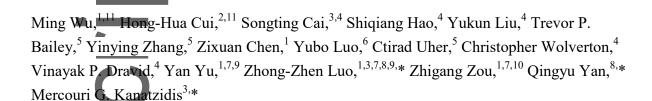
Weak electron-phonon coupling and enhanced thermoelectric performance in n-type

PbTe-Cu₂Se via dynamic phase conversion



¹Key Laboratory of Eco-materials Advanced Technology, College of Materials Science and Engineering, Fuzhou University, Fuzhou, 350108, P. R. China

²Mechanical and Electrical Engineering Practice Center, Fuzhou University, Fuzhou, 350108, P. R. China

³Department of Chemistry, Northwestern University, Evanston, Illinois 60208, United States ⁴Department of Materials Science and Engineering, Northwestern University, Evanston, Illinois 60208, United States

⁵Department of Physics, University of Michigan, Ann Arbor, Michigan 48109, United States ⁶State Key Laboratory of Materials Processing and Die & Mould Technology, School of Materials Science and Engineering, Huazhong University of Science and Technology, Wuhan 430074, P. R. China

⁷Fujian Science & Technology Innovation Laboratory for Optoelectronic Information of China, Fuzhou, Fujian 350108, P. R. China

⁸School of Materials Science and Engineering, Nanyang Technological University, 50 Nanyang Avenue 639798, Singapore

⁹Key Laboratory of Advanced Materials Technologies, International (HongKong Macao and Taiwan) Joint Laboratory on Advanced Materials Technologies, College of Materials Science and Engineering, Fuzhou University, Fuzhou, Fujian, 350108, P. R. China

This is the cuther canuscript accepted for publication and has undergone full peer review but has not been through the copyediting, typesetting, pagination and proofreading process, which may lead to difference betteen this version and the <u>Version of Record</u>. Please cite this article as <u>doi:</u> 10.1002/ac. 02203325.

¹⁰Eco-materials and Renewable Energy Research Center, College of Engineering and Applied Sciences, Nanjing University, Nanjing, 210093, P. R. China

¹¹These authors contributed equally: Ming Wu, Hong-Hua Cui

ABSTRACT

This study investigates Ga-doped n-type PbTe thermoelectric materials and the dynamic phase conversion process of the second phases via Cu_2Se alloying. Introducing Cu_2Se enhances its electrical transport properties while reducing its lattice thermal conductivity (κ_{lat}) via weak electron—phonon coupling. Cu_2Te and $CuGa(Te/Se)_2$ (tetragonal phase) nanocrystals precipitated during the alloying process, resulting in Te vacancies and interstitial Cu in the PbTe matrix. At room temperature, Te vacancies and interstitial Cu atoms serve as n-type dopants, increasing the carrier concentration and electrical conductivity from ~1.18 × 10^{19} cm⁻³ and ~1870 Scm⁻¹ to ~2.26 × 10^{19} cm⁻³ and ~3029 Scm⁻¹, respectively. With increasing temperature, the sample exhibits a dynamic change in Cu_2Te content and the generation of a new phase of $CuGa(Te/Se)_2$ (cubic phase), strengthening the phonon scattering and obtaining an ultralow κ_{lat} . $Pb_{0.975}Ga_{0.025}Te-3\%CuSe$ exhibits a maximum figure of merit of ~1.63 at 823 K, making it promising for intermediate-temperature device applications.

Keywords: thermoelectric, n-type PbTe, Cu₂Se alloying, electron–phonon coupling, dynamic phase conversion

INTRODUCTION

Thermoelectric materials, which can collect and convert waste heat directly into electricity, can enhance energy utilization and efficiency. The dimensionless thermoelectric figure of merit (ZT) quantifies the efficiency of thermoelectric materials as $ZT = \sigma S^2 T/(\kappa_{\rm ele} + \kappa_{\rm lat})$, where σ , S, T, $\kappa_{\rm ele}$ and $\kappa_{\rm lat}$ represent the electrical conductivity, Seebeck coefficient, working temperature (K), and electronic and lattice thermal conductivity, respectively. To maximize thermoelectric performance, an ideal thermoelectric material must possess a high power factor ($PF = \sigma S^2$), and a low total thermal conductivity ($\kappa_{\rm tot} = \kappa_{\rm ele} + \kappa_{\rm lat}$). However, σ , S, and $\kappa_{\rm ele}$ are strongly coupled and intertwined with the carrier concentration (n). More carriers, for example, result in higher σ ($\sigma = e\mu n$); $\kappa_{\rm ele}$ is governed by the Wiedemann–Franz relation ($\kappa_{\rm ele} = L\sigma T$, where L represents the Lorenz number); and S is inversely proportional to n. Therefore, decoupling the negatively correlated parameters is a primary goal for enhancing the ZT value.

A new generation of PbTe-based materials has exceptional p-type performance with ZT ~2.5 as the only commercially available intermediate temperature range thermoelectric material. However, advanced devices lag behind partly because the ZT for n-type PbTe is much lower due to the single conduction band (CB) structure versus the unique two-valence band (VB) structure for the p-type counterpart. Importantly, p- and n-type PbTe materials are required for module fabrication for the practical application of thermoelectric technology. Recently, effective approaches such as electronic band structure modulation (including midgap states via Ga doping and Ge alloying, deep defect level via In doping, shallow and deep level donor states via Ga doping, and band flattening, and band phonon scattering manipulations (including multiphase nanostructures, dislocations, li2]

discordant off-center behavior,^[1h, 13] and dual-site point defects,^[14] etc.) have been adopted to investigate the high performance n-type PbTe.

Previous research revealed that Ga doping can effectively increase the S via introducing midgap states in the band gap of PbTe, $^{[8, 10]}$ when compared to conventional n-type dopants including $L^{[15]}$ Sb, $^{[16]}$ Bi, $^{[16]}$ and La. $^{[17]}$ In contrast, Cu atoms have been demonstrated to fill the intrinsic Pb vacancies to improve carrier mobility ($\mu_{\rm H}$) and induce local disorder and to form nanoscale and microscale Cu-rich precipitates, thereby reducing $\kappa_{\rm lat}$. $^{[18]}$ Moreover, Cu atoms can form sub-nanostructure defects in the PbTe or SnTe $^{[19]}$ matrix as interstitials, resulting in locally disordered matrix lattices with sizes comparable to the phonon mean free path (0.1–10 nm) but much smaller than the PbTe carrier mean free path (100–1000 nm). $^{[20]}$ Therefore, these sub-nanostructure defects are beneficial for scattering phonons and maintaining electron mobility in the matrix.

In this study, we demonstrated that Cu_2Te and $CuGa(Te/Se)_2$ (tetragonal phase) nanocrystals precipitated in Ga-doped n-type PbTe with Cu_2Se alloying, resulting in the formation of Te vacancies and interstitial Cu and an increase in n. The content of the Cu_2Te second phase increased with increasing temperature, as measured using powder X-ray diffraction (PXRD). Moreover, the Cu_2Te second phase disappeared and the $CuGa(Te/Se)_2$ (cubic phase) appeared above 773 K. The dynamic phase conversion causes a significant reduction in κ_{lat} . As a result of the synergistically increased σ (from ~1870 Scm⁻¹ for Pb_{0.975}Ga_{0.025}Te to ~3029 Scm⁻¹ for x = 3) and μ_H , and the suppressed κ_{lat} (from ~1.70 Wm⁻¹K⁻¹ for x = 0 to ~1.36 Wm⁻¹K⁻¹ for x = 3 samples) via weakened electron—phonon coupling, a maximum ZT of ~1.63 at 823 K can be obtained in Pb_{0.975}Ga_{0.025}Te-3%Cu₂Se.

EXPERIMENTAL SECTION

Raw materials. The following high-purity commercial reagents were used: Pb wire (99.99%, American Elements, USA); Te shot (99.999%, American Elements, USA); Ga shot (99.999%, Sigma-Aldrich, USA); Se shot (99.999%, Canada); and granular Cu (99.9%, Sigma-Aldrich, USA).

Synthesis. Stoichiometric compositions of $Pb_{0.975}Ga_{0.025}Te-x\%Cu_2Se$ (x = 0, 1, 2, 3, 4, and 5) were melted by mixing the raw materials in flame-sealed 13 mm diameter fused quartz tubes under a residual pressure of $\sim 2 \times 10^{-3}$ Torr. The tubes were heated to 1373 K for 11 h and then soaked for 6 h before being quenched to room temperature. The tubes were shaken gently at 1373 K to achieve homogeneous mixing. To prepare a ~ 15 g ingot of $Pb_{0.975}Ga_{0.025}Te-3\%Cu_2Se$, the following amounts of raw materials were used: Pb (9 g, 43.44 mmol), Te (5.6846 g, 44.56 mmol), Ga (0.0777 g, 1.11 mmol), Se (0.1055 g, 1.34 mmol), and Cu (0.1699 g, 2.67 mmol).

Hall Coefficient Measurements. The Hall coefficient ($R_{\rm H}$) was measured on samples with dimensions of $<1 \times 3 \times 8$ mm³ held in an Ar-filled sample probe within a homemade apparatus using an AC 4-probe method. A linear research AC resistance bridge (LR-700) operating at 17 Hz was used to measure the Hall resistance. The signal was modulated using excitation fields of ± 0.5 Tesla provided by an air-bore Oxford superconducting magnet. The charge n and $\mu_{\rm H}$ were calculated as $n = 1/e|R_{\rm H}|$ and $\mu_{\rm H} = \sigma R_{\rm H}$, respectively, where e represents the electronic charge. Multiple measurements of the $R_{\rm H}$ at fixed temperatures revealed $\sim 5\%$ uncertainty in $R_{\rm H}$.

Thermal Conductivity. The thermal diffusivity (D) was measured directly on a Netzsch LFA457 with a continuous nitrogen flow using the laser flash analysis method. From the

SPSed pellets, a square-shaped sample of $\sim 6 \times 6 \times 1.5 \text{ mm}^3$ was cut and polished. To minimize emissivity errors, a thin layer of graphite spray coating was applied to squared sample surfaces. A Cowan model with pulse correction was used to analyze the D data. κ_{tot} was calculated using the expression $\kappa_{\text{tot}} = DC_p \rho$, where C_p and ρ represent the specific heat capacity and the density, respectively. The C_p data were taken from a previous study. The uncertainty of D in this measurement was determined to be within 5%. ρ was calculated as the mass of the samples divided by their volume. When all uncertainties from the D measurement, C_p calculation, and ρ test are considered, the combined relative uncertainty for ZT determination is $\sim 20\%$.

Scanning/Transmission Electron Microscopy characterization. Scanning/transmission electron microscopy (S/TEM) and STEM energy dispersive spectroscopy (EDS) were performed on a JEOL ARM300F microscope set to 300 kV. The electron-beam transmitted transmission electron microscopy (TEM) specimens were prepared using conventional methods including cutting, grinding, and Ar-ion milling (2.8 kV for ~0.5 h until a hole is formed, followed by ion cleaning at 0.3 kV for 1 h) at a low temperature (liquid nitrogen stage).

Electronic band-structure calculations. The relaxed geometries of various dopants in the PbTe host and the corresponding total energies were calculated using density functional theory (DFT) within the generalized gradient approximation of Perdew–Burke–Ernzerhof using the exchange-correlation functional with projector augmented wave potentials. Periodic boundary conditions and a plane wave basis set were used, as implemented in the Vienna ab initio simulation package. In all relaxation calculations, a basis set energy cutoff of 500 eV and a total energy numerical convergence criterion of \sim 3 meV/cation was adopted. For the k-point parameter, we used dense k-meshes with 4000 k-points per reciprocal atom in

the Brillouin zone. For the Ga-doped and Cu_2Se alloys, we first considered multiple configurations of one Cu and Ga substitution for two Pb atoms and Se substitution for Te in a $3 \times 3 \times 3$ PbTe supercell. For the band structure calculations, spin-orbit coupling is used to investigate the most energetically favorable configuration among the multiple possibilities.

Lattice vibration calculations. The frozen-phonon method was used to calculate the dynamical properties of relaxed structures. [24] The force-constant matrix is constructed using ab initio forces exerted on all cell atoms when a specific cell atom is slightly displaced from its equilibrium position. DFT calculations were performed with the VASP^[25] code and the electronic exchange-correlation functional GGA-PBE^[22]. Diagonalization of the corresponding dynamical matrix yielded the vibrational phonon modes. When compared to experiments with low-conductivity thermoelectric compounds, this method has recently been shown to produce accurate values of κ_{lat} . [26]

RESULTS AND DISCUSSION

Structural characterization. The PXRD patterns of the Pb_{0.975}Ga_{0.025}Te-x%Cu₂Se (x = 0, 1, 2, 3, 4, and 5) samples at room temperature are shown in Figure S1 (Supporting Information). With a space group of $Fm\overline{3}m$, all of the main peaks can be well indexed to the PbTe phase (JCPDS 78-1905). Furthermore, the tetragonal CuGa(Te/Se)₂ phase (T-phase, $I\overline{4}2d$, a = 6.0235 Å, c = 11.9398 Å, V = 433.20 Å³), became apparent with increasing Cu₂Se content, as indicated by the two red stars, which is consistent with the extremely low solid solubility of Cu in the PbTe matrix (Figure S2, Supporting Information). However, Xiao et al. reported that the I-doped PbTe-x%Cu₂Te samples exhibited no second phases with x < 5.5%. [18] The peaks shifted to a high-angle direction, indicating a solid solution of Se and Te, as confirmed

by PXRD patterns of CuGaTe_{2-x}Se_x (Figure S3, Supporting Information) and EDS mappings (discussed later). The lattice parameters gradually decrease with increasing Cu₂Se content due to Cu¹ (0.96 Å) and Se²⁻ (1.98 Å) substitutions at Pb²⁺ (1.20 Å) and Te²⁻ (2.21 Å) sites, respectively (Figure S4, Supporting Information).

Temperature-dependent PXRD was used to investigate the dynamic doping effect of Cu. The content of the Cu₂Te second phase increased and was detected at T > 473 K, as shown in Figure S5 (Supporting Information). Furthermore, the Cu₂Te phase disappeared, and the cubic CuGaTe₂ phase (C-phase, $F\overline{4}3m$, a = 6.1639 Å, V = 234.18 Å³) appeared above 723 K, as shown in Figure 1. According to the conservation of Cu atoms, equation (1) can be expressed as follows:

Cu₂Se \rightarrow 2%Cu₂Se (in matrix) + y%CuGa(Te/Se)₂ (T- and C-phase) + z%Cu₂Te (1) where $x + 2y + z \neq 1$.

The dissolution of Cu_2Se in the PbTe matrix produces Te vacancies. The $CuGa(Te/Se)_2$ (T-and C-phase) phases can introduce interstitial Cu into the matrix simultaneously. The vacancy of interstitial Cu and Te increase n and σ . In contrast, the dynamic phase transition, point defects, nano- and micro-sized Cu_2Te , and $CuGa(Te/Se)_2$ are beneficial for phonon scattering. The details are discussed later in this study.

Microstructural analysis and nanostructuring. S/TEM was used on the

Pb_{0.975}Ga_{0.025}Te-3%Cu₂Se sample to investigate the all-scale microstructure and its relationship to the change in transport properties. Figure 2a shows a typical high-angle annular dark field image of the specimen, which shows a significant amount of nanosized (10–200 nm) and micron-sized precipitates embedded in the matrix with bright contrast/high Z-contrast. According to the corresponding EDS mapping (Figure 2c–g), the second phases

are rich in Cu, Ga, and Se. In contrast, they were deficient in Pb, as indicated using yellow arrows in Figure 2a, which were assigned to CuGa(Te/Se)₂. We also discovered several precipitates that were high in Cu and Te but low in Pb and Ga, as indicated using the green arrows, which was assigned to Cu₂Te. The quantitative EDS analysis results of three typical regions (PbTe matrix, CuGa(Te/Se)₂, and Cu₂Te) are shown in Figure S7 (Supporting Information). It is worth noting that the Cu₂Te phases can only be found at grain boundaries; due to their small volume fraction in the sample, PXRD cannot detect signals from them. Figure 2b shows a bright-field TEM image of the specimen with mostly diffraction contrast, revealing a large number of dislocations around the nanoprecipitates, adding more scattering centers to the system.

A high-resolution TEM (HREM) image of a faceted nano precipitate embedded in a PbTe matrix is shown in Figure 2h. Along the [110] zone axis, the corresponding selected area electron diffraction (SAED) pattern from the matrix part belongs to the rock-salt PbTe phase $(Fm\bar{3}m)$. Because of the coherency and lattice misfit between the two phases, translational Moiré franges exist in the precipitate. Figure 2i depicts the corresponding SAED pattern. The bright spots in the PbTe matrix are highlighted in red along the [110] zone axis. Satellite spots were discovered around every PbTe spot, with the brightest spots indexed to the tetragonal CuGaTe₂ phase $(I\bar{4}2d)$ along the [110] zone axis. In this direction, the atomic planes of the two phases were perfectly aligned, with no rotation or twisting observed. The weaker spots from the satellites originate from the double diffraction between the two phases.

Charge Transport Properties

Hall Coefficient. According to the Hall measurements, the charge n gradually increases up to 3% with the addition of Cu_2Se , then spikes before showing a downward trend (Figure 3a). As in n-type doping, the formation of the second CuGa (Te/Se) $_2$ (T-phase) phase results in Te

vacancies in the PbTe matrix. The samples exhibited a degenerate semiconducting nature with a rapidly increasing n as the temperature increased (Figure 3b), particularly for the x = 5 samples. At T > 600 K, the n was more than twice that of the I-doped PbTe-x%Cu₂Te samples. The multiple inflection points in the temperature-dependent curves indicate competing doping mechanisms among the elemental components, namely Te, Cu, and Ga.

Electrical Conductivity. The temperature-dependent electrical conductivities of $Pb_{0.975}Ga_{0.025}Te$ -x% Cu_2Se (x = 0, 1, 2, 3, 4, and 5) are shown in Figure 4. All samples exhibited typical degenerate semiconductor behavior, with gradually decreasing σ as temperature increased. Over the entire T range, the Cu₂Se alloyed samples outperformed the Pb_{0.975}Ga_{0.025}Te samples in terms of σ . At room temperature, for example, the σ increased significantly from $\sim 1870 \text{ Scm}^{-1}$ for $Pb_{0.975}Ga_{0.025}Te$ to $\sim 3092 \text{ Scm}^{-1}$ for the x=1 sample. This represents $\sim 65\%$ increase over the Cu₂Se-free sample. The improved *n* values from $\sim 1.18 \times 10^{19}$ (for x = 0) to $\sim 1.27 \times 10^{19}$ cm⁻³ (for x = 1) and the enhanced mobilities $\mu_{\rm H}$ from ~990 (for x = 0) to ~1521 cm²V⁻¹s⁻¹ (for x = 1) are consistent with the enhanced σ , as shown in Figures 4c and 4d. Although Cu₂Se is a p-type semiconductor, ^[27] n increases significantly. This is attributed to the formation of Te vacancies and interstitial Cu. However, as the Cu₂Se fraction increases, the σ decreases owing to a decrease in μ_H from ~1521 (for x = 1) to ~999 cm²V⁻¹s⁻¹ (for x = 5). Furthermore, the x = 5 sample exhibited the lowest σ of ~2466 S cm⁻¹. Remarkably, there is a "hump" in the σ -T plots for the Cu₂Se alloyed samples at T > 623 K compared to Pb_{0.975}Ga_{0.025}Te. According to equation (1), increasing Cu₂Te reduces the total content of Cu₂Se in the matrix and CuGa(Te/Se)₂ (T and C phases). Consequently, the value of n decreases. Therefore, the abnormal signature of the "hump" for n and σ may result from competition between the previously mentioned decreased and increased n caused using Ga⁺ ionization (Figure S10, Supporting Information). Moreover, as Cu_2Te disappears at T > 773

K, n increases. The I-doped PbTe-x%Cu₂Te samples, in contrast, lack the distinctive "hump" signature in n, σ , and S (refer to the following section). [18]

Seebeck Coefficients. The temperature-dependent S of Pb_{0.975}Ga_{0.025}Te-x%Cu₂Se (x = 0, 1, 2, 3, 4, and 5) samples is shown in Figure 4b. The negative S agreed with the negative $R_{\rm H}$ (Figure S9, Supporting Information), indicating n-type properties. Because of the increased n, the Cu₂Se alloyed samples exhibit a lower absolute S value (~82.2–98.5 μ V K⁻¹ at room temperature and 190.8–215.8 μ V K⁻¹ at 873 K) than the Pb_{0.975}Ga_{0.025}Te samples (130.4 μ V K⁻¹ at room temperature and 252.4 μ V K⁻¹ at 873 K). In particular, for Cu₂Se alloyed samples, the S gradually decreased with increasing Cu₂Se up to 3% and then increased. Moreover, the S exhibits a similar signature to the "hump" as σ at T > 623 K for Cu₂Se alloyed samples. The $S = \frac{8\pi^2 k_{\rm B}^2}{3eh^2} m^* T \left(\frac{\pi}{3n}\right)^{2/3}$, where $k_{\rm B}$, e, h, and m^* represent the Boltzmann constant, electron charge, Plank constant, and density of states (DOS) effective mass, respectively, is consistent with the variation of n caused by the new phase generation as mentioned above (Figure 3b). Furthermore, the σ and S were stable during thermal cycling with repeated heating—cooling measurements, exhibiting good repeatability (Figure S11, Supporting Information).

Figure 3e depicts Pisarenko plots for the Pb_{0.975}Ga_{0.025}Te-x%Cu₂Se (x = 0, 1, 2, 3, 4, and 5) samples at room temperature. The green and purple curves represent theoretical values calculated by the single parabolic band model with DOS effective masses (m^*) of 0.3 m_e and 0.2 m_e (m_e is the free electron mass), respectively. The m^* decreases from ~0.3 m_e for Pb_{0.975}Ga_{0.025}Te to less than 0.2 m_e for x = 2 samples, then returns to 0.3 m_e as the Cu₂Se fraction increases. The lower S resulted from the lighter m^* and increased n, but the samples exhibited higher a and mobility than Cu₂Se free samples (Figure 3).

Power Factor. The temperature-dependent PFs of $Pb_{0.975}Ga_{0.025}Te-x\%Cu_2Se$ (x = 0, 1, 2, 3, 4, and 5) samples are shown in Figure 4f. Because of the low S in the low-temperature range (300–523 K), the Cu_2Se alloyed samples exhibited lower PF than the parent material $Pb_{0.975}Ga_{0.025}Te$. However, at T > 523 K, they exhibit superior PFs due to their higher σ . $Pb_{0.975}Ga_{0.025}Te-3\%Cu_2Se$, for example, exhibits a high value of ~19.0 μ Wcm⁻¹K⁻², which is ~30% higher than the Cu_2Se free sample $Pb_{0.975}Ga_{0.025}Te$ (~14.7 μ Wcm⁻¹K⁻²) at 873 K.

Thermal Conductivity. The temperature-dependent total thermal conductivities (κ_{tot}) for the Pb_{0.975}Ga_{0.025}Te-x%Cu₂Se (x = 0, 1, 2, 3, 4, and 5) samples are higher than those of the Cu₂Se free sample, as shown in Figure 5, due to the significant contribution from the increased κ_{ele} at T < 773 K. For the x > 1 sample, there is a small flat region in the κ_{tot} -T plot from 573 to 723 K. With increasing temperature, Pb_{0.975}Ga_{0.025}Te-4%Cu₂Se achieves the lowest κ_{tot} of ~1 W m $^{-1}$ K $^{-1}$ at 873 K.

To clarify the contributions of $\kappa_{\rm ele}$ and $\kappa_{\rm lat}$ in Pb_{0.975}Ga_{0.025}Te-x%Cu₂Se, the $\kappa_{\rm ele}$ is evaluated using the Wiedemann–Franz law, $\kappa_{\rm ele} = L\sigma T$, where L represents the Lorenz number calculated using the relationship: $L = (1.5 + \exp[-|S|/116]) \times 10^{-8} \text{ V}^2\text{K}^{-2}$. The L and $\kappa_{\rm ele}$ of Pb_{0.975}Ga_{0.025}Te-x%Cu₂Se are shown in Figures S11 (Supporting Information). The Pb_{0.975}Ga_{0.025}Te-x%Cu₂Se samples displayed a much higher $\kappa_{\rm ele}$ than that of Pb_{0.975}Ga_{0.025}Te over the entire measured temperature range, indicating that Cu₂Se alloying significantly enhanced σ . Furthermore, $\kappa_{\rm ele}$ exhibits the "hump" at T > 623 K.

The κ_{lat} of Pb_{0.975}Ga_{0.025}Te-x%Cu₂Se was calculated using the formula $\kappa_{lat} = \kappa_{tot} - \kappa_{ele}$, and is shown in Figure 5b. Over the entire T range, the κ_{lat} values of the Cu₂Se alloyed samples were significantly lower than those of Pb_{0.975}Ga_{0.025}Te. At room temperature, κ_{lat} for the x = 1 sample was drastically reduced from ~1.70 Wm⁻¹K⁻¹ for x = 0 to ~1.15 Wm⁻¹K⁻¹ (~32% reduction). The κ_{lat} of x > 2 samples exhibited anomalous behavior at high

temperatures, reaching an ultralow value of $\sim 0.05~\rm Wm^{-1}K^{-1}$ at 773 K. This value is significantly lower than the theoretical minimum $\kappa_{\rm lat}$ value of $\sim 0.36~\rm Wm^{-1}K^{-1}$, as determined using the Debye–Callaway model. As shown previously, the origin of the significantly reduced $\kappa_{\rm lat}$ for the Cu₂Se alloyed sample was systematically investigated using in situ PXRD and TEM. This abnormal $\kappa_{\rm lat}$ was caused by weak electron–phonon coupling via dynamic phase generation, point defects, dislocations, nano- and micro-sized Cu₂Te and CuGa(Te/Se)₂, and an overestimation of electronic component thermal conductivity governed by the Wiedemann–Franz law at temperatures above 673 K. Although the $\kappa_{\rm lat}$ value is difficult to evaluate precisely owing to the dynamic phase generation of Cu₂Te and CuGa(Te/Se)₂ (C-phase), $\kappa_{\rm tot}$ is veracious without that use of the Wiedemann–Franz law.

Figure of Merit. The ZT value increased with temperature, peaked at ~723 K, and then dropped off for the Cu₂Se-free sample, as shown in Figure 6. However, for the Cu₂Se alloyed samples, the ZT values maintained a rising trend up to 823 K and obtained higher values. Pb_{0.975}Ga_{0.025}Te-3%Cu₂Se sample has a peak ZT of ~1.63 due to increased PF and decreased κ_{tot} in the high-temperature range (T > 723 K). Cu₂Se alloy samples exhibit a higher ZT than Pb_{0.975}Ga_{0.025}Te and other high-performance n-type PbTe thermoelectrical materials. [30] Furthermore, an average ZT ($ZT_{\text{avg}} = \frac{1}{T_{\text{H}} - T_{\text{C}}} \int_{T_{\text{C}}}^{T_{\text{H}}} ZT \, dT$) of ~1.05 from $T_{\text{C}} = 400$ K to $T_{\text{H}} = 823$ K was obtained for Pb_{0.975}Ga_{0.025}Te-3%Cu₂Se, which is one of the best ZT_{avg} values for n-type PbTe. [11, 31/132]

Density functional theory calculations. We investigated the electronic band structure of CuGaTe₂ in PbTe to determine the dopant effect of Cu₂Se in PbTe and Ga-doped PbTe, as the above experiments revealed the relevant precipitates of Cu₂Te and CuGaTe₂. In addition to the strong alternating VBs, the CBs of Ga-doped and Cu₂Te alloyed PbTe differ significantly from those of Ga-doped PbTe, as shown in Figure 7. Owing to the tervalent Ga

substitution of Pb, the Ga 4s state contributes to a gap state across the Fermi level in Ga-doped PbTe. The Te 5p and Ga 4p states, in contrast, contributed to the DOS for the first CB minimum, as shown in Figure 7b. A critical effect for Ga-doped and Cu_2Se alloyed PbTe is an increase in density of state effective mass to 0.40 m_e , which is greater than the corresponding n-type effective mass of PbTe, which is $\sim 0.26 m_e$.

To investigate the effects of Ga doping and Cu₂Se alloying on the κ_{lat} of PbTe, we calculated phonon dispersion for pure PbTe, Ga-doped PbTe, and Ga-doped and Cu₂Se alloyed PbJe. The Cu, Ga, and Te atoms preferred to form a substitution Cu–Ga–Te complex (Figure S13a, Supporting Information) in the Ga-doped and Cu₂Se alloyed PbTe, where the Cu and Ga atoms were the nearest neighbors. We calculated the decrease in κ_{lat} after adding the Cu-Ga complex based on the phonon dispersion shown in Figure 8. κ_{lat} of Ga-doped and Cu₂Se alloyed in PbTe exhibited a significant decrease, similar to the previous study involving Ga-doped and Zn-alloyed PbTe. 26 The Cu-Ga complex caused ~35% reduction in $\kappa_{\rm lat}$ at 300 K, as shown in Figure S13b (Supporting Information). This significant decrease could be attributed to several factors. Beyond the mass fluctuation of the Cu and Ga introduction, the calculated average phonon velocity LA branch of the Cu-Ga complex case (~2750 ms⁻¹) was much lower than that of pure PbTe (3160 ms⁻¹). Moreover, the complex induced a relatively large local atomic strain, resulting in longer Pb–Te bond lengths than in the regular cases; therefore, relatively lower Pb vibration frequencies were induced. The low-frequency peaks of the Pb and Te projected phonon DOS (PDOS) are characterized by a left shift to a lower frequency, as shown in Figure 8b. Furthermore, in the case of the Cu–Ga complex, the Ga atom generates low-frequency optical vibration modes, as shown at the G point with a frequency of ~12 cm⁻¹. This model exhibited a small PDOS peak in the extremely low-frequency range (Figure 8b). This is attributed to the off-centered Cu-Ga

atom, which results in a relatively loose atomic environment and, therefore, a significant κ_{lat} reduction.

CONCLUSIONS

In summary, Cu₂Se can enhance the electrical transport properties of PbTe while also lowering the κ_{lat} yia dynamic phase generation. By introducing Te vacancies or interstitial Cu at room temperature, the Cu₂Se in the PbTe matrix and the generated CuGa(Te/Se)₂ (T-phase) increased the n and σ . With increasing temperature, the sample exhibited a dynamic change in Cu₂Te content and the formation of new phases of CuGa(Te/Se)₂ (C-phase). Point defects, dislocations, Cu₂Te and CuGa(Te/Se)₂ (T-phase) nanocrystals, dynamic phase generation of Cu₂Te and CuGa(Te/Se)₂ (C-phase) intense phonon scattering and extremely low $\kappa_{lat} \sim 0.05 \text{ Wm}^{-1}\text{K}^{-1}$. PbTe is promising for intermediate-temperature device applications due to its maximum ZT of ~ 1.63 at 823 K, which can be obtained for Pb_{0.975}Ga_{0.025}Te-3%Cu₂Se. A dynamic phase conversion is a novel approach that can be applied to various thermoelectric systems.

Conflicts of interest

The authors declare no conflict of interest.

ACKNOWLEDGMENTS

This work was supported in part by the National Key Research and Development Program of China (2020YFA0710303). At Northwestern work was supported in part by the Department of Energy, Office of Science, Basic Energy Sciences under grant DE-SC0014520, (sample preparation, synthesis, XRD, TE measurements, TEM measurements, DFT calculations). This study was supported in part by the National Natural Science Foundation of China (52102218, U1905215, and 52072076) and the Fujian Science & Technology Innovation Laboratory for Optoelectronic Information of China (2021ZZ127). The authors acknowledge the Minjiang Scholar Professorship (GXRC-21004), the EPIC facility of Northwestern University's NUANCE Center, which received support from the Soft and Hybrid Nanotechnology Experimental (SHyNE) Resource (NSF ECCS-1542205), the MRSEC program (NSF DMR-1720139) at the Materials Research Center, International Institute for Nanotechnology (IIN), Keck Foundation, State of Illinois, through IIN, and the Office of Science of the U.S. Department of Energy under Contract No. DE-AC02-06CH11357 and DE-AC02-05CH 1231. The authors also acknowledge the access to facilities for high-performance computational resources at Northwestern University and Singapore MOE AcRF Tier 1 RG128/21, Singapore A*STAR project A19D9a0096.

REFERENCES

[1] a) K. F. Hsu, S. Loo, F. Guo, W. Chen, J. S. Dyck, C. Uher, T. Hogan, E. K. Polychroniadis, M. G. Kanatzidis, *Science* 2004, 303, 818; b) G. Tan, L.-D. Zhao, M. G. Kanatzidis, *Chem. Rev.* 2016, 116, 12123; c) T. Zhu, Y. Liu, C. Fu, J. P. Heremans, J. G. Snyder, X. Zhao, *Adv. Mater.* 2017, 29, 1605884; d) J. He, T. M. Tritt, *Science* 2017, 357, 1369; e) A. J. Minnich, M. S. Dresselhaus, Z. Ren, G. Chen, *Energy Environ. Sci.* 2009, 2, 466; f) G. J. Snyder, E. S. Toberer, *Nat. Mater.* 2008, 7, 105; g) Z. Z. Luo, S. Hao, S. Cai, T. P. Bailey, G. Tan, Y. Luo, I. Spanopoulos, C. Uher, C. Wolverton, V. P. Dravid, Q. Yan, M. G. Kanatzidis, *J. Am. Chem. Soc.* 2019,

- 141, 6403; h) Z.-Z. Luo, S. Hao, X. Zhang, X. Hua, S. Cai, G. Tan, T. P. Bailey, R. Ma, C. Uher, C. Wolverton, V. P. Dravid, Q. Yan, M. G. Kanatzidis, *Energy Environ.* Sct. 2018. 11, 3220.
- [2] H. S. Kim, Z. M. Gibbs, Y. Tang, H. Wang, G. J. Snyder, *APL Mater.* **2015**, *3*, 041506
- a) G. J. Vineis, A. Shakouri, A. Majumdar, M. G. Kanatzidis, *Adv. Mater.* 2010, 22, 3970; b) M. G. Kanatzidis, *Chem. Mater.* 2010, 22, 648; c) J. R. Sootsman, D. Y. Chung, M. G. Kanatzidis, *Angew. Chem. Int. Ed.* 2009, 48, 8616; d) Z.-Z. Luo, S. Cai, S. Hao, T. P. Bailey, X. Hu, R. Hanus, R. Ma, G. Tan, D. G. Chica, G. J. Snyder, C. Uher, C. Wolverton, V. P. Dravid, Q. Yan, M. G. Kanatzidis, *Chem. Mater.* 2019, 31, 5943.
- [4] a) G. Tan, F. Shi, S. Hao, L.-D. Zhao, H. Chi, X. Zhang, C. Uher, C. Wolverton, V. P. Dravid, M. G. Kanatzidis, *Nat. Commun.* 2016, 7, 12167; b) Y. Wu, Z. Chen, P. Nan, F. Xiong, S. Lin, X. Zhang, Y. Chen, L. Chen, B. Ge, Y. Pei, *Joule* 2019, 3, 1276.
- [5] Q. Yan, M. G. Kanatzidis, *Nat. Mater.* **2022**, *21*, 503.
- [6] J. Zhang, D. Wu, D. He, D. Feng, M. Yin, X. Qin, J. He, Adv. Mater. 2017, 29, 1703148.
- [7] G. Tan, M. Ohta, M. G. Kanatzidis, *Philos. Tr. Soc.* **2019**, *377*, 20180450.
- [8] Z. -Z. Luo, S. Cai, S. Hao, T. P. Bailey, X. Su, I. Spanopoulos, I. Hadar, G. Tan, Y. Luo, J. Xu, C. Uher, C. Wolverton, V. P. Dravid, Q. Yan, M. G. Kanatzidis, *J. Am. Chem. Soc.* **2019**, *141*, 16169.
- [9] Q. Zhang, Q. Song, X. Wang, J. Sun, Q. Zhu, K. Dahal, X. Lin, F. Cao, J. Zhou, S. Chen, G. Chen, J. Mao, Z. Ren, *Energy Environ. Sci.* **2018**, *11*, 933.
- [10] X. Su, S. Hao, T. P. Bailey, S. Wang, I. Hadar, G. Tan, T. B. Song, Q. Zhang, C. Uher, C. Wolverton, *Adv. Energy Mater.* **2018**, *21*, 1800659.
- [11] Y. Xiao, H. Wu, J. Cui, D. Wang, L. Fu, Y. Zhang, Y. Chen, J. He, S. J. Pennycook, L.-D. Zhao, *Energy Environ. Sci.* **2018**, *11*, 2486.

- [12] a) Y. Xiao, Y. Wu, P. Nan, H. Dong, Z. Chen, Z. Chen, H. Gu, B. Ge, W. Li, Y. Pei, Chem 2020, 6, 523; b) Z. Chen, Z. Jian, W. Li, Y. Chang, B. Ge, R. Hanus, J. Yang, Y. Chen, M. Huang, G. J. Snyder, Y. Pei, Adv. Mater. 2017, 29, 1606768.
- [13] a) J. M. Hodges, S. Hao, J. A. Grovogui, X. Zhang, T. P. Bailey, X. Li, Z. Gan, Y. Y. Hu, C. Uher, V. P. Dravid, C. Wolverton, M. G. Kanatzidis, J. Am. Chem. Soc. 2018, 140, 18115; b) Z.-Z. Luo, S. Cai, S. Hao, T. Bailey, Y. Luo, W. Luo, Y. Yu, C. Uher, C. M. Wolverton, V. Dravid, Z. Zou, Q. Yan, M. G. Kanatzidis, Energy Environ. Sci. 2022, 15, 368.
- [14] L. Fu, M. Yin, D. Wu, W. Li, D. Feng, L. Huang, J. He, *Energy Environ. Sci.* **2017**, *10*, 2030.
- [15] A. D. LaLonde, Y. Pei, G. J. Snyder, Energy Environ. Sci. 2011, 4, 2090.
- [16] G. Tan, C. C. Stoumpos, S. Wang, T. P. Bailey, L.-D. Zhao, C. Uher, M. G. Kanatzidis, *Adv. Energy Mater.* 2017, 7, 1700099.
- [17] Y. Pei, Z. M. Gibbs, A. Gloskovskii, B. Balke, W. G. Zeier, G. J. Snyder, *Adv. Energy Mater.* **2014**, *4*, 1400486.
- [18] Y. Xiao, H. Wu, W. Li, M. Yin, Y. Pei, Y. Zhang, L. Fu, Y. Chen, S. J. Pennycook, L. Huang, J. He, L.-D. Zhao, *J. Am. Chem. Soc.* **2017**, *139*, 51.
- [19] W. Li, L. Zheng, B. Ge, S. Lin, X. Zhang, Z. Chen, Y. Chang, Y. Pei, Adv. Mater. 2017, 29, 1605887.
- [20] a) Y. Qin, Y. Xiao, L.-D. Zhao, APL Mater. 2020, 8, 010901; b) C. Zhou, Y. Yu, Y. K. Lee, C. Cojocaru-Mirédin, B. Yoo, S.-P. Cho, J. Im, M. Wuttig, T. Hyeon, I. Chung, J. Am. Chem. Soc. 2018, 140, 15535.
- [21] R. Blachnik, R. Igel, Z. Naturforsch., B: J. Chem. Sci. 1974, 29, 625.
- [22] J. P. Perdew, K. Burke, M. Ernzerhof, Phys. Rev. Lett. 1996, 77, 3865.
- [23] G. Kresse, J. Furthmüller, *Phys. Rev. B* **1996**, *54*, 11169.
- [24] L. Chaput, A. Togo, I. Tanaka, G. Hug, Phys. Rev. B 2011, 84, 094302.

- [25] G. Kresse, D. Joubert, Phys. Rev. B 1999, 59, 1758.
- [26] a) M. Asen-Palmer, K. Bartkowski, E. Gmelin, M. Cardona, A. P. Zhernov, A. V. Inyushkin, A. Taldenkov, V. I. Ozhogin, K. M. Itoh, E. E. Haller, *Phys. Rev. B* 1997, 56, 9431, b) Y. Zhang, E. Skoug, J. Cain, V. Ozoliņš, D. Morelli, C. Wolverton, *Phys. Rev. B* 2012, 85, 054306; c) D. T. Morelli, J. P. Heremans, G. A. Slack, *Phys. Rev. B* 2002, 66, 195304.
- [27] H. Liu, X. Shi, F. Xu, L. Zhang, W. Zhang, L. Chen, Q. Li, C. Uher, T. Day, G. J. Snyder, *Nat. Mater.* 2012, 11, 422.
- [28] Y. K. Koh, C. J. Vineis, S. D. Calawa, M. P. Walsh, D. G. Cahill, *Appl. Phys. Lett.* 2009, 94, 153101.
- [29] X. Qian, D. Wang, Y. Zhang, H. Wu, S. J. Pennycook, L. Zheng, P. F. P. Poudeu, L.-D. Zhao, J. Mater. Chem. A 2020, 8, 5699.
- [30] a) Y. Li, J. Zhang, S. Pan, Y. Jiang, K. Wang, J. Yang, Y. Pei, Q. Zhu, M. T. Agne, G. J. Snyder, Z. Ren, W. Zhang, J. Luo, Energy Environ. Sci. 2019, 12, 3089; b) X. Yu, H. Wu, W. Li, M. Yin, Y. Pei, Y. Zhang, L. Fu, Y. Chen, S. J. Pennycook, L. Huang, J. He, L. D. Zhao, J. Am. Chem. Soc. 2017, 139, 18732; c) Y. Xiao, Y. Wu, P. Nan, H. Dong, Z. Chen, Z. Chen, H. Gu, B. Ge, W. Li, Y. Pei, Chem 2020, 6, 523; d) Y. Han, Z. Chen, C. Xin, Y. Pei, M. Zhou, R. Huang, L. Li, J. Alloys Compd. 2014, 600, 91; e) X. Wang, X. Li, Z. Zhang, X. Li, C. Chen, S. Li, X. Lin, J. Sui, X. Liu, F. Cao, J. Yang, Q. Zhang, Mater. Today Phys. 2018, 6, 45; f) G. Tan, C. C. Stoumpos, S. Wang, T. P. Bailey, L.-D. Zhao, C. Uher, M. G. Kanatzidis, Adv. Energy Mater. 2017, 7, 1700099; g) H. Liu, Z. Chen, J. Tang, Y. Zhong, X. Guo, F. Zhang, R. Ang, ACS Appl. Mater. Interfaces 2020, 12, 52952; h) H.-T. Liu, Q. Sun, Y. Zhong, C.-L. Xia, Y. Chen, X.-L. Shi, Z.-G. Chen, R. Ang, Mater. Today Phys. 2022, 24, 100677.
- [31] a) M. Dutta, R. K. Biswas, S. K. Pati, K. Biswas, ACS Energy Lett. 2021, 6, 1625; b)
 L. You, J. Zhang, S. Pan, Y. Jiang, K. Wang, J. Yang, Y. Pei, Q. Zhu, M. T. Agne, G.
 J. Snyder, Z. Ren, W. Zhang, J. Luo, Energy Environ. Sci. 2019, 12, 3089; c) L. Fu,
 M. Yin, D. Wu, W. Li, D. Feng, L. Huang, J. He, Energy Environ. Sci. 2017, 10, 2030.

- [32] a) P.-Y. Deng, K.-K. Wang, J.-Y. Du, H.-J. Wu, Adv. Funct. Mater. 2020, 30, 2005479; b) J. Zhang, D. Wu, D. He, D. Feng, M. Yin, X. Qin, J. He, Adv. Mater. 2017, 29, 1703148.
- [33] a) P. Jood, M. Ohta, M. Kunii, X. Hu, H. Nishiate, A. Yamamoto, M. G. Kanatzidis, J. Mater. Chem. C 2015, 3, 10401; b) S. Wang, Y. Xiao, Y. Chen, S. Peng, D. Wang, T. Hong, Z. Yang, Y. Sun, X. Gao, L.-D. Zhao, Energy Environ. Sci. 2021, 14, 451; c) Z.-Z. Luo, X. Zhang, X. Hua, G. Tan, T. P. Bailey, J. Xu, C. Uher, C. Wolverton, V. P. Dravid, Q. Yan, M. G. Kanatzidis, Adv. Funct. Mater. 2018, 28, 1801617.

Figures

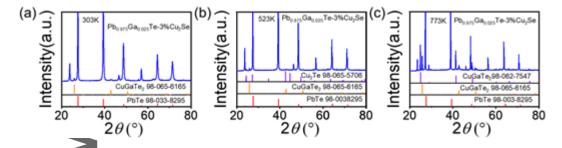


Figure 1. In-situ PXRD data of Pb_{0.975}Ga_{0.025}Te-3%Cu₂Se sample with the at 303 K (a), 523 K (b) and 773 K (c). The CuGaTe₂ 98-065-6165 belongs to a tetragonal phase with space group $I\overline{4}2d$, a = 6.0235(1) Å, c = 11.9398(0) Å, V = 433.20(1) Å³. The Cu₂Te 98-065-5706 crystallizes in the space group P6/mmm with a = 4.237 Å, c = 7.274 Å, V = 113.09 Å³. The CuGaTe₂ 98-062-7547 is a cubic phase with the space group $F\overline{4}3m$, a = 6.1639 Å, V = 234.18 Å³.



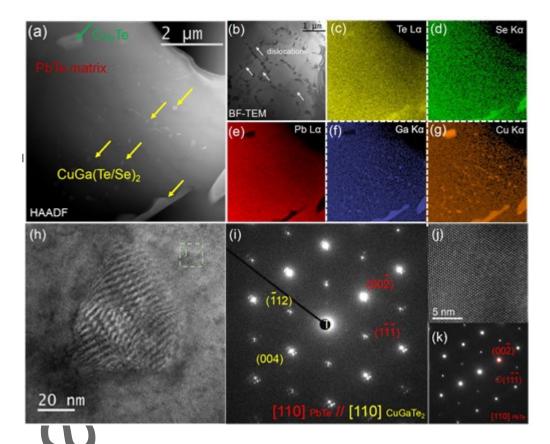


Figure 2. S/TEM analyses of Pb_{0.975}Ga_{0.025}Te-3%Cu₂Se before the performance characteristics (a) High angle annular dark-field image and (c)–(g) it is corresponding EDS mappings. (b) Bright-field TEM image of a partial area of (a). (h) High-resolution TEM (HREM) image of a typical coherent nano precipitate CuGaTe₂ sitting in the PbTe matrix. (i) Selected area electron diffraction (SAED) pattern of (h). (j) zoom-in version of the PbTe matrix part in (h). (k) SAED pattern in the (j) area.

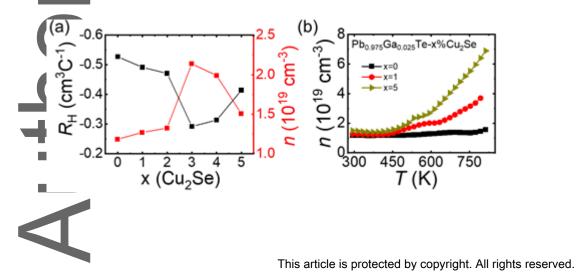


Figure 3. (a) Hall coefficient, $R_{\rm H}$ and Hall carrier concentration n for ${\rm Pb_{0.975}Ga_{0.025}Te-x\%Cu_2Se}$ (x = 0, 1, 2, 3, 4, and 5) at room temperature and (b) The temperature dependence of n for ${\rm Pb_{0.975}Ga_{0.025}Te-x\%Cu_2Se}$ (x = 0, 1, and 5).

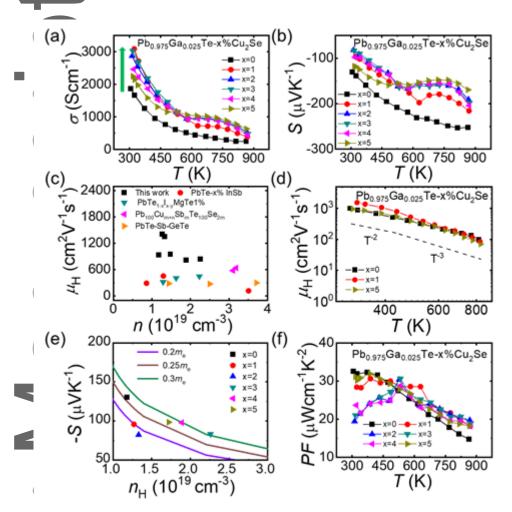


Figure 4. Electronic properties as a function of temperature for Pb_{0.975}Ga_{0.025}Te-x%Cu₂Se (x = 0, 1, 2, 3, 4, and 5), (a) Electrical conductivity, σ ; (b) Seebeck coefficient, S; (c) Comparison of carrier mobility, $\mu_{\rm H}$, in this work with the known high performance n-type PbTe based samples; [6, 33] (d) Temperature-dependent $\mu_{\rm H}$ for samples of Pb_{0.975}Ga_{0.025}Te-x%Cu₂Se (x = 0, 1, and 5); (e) Seebeck coefficient as a function of n at room temperature. The solid curves in (e) are the theoretical Pisarenko curves for n-type PbTe with an effective mass of electrons of 0.2 $m_{\rm e}$ (purple) and 0.3 $m_{\rm e}$ (green); and (f) Power Factor, PF.



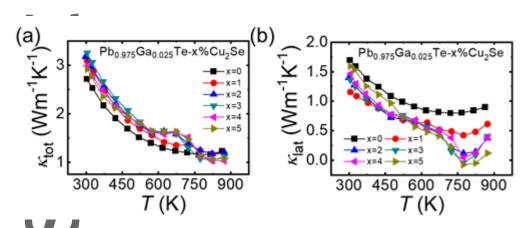


Figure 5. Thermal properties as a function of temperature for $Pb_{0.975}Ga_{0.025}Te-x\%Cu_2Se$ (x = 0, 1, 2, 3, 4, and 5), (a) Total thermal conductivity, κ_{tot} and (b) Lattice thermal conductivity, κ_{lat} .

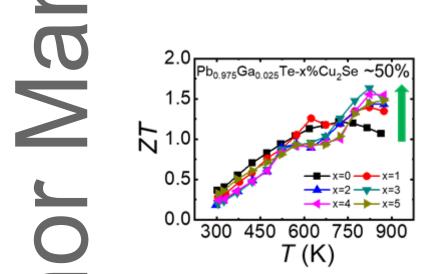


Figure 6. Figure of merit (ZT) values for $Pb_{0.975}Ga_{0.025}Te-x\%Cu_2Se$ (x = 0, 1, 2, 3, 4, and 5).

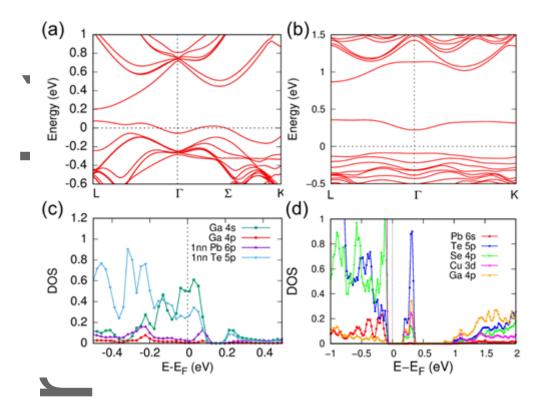


Figure 7. Electronic energy band structure of (a) Ga-doped PbTe and (c) Ga-doped and Cu₂Se-alloyed PbTe with the Ga off-centered model and corresponding projected density of states (b) and (d), respectively.

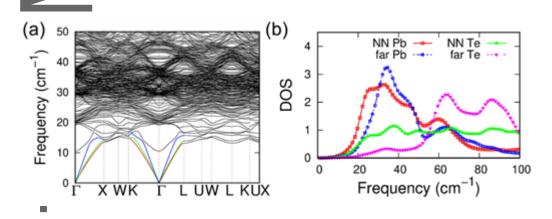
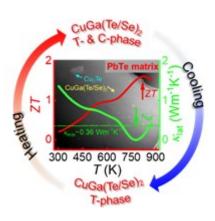


Figure 8. (a) Phonon dispersion curves of Ga-doped and Cu₂Se alloyed PbTe; (b) the projected phonon density of states (PDOS) of Ga-doped and Cu₂Se alloyed PbTe. "NN Pb"

and "NN Te" stand for the Pb and Te atoms that are next nearest neighbors to the Cu–Ga–Se dopants and have longer Pb–Te bond lengths within the supercell, whereas "far Pb" and "far Te" are the regular cases.

TOC

INUSC



A novel dynamic phase conversion was observed after Cu_2Se alloying in Ga-doped n-type PbTe. Cu_2Se alloying results in the Cu_2Te and tetragonal $CuGa(Te/Se)_2$ precipitates, leading to enhanced carrier concentration. With increasing temperature, the sample exhibits the dynamic change in Cu_2Te content and the generation of cubic $CuGa(Te/Se)_2$, obtaining an ultralow κ_{lat} and a high ZT of ~1.63.

