A Synthesis of (+)-7-Epiaustraline and (-)-7-Epialexine

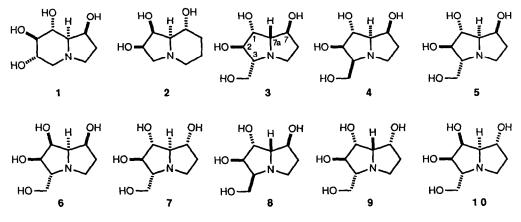
William H. Pearson*[†] and Jennifer V. Hines[‡]

Departments of Chemistry[†] and Medicinal Chemistry,[‡] The University of Michigan, Ann Arbor, MI 48109-1055

Key Words. (+)-7-Epiaustraline; (-)-7-epialexine; glycosidase inhibitors; polyhydroxylated pyrrolizidine alkaloids; epoxide

Abstract: Reductive cyclization of the azido epoxides 19α and 19β followed by deprotection afforded the HIV inhibitor (+)-7-epiaustraline 7 and (-)-7-epialexine 9. The formation of 7 proceeded with an unusual inversion of configuration at C-7.

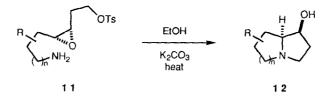
Polyhydroxylated indolizidine alkaloids such as castanospermine 1 and swainsonine 2 have attracted considerable interest in recent years due to their ability to inhibit glycosidases.¹ In addition to their use in the study of glycoprotein-processing enzymes, such alkaloids are promising anticancer and antiretroviral agents. More recently, polyhydroxylated pyrrolizidine alkaloids with similar biological activities have emerged from the independent investigations of research teams in the U.K. and the U.S.A. The isolation of alexine 3 from Alexa leiopetala was reported in 1988.² The isolation of similar alkaloids from Castanospermum australe quickly followed, including 3,7a-diepialexine (or 3-epiaustraline) 4,3,4 australine 5,5 1-epiaustraline (or 1,7adiepialexine) 6.6.7 and 7.7a-diepialexine (or 7-epiaustraline) 7.7 While alexine 3 and 3-epiaustraline 4 are generally poor inhibitors of glucosidases and galactosidases,^{2,4} they display amyloglucosidase inhibition which is on par with that of castanospermine,⁷ and alexine is an effective thioglucosidase inhibitor.⁸ Compounds 5-7 are also good amyloglucosidase inhibitors.^{5,7,9} Australine 5 inhibits glucosidase I, but not glucosidase II,⁹ and has recently been shown to exhibit antiviral activity.¹⁰ Modest glucosidase I, β -glucosidase, and α -mannosidase inhibition was observed for 1-epiaustraline 6.6 but it displayed good activity in a mouse gut digestive α glucosidase assay, as did 7-epiaustraline 7.7.11 An exciting recent report shows that australine 5, 1epiaustraline 6, and 7-epiaustraline 7 inhibit HIV-induced syncytia formation in JM cells.¹¹ The potential of the polyhydroxylated pyrrolizidine alkaloids as selective glycosidase inhibitors and as antiviral and antiretroviral agents makes them attractive targets for synthesis. In particular, the ability to prepare alternative stereoisomers of these alkaloids would be desirable, since the biological activity of these compounds varies substantially with their stereochemistry. We wish to report our initial efforts in this area, which have led to the synthesis of the naturally occurring alkaloid (+)-7-epiaustraline 7 (also known as 7,7a-diepialexine) which shows anti-HIV



activity,¹¹ and the non-natural compound (-)-7-epialexine 9.12 Our synthesis of 7 confirms its absolute configuration.⁷

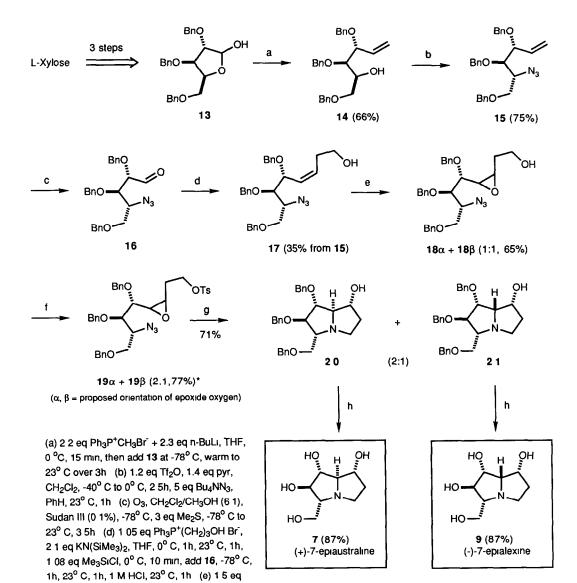
Synthetic efforts on the polyhydroxylated pyrrolizidine alkaloids have been sparse to date.¹³ In 1988, Fleet reported the synthesis of alexine **3** and the two non-natural alkaloids 3-epialexine **8** and 7-epialexine **9** from glucose.¹² Fleet has now completed the synthesis of **6** and the non-natural alkaloid 1,7,7a-triepialexine (or 1,7-diepiaustraline) **10**.¹⁴ Syntheses of related pyrrolizidine alkaloids without the 3-hydroxymethyl group have also been reported.^{15,16}

We have recently reported that compounds such as 11 undergo tandem epoxide opening/intramolecular alkylation to produce indolizidines 12 (n=2).¹⁷⁻¹⁹ The relative stereochemistry of the two newly formed stereocenters should be determined ultimately by the geometry of the alkene which is a precursor to 11 We now report that this strategy is useful for pyrrolizidine synthesis (12, n=1), and that the stereochemical outcome of this process is not as straightforward as was predicted.



Our synthesis began with 2,3,5-tri-O-benzyl-L-xylofuranose 13,20a which was prepared from L-xylose in 3 steps.^{20b} Wittig olefination produced 14, which was converted to azide 15 by displacement of a triflate with azide ion. The azide 15 was quite unstable, and underwent an intramolecular 1.3-dipolar excloaddition upon standing at room temperature. Hence, it was best to carry 15 to the next step without delay. Ozonolysis of 15 afforded the aldehyde 16, which was not purified due to its sensitivity, but was directly converted to the Z-alkene 17 by a very stereoselective Wittig reaction with a silvloxy-substituted vlide.²¹ Again, intramolecular 1,3-dipolar cycloaddition was observed if this azidoalkene was allowed to stand at room temperature. Epoxidation of the 17 with mCPBA produced a 1.1 mixture of epoxides 18α and 18β , which were not separated ²² Tosylation of $18\alpha/\beta$ gave $19\alpha/\beta$ as a 2.1 mixture of isomers, since one of the isomers of 18 underwent tosylation faster than the other, and the reaction could not be driven to completion Reduction of $19\alpha/\beta$ without debenzylation was possible by hydrogenolysis. The resultant amine was not isolated but was directly heated in refluxing EtOH containing K₂CO₃, affording a 2⁻¹ mixture of two pyrrolizidines 20 and 21 which were easily separated by flash chromatography At this point, the stereochemistry of 20 and 21 was unknown, but it was assumed that the methine hydrogens at C-7 and C-7a were cis in both isomers due to their cis relationship in alkene 17 and epoxides 18 and 19. Hydrogenolysis of the benzyl ethers of the separated pyrrolizidines 20 and 21 using a greater amount of palladium catalyst afforded two tetrahydroxypyrrolizidines, which were found to be (+)-7-epiaustraline 7 and (-)-7-epialexine 9 by comparison of their physical and spectral data to hterature values, including optical rotation 7.12 The formation of 7 was unexpected, since the methine hydrogens at C-7 and C-7a were trans rather than cis. We had expected australine 4 to result from the cyclization of 19 α . Isomerization of the alkene 17 or the epoxides 18 and 19 from *cis* to *trans* during their manipulation was ruled out by ¹H NMR spectroscopy. Given that the stereochemistry of the centers at C1-C3 is based on L-xylose, there are only four possible structures that may arise upon cyclization of the aminoepoxides 19. 3, 5, 7, and 9. Since the structures of 3, 5, and 9 have been established by X-ray crystallography 2,5,12 and since our data for 7 are not consistent with these structures, it appears that the assignment of 7 must be as shown. The literature assignment of 7 rests on NOE experiments and comparison with known structures.⁷ The absolute stereochemistry was not determined, but was proposed to be as shown. The unusual inversion of configuration at C-7 which occurs in the formation of 7 is currently under investigation

In summary, a relatively short route to (+)-7-epiaustraline 7 and (-)-7-epialexine 9 has been developed. The absolute stereochemistry of the anti-HIV alkaloid 7 has been confirmed to be as shown based on the stereochemistry of L-xylose. While the use of the tandem aminoepoxide opening/intramolecular alkylation approach is efficient, caution must be used in predicting the stereochemical outcome of this reaction.



mCPBA, CH_2Cl_2 , 0° C, 23° C, 24h (f) 2 eq pTsCl, 3 eq pyr, 0 1 eq DMAP, CH_2Cl_2 , -15° C, 48h *Based on 12% recovered **18** (g) 5 wt % of 10% Pd/C, $Et_2O/EtOH$ (2 1), 1 atm H₂, 23° C, 15h, filter, add 6 eq K₂CO₃, EtOH, reflux 20h, separate by flash chrom. (h) 300 wt % of 10% Pd/C, EtOH, 1 atm H₂, 23° C, 48h

Acknowledgments We thank the National Institutes of Health (GM-35572) and the University of Michigan (Rackham Predoctoral Fellowship to J.V.H.) for support of this research. J.V.H. was also supported in part by National Research Service Award T32-GM07767.

REFERENCES

- Reviews. (a) Howard, A. S., Michael, J. P. In The Alkaloids; Brossi, A., Ed.; Academic Press: New 1. York, 1986; Vol. 26, Ch. 3. (b) Elbein, A. D.; Molyneux, R. J. In Alkaloids Chemical and Biological Perspectives; Pelletier, S. W., Ed.; Wiley: New York, 1987; Vol. 5, Ch. 1, pp. 1-54. (c) Broquist, H. P. Ann Rev. Nutr 1985, 5, 391-409.
- 2. Nash, R. J.; Fellows, L. E.; Dring, J. V.; Fleet, G. W. J.; Derome, A E., Hamor, T. A.; Scofield, A M; Watkin, D. J. Tetrahedron Lett 1988, 29, 2487-2490.
- While these alkaloids may be named as stereoisomers of either alexine or australine, a suggestion has been 3 made⁶ that alkaloids with the R configuration at the bridgehead position C-7a be named as australines, while those with the S configuration at 7a be named as alexines.
- Nash, R. J.; Fellows, L. E.; Plant, A C, Fleet, G. W. J.; Derome, A. E; Baird, P D.; Hegarty, M P.; Scofield, A. M. Tetrahedron, 1988, 44, 5959-5964. Δ
- Molyneux, R. J., Benson, M; Wong, R Y; Tropea, J. E.; Elbein, A. D. J Nat Prod. 1988, 51, 1198-5. 1206
- 6. Harris, C. M.; Harris, T. M., Molyneux, R J., Tropea, J E., Elbein, A D. Tetrahedron Lett. 1989, 30. 5685-5688
- Nash, R. J.; Fellows, L. E.; Dring, J. V.; Fleet, G. W. J.; Girdhar, A.; Ramsden, N. G.; Peach, J. M.; 7. Hegarty, M. P.; Scofield, A. M. Phytochemistry 1990, 29, 111-114.
- 8 Scofield, A. M.; Rossiter, J. T.; Witham, P., Kite, G. C.; Nash, R. J.; Fellows, L. E. Phytochemistry 1990, 29, 107-109.
- Q Tropea, J. E.; Molyneux, R. J.; Kaushal, G. P; Pan, Y. T; Mitchell, M; Elbein, A D. Biochemistry 1989, 28, 2027-2034.
- 10 Elbein, A. D.; Tropea, J. E.; Molyneux, R. J. U. S. Pat Appl. US 289,907 Chem. Abstr. 1990, 113, P. 91444p
- 11 Fellows, L; Nash, R. PCT Int Appl WO GB Appl. 89/7,951. Chem Abstr. 1990, 114, 143777f
- 12. Fleet, G W. J.; Haraldsson, M., Nash, R J.; Fellow, L. E Tetrahedron Lett. 1988, 29, 5441-5444.
- 13 In this paper, the discussion is limited to polyhydroxylated pyrrolizidine alkaloids with a carbon substituent at C-3, rather than the more common type with a substituent at C-1 (i.e., the necine bases) For the latter, see: Robins, D. J. Nat Prod Rep. 1989, 6, 221-230
- 14 Choi, S.; Bruce, I.; Fairbanks, A. G., Fleet, G. W. J.; Jones, A. H., Nash, R. J ; Fellow, L. E. Tetrahedron Lett. 1991, 32, accompanying paper. We thank Professor Fleet for disclosure of his work prior to publication and for helpful discussions.
- 15. Carpenter, N. M , Fleet, G W. J., Cenci di Bello, I.; Winchester, B ; Fellows, L E.; Nash, R J *Tetrahedron Lett.* **1989**, *30*, 7261-7264 Burgess, K.; Henderson, I. *Tetrahedron Lett.* **1990**, *31*, 6949-6952.
- 17. Pearson, W H.; Bergmeier, S C J. Org Chem. 1991, 56, 1976-1978.
- 18 A similar strategy was used by Fleet, et al, for the synthesis of swainsonine and ring-contracted analogues.¹⁵ For other conceptually related double cyclizations using aminoepoxides, see ref. 19d-f.
- 19 For representative examples of related heterocyclic syntheses using intramolecular epoxide openings, see-(a) Bernotas, R. C.; Ganem, B Tetrahedron Lett 1984, 25, 165-168. (b) Pilard, S.; Vaultier, M. Tetrahedron Lett. 1984, 25, 1555-1556. (c) Adams, C E.; Walker, F. J.; Sharpless, K. B. J Org Chem 1985, 50, 420-422. (d) Setoi, H; Takeno, H.; Hashimoto, M. J. Org. Chem. 1985, 50, 3948-3950. (e) Setoi, H; Takeno, H.; Hashimoto, M. Tetrahedron Lett. 1985, 26, 4617-4620. (f) Kim, Y G, Cha, J. K. Tetrahedron Lett 1989, 30, 5721-5724.
- 20. (a) MacCoss, M.; Chen, A; Tolman, R L. Tetrahedron Lett 1985, 26, 4287-4290. (b) The procedure used was the same as that reported for the synthesis of 2.3.5-tri-O-benzyl-D-arabinofuranose Barker, R.; Fletcher, H. G J Org Chem. 1961, 26, 4605-4609; Tejima, S.; Fletcher, H. G. J Org Chem. 1963, 28, 2999-3004
- 21. (a) Takahashi, T.; Miyazawa, M.; Ueno, H.; Tsuji, J. Tetrahedron Lett 1986, 27, 3881-3884. (b) Salomond, W. G; Barta, M. A; Havens, J. L. J Org Chem 1978, 43, 790-792 For the preparation of Ph3P+(CH2)3OH Br-, see: (c) Kunz, H. Leibigs Ann Chem 1973, 2001-2009
- 22 For recent reports of similar epoxidations, see. (a) Wang, Z.; Schreiber, S. L. Tetrahedron Lett. 1990, 31, 31-34. (b) Erickson, S. D., Still, W. C. Tetrahedron Lett. 1990, 31, 4253-4256 (c) Coutts, S. J., Wittman, M. D.; Kallmerten, J Tetrahedron Lett 1990, 31, 4301-4304