

Analysis of tunable diode laser spectra of the ν_4 band of $^{12}\text{C}_3^{16}\text{O}_2$

W. H. Weber and P. D. Maker

Physics Department, Engineering and Research Staff, Ford Motor Company, Dearborn, Michigan 48121

C. W. Peters

Physics Department, University of Michigan, Ann Arbor, Michigan 48104

(Received 25 September 1975)

Current-tunable thin-film PbTe diode lasers have been used to obtain Doppler-limited resolution of the ν_4 asymmetric stretch band of C_3O_2 in the region 1565–1600 cm^{-1} . Roughly 70% of the interval was covered using different modes of three lasers, each mode giving a tuning range of 1–3 cm^{-1} . Absolute spectral calibration was obtained from NH_3 and H_2O , together with the channel spectrum of a 4-cm Ge etalon recorded simultaneously using a double beam technique. Rotational constants were determined for the ν_4 fundamental band and five different hot bands associated with the thermally excited low frequency bending mode ν_7 . The band origin for the ground state transition is $1587.390 \pm 0.002 \text{ cm}^{-1}$ and the ν_7 hot bands are shifted progressively to lower frequency. The systematic shift of the hot bands was verified independently with temperature difference spectra from a Fourier transform spectrometer.

I. INTRODUCTION

We describe in this paper the use of tunable, thin-film, PbTe diode lasers in a study of the ν_4 fundamental band of C_3O_2 in the region 1565–1610 cm^{-1} . Diode lasers made from Pb-salt semiconductors have been used previously in several spectroscopic studies exploiting their high spectral impurity.¹ Some examples are studies of Λ -type doubling,^{2,3} pressure broadening,^{4,5} Zeeman splittings,² Stark splittings,⁶ and nuclear hyperfine splittings.⁷ In each of these experiments the spectra obtained covered a range of at most only a few cm^{-1} . The work reported here is on the use of such lasers in the analysis of vibration-rotation bands in an extended spectral region of roughly 35 cm^{-1} . From spectra recorded with Doppler-limited resolution, the fundamental and five different hot bands of ν_4 of C_3O_2 have been identified, rotational constants for all of these bands have been determined, and over 300 lines have been assigned.

The fundamental vibrational frequencies of C_3O_2 are known fairly well from extensive Raman and infrared studies.^{8–14} The last fundamental to be observed directly was the lowest frequency bending vibration ν_7 reported by Carriera *et al.*¹⁵ to be at 22 cm^{-1} . The presence of this low frequency mode leads to a large number of hot bands associated with transitions whose lower states have one or more quanta of ν_7 thermally excited. Except in some of the combination bands in the near infrared, these hot bands tend to completely overlap one another, greatly complicating any attempts at rotational analyses. There has, in fact, been no previous analysis of rotational structure in any of the fundamentals of this molecule.

The first infrared study of C_3O_2 resolving the rotational structure was made on an unidentified combination band near 3200 cm^{-1} by Lafferty, Maki, and Plyler.¹⁶ They provided convincing evidence that the molecule's structure is linear with $D_{\infty h}$ symmetry and obtained molecular constants from analyses of a $\Sigma-\Sigma$ and a $\Pi-\Pi$ series. Recently, Mantz *et al.*¹⁷ have re-examined the near infrared spectrum using a Fourier

transform spectrometer with much higher resolution. They analyzed rotational structure in seven different $\Sigma-\Sigma$ bands and confirmed the assignment by Lafferty *et al.* of the 3169 cm^{-1} band as originating in the ground vibrational state. Different values for the ground state molecular constants were obtained by Mantz *et al.*,¹⁷ however, which they attributed to a misassignment of the J values in the earlier work.

We present results of analyses of the vibration-rotation spectrum of the ν_4 fundamental (species σ_u), which is the asymmetric C=C=C stretching mode. All of the observed transitions are of the type $\nu_4 1-0$ with a lower level having 0, 1, or 2 quanta of ν_7 (species π_u) thermally excited. Thus the bands will be of the type $\Sigma_u-\Sigma_g$, $\Pi_g-\Pi_u$, or $\Delta_u-\Delta_g$. Because of spin statistics, the $\Sigma-\Sigma$ bands will contain only even J 's, while the remaining bands will have both even and odd J 's. The even and odd J components of these bands will have slightly different molecular constants due to the l doubling.

Spectra with Doppler-limited resolution were obtained using current-tuned, PbTe diode lasers. Lower resolution spectra taken at different temperatures were obtained with a Fourier transform spectrometer, and these aided in the identification and analysis of the ground state transitions. The molecular constants B , D , and H were obtained for each band identified using a least squares computer fit to the standard polynomial expansion in J appropriate for a linear molecule. Our results agree with those of Mantz *et al.*¹⁷ for the ground and $2\nu_7^0$ states and with assignment 2 of Lafferty *et al.*¹⁶ for the ν_7^1 state. The D terms were found to be anomalously large compared with other similar molecules, suggesting that the primary effect of centrifugal distortion is a reduction in amplitude of the bending mode ν_7 .

II. EXPERIMENTAL

A. Diode lasers

The diode lasers were made from single crystal PbTe epitaxial films typically 4 μ thick grown on BaF_2

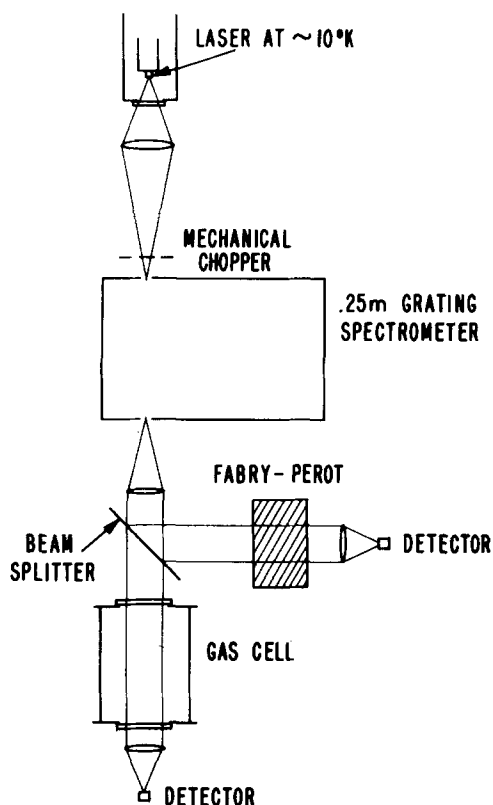


FIG. 1. Experimental arrangement for diode laser spectroscopy. The beam splitter is a thin silicon wafer, the Fabry-Perot is a 4-cm Ge etalon, the detectors are liquid-nitrogen-cooled Ge:Au photoconductors, and the windows and lenses are BaF_2 .

substrates. The laser cavities were 200 to 400 μ long. End facets were produced by photoresist etching. An evaporated Pb film electrode provided the minority carrier injection required for lasing. Continuous wave lasing was obtained at diode temperatures below about 20 K. The properties of these devices and details of their fabrication have been described previously.^{6,18} Tuning was accomplished by varying the current (50–250 mA) through the diode. The joule heating by the diode current produced a change of the temperature and consequently of the refractive index in the active PbTe region, leading to a change in the laser frequency. The continuous tuning range of a single mode was typically 1.5 to 2 cm^{-1} , which is somewhat larger than that normally obtained with bulk crystal Pb-salt lasers.^{1,5,19} Without this broader tuning range the identification of the various series of lines, whose spacings were typically 0.3 cm^{-1} , would have been very difficult. Normally 2–3 modes lase simultaneously in a particular diode and the frequencies of these modes shift somewhat when the Dewar is recycled from room temperature to liquid He temperature on different days. In addition, these diodes do not have a simple mode pattern, but tend to lase in two distinct regions some 10 to 15 cm^{-1} apart. The positions and separations of these two regions vary from diode to diode.

By using all the available modes of three different lasers, roughly 70% of a 35 cm^{-1} region extending from 1565 to 1600 cm^{-1} has been covered. There are five

gaps remaining in the spectrum, the largest being 5 cm^{-1} centered at 1595 cm^{-1} , while the other four are on the order of 1 cm^{-1} .

B. Calibration

A diagram of the apparatus is shown in Fig. 1. The 25 cm focal length grating spectrometer is used to select individual modes and to provide rough calibration. Relative frequency calibration is obtained from the fringes of a 4-cm thick uncoated Ge étalon, recorded simultaneously with the absorption spectrum using a double beam technique. The fringe separation was calculated at 5 cm^{-1} intervals using the measured thickness and the published refractive index values, including dispersion.²⁰ This quantity varied smoothly from 0.032770 cm^{-1} at 1565 cm^{-1} to 0.032759 cm^{-1} at 1600 cm^{-1} . From analysis of multiple records of the same spectral regions we found the rms uncertainty associated with measuring the separation between two strong lines to vary from about $\pm 0.0005 \text{ cm}^{-1}$ to 0.002 cm^{-1} , depending on the signal-to-noise ratio in the fringe and signal channels.

The double beam technique greatly improves the accuracy with which frequency differences can be measured. Although the frequency of a laser mode increases monotonically with the diode current, it has a nonlinear dependence on this quantity. In addition, there are temperature fluctuations in the Dewar which produce small but random changes in the laser frequency. Both of these difficulties preclude using the diode current to accurately monitor the laser frequency. Also, any disruption of the uniform progression of fringes served to indicate a changed laser mode structure and possible leakage of a second mode through the spectrometer.

Absolute calibration is obtained with H_2O and NH_3 lines.^{21,22} Since there is some arbitrariness in the choice of reference lines, we show in Table I a summary of the calibration points used in the various spectral regions. The separations of several closely

TABLE I. Summary of calibration lines used in various spectral regions. Line positions are taken from Refs. 21 and 22.

Spectral region (cm^{-1})	Calibration
1597–1600	NH_3 line # 349 at 1599.0696 cm^{-1} .
1590–1592	Average of $\alpha^{\text{P}}\text{P}(2,1)$ and $s^{\text{P}}\text{P}(2,1)$ NH_3 lines at 1590.8844 cm^{-1} .
1583–1589	$\alpha^{\text{R}}\text{P}(2,0)$ NH_3 line at 1586.8593 cm^{-1}
1576–1581	Linear interpolation between the average of the $\alpha^{\text{P}}\text{P}(3,3)$ and $s^{\text{P}}\text{P}(3,3)$ NH_3 lines at 1579.4838 cm^{-1} and the $5_{3,2} - 6_{2,5}$ H_2O line at 1577.5839 cm^{-1} .
1572–1576	Linear interpolation between the $s^{\text{P}}\text{P}(3,2)$ NH_3 line at 1575.8520 cm^{-1} and the $\alpha^{\text{P}}\text{P}(3,1)$ NH_3 line at 1572.4827 cm^{-1} .
1565–1569	$s^{\text{R}}\text{P}(3,0)$ NH_3 line at 1567.9901 cm^{-1} .

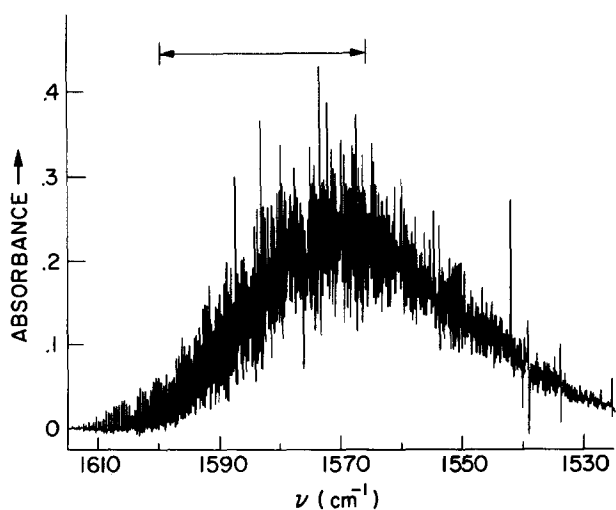


FIG. 2. Room temperature Fourier transform spectrum at 5 torr and 0.5 m path length in the ν_4 band of $^{12}\text{C}_3^{16}\text{O}_2$. The curve was obtained by straight line interpolation of digitized absorbance values computed at $\frac{1}{16}$ cm^{-1} intervals. The arrows indicate the range of the laser data. Most of the large spikes can be attributed to interference from strong ($\sim 100\%$ absorbing) H_2O lines which produced large errors in the ratioed absorbance data, e.g., the apparent dips at 1576 and 1539 cm^{-1} and the spikes at 1542 and 1534 cm^{-1} .

spaced NH_3 inversion doublets measured with our apparatus agreed (± 0.001 cm^{-1}) with the recent microwave results of Cohen and Poynter,²³ but they were systematically smaller (~ 0.02 cm^{-1}) than the reported infrared values.²² To compensate for these errors, apparently associated with deconvoluting the incompletely resolved grating spectra, we used the mean values of the inversion pairs for calibration.

In two spectral regions, each involving several overlapping charts, we determined the fringe spacing for calibration by measuring the separations of two widely spaced H_2O or NH_3 lines near the ends of the regions. This procedure was necessary since the end-point calibration line separations measured in terms of the calculated fringe spacing differed from the reported separations by a few parts per thousand in both spectral regions. At the moment we attribute these discrepancies to either or both or two possibilities: the index and dispersion of our Ge étalon differ from the published values, and/or there are small errors ($\sim 10^{-2}$ cm^{-1}) in some of the reported line positions.

C. Sample preparation

Carbon suboxide samples were obtained by the decomposition of malonic acid in a mixture of dry sand and phosphorus pentoxide. This mixture was heated in vacuum to 140°C and the reaction products collected in a liquid nitrogen trap. Fractional distillation was then used to obtain a pure liquid sample of C_3O_2 . This concentrated sample could be kept at 77 K indefinitely. Gas samples at 10 torr were kept for several months at -20°C with no sign of decomposition. The above procedure has been demonstrated to yield high purity C_3O_2 samples.¹³ The only impurity identified from the infrared spectra was a trace amount of CO_2 .

III. RESULTS

A. Fourier transform spectra

A Fourier transform spectrum of the ν_4 band of C_3O_2 is shown in Fig. 2 at a resolution of 0.063 cm^{-1} . These data were obtained with an Eocom Corp. model 7001 laser-controlled, scanning Michelson interferometer and a Digital Equipment Corp. PDP 11/40 computer. This instrument has been described previously²⁴ and will be described in more detail in a subsequent publication. The arrows at the top of Fig. 2 indicate the range of the laser data. The only resolvable series in this spectrum is the set of strong, single-channel lines which stand out above 1600 cm^{-1} . This series is the R branch of the ground state transition. The absence of correspondingly strong lines on the low frequency side of the band indicates a greater density of weaker lines in that region. This suggests that the hot bands are shifted to lower frequencies relative to the ground state transitions.

Figure 3 shows a Fourier transform spectrum of the difference between the absorbance at room temperature and 193 K. The cell pressure was 5 torr, which was low enough to prevent condensation upon cooling. The resolution is 0.63 cm^{-1} and all of the sharp structure observed is due to atmospheric H_2O interference. The increase in the room temperature absorbance on the low frequency side of the band is a further indication that the hot bands are progressively displaced to lower frequencies. This is opposite to the behavior of the hot bands in the combination band first studied by Lafferty *et al.*¹⁶ and later identified as $2\nu_2 + \nu_4$ by Mantz *et al.*¹⁷

B. Laser spectra

Laser spectra were recorded at low pressures (0.05–0.5 torr) to eliminate pressure broadening. The resolution under these conditions should be limited

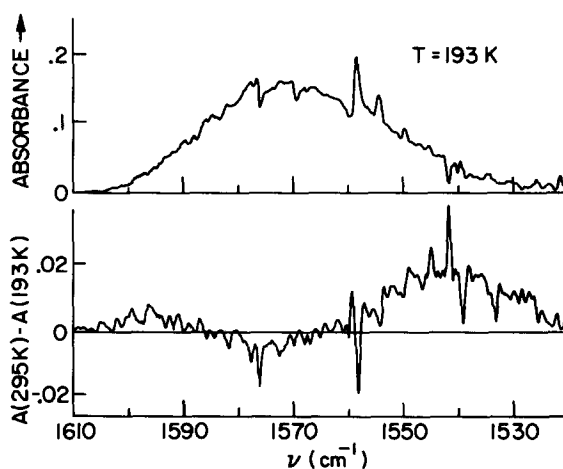


FIG. 3. The upper trace is a Fourier transform spectrum at 193 K of C_3O_2 at 5 torr and 0.5 m path length with 10 channels averaged together (0.63 cm^{-1} resolution) to remove the rotational structure. The lower trace is the difference between the room temperature and 193 K absorbance curves on the same sample. Most of the sharp structure is due to H_2O interference.

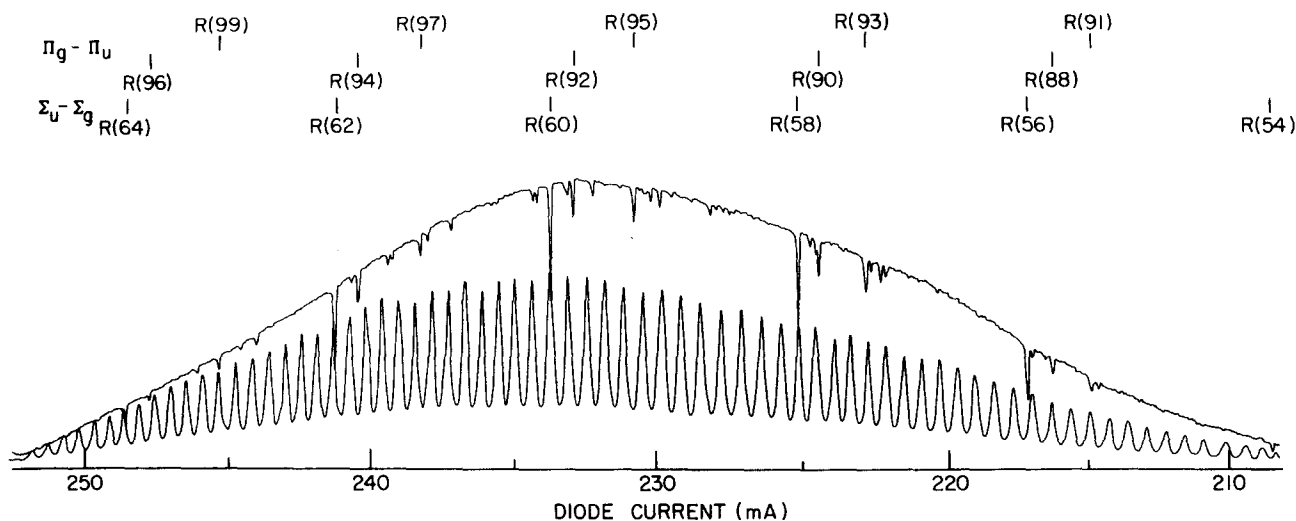


FIG. 4. A 2.3 cm^{-1} long diode laser spectrum of C_3O_2 at 0.1 torr and 3.2 m path length. The spectral region is $1597\text{--}1600\text{ cm}^{-1}$ and the calculated fringe spacing is 0.032759 cm^{-1} . The assignment of the odd- J $\Pi\text{--}\Pi$ series is tentative; it differs by 2 from that indicated in Table III.

by the laser linewidth $\Delta\nu_L$ and the natural Doppler width $\Delta\nu_D$ given by

$$\Delta\nu_D(\text{cm}^{-1}) = 7.17 \times 10^{-7} \nu(T/M)^{1/2} \approx 0.0024\text{ cm}^{-1}.$$

Although we have not attempted to measure $\Delta\nu_L$ directly, comparisons of the observed absorption linewidths with $\Delta\nu_D$ above indicate $\Delta\nu_L \lesssim 10^{-3}\text{ cm}^{-1}$. This result is consistent with the theoretical estimate $\Delta\nu_L \approx 10^{-4}\text{ cm}^{-1}$ for a typical laser mode emitting a few μW of power.¹

Figure 4 shows a 2.3 cm^{-1} laser scan at the highest frequency obtained with our lasers. The strongest lines in this figure are the same resolved lines appearing near 1600 cm^{-1} in Fig. 2, and we have identified these as the R branch of the ground state transition. Since we expect the hot bands to be systematically shifted to lower frequencies, we would identify the next strongest series of lines, of which there are two with

about equal intensities, to be the even and odd J components of the l -doubled $\Pi\text{--}\Pi$ series associated with ν_1^1 in the lower state.

A laser scan 1.7 cm^{-1} long near the band origin of the ground state transition is shown in Fig. 5. The first few R -branch lines for this series are indicated. An overlapping scan at lower frequencies shows the beginning of the P branch for the same series. The relative intensities of the small J lines and the gap between $R(0)$ and $P(2)$ of 1.5 times the normal spacing established the position of the band origin. This is the only band in which we have identified unambiguously all the lines on either side of the band origin. The determinations of band origins for the other series generally required trial and error fits to various possibilities, since the large density of weak lines occurring at lower frequencies usually made positive identification of the

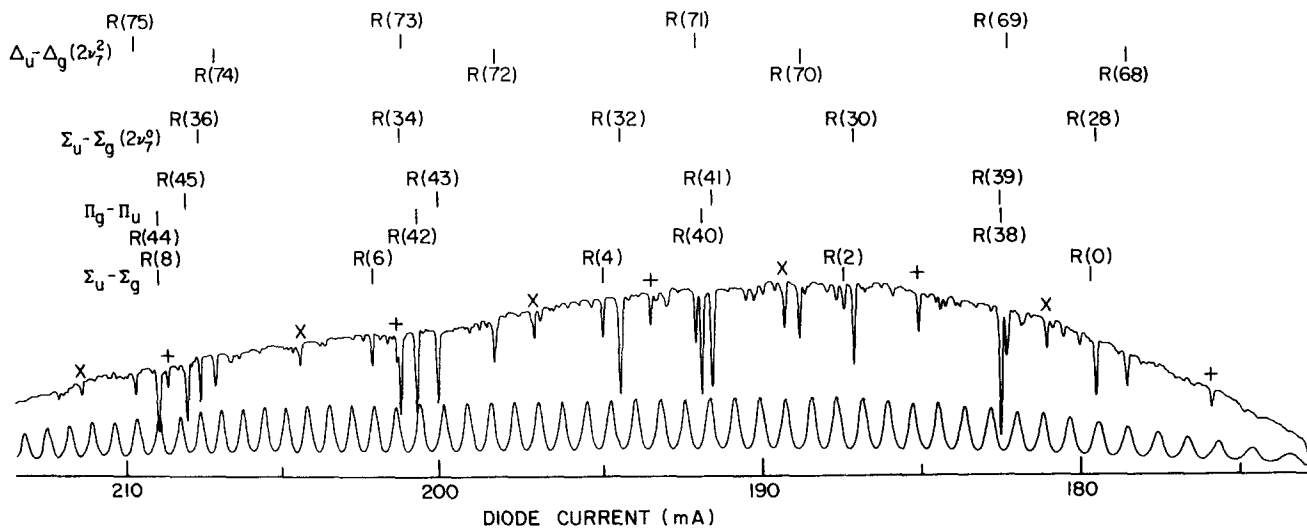


FIG. 5. A 1.7 cm^{-1} long diode laser spectrum of C_3O_2 at 0.1 torr and 3.2 m path length. The spectral region is $1587\text{--}1589\text{ cm}^{-1}$ and the calculated fringe spacing is 0.032762 cm^{-1} . The ground state $R(0)$ line appears as a weak shoulder on the high frequency side of the $2\nu_2^2$ $R(28)$ line.

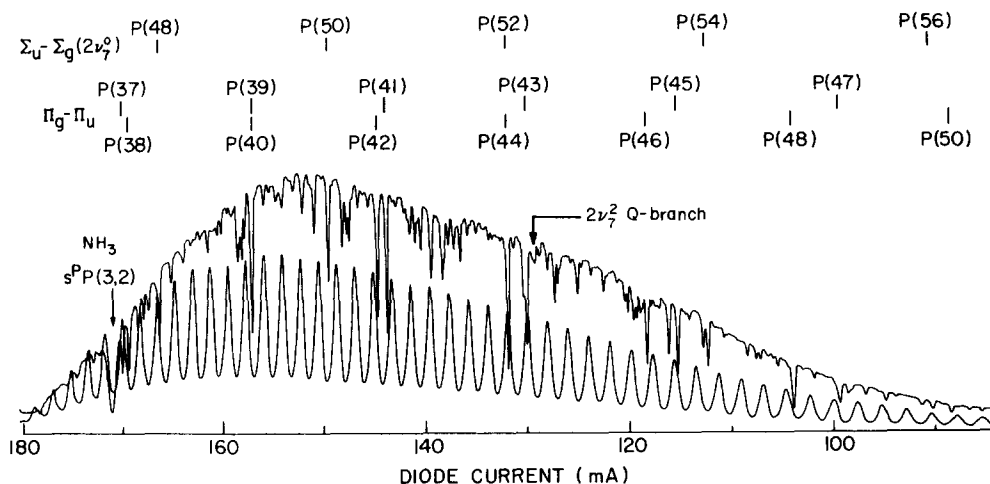


FIG. 6. A 1.6 cm^{-1} long diode laser spectrum of C_3O_2 at 0.1 torr and 3.2 m path length. The $\text{NH}_3\text{ s}^5\text{P}(3,2)$ line is at 1575.8520 cm^{-1} and the calculated fringe spacing is 0.032767 cm^{-1} . The NH_3 at several torr pressure is contained in a separate 10 cm cell in series with the C_3O_2 cell.

first few J lines impossible.

There are several other series apparent in Fig. 5, besides the extensions of the three already identified in Fig. 4. The strongest of these is a Σ - Σ band, and we have assigned it to a transition whose lower vibrational state is $2\nu_7^0$. Next in strength are two series, of about equal intensity and spacing, which are displaced from each other by nearly half the line spacing. We have identified these series as a Δ - Δ band associated with $2\nu_7^2$ in the lower state. This assignment is consistent with the expected systematic shift of the hot bands to lower ν and the expected small value for the l -doubling constant. The l -doubling constant should be extremely small in a Δ - Δ band, thus the two series will appear to be a single series with half the line spacing until quite large J values are reached.

There still remain two identifiable series in Fig. 5, which are marked with the + and \times symbols. These have roughly equal intensities with slightly different spacings. The assignment of these series is quite tentative, since we have been unable to identify the corresponding P branches. However, the most logical choices are a Π - Π or a Φ - Φ band, associated with $3\nu_7^1$ or $3\nu_7^3$, respectively. We presently favor the $3\nu_7^1$ assignment, since the l -doubling constant appears to be too large for the $l=3$ state.

Figure 6 shows a 1.5 cm^{-1} scan near the region of maximum average absorption. The P -branch lines for several of the series appearing in Fig. 5 are shown in this figure, as well as the $\text{NH}_3\text{ s}^5\text{P}(3,2)$ line at 1575.8520 cm^{-1} used for calibration. The diode current scale is linear, and a marked decrease in the fringe spacing is observed at high currents, indicative of the nonlinear dependence of the laser frequency on diode current. Due to the large density of weak lines we were able to identify only the strongest series in this spectral region and at lower frequencies. Even for these strong series, overlapping was often a problem. Note, for example, that both the $P(40)$ and $P(44)$ Π_g - Π_u lines in Fig. 6 are overlapped.

Also shown in Fig. 6 is our tentative assignment of the $2\nu_7^2$ Q branch. Weak Q branches are allowed for the $l \neq 0$ states, but assigning them is very difficult

since the lines decrease in intensity as J increases, the strongest being only 15% of the intensity of the strongest P - or R -branch lines.²⁵ The assignment in Fig. 6 was achieved only after the corresponding P and R branches had been found. The individual lines are not resolved but the shape, position, and intensity of the band are consistent with the Q -branch assignment.

IV. BAND ANALYSES AND MOLECULAR CONSTANTS

The molecular constants for each band were determined by fitting the observed line positions to the usual polynomial expansion in m :

$$\nu = \nu_0 + (B' + B'')m + (B' - B'')m^2 - 2(D' + D'')m^3 - (D' - D'')m^4 + 3(H' + H'')m^5 + (H' - H'')m^6, \quad (1)$$

where $m = -J$ for the P branch and $m = J + 1$ for the R branch. Approximate values for the molecular constants were obtained first by graphical methods using combination sums and differences. These values were then refined by a least-squares computer program. Since the gaps in the spectrum often limited the sums and differences to about half of the J values, the accuracy of the constants was greatly improved by the computer fit, which used all the observed lines.

A. Σ_u - Σ_g ground state

Table II gives the observed frequencies and the errors between the observed and calculated frequencies for the transitions originating in the ground vibrational state. Data for lines above 1600 cm^{-1} were taken from the Fourier transform spectra. These lines have a statistical uncertainty of approximately $\pm 0.02\text{ cm}^{-1}$, i.e., at least an order of magnitude greater than those measured with the lasers. Absolute calibration of the Fourier transform data was adjusted to agree with the same NH_3 lines used for calibrating the laser data.

The intensities measured with the Fourier transform spectrometer of the individual R lines near $J=60$ were found to increase 10%-20% when the sample was cooled to 193 K. This intensity change is consistent with the ground state assignment, since the rotational Boltzmann factor alone would predict a sizable decrease in intensity.

TABLE II. Observed frequencies (vac. cm^{-1}) of $\nu_4\ 1 \leftarrow 0$ ground state transitions. Lines above $1600\ \text{cm}^{-1}$ are determined from the Fourier transform spectrometer data. The rms error of the fit using the molecular constants in Table VIII is $\pm 0.0021\ \text{cm}^{-1}$ for the 58 lines below $1600\ \text{cm}^{-1}$ and $\pm 0.0071\ \text{cm}^{-1}$ for all 82 lines.

J	$R(J)$ OBS	O-C	$P(J)$ OBS	O-C
0	1587, 5398	-0.0028
2	1587, 8506	-0.0008	1587, 0887	-0.0005
4	1588, 1645	-0.0007	1586, 7898	-0.0036
6	1588, 4835	-0.0007	1586, 4999	-0.0028
8	1588, 8103	0.0020	1586, 2188	0.0016
10	1589, 1381	0.0007	1585, 9376	0.0008
12	1585, 6623	0.0008
14	1585, 3913	-0.0001
16	1585, 1267	0.0005
18	1590, 4989	-0.0043	1584, 8672	0.0011
20	1590, 8551	-0.0016	1584, 6129	0.0020
22	1591, 2153	0.0005	1584, 3612	0.0006
24	1591, 5797	0.0022	1584, 1180	0.0028
26	1591, 9488	0.0042	1583, 8766	0.0021
28	1592, 3216	0.0054	1583, 6396	0.0012
30	1583, 4080	0.0011
54	1597, 4861	-0.0006
56	1597, 9042	-0.0013	1580, 7629	0.0002
58	1598, 3238	-0.0026	1580, 5812	-0.0013
60	1598, 7477	-0.0017	1580, 4034	-0.0015
62	1599, 1708	-0.0034	1580, 2269	-0.0027
64	1599, 5951	-0.0055	1580, 0529	-0.0036
66	1600, 0300	0.0016	1579, 8835	-0.0019
68	1600, 4520	-0.0054	1579, 7149	-0.0012
70	1600, 8740	-0.0136	1579, 5482	-0.0001
72	1601, 2960	-0.0225	1579, 3796	-0.0024
74	1601, 7780	0.0279	1579, 2168	-0.0001
76	1602, 2000	0.0178	1579, 0519	-0.0009
78	1602, 6220	0.0075	1578, 8897	0.0003
80	1603, 0440	-0.0029	1578, 7277	0.0010
82	1603, 4660	-0.0132	1578, 5664	0.0020
84	1578, 4046	0.0023
86	1578, 2418	0.0016
88	1604, 7910	0.0179	1578, 0790	0.0012
90	1605, 2130	0.0102	1577, 9161	0.0010
92	1605, 6450	0.0136	1577, 7535	0.0018
94	1606, 0570	-0.0017	1577, 5910	0.0036
96	1606, 4790	-0.0055	1577, 4230	0.0009
98	1606, 9010	-0.0077	1577, 2549	-0.0007
100	1607, 3220	-0.0090	1577, 0852	-0.0024
102	1607, 7440	-0.0072	1576, 9154	-0.0026
104	1608, 1660	-0.0032	1576, 7457	-0.0009
106	1608, 5880	0.0031
108	1609, 0100	0.0120
112	1609, 7930	-0.0231
114	1610, 2150	-0.0058
116	1610, 6370	0.0146
120	1611, 4200	0.0043

B. $\Pi_g - \Pi_u\ \nu_4 + \nu_7^1 \leftarrow \nu_7^1$

The observed frequencies of the $\Pi - \Pi$ bands are shown in Tables III and IV. The J assignment of the $R(J \geq 89)$ lines in Table III is tentative, since these lines appear to be strongly perturbed. An alternative assignment, based only on the extrapolation back to the $R(53-63)$ region, is used to label this series in Fig. 4. Neither assignment gives a reasonable fit to Eq. (1), and for this reason we have omitted these lines from the determination of molecular constants. A Coriolis

interaction is probably the source of the perturbation of these lines. Two possible vibrations which have both the correct symmetry and approximately the right frequency for a Coriolis interaction with $\nu_4 + \nu_7^1$ are $2\nu_2$ and $3\nu_6^1 + \nu_7^1$. There is no evidence of perturbations in the even $-J\ \Pi - \Pi$ band given in Table IV, and all the lines are included in the fit for that band.

C. $\Delta_u - \Delta_g\ \nu_4 + 2\nu_7^2 \leftarrow 2\nu_7^2$

The observed frequencies for the $\Delta - \Delta$ bands are shown in Tables V and VI. The weak Coriolis perturbation apparent in the $R(61)$ and $P(63)$ lines of the odd $-J$ series in Table V allowed us to assign these two series with some certainty. Without relying on the fact that the perturbed levels have a common upper state, other possible assignments for the band origin

TABLE III. Observed frequencies (vac. cm^{-1}) of the $\nu_4 + \nu_7^1 \leftarrow \nu_7^1$ transitions. The $J \geq 80$ R lines appear to be highly perturbed and are not included in the computer fit. Even the assignment of J values for these lines is tentative, since the extrapolation to lower J did not yield an unambiguous assignment. The rms error of the fit using the molecular constants in Table VIII is $\pm 0.0023\ \text{cm}^{-1}$ for 57 lines.

J	$R(J)$ OBS	O-C	$P(J)$ OBS	O-C
5	1580, 1428	-0.0020
7	1579, 8485	-0.0019
9	1579, 5579	-0.0017
11	1579, 2699	-0.0024
13	1578, 9885	-0.0001
15	1583, 4499	0.0012	1578, 7103	0.0020
17	1583, 7844	0.0016	1578, 4348	0.0033
19	1584, 1202	0.0001	1578, 1609	0.0028
21	1584, 4615	0.0010	1577, 8918	0.0037
23	1584, 8035	-0.0004	1577, 6253	0.0040
25	1585, 1488	-0.0014	1577, 3609	0.0031
27	1585, 4974	-0.0020	1577, 0982	0.0008
29	1585, 8506	-0.0008	1576, 8396	-0.0005
31	1586, 2101	0.0040	1576, 5833	-0.0024
33	1586, 5644	0.0009
35	1586, 9205	-0.0030	1576, 0817	-0.0040
37	1587, 2835	-0.0025	1575, 8350	-0.0049
39	1587, 6496	-0.0014	1575, 5925	-0.0042
41	1588, 0176	-0.0009	1575, 3512	-0.0048
43	1588, 3887	0.0003	1575, 1148	-0.0031
45	1588, 7618	0.0012	1574, 8804	-0.0018
47	1589, 1381	0.0028	1574, 6489	0.0001
49	1574, 4211	0.0035
51	1574, 1918	0.0031
53	1590, 2728	-0.0009	1573, 9631	0.0012
55	1590, 6591	0.0010	1573, 7394	0.0022
57	1591, 0470	0.0018	1573, 5159	0.0014
59	1591, 4366	0.0016	1573, 2967	0.0027
61	1591, 8274	-0.0003	1573, 0754	-0.0000
63	1592, 2204	-0.0031	1572, 8615	0.0025
65	1572, 6447	0.0001
67	1572, 4325	0.0000
69	1572, 2206	-0.0019
71	1572, 0140	-0.0010
73	1571, 8087	-0.0012
89	1597, 7941	0.0002
91	1598, 1957	-0.0775
93	1598, 6002	-0.1633
95	1599, 0017	-0.2641
97	1599, 4036	-0.3776

TABLE IV. Observed frequencies (vac. cm^{-1}) of the $\nu_4 + \nu_7^d - \nu_7^d$ transitions. The rms error of the fit using the molecular constants in Table VIII is $\pm 0.0031\ \text{cm}^{-1}$ for 63 lines.

J	$R(J)$ OBS	O-C	$P(J)$ OBS	O-C
6	1579.9931	-0.0072
8	1579.7000	-0.0086
10	1579.4144	-0.0067
12	1579.1347	-0.0031
14	1583.3254	0.0048	1578.8588	-0.0000
16	1583.6633	0.0025	1578.5831	-0.0009
18	1584.0080	0.0030	1578.3140	0.0007
20	1584.3556	0.0025	1578.0477	0.0010
22	1584.7083	0.0032	1577.7875	0.0033
24	1585.0644	0.0035	1577.5288	0.0031
26	1585.4234	0.0029	1577.2744	0.0032
28	1585.7871	0.0034	1577.0218	0.0012
30	1586.1562	0.0057	1576.7737	-0.0000
32	1586.5213	0.0006	1576.5304	-0.0003
34	1586.8915	-0.0029
36	1587.2685	-0.0029	1576.0570	0.0015
38	1587.6491	-0.0025	1575.8213	-0.0019
40	1588.0312	-0.0037	1575.5928	-0.0015
42	1588.4176	-0.0035	1575.3675	-0.0012
44	1588.8079	-0.0024	1575.1456	-0.0008
46	1589.2020	-0.0002	1574.9275	0.0004
48	1574.7172	0.0064
50	1574.5008	0.0033
52	1590.3889	-0.0046	1574.2890	0.0021
54	1590.7923	-0.0031	1574.0801	0.0012
56	1591.1989	-0.0005	1573.8751	0.0016
58	1591.6074	0.0019	1573.6704	-0.0001
60	1592.0176	0.0040	1573.4678	-0.0020
62	1573.2692	-0.0020
64	1573.0729	-0.0018
66	1572.8775	-0.0025
68	1572.6845	-0.0026
70	1572.4956	-0.0003
72	1572.3028	-0.0033
74	1572.1161	-0.0016
76	1571.9316	0.0011
78	1571.7503	0.0058
88	1597.8626	0.0026
90	1598.2854	0.0028
92	1598.7065	0.0012
94	1599.1271	-0.0010
96	1599.5473	-0.0035

would have to be considered. As in the case of the $\Pi-\Pi$ bands, there are at least two vibrational states which might be causing a Coriolis perturbation in the $\Delta-\Delta$ bands: $2\nu_2 + \nu_7^1$ or $3\nu_6^1 + 2\nu_7^0$.

D. $\Sigma_u - \Sigma_g\ \nu_4 + 2\nu_7^0 \leftarrow 2\nu_7^0$

Table VII gives the observed frequencies for the transitions $\nu_4 + 2\nu_7^0 - 2\nu_7^0$. The band origin for this series was determined by trying several possible assignments and then choosing the one giving the best fit to Eq. (1). The intensity of the $\Sigma-\Sigma\ 2\nu_7^0$ band is substantially below that of the $\Sigma-\Sigma$ ground state band, and we can use this fact to obtain a rough estimate of the $2\nu_7^0$ vibrational energy. The line strengths of the $P(50)$ and $P(54)\ 2\nu_7^0$ lines, corrected for the expected J dependence, were $\approx 70\%$ of those for the similarly corrected $R(56)-R(62)$ ground state lines, recorded under identical conditions of pressure and scanning rate. If we

make the admittedly weak assumption that the transition moments for the two sets of lines are the same, then the intensity difference is due solely to a vibrational Boltzmann factor, which leads to the estimate $73 \pm 30\ \text{cm}^{-1}$ for the $2\nu_7^0$ state. This range of values is consistent with the value $61\ \text{cm}^{-1}$ found by Carreira *et al.*¹⁵ and Mantz *et al.*¹⁷ The large uncertainty is due primarily to the uncertainty in line strengths caused by weak overlapping lines.

E. Molecular constants

A summary of the molecular constants is given in Table VIII. The uncertainties quoted in the Table are rough estimates which include both the statistical uncertainty and some allowance for possible errors in the calibration points. The ground state contains the largest number of lines and the highest J values and is the only band for which the H terms are determined with any degree of accuracy. Our molecular constants for the ground state are in good agreement with those of Mantz *et al.*,¹⁷ but the agreement in the $2\nu_7^0$ state is not quite as good—our values of B'' and D'' being somewhat larger. For the ν_7^1 states our B'' values and the l -doubling constant $q_l'' = B_l'' - B_c'' = 3.6 (\pm 0.6) \times 10^{-4}\ \text{cm}^{-1}$ support Assignment 2 proposed by Lafferty *et al.*¹⁶ in their analysis of the $\Pi-\Pi$ bands.

TABLE V. Observed frequencies (vac. cm^{-1}) of the $\nu_4 + 2\nu_7^{2c} - 2\nu_7^{2c}$ transitions. The $J=62$ upper level appears to be perturbed and the two transitions associated with it are omitted from the fit. The rms error of the fit using the molecular constants in Table VIII is $\pm 0.0032\ \text{cm}^{-1}$ for 48 lines.

J	$R(J)$ OBS	O-C	$P(J)$ OBS	O-C
7	1574.0383	0.0046
9	1576.6750	-0.0004	1573.7436	0.0057
11	1577.0010	-0.0001	1573.4498	0.0044
13	1577.3346	0.0047	1573.1515	-0.0046
15	1577.6648	0.0031	1572.8709	0.0010
17	1577.9992	0.0026	1572.5842	-0.0027
19	1578.3372	0.0027	1572.3031	-0.0040
21	1572.0250	-0.0055
23	1579.0156	-0.0038	1571.7505	-0.0065
25	1579.3609	-0.0053
27
31	1580.4181	-0.0059
33	1580.7784	-0.0038
45	1568.9518	0.0022
47	1583.3649	0.0005	1568.7141	0.0024
49	1583.7433	0.0001	1568.4803	0.0038
51	1584.1256	0.0012	1568.2475	0.0037
53	1584.5089	0.0012	1568.0150	0.0014
55	1584.8948	0.0016	1567.7870	0.0012
57	1585.2850	0.0042	1567.5608	0.0004
59	1585.6748	0.0045	1567.3362	-0.0010
61	1586.0740	0.0124	1567.1139	-0.0022
63	1586.4531	-0.0016	1566.9019	0.0049
65	1586.8493	-0.0002	1566.6743	-0.0055
67	1587.2436	-0.0021	1566.4613	-0.0030
69	1587.6414	-0.0020	1566.2498	-0.0006
71	1588.0404	-0.0019	1566.0391	0.0011
73	1588.4409	-0.0014	1565.8299	0.0031
75	1588.8428	-0.0005
77	1589.2479	0.0028

TABLE VI. Observed frequencies (vac. cm^{-1}) of the $\nu_4 + 2\nu_7^d$ $\leftarrow 2\nu_7^d$ transitions. The rms error of the fit using the molecular constants in Table VIII is $\pm 0.0038\ \text{cm}^{-1}$ for 50 lines.

J	$R(J)$ OBS	O-C	$P(J)$ OBS	O-C
8	1573.8920	0.0084
10	1573.5939	0.0044
12	1577.1705	0.0076	1573.2948	-0.0039
14	1577.4988	0.0055	1573.0106	-0.0005
16	1577.8293	0.0025	1572.7250	-0.0017
18	1578.1646	0.0013	1572.4430	-0.0025
20	1578.5037	0.0007	1572.1630	-0.0046
22	1578.8442	-0.0016	1571.8889	-0.0041
24	1579.1895	-0.0021
26	1579.5330	-0.0075
28	1579.8881	-0.0044
30	1580.2417	-0.0058
32	1580.5999	-0.0056
44	1569.0899	0.0044
46	1568.8528	0.0036
48	1583.5780	0.0035	1568.6160	0.0001
50	1583.9622	0.0041	1568.3870	0.0016
52	1584.3477	0.0035	1568.1571	-0.0007
54	1584.7357	0.0030	1567.9323	-0.0006
56	1585.1266	0.0031	1567.7050	-0.0057
58	1585.5181	0.0015	1567.4910	0.0001
60	1585.9138	0.0020	1567.2726	-0.0009
62	1586.3086	-0.0003	1567.0556	-0.0026
64	1586.7050	-0.0028	1566.8432	-0.0019
66	1587.1064	-0.0018	1566.6341	0.0004
68	1587.5033	-0.0068	1566.4291	0.0051
70	1587.9094	-0.0038	1566.2201	0.0044
72	1588.3148	-0.0025	1566.0102	0.0016
74	1588.7231	0.0011	1565.7968	-0.0055
76	1589.1335	0.0064

V. DISCUSSION

Although the structure of the C_3O_2 molecule is unquestionably linear, the very low frequency of ν_7 will lead to a mean-squared structure that has significant bending. This bending can have an important influence on the molecular constants, an influence which becomes evident when we compare the data in Table VIII with results on other similar molecules and with results of electron diffraction measurements on C_3O_2 .

The rotational constant B for C_3O_2 can be calculated from

$$B = \frac{\hbar}{8\pi c} \frac{1}{m_{\text{O}} r_{\text{O}}^2 + m_{\text{C}} r_{\text{C}}^2}, \quad (2)$$

where r_{O} and r_{C} are the distances from the rotational axis to the oxygen and carbon atoms, respectively; m_{O} and m_{C} are the corresponding masses. Recently Almenningen *et al.*^{26,27} have measured by electron diffraction each of the atomic separations in C_3O_2 . Using their values at 237 K of $r_{\text{O}} = 2.3867\ \text{\AA}$, half of the O-O distance, and $r_{\text{C}} = 1.2567\ \text{\AA}$, half of the larger C-C distance, we find $B = 0.0766 (\pm 0.0005)\ \text{cm}^{-1}$. This value represents an average over all populated states, and it agrees well with the value of $0.0764\ \text{cm}^{-1}$ obtained by averaging the B'' values in Table VIII. If we had assumed a rigid linear structure with fixed C=C and C=O lengths of $1.2899\ \text{\AA}$ and $1.1645\ \text{\AA}$,^{26,27} respectively, then a value $B = 0.0725\ \text{cm}^{-1}$ would have

resulted, which is substantially below any of the observed values.

An estimate of ω_7 can be obtained from the relation²⁸

$$q_1'' \approx 2B^2/\omega_7. \quad (3)$$

Using the values from Table VIII, Eq. (3) yields $32 \pm 6\ \text{cm}^{-1}$ for the low frequency bending mode, compared with the observed value of $22\ \text{cm}^{-1}$. The agreement is not bad considering the fact that Eq. (3) assumes a harmonic potential for ν_7 , while the actual potential is expected to be highly anharmonic.^{15,27,29,30} The magnitude of q_1'' in the Δ states is expected to be $\approx 10^{-6}\ \text{cm}^{-1}$, i.e., a factor (B/ω_7) smaller than it is in the Π states.²⁸ Thus, within our accuracy, the B values in the Δ states should be equal, which is in agreement with the results in Table VIII.

The magnitude of the centrifugal distortion term D in C_3O_2 appears to be anomalously large compared with other similar molecules. In CO_2 , CS_2 , C_2H_2 , C_2D_2 , and C_2N_2 , all linear symmetric molecules, the D term is given to usually better than 20% accuracy by $4B^3/\omega_s^2$, where ω_s is the lowest symmetric stretching frequency. In C_3O_2 the same expression (with $\omega_s = 800\ \text{cm}^{-1}$) yields a D value at least a factor of ten below the observed

TABLE VII. Observed frequencies (vac. cm^{-1}) of the $\nu_4 + 2\nu_7^d$ $\leftarrow 2\nu_7^d$ transitions. The rms error of the fit using the molecular constants in Table VIII is $\pm 0.0019\ \text{cm}^{-1}$ for 50 lines.

J	$R(J)$ OBS	O-C	$P(J)$ OBS	O-C
2	1583.5863	0.0020
4	1583.8917	0.0024
6	1584.1941	-0.0000
8	1584.4997	0.0009
10	1584.8048	0.0014
12	1585.1071	-0.0006
14	1585.4110	-0.0009
16	1585.7155	-0.0004
18	1586.0180	-0.0017	1580.3698	-0.0051
20	1586.3236	0.0004
22	1586.6261	-0.0004	1579.7620	-0.0005
24	1586.9274	-0.0022	1579.4517	-0.0045
26	1587.2306	-0.0018	1579.1499	0.0001
28	1587.5346	-0.0004	1578.8431	-0.0002
30	1587.8371	-0.0002	1578.5385	0.0019
32	1588.1411	0.0017	1578.2321	0.0023
34	1588.4425	0.0014	1577.9265	0.0036
36	1588.7432	0.0006	1577.6203	0.0046
38	1589.0458	0.0020	1577.3086	0.0003
40	1589.3461	0.0014	1577.0010	0.0003
42	1576.6914	-0.0013
44
46	1576.0724	-0.0033
48	1590.5427	-0.0023	1575.7647	-0.0018
50	1590.8430	-0.0012	1575.4551	-0.0016
52	1591.1426	-0.0005	1575.1456	-0.0008
54	1591.4420	0.0004	1574.8356	0.0003
56	1591.7403	0.0006	1574.5247	0.0013
58	1592.0379	0.0005	1574.2146	0.0040
60	1573.8978	0.0010
62	1573.5797	-0.0022
64	1573.2638	-0.0019
66	1572.9468	-0.0012
68	1572.6309	0.0021

TABLE VIII. Molecular constants (vac. cm^{-1}) for the $\nu_4\ 1 \leftarrow 0$ transitions of $^{12}\text{C}_3\ ^{16}\text{O}_2$.

Transition	ν_0	B''	$B' - B'' (10^{-4})$	$D'' (10^{-8})$	$D' - D'' (10^{-8})$	$H'' (10^{-12})$	$H' - H'' (10^{-12})$
$\nu_4 \leftarrow 0^o$	1587.390 ± 0.002	0.075573 $\pm 2 \times 10^{-5}$ (0.075563) ^a	6.450 ± 0.04	4.183 ± 0.4 (3.817) ^a	1.928 ± 0.3	0.616 ± 0.1 (0.601) ^a	0.192 ± 0.1
$\nu_4 + \nu_7^{1c} \leftarrow \nu_7^{1c}$	1580.896 ± 0.003	0.076052 $\pm 3 \times 10^{-5}$ (0.07605) ^b	4.453 ± 0.08	6.604 ± 2.0	2.020 ± 0.6	7.400 ± 4.0	1.54 ± 1.0
$\nu_4 + \nu_7^{1d} \leftarrow \nu_7^{1d}$	1580.901 ± 0.003	0.076414 $\pm 3 \times 10^{-5}$ (0.07648) ^b	5.345 ± 0.06	3.460 ± 1.5	1.519 ± 0.3	0.849 ± 2.0	0.232 ± 0.2
$\nu_4 + 2\nu_7^{2c} \leftarrow 2\nu_7^{2c}$	1575.094 ± 0.003	0.076913 $\pm 3 \times 10^{-5}$	3.961 ± 0.08	4.226 ± 1.8	0.453 ± 0.4	1.640 ± 2.0	-0.276 ± 0.5
$\nu_4 + 2\nu_7^{2d} \leftarrow 2\nu_7^{2d}$	1575.092 ± 0.003 1575.098 ^c	0.076902 $\pm 3 \times 10^{-5}$	3.958 ± 0.08	4.616 ± 1.8	-0.113 ± 0.4	2.03 ± 2.0	-0.804 ± 0.5
$\nu_4 + 2\nu_7^0 \leftarrow 2\nu_7^0$	1583.126 ± 0.003	0.076327 $\pm 3 \times 10^{-5}$ (0.076280) ^a	-0.154 ± 0.04	4.829 ± 2.0 (1.76) ^a	0.124 ± 0.4	6.25 ± 5.0 (1.58) ^a	-0.29 ± 0.5

^aReference 17.^bAssignment 2, Ref. 16.^cFrom Q branch.

values. This result indicates that the primary effect of centrifugal distortion in C_3O_2 is a straightening out of the molecule, i. e., a stiffening of the restoring force and consequent reduction in amplitude of the ν_7 bending mode, rather than a stretching of the bonds.

The signs of most of the $B' - B''$ terms in C_3O_2 also appear to be anomalous compared with other linear molecules. For example, in the excitation of the asymmetric stretching modes ν_3 of CO_2 or ν_3 of C_2H_2 , the $B' - B''$ terms are all negative, while in a corresponding mode ν_4 of C_3O_2 , all the $B' - B''$ terms are positive, except for the $2\nu_7^0$ state. This result suggests a strong anharmonic coupling of ν_4 and ν_7 , leading to an increased bending mode amplitude in the ν_4 excited state. The same conclusion is supported by the systematic shift of the hot bands to lower frequencies.

Carreira *et al.*¹⁵ have determined a potential function for ν_7 by fitting their far infrared and Raman spectra. It would be of interest to compare the moments of inertia calculated from their model with the data in Table VIII. Also the band contours for a number of Raman and far infrared bands can now be calculated and checked against their assignments.

ACKNOWLEDGMENTS

The authors wish to thank Prof. K. N. Rao for sending us a listing of improved NH_3 line positions prior to publication. The assistance of C. M. Savage and L. P. Breitenbach in obtaining the Fourier transform spectra and of K. F. Yeung in obtaining some of the early laser spectra is gratefully acknowledged. The PbTe films used to make the lasers were grown by M. Mikkor.

¹A recent review of diode laser spectroscopy is given by A. R. Calawa, *J. Lumin.* **7**, 477 (1973).

²K. W. Nill, F. A. Blum, A. R. Calawa, and T. C. Harman,

Chem. Phys. Lett. **14**, 234 (1972).

³G. A. Antcliffe, S. G. Parker, and R. T. Bate, *Appl. Phys. Lett.* **21**, 505 (1972).

⁴F. A. Blum, K. W. Nill, P. L. Kelley, A. R. Calawa, and T. C. Harman, *Science* **177**, 694 (1972).

⁵G. A. Antcliffe and J. S. Wrobel, *Appl. Opt.* **11**, 1548 (1972).

⁶W. H. Weber, P. D. Maker, K. F. Yeung, and C. W. Peters, *Appl. Opt.* **13**, 1431 (1974).

⁷F. A. Blum, K. W. Nill, A. R. Calawa, and T. C. Harman, *Chem. Phys. Lett.* **15**, 144 (1972).

⁸W. Engler and K. W. F. Kohlrausch, *Z. Phys. Chem. Abt. B* **34**, 214 (1936).

⁹R. C. Lord and N. Wright, *J. Chem. Phys.* **5**, 642 (1937).

¹⁰D. A. Long, F. S. Murfin, and R. L. Williams, *Proc. R. Soc. London A* **223**, 251 (1954).

¹¹H. D. Rix, *J. Chem. Phys.* **22**, 429 (1954).

¹²B. P. Stoicheff, *Adv. Spectrosc.* **1**, 124 (1959).

¹³F. A. Miller and W. G. Fateley, *Spectrochim. Acta* **20**, 253 (1964).

¹⁴W. H. Smith and G. E. Leroi, *J. Chem. Phys.* **45**, 1767 (1967).

¹⁵L. A. Carreira, R. O. Carter, J. R. Durig, R. C. Lord, and C. C. Millionis, *J. Chem. Phys.* **59**, 1028 (1973).

¹⁶W. J. Lafferty, A. G. Maki, and E. K. Plyler, *J. Chem. Phys.* **40**, 224 (1964).

¹⁷A. W. Mantz, P. Connes, G. Guelachvili, and C. Amiot, *J. Mol. Spectrosc.* **54**, 43 (1975).

¹⁸W. H. Weber and K. F. Yeung, *J. Appl. Phys.* **44**, 4991 (1973).

¹⁹G. P. Montgomery, Jr. and J. C. Hill, *J. Opt. Soc. Am.* **65**, 579 (1975).

²⁰*American Institute of Physics Handbook*, edited by Dwight E. Gray (McGraw-Hill, New York, 1972), 3rd ed., Chap. 6, p. 30.

²¹J. S. Garing, H. H. Nielsen, and K. N. Rao, *J. Mol. Spectrosc.* **3**, 496 (1959).

²²A listing of improved line positions was provided by K. N. Rao (private communication, Sept. 1974).

²³E. A. Cohen and R. L. Poynter, *J. Mol. Spectrosc.* **53**, 131 (1974).

²⁴P. D. Maker, H. Niki, C. M. Savage, and L. P. Breitenbach, *Proceedings of the Thirtieth Symposium on Molecular Structure and Spectroscopy*, Columbus, Ohio, 16-20 June, 1975, p. 150.

²⁵G. Herzberg, *Infrared and Raman Spectra* (Van Nostrand,

New York, 1945), Chap. IV, p. 380.

²⁶A. Almenningen, S. P. Arnesen, O. Bastiansen, H. M. Seip, and R. Seip, *Chem. Phys. Lett.* **1**, 569 (1968).

²⁷A. Clark and H. M. Seip, *Chem. Phys. Lett.* **6**, 452 (1970).
We have made the corrections to the data in Ref. 26 indicated in this paper.

²⁸See, for example, C. H. Townes and A. L. Schawlow, *Microwave Spectroscopy* (McGraw-Hill, New York, 1955), p. 33.

²⁹F. A. Miller, D. H. Lemmon, and R. E. Witkowski, *Spectrochim. Acta* **21**, 1709 (1965).

³⁰R. L. Redington, *Spectrochim. Acta Part A* **23**, 1863 (1967).