhancement was better  
25 for smaller sized UCNPs.

---

1        Herein, it was clearly revealed that the maximum UCL enhancement factor occurred at a  
2 distance of approximately 10 nm (30 bp of the DNA sequence),  $\lambda_L$  absorption of 900 nm of  
3 the NRs, and 20 nm UCNPs in the tested tetramers. To the best of our knowledge, this is the  
4 first time that plasmon-enhanced UCL has been successfully achieved in solution.  
5 Plasmon-enhancement of UCL is ascribed to two effects: (1) increased excitation rate and  
6 efficiency arising from localized electric field enhancement by the collective oscillation  
7 coupling of NRs in the tetramers;<sup>[17, 24e]</sup> and (2) enhanced radiative decay rate and effective  
8 emission induced by the UCL emission coupling with plasmon resonance of NRs.<sup>[23, 24d, 37]</sup>  
9 Electromagnetic simulations were performed to further confirm the UCL enhancement. The  
10 electric field of the AUT<sub>750</sub> patterns assembled with 20 nm UCNPs, were calculated for three  
11 representative distances (3.7 nm, 10 nm, 15 nm) from the surface of the UCNPs to NRs. As  
12 shown in **Figure 4** (a–c), the electric field intensity of the tetramers gradually decreased with  
13 increasing distance (6.51 V m<sup>-1</sup> for a 3.7 nm gap, 5.86 V m<sup>-1</sup> for a 10 nm gap, 5.34 V m<sup>-1</sup> for  
14 a 15 nm gap), corresponding to the larger UCL enhancement factor of tetramers with a gap of  
15 10 nm (5.3-fold at 546 nm) than a gap of 15 nm (1.1-fold at 546 nm) (Figure 3g). In contrast,  
16 as the distance approached 3.7 nm, the NRET (declined with inverse fourth power of  
17 localized electric field<sup>[17, 24a]</sup>) overwhelmingly dominated,<sup>[1d]</sup> therefore the UCL intensity was  
18 significantly quenched, although with a higher electric field than the above two cases. When  
19 the distance of AUT<sub>750</sub> was set at 10 nm, the tetramers constructed with 35 nm UCNPs  
20 showed lower electric field intensity (5.22 V m<sup>-1</sup>) than the tetramers built with 20 nm  
21 UCNPs, producing a smaller UCL enhancement factor (Figures 3h and 4d). When the  
22 distance of the tetramers and the size of UCNPs were set at 10 nm and 20 nm, the AUT<sub>900</sub>  
23 possessed a higher electric field (12.0 V m<sup>-1</sup>) and thus resulted in stronger UCL than AUT<sub>750</sub>  
24 (Figures 3g and 4e).

---

1 Strong chiroptical and luminescent activity of the propeller-like assemblies enabled  
2 two-model biosensing of oligonucleotides, which improved the reliability and versatility of  
3 the method. Hepatitis A virus *Vall7* polyprotein gene (*HVA*), was selected as the target.<sup>[38]</sup> As  
4 illustrated in Figure 5a, NR-UCNP tetramers were fabricated with two partly-complementary  
5 DNA sequences (Table S1), by forming hairpin-like DNA structures between NR and UCNP.  
6 In the presence of oligonucleotide targets, hairpin-like DNA strands of the tetramers were  
7 extended, due to their specific biorecognition; and resulted in a longer gap length, which  
8 caused the reduction of both UCL and CD signal intensities. In the case of the chiroplasmonic  
9 method, AUT<sub>750</sub> was adopted, and the calibration curve was obtained with a difference in CD  
10 intensity ( $CD_{776\text{ nm}} - CD_{713\text{ nm}}$ ) (Figure 5 b and c). It exhibited an excellent linear response to  
11 the DNA target over the range of  $3.3 \times 10^{-8}$  to  $3.3 \times 10^{-5}$  nM. The limit of detection (LOD)  
12 was found to be 13.2 aM (Figure 5c, see Supporting Information for details). The UCL  
13 method was carried out using AUT<sub>750</sub> (Figure 5d). The standard curve was plotted with the  
14 UCL peak intensity at 546 nm as the *y* axis, and the logarithmic DNA concentration as the *x*  
15 axis. As shown in Figure 5e, the obtained LOD of 20.3 aM was almost two times higher than  
16 that of the CD technique with the AUT<sub>750</sub> assemblies, which was attributed to strong  
17 polarization rotation by the propeller-like geometry of the plasmonic nanoassemblies and  
18 bisignate characteristic of the CD spectra.<sup>[2a, 8]</sup> As a control, AUT<sub>900</sub> tetramers can also be  
19 used as a luminescence biosensor (Figure S11). Besides confirming the sensitivity, we  
20 evaluated the specificity of the developed method, by using non-complementary  
21 hybridization DNA sequences as negative controls. And the results (Figure S12) showed that  
22 no obvious changes of CD signal or luminescence was observed, indicating good specificity  
23 of the biosensor. Furthermore, we also evaluated the practical application of the developed  
24 biosensor in a complex biological matrix, such as human serum. The results in Table S2  
25 demonstrated excellent recoveries of analyte by the biosensor. In comparison with other  
26 conventional methods (such as ELISA, HPLC, plasmon resonance shift assay), therefore, the

---

1 developed method possessed better sensitivity and signal-to-noise ratio, as well as more  
2 accurate quantitative determination. In addition, dual signal-detection model of the method  
3 enabled the accurate and distinct identification of cancer biomarkers, which holds promising  
4 potential for early disease diagnosis.

5 In summary, these propeller-like nanoscale tetramers not only offer strong  
6 chiroplasmonic and enhanced upconversion luminescent properties, but also enable DNA  
7 detection with an unusually low LOD. Looking forward, the UCNPs-based chiroplasmonic  
8 assemblies show potential for efficient bioimaging and light-guided therapy *in vitro* and *in*  
9 *vivo*. In addition, the ultrasensitive bioanalysis nanoplatform may also satisfy the urgent  
10 requirements for early medical diagnosis, environmental monitoring, anti-counterfeiting and  
11 fingerprint forensics.

12

Author Manuscript

---

1 **Supporting Information**

2 The experimental details and supplementary figures and tables can be found in the Supporting  
3 Information.

4  
5 **Acknowledgements**

6 This work is financially supported by the National Natural Science Foundation of China (21471068,  
7 31400848, 21471128).

8

Author Manuscript

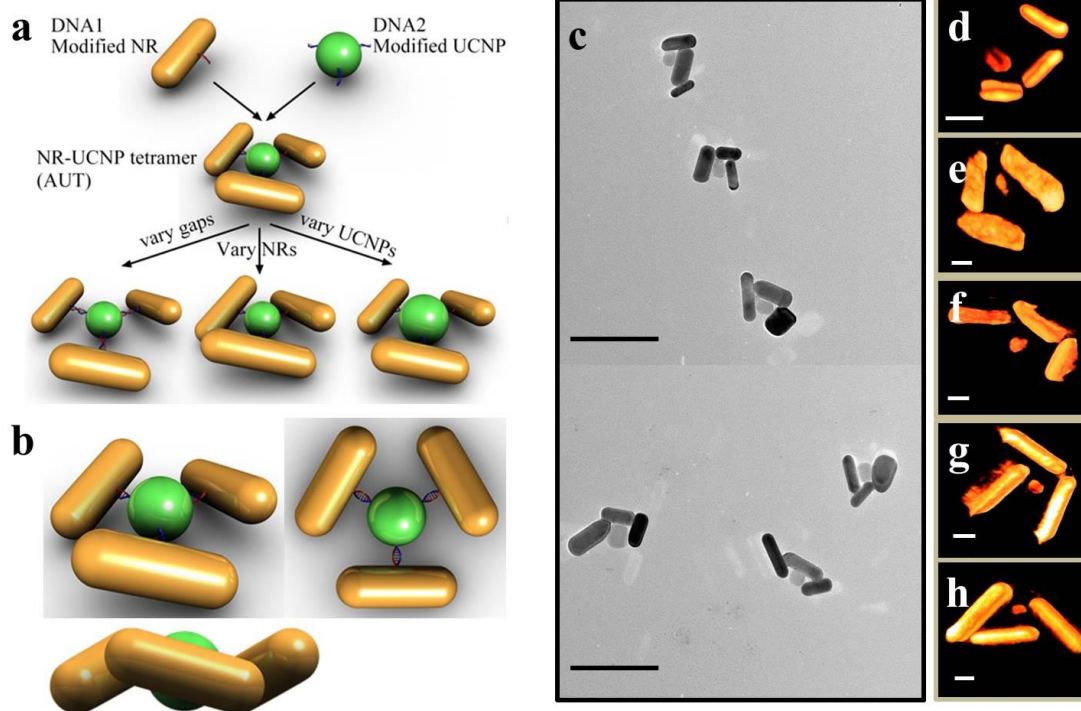
- 
- 1 [1] a) J. Yeom, B. Yeom, H. Chan, K. W. Smith, S. Dominguez-Medina, Joong H. Bahng, G. Zhao, W.-S.  
2 Chang, S.-J. Chang, A. Chuvilin, D. Melnikau, A. L. Rogach, P. Zhang, S. Link, P. Král, N. A. Kotov,  
3 *Nat. Mater.* **2015**, *14*, 66; b) R. Schreiber, N. Luong, Z. Fan, A. Kuzyk, P. C. Nickels, T. Zhang, D. M.  
4 Smith, B. Yurke, W. Kuang, A. O. Govorov, T. Liedl, *Nat. Commun.* **2013**, *4*, 2948; c) S. Ostovar pour,  
5 L. Rocks, K. Faulds, D. Graham, ParchaňskýVáclav, P. Bouř, E. W. Blanch, *Nat. Chem.* **2015**, *7*, 591; d)  
6 G. P. Acuna, F. M. Möller, P. Holzmeister, S. Beater, B. Lalkens, P. Tinnefeld, *Science* **2012**, 338, 506.
- 7 [2] a) W. Ma, H. Kuang, L. Xu, L. Ding, C. Xu, L. Wang, N. A. Kotov, *Nat. Commun.* **2013**, *4*, 2689; b) Z.  
8 Li, Z. Zhu, W. Liu, Y. Zhou, B. Han, Y. Gao, Z. Tang, *J. Am. Chem. Soc.* **2012**, *134*, 3322; c) J. Zhou, Q.  
9 Liu, W. Feng, Y. Sun, F. Li, *Chem. Rev.* **2015**, *115*, 395; d) W. Zheng, P. Huang, D. Tu, E. Ma, H. Zhu,  
10 X. Chen, *Chem. Soc. Rev.* **2015**, *44*, 1379.
- 11 [3] a) K. V. S. Ranganath, J. Kloesges, A. H. Schäfer, F. Glorius, *Angew. Chem. Int. Ed.*, **2010**, *49*, 7786; b)  
12 K. Sawai, R. Tatum, T. Nakahodo, H. Fujihara, *Angew. Chem. Int. Ed.*, **2008**, *47*, 6917.
- 13 [4] a) G. Chen, H. Agren, T. Y. Ohulchanskyy, P. N. Prasad, *Chem. Soc. Rev.* **2015**, *44*, 1680; b) X. Li, F.  
14 Zhang, D. Zhao, *Chem. Soc. Rev.* **2015**, *44*, 1346.
- 15 [5] T. Nakano, Y. Okamoto, *Chem. Rev.* **2001**, *101*, 4013.
- 16 [6] a) A. Kuzyk, R. Schreiber, Z. Fan, G. Pardatscher, E.-M. Roller, A. Högele, F. C. Simmel, A. O.  
17 Govorov, T. Liedl, *Nature* **2012**, *483*, 311; b) G. Singh, H. Chan, A. Baskin, E. Gelman, N. Repnin, P.  
18 Král, R. Klajn, *Science* **2014**, *345*, 1149; c) X. Lan, X. Lu, C. Shen, Y. Ke, W. Ni, Q. Wang, *J. Am. Chem.*  
19 *Soc.* **2015**, *137*, 457.
- 20 [7] a) W. Yan, L. Xu, C. Xu, W. Ma, H. Kuang, L. Wang, N. A. Kotov, *J. Am. Chem. Soc.* **2012**, *134*, 15114;  
21 b) A. J. Mastroianni, S. A. Claridge, A. P. Alivisatos, *J. Am. Chem. Soc.* **2009**, *131*, 8455; c) S. Li, L. Xu,  
22 W. Ma, X. Wu, M. Sun, H. Kuang, L. Wang, N. A. Kotov, C. Xu, *J. Am. Chem. Soc.* **2016**, *138*, 306.
- 23 [8] X. Wu, L. Xu, L. Liu, W. Ma, H. Yin, H. Kuang, L. Wang, C. Xu, N. A. Kotov, *J. Am. Chem. Soc.* **2013**,  
24 *135*, 18629.
- 25 [9] a) Z. Zhu, J. Guo, W. Liu, Z. Li, B. Han, W. Zhang, Z. Tang, *Angew. Chem.* **2013**, *125*, 13816; b) C. Tan,  
26 X. Qi, Z. Liu, F. Zhao, H. Li, X. Huang, L. Shi, B. Zheng, X. Zhang, L. Xie, Z. Tang, W. Huang, H.  
27 Zhang, *J. Am. Chem. Soc.* **2015**, *137*, 1565; c) W. Liu, Z. Zhu, K. Deng, Z. Li, Y. Zhou, H. Qiu, Y. Gao,  
28 S. Che, Z. Tang, *J. Am. Chem. Soc.* **2013**, *135*, 9659; d) C. Hao, L. Xu, W. Ma, X. Wu, L. Wang, H.  
29 Kuang, C. Xu, *Adv. Funct. Mater* **2015**, *25*, 5816.

- 
- 1 [10] L. Zhang, L. Qin, X. Wang, H. Cao, M. Liu, *Adv. Mater.* **2014**, *26*, 6959.
- 2 [11] S. H. Jung, J. Jeon, H. Kim, J. Jaworski, J. H. Jung, *J. Am. Chem. Soc.* **2014**, *136*, 6446.
- 3 [12] A. Ben-Moshe, B. M. Maoz, A. O. Govorov, G. Markovich, *Chem. Soc. Rev.* **2013**, *42*, 7028.
- 4 [13] P. Deria, C. D. Von Bargen, J.-H. Olivier, A. S. Kumbhar, J. G. Saven, M. J. Therien, *J. Am. Chem. Soc.*  
5 **2013**, *135*, 16220.
- 6 [14] L.-L. Li, Y. Lu, *J. Am. Chem. Soc.* **2015**, *137*, 5272.
- 7 [15] a) F. Vera, R. M. Tejedor, P. Romero, J. Barberá, M. B. Ros, J. L. Serrano, T. Sierra, *Angew. Chem.*  
8 **2007**, *119*, 1905; b) J. Barberá, L. Puig, P. Romero, J. L. Serrano, T. Sierra, *J. Am. Chem. Soc.* **2006**, *128*,  
9 4487; c) A. Martinez, L. Guy, J.-P. Dutasta, *J. Am. Chem. Soc.* **2010**, *132*, 16733; d) F. Grillo, V.  
10 Mugnaini, M. Oliveros, S. M. Francis, D.-J. Choi, M. V. Rastei, L. Limot, C. Cepek, M. Pedio, S. T.  
11 Bromley, *J. Phys. Chem. Lett.* **2012**, *3*, 1559.
- 12 [16] X. Shen, A. Asenjo-Garcia, Q. Liu, Q. Jiang, F. J. García de Abajo, N. Liu, B. Ding, *Nano Lett.* **2013**, *13*,  
13 2128.
- 14 [17] A. L. Feng, M. L. You, L. Tian, S. Singamaneni, M. Liu, Z. Duan, T. J. Lu, F. Xu, M. Lin, *Sci. Rep.* **2015**,  
15 *5*, 7779.
- 16 [18] a) Y. I. Park, K. T. Lee, Y. D. Suh, T. Hyeon, *Chem. Soc. Rev.* **2015**, *44*, 1302; b) X. Liu, C.-H. Yan, J. A.  
17 Capobianco, *Chem. Soc. Rev.* **2015**, *44*, 1299; c) B. Zhou, B. Shi, D. Jin, X. Liu, *Nat. Nanotechnol.* **2015**,  
18 *10*, 924; d) X. Zhao, L. Xu, M. Sun, W. Ma, X. Wu, H. Kuang, L. Wang, C. Xu, *Small* **2016**,  
19 DOI: 10.1002/sml.201503629.
- 20 [19] a) J.-N. Liu, W.-B. Bu, J.-L. Shi, *Acc. Chem. Res.* **2015**, *48*, 1797; b) R. Deng, F. Qin, R. Chen, W.  
21 Huang, M. Hong, X. Liu, *Nat. Nanotechnol.* **2015**, *10*, 237; c) R. Li, Z. Ji, J. Dong, C. H. Chang, X.  
22 Wang, B. Sun, M. Wang, Y.-P. Liao, J. I. Zink, A. E. Nel, T. Xia, *ACS Nano* **2015**, *9*, 3293.
- 23 [20] a) G. Chen, H. Qiu, P. N. Prasad, X. Chen, *Chem. Rev.* **2014**, *114*, 5161; b) D. Yang, P. a. Ma, Z. Hou, Z.  
24 Cheng, C. Li, J. Lin, *Chem. Soc. Rev.* **2015**, *44*, 1416; c) H. Dong, S.-R. Du, X.-Y. Zheng, G.-M. Lyu,  
25 L.-D. Sun, L.-D. Li, P.-Z. Zhang, C. Zhang, C.-H. Yan, *Chem. Rev.* **2015**, *115*, 10725; d) L. Cheng, C.  
26 Wang, L. Feng, K. Yang, Z. Liu, *Chem. Rev.* **2014**, *114*, 10869; e) M. Sun, L. Xu, W. Ma, X. Wu, H.  
27 Kuang, L. Wang, C. Xu, *Adv. Mater.* **2016**, *28*, 898.
- 28 [21] L.-D. Sun, Y.-F. Wang, C.-H. Yan, *Acc. Chem. Res.* **2014**, *47*, 1001.



- 
- 1 [22] a) H. Yuan, S. Khatua, P. Zijlstra, M. Yorulmaz, M. Orrit, *Angew. Chem. Int. Ed.*, **2013**, *52*, 1217; b) S.  
2 Khatua, P. M. R. Paulo, H. Yuan, A. Gupta, P. Zijlstra, M. Orrit, *ACS Nano* **2014**, *8*, 4440; c) A.  
3 Kinkhabwala, Z. Yu, S. Fan, Y. Avlasevich, K. Mullen, W. E. Moerner, *Nat. Photon.* **2009**, *3*, 654; d) P.  
4 Holzmeister, E. Pibiri, J. J. Schmied, T. Sen, G. P. Acuna, P. Tinnefeld, *Nat. Commun.* **2014**, *5*, 5356; e)  
5 A. Bek, R. Jansen, M. Ringler, S. Mayilo, T. A. Klar, J. Feldmann, *Nano Lett.* **2008**, *8*, 485.
- 6 [23] Y. Wang, T. Yang, M. T. Tuominen, M. Achermann, *Phys. Rev. Lett.* **2009**, *102*, 163001.
- 7 [24] a) W. Park, D. Lu, S. Ahn, *Chem. Soc. Rev.* **2015**, *44*, 2940; b) M. Saboktakin, X. Ye, U. K. Chettiar, N.  
8 Engheta, C. B. Murray, C. R. Kagan, *ACS Nano* **2013**, *7*, 7186; c) W. Zhang, F. Ding, S. Y. Chou, *Adv.*  
9 *Opt. Mater.* **2012**, *24*, 236; d) M. Saboktakin, X. Ye, S. J. Oh, S.-H. Hong, A. T. Fafarman, U. K.  
10 Chettiar, N. Engheta, C. B. Murray, C. R. Kagan, *ACS Nano* **2012**, *6*, 8758; e) Q.-C. Sun, H. Mundoor, J.  
11 C. Ribot, V. Singh, I. I. Smalyukh, P. Nagpal, *Nano Lett.* **2014**, *14*, 101.
- 12 [25] A. Kuzyk, R. Schreiber, H. Zhang, A. O. Govorov, T. Liedl, N. Liu, *Nat. Mater.* **2014**, *13*, 862.
- 13 [26] a) W. Ma, H. Kuang, L. Wang, L. Xu, W.-S. Chang, H. Zhang, M. Sun, Y. Zhu, Y. Zhao, L. Liu, C. Xu,  
14 S. Link, N. A. Kotov, *Sci. Rep.* **2013**, *3*, 1934; b) C. Zhou, X. Duan, N. Liu, *Nat. Commun.* **2015**, *6*, 8102.
- 15 [27] L. Xu, H. Kuang, C. Xu, W. Ma, L. Wang, N. A. Kotov, *J. Am. Chem. Soc.* **2012**, *134*, 1699.
- 16 [28] R. Qiao, C. Liu, M. Liu, H. Hu, C. Liu, Y. Hou, K. Wu, Y. Lin, J. Liang, M. Gao, *ACS Nano* **2015**, *9*,  
17 2120.
- 18 [29] P. K. Jain, S. Eustis, M. A. El-Sayed, *J. Phys. Chem. B* **2006**, *110*, 18243.
- 19 [30] N. Berova, L. D. Bari, G. Pescitelli, *Chem. Soc. Rev.* **2007**, *36*, 914.
- 20 [31] I. Lieberman, G. Shemer, T. Fried, E. M. Kosower, G. Markovich, *Angew. Chem. Int. Ed.*, **2008**, *47*,  
21 4855.
- 22 [32] a) L. Xu, W. Yan, W. Ma, H. Kuang, X. Wu, L. Liu, Y. Zhao, L. Wang, C. Xu, *Adv. Mater.* **2015**, *27*,  
23 1706; b) K. L. Young, M. B. Ross, M. G. Blaber, M. Rycenga, M. R. Jones, C. Zhang, A. J. Senesi, B.  
24 Lee, G. C. Schatz, C. A. Mirkin, *Adv. Mater.* **2014**, *26*, 653.
- 25 [33] B. Yeom, H. Zhang, H. Zhang, J. I. Park, K. Kim, A. O. Govorov, N. A. Kotov, *Nano Lett.* **2013**, *13*,  
26 5277.

- 
- 1 [34] a) T. Gibaud, E. Barry, M. J. Zakhary, M. Henglin, A. Ward, Y. Yang, C. Berciu, R. Oldenbourg, M. F.  
2 Hagan, D. Nicastro, R. B. Meyer, Z. Dogic, *Nature* **2012**, *481*, 348; b) H. Miyake, H. Tsukube, *Chem.*  
3 *Soc. Rev.* **2012**, *41*, 6977; c) M. Caricato, A. K. Sharma, C. Coluccini, D. Pasini, *Nanoscale* **2014**, *6*,  
4 7165; d) M. Caricato, A. Delforge, D. Bonifazi, D. Dondi, A. Mazzanti, D. Pasini, *Org. Biomol. Chem.*  
5 **2015**, *13*, 3593.
- 6 [35] Y. Zhao, L. Xu, W. Ma, L. Wang, H. Kuang, C. Xu, N. A. Kotov, *Nano Lett.* **2014**, *14*, 3908.
- 7 [36] a) X. Wu, L. Xu, W. Ma, L. Liu, H. Kuang, W. Yan, L. Wang, C. Xu, *Adv. Funct. Mater.* **2015**, *25*, 850;  
8 b) F. Lu, Y. Tian, M. Liu, D. Su, H. Zhang, A. O. Govorov, O. Gang, *Nano Lett.* **2013**, *13*, 3145.
- 9 [37] A. Rose, T. B. Hoang, F. McGuire, J. J. Mock, C. Ciraci, D. R. Smith, M. H. Mikkelsen, *Nano Lett.* **2014**,  
10 *14*, 4797.
- 11 [38] Y. W. C. Cao, R. Jin, C. A. Mirkin, *Science* **2002**, *297*, 1536.
- 12

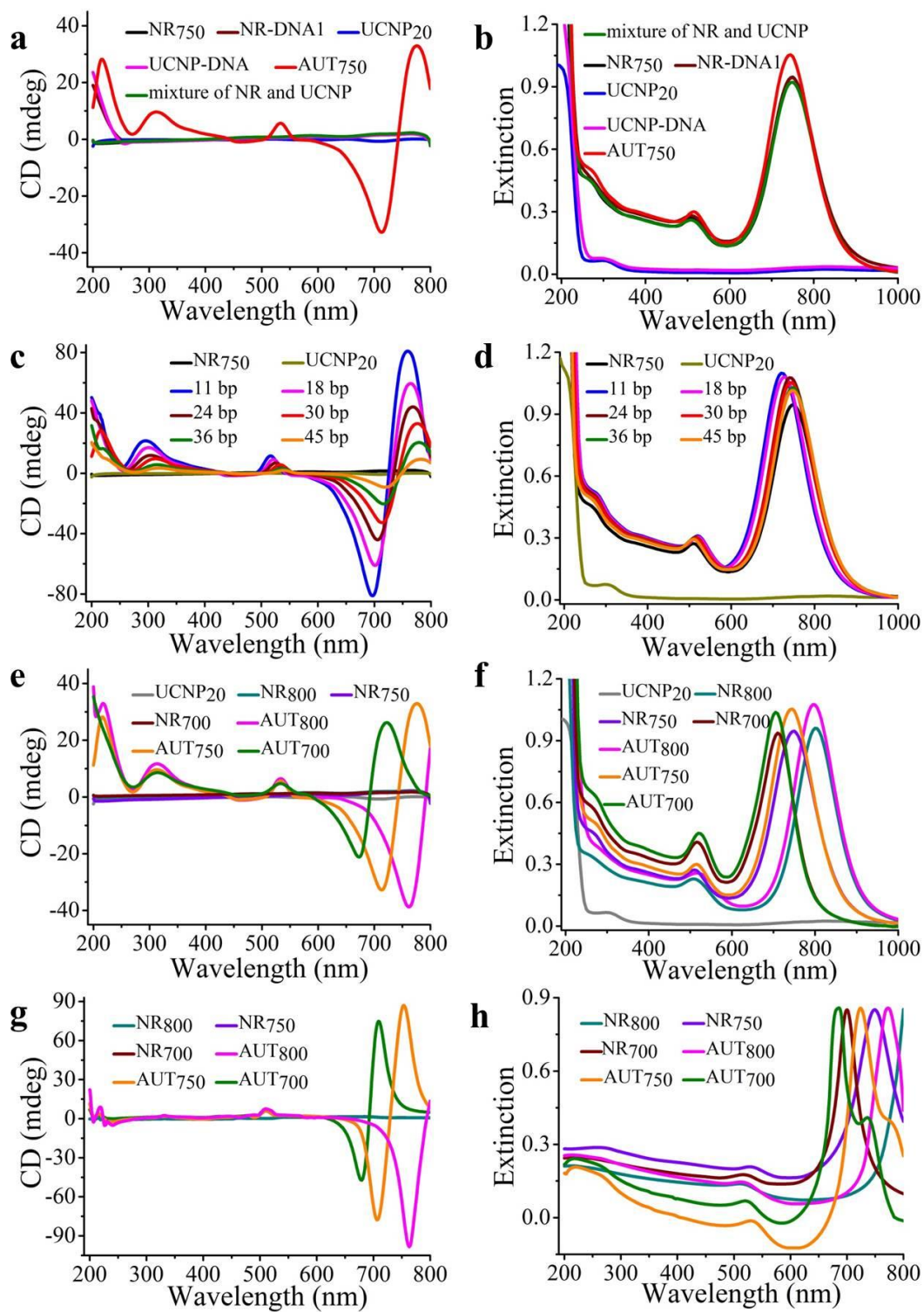


1

2 **Figure 1.** Structural Characteristics of NR-UCNP tetramer assemblies (AUT). (a) Schematic  
 3 illustration for the assembly. (b) The geometric structure illustrations of the tetramer. The illustrations  
 4 are the front view (upper left), top view (upper right), and the side view (lower left), respectively. (c-h)  
 5 Representative TEM images (c) and cryo-TEM tomographic reconstructions (d-h) of NR-UCNP  
 6 tetramer assembly. The different longitudinal absorption peak of NR was 750 nm (c,d), 700 nm (e); 800  
 7 nm (f); 850 nm (g); and 900 nm(h), respectively; scale bar in (c) and (d-h) is 100 and 20 nm,  
 8 respectively.

9

Author Manuscript



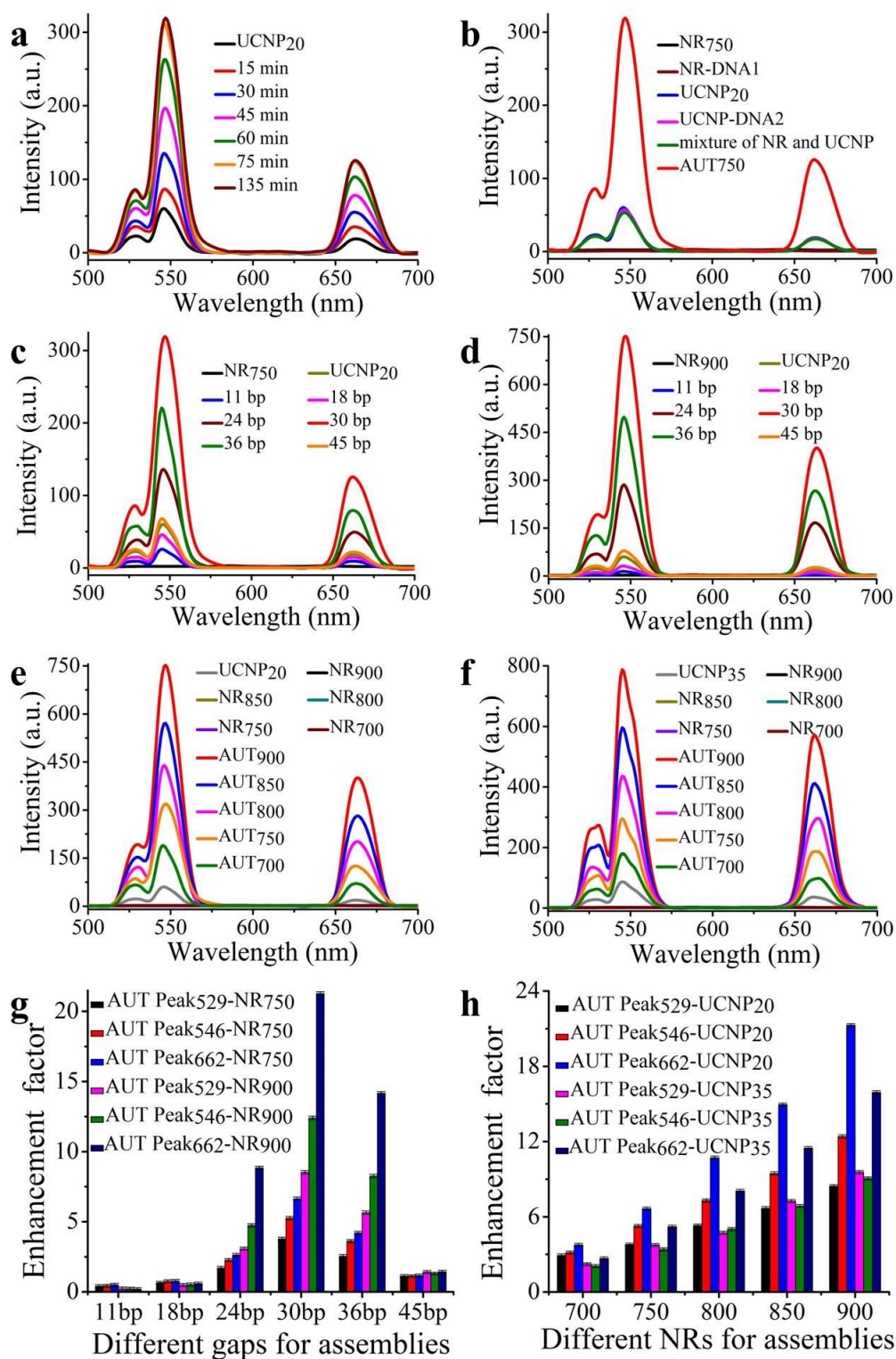
1

AI

This article is protected by copyright. All rights reserved.

---

1 **Figure 2.** Chiroptical activities of the NR-UCNP tetramer assemblies. (a-f) The experimental CD  
2 (a,c,e) and **UV-vis extinction (b,d,f) spectra** for NR-UCNP tetramer assembly with different lengths of  
3 DNA sequences (c,d), and different longitudinal absorption of NRs (e,f) and their controls (a,b). (g,h)  
4 The theoretical CD (g) and **UV-vis extinction (h) spectra** of NR-UCNP tetramer and their starting NRs.  
5 The NR-UCNP tetramers were assembled with 30 bp of DNA sequence, 20 nm of UCNP, and the  
6 longitudinal absorption peak of NR of 700 nm, 750 nm, and 800 nm.

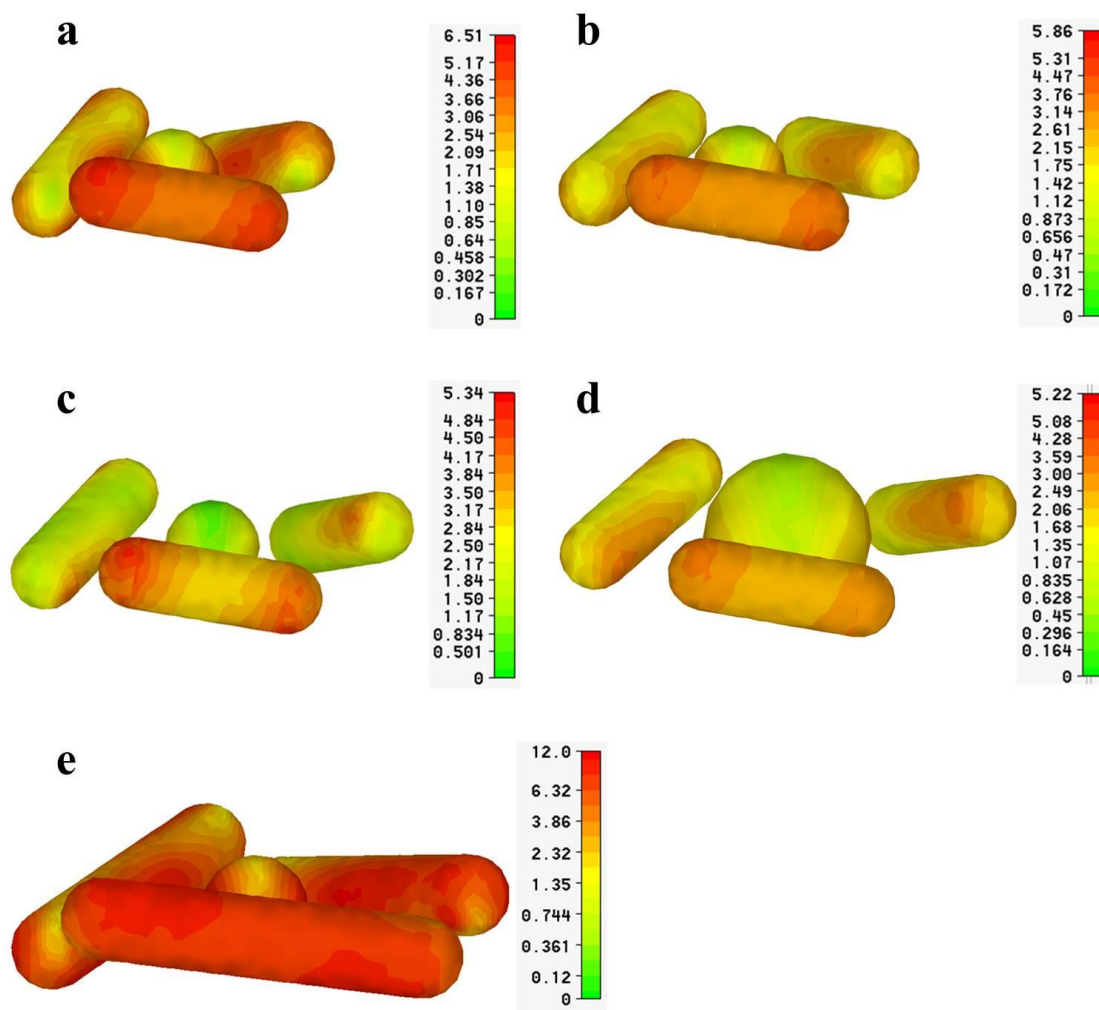


1

A

This article is protected by copyright. All rights reserved.

1 **Figure 3.** Up-conversion luminescence spectra of the NR-UCNP tetramer assembly and their controls  
 2 under excitation at 980 nm. (a) Dynamic luminescence spectra for the assembly. (b) Luminescent  
 3 spectra of NR-UCNP tetramers and their controls, (c,d) Assembly with different lengths of DNA  
 4 sequences with the longitudinal absorption peak of NRs of 750 nm (c), and 900 nm (d). (e,f) Assembly  
 5 with different NRs, where UCNPs were 20 nm (e), and 35 nm (f). (g,h) Luminescent enhancement  
 6 factor curves of c, d (g), and e, f (h).



7  
 8 **Figure 4.** The electromagnetic simulations of the NR-UCNP tetramer. (a,b,c) The electric field of  
 9 NR-UCNP tetramer assembled with 11 bp (a), 30 bp (b), and 45 bp (c) of DNA sequences, where the  
 10 longitudinal absorption peak of NR was 750 nm, and the size of UCNPs was 20 nm. (d) The electric field  
 11 of NR-UCNP tetramer assembled with 30 bp of DNA sequence, the longitudinal absorption peak of NR

---

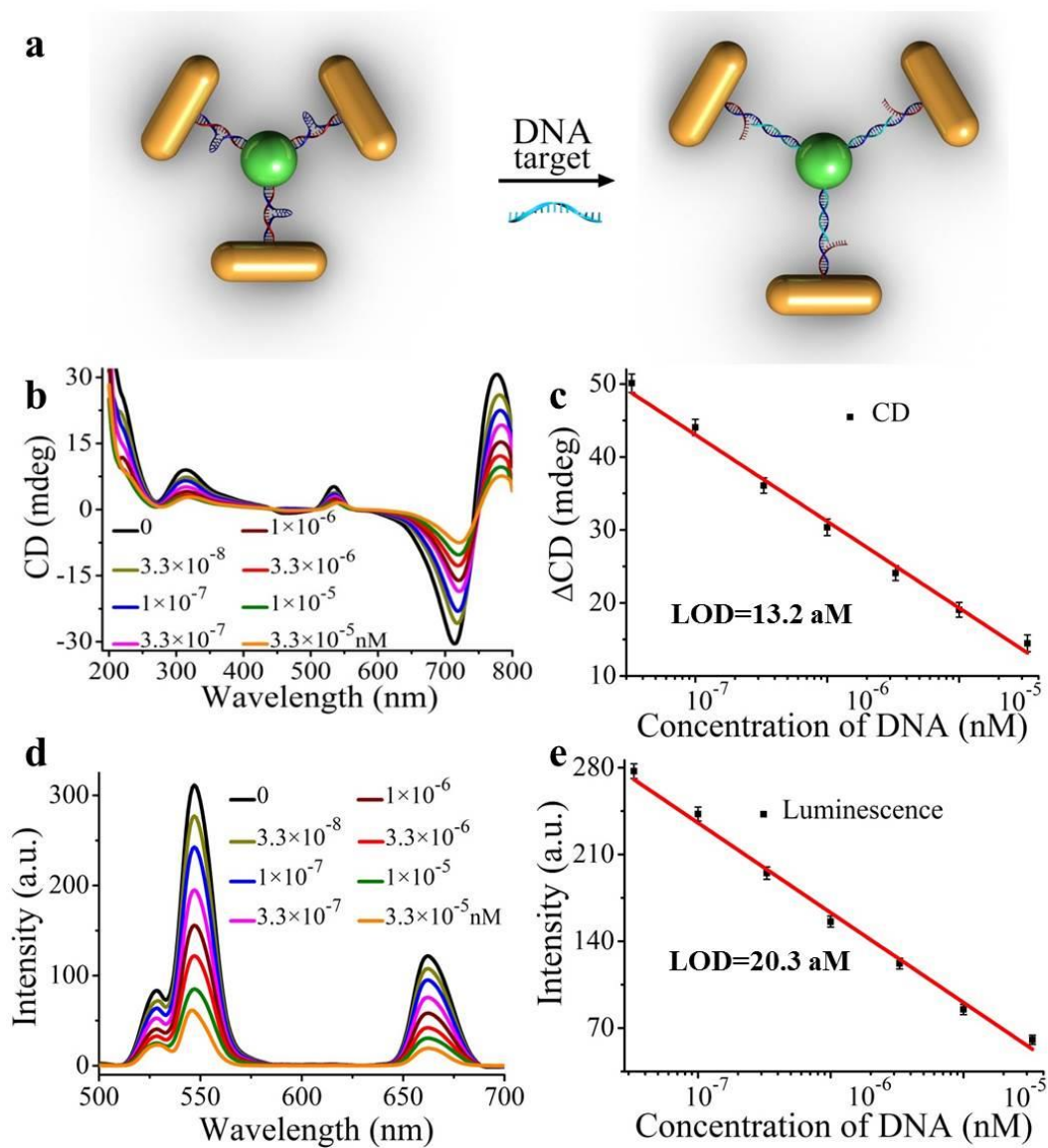
1 of 750 nm, and 35 nm of UCNP. (e) The electric field of NR-UCNP tetramer assembled with 30 bp of  
2 DNA sequence, the longitudinal absorption peak of NR of 900 nm, and 20 nm of UCNP.

3

Author Manuscript

This article is protected by copyright. All rights reserved.





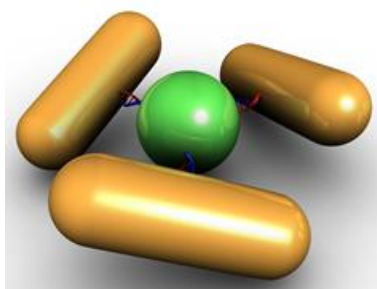
1

2 **Figure 5.** DNA detection by up-conversion luminescent and chiroplasmonic techniques with the  
 3 NR-UCNP tetramer assembly. (a) Schematic illustration for the DNA biosensing. (b,d) The CD (b) and  
 4 up-conversion luminescence (d) curves with increasing concentrations of DNA solution. (c,e) The CD  
 5 (c) and up-conversion luminescent (e) calibration curves for DNA detection. The longitudinal  
 6 absorption peak of NR using for assembly was 750 nm.

7

---

1 **TOC Figure:**



2

3 Propeller-like nanoscale assemblies with exceptionally intense chiroptical activity and strong  
4 luminescence, were prepared using gold nanorods and upconversion nanoparticles. The  
5 circular dichroism intensity of the tetramer reached 80.9 mdeg, with  $g$ -factor value of  $2.1 \times 10^{-2}$ .  
6 <sup>2</sup>. And the enhancement factor of upconversion luminescence was as high as 21.3 in aqueous  
7 phase. Attomolar bioanalysis of cancer biomarker with two model was also achieved, which  
8 holds promising potential for early disease diagnosis and environmental monitoring.

9

10 **Keywords:** nanoassemblies; upconversion nanoparticle; chiral; luminescence enhancement;  
11 aqueous phase; biosensing;

12

13

Author Manuscript