

RESEARCH ARTICLE

Metabolomics revealed the toxicity of cationic liposomes in HepG2 cells using UHPLC-Q-TOF/MS and multivariate data analysis

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Abstract

Cationic liposomes (CLs) are novel nonviral vectors widely used for delivering drugs or genes. However, applications of CLs are largely hampered by their cytotoxicity, partly because the potential mechanism underlying the cytotoxicity of CLs remains unclear. The aim of the present study was to explore the underlying mechanism of cytotoxicity induced by CLs on HepG2 cells. Differential metabolites were identified and quantified using ultra-liquid chromatography quadrupole time-of-flight mass spectrometry (UHPLC-Q-TOF/MS). The toxicity of CLs on HepG2 cells was evaluated by multivariate data analysis and statistics. Additionally, CCK-8 assay, heatmap, pathway and co-expression network were carried out to explore the relations between the metabolites and the pathways. The results showed a dose-dependent toxic effect of CLs on HepG2 cells, with an IC_{50} value of 119.9 $\mu\text{g}/\text{mL}$. Multivariate statistical analysis identified 42 potential metabolites between CLs exposure and control groups. Pathway analysis showed significant changes in pathways involving amino acid metabolism, energy metabolism, lipid metabolism and oxidative stress in the CLs exposure group vs the control group. Metabolites related to the above-mentioned pathways included phenylalanine, methionine, creatine, oxalacetic acid, glutathione, oxidized glutathione, choline phosphate and several unsaturated fatty acids, indicating that cells were disturbed in amino acid metabolism, energy and lipid supply when CLs exposure-induced injury occurred. It is concluded that CLs may induce cytotoxicity by enhancing reactive oxygen species *in vitro*, affect the normal process of energy metabolism, disturb several vital signaling pathways and finally induce cell death.

KEYWORDS

cationic liposome, cytotoxicity, metabolomics, UHPLC-Q-TOF/MS

1 | INTRODUCTION

Nonviral vectors have been widely used as carriers of drug delivery systems in the treatment of cancer (Andresen, Jensen, & Jorgensen, 2005). Among the nonviral vectors currently available, cationic liposomes (CLs) are considered most promising in that they transfer drugs through cellular membranes so that they can form electrostatic

complexes with plasmid DNA. Various cationic lipids have been synthesized and tested for use in nonviral gene delivery (Filion & Phillips, 1997). Although the transferring efficacy of CLs is substantially lower than that of viral vectors, CLs have some unique advantages including biodegradability, design flexibility and cell targeting capability (Cortesi et al., 2015; Gujrati et al., 2014). On the other hand, CLs also have some disadvantages, including high cytotoxicity and other adverse effects (Kagan, Bayir, & Shvedova, 2005), which greatly hamper wider applications of CLs. Meanwhile, the potential mechanism underlying cytotoxicity of CLs has not been well defined.

There are wide variations in toxicity of nongene-carrying CLs from different human cells. The capacity for nonspecific toxicity owing to

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Abbreviations: CL, cationic liposome; CoA, coenzyme A; DOTAP, 1,2-dioleoyl-3-trimethylammonium-propane; GSH, glutathione; NPs, nanoparticles; PCA, principal component analysis; PDI, polydispersity index; PLS-DA, partial least squares-discriminant analysis.

CLs themselves should be carefully considered when drug-containing liposomes are to be administered in humans (Romoren, Thu, Bols, & Evensen, 2004). On the one hand, the presence of high concentrations of negatively charged plasma proteins may affect the surface charge of CLs, or even aggregate CLs into micelles, causing significant cellular toxicity in the circulation process (Filion & Phillips, 1997). At the same time, some studies in the literature pointed out that the activity of cell membranes may disturb the integrity and the membrane function of cells or subcellular compartments, causing toxicity (Knudsen et al., 2015). Generally, CLs are mainly composed of natural phospholipids containing unsaturated fatty acids, which are easily oxidized, thus reducing the liquidity. Aggregation of CLs in cells prevents them from further going through biodegradation and excretion, resulting in unwanted cytotoxicity. Several different mechanisms, including oxidative stress, autophagy and certain physicochemical properties of nanoparticles (NPs), are reported to be associated with cellular toxicity (Wei et al., 2015; Xie & Wu, 2016). Many experiments have found that CLs also have toxicity in cells or animals, which hinders the clinical application of CLs (Filion & Phillips, 1997; Yang et al., 2016). Therefore, in-depth study of CLs is needed to explore the toxicity pathways and potential mechanisms underlying the vital role of CLs for the sake of widening the clinical application of CLs.

Metabolomics has emerged as a new discipline to study issues closely related to human health (Lindon et al., 2003). Meanwhile, research on the analysis of disease states as well as basic research to study biological processes or in predictive toxicology has gradually attracted wide attention. Metabolomics is concerned with the analysis of all small endogenous metabolites in order to investigate the relationship between metabolites and toxic stimulating factors (Robertson, 2005). It has become a forceful tool for prospecting new high-value compounds, providing insight into the potential toxicity mechanisms and might have remarkable usefulness for preventing cellular toxicity in the clinic (Tan et al., 2017). The liver is the main organ of metabolic clearance and has been recognized as the target site for drugs and NPs. Evidence has shown that NPs preferentially deposit in the liver under systemic exposure, resulting in prolonged retention within the organ (B. Wang, He, Zhang, Zhao, & Feng, 2013). Knowing that HepG2 is a hepatocyte carcinoma-derived cell line and thought to be a more realistic and relevant model of hepatocyte behavior, we chose HepG2 to carry out our investigation (Johnston et al., 2010). Additionally, an UHPLC-Q-TOF/MS-based metabolomics approach was carried out to analyze the potential mechanism underlying cellular toxicity of CLs exposure on HepG2 cells (G. Wang et al., 2016).

The aim of the present study was to evaluate the cellular toxicity of CLs on HepG2 cells, screen metabolites that undergo significant changes and discuss the potential toxicity mechanism of CLs. Based on the screened metabolites, we used the free online MetaboAnalyst software to analyze the obtained statistical data and further create a heatmap via multivariate statistical analysis, cluster grams and pathway analysis. The results demonstrate that metabolomics is a powerful method to gain insight into the toxicity mechanisms. To the best of our knowledge, the data shed light on the mechanisms underlying the cellular toxicity of CLs and might have important implications for developing safe gene delivery systems.

2 | MATERIALS AND METHODS

2.1 | Chemicals and equipment

Chemicals and equipment used in this study were Optima-grade methanol, formic acid, chloroform and acetonitrile of chromatographic grade (Merck); 1,2-dioleoyl-3-trimethylammonium-propane (DOTAP) (890890P), cholesterol and DSPE-PEG 2000 (Avanti Polar Lipids, AL, USA); 3-chloro-L-phenylalanine (Aladdin Reagent Co. Ltd); fetal bovine serum, penicillin, streptomycin phosphate-buffered saline and DMEM (Invitrogen, CA, USA); deionized water purified by Millipore purification system (Millipore, MA, USA); and Cell Counting Kit (CCK-8) (Dojindo Laboratories, Kumamoto, Japan). All of the other chemicals used in our research were of the highest analytic grade.

2.2 | Preparation and characterization of CLs

CLs were synthesized by using the classical film dispersion method as described previously (Kono, Iwasaki, Matsuoka, & Fujita, 2016; Yu et al., 2016). Briefly, DOTAP, cholesterol and DSPE-PEG 2000 at a molar ratio of 11:8:1 were dissolved in chloroform solution and mixed in a round-bottomed flask for about 1 h. Then the organic solvents were removed from the lipid solution with a rotary evaporator. Subsequently, the lipid film was hydrated with 10 mL phosphate-buffered saline to create the multilamellar liposomes. Then the nanoscale unilamellar vesicles were prepared by forcing out the multilamellar liposomes using a liposomes extruder LF-1 (Avestin, Inc, Canada) at 56°C. Extrusion was accomplished in a step-wise manner by performing membranes with progressively decreasing pore sizes (400, 200, 100 nm, Whatman Nuclepore membrane, Fisher Scientific, MA, USA), with 10 cycles per pore size. CLs were then stored at 4°C until use. The mean particle size, ζ -potential and polydispersity index (PDI) of the CLs were measured using a Malvern Zetasizer ZS90 (Malvern Instruments Ltd, UK).

Before measurement, newly prepared CLs were appropriately diluted 1:10 in H₂O and all of the operations were performed at room temperature after equilibration for 10 min. Data were expressed as an average of three measurements.

2.3 | Cell culture

The human hepatocellular carcinoma cell line HepG2 was maintained in DMEM supplemented with 10% (v/v) fetal bovine serum, penicillin (100 units/mL) and streptomycin (100 μ g/mL). Cells were routinely cultured under the standard conditions of 37°C, 5% CO₂ and 95% humidity (Thermo Scientific, Series II Water Jacket).

2.4 | CCK-8 assay

Cellular toxicity of CLs was assessed in HepG2 cells using the CCK-8 assay as described previously (Ko et al., 2015). During the experiment, HepG2 cells were seeded into a 96-well plate at a density of 1×10^4 cells per well and cultivated for 24 h to facilitate attachment to the surface, and then exposed to 100 μ L culture medium with a concentration range of CLs for another 24 h; untreated cells were used as control groups and each group was further divided into four parallel wells.

After treatment with CCK-8 solution for another 4 h at 37°C, the absorbance was measured at 450 nm using an absorbance microplate reader (Thermo Fisher Scientific Inc., Multiskan, MK3). Subsequently, cell viability was compared with that of the control using the following equation: $A_a/A_b \times 100$, where A_a and A_b are the absorbances of experimental samples and controls, respectively. Half-maximum inhibitory concentration (IC_{50}) values of CLs were calculated using Graphpad software.

2.5 | Sample collection and preparation

Cells were cultured overnight in a 100 mm dish at a density of 1×10^6 cells per dish, followed by incubation with a concentration of IC_{50} of CLs for 24 h. In each experiment, untreated cells were used as control and six biological replicates were performed. Next, cell lysates were obtained by adding 1 mL of cold methanol to each dish and mechanically harvested with a sterile plastic disposable cell scraper. Each dish was washed with an additional 500 μ L cold methanol that was collected into the respective Eppendorf® tube. Recovered cell lysate was vortexed for 1 min and further centrifuged at 13,000 g for 15 min at 4°C. The supernatant was collected and stored in a new 1.5 mL Eppendorf® tube at -80°C. The methanol solution was evaporated to dryness and the sample was re-suspended in 250 μ L pre-cooling reagent (methanol-chloroform, 9:1) containing 1 μ g/mL 3-chloro-L-phenylalanine as the internal standard. After being vortexed for 2 min, the sample was centrifuged at 13,000 g, 4°C for 15 min, and then added with 200 μ L supernatant in the sample vial for LC-MS analysis. A QC sample was accomplished by mixing aliquots from each supernatant (10 μ L per sample).

2.6 | Chromatography and mass spectrometry conditions

An Agilent Infinity 1290 series instrument (Agilent, CA, USA) equipped with an autosampler delivery system was used for UHPLC analysis. The chromatographic separation was performed on a HSS T₃ column (100 \times 2.1 mm, 2.5 μ m, Waters Corporation, USA). The mobile phase consisted of 0.05% formic acid (solvent A) and acetonitrile with 0.05% formic acid (solvent B). The column temperature was maintained at 25°C and the injection volume was 3 μ L. Post time was set at 5 min and the flow rate was 0.40 mL/min. The optimized elution condition was applied as follows: 0–2 min, 5% B; 2–17 min, 5–95% B; 17–19 min, 95% B.

The UHPLC system was connected to an Agilent 6538 accurate mass quadrupole time-of-flight mass spectrometer (Agilent Technologies, USA) with an electrospray ionization source operating in both positive and negative modes. The ESI source conditions were optimized as follows: capillary voltage, 4 kV in positive mode and 3.5 kV in negative mode; drying gas flow, 11 L/min; gas temperature, 350°C; nebulizer pressure, 45 psig; fragmentor voltage, 120 V; skimmer voltage, 60 V; collision energy, 10, 20, 30 and 40 eV. Accurate LC-MS mass spectra were recorded across an m/z range of 100–1200 Da. The MS/MS method was applied to further confirm the biomarkers.

2.7 | Data processing and statistical analysis

UHPLC-Q-TOF/MS spectra of all determined samples were recorded and converted into a common data file format (mzdata); subsequently, the collected data were extracted by molecular feature extraction with the retention time, m/z value and the corresponding intensities for all detected peaks in the dataset. The 80% rule was used to select the ions, and then normalization and autoscaling of the ion peaks were performed. To compare the differences of the two groups and find out the significant metabolites, Student's *t*-test of all the screened peaks was carried out using IBM SPSS Statistic Version 17.0 (IBM, New York, USA). The dataset containing all the peak area corresponding to the concentration of a certain metabolite was exported to the SIMCA-P software (version 11, Sweden) for principal component analysis (PCA) and partial least squares-discriminant analysis (PLS-DA) analysis. PLS-DA produced the key parameters: R^2X (the cumulative model variation in X), R^2Y (the cumulative model variation in Y) and Q^2 (the cumulative predicted variation). As these parameters approached 1.0, a model with predictive reliability appeared.

All data shown in the figures are expressed as the means \pm standard deviation (SD) and were assessed using the ANOVA or Student's *t*-test. Differences between the two groups were considered statistically significant using the least significant differences at the values of $p < 0.05$.

3 | RESULTS

3.1 | Characterization of CLs

CLs were characterized using a Zetasizer ZS90 to determine their diameter and size distribution. All samples were diluted 10-fold with ultrapure water. PDI is a measure of the uniformity of particle sizes present in the suspension, ranging from 0 (monodispersed) to 1 (very broad distribution); its value can reflect the homogenous size distribution of CLs. The physical parameters measured by the Malvern particle size analyzer are shown in Table 1. The size and ζ -potential distributions are shown in Figure S1 in the Supporting Information. Data are expressed as the means \pm standard deviation (SD). All measurements were carried out in triplicate at 25°C after 3 min equilibration. It was found that CLs were well dispersed and uniform in shape and size.

3.2 | Cellular toxicity of CLs

The potential effect of CLs on toxicity of HepG2 cells was investigated. The results demonstrated that the number of HepG2 cells decreased markedly in a concentration-dependent manner after exposure to CLs, with an IC_{50} value of 119.9 μ g/mL (Figure 1). In addition, no obvious cellular toxicity was observed when the dose of CLs was < 50 μ g/mL. However, when the dose of CLs was increased to the

TABLE 1 The physical characterizations of cationic liposomes (CLs)

Polymer	Size (nm)	PDI	ZP (mV)
CLs	152.2 \pm 4.04	0.117 \pm 0.013	8.38 \pm 4.96

PDI, Polydispersity index; ZP, ζ -potential; $n = 3$.

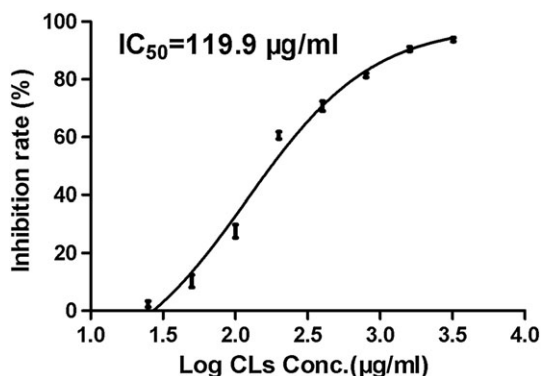


FIGURE 1 The inhibition rate of HepG2 cells treated with different concentrations of cationic liposomes (CLs)

highest level for 24 h, the inhibitory effect increased by ~96%, suggesting that CLs reduced the viability of HepG2 cells remarkably. To identify the altered level of metabolites more accurately, an IC_{50} value of 119.9 $\mu\text{g}/\text{mL}$ was chosen for subsequent experiments.

3.3 | Method validation

The UHPLC-Q-TOF/MS system was used to detect the cell samples. The representative total ion current chromatograms of both groups gained from positive and negative mode are shown in Figure S2.

A QC sample was applied to validate the system performance and obtain complementary product ion information. At the same time, five selected ions were calculated for the relative standard deviation (RSD) of the retention time and the peak area for validation. The RSD of retention time was <1% and the peak area was <5%, demonstrating that the method was precise, stable and reliable for subsequent analyses.

3.4 | Metabolomics analysis

To gain the whole metabolic information and investigate the alteration, the pattern recognition methods such as PCA and PLS-DA were performed on the resulting complex multivariate data set. Major metabolites were identified by high-resolution mass spectrometry and MS/MS experiments. The typical score plots between the control and CLs groups are shown in Figure 2. PCA was used to search the intrinsic variation in the dataset, and the results (Figure 2a and b) indicate that the clusters of the extracts from the CLs group were clearly separated from those of the relevant control, suggesting a significant alteration in the metabolic profile induced by CLs.

Knowing that PLS-DA is a supervised method mainly for discovering variation, it was used to identify metabolic differences between the two groups in this study. The pair-wise PLS-DA score plots in positive and negative ion modes are shown in Figure 2(c, d) and Figure S3, showing a significant difference between CLs and corresponding control groups. In the PLS-DA model, S-plots are usually used to identify the features contributing to group discrimination. Each spot represents an endogenous substance, and the spots located at the ends of the plot will contribute more to the differentiation between groups (Figure 2e and f).

3.5 | Identification of differential metabolites

UHPLC-MS and metabolomics data analyses were used to assess the difference in the metabolic profile within cells exposed to CLs vs the untreated control. The potential metabolites were picked out according to Variable Importance in the Project at the values of $VIP > 1.0$ and the S-plot. Some of these metabolites and the interpretation of significant ion peaks were preceded by searching metabolic databases including HMDB, KEGG and METLIN. In addition, these candidate metabolites from both groups were verified by comparison of the retention time and fragmentation pattern with authentic standards according to the MS/MS experiments. Subsequently, the literature and KEGG database (<http://www.genome.jp/kegg/>) were selected to interpret the pathways of the metabolites. Altogether 42 differential metabolites, including 23 in positive and 19 in negative modes, were identified, and the details are shown in Table 2.

3.6 | Function analysis and metabolic pathway

To visualize the level of metabolite change, a heatmap was built according to the peak area transformed data of metabolites. The heatmap based on the 42 metabolites showed significant differences between CLs and control groups (Figure S4), based on which a metabolic pathway analysis was further performed to evaluate their associated pathways. The result showed that the changed metabolites identified were primarily involved in amino acid metabolism, energy metabolism, lipids and oxidative stress. Figure 4 shows an overview of the pathway analysis, with the x-axis the pathway impact value calculated from pathway topological analysis and the y-axis corresponding to the $-\log(p)$ value obtained from pathway enrichment analysis, with the most impacted pathways colored. Consequently, the nine most significantly different pathways were considered closely related to the potential cellular toxicity induced by CLs. These pathways involved arachidonic acid metabolism, D-glutamine and D-glutamate metabolism, sphingolipids metabolism, pantothenate and coenzyme A (CoA) biosynthesis, glutathione (GSH) metabolism, TCA cycle, phenylalanine metabolism, glycerophospholipid metabolism and purine metabolism.

Based on the KEGG pathway database, we established the metabolic network (Figure 4), which clearly showed that CLs were related to the above-mentioned metabolites.

4 | DISCUSSION

To further explore cellular toxicity of CLs on HepG2 cells, we investigated the metabolic alterations in HepG2 cell extracts after 24 h continuous incubation with CLs by employing the UHPLC-Q-TOF/MS-based metabolomics approach. Subsequent analysis of the metabolite profiles of the samples from HepG2 cells provided a fingerprint of metabolic changes and highlighted the potential of metabolomics analysis in the evaluation of toxicity. About 42 metabolic biomarkers were highly likely to be associated with CLs exposure, because they showed better performance in terms of both specificity and sensitivity. These differential metabolites mostly included amino acids, lipids and fatty acids.

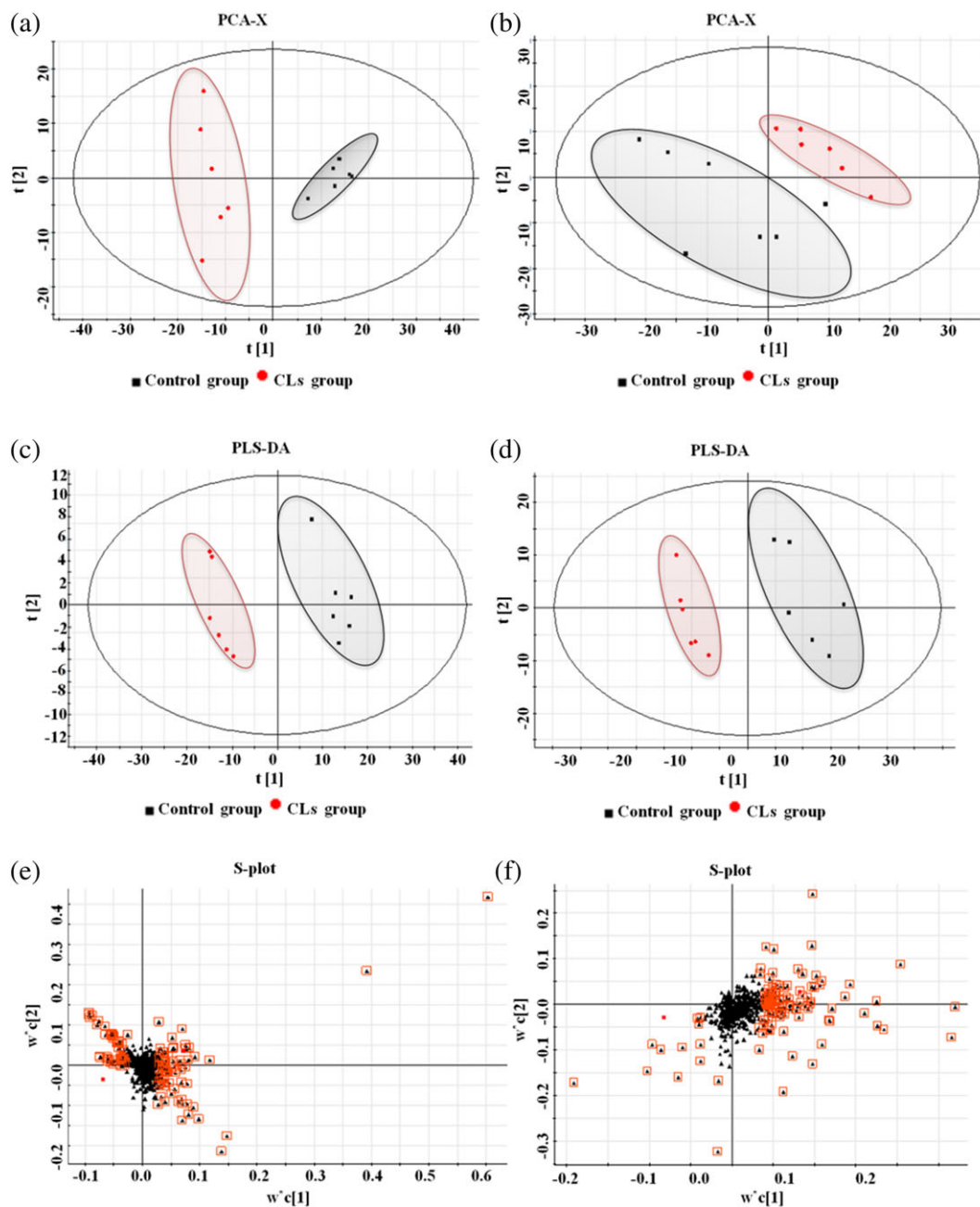


FIGURE 2 Principal component analysis (PCA) scores, partial least squares–discriminant analysis (PLS-DA) scores, S-plot between control and CLs groups in positive and negative mode. (a) PCA score in positive mode ($R^2X = 0.995$, $Q^2 = 0.793$); (b) PCA score in negative mode ($R^2X = 0.982$, $Q^2 = 0.765$); (c) PLS-DA score in positive mode ($R^2X = 0.855$, $R^2Y = 0.998$, $Q^2Y = 0.981$); (d) PLS-DA score in positive mode ($R^2X = 0.926$, $R^2Y = 0.999$, $Q^2Y = 0.964$); (e) S-plot in negative mode; (f) S-plot in positive mode

In our study, nine unique metabolic pathways were identified from CLs and control groups (Figure 3). The altered metabolites related to CLs exposure were mainly involved in amino acid metabolism, energy metabolism, oxidative stress metabolism and lipid metabolism (Figure 4).

4.1 | Amino acid metabolism

One recent study (Wei et al., 2015) reported that CLs could impair the activity of Na^+/K^+ -ATPase in cells, leading to cell anoxia and acute cell necrosis. Cell necrosis is known to affect endogenous metabolites.

Amino acids are the most fundamental substances in the organism and the most important substrate materials for synthetic proteins closely related to the life activities of cells.

Phenylalanine belonging to aromatic amino acid is a kind of essential amino acid and the precursor for tyrosine. On the one hand, phenylalanine is used to synthesize proteins in diverse cells; most phenylalanine *in vivo* is catalyzed by phenylalanine hydroxylase to generate tyrosine. In addition, tyrosine is further synthesized into essential hormones to participate in sugar and lipid metabolism. On the other hand, only a small proportion of phenylalanine is metabolized into phenylpyruvic acid (Du et al., 2016). Glutamic acids are substances

TABLE 2 Significant metabolomics profiles measured by UHPLC-Q-TOF/MS

No.	m/z	RT	VIP	Ion	Formula	Metabolites	Pathways	Fold change (CLs/C)	
1	175.1857	0.9805	1.791	[M - H] ⁻	C ₆ H ₁₃ N ₃ O ₃	Citrulline	Amino acids metabolism	1.373	↑##
2	307.323	0.9927	2.731	[M + H] ⁺	C ₁₀ H ₁₇ N ₃ O ₆ S	GSH	Glutathione metabolism	3.461	↑#
3	308.1819	0.9977	1.043	[M + H] ⁺	C ₉ H ₁₃ N ₂ O ₈ P	dUMP	Energy metabolism	0.514	↑#
4	612.631	1.0039	2.577	[M - H] ⁻	C ₂₀ H ₃₂ N ₆ O ₁₂ S ₂	Oxidized glutathione	Glutathione metabolism	3.155	↑#
5	135.1267	1.1637	1.167	[M + H] ⁺	C ₅ H ₅ N ₅	Adenine	Purine metabolism	0.534	↑##
6	131.1332	1.2292	5.225	[M + H] ⁺	C ₄ H ₉ N ₃ O ₂	Creatine	Energy metabolism	11.53	↑#
7	132.0716	1.2304	1.446	[M + H] ⁺	C ₄ H ₄ O ₅	Oxalacetic acid	Energy metabolism	0.913	↑#
8	164.158	1.7637	1.053	[M + H] ⁺	C ₉ H ₈ O ₃	Phenylpyruvic acid	Amino acids metabolism	0.446	↑##
9	181.1885	1.7645	2.248	[M + H] ⁺	C ₉ H ₁₁ NO ₃	L-tyrosine	Amino acids metabolism	2.095	↑#
10	352.4651	2.165	1.146	[M - H] ⁻	C ₂₀ H ₃₂ O ₅	Prostaglandin H2	Arachidonic acid metabolism	0.615	↑##
11	294.429	2.1922	1.049	[M - H] ⁻	C ₁₈ H ₃₀ O ₃	17-Hydroxylinolenic acid	Lipid metabolism	-0.519	↓##
12	166.1739	2.2008	1.803	[M + H] ⁺	C ₉ H ₁₀ O ₃	Phenyllactate	Others	1.297	↑##
13	463.2934	2.34073	2.341	[M - H] ⁻	C ₁₄ H ₁₈ N ₅ O ₁₁ P	Adenylsuccinic acid	Purine metabolism	2.481	↑#
14	439.609	2.8098	1.609	[M - H] ⁻	C ₂₃ H ₃₇ NO ₅ S	Leukotriene E4	Arachidonic acid metabolism	1.006	↑###
15	147.1293	2.8127	1.320	[M - H] ⁻	C ₅ H ₉ NO ₄	D-glutamic acid	Amino acids metabolism	0.706	↑###
16	89.0932	2.8134	1.038	[M - H] ⁻	C ₃ H ₇ NO ₂	Sarcosine	Others	-0.439	↓###
17	219.235	2.8583	2.496	[M + H] ⁺	C ₉ H ₁₇ NO ₅	Pantothenic acid	Others	2.298	↑##
18	355.4358	3.0647	3.065	[M - H] ⁻	C ₁₄ H ₂₃ N ₆ O ₃ S	S-Adenosylmethioninamine	Others	4.551	↑#
19	165.1891	3.652	4.972	[M + H] ⁺	C ₉ H ₁₁ NO ₂	L-phenylalanine	Amino acids metabolism	10.05	↑#
20	408.5714	4.0763	1.782	[M - H] ⁻	C ₂₄ H ₄₀ O ₅	Cholic acid	Others	1.252	↑###
21	567.3536	4.65051	4.651	[M - H] ⁻	C ₂₇ H ₅₄ NO ₉ P	Docosahexaenoic acid	Lipid metabolism	10.444	↑#
22	203.1158	5.329	1.785	[M - H] ⁻	C ₉ H ₁₇ NO ₄	Acetylcarnitine	Lipid metabolism	1.338	↑#
23	347.2212	5.4837	1.101	[M + H] ⁺	C ₁₀ H ₁₄ N ₅ O ₇ P	AMP	Energy metabolism	-0.606	↓#
24	458.3597	6.0405	1.104	[M + H] ⁺	C ₁₇ H ₂₃ N ₄ O ₉ P	FMNH2	Energy metabolism	-0.576	↓##
25	473.4393	6.0757	1.220	[M + H] ⁺	C ₂₀ H ₂₃ N ₇ O ₇	N5-formyl-THF	Others	-0.713	↓##
26	386.6535	8.9855	1.013	[M + H] ⁺	C ₂₇ H ₄₆ O	Cholesterol	Lipid metabolism	-0.306	↓###
27	299.2149	9.4478	1.322	[M + H] ⁺	C ₉ H ₁₈ NO ₈ P	D-4'-Phosphopantothenate	Others	0.578	↑###
28	496.3403	10.492	1.035	[M + H] ⁺	C ₂₄ H ₅₁ NO ₇ P	Palmitoylglycerophosphocholine	Lipid metabolism	-0.366	↓##
29	399.6077	10.582	1.988	[M + H] ⁺	C ₂₃ H ₄₅ NO ₄	L-Palmitoylcarnitine	Lipid metabolism	1.367	↑###
30	453.5503	10.582	2.403	[M - H] ⁻	C ₂₁ H ₄₄ NO ₇ P	LysoPE(16:0/0:0)	Lipid metabolism	2.651	↑#
31	454.6414	10.59	1.178	[M - H] ⁻	C ₂₉ H ₄₂ O ₄	Ubiquinone-4	Energy metabolism	0.635	↑#
32	481.7944	10.639	1.180	[M - H] ⁻	C ₃₀ H ₅₉ NO ₃	Ceramide (d18: 1/12:0)	Lipid metabolism	0.579	↑##
33	425.314	10.762	1.892	[M + H] ⁺	C ₁₂ H ₁₉ N ₄ O ₇ P ₂ S	Thiamin diphosphate	Energy metabolism	1.246	↑###
34	390.2345	11.257	1.247	[M - H] ⁻	C ₁₁ H ₁₉ O ₁₃ P	1-Phosphatidyl-D-myo-inositol	Lipid metabolism	0.606	↑###
35	523.6832	11.333	2.423	[M - H] ⁻	C ₂₆ H ₅₄ NO ₇ P	LysoPC(0:0/18:0)	Lipid metabolism	2.549	↑#
36	427.2011	11.408	1.052	[M + H] ⁺	C ₁₀ H ₁₅ N ₅ O ₁₀ P ₂	ADP	Energy metabolism	-0.351	↓###
37	665.441	11.804	1.835	[M + H] ⁺	C ₂₁ H ₂₉ N ₇ O ₁₄ P ₂	NADH	Energy metabolism	1.334	↑##
38	779.5218	12.123	1.218	[M + H] ⁺	C ₄₀ H ₇₇ NO ₁₁ S	3-O-Sulfogalactosylceramide	Lipid metabolism	0.501	↑###
39	735.9686	12.297	2.450	[M + H] ⁺	C ₃₈ H ₇₄ NO ₁₀ P	Ps (16:0/16:0)	Lipid metabolism	1.876	↑###
40	465.6273	12.301	1.875	[M - H] ⁻	C ₂₃ H ₅₀ N ₂ O ₅ P	LysoSM (d18:1)	Lipid metabolism	1.680	↑#
41	648.8913	12.549	1.282	[M + H] ⁺	C ₃₅ H ₆₉ O ₈ P	Pa (16:0/16:0)	Lipid metabolism	0.530	↑###
42	304.4669	14.845	2.091	[M - H] ⁻	C ₂₀ H ₃₂ O ₂	Arachidonic acid	Lipid metabolism	2.007	↑##

↓, Decreased; ↑, increased.

$p < 0.05$; ## $p < 0.01$; ### $p < 0.001$ (compared with control group).

CLs, Cationic liposomes group; C, control group.

Fold change was calculated as the logarithm of the average mass response (area) ratio between the two groups (fold change = \log_2 [CLs/C]).

required for the production of GSH, the most abundant natural antioxidant in the body. It was found in our study that the content of GSH was increased, probably owing to CLs-induced oxidative stress, which in turn increased glutamate and glycine to contribute to GSH

production. In addition, a panel of amino acids (phenylalanine, tyrosine, glutamic acid and citrulline) underwent significant changes in HepG2 cells compared with those in control group. These changes may be partly attributed to the disordered amino acid metabolism.

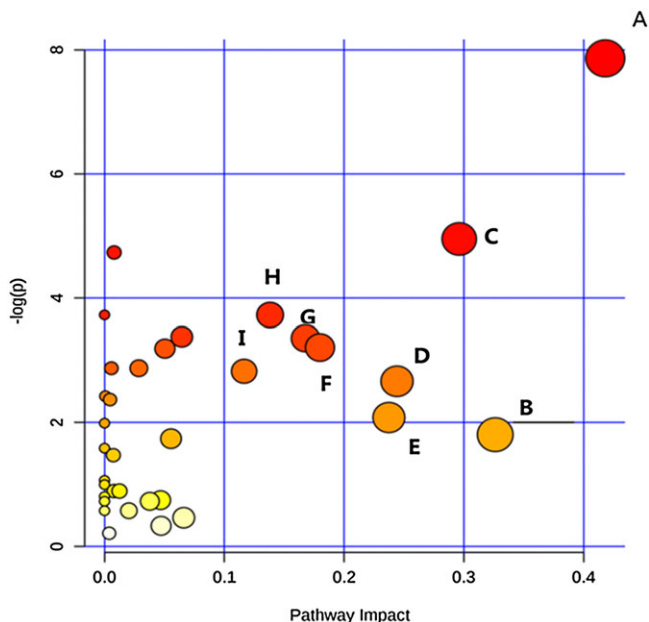


FIGURE 3 Impact on metabolic pathways suggested by the metabolites identified in HepG2 cells. (a) arachidonic acid metabolism; (b) D-glutamine and D-glutamate metabolism; (c) sphingolipid metabolism; (d) pantothenate and coa biosynthesis; (e) glutathione metabolism; (f) TCA cycle; (g) phenylalanine metabolism; (h) glycerophospholipid metabolism; (i) purine metabolism

4.2 | Energy metabolism

Energy metabolism is the key element in the regulation of life and death. Multiple metabolites such as creatine, oxalacetic acid, ATP and

NADH are known to play important roles in the cycle of energy metabolism.

Creatine is part of the cellular energy shuttle synthesized mainly in the liver by making use of arginine, glycine and methionine. It was found in our study that creatine was increased significantly in CLs group as compared with in the control group, indicating a perturbation of energy metabolism in HepG2 cells exposed to CLs. This finding, together with previous findings, may indirectly suggest that disturbed energy metabolism is involved in CLs exposure. Oxalacetic acid and thiamin diphosphate are important intermediates in TCA cycle. TCA cycle in association with oxidative phosphorylation is the major pathway in cellular energy production. On the other hand, phosphocholine is generated in a reaction catalyzed by choline kinase that converts ATP and choline into phosphocholine and ADP. Thus, change in phosphocholine implies that there is some disturbance in energy metabolism induced by CLs in HepG2 cells. At the same time, choline is a major constituent part of the cell membrane and plays an essential role in the integrity of cell membranes and lipid metabolism. The elevated concentrations of choline in the extracts denote the disruption of membrane fluidity caused by CLs. The production of energy partly depends on the oxygen-dependent electron transfer chain in cells; once the process is disturbed, the concentration of NADH, FADH₂ and ubiquinone-4 is altered in cells. In our study, AMP and ADP levels in CLs group were markedly decreased as compared with those in control group, suggesting that disturbed energy metabolism may be involved in CLs exposure. Pantothenic acid is the precursor of CoA, and also plays a significant role in lipid metabolism or as a prosthetic group in the TCA cycle (Li et al., 2010; Monteiro et al., 2015). The increased level of pantothenic acid

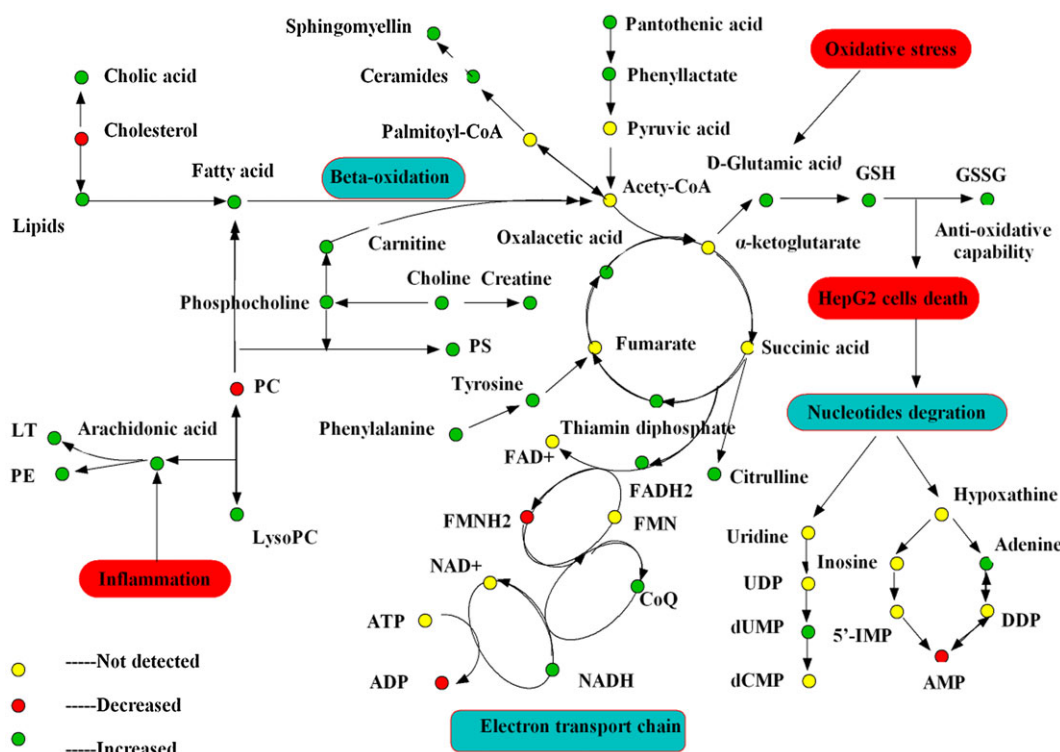


FIGURE 4 The network of metabolic pathways influenced by CLs in HepG2 cells. Identified differential metabolites in CLs group compared with the control group are labeled in red (down-regulated) green (up-regulated) and yellow (not detected), respectively

suggests that CLs exposure might induce the disturbance of coenzyme A and TCA cycle.

4.3 | Oxidative stress

GSH is the most abundant amino acid and plays an important role in anti-oxidation. It can protect some thiol-containing proteins or enzymes from being damaged by oxidants, especially peroxide (Hussain, Hess, Gearhart, Geiss, & Schlager, 2005). In the current investigation, we found significant alterations in the levels of GSH and oxidized glutathione in HepG2 cells, suggesting that CLs exposure could induce free radicals, which is consistent with the previous finding that liposome exposure could promote the production of reactive oxygen species (Gioria et al., 2016; Jia et al., 2009). Reactive oxygen species comprise a key signal to transmit environmental stimuli in cells in response to stress. The surface of the NPs could cause oxidative stress, thus increasing intracellular calcium and gene activation, or intracellular distribution of NPs may affect mitochondria, thereby generating oxidative stress (F. Wang et al., 2009).

Leukotriene (LT) and prostaglandin (PG) are minor metabolites of arachidonic acid under the action of inflammatory stimulation. They are also vital inflammatory mediators. It was found in our study that LTE4 and PGH2 levels were upregulated in CLs group, suggesting that CLs exposure may induce inflammation in HepG2 cells. Our experimental data demonstrate that oxidative stress caused by NPs could induce inflammation, resulting in DNA damage, which is consistent with the result of another study (Ma et al., 2011; Trinei et al., 2002). Adenine and dUMP are nucleotides, and substrates of DNA or RNA as well (Davies, Mendes, Smallbone, & Malys, 2012; Rai et al., 2009). In the CLs-exposed modes, we observed a significant increase in the content of the above nucleotides, indicating that CLs exposure triggered the inflammatory process and disturbed nucleotides.

4.4 | Lipid metabolism

Numerous studies have demonstrated that disturbance of lipid metabolism plays a pivotal role in the progression of toxicity of NPs exposure (Arisaka, Takano, Negishi, Arima, & Aramaki, 2011; Rajiv et al., 2016; Tian et al., 2016). An earlier transcriptomic study from our group found significant perturbations in mRNA associated with lipid metabolism in HepG2 cell extracts after CLs exposure. Both phosphatidylcholine (PC) and glycerophosphorylcholine (GPC) are derivatives of choline and the two main storage forms of choline in the cytoplasm. Choline-like metabolites are involved in several important pathways, playing signaling and structural roles in cell membranes and immune response. It was found in this study that the contents of these two metabolites were both significantly up-regulated in the CLs group, indicating that the increase in these metabolites may help remove damaged cells after CLs exposure. Three unsaturated fatty acids (arachidonic acid, 17-hydroxylinolenic acid and docosahexaenoic acid) were also increased markedly in the CLs group. Arachidonic acid, which is known to possess inflammatory response-mediating properties, is a polyunsaturated essential fatty acid and a vital component of phosphatides. However, arachidonic acid has cytotoxicity and may work as a physiological mediator of cell death. 17-Hydroxylinolenic acid and docosahexaenoic

acid are two important polyunsaturated fatty acids that play a vital role in many metabolic and cell signaling processes. We also observed a significant increase in some metabolites including ceramide and cholesterol in the CLs group vs the control group. Therefore, the significant perturbations in these lipid metabolites may be closely related to lipid metabolic dysfunction of CLs-induced cell death.

The process of fatty acid degradation, also known as beta-oxidation, converts an aliphatic compound into a set of activated acetyl units (acetyl CoA) that can be processed by a TCA cycle. The carnitine cycle is the first step for fatty acid oxidation, in which fatty acyl CoA enters the mitochondria as fatty acyl carnitines via carnitine transport. Numerous disorders involving energy production and excretion of unusual acylcarnitines have been described. In this study, acetylcarnitine and palmitoylcarnitine were decreased in the CLs group, indicating that the abnormality of fatty acid metabolism might be associated with the toxicity of CLs. Bile acids are physiological detergents that facilitate the excretion, absorption and transport of fats and sterols. The increased concentration of cholic acid in CLs group was observed, suggesting that CLs could disturb bile biosynthesis and cholesterol metabolism.

4.5 | Other metabolisms

Sarcosine, S-adenosyl-L-methionine (SAM) and N5-Formyltetrahydrofolate (N5-formyl-THF) are sources of methyl groups. The methylation process helps cells to respond to stress, detoxify xenobiotics and generate methylated compounds. The lack of methyl will affect the methylation of DNA and RNA, thus interfering with nucleic acid synthesis and affecting the normal metabolism of overall circulation. The change in methyl, as reflected in the concentrations of these metabolites with methyl sources such as choline, sarcosine, SAM and N5-Formyl-THF, suggests that methylation plays a vital role in detoxification after CLs exposure.

5 | CONCLUSIONS

Recent advances in metabolomics technologies have made it possible to analyze a wide range of biomolecules qualitatively and quantitatively, and determine their relative abundance in different biological conditions. In the present research, a UHPLC-Q-TOF/MS-based metabolomics method coupled with multivariate data analysis was employed to evaluate the toxicological effect of CLs *in vitro*. A total of 42 differentially expressed metabolites induced by CLs were screened between CLs and control groups. These metabolites belonged to the classifications amino acids, fatty acids and lipids and were mainly associated with the distribution of some metabolic pathways including amino acid metabolism, energy metabolism and the metabolism of phospholipid and fatty acid. Our results highlight that CLs may exert their toxicity via different pathways, although this needs to be verified by more studies.

Our findings demonstrate that the metabolomics approach can be employed as a tool to understand the toxicity mechanism of CLs. In addition, these results may serve as useful clues for further investigations needed to better understand the toxicity mechanisms of CLs.

Our ongoing study will construct a molecular network to further explore the mechanism underlying the cytotoxicity of CLs by transcriptomics and proteomics analysis.

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CONFLICT OF INTEREST

The authors declare that they have no conflict of interests.

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