# 209. Thioalkylation of Meldrum's Acid: Protected Alkylidene Derivatives of Isopropylidene Malonate 

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Thioalkylated Meldrum's acid is easily available by treatment of Meldrum's acid with an aldehyde and thiophenol in the presence of catalytical amounts of piperidinium acetate ( $\rightarrow \mathbf{1 - 6}$, Table 1). The adducts $\mathbf{1 - 6}$ are crystalline, stable compounds and they can be caused to react directly with nucleophiles and dienes (see $\mathbf{3} \rightarrow \mathbf{7 - 1 2}$, Scheme l). The regeneration of the parent olefin is effected thereby by simply dissolving the adduct under neutral or basic conditions. Extension of this method to thiocarboxylic acids allowed the preparation of the corresponding formaldehyde derivatives 13 and 15 (Table 3).

Introduction. - Isopropylidene alkylidenemalonates i [1] have achieved considerable interest as highly reactive Michael acceptors [2], electron-deficient dienophiles [3], strongly polarized heterodienes [4], and as precursors of the corresponding saturated compounds [5] as well as of methylene ketenes [6].

i

ii

iii

iv

$v$

Our own interest in these structures stems from their potential as unique alkylating bioprobes [7]. Besides, their reactivity provides special insight into the dynamic character of vinylogous nucleophilic additions.

A simple synthetic approach to alkylidenemalonates i, the Knoevenagel condensation with Meldrum's acid ( $=2,2$-dimethyl-1,3-dioxane-4,6-dione), has been shown to be efficient in the case of aromatic and $\alpha$-branched aldehydes [8] as well as for certain ketones [8] [9] or imines [10] [11]. In the reaction of simple aldehydes, however, the olefins tend to be trapped by Meldrum's acid [12], or, as in the case of butyraldehyde, to undergo
a further Knoevenagel reaction followed by the Michael addition'). Several methods have been developed to overcome this problem. Thus, the olefins have been trapped in situ with dienes [3a,b], olefins [4], reducing agents [5a,b], indole [2b,d], methoxide [13], and secondary amines [14]. For heteronucleophiles, the addition is reversible, and the olefin can be regenerated by treatment with acid, thus providing an easy access to the alkylidene derivatives $\mathbf{i}$. More recently, compounds of type $\mathbf{i}$ have also been prepared by addition of metallorganic reagents to olefins iv and $\mathbf{v}$ [15] [16].

A limitation, however, of these methods is the use of fairly strong bases combined with the requirement of anhydrous reaction conditions. We have found thio derivatives to be the intermediates of choice for many of these trapping reactions.

Results. - Good yields of the crystalline adducts $1 \mathbf{6}$ were obtained by simply mixing Meldrum's acid, aldehyde, and thiophenol in the presence of piperidinium acetate in MeCN , quenching the reaction with an excess of aq. citric-acid solution, and filtering the product (Table 1). Similar results could be obtained by using a variety of different solvents and bases ${ }^{2}$ ). Due to its high volatility combined with the discrete melting points and crystalline character of the adducts, thiophenol seemed to be the most appropriate thiol. The method seems to be generally applicable to any aldehyde, except formaldehyde.

Table 1. Synthesis of /1-( Phenylthio) alkyl/malonates 1-6

|  |  | R | M.p.[ ${ }^{\circ}$ ] | Yield [\%] |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |

Compounds 1-6 were stored at room temperature for several months without decomposition. In solution, partial dissociation occurred as shown in the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectra ${ }^{3}$ ). Depending on purity and solvent, characteristic equilibrium ratios of adduct and olefin/ thiophenol were observed by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (Table 2).

The versatility of compounds $1-6$ is demonstrated in the case of the butyraldehyde adduct 3 (Scheme 1). Due to the tendency to eliminate thiophenol, adduct 3 showed the same reactivity as the free olefin 7 yielding e.g. the epoxide 8 with $\mathrm{H}_{2} \mathrm{O}_{2}$. In addition, adduct 3 was observed to undergo an oxidative cyclization with Meldrum's acid: On trying to eliminate the thiophenol moiety by oxidation with sodium metaperiodate, we obtained cyclopropane derivative 9 as a by-product; the yield of 9 was strongly improved by adding 1 equiv. of Meldrum's acid. This suggests that in the formation of 9 from 7, olefin 7 undergoes addition of $\mathrm{H}_{2} \mathrm{O}$, followed by a retro-aldol reaction to generate

[^0]Table 2. Ratios Adduct/Olefin + Thiophenol


| Adduct | $\mathbf{1}$ | $\mathbf{2}$ | $\mathbf{3}$ | $\mathbf{4}$ | $\mathbf{5}$ | $\mathbf{6}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| R | Me | Et | Pr | $\mathrm{i}-\mathrm{Pr}$ | $\mathrm{CH}_{3}\left(\mathrm{CH}_{2}\right)_{5}$ | Ph |
| Adduct/olefin + thiophenol $\left.{ }^{\text {a }}\right)$ | $5.5: 1$ | $2.5: 1$ | $1: 1$ | $1: 0$ | $1: 0$ | $15: 1$ |

${ }^{\text {a }}$ ) The ratios were determined by integration of the olefin peak ( $7.9-8.1 \mathrm{ppm}$ ) and the methine protons 3.7-4.1 $((\mathrm{H}-\mathrm{C}(1))$ and $3.9-5.2(\mathrm{H}-\mathrm{C}(2)) \mathrm{ppm})$.

a) $\mathrm{K}_{3}\left[\mathrm{Fe}(\mathrm{CN})_{6}\right], \mathrm{KOH}$; b) $\mathrm{H}_{2} \mathrm{O}_{2}, \mathrm{MeCN}$; c) Meldrum's acid, $\mathrm{NaIO}_{4}, \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}$; d) $\mathrm{NaBH}_{4}$, THF/ EtOH ; e) $\mathrm{MeNO}_{2}, \mathrm{Bu}_{4} \mathrm{NOH}, \mathrm{MeOH} / \mathrm{THF} ; ~ f$ ) 2,3-dimethyl-1,3-butadiene, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$.
unsubstituted Meldrum's acid. This then adds to olefin 7 as a radical ${ }^{4}$ ), generated by H -transfer to a phenylthio radical. The radical produced by this addition might then undergo a further H -abstraction followed by a ring closure, to give 9 . The structure is in agreement with elementary analysis as well as spectral data. In addition, we have saponified 9 to the corresponding known tetraacid [18].

Reaction of 3 with $\mathrm{NaBH}_{4}, \mathrm{MeNO}_{2}$, or 2,3-dimethyl-1,3-butadiene gave the alkyl derivative 10, the nitro compound 11, and the Diels-Alder adduct 12, respectively.

As mentioned above, probably the most valuable compound, the formaldehyde/thiophenol adduct, was not stable enough to be isolated in pure form. On trying to diversify

[^1]our method, we found, to our astonishment, that the stability of thiocarboxylic-acid adducts was even higher then the one of the corresponding thiophenol adducts. Using thioacetic acid, up to $75 \%$ of formaldehyde derivative 13 were obtained (Table 3). Adducts 14 and 15 were prepared similarly. The intermediate on the way to 13 , the methylidene-substituted Meldrum's acid vi, is initially not only trapped by the thioacid,

Table 3. Synthesis of [1-(Thiocarboxy)alkyl/malonates 13-15

|  |  |  |  | $\mathrm{R}, \mathrm{R}^{\prime}$ | M.p. [ $\left.{ }^{\circ}\right]$ | Yield [\%] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | 13 | $\mathrm{H}, \mathrm{Me}$ | 108-109 (dec.) | 75 |
|  |  |  | 14 | $\mathrm{Me}, \mathrm{Me}$ | 105 | 94 |
|  |  |  | 15 | H, Ph | 110-111 (dec.) | 68 |

Scheme 2

a) Meldrum's acid; b) AcSH ; c) HCHO.

a) 2,3-Dimethyl-1,3-butadiene, DMSO; b) 1,3-butadienyl acetate, DMSO; c) 2-methoxypropene, $\mathrm{K}_{2} \mathrm{CO}_{3}$, MeCN ; d) $\mathrm{H}^{+}$; e) morpholinocyclopentanone enamine, MeCN ; f) $\mathrm{RCH}_{2} \mathrm{NO}_{2}, \mathrm{Bu}_{4} \mathrm{NOH}$, THF; g) Meldrum's acid, $\mathrm{NaIO}_{4}, \mathrm{MeCN}, \mathrm{H}_{2} \mathrm{O}$; h) morpholine, MeCN .
but also by Meldrum's acid to give the known diisopropylidene methylenedimalonate (vii) [12a]. Thus, on quenching the reaction at an early stage, different mixtures of Michael adduct vii, thioacetic-acid adduct 13, and formaldehyde adduct viii (Scheme 2) were obtained. On longer reaction times using 1.3 equiv. of formaldehyde, only $\mathbf{1 3}$ was isolated ${ }^{5}$ ).

In addition to the well known reaction of methylidene-substituted Meldrum's acid with dienes $(\rightarrow \mathbf{1 6}, \mathbf{1 7})$ [3a, b] [15], 13 reacted with enol ethers [4c] and nucleophiles such as nitronates and enamines under mild conditions ( $\rightarrow \mathbf{1 8} \mathbf{2 1}$; Scheme 3). The parent olefin could, thereby, be generated under neutral or basic conditions. In analogy to the transformation $3 \rightarrow 9$ (see Scheme 1), 13 gave 22. Reaction of $\mathbf{1 3}$ with morpholine, finally, gave a good yield of the Mannich adduct 23, as stable compound that should also allow the generation of the methylidene-substituted Meldrum's acid under acidic conditions [14].

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## Experimental Part

1. General. $\mathrm{H}_{2} \mathrm{O}$-sensitive reactions were carried out in oven-dried flasks ( $120^{\circ}$ ) under $\mathrm{N}_{2}$. THF was distilled over $\mathrm{Na} /$ benzophenone just prior to use. MeCN was stored over $4-\AA$ molecular sieves. Solns. were dried ( $\mathrm{MgSO}_{4}$ ) and evaporated $<40^{\circ}$ in a Büchi rotary evaporator. TLC: Merck precoated silica gel 60 F- 254 plates, detection by UV and phosphomolybdic acid. M.p. (of recrystallized products, uncorrected): Thomas-Hoover-Uni-Melt apparatus. IR : Nicolet $60-S X ;$ K.Br pellets. ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-NMR: Bruker AM-300 or Bruker WM-360; chemical shifts $\delta$ in ppm rel. to TMS as internal standard, coupling constants $J$ in Hz ; the solvent used for ${ }^{13} \mathrm{C}$-NMR and ${ }^{1} \mathrm{H}-\mathrm{NMR}$ was always the same. MS: Finnigan $4021 G C M S / D S$ by direct probe sample introduction; chemical ionization (CI) was accomplished with $\mathrm{NH}_{3}$. Elemental analyses were performed by Spang Microanalytical Laboratories, Eagle Harbor, Mi.
2. Synthesis of 1-6. 2.1. General Procedure. Crystalline piperidinium acetate ( 0.1 equiv.) was added with stirring to a cooled soln. ( $5^{\circ}$ ) of Meldrum's acid ( = isopropylidene malonate; 1 equiv.), aldehyde, and thiophenol ( 1.05 equiv. each) in MeCN. After 1 h , the cooling bath was removed and stirring was continued for $2-4 \mathrm{~h}$ (TLC). The reaction was quenched by slowly adding an excess of aq. $10 \%$ citric-acid soln. The product was filtered, washed sequentially with $\mathrm{H}_{2} \mathrm{O}$ and $\mathrm{Et}_{2} \mathrm{O}$ /pentane $1: 5$ and finally dried under high vacuum. Care should be taken in drying and recrystallizing of the products since heating over $40^{\circ}$ can cause decomposition!
2.2. Isopropylidene 2-(Phenylthio)propane-1,1-dicarboxylate ( $=2,2$-Dimethyl-5-[1-(phenylthio)ethyl $]-1,3$ -dioxane-4,6-dione; 1). Meldrum's acid ( $2.00 \mathrm{~g}, 13.9 \mathrm{mmol}$ ), thiophenol, and acetaldehyde in 30 ml of MeCN at $5^{\circ}$ in a closed flask: $3.54 \mathrm{~g}(91 \%)$ of $\mathbf{1}$. M.p. (AcOEt/hexane) $102^{\circ}$. IR: $1779 \mathrm{~m}, 1747 \mathrm{~s}, 1386 \mathrm{~s}, 1328 \mathrm{~s}, 1317 \mathrm{~s}, 1270 \mathrm{~m}, 1233 \mathrm{~m}$, $1205 \mathrm{~m}, 1188 \mathrm{~m}, 1055 \mathrm{~m}, 1026 \mathrm{~m}, 985 \mathrm{~m}, 876 \mathrm{~m}, 752 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.55-7.21(\mathrm{~m}, 5 \mathrm{arom} . \mathrm{H}) ; 4.10(\mathrm{dq}, J=2.6$, 7.1, $\mathrm{H}-\mathrm{C}(2)) ; 3.84(d, J=2.6, \mathrm{H}-\mathrm{C}(1)) ; 1.77(s, \mathrm{Me}) ; 1.75(s, \mathrm{Me}) ; 1.60(d, J=7.1,3 \mathrm{H}-\mathrm{C}(3))$; olefin ( + thiophenol): $8.03(q, J=7.5, \mathrm{H}-\mathrm{C}(2)) ; 2.49(d, J=7.5,3 \mathrm{H}-\mathrm{C}(3)) ; 1.74(s, 2 \mathrm{Me})$; ratio $5.5: 1 .{ }^{13} \mathrm{C}-\mathrm{NMR}$ : $163.8 ; 163.4 ; 135.3 ; 132.1 ; 129.2 ; 127.7 ; 105.2 ; 52.0 ; 42.9 ; 28.4 ; 27.2 ; 19.2 . \mathrm{MS}: 280\left(14, M^{+}\right), 113$ (18), 112 (19), 110 (50), 109 (19), 84 (15), 69 (100), 68 (37). Anal. calc. for $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{O}_{4} \mathrm{~S}$ (280.34): C 59.98, H 5.75, S 11.44; found: C 59.89, H 5.73, S 11.48.
2.3. Isopropylidene 2-(Phenylthio)butane-1,1-dicarboxylate ( $=2,2$-Dimethyl-5-[1-(phenylthio, propyl]-1,3-dioxane-4,6-dione; 2). Meldrum's acid ( $2.00 \mathrm{~g}, 13.9 \mathrm{mmol}$ ), thiophenol, and propionaldehyde in 20 ml of MeCN : $3.61 \mathrm{~g}\left(88 \%\right.$ ) of 2. M.p. (AcOEt/hexane) $103-104^{\circ}$. IR: $1778 \mathrm{~m}, 1748 \mathrm{~s}, 1386 \mathrm{~m}, 1322 \mathrm{~m}, 1292 \mathrm{~m}, 1209 \mathrm{~m}, 1181 \mathrm{~m}$, $1067 \mathrm{~m}, 882 \mathrm{~m}, 751 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.55-7.15(\mathrm{~m}, 5 \operatorname{arom} . \mathrm{H}) ; 3.88(d d d, J=2.3,5.8,8.2, \mathrm{H}-\mathrm{C}(2)) ; 3.84(d$, $J=2.3, \mathrm{H}-\mathrm{C}(1)) ; 2.18-2.08(m, 1 \mathrm{H}) ; 1.97-1.87(m, 1 \mathrm{H}) ; 1.78(s, \mathrm{Me}) ; 1.75(s, \mathrm{Me}) ; 1.08(t, J=7.3,3 \mathrm{H}-\mathrm{C}(4))$; olefin (+thiophenol): $7.91(t, J=7.5, \mathrm{H}-\mathrm{C}(2)) ; 2.95(d q, J=7.5,7.5,2 \mathbf{H}-\mathrm{C}(3)) ; 1.75(s, \mathrm{Me}) ; 1.21(t, J=7.5,3$ H-C(4)); ratio 2.5:1. ${ }^{13} \mathrm{C}-\mathrm{NMR}: 164.1 ; 164.0 ; 135.8 ; 131.8 ; 129.2 ; 127.5 ; 105.3 ; 51.1 ; 50.3 ; 28.4 ; 27.6 ; 27.4 ; 12.9$. MS: 294 (18, $M^{+}$), 127 (15), 110 (54), 109 (29), 108 (28), 83 (100), 66 (20). Anal. calc. for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{O}_{4} \mathrm{~S}$ (294.37): C 61.20, H 6.16, S 10.89 ; found: C 61.09, H 6.16, S 10.83 .

[^2]2.4. Isopropylidene 2-(Phenylthio)pentane-1,1-dicarboxylate ( $=2,2$-Dimethyl-5-[1-(phenylthio)butyl]-1,3-dioxane-4,6-dione; 3). Meldrum's acid ( $10.00 \mathrm{~g}, 69.4 \mathrm{mmol}$ ), thiophenol, and butyraldehyde in 20 ml of MeCN : $1970 \mathrm{~g}(92 \%)$ of 3. M.p. (AcOEt/hexane) $83-84^{\circ}$. IR: $1784 \mathrm{~s}, 1745 \mathrm{~s}, 1580 \mathrm{~m}, 1481 \mathrm{~s}, 1462 \mathrm{~m}, 1438 \mathrm{~s}, 1339 \mathrm{~m}, 1330 \mathrm{~s}$, $1306 s, 1272 s, 1215 s, 1210 s, 1055 s, 1004 m, 983 m, 738 m, 728 m, 690 m .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.53-7.13(\mathrm{~m}, 5 \mathrm{arom} . \mathrm{H})$; $3.98(d d d, J=2.3,5.4,7.8, \mathrm{H}-\mathrm{C}(2)) ; 3.83(d, J=2.3, \mathrm{H}-\mathrm{C}(1)) ; 2.17-2.03(m, 1 \mathrm{H}) ; 1.87-1.73(\mathrm{~m}, 1 \mathrm{H}) ; 1.78(\mathrm{~s}$, $\mathrm{Me}) ; 1.74(\mathrm{~s}, \mathrm{Me}) ; 1.79-1.60(\mathrm{~m}, \mathrm{l} \mathrm{H}) ; 1.48-1.36(\mathrm{~m}, 1 \mathrm{H}) ; 0.91(t, J=7.4,3 \mathrm{H}-\mathrm{C}(5))$; olefin 7 ( + thiophenol): see 3.1; ratio 1:1. ${ }^{13}$ C-NMR: 163.9 (2); 135.7; 131.7; 129.1; 127.4; 105.2; $51.3 ; 48.1 ; 36.3 ; 28.3 ; 27.3 ; 21.2 ; 13.6$. MS: $308\left(4, M^{+}\right), 141(22), 140(16), 123(15), 122(62), 110(100), 109(28), 97(23), 94(34), 84(21), 68(48), 66$ (54). Anal. calc. for $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{O}_{4} \mathrm{~S}(380.40)$ : C 62.31, H 6.54, S 10.40; found: C 62.23, H 6.62, S 10.45.
2.5. Isopropylidene 3-Methyl-2-(phenylthio)butane-1,1-dicarboxylate ( $=2,2$-Dimethyl-5-(2-methyl-I-(phenylthio)propyl]-1,3-dioxane-4,6-dione; 4). Meldrum's acid ( $2.00 \mathrm{~g}, 13.9 \mathrm{mmol}$ ), thiophenol, and isobutyraldehyde in 5 ml of $\mathrm{MeCN}: 1.93 \mathrm{~g}\left(90 \%\right.$ ) of 4. M.p. (AcOEt/hexane) $81-82^{\circ}$. IR: 2974m, $1783 \mathrm{~s}, 1742 \mathrm{~s}, 1483 \mathrm{~m}, 1440 \mathrm{~m}$, $1392 \mathrm{~m}, 1384 \mathrm{~m}, 1371 \mathrm{~m}, 1324 \mathrm{~s}, 1289 \mathrm{~s}, 1219 \mathrm{~m}, 1203 \mathrm{~s}, 1064 \mathrm{~m}, 886 \mathrm{~m}, 741 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.51-7.22(\mathrm{~m}, 5 \mathrm{arom}$. $\mathrm{H}) ; 3.95(d d, J=1.7,10.6, \mathrm{H}-\mathrm{C}(2)) ; 3.73(d, J=1.7, \mathrm{H}-\mathrm{C}(1)) ; 2.62-2.46(m, \mathrm{H}-\mathrm{C}(3)) ; 1.80(s, \mathrm{Me}) ; 1.74(s, \mathrm{Me}) ;$ $1.18(d, J=6.6, \mathrm{Me}-\mathrm{C}(3)) ; 1.02(d, J=6.7, \mathrm{Me}-\mathrm{C}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}: 164.8 ; 164.1 ; 136.3 ; 131.1 ; 129.1 ; 127.2 ; 105.3$; $55.6 ; 49.7 ; 32.6 ; 28.4 ; 27.7 ; 22.0 ; 21.3$. MS: $308\left(3, M^{+}\right), 123$ (18), 122 (32), $110(47), 109$ (67), 97 (67), 69 (22), 66 (35), 43 (100). Anal. calc. for $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{O}_{4} \mathrm{~S}(308.40)$ : C 62.31, H 6.54, S 10.40 ; found: C $62.35, \mathrm{H} 6.61, \mathrm{~S} 10.32$.
2.6. Isopropylidene 2-(Phenylthio)octane-1,1-dicarboxylate ( $=2,2$-Dimethyl-5-[1-(phenylthio)heptyl]-1,3-dioxane-4,6-dione; 5). Meldrum's acid ( $1.00 \mathrm{~g}, 6.9 \mathrm{mmol}$ ), thiophenol, and heptanal in 10 ml of $\mathrm{MeCN}: 2.14 \mathrm{~g}$ ( $88 \%$ ) of 5. M.p. (AcOEt/hexane) $96-97^{\circ}$. IR: 2925s, $1785 m, 1747 s, 1482 m, 1397 m, 1388 m, 1338 s, 1330 s, 1310 m$, $1272 \mathrm{~m}, 1219 \mathrm{~m}, 1203 \mathrm{~s}, 1066 \mathrm{~m}, 996 \mathrm{~m}, 982 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) ; 7.50-7.21(\mathrm{~m}, 5$ arom H$) ; 3.96(d d d, J=2.4,5.6$, $9.5, \mathrm{H}-\mathrm{C}(2)) ; 3.82(d, J=2.4, \mathrm{H}-\mathrm{C}(1)) ; 2.15-2.05(m, 1 \mathrm{H}) ; 1.88-1.78(m, 1 \mathrm{H}) ; 1.78(s, \mathrm{Me}) ; 1.74(s, \mathrm{Me})$; $1.67-1.55(m, 1 \mathrm{H}) ; 1.43-1.18(m, 7 \mathrm{H}) ; 0.87(t, J=7.5,3 \mathrm{H}-\mathrm{C}(8)) .{ }^{13} \mathrm{C}-\mathrm{NMR}: 164.0(2) ; 135.7 ; 131.8 ; 129.2 ; 127.4$; $105.2 ; 51.3 ; 48.4 ; 34.2 ; 31.6 ; 28.8 ; 28.3 ; 28.0 ; 27.4 ; 22.5 ; 14.0$. MS: $350\left(2, M^{+}\right), 194(9), 164(19), 139(37), 110(80)$, $109(46), 108(37), 69(40), 68(54), 67(18), 66(37), 55(100)$. Anal. calc. for $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{~S}(350.48): \mathrm{C} 65.11, \mathrm{H} 7.48, \mathrm{~S}$ 9.15; found: C 65.23, H 7.42, S 9.22.
2.7. Isopropylidene 2-Phenyl-2-(phenylthio)ethane-1,1-dicarboxylate ( $=2,2$-Dimethyl-5-( $\alpha$-(phenylthio)-benzyl/-1,3-dioxane-4,6-dione; 6). Meldrum's acid ( $1.00 \mathrm{~g}, 6.9 \mathrm{mmol}$ ), thiophenol, and benzaldehyde in 10 ml of MeCN: 2.10 g ( $88 \%$ ) of 6. M.p. (AcOEt/hexane) $98^{\circ}$. IR: $3060 \mathrm{w}, 1789 \mathrm{~s}, 1740 \mathrm{~s}, 1583 \mathrm{~m}, 1496 \mathrm{~m}, 1438 \mathrm{~m}, 1393 \mathrm{~s}, 1386 \mathrm{~s}$, $1345 s, 1332 s, 1231 s, 1207 s, 1069 s, 738 s, 709 s .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.62-7.20(\mathrm{~m}, 2 \mathrm{Ph}) ; 5.21(d, J=2.7, \mathrm{H}-\mathrm{C}(2))$; $4.13(d, J=2.7, \mathrm{H}-\mathrm{C}(1)) ; 1.67(s, \mathrm{Me}) ; 1.43(s, \mathrm{Me}) .{ }^{13} \mathrm{C}-\mathrm{NMR}: 163.6 ; 163.5 ; 138.4 ; 135.4 ; 131.7 ; 129.1(2) ; 128.6$; $128.2 ; 127.7 ; 105.5 ; 53.2 ; 51.9 ; 28.2 ; 27.7$. MS: $342\left(0.3, M^{+}\right), 175(19), 174(55), 146$ (17), 110 (100), 109 (25), 102 (28), 84 (18), 66 (38). Anal. calc. for $\mathrm{C}_{19} \mathrm{H}_{18} \mathrm{O}_{4} \mathrm{~S}$ (342.41): C 66.65, H 5.30, S 9.36 ; found: C 66.69, H 5.21, S 9.43.
3. Transformations of 3. 3.1. Isopropylidene 1-Penten-1,1-dicarboxylate ( $=5$-Butylidene-2,2-dimethyl-1,3-di-oxane-4,6-dione; 7). A soln. of $\mathbf{3}(6.00 \mathrm{~g}, 19.5 \mathrm{mmol})$ in $\mathrm{MeCN}(15 \mathrm{ml})$ was shaken with an aq. KOH soln. $(2 \mathrm{~m}, 80$ $\mathrm{ml})$ in a separatory funnel. A cooled, aq. soln. of $\mathrm{K}_{3}\left[\mathrm{Fe}(\mathrm{CN})_{6}\right](7.20 \mathrm{~g}, 21.9 \mathrm{mmol}$, in 80 ml$)$ was added with shaking. After 2 washings with $\mathrm{Et}_{2} \mathrm{O}(100 \mathrm{ml})$, the aq. phase was added dropwise to a $10 \% \mathrm{HCl}$ soln. ( 150 ml ) at $5^{\circ}$. Extraction with $\mathrm{Et}_{2} \mathrm{O}$ ( 60 ml ), drying, and bulb-to-bulb distillation ( $100^{\circ} / 0.1$ Torr) yielded $3.24 \mathrm{~g}(84 \%$ ) of pure 7 as a colorless oil ([13]: m.p. 33-36 ${ }^{\circ}$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.89(t, J=7.5, \mathrm{H}-\mathrm{C}(2)) ; 2.88$ ( $\left.q^{\prime}, J=7.5,2 \mathrm{H}-\mathrm{C}(3)\right) ; 1.70$ $(s, 2 \mathrm{Me}) ; 1.66-1.52(\mathrm{~m}, 2 \mathrm{H}-\mathrm{C}(4)) ; 0.97(t, J=7.5,3 \mathrm{H}-\mathrm{C}(5))$.
3.2. Isopropylidene 1,2-Epoxypentane-1,1-dicarboxylate ( $=2,2$-Dimethyl-3'-propylspirol 1,3-dioxane-5, $2^{\prime}$ oxirane $-4,6$-dione; $\mathbf{8}) . \mathrm{H}_{2} \mathrm{O}_{2}(30 \%, 6 \mathrm{ml})$ was added at r.t. to a soln. of $3(3.00 \mathrm{~g}, 9.7 \mathrm{mmol})$ in $\mathrm{MeCN}(30 \mathrm{ml})$. After $30 \mathrm{~min}, \mathrm{Et}_{2} \mathrm{O}(50 \mathrm{ml})$ was added. The org. phase was separated and washed with brine ( 30 ml ). Drying, concentration under vacuum, and recrystallization from $\mathrm{Et}_{2} \mathrm{O} /$ hexane yielded $1.52 \mathrm{~g}(73 \%)$ of 8 as colorless needles. M.p. $60-61^{\circ}$. IR: $2957 \mathrm{w}, 1791 \mathrm{~s}, 1766 \mathrm{~s}, 1394 \mathrm{~m}, 1378 \mathrm{~m}, 1348 \mathrm{~m}, 1278 \mathrm{~m}, 1229 \mathrm{~m}, 1219 \mathrm{~m}, 1205 \mathrm{~m}, 1166 \mathrm{~m}, 937 \mathrm{~m}, 920 \mathrm{~m}$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 3.65(t, J=6.1, \mathrm{H}-\mathrm{C}(1)) ; 1.83(s, 2 \mathrm{Me}) ; 1.81-1.76(m, 2 \mathrm{H}) ; 1.70-1.52(m, 2 \mathrm{H}) ; 1.02(t$, $J=7.2,3 \mathrm{H}-\mathrm{C}(5)) .{ }^{13} \mathrm{C}-\mathrm{NMR}: 163.7 ; 161.8 ; 105.8 ; 68.4 ; 55.2 ; 28.7 ; 28.0 ; 27.7 ; 19.5 ; 13.7$. CI-MS: 233 ( 13 , $\left.M^{+}+19\right), 232\left(100, M^{+}+18\right), 136(4), 94(6), 76(10)$. Anal. calc. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{O}_{5}(214.22)$ : C $56.07, \mathrm{H} 6.59$; found: C 56.03, H 6.55.
3.3. 1,1:2,2-Diisopropylidene 3-Propylcyclopropane-1,1,2,2-tetracarboxylate ( $=2,2,2^{\prime \prime}, 2^{\prime \prime}$-Tetramethyl-3'-propyldispirol [ 1,3 -dioxane- $5,1^{\prime}$-cyclopropane- $2^{\prime}, 5^{\prime \prime}-1^{\prime \prime}, 3^{\prime \prime}$-dioxane $]-4,4^{\prime \prime}, 6,6^{\prime \prime}$-tetrone ; 9 ). A soln. of $\mathrm{NaIO}_{4}(3.00 \mathrm{~g}, 14.0$ mmol) in $\mathrm{H}_{2} \mathrm{O}(40 \mathrm{ml})$ was added dropwise at $5^{\circ}$ within 10 min to a stirred soln. of $3(4.0 \mathrm{~g}, 13.0 \mathrm{mmol})$ and Meldrum's acid ( $1.90 \mathrm{~g}, 13.2 \mathrm{mmol}$ ) in $\mathrm{MeCN}(40 \mathrm{ml})$. The mixture turned red and a precipitate formed. After a further $15 \mathrm{~min}, \mathrm{H}_{2} \mathrm{O}$ was added, and the solid was filtered and washed carefully with $\mathrm{H}_{2} \mathrm{O}(100 \mathrm{ml})$ and $\mathrm{Et}_{2} \mathrm{O}(100$ ml ). Drying under high vacuum yielded $4.01 \mathrm{~g}(91 \%)$ of 9 . The diphenyl disulfide was easily recovered out of the
org. washings. The recrystallized (AcOEt /hexane) product decomposed on heating at $\mathrm{ca} .191^{\circ}$. IR: $180 \mathrm{~lm}, 1764 \mathrm{~s}$, $1398 \mathrm{~m}, 1386 \mathrm{~m}, 1282 \mathrm{~s}, 1269 \mathrm{~s}, 1251 \mathrm{~m}, 1232 \mathrm{~m}, 1205 \mathrm{~s} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 3.23(t, J=7.8, \mathrm{H}-\mathrm{C}(3)) ; 2.05$ (' $q$ ', $J=7.6$ (the 2 center peaks show up as $t$ 's, $J=2.1$ ), $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2}$ ); $1.84(\mathrm{~s}, 2 \mathrm{Me}$ ); $1.77(\mathrm{~s}, 2 \mathrm{Me}$ ); 1.63 ('sext.', $J=7.5$ (the 2 center peaks show up as $t$ 's, $J=2.1$ ), $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2}\right) ; 1.03\left(t, J=7.5, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}-\mathrm{NMR}: 161.7 ; 159.2$; $106.1 ; 42.3 ; 40.7 ; 28.1 ; 26.9 ; 25.2 ; 21.3 ; 13.6$. CI-MS: 359 (18, $M^{+}+19$ ), 358 ( $100, M^{+}+18$ ), 136 (4), 93 (6). Anal. calc. for $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{O}_{8}$ (340.33): C 56.47 , H 5.92; found: C $56.26, \mathrm{H} 5.92$.
3.4. Isopropylidene Pentane-1,1-dicarboxylate ( $=5$-Butyl-2,2-dimethyl-1,3-dioxane-4,6-dione; 10). $\mathrm{NaBH}_{4}$ $(1.00 \mathrm{~g}, 26.4 \mathrm{mmol})$ was added in 3 portions to a soln. at $5^{\circ}$ of $\mathbf{3}(2.00 \mathrm{~g}, 6.5 \mathrm{mmol})$ in THF/EtOH $10: 1(20 \mathrm{ml})$. Upon adding the Ist portion of $\mathrm{NaBH}_{4}$, the mixture turned deeply yellow. Then, the color faded. After 1 h stirring, the suspension was acidified with $10 \% \mathrm{HCl}$ soln. and extracted with $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{ml})$. The org. phase was washed with brine ( 50 ml ), dried, and evaporated (drying under high vacuum) to yield $1.24 \mathrm{~g}(95 \%)$ of 10 . M.p. $59-61^{\circ}([5 \mathrm{~b}]$ : $58-60^{\circ}$ ).
3.5. Isopropylidene 2-(Nitromethyl)pentane-1,1-dicarboxylate ( $=1$-(Nitromethyl)butyl-2,2-dimethyl-1,3-di-oxane-2,6-dione; 11). $\mathrm{Bu}_{4} \mathrm{NOH}$ (Aldrich, 1 m in $\mathrm{MeOH} ; 14 \mathrm{ml}$ ) was added dropwise at $5^{\circ}$ to a soln. of 3 ( $2.00 \mathrm{~g}, 6.5$ $\mathrm{mmol})$ and nitromethane ( $1.00 \mathrm{~g}, 16.4 \mathrm{mmol}$ ) in THF ( 10 ml ). After 2 h at r.t., $10 \% \mathrm{HCl}$ soln. ( 50 ml ) was added. Extraction with $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{ml})$, washing of the org. phase with brine ( 50 ml ), drying, concentration, and recrystallization from $\mathrm{Et}_{2} \mathrm{O} /$ hexane yielded $1.41 \mathrm{~g}(84 \%)$ of 11 . M.p. $98-99^{\circ}$. IR: $2883 \mathrm{~m}, 1778 \mathrm{~s}, 1743 \mathrm{vs}, 1735 \mathrm{vs}, 1545 \mathrm{vs}, 1398 \mathrm{~m}$, $1387 \mathrm{~m}, 1358 \mathrm{~m}, 1328 \mathrm{~s}, 1293 \mathrm{~m}, 1241 \mathrm{~m}, 1206 \mathrm{~m}, 1065 \mathrm{~m}, 1011 \mathrm{~m}, 874 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 4.95(\mathrm{dd}, \mathrm{J}=8.3,11.2$, $1 \mathrm{H}-\mathrm{C}(3)) ; 4.56(d d, J=3.8,11.2,1 \mathrm{H}-\mathrm{C}(3)) ; 3.90(d, J=2.1, \mathrm{H}-\mathrm{C}(1)) ; 3.32-3.22(\mathrm{~m}, 14$ lines, $\mathrm{H}-\mathrm{C}(2)) ; 1.81(s$, $\mathrm{Me}) ; 1.78(s, \mathrm{Me}) ; 1.62-1.30(m, 4 \mathrm{H}) ; 0.94(t, J=7.2,3 \mathrm{H}-\mathrm{C}(5))$. ${ }^{13} \mathrm{C}-\mathrm{NMR}: 164.1$ (2); 105.5; 76.0; 47.0; 36.2; $31.5 ; 28.2 ; 26.8 ; 20.6 ; 13.7$. CI-MS: 278 ( $15, M^{+}+19$ ), 277 ( $100, M^{+}+18$ ), 219 (4), 192 (6). Anal. calc. for $\mathrm{C}_{11} \mathrm{H}_{17} \mathrm{NO}_{6}(259.26): \mathrm{C} 50.96, \mathrm{H} 6.61, \mathrm{~N} 5.40$; found: C $51.05, \mathrm{H} 6.82$, N 5.43.
3.6. Isopropylidene 3,4-Dimethyl-6-propy/cyclohex-3-ene-1,1-dicarboxylate ( $=2^{\prime}, 2^{\prime}, 3,4$-Tetramethyl-6-propylspiro [ 3 -cyclohexene-1, $5^{\prime}-I^{\prime}, 3^{\prime}$-dioxane $/-4^{\prime}, 6^{\prime}$-dione; 12). A soln. of 2,3 -dimethyl-1,3-butadiene ( $1.50 \mathrm{~g}, 18.3 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{ml})$ was added, at $5^{\circ}$, to a soln. of $3(2.00 \mathrm{~g}, 6.5 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{ml})$. After 6 h at r.t., the mixture was evaporated and recrystallized from $\mathrm{Et}_{2} \mathrm{O} /$ hexane $\left(-15^{\circ}\right): 1.31 \mathrm{~g}(72 \%)$ of 12 . M.p. $97-99^{\circ}$. $1 \mathrm{R}: 2959 w, 2874 w$, $1783 \mathrm{~s}, 1744 \mathrm{~s}, 1396 \mathrm{~m}, 1388 \mathrm{~m}, 1340 \mathrm{~m}, 1330 \mathrm{~s}, 1305 \mathrm{~m}, 1272 \mathrm{~m}, 1214 \mathrm{~m}, 1202 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 2.70(d, J=17.2$, $\mathrm{H}-\mathrm{C}(2)) ; 2.35-2.22(m, 1 \mathrm{H}) ; 2.17(d, J=17.2, \mathrm{H}-\mathrm{C}(2)) ; 2.12-1.91(\mathrm{~m}, 2 \mathrm{H}) ; 1.67(s, 2 \mathrm{Me}) ; 1.62(\mathrm{~s}, \mathrm{Me}) ; 1.57(s$, $\mathrm{Me}) ; 1.46-1.34(\mathrm{~m}, 1 \mathrm{H}) ; 1.27-1.02(\mathrm{~m}, 2 \mathrm{H}) ; 0.85-0.77(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$-NMR: $171.9 ; 167.0 ; 125.4 ; 119.4 ; 104.2 ; 52.9$; $40.8 ; 38.9 ; 34.2 ; 33.7 ; 29.9 ; 28.1 ; 19.9 ; 18.8 ; 18.1 ; 13.6$ MS: $280\left(8, M^{+}\right), 222(14), 152(13), 151$ (100), $150(37), 149$ (40), 107 (91), 91 (39). Anal. calc. for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{4}(280.36$ ): C $68.55, \mathrm{H} 8.63$; found: C 68.68, H 8.55.
4. Synthesis of 3-15. 4.1. General Procedure. A soln. of Meldrum's acid (1 equiv.), aldehyde (1.0-1.3 equiv.), thiocarboxylic acid ( 1.05 equiv.), and piperidinium acetate ( 0.1 equiv.) in MeCN was stirred for $c a .20 \mathrm{~h}$ at r.t. The mixture was worked up as described for 1-6. In the case of 13 and 14, a precipitate, mainly diisopropylidene methylenedimalonate, formed after $c a .2 \mathrm{~h}$.
4.1. Isopropylidene 2-(Acetylthio)ethane-1,1-dicarboxylate ( $=5$ - ( (Acetylthio) methyl $]-2,2$-dimethyl-1,3-di-oxane-4,6-dione; 13). Meldrum's acid ( $10.00 \mathrm{~g}, 69.4 \mathrm{mmol}$ ), thioacetic acid (Aldrich, $90 \% ; 6.0 \mathrm{ml}, 75.6 \mathrm{mmol}$ ) and formalin (Aldrich, $37 \% ; 6.9 \mathrm{ml}, 92.1 \mathrm{mmol}$ ) in 10 ml of MeCN : 12.05 g ( $75 \%$ ) of 13. M.p. (AcOEt/hexane): $108-109^{\circ}$ (dec.). IR: $1785 m, 1743 \mathrm{~s}, 1679 \mathrm{~s}, 1398 \mathrm{~m}, 1389 \mathrm{~m}, 1362 \mathrm{~m}, 1341 \mathrm{~s}, 1291 \mathrm{~s}, 1275 \mathrm{~m}, 1205 \mathrm{~m}, 1096 \mathrm{~m}, 1065 \mathrm{~m}$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 3.95(t, J=5.5, \mathrm{H}-\mathrm{C}(1)) ; 3.51(d, J=5.5,2 \mathrm{H}-\mathrm{C}(2)) ; 2.32(s, \mathrm{AcS}) ; 1.82(s, \mathrm{Me}) ; 1.75(s, \mathrm{Me})$. ${ }^{13}$ C-NMR: $194.5 ; 163.8(2) ; 105.4 ; 47.1 ; 30.1 ; 28.3 ; 26.2 ; 24.4$. CI-MS: $174\left(40, M^{+}+18-\mathrm{AcSH}\right), 133$ (7), 116 (6), 76 (100). Anal. calc. for $\mathrm{C}_{9} \mathrm{H}_{12} \mathrm{O}_{5} \mathrm{~S}(232.25)$ : C 46.54, H 5.21, S 13.81 ; found: C 46.48, H 5.26, S 13.86 .
4.2. Isopropylidene 2-( Acetylthio)propane-1,1-dicarboxylate ( $=5-[1-($ Acetylthio) ethyl $]-2,2-d i m e t h y l-1,3-d i-$ oxane-4,6-dione; 14). Meldrum's acid ( $2.00 \mathrm{~g}, 13.9 \mathrm{mmol}$ ), thioacetic acid ( $1.2 \mathrm{ml}, 15.1 \mathrm{mmol}$ ), and acetaldehyde ( $0.9 \mathrm{ml}, 16.1 \mathrm{mmol}$ ) at $5^{\circ}$ in 10 ml of MeCN in a tightly stoppered flask: 3.21 g ( $94 \%$ ) of 14. M.p. (AcOEt $/$ hexane) 105. IR: $1777 \mathrm{~m}, 1737 \mathrm{~s}, 1684 \mathrm{~m}, 1389 \mathrm{~m}, 1338 \mathrm{~m}, 1320 \mathrm{~m}, 1067 \mathrm{~m}, 987 \mathrm{~m}, 882 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 4.50(\mathrm{dq}, J=2.9$, $7.1, \mathrm{H}-\mathrm{C}(2)) ; 3.94(d, J=2.9, \mathrm{H}-\mathrm{C}(1)) ; 2.29(s, \mathrm{AcS}) ; 1.78(s, \mathrm{Me}) ; 1.73(s, \mathrm{Me}) ; 1.39(d, J=7.1,3 \mathrm{H}-\mathrm{C}(3))$. ${ }^{13}$ C-NMR: 195.1; 163.3; 163.0; 105.2; 51.4; 35.3; 30.1; 28.3; 26.4; 17.2. CI-MS: 232 (7), 189 (9), 188 (100, $\left.M^{+}+18-\mathrm{AcSH}\right), 147(11), 76(82)$. Anal. calc. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{O}_{\mathrm{S}} \mathrm{S}(246.28): \mathrm{C} 48.77, \mathrm{H} 5.73, \mathrm{~S} 13.02$; found: C 48.84, H 5.87, S 13.00 .
4.3. Isopropylidene 2-(Benzoylthio)ethane-1,1-dicarboxylate ( $=5$-( (Benzoylthio)methyl]-2,2-dimethyl-1,3-dioxane-4,6-dione; 15). Meldrum's acid ( $3.00 \mathrm{~g}, 20.8 \mathrm{mmol}$ ), thiobenzoic acid ( $2.5 \mathrm{ml}, 21.2 \mathrm{mmol}$ ), and formalin (Aldrich, $37 \%$; $1.50 \mathrm{ml}, 20.0 \mathrm{mmol}$ ): $4.16 \mathrm{~g}\left(68 \%\right.$ ) of 15 , after recrystallization from EtOAc/hexane ${ }^{6}$ ). M.p. $110-111^{\circ}$ (dec.). IR : $1790 \mathrm{~m}, 1751 \mathrm{~s}, 1671 \mathrm{~m}, 1665 \mathrm{~m}, 1385 \mathrm{~m}, 1332 \mathrm{~m}, 1271 \mathrm{~m}, 1207 \mathrm{~m}, 1177 \mathrm{~m}, 1066 \mathrm{~m}, 928 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}$

[^3]$\left(\mathrm{CDCl}_{3}\right): 7.96(d, J=7.9,2 \operatorname{arom} . \mathrm{H}) ; 7.58(t, J=7.9,1$ arom. H); $7.45(t, J=7.9,2$ arom. H$) ; 3.99(t, J=5.6$, $\mathrm{H}-\mathrm{C}(1)) ; 3.75(d, J=5.6,2 \mathrm{H}-\mathrm{C}(2)) ; 1.83(s, \mathrm{Me}) ; 1.78(s, \mathrm{Me})$. CI-MS: $312\left(6, M^{+}+18\right), 228(13), 191(27), 174$ (100), 173 (23), 156 (45), 139 (64), 136 (20), 76 (21). Anal. calc. for $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{O}_{5} \mathrm{~S}$ (294.33): C 57.13, H 4.79, S 10.89; found: C 57.17, H 4.83, S 10.91.
5. Transformations of 13. 5.1. Isopropylidene 3,4-Dimethylcyclohex-3-ene-1,1-dicarboxylate $\left(=2^{\prime}, 2^{\prime}, 3,4-\right.$ Tetramethylspiro 3 -cyclohexene-1, $5^{\prime}-1^{\prime}, 3^{\prime}$-dioxane $1-4^{\prime}, 6^{\prime}$-dione; 16). At r.t., 2,3-dimethyl-1,3-butadiene ( $2.00 \mathrm{~g}, 24.4$ $\mathrm{mmol})$ was added to a soln. of $13(2.00 \mathrm{~g}, 8.6 \mathrm{mmol})$ in DMSO $(3 \mathrm{ml})$. After 16 h , the mixture was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{ml})$ and washed with $\mathrm{H}_{2} \mathrm{O}$. Drying and recrystallization from $\mathrm{Et}_{2} \mathrm{O}$ /hexane gave $0.99 \mathrm{~g}(48 \%)$ of 16 . M.p. $88^{\circ}$ ([3a]: 87-88 $)$.
5.2. Isopropylidene 2-Acetoxycyclohex-3-ene-1,1-dicarboxylate ( $=2$-Acetoxy- $2^{\prime}, 2^{\prime}$-dimethylspirof 3 -cyclo-hexene-1, $5^{\prime}-I^{\prime}, 3^{\prime}$-dioxane $]-4^{\prime}, 6^{\prime}$-dione; 17). A mixture of $13(2.0 \mathrm{~g}, 8.6 \mathrm{mmol})$ and 1,3 -butadienyl acetate [20] ( 1.90 g , $17.1 \mathrm{mmol})$ in DMSO $(3 \mathrm{ml})$ was stirred overnight. The mixture was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{ml})$ and twice washed with brine ( 40 ml ). Drying, evaporation of the solvent, trituration of the resulting oil with $\mathrm{Et}_{2} \mathrm{O}$, and filtering yielded $1.28 \mathrm{~g}(55 \%)$ of 17 as a single isomer. M.p. (AcOEt/hexane; premelted sample) $156-157^{\circ}$. IR: $1787 \mathrm{~m}, 1751 \mathrm{~s}$, $1396 m, 1385 m, 1321 m, 1302 m, 1274 m, 1237 s, 1200 s, 1031 m, 937 m$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 5.92-5.83(m, \mathrm{H}-\mathrm{C}(2)$, $\mathrm{H}-\mathrm{C}(3)) ; 5.60-5.53(m, \mathrm{H}-\mathrm{C}(4)) ; 2.32-2.08(m, 4 \mathrm{H}) ; 1.93(s, \mathrm{AcO}) ; 1.65(s, 2 \mathrm{Me}) .{ }^{13} \mathrm{C}-\mathrm{NMR}: 169.3 ; 169.1 ; 163.8 ;$ 129.4; 123.4; 104.9; 71.9; 51.3; 30.9; 28.8; 28.5; 21.7; 20.5. CI-MS: 287 (14, $M^{+}+19$ ), 286 ( $100, M^{+}+18$ ), 228 (15), 184 (40), 141 (54), 124 (47), 107 (69). Anal. calc. for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{6}$ (268.27): C $58.20, \mathrm{H} 6.01$; found: C $58.02, \mathrm{H}$ 6.06 .
5.3. Isopropylidene 4-Oxopentane-1,1-dicarboxylate ( $=2,2$-Dimethyl-5-(3-oxobutyl)-1,3-dioxane-4,6-dione; 18). To a soln. of $13(2.00 \mathrm{~g}, 8.6 \mathrm{mmol})$ in $\mathrm{MeCN}(20 \mathrm{ml}), 2$-methoxypropene ( $2.0 \mathrm{~g}, 27.7 \mathrm{mmol}$ ) and finely powdered anh. $\mathrm{K}_{2} \mathrm{CO}_{3}(1.50 \mathrm{~g}, 10.7 \mathrm{mmol})$ were simultaneously added. After efficient stirring for $c a .1 \mathrm{~h}$ (TLC), the excess enol ether was removed under vacuum and the suspension acidified by dropwise addition of $10 \% \mathrm{aq} . \mathrm{HCl}$ soln. ( 10 ml ). The product was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{ml})$, the org. phase dried, filtered, and evaporated. Trituration of the resulting oil with $\mathrm{Et}_{2} \mathrm{O}$ /hexane $1: 2$ and filtering gave $1.37 \mathrm{~g}(74 \%)$ of 18 as colorless leaflets. M.p. (AcOEt/hexane) $118^{\circ}$. IR: $1787 \mathrm{~s}, 1749 \mathrm{~s}, 1709 \mathrm{~s}, 1384 \mathrm{~s}, 1354 m, 1300 \mathrm{~s}, 1227 \mathrm{~m}, 1204 \mathrm{~m}, 1165 \mathrm{~m}, 1050 \mathrm{~m}, 1011 \mathrm{~m}, 987 \mathrm{~m}$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 3.84(t, J=5.6, \mathrm{H}-\mathrm{C}(1)) ; 2.70(t, J=7.1,2 \mathrm{H}-\mathrm{C}(3)) ; 2.23\left(q^{\prime}, J=6.7,2 \mathrm{H}-\mathrm{C}(2)\right) ; 2.10(s, 3$ $\mathrm{H}-\mathrm{C}(5)$ ); 1.76 ( $s, \mathrm{Me}$ ); 1.71 ( $s, \mathrm{Me}$ ). ${ }^{13} \mathrm{C}-\mathrm{NMR}: 207.7$; $165.1 ; 104.9 ; 44.6 ; 39.2 ; 29.8 ; 28.4 ; 26.2 ; 20.0$. CI-MS: 233 $\left(12, M^{+}+19\right), 232\left(100, M^{+}+18\right), 147$ (5). Anal. calc. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{O}_{5}(214.22)$ : C $56.07, \mathrm{H} 6.59$; found: C 55.98 , H6.75.
5.4. Isopropylidene 2-(2-Oxocyclopentyl)ethane-1,1-dicarboxylate ( $=2,2$-Dimethyl-5- $/(2$-oxocyclopentyl)-methyll-I,3-dioxane-4,6-dione; 19). To a soln. of $13(2.00 \mathrm{~g}, 8.6 \mathrm{mmol})$ at $5^{\circ}$ in $\mathrm{MeCN}(20 \mathrm{ml})$, morpholinocyclopentanone enamine ( $3.00 \mathrm{~g}, 20.1 \mathrm{mmol}$ ) was added within 15 min . The mixture turned yellow and a precipitate formed. After stirring for further 15 min at $5^{\circ} 10 \%$ aq. HCl soln. ( 20 ml ) was added. The colorless clear soln. was stirred for 1 h at r.t. The product was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{ml})$. On standing, a further amount of product crystallized out of the aq. soln. Drying $\left(\mathrm{MgSO}_{4}\right)$, evaporation of the solvent, drying under high vacuum, and trituration with pentane yielded $1.75 \mathrm{~g}\left(85 \%\right.$ ) of 19 as colorless crystals. M.p. $143^{\circ}$ (AcOEt/hexane, dec.). IR: $1791 m, 1745 s, 1731 s, 1384 m, 1366 m, 1343 m, 1324 m, 1305 m, 1284 m, 1208 m, 1198 m .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 4.42(d d$, $J=3.4,7.8, \mathrm{H}-\mathrm{C}(\mathrm{l})) ; 2.65-2.52(\mathrm{~m}, 1 \mathrm{H}) ; 2.35-1.94(m, 7 \mathrm{H}) ; 1.81(s, \mathrm{Me}) ; 1.73(s, \mathrm{Me}) ; 1.63-1.48(m, 1 \mathrm{H})$. ${ }^{13}$ C-NMR: 214.0; 165.7; 165.4; 104.9; 44.2; 43.3; 38.4; 30.5; 28.5; 26.2 (2); 20.5. CI-MS: 259 (13, $\left.M^{+}+19\right), 258$ (100, $\left.M^{+}+18\right), 200(9), 173(8), 156$ (10). Anal. calc. for $\mathrm{C}_{12} \mathrm{H}_{16} \mathrm{O}_{5}$ (240.26): C 59.99, H 6.71; found: C 59.79, H 6.83 .
5.5. Isopropylidene 3-Nitropropane-1,1-dicarboxylate ( $=5$-(2-Nitroethyl)-2,2-dimethyl-1,3-dioxane-4,6dione; 20). $\mathrm{Bu}_{4} \mathrm{NOH}$ (Aldrich, 1 m in $\left.\mathrm{MeOH} ; 20 \mathrm{ml}\right)$ was added dropwise within 15 min at $5^{\circ}$ to a soln. of $\mathbf{1 3}(2.00 \mathrm{~g}$, $8.6 \mathrm{mmol})$ and nitromethane $(1.50 \mathrm{~g}, 24.6 \mathrm{mmol})$ in $\mathrm{MeCN}(10 \mathrm{ml})$. After stirring for 15 min at $5^{\circ}$, the mixture was acidified with $10 \% \mathrm{HCl}$ soln. ( 50 ml ). The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(40 \mathrm{ml})$, the org. phase concentrated and triturated with $\mathrm{H}_{2} \mathrm{O}$, and the resulting crystalline slurry filtered, washed sequentially with $\mathrm{H}_{2} \mathrm{O}$ and $\mathrm{Et}_{2} \mathrm{O}$ /pentane $1: 1$, and dried under high vacuum to yield $1.39 \mathrm{~g}(74 \%)$ of $\mathbf{2 0}$ as a colorless powder. M.p. $135-136^{\circ}$ (dec.). IR: $1785 m, 1734 s, 1555 s, 1388 m, 1375 m, 1359 m, 1302 m, 1201 m, 1071 m, 979 m, 878 m .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 4.71(t$, $J=7.1,2 \mathrm{H}-\mathrm{C}(3)) ; 4.52(t, J=5.0, \mathrm{H}-\mathrm{C}(1)) ; 2.53\left({ }^{\prime} q^{\prime}, J=6.3,2 \mathrm{H}-\mathrm{C}(2)\right) ; 1.80(s, \mathrm{Me}) ; 1.67(s, \mathrm{Me}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ : 165.3; $105.2 ; 72.7 ; 43.5 ; 28.0 ; 25.6 ; 22.8$. CI-MS: $236\left(10, M^{+}+19\right), 235\left(100, M^{+}+18\right)$. Anal. calc. for $\mathrm{C}_{8} \mathrm{H}_{11} \mathrm{NO}_{6}$ (217.18): C 44.24, H 5.11, N 6.45; found: C 44.36, H 5.25 , N 6.43.
5.6. Isopropylidene 3-Nitrobutane-1,1-dicarboxylate $\quad(=5$-(2-Nitropropyl)-2,2-dimethyl-1,3-dioxane-4,6dione; 21). As described for 20 using nitroethane ( $1.65 \mathrm{~g}, 22.0 \mathrm{mmol}$ ): $1.42 \mathrm{~g}(71 \%)$ of $\mathbf{2 1}$. M.p. $132^{\circ}$ (dec.). IR: $1776 m, 1740 s, 1542 \mathrm{~s}, 1397 \mathrm{~m}, 1388 \mathrm{~m}, 1203 \mathrm{~m}, 1063 \mathrm{~m}, 991 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CD}_{3} \mathrm{CN}\right): 4.92-4.80(\mathrm{~m}, 14$ lines, $\mathrm{H}-\mathrm{C}(3))$; $3.73(d d, J=4.5,6.9, \mathrm{H}-\mathrm{C}(1)) ; 2.59(d d d, J=4.5,9.0,15.0,1 \mathrm{H}-\mathrm{C}(2)) ; 2.18(d d d, J=4.5,6.9,15.0,1 \mathrm{H}-\mathrm{C}(2))$;
1.63 ( $s, \mathrm{Me}$ ) $; 1.56(s, \mathrm{Me}) ; 1.42(d, J=6.7,3 \mathrm{H}-\mathrm{C}(4)) .{ }^{13} \mathrm{C}-\mathrm{NMR}: 166.1 ; 165.9 ; 106.5 ; 81.4 ; 44.3 ; 31.6 ; 28.6 ; 26.4$; 19.8. CI-MS: $250\left(12, M^{+}+19\right), 249\left(100, M^{+}+18\right)$. Anal. calc. for $\mathrm{C}_{9} \mathrm{H}_{13} \mathrm{NO}_{6}(231.20)$ : C $46.75, \mathrm{H} 5.67, \mathrm{~N} 6.06$; found: C $46.75, \mathrm{H} 5.77$, N 6.08 .
5.7. 1,1:2,2-Diisopropylidene Cyclopropane-1,1,2,2-tetracarboxylate $\left(=2,2,2^{\prime \prime}, 2^{\prime \prime}\right.$-Tetramethyldispiro[1,3-di-oxane-5, $1^{\prime}$-cyclopropane- $2^{\prime}, 5^{\prime \prime}-1^{\prime \prime}, 3^{\prime \prime}$-dioxane $]-4,4^{\prime \prime}, 6,6^{\prime \prime}$-tetrone; 22). Treatment of $13(3.02 \mathrm{~g}, 13.0 \mathrm{mmol})$ in the presence of Meldrum's acid $(1.90 \mathrm{~g}, 13.2 \mathrm{mmol})$ with $\mathrm{NaIO}_{4}(2.90 \mathrm{~g}, 13.6 \mathrm{mmol})$ as described for 9 yielded 2.21 g ( $57 \%$ ) of 22, after washing with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{ml})$. The recrystallized ( MeCN ) 22 decomposed on heating at $206^{\circ}$. IR: $1804 m, 1758 \mathrm{~s}, 1398 \mathrm{~m}, 1284 \mathrm{~s}, 1253 \mathrm{~m}, 1209 \mathrm{~m}, 1197 \mathrm{~m}, 1058 \mathrm{~m}, 971 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CD}_{3} \mathrm{CN}\right): 2.75(\mathrm{~s}, 2 \mathrm{H}-\mathrm{C}(3))$; $1.67(s, 2 \mathrm{Me}) ; 1.60(s, 2 \mathrm{Me}) .{ }^{13} \mathrm{C}$-NMR: 161.8; 108.0; 40.5; 27.8; 27.4; 27.1. MS: $317\left(15, M^{+}+19\right), 316$ ( 100 , $\left.M^{+}+18\right), 93$ (7). Anal. calc. for $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{O}_{8}$ (298.25): C 52.35, H 4.73; found: C 52.24, H 4.87.
5.8. Isopropylidene 2-Morpholinoethane-1,1-dicarboxylate ( $=2,2$-Dimethyl-5-(morpholinomethyl)-1,3-dio-xan-4,6-dione; 23). Morpholine ( $1.74 \mathrm{~g}, 20.0 \mathrm{mmol}$ ) was added to a soln. of $13(2.00 \mathrm{~g}, 8.6 \mathrm{mmol})$ at $5^{\circ}$ in $\mathrm{MeCN}(10$ ml ). After 30 min , the crystals were filtered and washed sequentially with $\mathrm{MeCN}, \mathrm{Et}_{2} \mathrm{O}$, and pentane to yield 1.72 g ( $82 \%$ ) of 23. M.p. $132-133^{\circ}$ (dec.). IR: $1691 w, 1593 s, 1525 w, 1456 w, 1408 m, 1388 m, 1374 m, 1259 m, 1123 m, 935 m$, $909 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 10.16$ (br. $\left.s, \mathrm{NH}\right) ; 4.08-3.94(\mathrm{~m}, 4 \mathrm{H}) ; 3.87$ ( $\left.{ }^{\prime} d^{\prime}, J=3.6,2 \mathrm{H}\right) ; 3.37$ ( $d$ ' $, J=12.5,2 \mathrm{H}$ ); 3.05-2.91 ( $\mathrm{m}, 2 \mathrm{H}$ ) ; $1.60(\mathrm{~s}, 2 \mathrm{Me}) .{ }^{13} \mathrm{C}-\mathrm{NMR}: 167.6 ; 102.1 ; 66.8 ; 63.6 ; 55.4 ; 51.3 ; 26.1$. CI-MS: $244\left(0.3, M^{+}+1\right)$, 105 (31), 88 (100). Anal. calc. for $\mathrm{C}_{11} \mathrm{H}_{17} \mathrm{NO}_{5}$ (243.26): C 54.31, H 7.04, N 5.76 ; found: C $54.16, \mathrm{H} 7.08, \mathrm{~N} 5.77$.

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[^0]:    ${ }^{1}$ ) On reacting Meldrum's acid with butyraldehyde in MeCN in the presence of catalytical amounts of piperidine, we obtained a $52 \%$ yield of the dimer ii (m.p. $117-118^{\circ}$ ), probably the same compound that was believed to be the Michael adduct iii [13]. Reduction of dimer ii with $\mathrm{NaBH}_{4}$ [5c] gave the corresponding alkane (m.p. 65-67 ; dec.).
    ${ }^{2}$ ) E.g., with THF or $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ as solvent and proline or hydrazine as catalyst.
    ${ }^{3}$ ) The same phenomenon has been observed in the case of adducts of cyclic tertiary amines [17].

[^1]:    ${ }^{4}$ ) The Michael adduct of Meldrum's acid to olefin 7 could not be cyclized with sodium metaperiodate in the presence of thiophenol.

[^2]:    ${ }^{5}$ ) An excess of formaldehyde was required to assure complete conversion of vii to $\mathbf{1 3}$ (and viii).

[^3]:    ${ }^{6}$ ) The crude product contained diisopropylidene methylenedimalonate.

