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COMPUTER SIMULATION OF A LIVING CELL:
MULTILEVEL CONTROL SYSTEMS

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SUMMARY

A simple bacterial cell (Escherichia coli) has been modeled, and the input-output behavior of the model has been simulated as a program in FORTRAN IV for an IBM 360/67 digital computer.

Automata theoretic analysis of the homomorphic model underlying the computer simulation enables us to investigate the information content and complexity of the simulation and of the measurement space representing our data base.

The simulated cell is able to adjust its enzymes and DNA to grow in different chemical environments using allosteric modification of enzymes, and repression of RNA synthesis. It grows at realistic rates and achieves limited metabolic stability.

INTRODUCTION

Availability of experimental and theoretical analyses of the systems operative in living and reproducing organisms, as well as excellent presentations of powerful and convenient computer techniques make a computer simulation of a living cell a logical endeavor at this time.

In particular much is known about the biochemical behavior of bacterial cells. [Biological laboratory studies have been done elucidating control of DNA replication in bacteria (Eberle and Lark, 1969; Clark, 1968). The 1968 Cold

Spring Harbor Symposium in Quantitative Biology was devoted to papers concerning replication of DNA in micro-organisms (edited by Frisch, 1969). Lark has reviewed initiation and control of DNA synthesis (1969), and the subject is covered in books by Mandelstam and McQuillen (1968), Hayes (1968), and Davis et. al. (1968).

Control of Enzyme production by repression of messenger RNA production by the DNA is still being studied (Umbarger, 1969), and original hypotheses modified. Modification of enzymes already present in the cell, as a control mechanism, is reviewed by Umbarger (1969), and Batta (1960), and the relationship of this type of control to energy relationships in the cell has been reviewed by Atkinson (1966), discussed in *Control of Energy Metabolism* edited by Chance et. al. (1965), and investigated by Murray and Atkinson (1968). The importance of the ATP/ADP ratio in energy control relationships is becoming apparent in these works. These subjects, as well as convenient presentation of data on metabolic pathways, appear in books by Bernhard (1968), Westley (1969), Reiner (1968, 1969) and Mahler and Cordes (1966).

Mathematical analyses of enzyme modification utilizing computer techniques have been published by Walter (1969a, 1969b), Cennamo (1969), Griffith (1968), Heinmets (1964), and Yeisley and Pollard (1964). Relationships have been drawn between regulatory mechanisms in microbial cells, and in higher cells by Mitchison (1969), Tsanev and Sendov (1969), Comings and Kakefuda (1968), Britten and Davidson (1969), Gause (1966), and Heinmets (1966), making plausible the extension of computer simulation studies of a bacterial cell to studies of cancer in higher organisms (Tsanev and Sendov, 1969). Weinberg and Berkus (1969), Weinberg (1969a, b, c) and Stahl (1967) have modeled living cells as computer

programs. Formal aspects of self-reproducing systems are described in Waddington (1969), Burks (1969), Codd (1968), Mesarovic (1968), and von Neumann and Burks (1966). Techniques for automata theoretic numerical analysis, and also computer simulation are well described in Gordon (1969), Mize and Cox (1968), IBM Corporation Scientific Subroutine Package (1969), Knuth (1969), Wendroff (1969), Ginzburg (1968), Kalman, et. al. (1969) and Ulam (1966).

The connection between molecular controls and evolutionary mechanisms has been outlined for a computer simulation by Weinberg and Berkus (1969) taking advantage of basic genetic mechanisms (Strickberger, 1968, Kimura, 1964), and theoretical analyses connecting econometric studies (Gale, 1967) to a general theory of adaptive systems (Holland, 1969a, b).]

The present computer simulation of a living cell is the first effort to compare the predictions of hypotheses concerning a complete, functional cell with detailed laboratory data.

In our attempt to model a simple cell at the biochemical level we had to confront the complexity of the metabolic pathways present in even the simplest of cells. More than 3,000 different kinds of molecules are present in a complex spatial and functional relationship. This complexity had to be drastically reduced to permit a simulation involving five hundred instructions in FORTRAN IV for a 360/67 digital computer. To construct our simulation we lumped molecules into pools, and considered 1) metabolic topology, 2) functional relationships among cell structures and chemicals, and 3) type of experimental data available in the literature. This process of lumping is further described and justified from a systems viewpoint in the appendix.

Metabolic topology was considered, and an attempt was made to lump together as a single entity only those molecules which could be drawn adjacent to one

another on the metabolic map (Figures 1, 2). For example, in a pathway $A \rightarrow B \rightarrow C$, chemicals, A, B, C, might be partitioned into A, and BC, but would not be lumped into AC, and B.

Functional relationships between groups of molecules were extremely important, and the molecules lumped together in any one model entity were, in some way, functionally a unit (Figure 3). Thus, all the molecules produced in the breakdown of sugar to CO_2 and water to produce energy were lumped in this model since they could be considered functionally as molecules intermediate in a chemical pathway used to produce energy. Later models will employ more refined partitions, for example to capture the subtle and important relationships between molecules at different points in the glycolytic pathway citric acid cycle and cytochrome system.

Experimental data were often available for large chemical pools, e.g. products of glycolysis and the citric acid cycle, making these separate entities logical candidates for grouping into single entities in this model of the cell.

DESCRIPTION OF THE MODEL

The simulation can be described by its state together with its transition function (Figures 1-5). The state of the cell is described 1) by the concentrations of thirty internal chemical pools, 2) by the genetic apparatus, and 3) by the cell volume. The transition function used to obtain the state of the cell at the next time step of the simulation from a given time step consists of difference equations and Boolean expressions describing 1) enzyme catalyzed chemical reactions, 2) allosteric modification of enzymes, 3) repression of

messenger RNA-production, 4) self-replication of DNA under genetic controls, and 5) permeability of the cell to the chemical pools represented in the simulation (Figure 3). A detailed description of an earlier version of the simulation is available (Weinberg and Berkus, 1969b).

Input to the simulation consists of the concentrations of chemicals in the liquid environment in which the cell is growing. Output from the simulation consists of state descriptions during successive time increments. Comparison between simulation output and experimental data from the real world (Figures 5-15) enables us to judge the validity of the hypotheses used to write the simulation, to modify the hypotheses used to write the simulation in order to make the simulation more realistic, and to suggest critical real world experiments.

The equations in the transition function are general, thus allowing us to simulate cell behavior in many different environments, and in changes from one environment to another. This is especially significant since we wish to test the stability of the model under a variety of conditions.

RESULTS OF THE SIMULATION

The environment simulated was liquid growth medium at a temperature of 37 degrees Centigrade, with an abundance of oxygen. The simulated experiments presented were in 1) minimal medium (medium containing glucose, ammonium salt, and minerals) and 2) broth (a rich medium containing additional amino acids, nucleosides and vitamins (Weinberg and Berkus, 1969b)). The simulated cell grew faster in broth than in minimal medium, in agreement with laboratory data. This is a reasonable result, since addition of growth products implies that there are fewer molecules to be synthesized by the cell itself.

The simulated cell produced chemicals and cell mass at a logarithmic rate, but duplicated in a stepwise fashion (Figure 5) just as the real cell does. Since the simulated cell produced these smooth growth curves from a complex interaction of many equations, the growth curves are a good preliminary confirmation of the models used to write the simulation.

The simulated cell employed repression to control the production of its enzymes (Figure 21, Appendix). Repression operated at the DNA level. For example EK2 was the enzyme pool needed for producing amino acids from carbohydrates. EK2 was produced under control of DNA by way of the RNA pools as long as the amino acid pool concentration was below a certain critical level. DNA directed the production of messenger RNA specific for the production of EK2. EK2 was produced by hooking together amino acids attached to transfer RNA. This hooking was done by messenger RNA attached to ribosomes. If the amino acid level rose above the critical level, production by DNA of messenger RNA responsible for EK2 production was sharply curtailed by the nature of the Boolean equations in the transition function. The messenger RNA already present rapidly decayed, and almost no new messenger RNA for EK2 production was formed.

Of course if the amino acid concentration fell too low, insufficient amino acids were available for hooking together into the EK2 enzyme; production of all enzymes was blocked in the event of extreme scarcity of amino acids as a result of the form of the differential equations concerning their production.

The simulated cell employed feedback inhibition to control the activity of the enzymes already present (Weinberg and Berkus, 1969b). For example, EK2,

the enzyme for production of amino acids from carbohydrates, appeared in three different forms: pure enzyme, enzyme with one molecule of amino acid attached to it, and enzyme with two molecules of amino acid attached to it. These three forms of EK2 had different catalytic ability. The relative amount of EK2 in each form determined the activity of the EK2 present in the cell in terms of its efficiency in converting carbohydrate into amino acids. The percentage of EK2 in each of the three forms was determined by the number of amino acid molecules per cell volume unit (one cell volume unit was taken as the volume of a cell growing rapidly in mineral glucose medium with ammonium salt). The higher the amino acid concentration in the simulated cell, the greater was the percentage of EK2 in its low activity form, and the less effective was the EK2 in production of amino acids from carbohydrates.

We tested the ability of the simulated cell to grow in a medium it had never "seen" before by simulating a shift down to low glucose minimal medium with 10^{-4} times the usual glucose concentration found in minimal medium (Figure 15). The simulated cell decreased its growth rate in this shift down, just as the real cell does (Moser, 1958).

The cell could also adjust to shifts up from minimal medium to broth, and back down from broth to minimal medium when it was using its feedback controls. It is significant that without feedback controls the orderly shift up from minimal medium to broth was not possible (Figures 6-10).

The simulation experiments to determine the function of the feedback controls was performed as follows: the simulated cell was shifted down from growth in broth to growth in minimal medium, and growth was followed for ten seconds. The results of the shift were plotted along with a broth control for

the simulated behavior obtained with and without feedback equations. Similarly, shifts up from minimal medium to broth were studied. Each graph represents measurements of some pool (such as ATP) during the shift, and during the corresponding control run.

The simulated data agreed well with laboratory data when feedback controls were present (Figures 6-10) but the simulated cell without its feedback controls was no longer able to realistically handle shifts up from minimal medium to broth (Figures 11-14). This suggested that feedback in the real cell was evolved to handle shift up situations since the normal pathways are not stable in this condition without feedback.

The simulated cell with feedback controls maintained stability through rapid oscillation of concentrations about equilibrium points, a phenomenon well known in the literature on feedback control.

Oscillation of concentration to maintain equilibrium was strikingly illustrated by ATP and ADP concentrations during a shift up from minimal medium to broth (Figures 6, 7). The "restoring force" effected by the feedback equations enabled the cell to maintain equilibrium concentrations of ATP and ADP. In contrast the concentration of ATP was too high and the concentration of ADP was too low after a simulated shift up without feedback controls in the simulated cell.

Similarly an overshoot in ribosomal RNA concentration was quickly corrected by the simulated cell with feedback controls, while a similar simulation experiment on a cell without feedback controls produced so much ribosomal RNA that a real cell would lyse (Figure 10).

CONCLUSIONS

Preliminary conclusions drawn are 1) Shifts from poor to rich medium are more of a challenge to the cell than shifts from rich to poor medium. The shift up to rich medium requires elaborate feedback control mechanisms, whereas the shift down to poor medium does not require feedback control as strongly, but can be handled by metabolic topology. 2) Oscillations occur about equilibrium concentrations; fixed equilibrium concentrations are not maintained in the simulation. However, this agrees with experimental observations which indicate that real cells are constantly oscillating biochemical systems and suggests, indeed, that the oscillating concentrations produced by feedback control systems are necessary for the flexibility characteristic of living systems.

APPENDIX

Formally, in constructing a model on which to base a simulation a homomorphic mapping is often used to reduce a complex system description to a relatively simple one (Figure 17). Such a homomorphism is essentially a partition of the state space of the real system which preserves the transition function of the system just as a group homomorphism preserves the group multiplication. In practice, the mapping between real system and model is never truly a homomorphism--indeed when divergence between the behavior of the partitioned real system and the behavior of the model is detected this may initiate a search for more truly homomorphic and hence more adequate models (Ulam, 1966)

The kind of partition considered may depend on several factors. The measuring instruments with which experimental observations are made impose certain equivalences which cannot be cut across. The partition must relate to experimental data which is actually or potentially available. It must also be fine enough to maintain distinctions between the parts of the system which are of primary interest.

When a simulation by computer is involved, as it is in this paper, additional restrictions are placed on the kind of model that can be considered. The model must not exceed the information processing capabilities of real computers. Suppose for example we momentarily consider describing the state of a biological cell by listing the states of each of the elementary atomic particles of which the cell is composed. The sheer enormity of the number of such particles would relegate such an approach to the realm of wishful thinking in two ways: one, we would hardly have enough data storage capacity to keep track of such a long list, and two, we would not have enough program storage capacity to specify how each atomic state changes as a function of the prior states of the atoms which

influence it. (Moreover, the time required to run such a program would exceed a scientist's patience if not his lifetime).

This paper concerns a model of a bacterial cell (E. coli) which has been constructed to enable computer simulation of the cell behavior in its living form. We believe that our model is best understood as an attempt, operating under the constraints on model making just mentioned, to achieve a truly homomorphic mapping of a real world system. (We suggest that it is fruitful to deal with the modelling process in general in this way but we do not further argue this proposition.) Accordingly, we devote some time to an exposition of the system theoretic concepts underlying the idea of models as attempted homomorphisms. This development is briefly sketched here and will be more fully expanded in subsequent publications (Weinberg and Zeigler, to be published).

In its most basic form, a system is defined as a set of states S , together with a transition function $\tau: S \rightarrow S$. τ describes the behavior of the system over time by indicating which next state is to follow the present state. Thus if the state at time t is $s(t)$ then the state at time $t+1$, $s(t+1) = \tau(s(t))$.

The state space S is usually described as a cartesian product of component state sets i.e. $S = \prod_{\alpha \in D} S_{\alpha}$ where D is called a set of co-ordinates (or entities) and S_{α} is the state set (or attribute set) of co-ordinate α . In Figure 1 we list a number of possible state spaces and indicate the form a transition function might take in each. In Figures 2 and 3 we specify in more detail the state space and transition function of the present model.

A homomorphism from a system (S, τ) to a system (S', τ') is a map h from S onto S' such that for all $s \in S$

$$h(\tau(s)) = \tau'(h(s))$$

Thus, a homomorphism preserves the transition function and guarantees that every

state trajectory in (S, τ) has a corresponding state trajectory in (S', τ') .

As we have stated, model making may be identified as an attempt to obtain a truly homomorphic image of a real system which among other things, is simple enough to implement on a computer. The process of going from lower level to higher level functional units may be viewed in this light. We cannot implement an atomic state model of an E. coli (even though presumably such a simulation would be maximally informative) because such a model would require an information processing capacity well beyond that possessed by any man made computer. At the molecular and concentration levels of Figure 18 the same thing would be true. Notice that each of these levels arises by grouping together co-ordinates at a lower level to form higher level units. It is only by continuing this process one more step that we are able to arrive at models of a cell simple enough to implement on a computer. It is still possible at this level to construct models which can be meaningfully tested against real cellular behavior as consideration of our present simulation has demonstrated.

Partitioning the co-ordinates of system to achieve a simpler system can be given a mathematical formulation (Zeigler, to be published). We indicate the conditions under which such a partition will yield a homomorphic image system. We also show how the complexity of the system (as determined by measures relevant to computer implementation) can be made to decrease in this way. The point to be made, however, is that such a partitioning can be justified mathematically and experimentally at any level, not only the levels-- atomic, molecular, etc.--traditionally accepted.

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Figure 1. *Natural Groupings. Chemical species in one large block were represented in the simulation as one pool.*

Figure 2. *Flow of Materials.*

Figure 3. *Model of a Living Cell Used for the Computer Simulation.*

Figure 4. *Differential Equations. Quantity to the left of = is the change in amount of the substance; e. g., DDNA represents the change in the amount of DNA in one time increment DT. The differential equation underlying the first equation is*

$DDNA = K(6) * NUC * EK(6) * ATP * DT$ for a discrete time interval DT . As DT approaches 0, we get the underlying continuous differential equation

$$\lim_{DT \rightarrow 0} D(DNA)/DT = d(DNA)/dt = K(6) * NUC * EK(6) * ATP$$

Figure 5. *Logarithm of Various Quantities in a Growing Culture. Logarithmic Increase in Cell Mass over Time, Stepwise Increase in Number of Cell. Comparative magnitude of various quantities are a function of the scaling factors used in order to plot all quantities on one graph. A cell doubles after a DNA replication cycle. The doubling takes a relatively short time, as indicated by the sudden, stepwise increase in "TOTAL NUMBER OF CELLS", where as "TOTAL DNA" increases throughout the replication cycle.*

Figure 6. *ATP during Simulated Shift Up.*

Figure 7. *ADP during Simulated Shift Up.*

Figure 8. *DNA during Shift Up. Well aerated liquid cultures at 37°C were used for simulated and laboratory data. Symbols: □○ simulated shift from minimal glucose to broth at time zero; ○ simulated minimal . X laboratory shift up; X laboratory minimal glucose.*

Figure 9. *Protein during Shift Up.* Well aerated liquid cultures were used for laboratory and experimental data. Symbols: \square \bigcirc simulated shift from minimal glucose to broth at time zero; \bigcirc simulated minimal glucose control; \times laboratory shift up; $+$ laboratory minimal.

Figure 10. *Ribosomal RNA during Shift Up.* Well aerated liquid cultures at 37°C were used for laboratory and simulated data. Symbols: \bigcirc simulated minimal glucose liquid medium; \square \bigcirc simulated shift from broth to minimal glucose at time zero; \times laboratory shift up; $+$ laboratory minimal glucose.

Figure 11. *ATP during Simulated Shift Down.*

Figure 12. *DNA during Shift Down.* Well aerated liquid cultures at 37°C were used for laboratory and experimental data. Symbols: \bigcirc simulated shift down from broth to minimal glucose at time zero; \bigcirc broth control; \times laboratory shift down; \times laboratory control.

Figure 13. *Ribosomal RNA during Shift Down from Broth to Minimal Glucose.* Well aerated liquid cultures at 37°C were used for simulated and laboratory data. Symbols: \bigcirc nutrient broth; \bigcirc shifted to minimal glucose from nutrient broth at time zero; \times laboratory shift down; \times laboratory control.

Figure 14. *Protein/Cell during Shift Down.* Well aerated liquid cultures at 37°C were used for laboratory and simulated data. Symbols: \bigcirc simulated shift from broth to minimal glucose at time zero; \bigcirc \diamond simulated broth; $+$ laboratory shift down; \times laboratory broth.

Figure 15. *Simulated Growth in Low Glucose Concentration.* Well aerated liquid cultures at 37°C were used for simulated and laboratory data. Symbols: all cultures were grown in minimal medium; \square laboratory, 40 mg glucose per liter; Δ laboratory, $4 \cdot 10^{-4}$ mg glucose per liter; \bigcirc simulated 4 mg glucose per liter to $4 \cdot 10^{-4}$ mg glucose per liter at time zero; \bigcirc simulator HIGH.

Figure 16. *Formal Definitions.*

Figure 17. *Useful Homomorphisms.*

Figure 18. *Summary of Program.*

Figure 19. *Growth Cycle.*

Figure 20. *Repression and Allosteric Inhibition. Repression is obtained by adjustment of $KK8K(INTGR)$. Allosteric inhibition is obtained through adjustment of $C(INTGR)$.*

Table 1. *Variables in Program.*

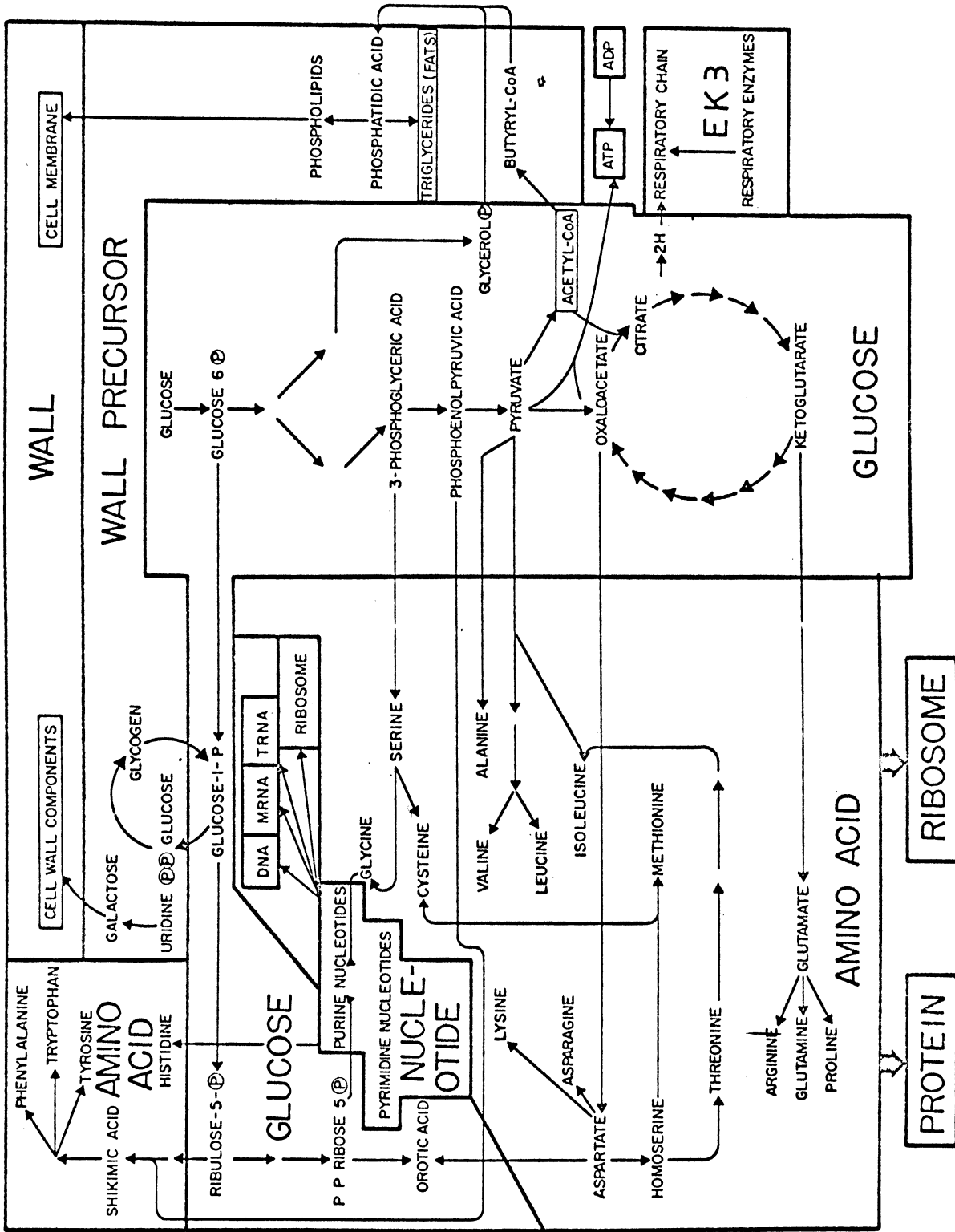


figure 1

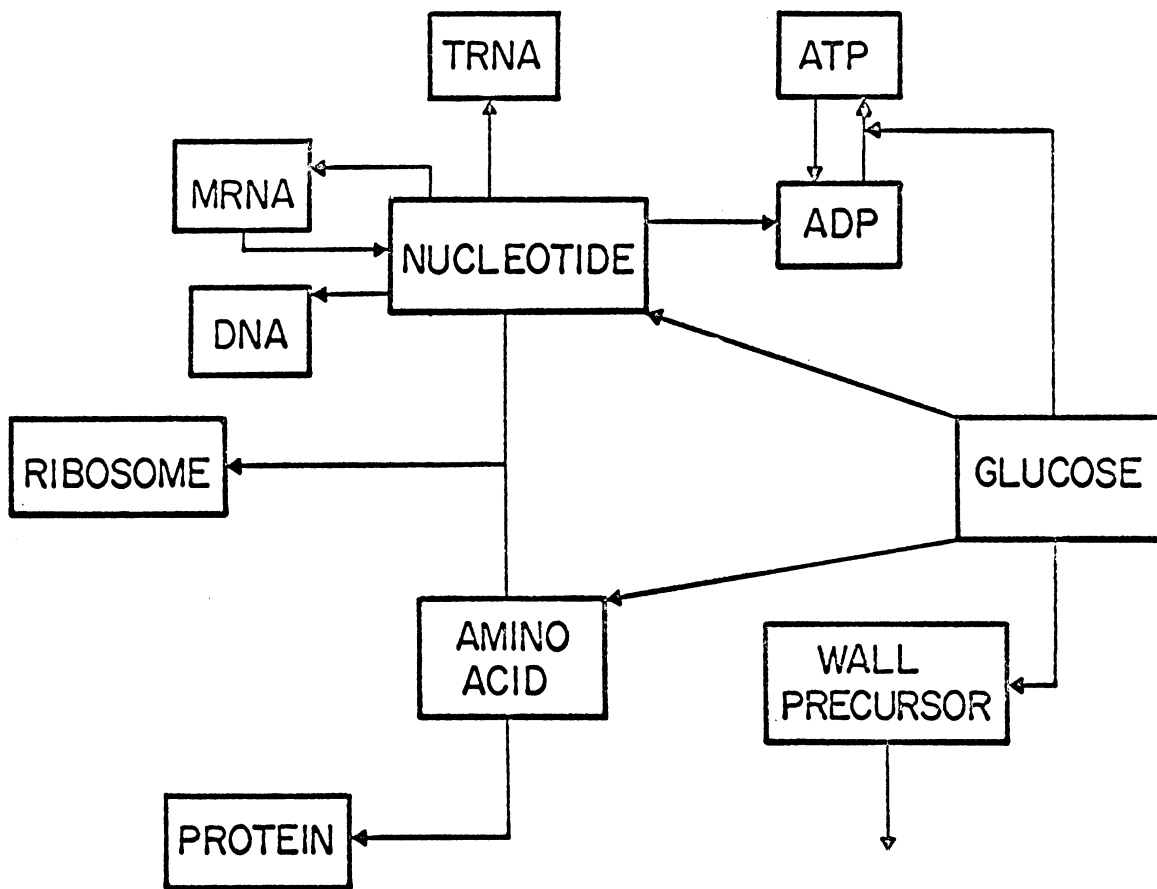


figure 2

STATE

COORDINATE = ENTITY	RANGE OF VALUE OF COORDINATES = ATTRIBUTE OF ENTITY
Pools of Chemicals, PRDC(1),..., PRDC(10)	Concentration of pool
Enzymes, EK(1),...,EK(10)	Concentration of enzyme
Messenger RNA, RNK(1),..., RNK(10)	Concentration of RNA
Genetic Apparatus	Amount of DNA, site of replication, number of genes in cell for producing sites for replication
Cell Volume	Total volume of the cell
Cell Number	Number of cells represented in the culture

TRANSITION FUNCTION

(for calculating the state of the system in the next time step from the present state)

- A. *The differential and boolean equations* relating concentrations of variables at a given time to the concentrations of those variables DT seconds later. e.g. for AA, the amino acid pool, one needs enzyme EK(2) to catalyze the production of AA from glucose, and one uses ATP as an energy source. At the same time, AA is lost as it is used for the production of RIB and PRTN.
1.
$$DAA = \underset{\text{production of AA from}}{K(2)*GLUC*DT*EK(2)*ATP} - \underset{\text{loss of AA to}}{1.E6/102.*DRIB} - \underset{\text{loss of AA to PRTN}}{(4.E4/102.)*DPRTN}$$
 2. RNK(2) produced EK(2) from AA under the direction of DNA, using ATP for energy.
 3. $DEK(2) = K(7)*AA(RNK(2)/MRNAO)*DT*EK(7)*ATP$
 4. RNK(2) itself was produced from NUC under the direction of DNA, catalyzed by EK8, using ATP for energy. RNK(2) decayed spontaneously at the same time, producing some loss of RNK(2) already present.
 5.
$$DRNK(2) = \underset{\text{production of RNK(2)}}{(K8K(2)*NUC*DNA*EK(8)*ATP} - \underset{\text{decay of RNK(2)}}{KDRNK*RNK(2)*DT}$$
- B. *Allosteric modification* of enzymes simulated by modifying the rate constant which characterizes all different forms of any enzyme associated with a particular reaction.
- C. *Repression* of messenger RNA directing the production of a particular enzyme
- D. *Genetic behavior* of DNA in response to the state of the cell
- E. *Permeability*

figure 3

```

91 DDNA = K(6)*NUC*DT*EK(6)*ATP
DDNA1 = K(6)*(T/DBLE)*.5*IN1*NUC*DT*EK(6)*ATP
DDNA2 = K(6)*(T/DBLE)*.5*IN2*NUC*DT*EK(6)*ATP
DDNA3 = K(6)*(T/DBLE)*.5*IN3*NUC*DT*EK(6)*ATP
C**** 40 MINUTES TO REPLICATE .5 * DNA
DO 100 I = 1,10
100 DRNK(I) = (K8K(I)*NUC*DNA*EK(8)*ATP - KDRNK*RNK(I))*DT
C****
DMRNA = DRNK(1) + DRNK(2) + DRNK(3) + DRNK(4) + DRNK(5) + DRNK(6)
1 + DRNK(8) + DRNK(9) + DRNK(10) + DRNK(7)
C****
DTRNA = K(10)*NUC*DT*EK(10)*ATP
DRIB = K(9)*NUC*AA*DT*EK(9)*ATP
DRNA = DMRNA + .25*DTRNA + .75*DRIB
DWALL = K(4)*GLUC*DT*EK(4)*ATP
C****
DO 101 I = 1,10
101 DEK(I) = K(7)*AA*(RNK(I)/MRNA0)*DT*EK(7)*ATP
C****
DPRTN = DEK(1) + DEK(2) + DEK(3) + DEK(4) + DEK(5) + DEK(6)
1 + DEK(8) + DEK(9) + DEK(10) + DEK(7)
C****
DNUC = -(2.5E9/660.)*DDNA - (1.E6/660.)*DMRNA
1 +K(1)*GLUC*DT*EK(1)*ATP - (2.5E4/660.)*DTRNA
2 -(2.E6/660.)*DRIB - K(5)*NUC*DT*EK(5)*ATP
C****
DAA = K(2)*GLUC*DT*EK(2)*ATP - 1.E6/102.*DRIB - (4.E4/102.)*DPRTN
DATP = K(3)*GLUC*DT*ATP*EK(3) - DNAP*DDNA - MRNAP*DMRNA
1 - TRNAP*DTRNA - RIBP*DRIB - PRNTP*DPRTN - WALLP*DWALL
2 - (AAP*K(2)*GLUC*EK(2)*ATP + NUCP*K(1)*GLUC*EK(1)*ATP + 2*K(5)*NUC
3 *EK(5)*ATP)*DT
C****
DADP = -DATP + K(5)*NUC*DT*EK(5)*ATP
DVOL = K(14)*WALL*DT
C**** INCREASE IN VOLUME PER UNIT INCREASE IN CELL WALL

```

Figure 4

Differential Equations: quantity to the left of = is the change in amount of the substance; e.g., DDNA represents the change in the amount of DNA in one time increment DT. The differential equation underlying the first equation is

$DDNA = K(6)*NUC*EK(6)*ATP*DT$ for a discrete time interval DT. As DT approaches 0, we get the underlying continuous differential equation

$$\lim_{DT \rightarrow 0} D(DNA)/DT = d(DNA)/dt = K(6)*NUC*EK(6)*ATP$$

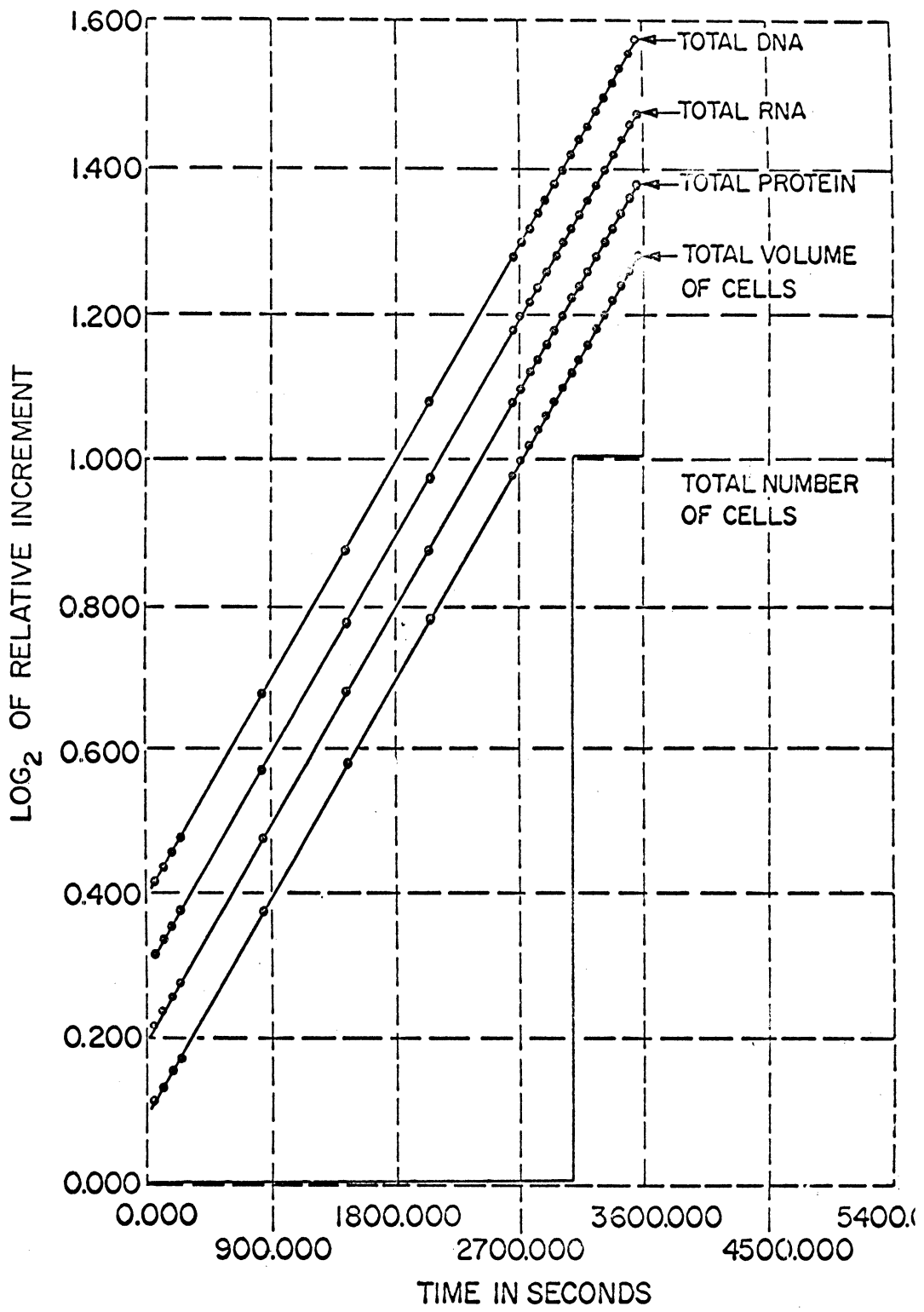


figure 5

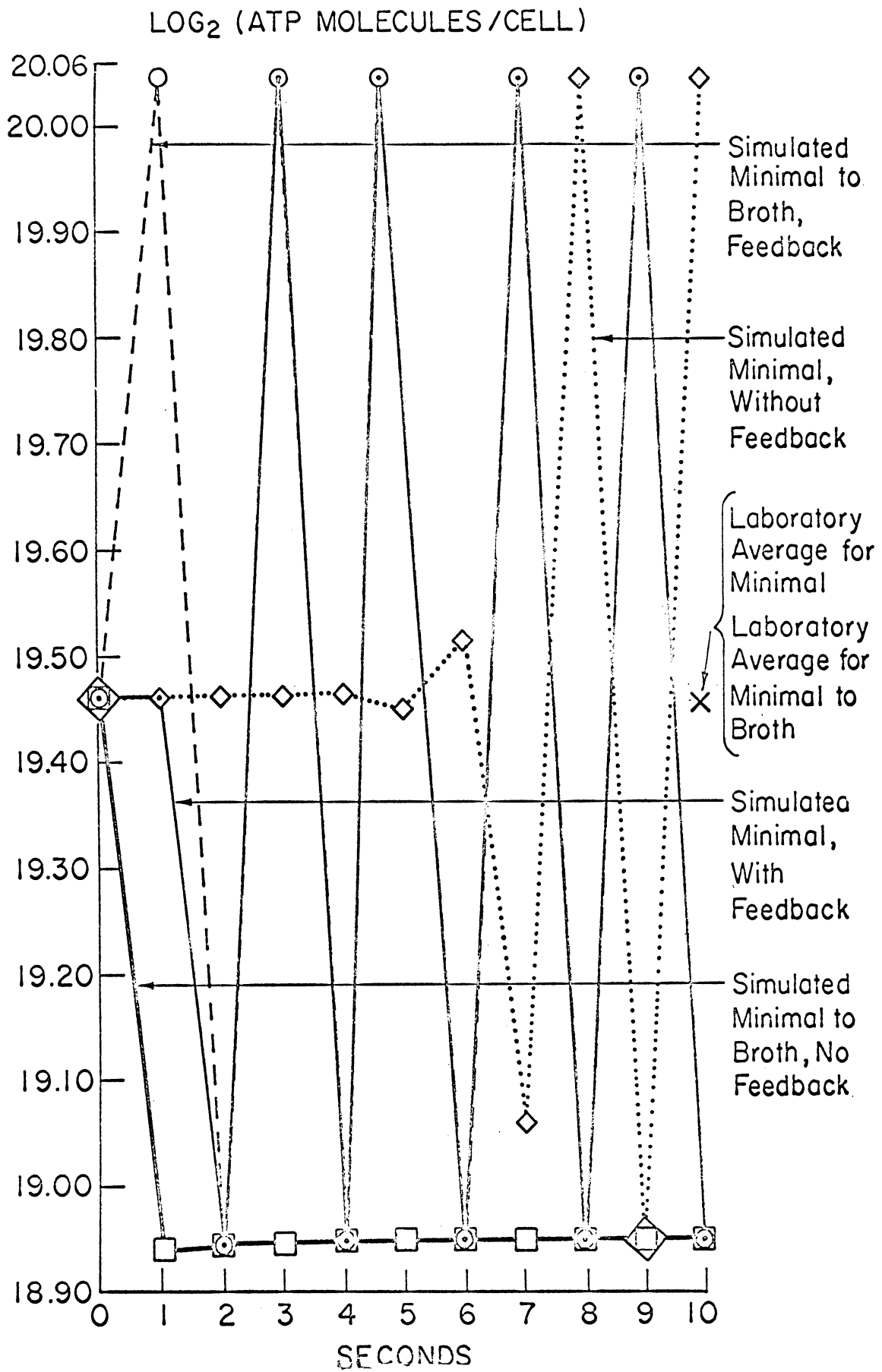


Figure 6. ATP during Simulated Shift Up.

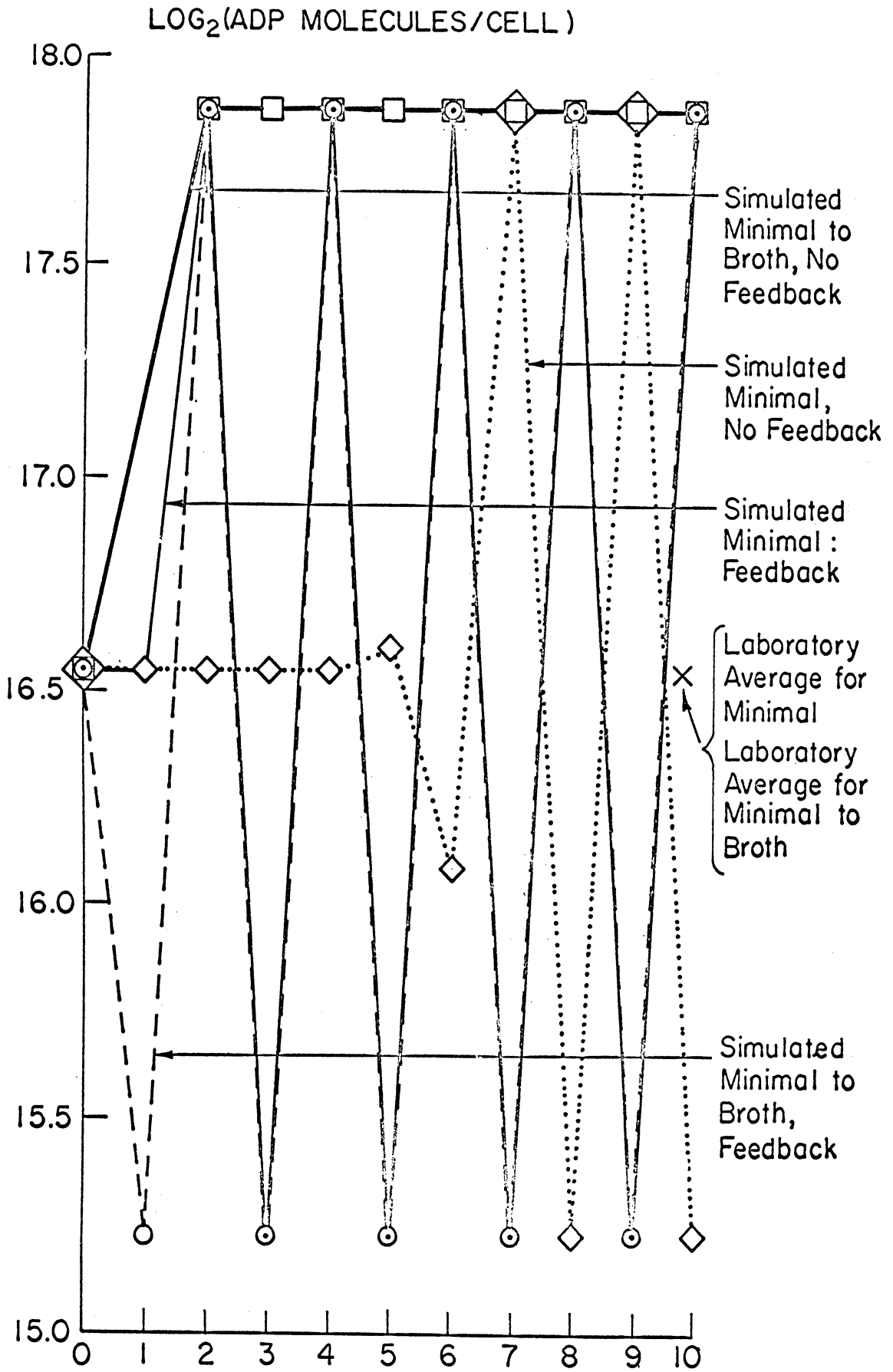


Figure 7. ADP during Simulated Shift Up.

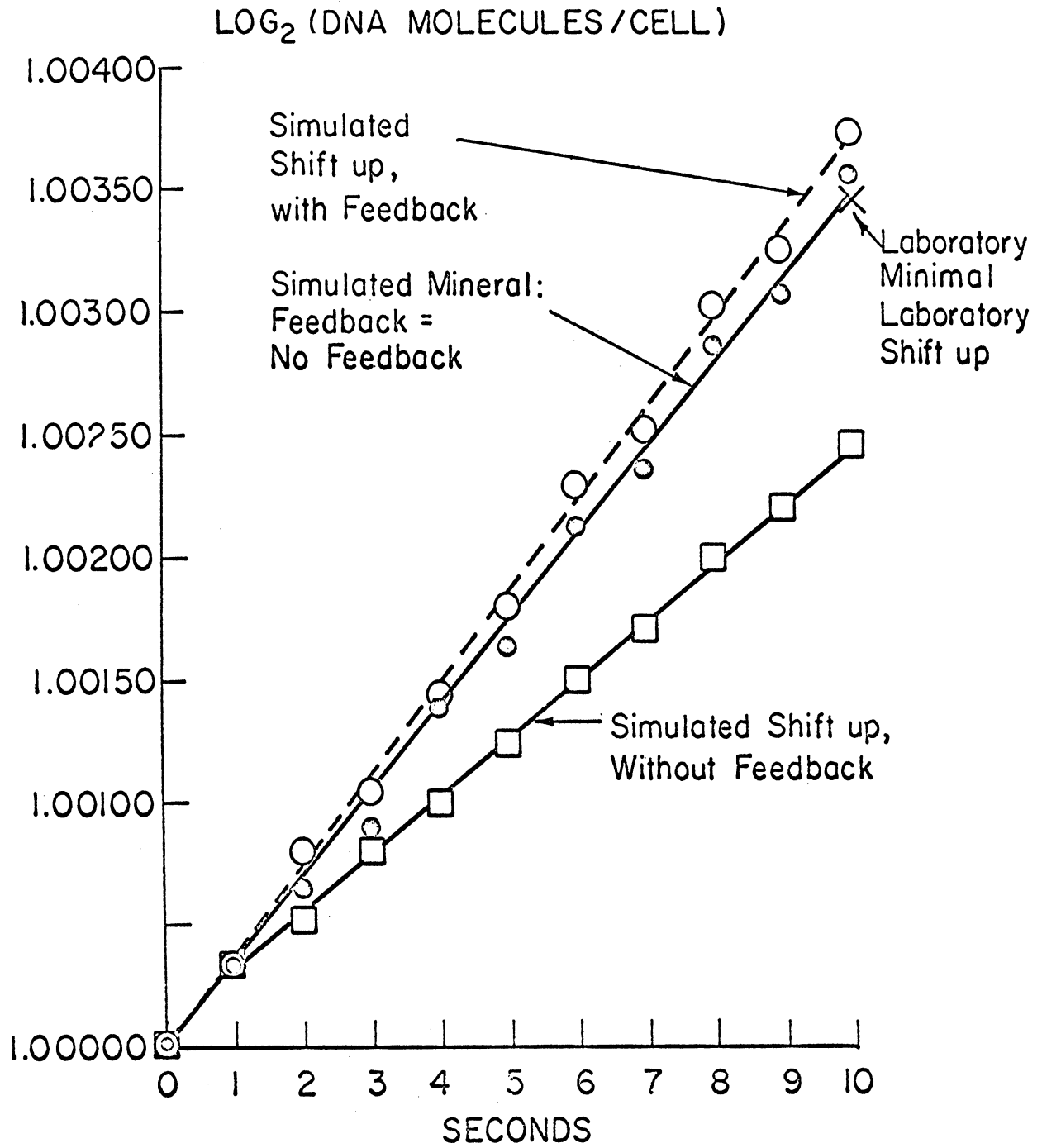


figure 8

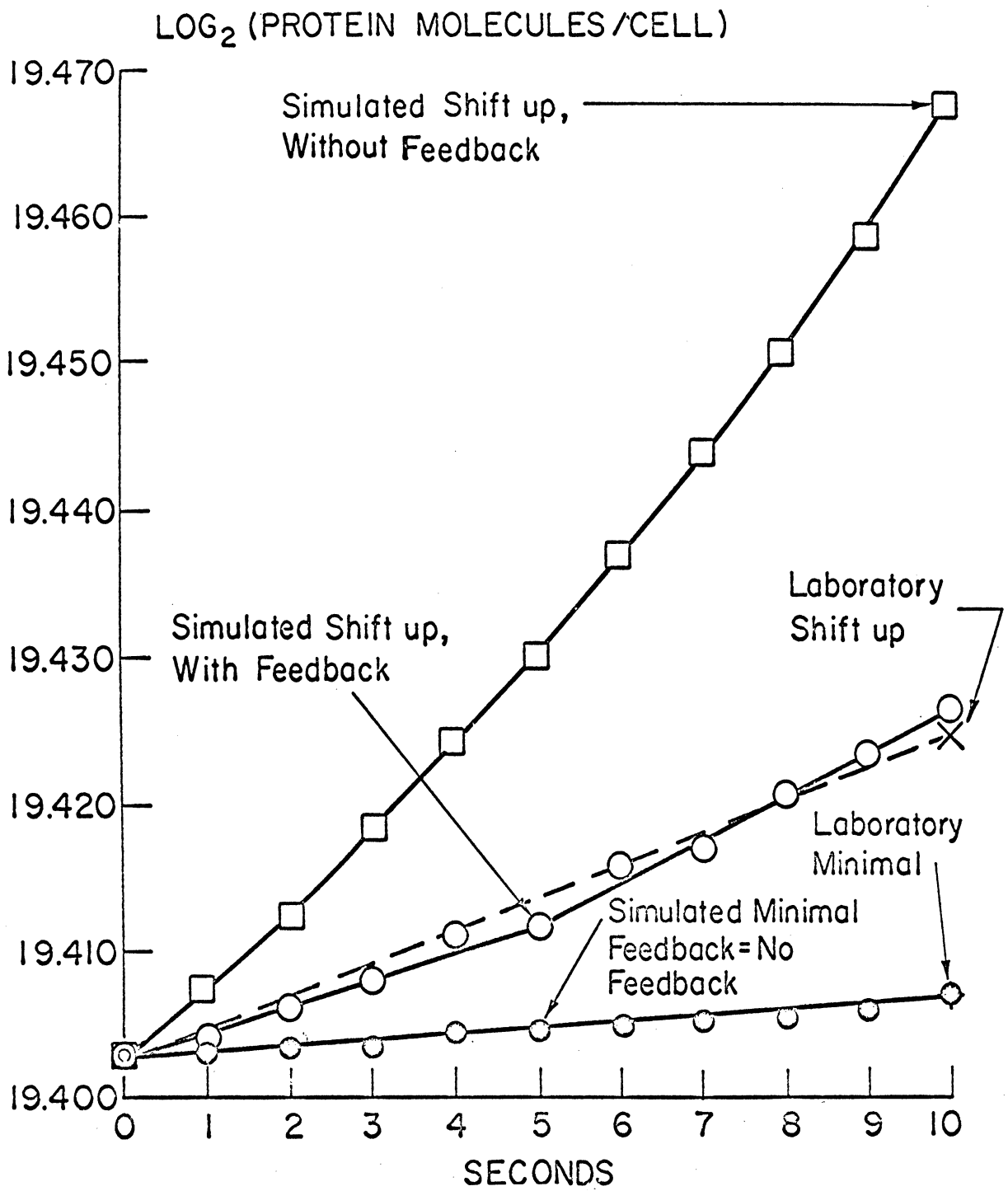


figure 9

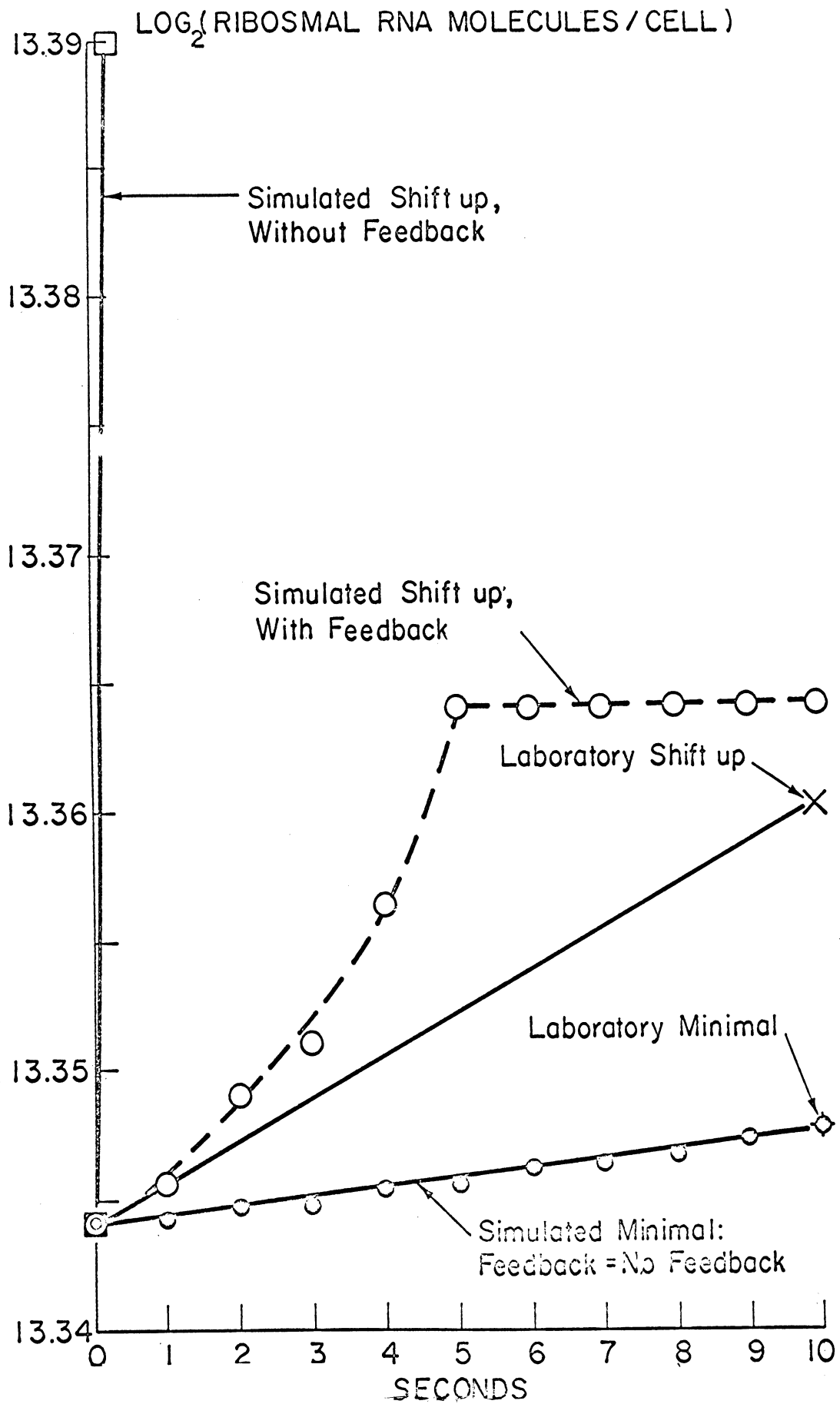


Figure 10

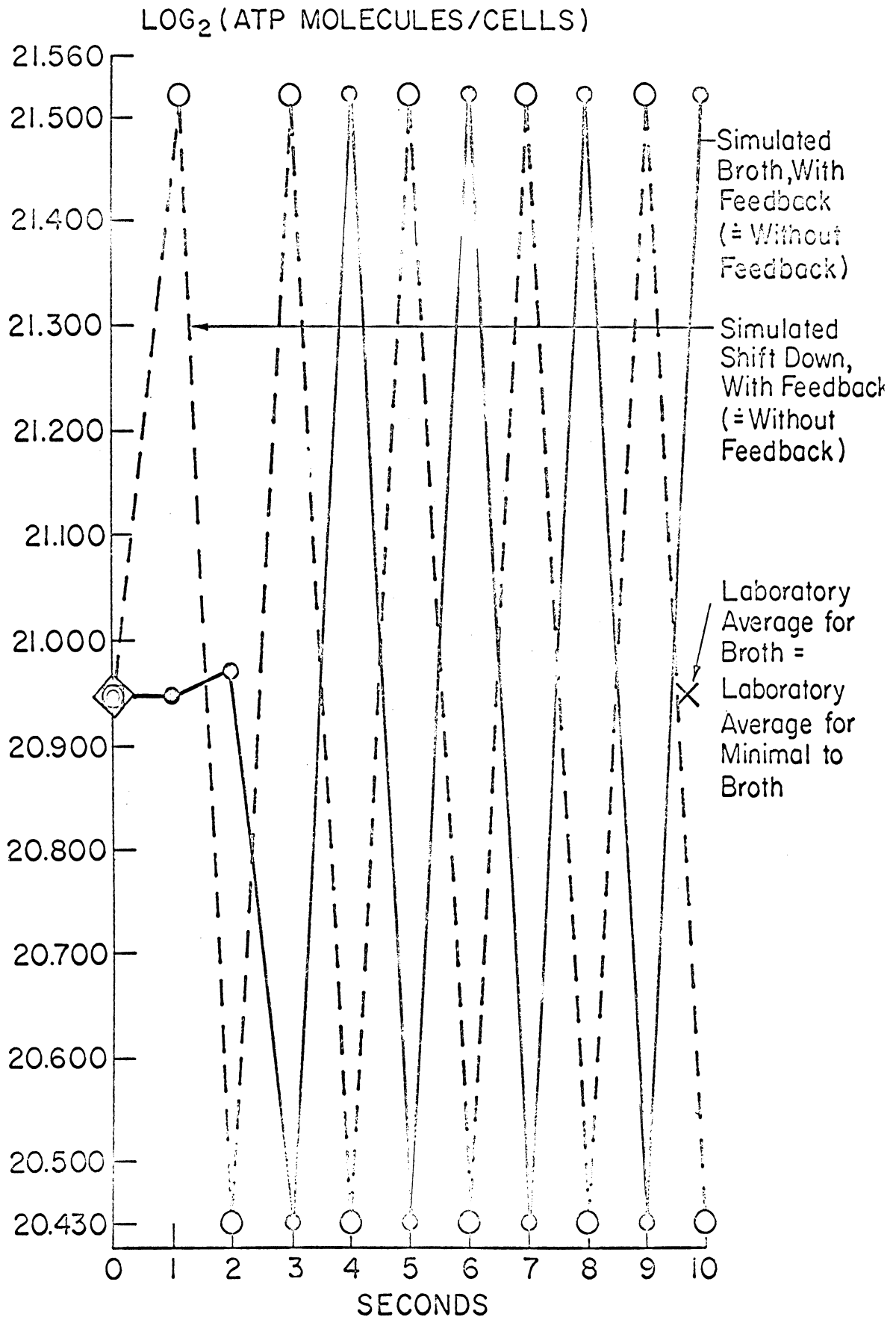


Figure 11.

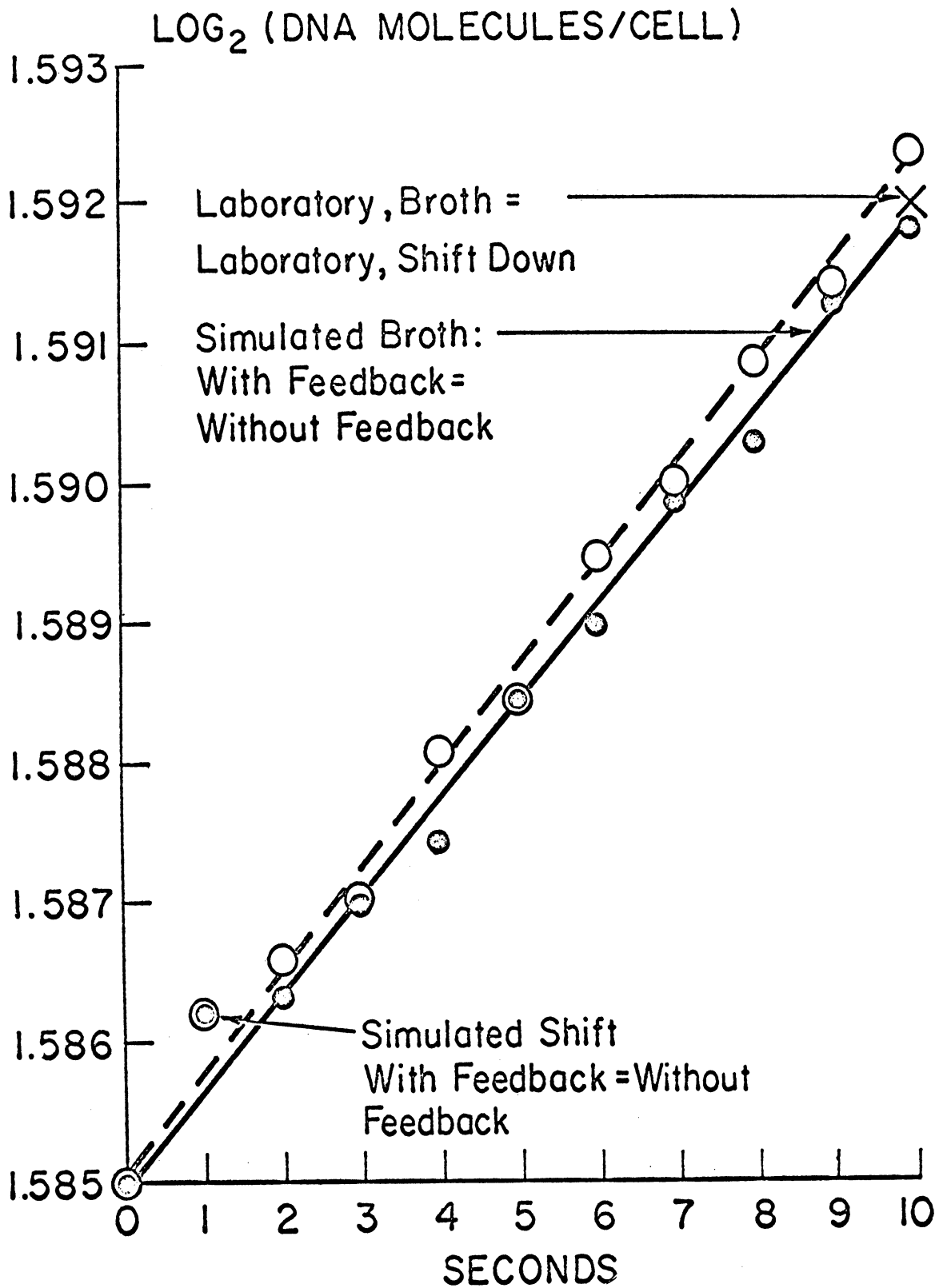


figure 12

LOG₂ (RIBOSOMAL RNA MOLECULES/CELL)

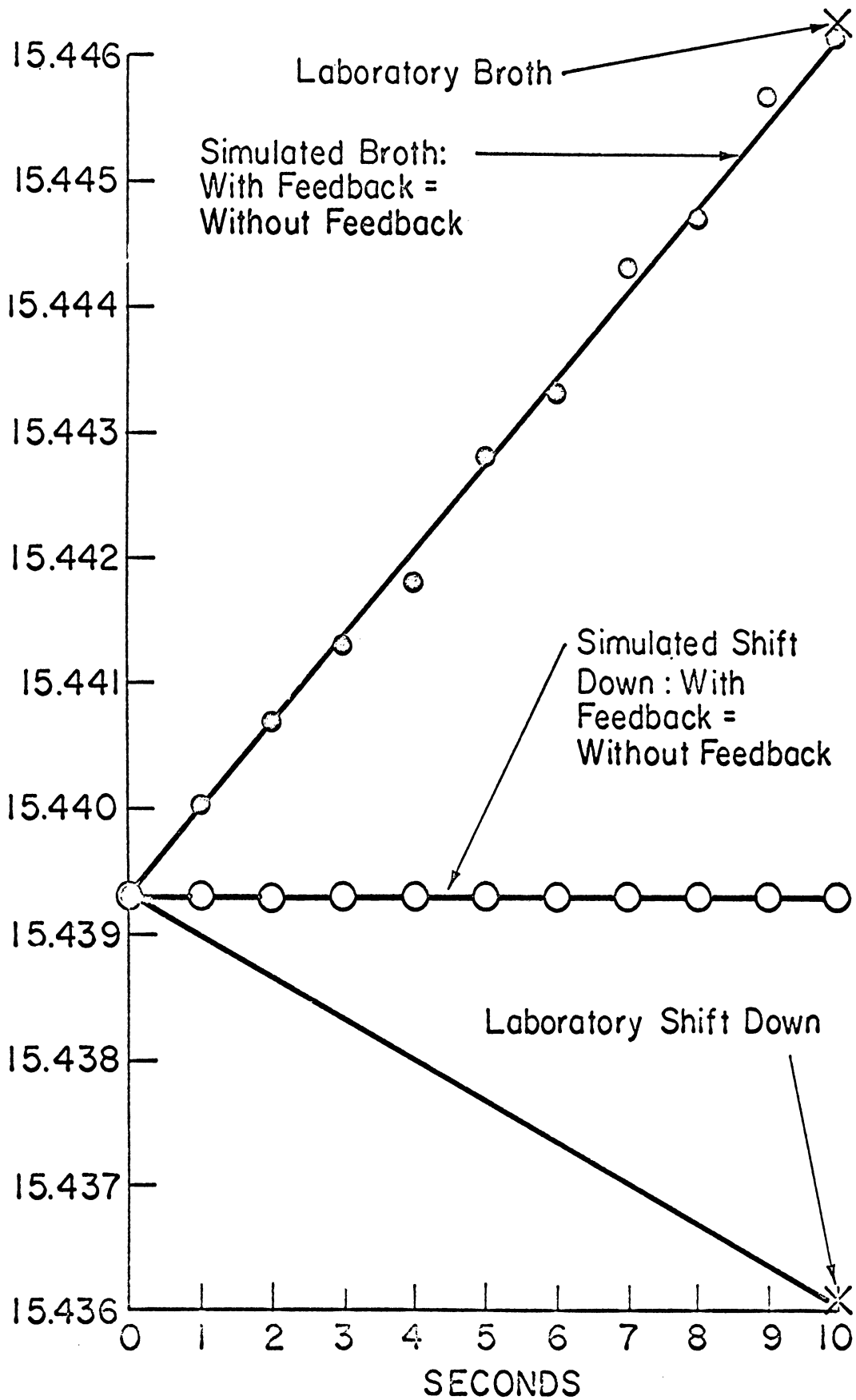


figure 13

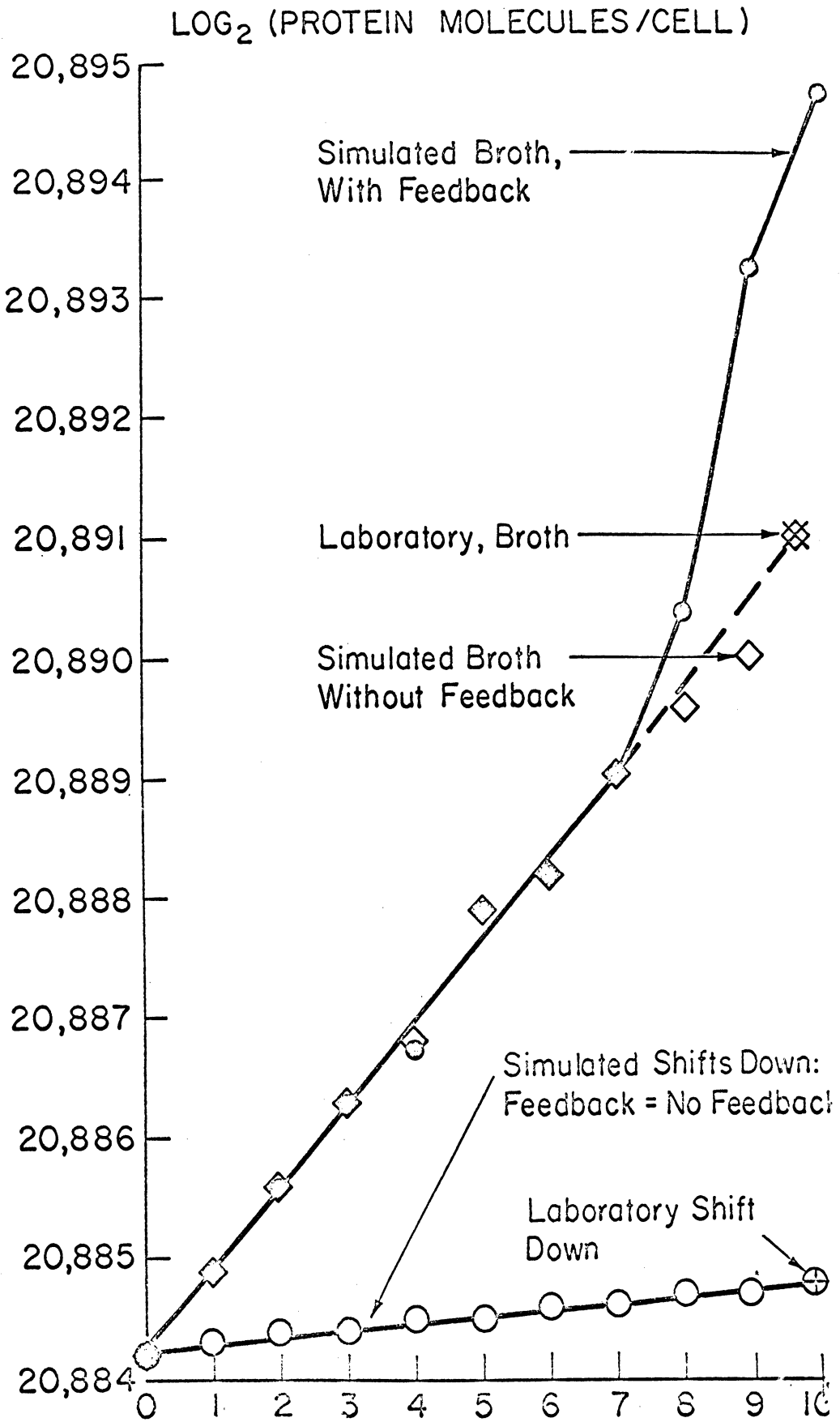


figure 14

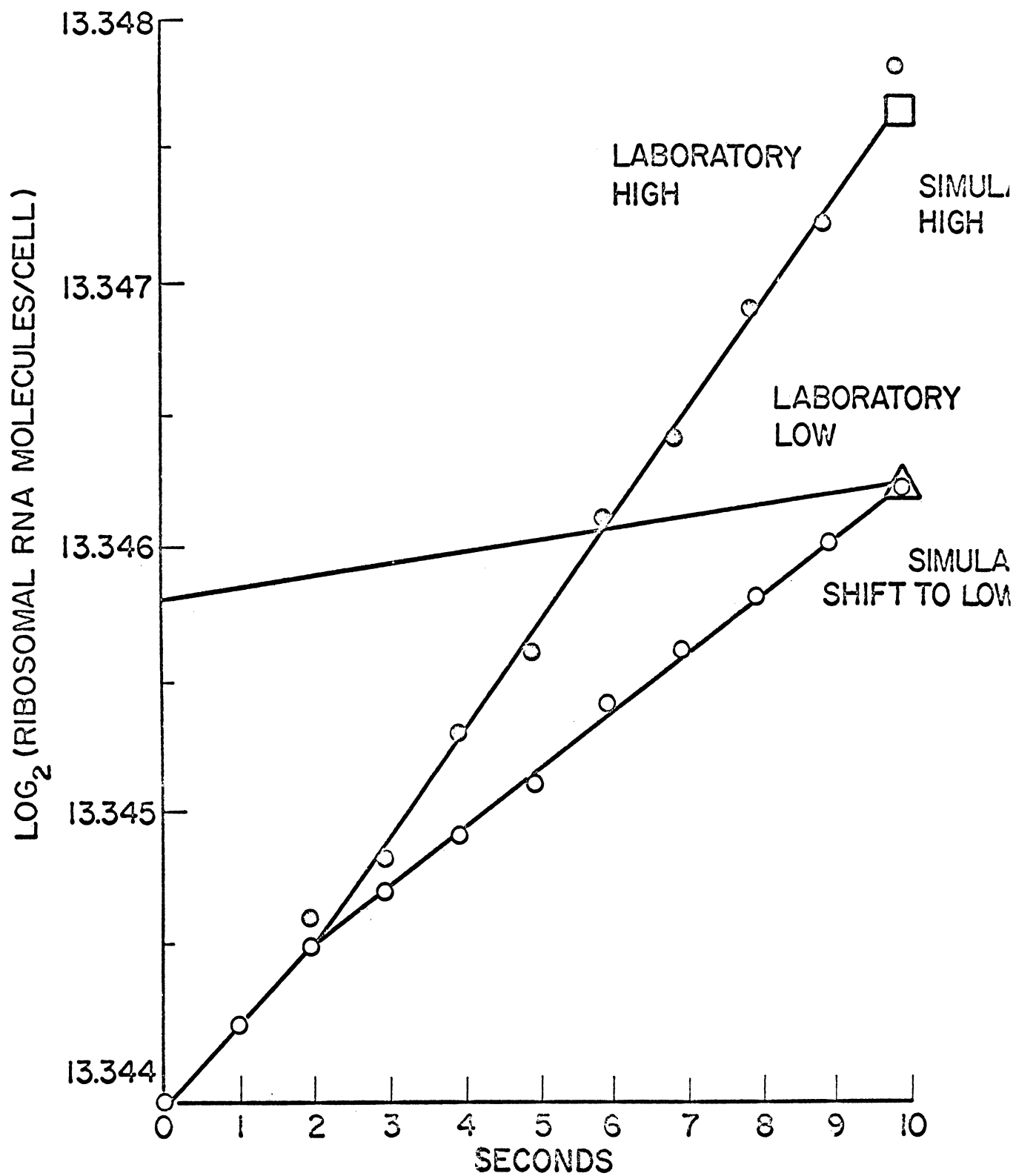


figure 15

(S, τ) is a system where

S - state space

$\tau : S \rightarrow S$ (transition function)

$$s_{t+1} = \tau(s_t)$$

S , given as a cartesian product

$$S = \prod_{\alpha \in \mathcal{D}} S_{\alpha}$$

where \mathcal{D} - set of co-ordinates (entities)

S_{α} - state set (attribute set) of co-ordinate α

Homomorphism h from (S, τ) to (S', τ')
is a map $h: S \rightarrow S'$ such that for all $s \in S$

$$h(\tau(s)) = \tau'(h(s))$$

Homomorphisms may be obtained by partitioning
co-ordinate set \mathcal{D} such that the blocks, or
higher level co-ordinates preserve the transition
function.

figure 16

STATE SPACE	CO-ORDINATES	CO-ORDINATE SETS	TRANSITION FUNCTION
Atomic	Electrons, nuclei	Position, momentum	Schrodinger's equation (quantum mechanics)
Molecular	Molecules	Shape, active site, energy level	e.g. Koch probabilistic model
Concentration	Molecule type e.g. ADP, DNA	Number of molecules of ADP/cell etc.	Mathematical and logical equations based on chemical kinetics e.g. Chance et al
Higher level groupings	Bio-chemical pools e.g. amino acids	Number of molecules of pool/cell	Mathematics and logical equations based on chemical kinetics and Molecular control mechanisms e.g. repression, allosteric inhibition (Weinberg, Goodwin)

Figure 17.

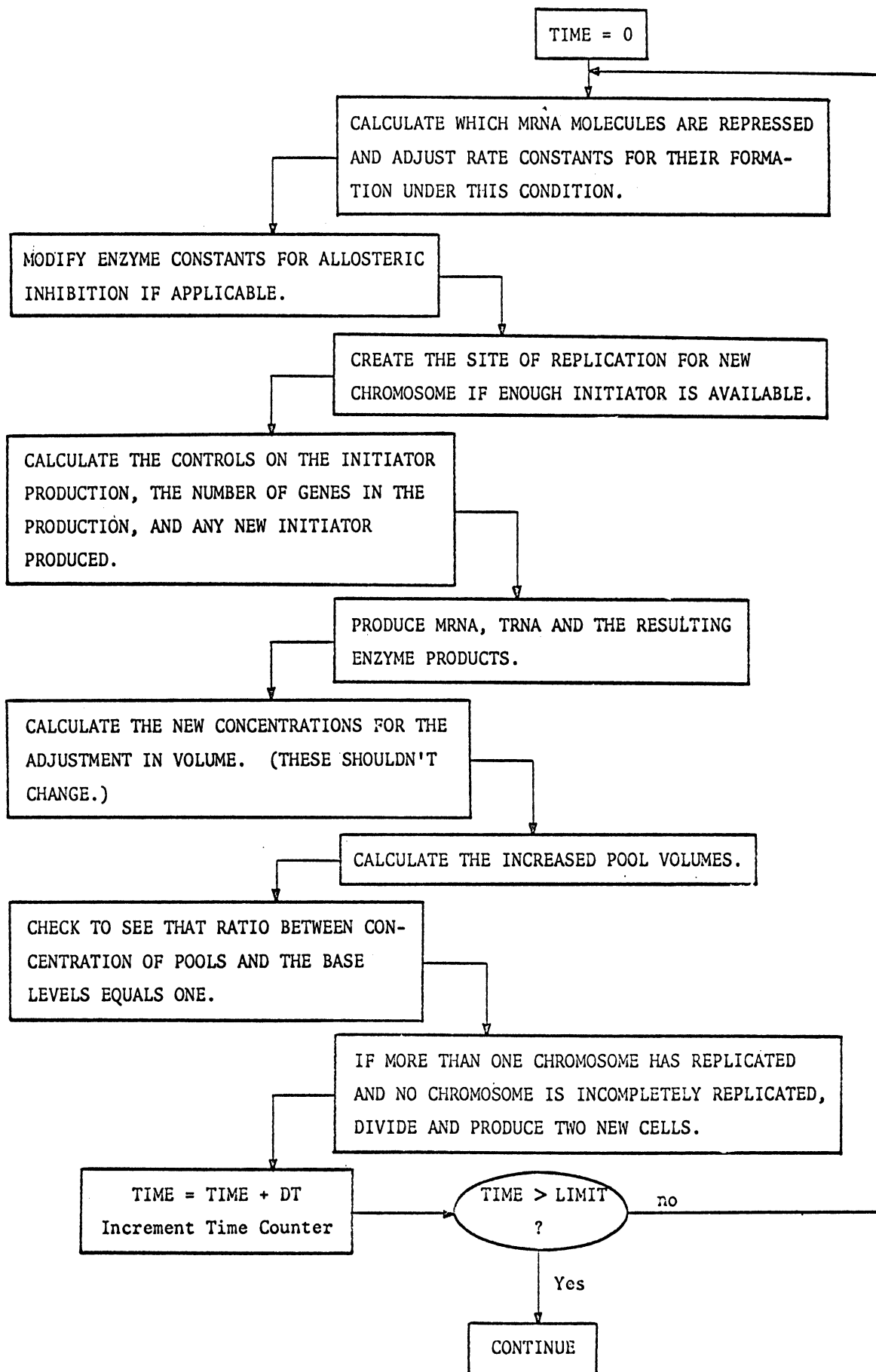


Figure 18. Summary of Program.

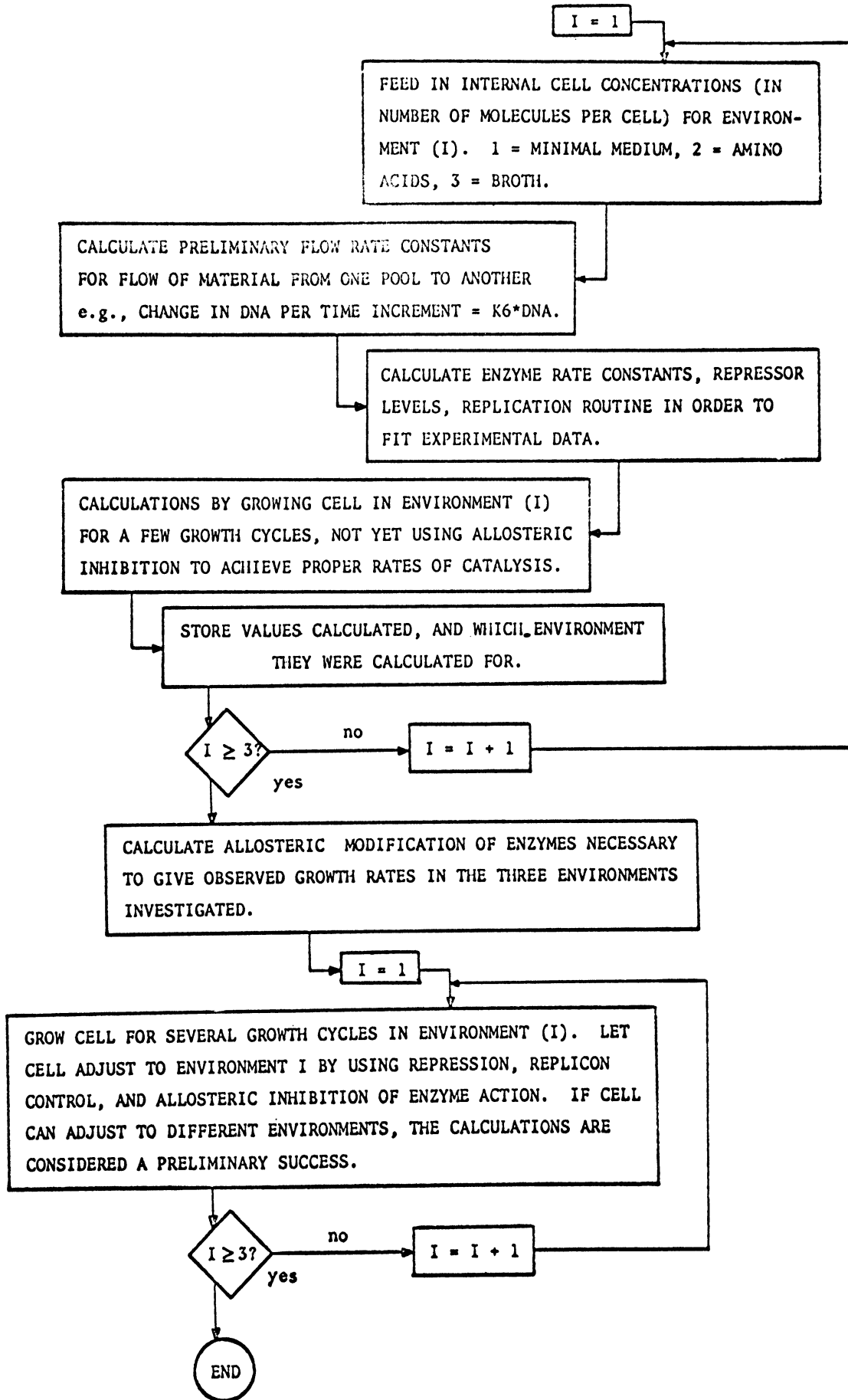


figure 19

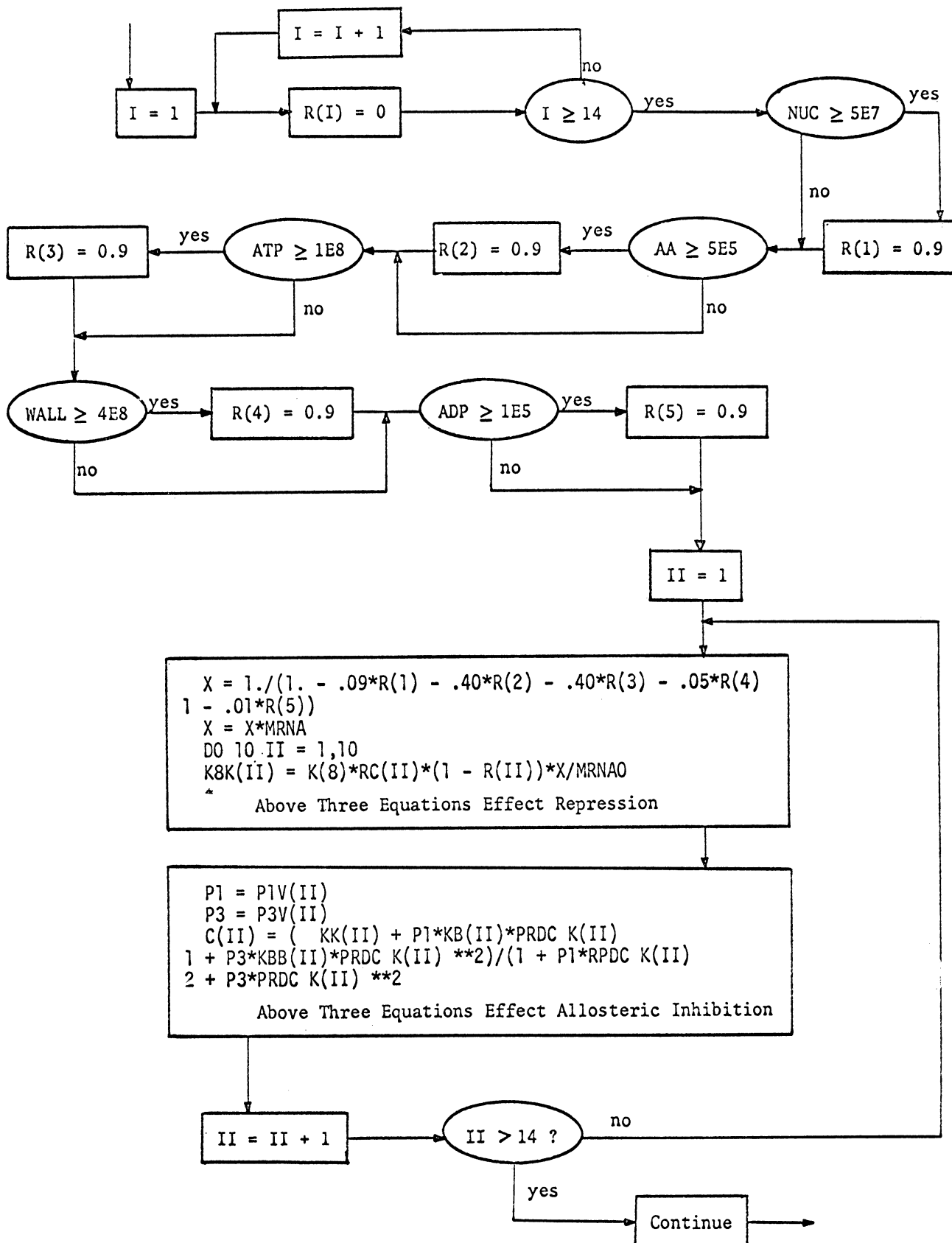


Figure 20 Repression and Allosteric Inhibition.
 Repression is obtained by adjustment

TABLE 1

VARIABLES IN PROGRAM: A = array
 O = floating point
 I = integer

A2	A	0 arrays used in solve function to obtain rate constants used
A3	A	0 in allosteric inhibition
AAO		0 amino acid concentration at time zero
AAP		0 ATP molecules used to make 1 amino acid molecule
AAO		0 amino acid concentration
ADJST		0 adjustment factor for concentrations from volume increase
ADPO		0 ADP concentration at time 0
ADP		0 ADP concentration
ATPO		0 ATP concentration at time zero
ATP		0 ATP concentration
ATPSB	A	0 array to store ATP concentrations in different environments
BROTH		1 equals 1, if cell growing in broth
CAA		1 equals 1 if cell growing in casamino acid
CHROMO		0 number of chromosomes at time 0
CNTRL		1 equals 1 if cell using metabolic controls to adjust growth rate
COUNT		0 number of growth cycles made
CRAZY		1 used as a logical variable
C(I)	A	0 enzyme rate constants
DAA		0 change in amino acid concentration
DADP		0 change in ADP concentration
DAPO2		0 change in ATP concentration from literature
DATP		0 change in ATP concentration calculated from rate constants in one time step
DDNA1		0 change in amount chromosome 1 in one time increment
DDNA2		0 change in amount of chromosome 2 in one time increment
DDNA3		0 change in amount of chromosome 3 in one time increment
DDNA		0 change in total DNA in one time increment
DEK(1)		0 change in enzymes for nucleotide production in one time increment
DEK(2)		0 change in enzymes for amino acid production in one time increment
DEK(3)		0 change in enzymes for glycolysis production in one time increment
DEK(4)		0 change in enzymes for wall production in one time increment
DEK(5)		0 change in enzymes for ADP, ATP synthesis in one time increment
DEK(6)		0 change in enzymes for DNA synthesis in one time increment
DEK(7)		0 change in enzymes for protein production in one time increment
DEK(8)		0 change in enzymes for mRNA synthesis in one time increment
DEK(9)		0 change in enzymes for ribosome synthesis in one time increment

DEK(10) 0 change in enzymes for TRNA production in one time increment
 DIN 0 change in initiator concentration in one time increment
 DNAO 0 DNA at time zero
 DNA1 0 chromosome 1 "concentration", i.e., amount/volume of cell
 DNA1Z 0 chromosome 1 at zero time
 DNA1T 0 total chromosome 1

 DNA2 0 chromosome 2 "concentration"
 DNA2T 0 total chromosome 2
 DNA2Z 0 chromosome 2 at zero time
 DNA3 0 chromosome 3 "concentration"
 DNA3T 0 total chromosome 3

 DNA3Z 0 chromosome 3 at zero time
 DNAP 0 ATP used per DNA molecule synthesized
 DNA 0 DNA
 DNASB A 0 array to save concentrations of DNA in different environments
 DNUC 0 change in nucleotide concentration

 DBLE 0 time for cell to go through one reproductive cycle
 DPRTN 0 change in protein in one time increment
 DRIB 0 change in ribosome in one time increment
 DMRNA 0 change in MRNA in one time increment
 DRNA 0 change in total RNA in one time increment

 DRNK(i) A 0 change in MRNA for enzyme EK(i) in one time increment.
 i ranges from 1 to 10.
 DT 0 length of one time increment, = differential
 DUM1 0 dummy variable in solve function
 DUM2 0 dummy variable in solve function
 DUM3 0 dummy variable in solve function

 DVOL 0 change in cell volume in one time increment
 DWALL 0 change in cell membrane and cell wall in one time increment
 DPRDK A 0 array of change in product concentration in one time increment
 PRD A 0 the stored array of the previous four product values, for
 predictor corrector
 DPRD A 0 array of the four previous D(product) values for the predictor
 corrector

 PPRD A. 0 current array of the predictor values of products
 CPRD A 0 current array of corrector values of products
 EK(1) 0 concentration of enzymes for nucleotide production
 EKZ(1) 0 concentration of enzymes for nucleotide production at zero time
 EK(2) 0 concentration of enzymes for amino acid production
 EKZ(2) 0 concentration of enzymes for amino acid production at zero time

where 3 indicates glycolysis
 4 indicates cell wall production
 5 indicates ADP, ATP production
 6 indicates DNA production
 7 indicates protein production
 8 indicates MRNA production
 9 indicates ribosome production
 10 indicates TRNA production

FACTR 0 factor by which chromosomes multiply in one reproductive cycle
 GLUCO 0 glucose concentration at zero time
 GLUC 0 glucose concentration
 ID 1 integer variable in RPLACE routine
 IN11 0 site for replication of chromosome 11, = 1 if it is present

 IN11Z 0 site for replication of chromosome 11 at zero time
 IN1 0 site for replication of chromosome 1, = 1 if it is present
 IN1Z 0 site for replication of chromosome 1 at zero time
 IN21 0 site for replication of chromosome 21
 IN21Z 0 site for replication of chromosome 21 at zero time

 IN2 0 site for replication of chromosome 2
 IN2Z 0 site for replication of chromosome 2 at zero time
 IN31 0 site for replication of chromosome 31
 IN31Z 0 site for replication of chromosome 31 at zero time
 IN3 0 site for replication of chromosome 3

 IN3Z 0 site for replication of chromosome 3 at zero time
 IN 0 concentration of initiator in cytoplasm
 II 1 an integer variable
 INZ 0 initiator concentration at zero time
 K(1) 0 preliminary rate constant for nucleotide production

 K(2) 0 preliminary rate constant for amino acid production
 K(3) 0 preliminary rate constant for glycolysis
 K(4) 0 preliminary rate constant for cell wall production
 K(5) 0 preliminary rate constant for ADP production
 K(6) 0 preliminary rate constant for DNA production

 K(7) 0 preliminary rate constant for protein production
 K(8) 0 preliminary rate constant for MRNA production
 K(9) 0 preliminary rate constant for ribosome production
 K(10) 0 preliminary rate constant for TRNA production
 K(14) 0 preliminary rate constant for volume increase as a function
 of wall

 KDRNK 0 rate constant for MRNA decay
 K8K(i) A 0 rate constant for MRNA EK(i)
 K8KZ(i) A 0 rate constant for MRNA for EKZ(i)
 KBB(i) A 0 rate constant for allosterically inhibited enzyme EK(i)
 with two molecules of product attached to the enzyme
 KB A 0 array of rate constants of allosterically inhibited enzymes
 with one molecule of product attached to the enzyme

KIN 0 preliminary rate constant for initiator production
 K8K(i) A 0 rate constant for production of EK(i)
 KK(i) A 0 rate constant for uninhibited enzyme EK(i)
 K(i) A 0 array to store preliminary rate constants, used for each
 environment
 LN2 0 natural logarithm of 2

 L 1 integer variable for calling on solve function
 M 1 integer variable for printing loop
 MRNAO 0 MRNA concentration at time zero
 MRNAP 0 ATP per MRNA molecule produced
 MRNA 0 MRNA concentration

 MULT 0 number of genes producing initiator
 NO 0 number of cell in population (doubles when cell divides)
 NUCO 0 molecules of nucleotide at zero time
 NUCP 0 molecules of ATP to make one nucleotide
 NUC 0 concentration of nucleotide

 P1 0 rate constant
 P1V A 0 array of equilibrium rate constants for enzymes
 P3 0 equilibrium rate constant for two molecule allosteric inhibition
 P3V A 0 array of equilibrium rate constants for two molecule
 allosteric inhibition
 PRDCO A 0 array equivalenced to products at zero time

 PRDCK A 0 array equivalenced to products
 PRDC A 0 array for storing concentrations of products in different
 environments
 PRDC(1) 0 NUC
 PRDC(2) 0 AA
 PRDC(3) 0 ATP

 PRDC(4) 0 WALL
 PRDC(5) 0 ADP
 PRDC(6) 0 DNA
 PRDC(7) 0 PRTN
 PRDC(8) 0 MRNA

 PRDC(9) 0 RIB
 PRDC(10) 0 TRNA
 PRDC(11) 0 GLUC
 PRDC(14) 0 VOL
 PRTNO 0 protein concentration at zero time

 PRTNP 0 ATP molecules used per protein molecule formed
 PRTN 0 protein concentration
 RAA 0 ratio of amino acid concentration to a base level
 RADP 0 ratio of ADP concentration to a base level
 RATP 0 ratio of ATP concentration to a base level

RC	A	0 array of repression constants for mRNA repression
RDNA1		0 ratio of chromosome 1 concentration to a base level
RDNA2		0 ratio of chromosome 2 concentration to a base level
RDNA		0 ratio of DNA concentration to a base level
REK(i)	A	0 ratio of EK(i) concentration to a base level, $i = 1, \dots, 10$
RIBO		0 ribosome concentration at time zero
RIBP		0 ATP used per ribosome made
RIB		0 ribosome concentration
RNAO		0 RNA concentration at time zero
RNA		0 RNA concentration
TRNAO		0 transfer RNA concentration at time zero
TRNAP		0 ATP per transfer RNA molecule made
TRNA		0 transfer RNA concentration
RNK(i)	A	0 concentration of mRNA for enzyme EK(i), $i = 1, \dots, 10$
RNKZ(i)	A	0 concentration at zero time of mRNA for EKZ(i), $i = 1, \dots, 10$
RNUC		0 ratio of nucleotide concentration to a base level
RON		1 used as a logical variable turning repression on
RPRTN		0 ratio of protein concentration to a base level
RRIB		0 ratio of ribosome concentration to a base level
RMRNA		0 ratio of mRNA concentration to a base level
RRNA		0 ratio of RNA concentration to a base level
RTRNA		0 ratio of TRNA concentration to a base level
RRNK(i)	A	0 ratio of RNK(i) concentration to a base level, $i = 1, \dots, 10$
R	A	0 array for repression constants
RVOL		0 ratio of new volume to old volume at end of one time increment
RWALL		0 ratio of pool for wall to a base level in terms of concentration
SUM	A	0 array used in solve function
T		0 generation time in seconds
VOLO		0 volume of cell at time zero
VOLN		0 volume at end of one time increment
VOL		0 volume
WALLO		0 concentration of pool for wall production at time zero
WALLP		0 ATP molecules used per molecule of cell wall produced
WALL		0 concentration of pool for wall production
X		0 variable used in repression routine
XK(i,j)	A	0 value of K(i) in environment (j)
XEK(i,j)	A	0 value of EK(k) in environment (j)
XK8(i,j)	A	0 value of K8K(i) in environment (j)

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