# Antagonism by Chloramphenicol of Carbon Tetrachloride Hepatotoxicity

Examination of Microsomal Cytochrome P-450 and Lipid Peroxidation 1

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Administration of chloramphenicol early in CCl, intoxication prevents lipid peroxidation of endoplasmic reticulum membranes. Conversely, a sulfamoyl analog, Tevenel, was ineffective in preventing the lipoperoxidative process. Likewise, in an *in vitro* microsomal system chloramphenicol inhibited a lipid peroxidation process and Tevenel did not. However, both compounds bind to cytochrome P-450. Chloramphenicol did not maintain cytochrome P-450 levels after CCl<sub>4</sub> administration nor did it depress cytochrome P-450 levels in untreated animals. The data obtained indicate that chloramphenicol may prevent lipid peroxidation either by inhibiting CCl<sub>4</sub> metabolism or by acting as a free radical sequestering agent.

### INTRODUCTION

It is now generally accepted that metabolism of CCl<sub>4</sub> is essential in the pathogenesis of carbon tetrachloride-induced liver injury. Since the major stable products of CCl<sub>4</sub> metabolism, CO<sub>2</sub> and CHCl<sub>3</sub>, are relatively nontoxic, and since the CCl<sub>4</sub> molecule itself must be cleaved before its toxicological potentiality can be expressed, it seems clear that events intimately associated with the cleavage of the CCl<sub>3</sub>-Cl bond must be responsible for initiating the pathological consequences. Metabolism of CCl<sub>4</sub> by the MFO<sup>2</sup> system in the hepatic endoplasmic reticulum is postulated to result in homolytic cleavage of the carbon-chlorine bond, producing CCl<sub>3</sub> · and Cl · free radicals. The disruption of cellular function is attributed to the reaction of CCl<sub>3</sub> · with either (1) membrane lipids promoting an autocatalytic peroxidative process and destroying the membrane (Rccknagel, 1967; Glende, 1972b) or (2) neighboring proteins and other macromolecules altering their cellular function (Diáz Goméz *et al.*, 1973; Castro *et al.*, 1975). However, studies of anaerobic metabolism of CCl<sub>4</sub> *in vitro* indicate that peroxidation of microsomal lipids appears to be obligatory for loss of microsomal enzyme

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<sup>2</sup> Abbreviations used: D-CAPS, D-threo-chloramphenicol, succinyl sodium salt; D-CAP, Doptical isomer of threo-chloramphenicol; L-CAP, L-isomer of threo-chloramphenicol; ip, intraperitoneally; DMSO, dimethyl sulfoxide; EDTA, ethylenediaminetetraacetate; ER, endoplasmic reticulum; MFO, mixed function oxidase.

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Insight into the mechanism of action of  $CCl_4$  has been facilitated through use of agents which prevent development of the necrosis associated with  $CCl_4$  administration. Recently chloramphenicol (D-CAP), a broad-spectrum antibiotic, has been shown to function as an antagonist of hepatotoxicity, particularly the mitochondrial damage, induced by  $CCl_4$  (Huyen and Banaschak, 1973; Brabec and Bernstein, 1975; Brabec *et al.*, 1976). The protective action of *D*-CAP does not appear to be associated with its antibiotic activity, nor with its inhibition of mitochondrial protein synthesis. Tevenel, a structural analog, has similar antibiotic properties but does not protect from  $CCl_4$  toxicity. Conversely, an optical isomer of chloramphenicol (L-CAP) which is inactive as an antibiotic and as an inhibitor of mitochondrial protein synthesis does protect mitochondria from  $CCl_4$ induced mitochondrial damage (Brabec and Bernstein, 1975; Brabec *et al.*, 1976). In addition, interaction of chloramphenicol with the microsomal MFO system has been reported (Grogan *et al.*, 1972; Huyen *et al.*, 1974; Dixon and Fouts, 1962).

In light of the action of D-CAP on CCl<sub>4</sub>-induced liver injury, a study was conducted to determine whether D-CAP's protective effect was exerted at the site of ER injury during early stages of CCl<sub>4</sub> intoxication and, if so, what parameters of ER damage were altered. A further aim was to reveal those processes, now marked by their sensitivity to interruption by D-CAP, that lead to mitochondrial damage.

## METHODS

Adult (250 to 350-g) male CFN strain rats were maintained on a commercial diet of rat chow and water *ad libitum*. Carbon tetrachloride was administered ip without a carrier at selected doses. The succinyl derivative of chloramphenicol, dissolved in water, was administered ip at 100 mg/kg body weight, while D-CAP and Tevenel were dissolved in propylene glycol or DMSO. Unless otherwise noted, CCl<sub>4</sub> was administered followed immediately by D-CAP, D-CAPS, or Tevenel.

Microsomes were prepared by differential centrifugation of a 10% liver homogenate in an isolation medium containing 0.25 M sucrose, 10 mM Tris, 10 mM KCl, and 1 mM EDTA (pH 7.0). Mitochondria, nuclei, and cell debris were removed by centrifugation at  $8800g_{av} \times 10$  min. The supernatant fraction was then centrifuged at  $103,000g_{av} \times 60$  min in a Spinco Model L ultracentrifuge.

Microsomal cytochrome P-450 content was determined by the method of Omura and Sato (1967). Animals were sacrificed by decapitation 15 and 20 hr after administration of CCl<sub>4</sub> (1.2 ml/kg) and D-CAPS. Microsomal protein was diluted to a concentration of 1 mg/ml in a buffer containing 0.15 M Tris, and 25 mM KCl, pH 7.5. Activity was measured using the split-beam mode of an Aminco DW-2 UV-VIS spectrophotometer.

Chloramphenicol interaction with hepatic microsomal cytochrome P-450 was analyzed by spectral binding studies. The ability of the drug to displace carbon monoxide from reduced P-450 was ascertained by the method described by Schenkman *et al.* (1967). Liver microsomes were diluted in a 25 mM KCl and 0.15 M Tris buffer, pH 7.5, to a protein concentration of 2.5 mg/ml. Livers were perfused with isolation medium prior to fractionation.

Binding studies with hexobarbital were carried out by following the spectral change between 420 and 500 nm as described by Schenkman *et al.* (1973). Liver microsomes were suspended in 0.2 M sodium phosphate buffer, pH 7.5, to a protein concentration of 2.6 mg/ml. p-CAP or one of its analogs was added to both experimental (containing hexobarbital) and reference cuvettes in concentrations ranging from 0.25 to 5.0 mM. Microsomes used in the spectral binding experiments were prepared from rats pretreated with phenobarbital, 100 mg/kg ip, for 5 days and sacrificed 48 hr after the last injection.

The NADPH- and NADH-cytochrome c reductase activities were measured in the presence of p-CAP or p-CAPS by the method described by Williams and Kamin (1962). The rate of reduction of cytochrome c by NADH and NADPH was followed on a Beckman Gilford Model 2000 recording spectrophotometer.

Determination of lipid peroxidation *in vivo* was measured by conjugated diene formation as described by Recknagel and Goshal (1966). Animals were sacrificed 30 min after  $CCl_4$  (1.2 ml/kg) administration. Lipids were extracted from 3.5 g of liver. Spectral scans were made against a methanol blank over the range of 220–300 nm on a Beckman DB spectrophotometer. Difference spectra were obtained by subtraction of scans of nonperoxidized lipid extracts.

Lipid peroxidation *in vitro* was measured by following  $O_2$  consumption in an Fe-ADP system as described by Ernster and Nordenbrand (1967).  $O_2$  uptake was measured with a Clark electrode at 30°C. p-CAP and Tevenel, dissolved in DMSO, were tested over concentrations ranging from 5 to 20 mM. The drug was added after a period of thermoequilibration of buffer and microsomes (2.6 mg/ml of microsomal protein). The reaction was initiated by addition of NADPH.

Protein was analyzed by the Biuret method (Gornall et al., 1949).

Statistical analysis: Comparison of mean values was determined by one-way analysis of variance (Steele and Torrie, 1960). The significance of the difference between control and experimental means was performed by Dunnett's procedure (Dunnett, 1955). The significance of difference between treatments was determined by one degree of freedom contrasts (Steele and Torrie, 1960).

# RESULTS

Two major ER-associated events which are integral with the development of  $CCl_4$ -induced hepatotoxicity are loss of microsomal enzyme activity and the process of lipid peroxidation.

Using conjugated diene formation in lipids extracted from hepatic microsomes as an index of *in vivo* lipid peroxidation, we found a marked diene conjugation absorption after 30 min of poisoning by  $CCl_4$ . Diene absorption in lipids from p-CAP-CCl<sub>4</sub>-treated rats was reduced by 71%. Tevenel treatment did not alter the degree of  $CCl_4$ -induced lipid peroxidation (Fig. 1).

Since chloramphenicol's effectiveness as a mitochondrial protective agent has been shown to be a function of time after delivery of  $CCl_4$  (Brabec *et al.*, 1976) the ability of p-CAP to interrupt lipid peroxidation when administered 4.5 hr after a single dose of  $CCl_4$  was tested. As illustrated in Fig. 2, p-CAP did not prevent the formation of conjugated dienes in membrane lipids of the ER when delivered at this time.

To further evaluate the effects of p-CAP on lipoperoxidation, studies of the process were conducted *in vitro*. p-CAP inhibited peroxide formation as mea-

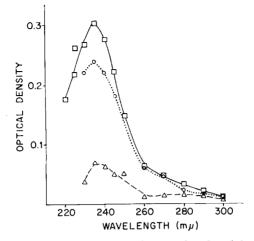


FIG. 1. Effect of chloramphenicol and Tevenel on CCl<sub>4</sub>-induced lipid peroxidation. Diene conjugation absorption of rat liver microsomal lipids. Rats were treated ip with CCl<sub>4</sub> (1.2 ml/kg), CCl<sub>4</sub> and p-CAPS (100 mg/kg), or CCl<sub>4</sub> and Tevenel (100 mg/kg), 30 min prior to sacrifice. Lipids from 3.5 g of rat liver were extracted in a chloroform:methanol mixture (2:1). The recovered lipids were diluted in methanol to a concentration of 1 mg/ml and wavelength scans were determined against a methanol blank over the range of 220–300 nm. Difference spectra were plotted: (\_\_\_\_) CCl<sub>4</sub>; (.....) Tevenel + CCl<sub>4</sub>; (----) p-CAPS + CCl<sub>4</sub>.

sured in an Fe-ADP-stimulated microsomal lipid peroxidation system (Fig. 3). In correlation with the experiments *in vivo*, Tevenel did not significantly affect the process of lipid peroxidation *in vitro*. p-CAP's effectiveness in inhibiting the peroxidative process *in vitro* displayed a requirement for an incubation period.

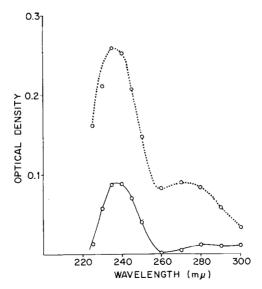


FIG. 2. Appearance of conjugated dienes in rat liver microsomal lipids following chloramphenicol treatment 5 hr after CCl<sub>4</sub> administration. Rats were treated ip with CCl<sub>4</sub> (1.2 ml/kg), followed 4.5 hr later by D-CAPS (100 mg/kg). Thirty minutes later the animals were sacrificed. Lipid extraction was performed as described in Fig. 1. (——) CCl<sub>4</sub>; (.....) D-CAPS and CCl<sub>4</sub>.

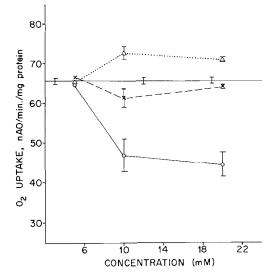


FIG. 3. Effect of chloramphenicol and Tevenel on *in vitro* lipid peroxidation as measured in an Fe-ADP-stimulated microsomal system. Liver microsomes were suspended in a 0.15 M Tris, 25 mM KCl buffer, pH 7.5, to a protein concentration of 2.6 mg/ml. The drug or carrier was added prior to initiating the peroxidative process. Results are expressed as means of the values  $\pm$  SE. DMSO values are equal to the volume in which the drug was suspended. The control is shown as a horizontal black line. (....) Tevenel; (----) DMSO; and (----) p-CAP. Comparison of the 10 means was performed by a one-way analysis of variance. p-CAP at concentrations of 10 and 20 mM was significantly different (P < 0.05) from the control using Dunnett's procedure.

p-CAP was ineffective as an inhibitor when administered after the peroxidative process had been initiated.

Since D-CAP decreases microsomal lipid peroxidation, it might likewise prevent the loss of cytochrome P-450 if these two events are causally related (Glende, 1972a). As illustrated in Table I, D-CAP when administered with CCl<sub>4</sub> did not significantly restore or prevent the loss of cytochrome P-450 at 15–20 hr after exposure to CCl<sub>4</sub>. The levels of hepatic microsomal cytochrome P-450 in D-CAPprotected animals were not significantly different from those in CCl<sub>4</sub>-poisoned animals when compared at corresponding periods of exposure.

The decrease in lipid peroxidation caused by p-CAP suggests an effect on the metabolism of  $CCl_4$ . This could be due to an interference with the electron flow from NADPH to cytochrome *P*-450. However, microsomal electron transport, as measured by the activity of NADPH-cytochrome *c* reductase, was not altered in the presence of p-CAP or p-CAPS (Table II).

The possible destruction of cytochrome P-450 by p-CAP within 5 hr of CCl<sub>4</sub> was investigated, as studies have shown that a decrease in the levels of cytochrome P-450 will reduce hepatotoxicity by CCl<sub>4</sub> (Dambrauskas and Cornish, 1970; Seawright and McLean, 1967; Smuckler and Hutten, 1966). It was found that chloramphenicol did not depress cytochrome P-450 levels in rats when administered up to 6 hr prior to sacrifice (control,  $1.275 \times 10^{-3} \ \mu M/mg$  of protein; 6 hr,  $1.062 \times 10^{-3} \ \mu M/mg$  of protein).

An interaction by D-CAP with cytochrome P-450 has been reported (Grogan et al., 1972; Huyen et al., 1974; Dixon and Fouts, 1962). To determine if this

$Treatment^a$		Cytochrome $P$ -450 ( $\mu M/mg$ of protein/ml)	
$\operatorname{CCl}_4{}^b$			
15 h	ır (3)	$1.39  imes 10^{-4} \pm 0.12  imes 10^{-4}$	
20 h	ır (6)	$1.29 imes 10^{-4}\pm 0.47 imes 10^{-4}$	
$CCl_4$ -	+ D-CAPS <sup>b</sup>		
15 h	ır (3)	$1.60 imes 10^{-4}\pm 0.19 imes 10^{-4}$	
20 h	ır (7)	$1.46  imes 10^{-4} \pm 0.56  imes 10^{-4}$	
Contro	$ol^{c}(8)$	$3.70 \times 10^{-4} \pm 1.50 \times 10^{-4}$	

 TABLE I

 Effect of Chloramphenicol and Carbon Tetrachloride on Cytochrome P-450

<sup>a</sup> Carbon tetrachloride (1.2 ml/kg) alone or with D-CAPS (100 mg/kg) was given ip to rats and the animals were sacrificed 15 or 20 hr later. A liver suspension (1.0 mg/ml) was diluted in a 0.15 MTris, 25 mM KCl, pH 7.5, buffer. The number of rats used is in parentheses. Results are expressed as means of the values  $\pm$ SE. A one-way analysis of variance was used to compare the five groups. Five one degree of freedom contrasts determined the significance of difference between treatments.

<sup>b</sup> Differences between cytochrome P-450 levels in the CCl<sub>4</sub>-treated and CCl<sub>4</sub> + p-CAPS-treated animals were not significant (P > 0.10). Interaction between drug and time was not significant (P > 0.10). Consequently, corresponding values at 15 and 20 hr were pooled.

<sup>c</sup> The cytochrome P-450 levels in both the CCl<sub>4</sub>- and CCl<sub>4</sub> + p-CAPS-treated animals were significantly different from the control (P < 0.001).

interaction could influence the activity of cytochrome P-450 and therefore CCl<sub>4</sub> metabolism, spectral binding studies were done. When p-CAP was incubated with oxidized liver microsomes, neither Types I, II, nor reverse Type I spectra were observed. However, reaction with the heme moiety of the cytochrome P-450 molecule resulting in formation of a ferrihemochrome is characteristic of a Type II compound (Schenkman *et al.*, 1967; Omura and Sato, 1964). As shown in Fig. 4, p-CAP and Tevenel both have the capacity to displace carbon monoxide from the reduced cytochrome P-450 hemochrome. Binding studies with hexobarbital revealed that both p-CAP and Tevenel were noncompetitive inhibitors of hexobarbital binding to the substrate site of cytochrome P-450 (Fig. 5). In this case, Tevenel was less effective than p-CAP.

## DISCUSSION

Activation of carbon tetrachloride by the MFO system with homolytic cleavage of the molecule into  $CCl_3$  and Cl species is accepted by most investigators to be the mechanism which triggers the development of hepatonecrosis. Evidence supports the view that reactive free radicals initiate a lipoperoxidative process on organelle membranes as well as bind directly to macromolecules. Which reaction is the more lethal is not known. However, *in vitro* studies have shown that  $CCl_4$ -induced loss of some microsomal enzyme activity depends on lipid peroxidation and not covalent binding of  $CCl_4$  cleavage products (Glende *et al.*, 1976).

In our studies, chloramphenicol inhibited  $CCl_4$ -induced lipid peroxidation in vivo and Fe-ADP-induced lipid peroxidation in vitro, while a structural analog, Tevenel, did not demonstrate any inhibition. Inhibition by p-CAP of lipid peroxidation may be explained by one of two mechanisms: (1) p-CAP

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<b>D-CAP</b> (mM)	$\begin{array}{c} \textbf{NADPH} \\ (\mu \text{moles of} \\ \text{cytochrome } c \\ \text{reduced/min}) \end{array}$	NADH ( $\mu$ moles of cytochrome <i>c</i> reduced/min)	
 	$1.39 imes10^{-3}$	$1.53 \times 10^{-3}$	
1	$1.15 imes10^{-3}$	$1.51 imes10^{-3}$	
7.5	_	$1.69 imes10^{-3}$	
10.0	$1.39 imes10^{-3}$	<u> </u>	

TABLE II Effect of Chloramphenicol on NADPH- and NADH-Cytochrome c Reductase Activity<sup>a</sup>

<sup>a</sup> Liver microsomes were suspended in a 0.15 *M* Tris, 25 m*M* KCl buffer, pH 7.5. The cuvettes contained 34  $\mu$ moles of cytochrome *c*, 1 m*M* cyanide, and enzyme. Aliquots of p-CAP ranging from 1 to 10 m*M* were added and the reaction was initiated by addition of 100  $\mu$ moles of either NADPH or NADH. Reactions were run in duplicate.

could directly interrupt the peroxidative process by acting as a sequestering agent for  $CCl_3$  and Cl molecules, or (2) D-CAP could alter lipid peroxidation by blocking the activation of  $CCl_4$ .

In support of (1), inhibition of electron transport by D-CAP in bacteria has been attributed to the formation of a benzoyl free radical which has the potential to function as an electron sink (Hansch *et al.*, 1973). A benzoyl free radical may be generated by the microsomal MFO system. The inability of Tevenel to prevent lipid peroxidation could be due to the inability of the sulfamoyl group to stabilize a benzoyl free radical.

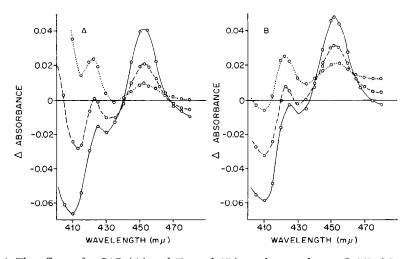


FIG. 4. The effects of D-CAP (A) and Tevenel (B) on the cytochrome P-450-CO complex. A microsomal suspension (2.5 mg/ml) was diluted in a 0.15 M Tris, 25 mM KCl buffer, pH 7.5, and divided into two cuvettes. A baseline was scanned. The contents of the experimental cuvette were gassed with CO for 1 min and then 0.17 mM NADPH was added to each cuvette (solid line). Aliquots of D-CAP or Tevenel in propylene glycol were added to the experimental cuvette and the difference spectra were recorded. (A) (-----) 6.25 mM D-CAP; (.....) 18.75 mM D-CAP. (B) (----) 6.6 mM Tevenel; (.....) 19.8 mM Tevenel.

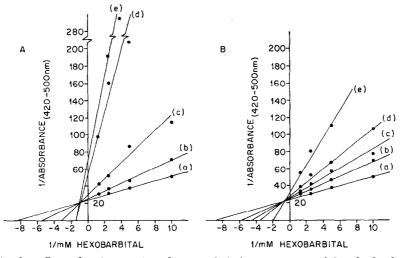


FIG. 5. The effect of D-CAP (A) and Tevenel (B) on titration of hexobarbital-induced Type I spectral changes. Liver microsomes were suspended in 0.2 M sodium phosphate buffer, pH 7.5, to a protein concentration of 2.6 mg/ml containing 2.4  $\mu M$  cytochrome P-450. D-CAP and Tevenel were suspended in propylene glycol. The difference spectra were recorded betwen 420 and 500 nm. Line (a), no additions; line (b), propylene glycol; line (c), 0.25 mM D-CAP or Tevenel; line (d), 1.25 mM D-CAP or Tevenel; line (e), 5.0 mM D-CAP or Tevenel.

D-CAP could decrease the rate of conversion of CCl<sub>4</sub> to CHCl<sub>3</sub>· through inhibition or destruction of the CCl4-metabolizing system. This process would decrease the number of  $CCl_3$  molecules generated, and therefore the subsequent lipoperoxidative process and covalent binding of CCl4 metabolites to cellular proteins. Our evidence suggests that D-CAP and its analogs bind to several sites on cytochrome P-450. Both D-CAP and Tevenel promote a reduction in the CO binding spectrum of cytochrome P-450, indicating a binding with the heme iron of cytochrome P-450. A Type I interaction can be detected in binding studies with hexobarbital by p-CAP and Tevenel. The ability of p-CAP to bind as a substrate is not unexpected, as Glazko (1966) has reported that D-CAP is metabolized by the liver. The noncompetitive inhibition exhibited by both p-CAP and Tevenel indicates interaction at a locus other than the substrate binding site. The difficulty one experiences in obtaining a conventional spectrum from incubating p-CAP and microsomes probably relates to the number of cytochrome P-450 sites on which p-CAP interacts. The significance of Tevenel interaction with cytochrome P-450 is paradoxical, as Tevenel is ineffective in ameliorating CCl<sub>4</sub> hepatotoxicity. The mechanism of protection depends on a reaction that should be evident with D-CAP and absent with Tevenel. However, in binding studies Tevenel was less effective than D-CAP.

The locus of CCl<sub>4</sub> activation is ill-defined. Some evidence argues that CCl<sub>4</sub> activation occurs at cytochrome *P*-450. However, Slater and Sawyer (1971a, b) propose that the reduction of CCl<sub>4</sub> occurs at the flavoprotein, cytochrome *P*-450 reductase. An alteration of the metabolism of CCl<sub>4</sub> by D-CAP could occur therefore through inhibition of this portion of the MFO system. Huyen *et al.* (1974) have reported a decline in reductase activity in the presence of D-CAP when

assayed by the formation of a reduced cytochrome P-450-CO complex. However, when tested with cytochrome c as an artificial electron acceptor, chloramphenicol did not alter reductase activity. The inhibition reported by Huyen *et al.* could be due to the binding of the heme iron by p-CAP. The possibility that p-CAP was decreasing CCl<sub>4</sub> metabolism by causing a decrease in cytochrome P-450 levels was directly tested and ruled out, as p-CAP did not significantly alter cytochrome P-450 levels in a time interval consistent with the protective period.

Despite the reduction of lipid peroxidation by p-CAP, chloramphenicol did not prevent the loss of cytochrome P-450 associated with CCl<sub>4</sub> intoxication as measured at 15 and 20 hr. Destruction of cytochrome P-450 did not coincide with the appearance of lipid peroxidation following administration of CCl<sub>4</sub>. For this reason, measurement of cytochrome P-450 was taken at a later stage of injury (15–20 hr) when cytochrome P-450 levels were substantially reduced. Ota *et al.* (1975) have suggested that CCl<sub>4</sub> irreversibly binds with cytochrome P-450, thus inhibiting MFO activity. p-CAP may lack the capacity to reverse this CCl<sub>4</sub>-cytochrome P-450 binding. Conversely, the loss of cytochrome P-450 during p-CAP inhibition of lipid peroxidation may simply reflect either the sensitivity of this enzyme to even the limited amount of lipid peroxidation that occurs in the presence of p-CAP or a delay in the degradative process. However, a delayed onset of lipid peroxidation is unlikely, as mitochondrial damage does not occur in the animal protected by p-CAP.

In light of p-CAP's amelioration of CCl<sub>4</sub>-induced mitochondrial damage, the diminished lipid peroxidation, and a reduction of mortality, the process responsible for necrotic development likely involves a peroxidative mechanism and mitochondrial damage. The above observations allow several routes for mitochondrial damage to be considered. (1) The process of lipid peroxidation could result in direct attack on mitochondrial membranes. (2) Mitochondrial damage may occur as a result of disruption of the ER, with a subsequent failure to synthesize mitochondrial components necessary for normal maintenance of homeostasis. Further delineation of these routes of mitochondrial damage are being attempted.

The results of this study suggest that chloramphenicol, a broad-spectrum antibiotic, prevents a lipoperoxidative process induced by  $CCl_4$  intoxication *in vivo* and inhibits an Fe-ADP-stimulated microsomal lipid peroxidation system *in vitro*. By determining the locus of the MFO system where D-CAP exerts its protective action, further insight into the etiology of CCl<sub>4</sub>-induced hepatonecrosis and other free radical-associated injury may be attained.

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