

## Cadmium Concentrations in Recent Snow and Firn Layers in the Canadian Arctic

J. O. Nriagu,\* G. S. Lawson, D. J. Gregor

Environment Canada, National Water Research Institute, Box 5050, Burlington, Ontario L7R 4A6, Canada

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It is becoming increasingly evident that long-range transported pollutants can have a devastating effect on the fragile ecosystems of the Canadian Arctic region (Muir et al., 1992; Thomas et al., 1992). The phenomenon of Arctic haze which prevails during the late winter-early spring period results in the flux of elevated quantities of toxic pollutants to the region (Nriagu et al., 1991; Landsberger et al., 1992). Although there have been studies of the atmospheric concentrations of pollutants in the region (Gregor, 1991; Barrie et al., 1992), there are no reliable measurements of the accumulation of trace metals in the snow and ice layers. Analysis of dated snow and ice layers in Greenland has furnished a wealth of information on the history of global air pollution with toxic metals (Wolff and Peel, 1988). Similar studies in Arctic Canada should provide additional insights into the hemispheric transport processes and the sources of trace metals in this polar region. This report presents the first set of data on the concentrations of cadmium in recent snow layers (deposited during the past 7 years) of the Agassiz Ice Cap, Ellesmere Island, Canada. The data were obtained using an ultra-clean sampling and analysis procedures.

## MATERIALS AND METHODS

The Agassiz Ice Cap is located in north-central Ellesmere Island, Northwest Territories, Canada. The sampling site was approximately 1800 m above sea level. Features that dictated the choice of this site included: (a) snow accumulation represents about 75% of the average annual precipitation; (b) summer melting affects only 3% of the winter snow layer so that the redistribution of material between adjacent snow layers is negligible; (c) individual annual conductivity profiles are preserved in the historical records implying that the blurring of seasonal stratification by snow drift and seasonal melt is minimal (see Barrie et al., 1985; Gregor, 1991).

The sample pit was located upwind and approximately 1.7 Km NW from the base camp (80°40'N 73°30'W). The pit, about one square meter, was dug in full clean-room coveralls using a stainless steel shovel with a fibreglass handle. The digging was accomplished over two days (2-3 hrs/day). Any snow removed was discarded downwind away from the pit. All sampling equipment, except the

<sup>\*</sup>Present address: Department of Environmental & Industrial Health, School of Public Health, University of Michigan, Ann Arbor, MI 48109

Correspondence to: J. O. Nriagu

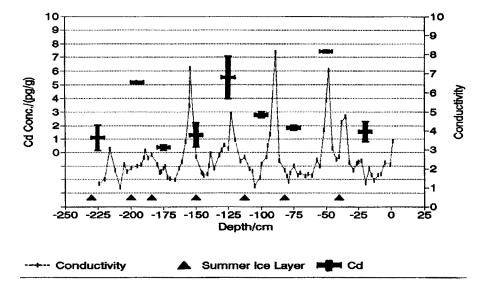


Figure 1. Conductivity and cadmium concentrations in the snow and firn layers deposited during the past 7 years at the Agassiz Ice Cap. Figure 1.

shovel, consisted of plastics which were acid-leached. Sampling down the side of the pit was accomplished using a Teflon coring tube. The tube corer was pressed into a freshly scraped surface (using a Teflon scraper). The sample was broken up in the corer by squeezing its sides and the sample was then poured into low density linear polyethylene (LDPE) 1 L wide-mouth bottles. The samples were double bagged then returned to the laboratory for acidification to pH 3 or pH 1.5 and analysis. All plastic equipment, including corer bottles, scrapers, and sample bottles were rigorously cleaned in successive acid baths as described previously (Nriagu et al., 1993).

All the sample handling and analysis were done in a Class 100 laboratory dedicated to research on trace metals. The cadmium content of each sample was determined by multiple injection of 10x30 uL samples into a graphite furnace AA instrument with Zeeman background correction (Wolff and Peel, 1988; Gorlach and Boutron, 1989). Simple field blanks, in triplicate, were obtained to determine possible changes in samples due to transport and storage. The measurements were below the detection limit of the instrumental technique for Cd (approx. 0.3 pg/g). Sampling blanks for the coring procedure were not obtained so that blank correction due to procedural contamination has not been applied to the data being presented here. Replicates, however, were taken to test the variability between samples. These are shown as vertical error bars in Figure 1. The horizontal error bars in the figures represent the diameter of the coring bottles (10 cm) since the concentrations are in fact mean values for the 10 cm time periods.

## RESULTS AND DISCUSSION

Barrie et al (1985) observed a strong seasonality in the conductivity and acidity in the snow layers at the Agassiz ice cap, which was closely correlated with seasonal fluctuations in air pollution in the region. Peaks in the conductivity and acidity values during the "Arctic winter" (late winter-early spring) were attributed to the deposition of pollutants in the Arctic haze. Strong seasonal fluctuations in conductivity are also observed in this study and have, in fact, been used to date the samples (Figure 1). Although there are marked differences in the year-to-year peak values, the conductivity during the off-peak periods remains low and shows no definitive trend during the last seven years.

Since continuous whole-year samples were not analyzed, it is impossible to define the detailed profile of cadmium concentration down the snow layers (Figure 1). Nevertheless, the Cd concentrations do indicate seasonal variations with peak values during the late winter-early spring period (Figure 1). The Cd concentrations during the peak and off-peak periods average 6.1 and 1.2 ng/kg respectively; the difference between the background and peak Cd concentrations is nearly 6-fold. Late winter-early spring maxima likewise have been found in the concentrations of airborne trace metals at various locations in Arctic Canada (Barrie and Hoff, 1985; Cheng et al., 1991). The record of Cd deposition preserved in the snow and ice layers thus seems consistent with the general features of atmospheric air pollution in the region.

We are not aware of any published data which show comparable seasonal variations in the Cd content of recently deposited snow layers in the Arctic polar region. Wolff and Peel (1988) could not find any seasonal difference in the Cd concentrations in the snow layers of southern Greenland. The off-peak (background) Cd levels at Agassiz are similar to the 1.1-1.6 ng/kg in surface snow samples collected in May 1987 at the Summit Station (elevation ~3230 m) in central Greenland (Gorlach and Boutron, 1989). The average annual Cd concentration in recent snow at the Dye 3 Station (elevation ~2480 m) in Greenland was found to be 0.74 ng/kg (Wolff and Peel, 1988). The higher Cd concentrations (2.6-6.0 ng/kg) in the accumulated snow in the Greenland-Spitsbergen region reported by Mart (1983) may be related to the fact that the sampling stations were at sea level. In view of the differences in elevation and location in relation to the sources of pollutants in the Arctic haze (Barrie et al., 1992), it is rather surprising that the Cd concentrations in deposited snow are similar in Greenland and the Ellesmere Island.

The average annual Cd concentration in recently deposited snow in the Antarctic Peninsula is only 0.07 ng/kg (Suttie and Wolff, 1991). By comparison with the data at our study site, it would appear that a large fraction of the background Cd burden in the Agassiz area is probably derived from anthropogenic sources. Data on other trace metals will be required to resolve the question of Canadian (local) versus transboundary sources of metal pollution at the Agassiz ice cap.

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