### Acid-base Catalysis in the Mechanism of Thioredoxin Reductase from Drosophila melanogaster

by

**Hsin-Hung Huang** 

A dissertation submitted in partial fulfillment of the requirements for the degree of Doctor of Philosophy (Toxicology) in The University of Michigan 2008

Doctoral committee:

Professor David P. Ballou, Co-Chair Professor Craig Harris, Co-Chair Professor Charles H. Williams, Jr. Professor Martin A. Philbert Associate Professor Ursula H. Jakob To my Mom, Mrs. Hui-Li Tsai, and Dad Mr. Yao-Hsin Huang for their enormous love

Also to my Wife, Pei-Pei Chiang and my Daughter, Chelsea Huang for their encouragement and support

### Acknowledgements

It was really fun to learn enzymology and to conduct my graduate research in the Ballou and Williams' lab. I am so fortunate to receive assistance from many persons in the past four and a half years.

First and foremost, I must show my great gratitude to my two mentors: Drs. David P. Ballou and Charles H. Williams, Jr. I appreciate having the opportunity to work in their laboratory and am grateful for having their training, which has made me become a biochemist. The work of this thesis would have been impossible without their encouragement and guidance. I also thank my doctoral committee members: Drs. Craig Harris, Ursula H. Jakob and Martin A. Philbert; they gave me so many useful suggestions.

I feel grateful to Mr. L. David Arscott: he taught me the knowledge of enzyme kinetics and gave me tremendous support. In addition, I have to thank two former lab members, Mrs. Sailaja Pullela and Dr. Zhiyong Cheng. Sailaja helped me prepare enzymes, which was really helpful to expedite my experiments; Dr. Cheng contributed several useful suggestions for my work. My thanks also go to Drs. Sumita Chakraborty, Mike Tarasev, and Tatyana Spolitak. They created a friendly lab environment and gave me a lot of help. My sincere thanks go to my grandmother, parents, sisters and parents-in law for giving me strong support. Five and a half years ago, I came to Ann Arbor alone. Right now, I have my family here. It is quite amazing that we are married in Ann Arbor

and were together to welcome our daughter, Chelsea. I have to acknowledge my wife, Pei-Pei because of her company and encouragement. She gave me warmth when I needed. Chelsea let me feel blessed using her wonderful smile.

### Preface

This dissertation is composed of four chapters. The first chapter contains an introduction to the physiological functions of the thioredoxin (Trx) system, the various intracellular antioxidant systems, and the roles of the Trx system in toxicology. The Trx system is made up of Trx and thioredoxin reductase (TrxR) and comprises the major antioxidant system in dipteran insects such as Drosophila melanogaster and Anopheles gambiae, a vector of malarial parasites. Because TrxRs from D. melanogaster and A. gambiae are virtually identical, TrxR from D. melanogaster (DmTrxR) offers an excellent model for TrxRs from dipteran insects. The dithiol-disulfide interchange reaction is involved in the catalysis of DmTrxR. A dyad of His-464' and Glu-469' in DmTrxR is proposed to facilitate the formation of thiolate anion to initiate the interchange reaction. Thus, in the first chapter, the potential roles of His-464' and Glu-469' in DmTrxR are also discussed. The second chapter describes studies of the function of His-464' in the acid-base catalysis involved in DmTrxR. The results showed that this histidine residue is crucial to the catalysis of DmTrxR; it acts as the immediate base catalyst to facilitate the formation of thiolate anions and also stabilizes the thiolate anions by ionic interactions. His-464' is involved in both the reductive and oxidative halfreactions. The content of the second chapter is from a manuscript that is in press in The third chapter describes studies of the function of Glu-469' in acid-Biochemistry. base catalysis of DmTrxR. The results show that Glu-469' is important but not crucial;

the function of Glu-469' is to facilitate the proper positioning of His-464' toward the interchange thiol (Cys-57). The content of the third chapter will form the basis of a second manuscript when two further experiments have been completed. In the fourth chapter, the conclusions of this study and potential future work are addressed.

## **Tables of Contents**

Dedication	ii
Acknowledgements	iii
Preface	v
List of Figures	X
List of Tables	xii
List of Appendices	xiii
List of Abbreviations	xiv
Abstract	xvi
Chapter 1: Introduction: oxidative stress, cellular antioxidant systems, and the impo	ortance
of the thioredoxin system in cells	
A. The functions of thioredoxin and thioredoxin reductase: the thioredoxin system.	
1. The function of Trx in antioxidant systems	
2. Trx as an enzyme cofactor	
3. The function of Trx in the immune system	
4. Function of Trx in redox regulation of signal transduction	
5. The function of Trx in apoptosis	
B. Diseases in which Trx is thought to play a role.	
1. Cancer	
2. Rheumatoid arthritis and Sjögren syndrome	
3. Cardiovascular diseases	
4. Human immunodeficient virus infection	
5. Reperfusion injury	
C. The redox environment in cells	
1. Reactive oxygen species and oxidative stress	
2. Superoxide anion	
3. Hydrogen peroxide	
4. Hydroxyl radical	
5. Nitric oxide	
D. Roles of the thioredoxin-thioredoxin reductase system in toxicology	17
1. Malaria	
2. Thioredoxin reductase from dipteran insects	22
E. Biochemical properties of Trx and peroxiredoxin	
1. Mechanism of thioredoxin reactions	
2. Structure of thioredoxin	24
3. Peroxiredoxin	
F. Thioredoxin reductase (TrxR) and thioredoxin-glutathione reductase (TGR)	
1. Low molecular weight thioredoxin reductase	
2. High molecular weight thioredoxin reductase	

3. The catalytic mechanism and structural properties of DmTrxR	30
a. The function of the histidine residue in the catalytic mechanism	
b. The function of glutamate residue in the catalytic mechanism	36
Chapter 2: The function of His-464' in acid-base catalysis of thioredoxin reductas	se from
Drosophila melanogaster	
A. Abstract	
B. Introduction	
C. Materials and Methods	
1. Chemicals	60
2. DNA sequencing of wild-type DmTrxR-1 and the histidine mutant	60
3. Preparation of wild-type DmTrxR-1, the histidine mutant, and DmTrx-2	
4. Determination of the activities of DmTrxR and of the histidine variant	61
5. Steady-state kinetics of wild-type DmTrxR and the histidine variant over a	range
of pH.	
6. Titrations of wild-type DmTrxR and the histidine variant with NADPH	62
7. The kinetics of the reductive half-reactions of wild-type DmTrxR and the	
histidine variant using stopped-flow spectrophotometry	63
8. The kinetics of the oxidative half-reactions of wild-type DmTrxR and the	
histidine variant using stopped-flow spectrophotometry	63
D. Results and Discussion	
1. Effect of pH on the activity of wild-type and variant TrxR	
2. Titrations of wild-type and variant TrxR	
3. Effect of pH on the reductive half-reaction of wild-type and variant DmTrx	
4. Effect of pH on the oxidative half reactions of wild-type and variant TrxR.	
E. Concluding remarks	87
Chapter 3: The function of Glu-469' as an acid-base catalyst in thioredoxin reducta	ise
from Drosophila melanogaster	
A. Abstract	
B. Introduction	
C. Materials and Methods	
1. Chemicals	
2. Site-directed mutagenesis	
3. Preparation of the glutamate variants and DmTrx-2	
4. The determination of activities of the glutamate variant	
5. Steady-state kinetics of the glutamine variants over a range of pH	
6. The kinetics of the reductive half-reactions of the glutamine variants usin	
stopped-flow spectrophotometry	-
D. Results and discussion	
1. The activity of the glutamate variants	
2. Effect of pH on the steady-state kinetics of the glutamate variants	
3. Effect of pH on the reductive half-reaction of E469'Q and E469'A DmT	
E. Concluding remarks	115

Chapter 4: Conclusions and Potential Future Work	119
1	
Appendices	129

# List of Figures

Fig. 1.1 The life cycle of malarial parasites21
Fig. 1.2 Stereo view of the structure of a thioredoxin fold25
Fig. 1.3 Stereo view showing the relative positions of amino acid residues in the active site of DmTrxR
Fig. 2.1 Stereo view showing the relative positions of amino acid residues in the active site of DmTrxR
Fig. 2.2 The pH profiles of Vmax for wild-type DmTrxR(A) and H464'Q DmTrxR(B).66
Fig. 2.3 The pH profiles of Vmax/Km for wild-type DmTrxR(A) and H464'Q DmTrxR(B)
Fig. 2.4 The turnover numbers of wild-type DmTrxR and H464'Q DmTrxR in the presence of imidazole-imidazolium at pH 7.5 and at pH 8.5
Fig. 2.5 Titrations of wild-type with NADPH at pH 871
Fig. 2.6 Titration of wild-type DmTrxR with NADPH at pH 7, 8, and 972
Fig. 2.7 Titration of H464'Q DmTrxR with NADPH at pH 874
Fig. 2.8 Titration of H464'Q DmTrxR with NADPH at pH 7, 8, and 975
Fig. 2.9 Spectra and kinetics observed in the reductive half-reactions of wild-type DmTrxR
Fig 2.10 Spectra and kinetics observed in the reductive half-reactions of H464'Q DmTrxR
Fig.2.11 The oxidative half-reaction of wild-type DmTrxR at pH 7 under anaerobic conditions carried out in the double mixing stopped-flow instrument
Fig. 2.12 The oxidative half-reaction of H464'Q DmTrxR at pH 7 under anaerobic conditions carried out in the double mixing stopped-flow instrument

Fig. 3.1 Stereo view showing the relative positions of amino acid residues in the active site of DmTrxR
Fig 3.2 The pH profiles of $V_{\text{max}}$ for wild-type , E469'Q and E469'A DmTrxR107
Fig. 3.3 The pH profiles of $V_{\text{max}}/K_m$ for wild-type DmTrxR and E469'Q DmTrxR108
Fig. 3.4 Spectra and kinetics observed in the reductive half-reaction of E469'A DmTrxR
Fig. 3.5 Spectra and kinetics observed in the reductive half-reaction of E469'Q DmTrxR
Fig. 3.6 The kinetics at 450 nm in the reductive half-reaction of E469'A DmTrxR at pH 7 (trace 1) and pH 9 (trace 2) cf. wild-type in Fig. 2.9
Fig. 3.7 The kinetics at 450 nm in the reductive half-reaction of E469'Q DmTrxR at pH 7 (trace 1) and pH 9 (trace 2) cf. wild-type in Fig. 2.9
Fig. A.1.1 The structures of tocopherol and tocotrienol142
Fig. A.2.1 Alignment of high M <sub>r</sub> TrxRs147

## List of Tables

Table 2.1 The apparent rate constants measured during the reductive half-reactions of wild-type and H464'Q DmTrxR, at different pH values, at 25 °C78
Table 2.2 The apparent rate constants of the oxidative half-reactions of wild-type and H464'Q DmTrxR, at different pH values, at 25 °C
Table 3.1 The apparent rate constants measured during the reductive-half reactions of wild-type, E469'Q and E469'A DmTrxR, at different pH values, at 25°C111
Table A.2.1 The primer for site- site-directed mutagenesis

# List of Appendices

Appendix 1	. 129
Appendix 2	. 146

### List of Abbreviations

- AP-1: activator protein 1
- ASK-1: apoptotic signal kinase-1
- CTC: charge-transfer complex (identified as donor-acceptor)
- DmTrxR: thioredoxin reductase from Drosophila melanogaster
- DTNB: 5,5'-dithiobis-(2-nitrobenzoic acid)
- EH<sub>2</sub>: 2-electron-reduced enzyme
- EH<sub>4</sub>: 4-electron reduced enzyme
- FAD: flavin adenine dinucleotide
- GPx: glutathione peroxidase
- GSH: glutathione
- GSSG: glutathione disulfide
- GR: glutathione reductase
- Grx: glutaredoxin
- GST: glutathione-S-transferase
- HIF-1 $\alpha$ : hypoxia-inducible factor 1  $\alpha$
- LBHB: low-barrier hydrogen bond
- MDS: mixed disulfide
- NF-κB: nuclear factor- κB
- PfTrxR: thioredoxin reductase from Plasmodium falciparum

Prx: perodiredoxin

Ref-1: redox factor-1

TNB: thionitrobenzoate

TNF: tumor necrosis factor

Trx: thioredoxin

TrxR: thioredoxin reductase

### Abstract

Thioredoxin reductase (TrxR) catalyses the reduction of thioredoxin (Trx) by NADPH. Like other members of the pyridine nucleotide-disulfide family, TrxR is a homodimer. The catalytically active unit in the enzyme from *Drosophila melanogaster* (DmTrxR) consists of three redox centers: FAD and an N-terminal Cys-57/Cys-62 redoxactive disulfide from one monomer, and a Cys-489'/Cys-490' C-terminal redox-active disulfide from the second monomer. Because dipteran insects such as *D. melanogaster* lack glutathione reductase, glutathione disulfide must be reduced by Trx, making DmTrxR particularly important in this organism. DmTrxR is used as a model for the enzyme from a malaria vector, *Anopheles gambiae*. Based on the structures and mechanisms of other family members, a dyad of His-464' and Glu-469' acts as the acidbase catalyst of the dithiol-disulfide interchange reactions required in the catalysis of DmTrxR.

The functions of His-464' and Glu-469' in the catalytic mechanism of DmTrxR were investigated. His-464' was shown to be critical to catalysis by DmTrxR; thus, H464'Q retains only 2% of the wild-type activity. The pH dependence of  $V_{\text{max}}$  for wild-type DmTrxR has apparent p $K_a$  values of 6.4 and 9.3, whereas H464'Q DmTrxR has an observable p $K_a$  only at 6.4, indicating that the p $K_a$  at pH 9.3 is contributed by His-464'. The macroscopic p $K_a$  at pH 6.4 has been assigned to Cys-57 and Cys-490'; the thiolate of Cys-57 is the nucleophile in the internal dithiol-disulfide interchange reaction and the

thiolate of Cys-490' is the nucleophile in the reduction of Trx. The rates of both the reductive and oxidative half reactions are markedly smaller in H464'Q DmTrxR than those of wild-type enzyme, indicating that His-464' is involved in both half reactions. The pH dependence of the steady-state kinetics shows that the basicity of His-464' decreases in the glutamate variants, as predicted. The reductive half-reactions of two glutamate variants are slower than those of wild-type enzyme.

Malaria causes serious public health problems in the world. It is hoped that differences among TrxRs from human, *Plasmodium falciparum* (the causative agent) and *Diptera* (the vector) will be useful for developing differential inhibitors, useful as prophylactics.