

Examination of the role of methylenetetrahydrofolate reductase in incorporation of methyltetrahydrofolate into cellular metabolism

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ABSTRACT

Most mammalian cells receive exogenous folate from the bloodstream in the form of 5-methyltetrahydropteroylmonoglutamate ($\text{CH}_3\text{-H}_4\text{PteGlu}_1$). Because this folate derivative is a very poor substrate for foylpolylglutamate synthetase, the enzyme that adds glutamyl residues to intracellular folates, $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ must first be converted to tetrahydropteroylmonoglutamate ($\text{H}_4\text{PteGlu}_1$), 10-formyltetrahydropteroylmonoglutamate ($\text{CHO-H}_4\text{PteGlu}_1$), or dihydrofolate (H_2folate), which are excellent substrates for foylpolylglutamate synthetase. Polyglutamylation is required both for retention of intracellular folates and for efficacy of folates as substrates for most folate-dependent enzymes. Two enzymes are known that will react with $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ in vitro, methylenetetrahydrofolate reductase and methyltetrahydrofolate-homocysteine methyltransferase (cobalamin-dependent methionine synthase). These studies were performed to assess the possibility that methylenetetrahydrofolate reductase might catalyze the conversion of $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ to $\text{CH}_2\text{-H}_4\text{PteGlu}_1$. $\text{CH}_2\text{-H}_4\text{PteGlu}_1$ is readily converted to $\text{CHO-H}_4\text{PteGlu}_1$ by the action of methylenetetrahydrofolate dehydrogenase/methenyltetrahydrofolate cyclohydrolase, and these enzyme activities show very little preference for foylpolylglutamate substrates as compared with foylmonoglutamates. We conclude from in vitro studies of the enzyme that methylenetetrahydrofolate reductase cannot convert $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ to $\text{CH}_2\text{-H}_4\text{PteGlu}_1$ under physiological conditions and that uptake and retention of folate will be dependent on methionine synthase activity. — GREEN, J. M.; BALLOU, D. P.; MATTHEWS, R. G. Examination of the role of methylenetetrahydrofolate reductase in incorporation of methyltetrahydrofolate into cellular metabolism. *FASEB J.* 2: 42-47; 1988.

Key Words: methyltetrahydrofolate • foylpolylglutamate • methylenetetrahydrofolate reductase • folate uptake

INTRACELLULAR FOLATES ARE PRESENT predominantly as polyglutamate derivatives with a chain of glutamyl residues in γ linkage. The polyglutamate tail is required for intracellular retention of folates. It also constitutes an important recognition site for folate-dependent enzymes; many such enzymes exhibit much lower K_m values for folate polyglutamate substrates than for the corresponding monoglutamates. Most mammalian cells receive exogenous folate as the monoglutamate of 5-methyltetrahydrofolate ($\text{CH}_3\text{-H}_4\text{PteGlu}_1$),¹ the form of folate that is transported in the bloodstream. Tetrahydropteroylmonoglutamate ($\text{H}_4\text{PteGlu}_1$), 7,8-dihydropteroylmonoglutamate ($\text{H}_2\text{PteGlu}_1$), and 10-formyltetrahydropteroylmonoglutamate ($\text{CHO-H}_4\text{PteGlu}_1$) are all excellent substrates for foylpolylglutamate synthetase (EC 6.3.2.17) from pig liver (1). In contrast, $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ is a very poor substrate for foylpolylglutamate synthetase, the enzyme that adds glutamyl residues to intracellular folate derivatives (1). Incorporation of $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ into cellular metabolism therefore requires conversion of $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ into $\text{H}_4\text{PteGlu}_1$ or 5,10-methylenetetrahydropteroylmonoglutamate ($\text{CH}_2\text{-H}_4\text{PteGlu}_1$), with subsequent conversion of $\text{CH}_2\text{-H}_4\text{PteGlu}_1$ into $\text{H}_4\text{PteGlu}_1$, $\text{H}_2\text{PteGlu}_1$, or $\text{CHO-H}_4\text{PteGlu}_1$. The methylenetetrahydrofolate dehydrogenase/methenyltetrahydrofolate cyclohydrolase activities (EC 1.5.1.5/EC 3.5.4.9) of the trifunctional enzyme from pig liver, which catalyzes reactions 3A, 3B, and 3C of Fig. 1, show a very low discrimination between polyglutamate and monoglutamate substrates

¹Abbreviations: $\text{CH}_3\text{-H}_4\text{PteGlu}_n$, 5-methyltetrahydropteroylpolyglutamate with n glutamyl residues; $\text{CH}_2\text{-H}_4\text{PteGlu}_n$, 5,10-methylenetetrahydropteroylpolyglutamate with n glutamyl residues; $\text{H}_4\text{PteGlu}_n$, tetrahydropteroylpolyglutamate with n glutamyl residues; $\text{CHO-H}_4\text{PteGlu}_n$, 10-formyltetrahydropteroylpolyglutamate with n glutamyl residues; $\text{H}_2\text{PteGlu}_n$, 7,8-dihydropteroylpolyglutamate with n glutamyl residues; H_2folate , dihydrofolate; $\text{CH}_3\text{-H}_4\text{folate}$, 5-methyltetrahydrofolate; $\text{CH}_2\text{-H}_4\text{folate}$, 5,10-methylenetetrahydrofolate; AdoMet, S-adenosylmethionine; AdoHCy, S-adenosylhomocysteine.

inhibit the conversion of $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ to $\text{CH}_2\text{-H}_4\text{PteGlu}_1$, then a physiological role for methylenetetrahydrofolate reductase in catalyzing this conversion is unlikely. The experiments to be described provide evidence that physiological concentrations of AdoMet and/or NADPH severely inhibit reactions 2 and 3 *in vitro*.

RESULTS AND DISCUSSION

Effect of AdoMet on half-reactions necessary to support enzyme-mediated exchange of reducing equivalents between $\text{CH}_2\text{-H}_4\text{folate}$ and $\text{CH}_3\text{-H}_4\text{folate}$

The absorbance at 450 nm of the enzyme-bound FAD of methylenetetrahydrofolate reductase provides a convenient probe of the oxidation state of the enzyme. This permits us to measure the reductive and oxidative half-reactions by stopped-flow techniques. The reduction of methylenetetrahydrofolate reductase by $100\ \mu\text{M}$ $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ in the presence and absence of AdoMet is shown in Fig. 2. Both curves are monophasic; the first-order rate constant for reduction of the enzyme by saturating $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ is $6.1\ \text{s}^{-1}$. In the presence of added AdoMet this value is $0.002\ \text{s}^{-1}$. Thus the rate of reduction of the enzyme in the presence of AdoMet is 3000-fold slower than in its absence. Figure 3 shows the oxidation of reduced methylenetetrahydrofolate reductase by saturating $\text{CH}_2\text{-H}_4\text{PteGlu}_1$. In the absence of AdoMet, this reaction is clearly biphasic. The fast phase, which accounts for 67% of the change in ab-

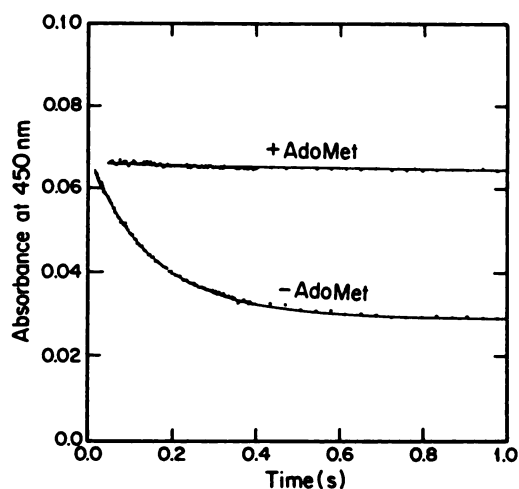


Figure 2. Stopped-flow measurement of the reduction of methylenetetrahydrofolate reductase by $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ in the presence and absence of $110\ \mu\text{M}$ AdoMet. Methylenetetrahydrofolate reductase was purified from pig liver by a method previously described (7). AdoMet, (6*R*)- H_4folate , and (6*S*)- $\text{CH}_3\text{-H}_4\text{folate}$ were prepared and/or purified as described before (8, 9). Buffered enzyme, $6\ \mu\text{M}$ in FAD, was mixed with an equal volume of buffer containing $100\ \mu\text{M}$ (6*S*)- $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ added as (6-*R,S*)- $\text{CH}_3\text{-H}_4\text{PteGlu}_1$. [All experiments were conducted in 50 mM potassium phosphate buffer, pH 7.2, 0.3 mM EDTA, and 10% glycerol (wt/vol).] Absorbance was monitored at 450 nm in the 2-cm pathlength of a stopped-flow apparatus, which was maintained under anaerobic conditions at 25°C. Anaerobiosis was achieved and maintained as described before (see ref 5).

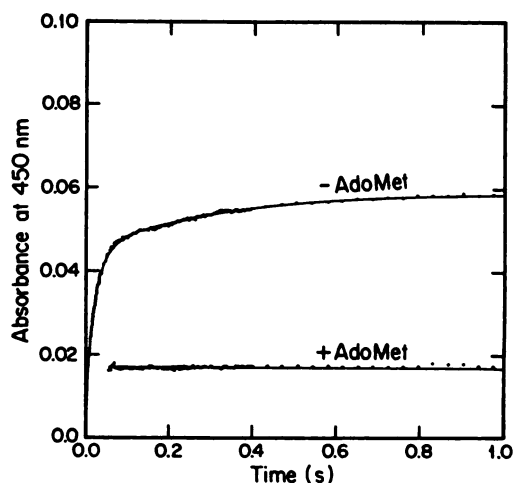


Figure 3. Stopped-flow measurement of the oxidation of reduced methylenetetrahydrofolate reductase by $\text{CH}_2\text{-H}_4\text{PteGlu}_1$ in the presence or absence of $110\ \mu\text{M}$ AdoMet. Buffered enzyme, $6\ \mu\text{M}$ in FAD, was anaerobically titrated with a stoichiometric amount of NADPH to form reduced enzyme. Reduced enzyme was then mixed with an equal volume of buffer containing $100\ \mu\text{M}$ (6*R*)- $\text{CH}_2\text{-H}_4\text{PteGlu}_1$, added as (6-*R,S*)- $\text{CH}_2\text{-H}_4\text{PteGlu}_1$. [All experiments were conducted in 50 mM potassium phosphate buffer, pH 7.2, 0.3 mM EDTA, and 10% glycerol (wt/vol).] $\text{CH}_2\text{-H}_4\text{PteGlu}_1$ was prepared by preincubating 13 mM formaldehyde and $200\ \mu\text{M}$ (6*R,S*)- H_4folate together in buffer containing 26 mM 2-mercaptoethanol for 10 min before the experiment.

sorbance, is characterized by a first-order rate constant of $47\ \text{s}^{-1}$, and the value calculated for the slow phase is $3\ \text{s}^{-1}$. In the presence of AdoMet, the oxidation of reduced enzyme by $100\ \mu\text{M}$ $\text{CH}_2\text{-H}_4\text{PteGlu}_1$ is apparently monophasic for 600 s; the rate constant for this half-reaction is $0.0037\ \text{s}^{-1}$.

We conclude that AdoMet is a potent inhibitor of the transfer of reducing equivalents between the enzyme-bound flavin of methylenetetrahydrofolate reductase and folate substrates.

Effects of AdoMet and NADPH on $\text{CH}_3\text{-H}_4\text{folate}$ /menadione oxidoreductase activity

Although menadione is not a physiological electron acceptor we have assumed that inhibition studies of the $\text{CH}_3\text{-H}_4\text{folate}$ /menadione oxidoreductase activity could provide insight into the possibility that methylenetetrahydrofolate reductase might catalyze a similar reduction of an unknown high-potential electron acceptor in the cell, with concomitant oxidation of $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ to $\text{CH}_2\text{-H}_4\text{PteGlu}_1$. For these studies we used two methods of measuring $\text{CH}_3\text{-H}_4\text{folate}$ /menadione oxidoreductase activity: spectrophotometric measurement and radiochemical assay. The spectrophotometric method offers the advantages of a continuous assay; however, it is useful only over a limited range of $\text{CH}_3\text{-H}_4\text{folate}$ concentrations. Concentrations of $\text{CH}_3\text{-H}_4\text{folate}$ below $5\ \mu\text{M}$ produce nonlinear initial rates owing to substrate depletion during the assay, whereas concentrations above $70\ \mu\text{M}$ possess too much background absorbance for spectrophotometric measurement. The spectrophotometric assay was used to measure the K_m for $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ ($33\ \mu\text{M}$, data not

shown). The radiochemical assay was used to measure the K_m for $\text{CH}_3\text{-H}_4\text{PteGlu}_6$ ($1.9 \mu\text{M}$, data not shown) and to determine the relative values for V_{\max} with $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ (set at 1.0) and $\text{CH}_3\text{-H}_4\text{PteGlu}_6$ (measured as 0.52). For reaction 2 to be of physiological significance, $\text{CH}_3\text{-H}_4\text{PteGlu}_1$, which is present at very low concentrations (1), would be required to compete successfully with $\text{CH}_3\text{-H}_4\text{folate}$ polyglutamates, which are present at much higher concentrations, for oxidation by methylenetetrahydrofolate reductase. The magnitude of the V_{\max}/K_m ratio for a substrate is usually proportional to the effectiveness of that substrate in competition with other substrates. Our studies show that V_{\max}/K_m for $\text{CH}_3\text{-H}_4\text{PteGlu}_6$ is 11.5-fold higher than V_{\max}/K_m for $\text{CH}_3\text{-H}_4\text{PteGlu}_1$. These results suggest that intracellular $\text{CH}_3\text{-H}_4\text{folate}$ polyglutamates may significantly inhibit the oxidation of $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ by methylenetetrahydrofolate reductase.

The radiochemical assay was also used to measure the effect of NADPH on $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ /menadione oxidoreductase activity in the presence or absence of AdoMet. Either NADPH ($100 \mu\text{M}$) or AdoMet ($50 \mu\text{M}$) inhibits the reactions; when they are included together in the preincubation, inhibition is virtually complete (data not shown). Thus, the presence of NADPH does not protect the $\text{CH}_3\text{-H}_4\text{folate}$ /menadione oxidoreductase activity of the enzyme from inhibition by AdoMet.

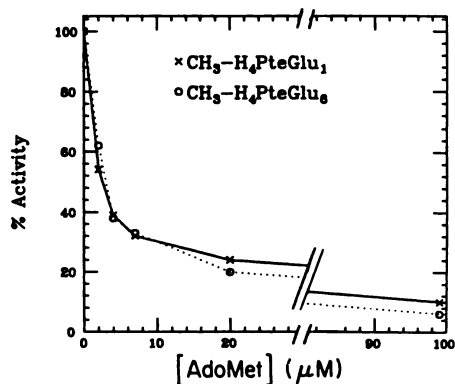


Figure 4. Comparison of the effect of AdoMet on the $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ /menadione and $\text{CH}_3\text{-H}_4\text{PteGlu}_6$ /menadione oxidoreductase activities of methylenetetrahydrofolate reductase. Cuvets (1 ml) containing buffer, $\text{CH}_3\text{-H}_4\text{folate}$, and menadione were placed in the sample and reference beams of a Cary 118 recording spectrophotometer; the back cuvet contained no enzyme, and the reaction in the front cuvet was initiated with substrate. Assay mixtures containing buffer, methylenetetrahydrofolate reductase, and varying amounts of AdoMet were preincubated for 10 min at 25°C . [All experiments were conducted in 50 mM potassium phosphate buffer, pH 7.2, 0.3 mM EDTA, and 10% glycerol (wt/vol).] ($6S$)- $\text{CH}_3\text{-H}_4\text{folate}$ ($25 \mu\text{M}$) and menadione ($121 \mu\text{M}$) were quickly added to initiate the reaction, and the increase in absorbance at 312 nm was monitored; the change in absorbance at 312 nm represents the combined effects of $\text{CH}_3\text{-H}_4\text{folate}$ and menadione being converted to $\text{CH}_2\text{-H}_4\text{folate}$ and menadiol ($\Delta\epsilon = 8300 \text{ M}^{-1} \text{ cm}^{-1}$). Blank rates caused by the reaction of residual 2-mercaptoethanol with menadione were subtracted; 2-mercaptoethanol was present to stabilize the $\text{CH}_3\text{-H}_4\text{folate}$ stocks, and the concentration in the final reaction mixture was less than 1 mM. Assays were performed in duplicate or triplicate and the averaged data are shown. Activities are expressed as the percentage of the activity observed in the absence of AdoMet.

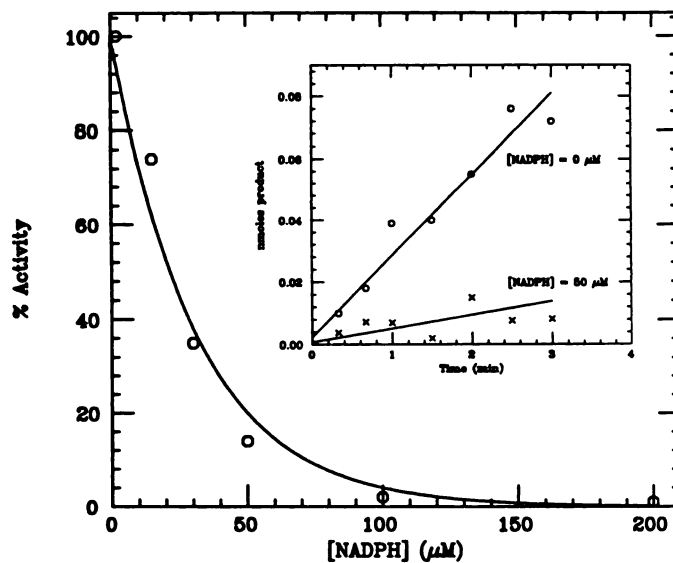


Figure 5. Effect of NADPH on the exchange of reducing equivalents between $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ and $\text{CH}_2\text{-H}_4\text{PteGlu}_6$. Assay mixtures, 0.4 ml, contained buffer, 20 mM 2-mercaptoethanol, 3 mM formaldehyde, $50 \mu\text{M}$ tritiated ($6S$)- $\text{CH}_3\text{-H}_4\text{PteGlu}_1$, and $25 \mu\text{M}$ $\text{H}_4\text{PteGlu}_1$. [All experiments were conducted in 50 mM potassium phosphate buffer, pH 7.2, 0.3 mM EDTA, and 10% glycerol (wt/vol).] The [^3H]- $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ had a specific radioactivity of 68,000 dpm/nmol. Assays were initiated by the addition of enzyme, and were heat-quenched at 98°C for 5 min after addition of 1.25 ml of dimedone in 1 M sodium acetate, pH 4.5. Samples were cooled on ice and the dimedone-formaldehyde complex was extracted with toluene and counted in organic counting scintillant (Amersham, Arlington Heights, IL). Blanks containing no enzyme were subtracted from the other values for each set. The primary data of disintegrations per minute ^3H in product vs. time for each concentration of NADPH were fitted by linear regression. Slopes of these lines were used to calculate the percent activity relative to that observed with no NADPH added; the percent activity remaining is plotted vs. the concentration of NADPH present in the assay.

Figure 4 shows the effect of AdoMet on $\text{CH}_3\text{-H}_4\text{folate}$ /menadione oxidoreductase activity. AdoMet inhibits $\text{CH}_3\text{-H}_4\text{folate}$ /menadione oxidoreductase activity with either monoglutamyl or hexaglutamyl substrates. The effect of AdoMet on $\text{CH}_3\text{-H}_4\text{folate}$ /menadione oxidoreductase activity is similar to its effect on NADPH/menadione oxidoreductase activity (10). Half-maximal inhibition of NADPH/menadione oxidoreductase activity measured under the same conditions was seen at 2–3 μM , whereas it occurs at 3 μM with the $\text{CH}_3\text{-H}_4\text{folate}$ /menadione oxidoreductase reaction. The degree of inhibition seen at 100 μM AdoMet is, however, less than that seen in the stopped-flow study shown in Fig. 2, and may indicate that the presence of $\text{CH}_3\text{-H}_4\text{folate}$ and/or menadione slightly protects the enzyme against inhibition by AdoMet.

Measurement of the exchange of reducing equivalents between the folate substrates of methylenetetrahydrofolate reductase in the presence of NADPH and/or AdoMet

The effect of NADPH on the exchange of reducing equivalents between $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ and $\text{CH}_2\text{-H}_4\text{PteGlu}_6$ was studied, and the data are shown in Fig. 5.

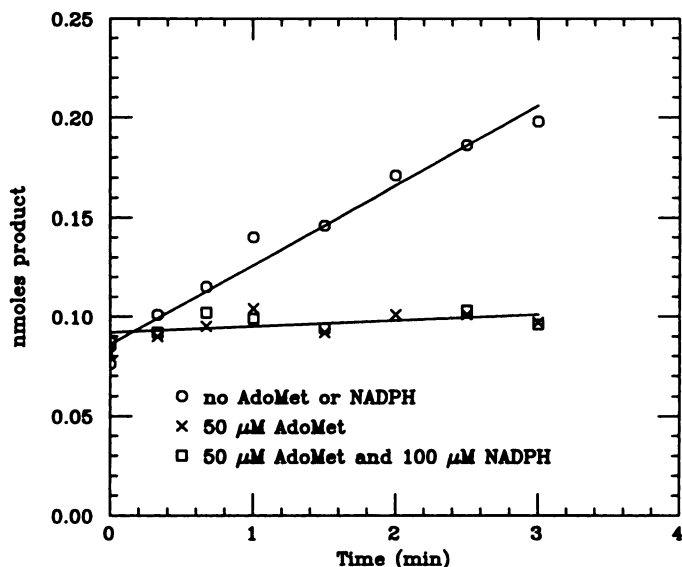


Figure 6. Effect of AdoMet and NADPH on the exchange of reducing equivalents between $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ and $\text{CH}_2\text{-H}_4\text{PteGlu}_6$. Assay mixtures, 0.4 ml, contained buffer, 50 μM tritiated (6*R*)- $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ (68,000 dpm/nmol), and 25 μM $\text{CH}_2\text{-H}_4\text{PteGlu}_6$, and AdoMet and NADPH were added as indicated. [All experiments were conducted in 50 mM potassium phosphate buffer, pH 7.2, 0.3 mM EDTA, and 10% glycerol (wt/vol).] Assays were initiated with the addition of enzyme, and were heat-quenched and performed as described before. Enzyme was preincubated for 10 min at 25°C in buffer containing the same amount of AdoMet and/or NADPH present in that assay. In this experiment the blanks were not subtracted. Percent activity (relative to the assay containing no added NADPH and AdoMet) was calculated from slopes: the reaction with no additions represents 100% activity; the reaction with 50 μM AdoMet exhibited $12 \pm 5\%$ activity; the reaction with 50 μM AdoMet and 100 μM NADPH exhibited $8 \pm 5\%$ activity.

The rate of exchange between monoglutamate and hexaglutamate substrates is one-fourth as slow as that between the two monoglutamate substrates (data not shown). The curves of percentage of activity vs. NADPH concentration for the $\text{CH}_3\text{-H}_4\text{PteGlu}_1/\text{CH}_2\text{-H}_4\text{PteGlu}_1$ and $\text{CH}_3\text{-H}_4\text{PteGlu}_1/\text{CH}_2\text{-H}_4\text{PteGlu}_6$ exchange reactions are nearly identical. This implies that NADPH competes with $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ for reduction of the enzyme and that the nature of the second substrate does not alter the apparent inhibition of exchange by NADPH.

Although AdoMet and NADPH both individually inhibit exchange, previous studies have shown that in combination these may act as antagonists. Specifically, high concentrations of NADPH can decrease the inhibitory effect of AdoMet on the NADPH/menadione oxidoreductase activity of methylenetetrahydrofolate reductase (10). In Fig. 6 we show that a combination of NADPH (100 μM) and AdoMet (50 μM) leads to the same inhibition of exchange as does AdoMet (50 μM) alone.

These studies establish that physiological concentrations of NADPH [200 μM (11)] and/or AdoMet [70 μM (12)] profoundly inhibit the $\text{CH}_3\text{-H}_4\text{folate/menadione}$ oxidoreductase activity (reaction 2) and the exchange of reducing equivalents between $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ and $\text{CH}_2\text{-H}_4\text{PteGlu}_6$ (reaction 3) catalyzed by methylenetetrahydrofolate reductase. The turnover number for NADPH/

$\text{CH}_2\text{-H}_4\text{folate}$ oxidoreductase activity is 5900 min^{-1} , whereas the turnover number for $\text{CH}_3\text{-H}_4\text{PteGlu}_1/\text{menadione}$ oxidoreductase activity is 300 min^{-1} , and that for the $\text{CH}_3\text{-H}_4\text{PteGlu}_1/\text{CH}_2\text{-H}_4\text{PteGlu}_6$ exchange reaction, reaction 3, is 72 min^{-1} ; thus, the uninhibited reactions that could lead to oxidation of $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ are significantly slower than the NADPH/ $\text{CH}_2\text{-H}_4\text{folate}$ oxidoreductase reaction. Intracellular NADPH and AdoMet should further reduce the velocity of these reactions in vivo, as they do in vitro.

We conclude that methylenetetrahydrofolate reductase will not provide a viable mechanism that can help the cell convert $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ into a better substrate for folylpolyglutamate synthetase. Despite its rather high K_m for $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ (6 μM) and its strong preference for $\text{CH}_3\text{-H}_4\text{PteGlu}_6$ over $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ (6), methionine synthase must be responsible for the introduction of $\text{CH}_3\text{-H}_4\text{PteGlu}_1$ into cellular metabolism. Because rapidly growing cells will require more exogenous folate than more slowly growing cells, this means that methionine synthase may be an attractive target for chemotherapeutic intervention. [FJ]

This work was supported in part by U.S. Public Health Service grant GM 30885 from the National Institute of General Medical Sciences.

REFERENCES

1. CICHOWICZ, D. J.; SHANE, B. Mammalian folyl- γ -glutamate synthetase. 2. Substrate specificity and kinetic properties. *Biochemistry* 26: 513-521; 1987.
2. ROSS, J.; GREEN, J.; BAUGH, C. M.; MACKENZIE, R. E.; MATTHEWS, R. G. Studies on the polyglutamate specificity of methylenetetrahydrofolate dehydrogenase from pig liver. *Biochemistry* 23: 1796-1801; 1984.
3. KUTZBACH, C.; STOKSTAD, E. L. R. Mammalian methylenetetrahydrofolate reductase: partial purification, properties, and inhibition by S-adenosylmethionine. *Biochim. Biophys. Acta* 250: 459-477; 1971.
4. VANONI, M. A.; MATTHEWS, R. G. Kinetic isotope effects on the oxidation of reduced nicotinamide adenine dinucleotide phosphate by the flavoprotein methylenetetrahydrofolate reductase. *Biochemistry* 23: 5272-5279; 1984.
5. VANONI, M. A.; BALLOU, D. P.; MATTHEWS, R. G. Methylenetetrahydrofolate reductase: steady state and rapid reaction studies on the NADPH-menadione oxidoreductase activities of the enzyme. *J. Biol. Chem.* 258: 11510-11514; 1983.
6. MATTHEWS, R. G.; GHOSE, C.; GREEN, J. M.; MATTHEWS, K. D.; DUNLAP, R. B. Folyl-polyglutamates as substrates and inhibitors of folate-dependent enzymes. *Adv. Enzyme Regul.* 26: 157-171; 1987.
7. MATTHEWS, R. G. Methylenetetrahydrofolate reductase from pig liver. *Methods Enzymol.* 122: 372-381; 1986.

8. SUMNER, J.; JENCKS, D. A.; KHANI, S.; MATTHEWS, R. G. Photoaffinity labeling of methylenetetrahydrofolate reductase with 8-azido-S-adenosylmethionine. *J. Biol. Chem.* 261: 7697-7700; 1986.
9. MATTHEWS, R. G.; ROSS, J.; BAUGH, C.; COOK, J.; DAVIS, L. Interactions of pig liver serine hydroxymethyltransferase with methyltetrahydropteroylpolyglutamate inhibitors and with tetrahydropteroylpolyglutamate substrates. *Biochemistry* 21: 1230-1238; 1982.
10. JENCKS, D. A.; MATTHEWS, R. G. Allosteric inhibition of methylenetetrahydrofolate reductase by adenosylmethionine: effects of adenosylmethionine and NADPH on the equilibrium between active and inactive forms of the enzyme and on the kinetics of approach to equilibrium. *J. Biol. Chem.* 262: 2485-2493; 1987.
11. CONWAY, J. G.; KAUFFMAN, F. C.; THURMAN, R. G. Genetic regulation of NADPH supply in perfused mouse liver. *J. Biol. Chem.* 258: 3825-3831; 1983.
12. HOFFMAN, D. R.; MARION, D. W.; CORNATZER, W. E.; DUERRE, J. A. S-Adenosylmethionine and S-adenosylhomocysteine metabolism in rat liver: effects of L-methionine, L-homocysteine, and adenosine. *J. Biol. Chem.* 255: 10822-10827; 1980.

*Received for publication August 27, 1987.
Accepted for publication September 23, 1987.*