

Arctic amplification of global warming strengthened by sunlight oxidation of permafrost carbon to CO₂

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Key Points:

- Wavelength dependence of permafrost dissolved organic carbon (DOC) photomineralization revealed a high lability to visible light.
- Iron catalyzes the photomineralization of old permafrost DOC (> 4,000 a BP) derived from soil lignin and tannin to carbon dioxide (CO₂).
- Photomineralization rates of permafrost DOC to CO₂ are double that of modern DOC, which will increase future arctic amplification.

Abstract

Once thawed, up to 15% of the ~1,000 Pg of organic carbon (C) in arctic permafrost soils may be oxidized to carbon dioxide (CO₂) by 2100, amplifying climate change. However, predictions of this amplification strength ignore the oxidation of permafrost C to CO₂ in surface waters (photomineralization). We characterized the wavelength dependence of permafrost dissolved organic carbon (DOC) photomineralization and demonstrate that iron catalyzes photomineralization of old DOC (4,000-6,300 a BP) derived from soil lignin and tannin. Rates of CO₂ production from photomineralization of permafrost DOC are two-fold higher than for modern DOC. Given that model predictions of future net loss of ecosystem C from thawing permafrost do not include the loss of CO₂ to the atmosphere from DOC photomineralization, current predictions of an average of 208 Pg C loss by 2299 may be too low by ~14%.

Plain Language Summary

The thawing of organic carbon stored in arctic permafrost soils, and its oxidation to carbon dioxide (a greenhouse gas), is predicted to be a major, positive feedback on global warming. However, current estimates of the magnitude of this feedback do not include the oxidation of permafrost soil organic carbon flushed to sunlit lakes and rivers. Here we show that ancient dissolved organic carbon (> 4,000 years old) draining permafrost soils is readily oxidized to carbon dioxide by sunlight. As a consequence, current estimates of additional global warming from the permafrost carbon feedback are too low.

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35 1 Introduction

Current estimates are that 5 to 15% of the ~1,000 Pg of the soil organic carbon (C) stored in surface permafrost soils could be emitted as greenhouse gases by 2100 given the current trajectory of global warming (Schuur et al., 2015; Plaza et al., 2019), with additional C lost in lateral transfer from soils to surface waters (Plaza et al., 2019). Models assessing the sensitivity of the climate system to thawing permafrost soils estimate that decomposition of organic C in these soils could result in 0.3 °C to 0.4 °C additional global warming (i.e., arctic amplification) by 2100 to 2299, respectively (McGuire et al., 2018). However, none of these predictions include the oxidation of organic C upon export to sunlit surface waters.

Oxidation of dissolved organic carbon (DOC) to carbon dioxide (CO₂) by sunlight (photomineralization) currently accounts for up to 30% of the CO₂ emitted to the atmosphere from arctic surface waters (Cory et al., 2014). As permafrost DOC is exported to sunlit waters, its oxidation to CO₂ will depend on whether permafrost DOC is labile to photomineralization, which is currently debated (Stubbins et al., 2016; Ward & Cory, 2016; Selvam et al., 2017; Shirokova et al., 2019). The lability of terrestrially-derived DOC to photomineralization is hypothesized to depend on iron and DOC chemical composition (Miles & Brezonik, 1981; Gao & Zepp, 1998; Xie et al., 2004; Ward & Cory, 2016; Gu et al., 2017). To test these hypothesized controls, we made the first direct measurements of the amount, source, and age of CO₂ produced from photomineralization of permafrost DOC collected on younger and older glacial surfaces, and from two common vegetation types in the Arctic (Table S1; Ping et al., 1998; Trusiak et al., 2019). Here we show that (i) the lability of permafrost DOC to photomineralization depends on sunlight wavelength, (ii) iron controls the lability of permafrost DOC to photomineralization, and (iii) old carboxylic acid C (4,000 to 6,300 a BP) derived from lignin and tannin is mineralized to CO₂ by sunlight. Collectively, our results support the inclusion of photomineralization in model predictions and experimental studies of arctic amplification of climate change.

60 2 Methods

2.1 Permafrost soil collection

Soils were collected from the frozen permafrost layer (> 60 cm below the surface) at five sites underlying moist acidic tussock or wet sedge vegetation, and on three glacial surfaces on the North Slope of Alaska during summer 2018 (Table S1; Mull & Adams, 1989; Walker et al., 2005; Hobbie & Kling, 2014). See the Supporting Information for soil collection protocols, including precautions to minimize radiocarbon (¹⁴C) contamination, and for a summary of the experimental design. DOC was leached from each permafrost soil (leachate) at the Woods Hole Oceanographic Institution (WHOI) as described in the Supporting Information.

2.2 $\Delta^{14}\text{C}$ and $\delta^{13}\text{C}$ analyses of DOC

The ¹⁴C and stable carbon (¹³C) isotopic compositions of DOC were analyzed from each permafrost leachate at the National Ocean Sciences Accelerator Mass Spectrometry (NOSAMS) facility at WHOI (Table S2) following Beaupré et al. (2007). Each permafrost leachate was diluted with UVC-oxidized MilliQ water (Millipore Simplicity ultraviolet, UV, system; 1.5 hr; 1200 W medium pressure mercury arc lamp) to achieve a total C mass between 800 and 2000 µg. The diluted permafrost leachate was acidified with UVC-oxidized trace-metal grade phosphoric acid (85%) to pH < 2 in a precombusted quartz reactor (450 °C; 4 hr) and the

dissolved inorganic carbon (DIC) was purged with high-purity helium gas in the dark. The DOC was then oxidized with UVC light to DIC for 4 hr (1200 W medium pressure mercury arc lamp), and the resultant CO₂ was extracted cryogenically. On average, 1370 ± 240 μg of C were extracted from each permafrost leachate (± 1 standard error, SE; *n* = 6; Table S2). A subsample of the CO₂ was analyzed for ¹³C using a VG Prism-II or Optima stable isotope ratio mass spectrometer (instrumental precision of 0.1‰; Coplen et al., 2006), and the δ¹³C (‰) was calculated as follows:

$$\delta^{13}\text{C} = ({}^{13}\text{R}_{\text{sample}}/{}^{13}\text{R}_{\text{standard}} - 1) \quad (1)$$

where ¹³R is the isotope ratio of a sample or standard (VPDB), as defined by:

$${}^{13}\text{R} = ({}^{13}\text{C}/{}^{12}\text{C}) \quad (2)$$

The remaining CO₂ was reduced to graphite with H₂ and an iron catalyst, and then analyzed for ¹⁴C isotopic composition using an accelerator mass spectrometer at the NOSAMS facility (Longworth et al., 2015). The Δ¹⁴C (‰) and radiocarbon age of DOC were calculated from the fraction modern (Stuiver & Polach, 1977; McNichol et al., 2001) using the oxalic acid I standard (NIST-SRM 4990). Δ¹⁴C analyses of DOC had an instrumental precision of 2-6‰ (Longworth et al., 2015; McNichol et al., 2001).

DOC leached from one permafrost soil (Toolik moist acidic tundra) was prepared and analyzed for ¹⁴C and ¹³C twice to quantify the standard error of duplicate analyses (Table S2). Δ¹⁴C and δ¹³C analyses of DOC had standard errors of 1‰ and 0.1‰, respectively (1 SE; *n* = 2; Table 1). A procedural blank was quantified manometrically by oxidizing MilliQ water with UVC light in a precombusted quartz reactor (450 °C; 4 hr) for 1.5 hr, acidifying to pH < 2, and purging the DIC as described above. The procedural blank was 4 μg of C, which was < 0.5% of the total C masses extracted from the permafrost leachates.

2.3 Apparent quantum yield spectra

The CO₂ produced from photomineralization of permafrost DOC was measured as a function of sunlight wavelength. The lability of DOC to photomineralization is defined as the apparent quantum yield spectrum (CO₂ produced per mol photon absorbed by DOC; hereafter called the yield spectrum, φ_{PM,λ}). Yield spectra of permafrost DOC were directly measured for the first time with a custom-built high-powered (≥ 100 mW), narrow-banded (± 10 nm) light-emitting diode (LED) system from soils collected in 2018 (Figure S1). Each permafrost leachate was equilibrated to room temperature (~24 hr) and then placed in 20 gas-tight, flat-bottomed precombusted (450 °C; 4 hr) 12 mL quartz vials with butyl rubber septa and GL-18 caps (light-exposed vials) and four gas-tight precombusted (450 °C; 4 hr) 12 mL borosilicate exetainer vials (dark control vials; Labco, Inc.). Vials were placed in an inner aluminum housing (painted black matte to minimize light scattering), with the flat bottom facing upward toward the light source, and then exposed to ≥ 100 mW, narrow-banded (± 10 nm) LEDs at 278, 309, 348, 369, and 406 nm alongside the dark controls for 12 or 30 hr (Table S3). The LEDs were tuned such that each permafrost leachate absorbed the same amount of light at each wavelength. After LED exposure, light-exposed and dark control waters were immediately analyzed for DIC (Apollo SciTech, Inc.) and for chromophoric dissolved organic matter (CDOM; Cory et al., 2014). The experiment above was then repeated for the analysis of photochemical oxygen (O₂) consumption to quantify the apparent quantum yield spectra of photo-oxidation (φ_{PO,λ}) from each permafrost DOC.

120 Dissolved O₂ was measured in each light-exposed and dark control vial on a membrane inlet mass spectrometer (Bay Instruments; Kana et al., 1994).

At each LED wavelength, $\phi_{\text{PM},\lambda}$ and $\phi_{\text{PO},\lambda}$ were calculated as the concentration of DIC produced and O₂ consumed, respectively, divided by the light absorbed by CDOM. The amount of light absorbed by CDOM (mol photon m⁻² nm⁻¹) was quantified for each vial exposed to a LED using absorption coefficients of CDOM ($a_{\text{CDOM},\lambda}$) and the photon flux spectrum (Cory et al., 2014). The photon flux spectrum was quantified from the solar irradiance spectrum from each LED source, which was measured by radiometry and chemical actinometry (see Supporting Information). $\phi_{\text{PM},\lambda}$ and $\phi_{\text{PO},\lambda}$ are reported as the average \pm 1 SE of experimental replicate vials ($n = 4$).

2.4 $\Delta^{14}\text{C}$ and $\delta^{13}\text{C}$ of CO₂ produced from light

130 The $\Delta^{14}\text{C}$ and $\delta^{13}\text{C}$ of DIC produced following exposure of DOC to UV and visible light were quantified from permafrost leachates prepared from each permafrost soil collected in 2018, except for Sagwon moist acidic tundra (Figure S1). Each permafrost leachate was equilibrated to room temperature and then placed in up to four precombusted (450 °C; 4 hr) 600 mL quartz flasks with ground glass stoppers and no headspace. The flasks were exposed to custom-built LED arrays consisting of ten \geq 100 mW, narrow-banded (\pm 10 nm) 309 or 406 nm chips alongside one or two foil-wrapped dark control flasks (Table S2). Exposure times ranged from 8 to 25 hr to achieve similar concentrations of DIC produced from each permafrost DOC sample and at each wavelength (Table S4).

140 After LED exposure, foil-wrapped light-exposed and dark control flasks were immediately transferred to foil-wrapped, precombusted 500 mL borosilicate glass bottles (450 °C; 4 hr) in a N₂-filled glove bag, preserved with saturated mercuric chloride, and plugged with gas-tight ground glass stoppers (McNichol et al., 1994). Those bottles were stored in the dark at room temperature for \leq 1 week until preparation for carbon isotope analyses at the NOSAMS facility. Bottles were kept foil-wrapped while each water sample was acidified with trace-metal grade phosphoric acid (85%) to pH $<$ 2 and stripped of DIC with high-purity N₂ gas. The resultant CO₂ was trapped and purified cryogenically and its concentration was quantified manometrically. The ¹⁴C and ¹³C of the CO₂ were analyzed at the NOSAMS facility (Table S2) and converted to $\Delta^{14}\text{C}$ and $\delta^{13}\text{C}$ values as described above. $\Delta^{14}\text{C}$ analyses of DIC had an instrumental precision of 1-2‰ (Longworth et al., 2015; McNichol et al., 2001). The reported precision of $\delta^{13}\text{C}$ is 0.1‰ (Coplen et al., 2006).

The $\Delta^{14}\text{C}$ and $\delta^{13}\text{C}$ of CO₂ produced from the photomineralization of DOC were calculated as follows:

$$\Delta^{14}\text{C-CO}_{2,\lambda} = \frac{(\Delta^{14}\text{C-DIC}_{\text{Light},\lambda} * [\text{DIC}]_{\text{Light},\lambda}) - (\Delta^{14}\text{C-DIC}_{\text{Dark}} * [\text{DIC}]_{\text{Dark}})}{([\text{DIC}]_{\text{Light},\lambda} - [\text{DIC}]_{\text{Dark}})} \quad (3)$$

$$\delta^{13}\text{C-CO}_{2,\lambda} = \frac{(\delta^{13}\text{C-DIC}_{\text{Light},\lambda} * [\text{DIC}]_{\text{Light},\lambda}) - (\delta^{13}\text{C-DIC}_{\text{Dark}} * [\text{DIC}]_{\text{Dark}})}{([\text{DIC}]_{\text{Light},\lambda} - [\text{DIC}]_{\text{Dark}})} \quad (4)$$

155 The $\Delta^{14}\text{C}$ and $\delta^{13}\text{C}$ of CO₂ produced in each light-exposed flask were calculated relative to one or two dark controls (Tables S2, S4) and are reported as the average \pm 1 SE of replicate values

for the experiments conducted alongside two dark controls (Table 1). The concentration, $\Delta^{14}\text{C}$, and $\delta^{13}\text{C}$ of DIC in the dark controls are reported as the average \pm 1 SE of replicate flasks ($n = 2$; Table S4). The ^{14}C age of CO_2 produced is the age of DIC in light-exposed flask minus the ^{14}C age of DIC in the dark control (Table 1). This approach to quantify the $\Delta^{14}\text{C}$ and $\delta^{13}\text{C}$ of CO_2 produced from photomineralization of organic C was previously described in detail for polystyrene (Ward et al., 2019). In this previous study, experimental reproducibility of $\Delta^{14}\text{C}$ and $\delta^{13}\text{C}$ of CO_2 produced from photomineralization was 5‰ and 0.1‰, respectively (\pm 1 SE; $n = 3$).

3 Results and Discussion

All permafrost DOC was labile to photomineralization at all wavelengths measured (Figure 1a), and the yield spectrum always decreased exponentially with increasing wavelength from the UV to the visible ($p < 0.05$; Figure 1a). The magnitude of the photomineralization yield varied up to 8-fold among permafrost DOC samples (Figure 1a) and was significantly, positively correlated with the concentration of dissolved iron ($p < 0.001$ as shown at 309 nm in Figure 1b; see Supporting Information). There were no significant correlations of the photomineralization yield with dissolved cations other than iron or with any measure of DOC concentration or composition (see Supporting Information).

Our results are the first to demonstrate in natural samples that the lability of permafrost DOC to photomineralization is controlled by dissolved iron. Although photomineralization of terrestrially-derived DOC has been shown to increase with addition or decrease with removal of iron in the laboratory (Miles & Brezonik, 1981; Gao & Zepp, 1998; Xie et al., 2004; Gu et al., 2017), evidence is lacking for a mechanism by which iron enhances photomineralization. Here we present new experimental evidence and a synthesis of literature results that collectively support a mechanism of iron-catalyzed photo-decarboxylation of lignin- and tannin-derived carboxylic acids within old permafrost DOC.

Iron is hypothesized to catalyze the photo-decarboxylation of organic acids by a Ligand-Metal-Charge-Transfer reaction (Mangiante et al., 2017) where Fe(III) is a cyclic catalyst that is photo-reduced to Fe(II) while the C in carboxylic acids (hereafter “carboxyl C”) is oxidized to CO_2 (Miles & Brezonik, 1981; Gao & Zepp, 1998; Xie et al., 2004). Two lines of evidence from our study strongly support iron-catalyzed photo-decarboxylation of permafrost DOC to CO_2 . First, loss of carboxyl C (quantified by ^{13}C nuclear magnetic resonance) upon exposure of permafrost DOC to sunlight was significantly, positively correlated with the dissolved iron concentration ($p < 0.05$; Figure S2), as expected if photo-decarboxylation is the mechanism of CO_2 production. The only other study that quantified photochemical loss of carboxyl C from permafrost DOC concluded that it accounted for up to 90% of the CO_2 produced from photomineralization (Ward & Cory, 2016). However, this prior study used DOC from one site (with only one iron concentration) and thus was unable to link iron abundance to DOC photo-decarboxylation. Second, the ratio of photochemical CO_2 production per dissolved O_2 consumption by DOC was ≥ 1 for all permafrost DOC that also contained $> 1 \mu\text{M}$ total dissolved iron (Figure S3). A ratio ≥ 1 for photochemical CO_2 produced per O_2 consumed is considered evidence for photo-decarboxylation because this reaction is expected to proceed with a stoichiometry of $\geq 2:1$ mol CO_2 produced per mol O_2 consumed (Miles & Brezonik, 1981; Xie et al., 2004). While ratios ≥ 1 for photochemical CO_2 produced per O_2 consumed have previously been observed in high-iron waters (Miles & Brezonik, 1981; Xie et al., 2004; Cory et al., 2015),

200 here we show ratios ≥ 1 for DOC from various permafrost soils concurrent with photochemical loss of carboxyl C.

Isotopic signatures of the CO₂ produced by sunlight indicate that iron is catalyzing the oxidation of carboxyl C attached to organic matter derived from lignin and tannin. Photochemical production of ¹³C-depleted CO₂ (Table 1) increased significantly with the ratio of photochemical CO₂ produced per mol of O₂ consumed ($p < 0.05$; Figure S4). Therefore, as photo-decarboxylation accounts for more of the total CO₂ produced (as indicated by increasing CO₂/O₂; Figure S4), the source of the CO₂ is increasingly ¹³C-depleted carboxyl C, such as that derived from lignin and tannin (Ball & Aluwihare, 2014). Photochemical production of ¹³C-depleted CO₂ is interpreted as resulting from photomineralization of lignin- or tannin-derived DOC (Spencer et al., 2009; Franke et al., 2012) because lignin and tannin are relatively more depleted in ¹³C compared to other fractions of DOC (Benner et al., 1987; see Supporting Information). In addition, the ¹³C enrichment of DOC remaining after photomineralization has been correlated with photochemical loss of lignin (Spencer et al., 2009), and high-resolution mass spectrometry revealed that lignin- and tannin-derived compounds within permafrost DOC are preferentially degraded by sunlight compared to other fractions of DOC (Ward & Cory, 2016; Ward et al., 2017). Thus, our results indicate that iron-catalyzed photo-decarboxylation of lignin and tannin in permafrost DOC is producing CO₂ (Table 1, Figures S2, S4).

The carboxyl C derived from lignin and tannin that was photomineralized to CO₂ was old, from 4,000 to 6,300 a BP (Table 1, Figure 2). The $\Delta^{14}\text{C}$ composition of CO₂ produced from photomineralization of permafrost DOC (-546‰ to -397‰) was always $\leq 70\%$ different from the initial, bulk $\Delta^{14}\text{C}$ -DOC signature (-585‰ to -411‰; Table 1, Figure 2). The linear relationship between the initial, bulk permafrost $\Delta^{14}\text{C}$ -DOC and the $\Delta^{14}\text{C}$ -CO₂ produced by photomineralization ($p < 0.05$; Figure 2) indicates that the bulk age of permafrost DOC was a strong predictor of the age of DOC photomineralized to CO₂. Collectively, our results demonstrate that old carboxyl C (4,000 to 6,300 a BP) derived from lignin and tannin and associated with iron is photomineralized to CO₂.

The presence of iron may explain contrasting literature results from high (Ward & Cory, 2016; Selvam et al., 2017; Shirokova et al., 2019) to little or no (Stubbins et al., 2016; Shirokova et al., 2019) lability of permafrost DOC to photomineralization. For example, undetectable photomineralization of permafrost DOC from Russian arctic thaw slumps (Stubbins et al., 2016) may have been due to the 100-fold dilution of the DOC with deionized water. Although dissolved iron was not reported (Stubbins et al., 2016), dilution likely also resulted in the precipitation of iron (oxy)hydroxides and thus lower dissolved iron concentrations (Miller et al., 2009). In a study of Russian arctic surface waters that likely contained permafrost DOC (Shirokova et al., 2019), up to 13% of the DOC pool was photomineralized to CO₂, consistent with the presence of dissolved iron (3-7 μM). Provided that all permafrost DOC contains carboxyl C (Ward & Cory, 2016; Feng et al., 2017; Ward et al., 2017) and that permafrost soils generally contain high levels of leachable iron (Ping et al., 1998; Herndon et al., 2015; Heslop et al., 2019; Trusiak et al., 2019), arctic permafrost DOC is labile to photomineralization in proportion to the iron present. Given that the export of iron is currently strongly, positively correlated with DOC export from arctic soils to surface waters (Trusiak et al., 2019), we expect that iron and DOC export may continue to co-vary as permafrost soils thaw. Therefore, we predict that the yield spectrum of permafrost DOC will be within the range reported here in the future (Figure 1a).

245 The photomineralization yield spectra of permafrost DOC directly measured in this study
have significantly shallower spectral slopes compared to those quantified indirectly for arctic
surface water DOC (two-tailed, unpaired t-test; $p < 0.001$; Figure S5). Thus, permafrost DOC
has relatively lower lability to photomineralization at UV wavelengths and higher lability at
visible wavelengths (different at 95% confidence interval; Figure S5). If permafrost DOC
250 comprises 100% of the DOC in surface water, photomineralization rates will increase by two-
fold compared to current rates (Figure 3a) due to the higher lability of permafrost DOC at visible
wavelengths multiplied by the ~10-fold greater photon flux in the visible versus the UV light
region (Table S5). It follows that photomineralization rates increase in proportion to the
permafrost DOC exported to surface waters (Figure 3b).

255 **4 Implications**

The uncertainty in model predictions of future ecosystem C gain or loss crosses zero
(McGuire et al., 2018). For example, under the RCP8.5 scenario a net ecosystem loss of C of
 208 ± 307 Pg C is predicted by 2299 (average ± 1 SD). The large uncertainty in model
predictions of permafrost C storage in this scenario includes a 20% probability that the net C
260 storage is between +100 (gain) or -100 (loss) Pg C (see Supporting Information).
Photomineralization of DOC to CO₂ is always a loss to the atmosphere, and as the net C gain or
loss for any particular year or over time nears zero, the relative importance of
photomineralization increases. Given that photomineralization rates of permafrost DOC are
nearly two-fold higher than for modern DOC in arctic surface waters (Figure 3a), and assuming
265 from 2010 to 2299 75% of DOC in surface waters was delivered from permafrost soils, then the
photomineralization rates of $20 \text{ g C m}^{-2} \text{ y}^{-1}$ reported in Cory et al. (2014) would increase to $39 \text{ g C m}^{-2} \text{ y}^{-1}$.
Using a surface area of water in permafrost regions of 6% (Cooley et al., 2019), ~9 Pg
CO₂ could be produced from the photomineralization of permafrost DOC by 2299 (see
Supporting Information). In addition, if potentially more than half of future terrestrial C losses
270 are lateral in hydrologic flow (Plaza et al., 2019), a pathway missing from models used in
McGuire et al. (2018), photomineralization of that C would occur upon exposure to sunlight. For
example, taking the predicted net ecosystem loss of 208 Pg C by 2299 under RCP8.5 (McGuire
et al., 2018), potentially another ~100 Pg C could be lost laterally and produce ~21 Pg C as CO₂
from photomineralization in surface waters (Cory et al., 2014; see Supporting Information). CO₂
275 from photodegradation of permafrost DOC is conservative because it does not account for (a)
increased lability of permafrost DOC to microbial respiration following exposure to sunlight
(Cory et al., 2013; Ward et al., 2017), and (b) increased annual sunlight exposure due to more
ice-free days for surface waters in a warmer Arctic (Šmejkalová et al., 2016). Therefore,
reducing the uncertainty on whether permafrost thaw will be a net sink or source of C to the
280 atmosphere requires representing processes such as photochemistry in models of the future arctic
C balance.

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degradation-dissolved-organic-carbon/data-photo-coupled](https://arc-lter.ecosystems.mbl.edu/arctic-coupled-biological-and-photochemical-degradation-dissolved-organic-carbon/data-photo-coupled)).

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Figure 1. Controls on the lability of permafrost DOC to photomineralization. (a) Wavelength-dependent apparent quantum yield spectrum for photomineralization ($\phi_{\text{PM},\lambda}$) of permafrost DOC. Each data series was fit with a least-squares exponential model where $R^2 > 0.83$, $p < 0.05$. (b) Apparent quantum yield for photomineralization at 309 nm ($\phi_{\text{PM},309}$) versus total dissolved iron concentration in permafrost leachates prior to light exposure. Closed symbols indicate $\phi_{\text{PM},309}$ measured following LED exposure at 309 nm. Open symbols indicate $\phi_{\text{PM},309}$ estimated from an exponential fit following exposure to broadband light (see Methods). Data in (b) were fit using a least-squares regression where $R^2 = 0.87$, t-statistic = 7.8, $p < 0.001$, excluding the open red symbol (see Supporting Information). Open symbols for Innavait moist acidic tundra were previously reported (Ward & Cory, 2016). All values are the average ± 1 SE of experimental replicates ($n = 2$ and 4 for open and closed symbols, respectively; see Methods). $\phi_{\text{PM},\lambda}$ at other wavelengths versus dissolved iron are reported in the Supporting Information.

Figure 2. $\Delta^{14}\text{C}$ of bulk permafrost DOC was a strong predictor of the $\Delta^{14}\text{C}\text{-CO}_2$ produced from photomineralization of DOC. $\Delta^{14}\text{C}\text{-CO}_2$ produced from exposure of permafrost DOC to UV (309 nm, diamond symbols) and visible (406 nm, square symbols) light versus $\Delta^{14}\text{C}$ of initial, bulk permafrost DOC plotted with the 1:1 line. When all data are fit using a least-squares regression, $R^2 = 0.66$, t-statistic = 3.4, $p < 0.05$. Values for photochemically produced $\Delta^{14}\text{C}\text{-CO}_2$ are shown as the average ± 1 SE of experimental replicates ($n = 2$).

Figure 3. Photomineralization rates were higher for permafrost DOC than for surface water DOC due to higher lability in the visible light region. (a) Wavelength-dependent water column rates of photomineralization for Innavait moist acidic tundra permafrost DOC and Innavait Creek DOC (turquoise versus black line, respectively). Solid lines show the average photomineralization rate spectrum and the similar color shading shows the upper and lower 95% confidence intervals. Rates of photomineralization were calculated using the different wavelength-dependent apparent quantum yields for photomineralization ($\phi_{\text{PM},\lambda}$) for Innavait moist acidic tundra permafrost DOC or Innavait Creek DOC (Figure S5; see Supporting Information). The red dashed and solid lines mark the wavelength of peak photomineralization rates (330 nm versus 402 nm) for Innavait Creek and Innavait moist acidic tundra permafrost DOC, respectively. (b) Calculated photomineralization rates increased with an increasing fraction of permafrost DOC in the surface water DOC pool. Percent increase in photomineralization rates as permafrost DOC comprises 0-100% of the DOC pool in Innavait Creek, Kuparuk River, and Toolik Lake (compared to no permafrost DOC in the DOC pool). Only the $\phi_{\text{PM},\lambda}$ was varied in the water column rate calculations, using a 'mixed' $\phi_{\text{PM},\lambda}$ calculated as a mixture of the $\phi_{\text{PM},\lambda}$ for permafrost DOC with the $\phi_{\text{PM},\lambda}$ for the surface water DOC (see Supporting Information). All values are shown as the average ± 1 SE of calculated rates for surface water DOC mixed with each of the five permafrost DOCs in this study ($n = 5$).

540 **Table 1.** $\Delta^{14}\text{C}$ and $\delta^{13}\text{C}$ of initial, bulk permafrost DOC and the CO_2 produced by UV and visible light.

	Light source	Toolik moist acidic tundra	Imnavait moist acidic tundra	Imnavait wet sedge tundra	Sagwon wet sedge tundra	Sagwon moist acidic tundra
$\Delta^{14}\text{C}$ -DOC (‰)		-436 ± 1	-474	-585	-411	-519
^{14}C age of DOC (a BP)		$4,520 \pm 5$	$5,080$	$6,990$	$4,300$	$5,890$
$\delta^{13}\text{C}$ -DOC (‰)		-25.5 ± 0.1	-25.4	-26.1	-27.8	-26.9
$\Delta^{14}\text{C}$ - CO_2 produced (‰)	UV	-487 ± 4	-465 ± 8	-546 ± 11	-397 ± 1	N/A
	Visible	-506	-454 ± 8	-538 ± 13	-402	N/A
^{14}C age of CO_2 produced (a BP)	UV	$5,300 \pm 55$	$4,950 \pm 110$	$6,270 \pm 190$	$4,000 \pm 5$	N/A
	Visible	$5,600$	$4,800 \pm 120$	$6,150 \pm 220$	$4,070$	N/A
$\delta^{13}\text{C}$ - CO_2 produced (‰)	UV	-30.2 ± 0.4	-31.0 ± 0.4	-31.8 ± 1.8	-35.8 ± 0.6	N/A
	Visible	-29.8	-28.3 ± 0.4	-29.2 ± 2.0	-36.0	N/A

Note. The $\Delta^{14}\text{C}$ and $\delta^{13}\text{C}$ of the CO_2 produced from photomineralization of permafrost DOC are reported following exposure to LEDs at 309 (UV) or 406 nm (visible). When available, all values are reported as the average ± 1 SE of experimental replicates ($n = 2$).





