

Microstructure of BaRuO₃ thin films grown on (001) SrTiO₃

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BaRuO₃ thin films with hexagonal 4*H* structure were grown on (001) SrTiO₃ by a 90° off-axis rf-sputtering technique. The thin films were epitaxially grown on the (001) surface of SrTiO₃, with (2023) planes parallel to the surface of the substrate. Within the growth plane, the film consists of four different crystallographic orientations with respect to the substrate, defined by the surface symmetry of the (001) SrTiO₃ substrate. BaRuO₃ grains of all four orientations show an anisotropic shape elongated along the $[\bar{1}2\bar{1}0]$ direction. The reason for the anisotropic growth is that the lattice mismatch between BaRuO₃ and SrTiO₃ is smaller along the $[\bar{1}2\bar{1}0]$ direction of SrTiO₃ in comparison to that along its perpendicular direction. Stacking faults and intergrowths of the 9*R* structure were observed in small local regions of the film. © 2000 American Institute of Physics. [S0003-6951(00)03833-X]

Ternary ruthenium oxides (ARuO₃; A = Ba, Sr, and Ca) have attracted increasing interest due to their unique electrical and magnetic properties, which exhibit potential applications in microelectronic devices.¹⁻³ The structure of ARuO₃ can be considered as a close-packed stacking of AO₃ layers in which the atomic configuration is similar to the (111) atomic plane of the cubic perovskite. The cubic close-packed stacking of AO₃ layers results in a cubic-like perovskite. In this perovskite structure, oxygen atoms form octahedra which are corner shared and filled by small Ru atoms. CaRuO₃ and SrRuO₃ belong to this structure.⁴ This cubic-like structure gives rise to oxygen intermediate Ru–O–Ru bonding. In contrast, a hexagonal close-packed stacking of AO₃ layers will lead to a hexagonal structure (2*H*). In the hexagonal 2*H* structure RuO₆ octahedra are only face shared between different layers, giving rise to direct Ru–Ru bonding. In addition to these two extreme cases, a number of polymorphic structures can be found by mixing cubic and hexagonal stacking sequences of AO₃ layers. For instance, three different hexagonal phases were reported for BaRuO₃.^{5,6} A four-layer hexagonal structure of BaRuO₃(4*H*) can be formed by stacking AO₃ layers with a sequence of CHCHCH where “C” and “H” represent cubic and hexagonal close-packing, respectively. Similarly, the six-layer hexagonal phase of BaRuO₃(6*H*) displays a stacking sequence of CCHCCH, and the nine-layer rhombohedral phase (9*R*) has a stacking sequence of CHHCHHCHH. It was found that the 9*R* phase is the most stable phase existing at atmospheric pressure with a space group of $R\bar{3}m$ (No. 166) and lattice constants $a = 5.75 \text{ \AA}$ and $c = 21.6 \text{ \AA}$.⁶ As pressure is increased, the 9*R* structure transforms through the 4*H* structure with a space group $P6_3/mmm$ (No. 194) and lattice constants $a = 5.73 \text{ \AA}$ and $c = 9.5 \text{ \AA}$ (Ref. 7) into the 6*H* structure with a space group $P6_3/mmm$ and lattice constants $a = 5.71 \text{ \AA}$ and $c = 14.0 \text{ \AA}$.⁸ A cubic-like perov-

skite polymorph of BaRuO₃ was predicted to exist as pressure is increased to approximately 120 kbar.⁸

Although the 4*H* phase is considered to be metastable, it has been successfully synthesized utilizing single-crystal-growth methods.^{7,9} Recently, Lee *et al.*¹⁰ reported the synthesis and electrical transport and magnetic properties of epitaxial thin films of *c*-axis oriented metastable 4*H* BaRuO₃ thin films on (111) SrTiO₃ substrates. It was found that the electrical properties of the 9*R* phase differ significantly from the 4*H* phase below ~100 K. This variation of electrical properties was ascribed to the different percentage of direct Ru–Ru bonds and oxygen intermediate Ru–O–Ru bonds in the 9*R* and 4*H* phases.⁹ To separate the roles of direct Ru–Ru and Ru–O–Ru bonds in determining the physical properties, it is desirable to synthesize BaRuO₃ with cubic-like and pure hexagonal close-packed structures.

It is well accepted that structural stabilization through thin-film epitaxy is a feasible way to form metastable phases of material.¹¹ Recently, Fukushima *et al.*¹² reported epitaxial growth of pseudocubic perovskite BaRuO₃ thin films on (001) SrTiO₃ substrates by means of rf sputtering. However, both x-ray diffraction¹³ and the present transmission electron microscopy (TEM) studies of BaRuO₃ thin films, which were grown on (001) SrTiO₃ by 90° off-axis rf sputtering, revealed that the film has the 4*H* or/and 9*R* hexagonal structure rather than the pseudocubic perovskite structure.

In the present work, BaRuO₃ films were grown on (001) SrTiO₃ substrates using 90° off-axis sputtering. Details of the growth condition can be found elsewhere.¹⁰ The TEM specimens were prepared by conventional cross-section TEM sample preparation procedures which ended with ion milling TEM studies were conducted with a JEOL 4000EX high-resolution transmission electron microscope (HRTEM) which has a point resolution of ~0.17 nm.

Figure 1(a) is a low-magnification bright-field TEM image showing a cross-section view of a BaRuO₃ thin film grown on (001) SrTiO₃. The mean-film thickness was determined to be ~320 nm. As shown in Fig. 1(a), the film has a

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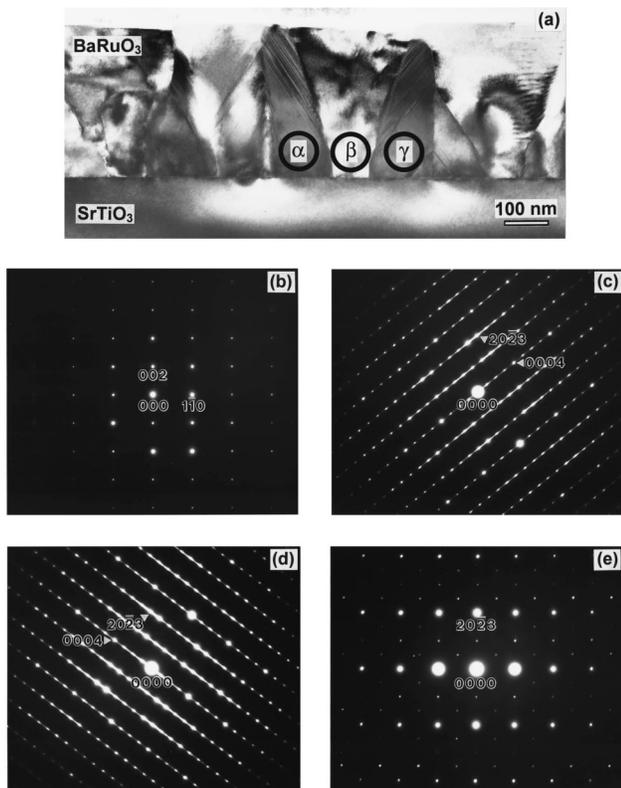


FIG. 1. (a) Bright-field cross-section TEM image of a BaRuO₃ film showing different columnar grains. (b) Electron-diffraction pattern along the [110] zone of the SrTiO₃ substrate. (c) $[\bar{1}2\bar{1}0]$ zone diffraction pattern of 4H BaRuO₃ recorded from the grain marked by α in (a). (d) $[\bar{1}2\bar{1}0]$ zone diffraction pattern taken from the grain marked by γ in (a). (e) $[\bar{3}0\bar{3}4]$ zone diffraction pattern from the grain marked by β in (a).

sharp interface with the substrate, but it shows a rough surface with the amplitude of height fluctuation of ~ 60 nm. It is also seen that the film consists of columnar grains with a mean size of ~ 300 nm, which is comparable with the film thickness. To identify the crystallographic orientation relationships between these BaRuO₃ grains and the SrTiO₃ substrate, selected-area electron-diffraction studies were carried out. Figure 1(b) is an electron-diffraction pattern taken from the SrTiO₃ substrate along the [110] zone axis. Figures 1(c), 1(d), and 1(e) are electron-diffraction patterns taken from areas marked by α , γ , and β in Fig. 1(a), respectively. Figures 1(c) and 1(d) are identified to be the $[\bar{1}2\bar{1}0]$ and $[1\bar{2}10]$ zone diffraction patterns of the hexagonal 4H structure of BaRuO₃, respectively. This means that grain α has a rotation of 180° around the film normal with respect to grain γ . Figure 1(e) was determined to be either the $[\bar{3}0\bar{3}4]$ or $[30\bar{3}\bar{4}]$ zone-axis diffraction pattern of the 4H BaRuO₃. The weak spots in Fig. 1(e) are due to dynamic double diffractions of the crystal and from high-order Laue zones. By carefully checking the whole specimen, the electron-diffraction pattern of each grain observed in the specimen can be identified to be one of these four types of diffraction patterns. Comparing the diffraction patterns of BaRuO₃ films [Figs. 1(c)–1(e)] with that of the SrTiO₃ substrate [Fig. 1(b)], it is concluded that all grains in the BaRuO₃ film are aligned with their (20 $\bar{2}$ 3) plane parallel to the (001) surface of the substrate. This indicates that the BaRuO₃ thin film has the hexagonal 4H structure and grows with the (20 $\bar{2}$ 3) plane parallel to the (001) surface of SrTiO₃.

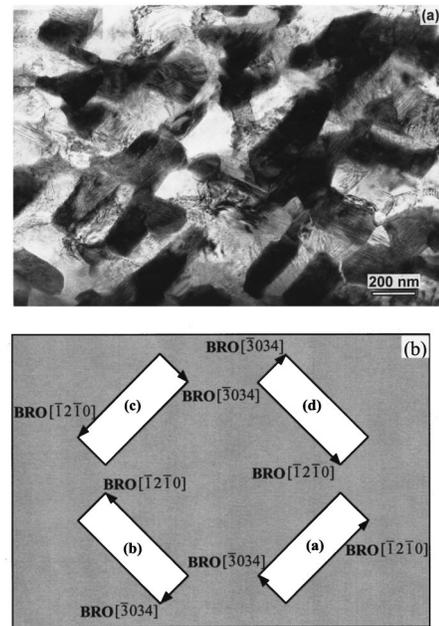


FIG. 2. (a) Plan-view TEM micrograph showing the microstructure of a BaRuO₃ film grown on (001) SrTiO₃. Note the elongated shape of the BaRuO₃ grains. (b) Schematic showing the BaRuO₃ grain arrangements on the (001) surface of SrTiO₃. The $[\bar{3}0\bar{3}4]$ and $[\bar{1}2\bar{1}0]$ direction of BaRuO₃ grains are parallel to the [110] and $[\bar{1}\bar{1}0]$ direction of SrTiO₃, respectively.

BaRuO₃ grains with the $[\bar{3}0\bar{3}4]$ or $[30\bar{3}\bar{4}]$ direction parallel the SrTiO₃ [110] direction cannot be distinguished in the $[\bar{3}0\bar{3}4]$ and $[30\bar{3}\bar{4}]$ electron-diffraction patterns taken from a TEM specimen cut along (110) SrTiO₃, because these two patterns appear the same, as shown in Fig. 1(e). However, these two orientated grains can be unambiguously distinguished from each other by tilting the specimen within the TEM. Alternatively, they are distinct in the electron-diffraction pattern of a specimen cut from $(\bar{1}10)$ SrTiO₃, in which the $[\bar{1}2\bar{1}0]$ and $[1\bar{2}10]$ zone-axis diffraction patterns are obtained, as shown in Figs. 1(c) and 1(d). Detailed TEM studies of both specimens showed that these four differently oriented grains in the film are randomly distributed with the same volume fraction. This was further confirmed by TEM observations of a plan-view specimen. Figure 2(a) is a plan-view TEM image of the same film as in Fig. 1. BaRuO₃ grains in the film are elongated along the $[\bar{1}2\bar{1}0]$ direction in the plane, as shown in Fig. 2(b). This implies the existence of an anisotropic driving force for BaRuO₃ grain growth on (001) SrTiO₃.

It should be noted that the diffraction spots in Figs. 1(c) and 1(d) are strongly streaked along the c direction of the 4H hexagonal structure. This results from a high density of stacking faults existing in some local regions of the BaRuO₃ film. The stacking faults appear as fine fringes in the cross-section TEM images, as shown in Fig. 1(a). Furthermore, high-resolution transmission electron microscopy studies reveal that the atomic stacking sequence in many (HRTEM) faulted regions, where stacking faults are periodically distributed along the c axis, is in accord with that of the 9R structure. Therefore, the intergrowth of the 9R phase occurs occasionally in some small regions. Figure 3 shows a cross-section HRTEM image of a highly defective region in the film. The isolated stacking faults marked by arrows are in fact one third or two thirds of one unit cell of the 9R struc-

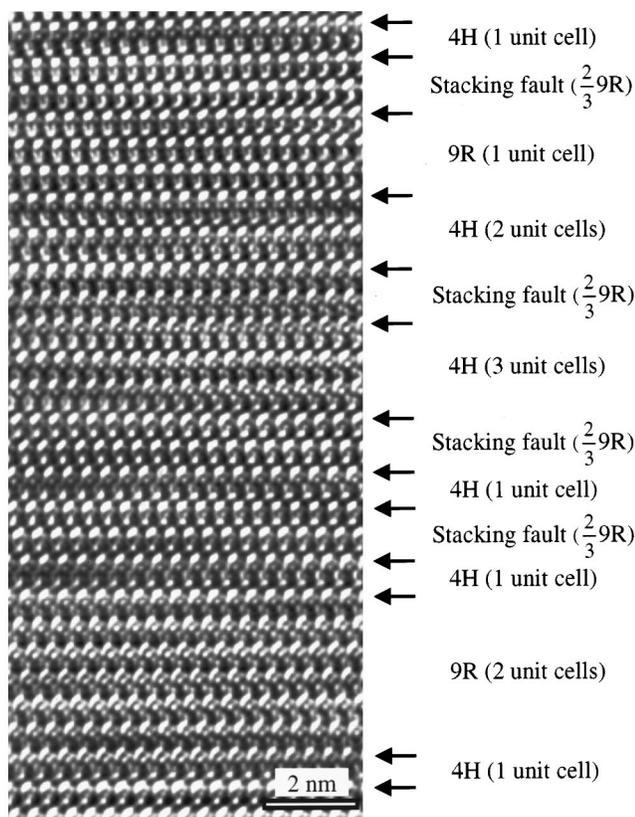


FIG. 3. HRTEM image showing the existence of stacking faults in a small region of the BaRuO₃ film. Segments with a thickness of $2/3$, and one and two unit cells of the 9R BaRuO₃ structure along the c axis are marked.

ture along the c axis. As a result of the connection of these stacking faults, 9R phases of one or two unit cells along the c axis exist, as indicated by the arrows in Fig. 3. The occurrence of stacking faults and intergrowths of the 9R structure in the film is most likely due to the subtle difference between the formation enthalpies of 9R and 4H phases. The intergrowths are favorable because the free energy reduces with increasing the entropy in the system by the randomly layered mixture of 9R and 4H phases. However, the intergrowths of the 6R structure were not observed. This may suggest that the 6R phase has a greater formation enthalpy than the 4H and 9R phases. This is consistent with a previous study³ which shows that the 6R structure is the most unstable polymorph of BaRuO₃.

The film microstructure and grain orientation relationships observed in TEM can be understood by considering the crystallographic structure of the BaRuO₃ film and the SrTiO₃ substrate. SrTiO₃ is a cubic perovskite with the space group $Pm\bar{3}m$ and lattice constant 3.905 Å. The (001) surface of SrTiO₃ has a fourfold rotation symmetry around the surface normal. When the 4HBaRuO₃ phase forms on the (001) surface of SrTiO₃ with (20 $\bar{2}$ 3) parallel to the substrate surface, there are four different crystallographic orientations in the growth plane:

- (a) BaRuO₃[$\bar{3}034$] \parallel SrTiO₃[110] and BaRuO₃[$\bar{1}2\bar{1}0$] \parallel SrTiO₃[$1\bar{1}0$],
- (b) BaRuO₃[$\bar{3}034$] \parallel SrTiO₃[$\bar{1}10$] and BaRuO₃[$\bar{1}2\bar{1}0$] \parallel SrTiO₃[110],

- (c) BaRuO₃[$\bar{3}034$] \parallel SrTiO₃[$\bar{1}\bar{1}0$] and BaRuO₃[$\bar{1}2\bar{1}0$] \parallel SrTiO₃[$\bar{1}10$], and
- (d) BaRuO₃[$\bar{3}034$] \parallel SrTiO₃[$1\bar{1}0$] and BaRuO₃[$\bar{1}2\bar{1}0$] \parallel SrTiO₃[$\bar{1}\bar{1}0$].

These orientation relationships are in agreement with TEM observations described previously. These four orientations are crystallographically equivalent, thus, they should have an equal probability to appear in the thin film. The observed in-plane elongated grain growth may be a result of the different lattice mismatches along the two orthogonal directions between the BaRuO₃ film and the SrTiO₃ substrate. Periods of BaRuO₃[$\bar{1}2\bar{1}0$] and SrTiO₃[110] are 0.573 and 0.552 nm, respectively. Moreover, the atomic arrangements along BaRuO₃[$\bar{1}2\bar{1}0$] are similar to those along SrTiO₃[110]. In contrast, the period of BaRuO₃[$\bar{3}034$] is 4.83 nm and the atomic arrangements along this direction differ significantly from those along SrTiO₃[110]. BaRuO₃ should prefer growing along the direction which leads to a lower interfacial misfit energy. As a consequence, the BaRuO₃ grains grow with an elongation along [$\bar{1}2\bar{1}0$].

In conclusion, it was found that the BaRuO₃ thin film grown on (001) SrTiO₃ by 90° off-axis sputtering has the hexagonal 4H structure. Transmission electron microscopy studies revealed that the films are grown epitaxially with the (20 $\bar{2}$ 3) plane parallel to the (001) surface of the SrTiO₃ substrate. In the growth plane there exists four crystallographic equivalent orientations defined by the fourfold rotation symmetry of the (001) SrTiO₃ substrate surface. It was also found that BaRuO₃ grains show an elongation along the [$\bar{1}2\bar{1}0$] direction which results from the nature of structural mismatches between the BaRuO₃ grains and the SrTiO₃ substrate. Stacking faults and intergrowths of the 9R structure were observed in small local regions of the film.

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