

# TRANSITION PROBABILITIES IN HIGHLY IONIZED $p^2$ AND $p^4$ CONFIGURATIONS

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**Abstract**—Line strengths of magnetic dipole and electric quadrupole transitions in the configurations  $2p^2$ ,  $2p^4$ ,  $3p^2$  and  $3p^4$  are calculated using interpolated values of spin-orbit interaction parameters and unperturbed energies of the levels. Probabilities of the important ground state transitions are tabulated in the four isoelectronic sequences through the following net charges:  $17(2p^2)$ ,  $13(2p^4)$ ,  $15(3p^2)$ ,  $13(3p^4)$ .

We present the results of calculations of probabilities of magnetic dipole and electric quadrupole transitions for high members of the carbon, oxygen, silicon and sulphur isoelectronic sequences. The computations are based on interpolations of the form

$$\theta^{1/3} = \alpha Z + \beta \quad \sigma^{1/3} = \alpha' Z + \beta'$$

where, in Garstang's<sup>(1)</sup> notation,

$$\theta = \frac{\zeta}{DP} \quad \sigma = \frac{\zeta}{SP}$$

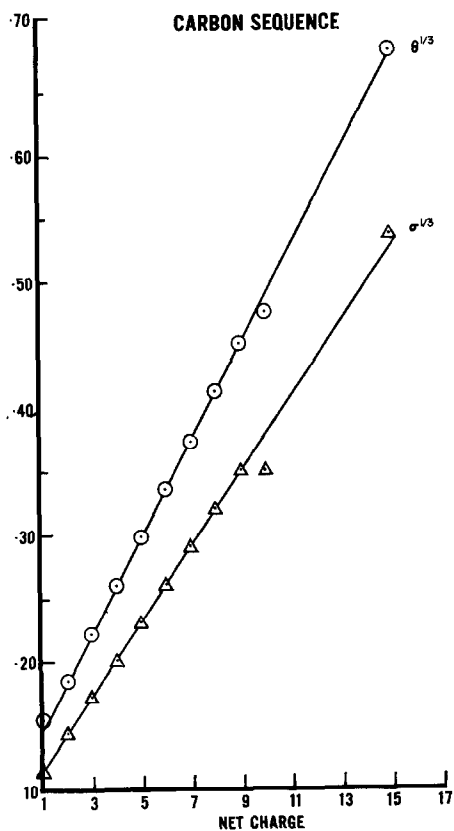
TABLE 1. CARBON SEQUENCE

Ion	Magnetic dipole transition probabilities (sec <sup>-1</sup> )					Electric quadrupole transition probabilities (sec <sup>-1</sup> )			
	<sup>1</sup> S <sub>0</sub> - <sup>3</sup> P <sub>1</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>2</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>1</sub>	<sup>3</sup> P <sub>2</sub> - <sup>3</sup> P <sub>1</sub>	<sup>3</sup> P <sub>1</sub> - <sup>3</sup> P <sub>0</sub>	<sup>1</sup> S <sub>0</sub> - <sup>1</sup> D <sub>2</sub>	<sup>1</sup> S <sub>0</sub> - <sup>3</sup> P <sub>2</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>0</sub>	<sup>3</sup> P <sub>2</sub> - <sup>3</sup> P <sub>0</sub>
C*	0.226	0.323	0.478	0.627	0.778	0.50	0.419	0.755	0.1220
N+*	0.134	0.230	0.3103	0.574	0.821	1.08	0.316	0.642	0.1113
O+2*	0.23	0.121	0.271	0.497	0.426	1.6	0.371	0.519	0.1035
F+3*	1.1	0.198	0.134	0.378	0.220	2.1	0.323	0.564	0.0950
Ne+4*	4.2	0.38	0.14	0.245	0.213	2.6	0.268	0.419	0.0852
Na+5	13.4	1.20	0.44	0.121	0.160	3.1	0.169	0.447	0.0738
Mg+6	37	3.3	1.26	0.179	0.126	3.5	0.136	0.3107	0.0623
Al+7	93	8.4	3.4	0.26	0.194	4.0	0.179	0.222	0.05118
Si+8	210	19.1	8.1	0.76	0.31	4.4	0.150	0.245	0.052
P+9	370	33	15.0	1.98	0.96	4.5	0.22	0.375	0.0420
S+10	860	82	40	4.7	2.7	4.7	0.47	0.3154	0.0469
Cl+11	1610	157	82	10.3	7.1	4.9	0.80	0.326	0.0322
Ar+12	2900	290	166	23	17.7	4.9	1.31	0.344	0.0266
K+13	4900	510	320	42	42	4.9	2.0	0.372	0.02185
Ca+14†	7800	860	580	77	95	4.9	3.1	0.114	0.0248
Sc+15	13,000	1400	1090	138	200	5.0	4.7	0.1167	0.0126
Ti+16	20,000	2200	1940	260	410	4.9	6.7	0.124	0.0133

\* Garstang.<sup>(1)</sup>

† Line strengths from Garstang.<sup>(8)</sup>

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FIGS. 1-4. THE VARIATION OF  $\theta^{1/3}$  AND  $\sigma^{1/3}$  ALONG FOUR ISOELECTRONIC SEQUENCES.

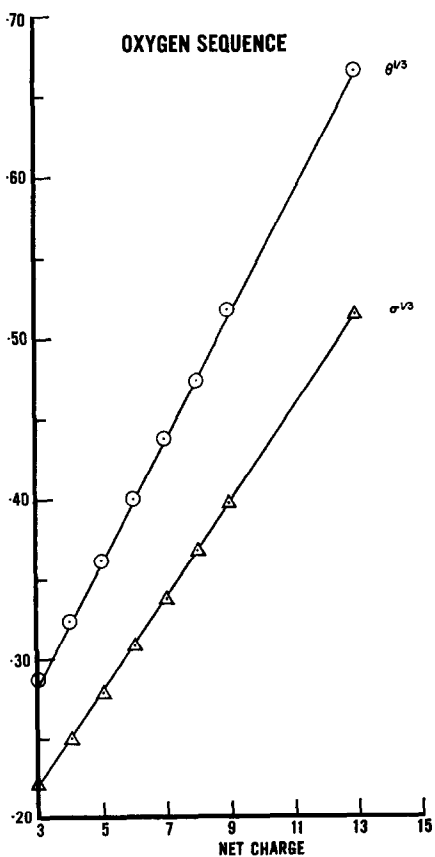


FIG. 2.

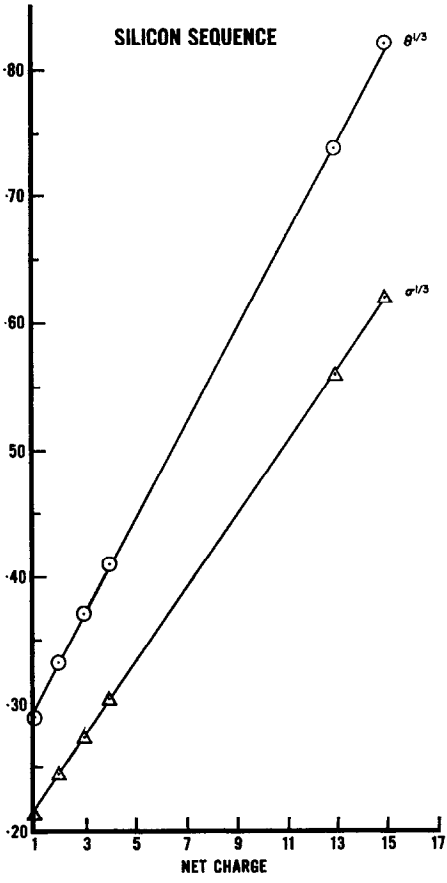


FIG. 3.

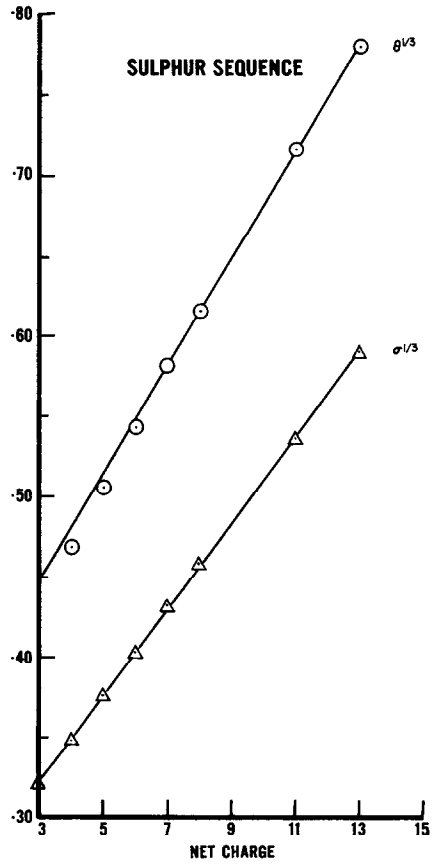


FIG. 4.

TABLE 2. OXYGEN SEQUENCE

Ion	Magnetic dipole transition probabilities (sec <sup>-1</sup> )					Electric quadrupole transition probabilities (sec <sup>-1</sup> )			
	<sup>1</sup> S <sub>0</sub> - <sup>3</sup> P <sub>1</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>2</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>1</sub>	<sup>3</sup> P <sub>0</sub> - <sup>3</sup> P <sub>1</sub>	<sup>3</sup> P <sub>1</sub> - <sup>3</sup> P <sub>2</sub>	<sup>1</sup> S <sub>0</sub> - <sup>1</sup> D <sub>2</sub>	<sup>1</sup> S <sub>0</sub> - <sup>3</sup> P <sub>2</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>0</sub>	<sup>3</sup> P <sub>0</sub> - <sup>3</sup> P <sub>2</sub>
O*	0.178	0.369	0.222	0.417	0.489	1.28	0.37	0.511	0.9103
F**	0.49	0.144	0.1138	0.318	0.389	2.1	0.316	0.541	0.818
Ne <sup>+2*</sup>	2.2	0.20	0.160	0.3113	0.361	2.8	0.351	0.312	0.720
Na <sup>+3</sup>	7.6	0.66	0.197	0.355	0.130	3.5	0.121	0.430	0.6148
Mg <sup>+4</sup>	23	1.96	0.56	0.122	0.125	4.2	0.127	0.467	0.588
Al <sup>+5</sup>	61	5.3	1.44	0.171	0.45	4.8	0.156	0.3132	0.543
Si <sup>+6</sup>	148	12.9	3.3	0.194	1.45	5.5	0.105	0.325	0.4182
P <sup>+7</sup>	340	29	6.8	0.47	4.2	6.2	0.178	0.346	0.468
S <sup>+8</sup>	710	65	14.2	0.95	11.4	6.9	0.33	0.376	0.322
Cl <sup>+9</sup>	1410	129	25	1.95	28	7.7	0.51	0.3125	0.368
Ar <sup>+10</sup>	2700	250	44	3.0	67	8.6	0.78	0.320	0.3181
K <sup>+11</sup>	4900	460	71	3.6	149	9.6	1.14	0.330	0.344
Ca <sup>+12†</sup>	8560	840	107	3.6	330	11.0	1.60	0.342	0.311

\* Garstang.<sup>(1)</sup>† Garstang.<sup>(1)</sup>

TABLE 3. SILICON SEQUENCE

Ion	Magnetic dipole transition probabilities (sec <sup>-1</sup> )					Electric quadrupole transition probabilities (sec <sup>-1</sup> )			
	<sup>1</sup> S <sub>0</sub> - <sup>3</sup> P <sub>1</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>2</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>1</sub>	<sup>3</sup> P <sub>2</sub> - <sup>3</sup> P <sub>1</sub>	<sup>3</sup> P <sub>1</sub> - <sup>3</sup> P <sub>0</sub>	<sup>1</sup> S <sub>0</sub> - <sup>1</sup> D <sub>2</sub>	<sup>1</sup> S <sub>0</sub> - <sup>3</sup> P <sub>2</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>0</sub>	<sup>3</sup> P <sub>2</sub> - <sup>3</sup> P <sub>0</sub>
Si	0.1355	0.269	0.3964	0.4417	0.818	0.798	0.3114	0.3624	0.3356
P**	0.22	0.168	0.3625	0.374	0.824	1.95	0.3630	0.32	0.3606
S <sup>+3*</sup>	0.85	0.1635	0.1246	0.3241	0.367	2.54	0.163	0.387	0.747
Cl <sup>+3*</sup>	2.61	0.195	0.1796	0.3819	0.211	3.15	0.38	0.422	0.328
Ar <sup>+4*</sup>	6.82	0.512	0.221	0.1271	0.3796	3.78	0.311	0.451	0.3132
K <sup>+5</sup>	15.7	1.14	0.53	0.176	0.126	4.1	0.141	0.3110	0.354
Ca <sup>+6</sup>	34	2.5	1.22	0.197	0.176	4.3	0.25	0.321	0.3189
Sc <sup>+7</sup>	67	5.0	2.7	0.46	0.21	4.5	0.41	0.338	0.460
Ti <sup>+8</sup>	125	9.5	5.6	0.98	0.53	4.6	0.66	0.365	0.3174
V <sup>+9</sup>	220	17.5	11.3	1.92	1.28	4.6	1.02	0.3108	0.346
Cr <sup>+10</sup>	380	31	22	3.5	3.0	4.5	1.54	0.3170	0.3116
Mn <sup>+11</sup>	630	53	41	6.0	6.5	4.4	2.3	0.326	0.327
Fe <sup>+12</sup>	1000	87	72	9.6	13.8	4.3	3.2	0.341	0.361
Co <sup>+13</sup>	1570	145	132	14.3	28	4.1	4.7	0.351	0.3132
Ni <sup>+14</sup>	2400	230	230	22	56	3.9	6.7	0.361	0.328

\* Czyzak and Krueger.<sup>(3)</sup>

Values of the spin orbit interaction parameter,  $\zeta$ , and the intervals,  $DP$  and  $SP$ , between the unperturbed terms were obtained from Garstang,<sup>(1)</sup> Edlen,<sup>(2)</sup> Rohrlich.<sup>(4)</sup> Figures 1-4 show the variation of the known values of  $\theta^{1/3}$  and  $\sigma^{1/3}$  along the four sequences together with the interpolations used in the calculations.

Estimates of the quadrupole moment

$$s_q = \frac{2}{5} \int_0^\infty r^2 p^2 dr,$$

required for the determination of the electric quadrupole line strengths, were obtained by extrapolating

$$s_q^{-1/2} = \alpha''Z + \beta''$$

TABLE 4. SULPHUR SEQUENCE

Ion	Magnetic dipole transition probabilities (sec <sup>-1</sup> )					Electric quadrupole transition probabilities (sec <sup>-1</sup> )			
	<sup>1</sup> S <sub>0</sub> - <sup>3</sup> P <sub>1</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>2</sub>	<sup>1</sup> D <sub>3</sub> - <sup>3</sup> P <sub>1</sub>	<sup>3</sup> P <sub>0</sub> - <sup>3</sup> P <sub>1</sub>	<sup>3</sup> P <sub>1</sub> - <sup>3</sup> P <sub>2</sub>	<sup>1</sup> S <sub>0</sub> - <sup>1</sup> D <sub>2</sub>	<sup>1</sup> S <sub>0</sub> - <sup>3</sup> P <sub>2</sub>	<sup>1</sup> D <sub>2</sub> - <sup>3</sup> P <sub>0</sub>	<sup>3</sup> P <sub>0</sub> - <sup>3</sup> P <sub>2</sub>
S*	0.350	0.1273	0.2800	0.295	0.2139	1.78	0.2731	0.495	0.709
Cl <sup>+</sup> *	1.34	0.10	0.1292	0.2143	0.2753	2.29	0.118	0.4120	0.478
Ar <sup>+</sup> 2*	4.02	0.32	0.1829	0.2512	0.1306	3.10	0.425	0.4290	0.272
K <sup>+</sup> 3	10.4	0.83	0.20	0.146	0.104	3.9	0.186	0.460	0.4122
Ca <sup>+</sup> 4	24	1.92	0.43	0.135	0.31	4.6	0.156	0.2112	0.445
Sc <sup>+</sup> 5	50	4.1	0.83	0.171	0.83	5.3	0.26	0.2191	0.2142
Ti <sup>+</sup> 6	97	8.4	1.53	0.133	2.0	6.1	0.45	0.29	0.241
V <sup>+</sup> 7	181	16.0	2.6	0.21	4.7	6.9	0.66	0.246	0.2108
Cr <sup>+</sup> 8	320	29	4.0	0.30	10.3	7.6	0.92	0.272	0.226
Mn <sup>+</sup> 9	560	50	6.1	0.29	22	8.5	1.24	0.2111	0.256
Fe <sup>+</sup> 10	920	86	8.8	0.28	43	9.3	1.60	0.2168	0.121
Co <sup>+</sup> 11	1470	145	12.4	0.164	84	10.2	2.1	0.225	0.124
Ni <sup>+</sup> 12	2300	240	16.8	0.172	156	11.0	2.5	0.237	0.148
Cu <sup>+</sup> 13	3400	380	22		280	11.8	3.0		
Zn <sup>+</sup> 14	4900	610	29		490	12.7	3.4		

\* Czyzak and Krueger.<sup>(5)</sup>

TABLE 5. EXTRAPOLATED QUADRUPOLE MOMENTS

Net charge	2p <sup>2</sup>	2p <sup>4</sup>	3p <sup>2</sup>	3p <sup>4</sup>
4		0.234		0.804
5		0.176		0.649
6	0.177	0.138	0.671	0.536
7	0.138	0.110	0.543	0.449
8	0.111	0.1905	0.449	0.382
9	0.1912	0.1756	0.377	0.329
10	0.1760	0.1641	0.321	0.286
11	0.1645	0.1550	0.277	0.252
12	0.1554	0.1477	0.241	0.223
13	0.1481	0.1418	0.212	0.198
14	0.1420		0.188	0.178
15	0.1372		0.167	0.161
16	0.1331			
17	0.1296			

using values of  $s_q$  determined by Garstang<sup>(1)</sup> and Czyzak and Krueger<sup>(5)</sup> from self-consistent field wave functions. The extrapolated quadrupole moments, which provide the greatest source of errors in our transition probabilities, are given in Table 5. When better values of  $s_q$  become available the electric quadrupole probabilities can be accordingly revised. In the case of Si, a quadrupole moment,  $s_q = 3.40$ , was calculated from the wave function of neutral silicon given by Herman and Skillman.<sup>(7)</sup>

Magnetic dipole and electric quadrupole probabilities were determined by the usual methods<sup>(1,6)</sup> using the wave numbers estimated by Rohrlich and Pecker,<sup>(3)</sup> and are given in Tables 1-4. These tables contain also the calculations of Garstang<sup>(1,8,9)</sup> and of Czyzak and Krueger<sup>(5)</sup> for some members of the sequences.

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**Резюме**—Вычисляются силовые линии магнитного диполя и электрические четырехполюсные переходы в конфигурации  $2p^2$ ,  $2p^4$ ,  $3p^2$  и  $3p^4$  используя интерполированные значения параметров спино-орбитального взаимодействия, а также невозмущенные энергии уровней. Возможность значительных переходов наземного состояния расположены в виде диаграммы в четырех изоэлектронных последовательностях посредством следующих зарядов нетто:  $17(2p^2)$ ,  $13(2p^4)$ ,  $15(3p^2)$ ,  $13(3p^4)$ .