



REVIEW ARTICLE

10.1002/2014GC005305

Special Section:

Development of Isotopic Proxies for Paleoenvironmental Interpretation: A Carbon Perspective (DIPPI-C)

Key Points:

- Photic zone recycling of organic carbon is responsible for their low $\delta^{13}\text{C}$ values
- Black shales deposited during periods of strong surface ocean stratification
- Periods of greenhouse climate established conditions for black shale deposition

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Citation:

Meyers, P. A. (2014), Why are the $\delta^{13}\text{C}_{\text{org}}$ values in Phanerozoic black shales more negative than in modern marine organic matter?, *Geochem. Geophys. Geosyst.*, 15, 3085–3106, doi:10.1002/2014GC005305.

Received 20 FEB 2014

Accepted 12 JUN 2014

Accepted article online 16 JUN 2014

Published online 12 JUL 2014

Why are the $\delta^{13}\text{C}_{\text{org}}$ values in Phanerozoic black shales more negative than in modern marine organic matter?

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Abstract The $\delta^{13}\text{C}_{\text{org}}$ values of Phanerozoic black shales average -27‰ , whereas those of modern marine organic matter average -20‰ . The black shale isotopic values mimic those of continental organic matter, yet their organic geochemical properties mandate that they contain predominantly marine organic matter. Hypotheses that proposed to explain the low $\delta^{13}\text{C}$ values of black shales include diagenetic losses of isotopically heavier organic matter components, releases of isotopically light carbon from methane clathrates or extensive magmatic events, greater photosynthetic discrimination against ^{13}C during times of higher atmospheric $p\text{CO}_2$, and greenhouse climate stratification of the surface ocean that magnified photic zone recycling of isotopically light organic matter. Although the last possibility seems contrary to the vertical mixing that leads to the high productivity of modern oceanic upwelling systems, it is consistent with the strongly stratified conditions that accompanied deposition of the organic carbon-rich Pliocene-Pleistocene sapropels of the Mediterranean Sea. Because most Phanerozoic black shales contain evidence of photic zone anoxia similar to the sapropels, well-developed surface stratification of the oceans was likely involved in their formation. Existence of isotopically light land plant organic matter during several episodes of extensive magmatism that accompanied black shale deposition implies massive release of mantle CO_2 that added to the greenhouse conditions that favored oceanic stratification. The ^{13}C depletion common to most Phanerozoic black shales apparently resulted from a greenhouse climate associated with elevated atmospheric $p\text{CO}_2$ that led to a strongly stratified ocean and photic zone recycling of organic matter in, augmented by magmatic CO_2 releases.

1. Introduction

The fate of over 99% of the organic matter produced in the oceans is oxidation and remineralization and not accumulation in the seafloor [Müller and Suess, 1979]. However, large amounts of organic matter can sometimes accumulate in ocean sediments. In the upwelling systems of the modern oceans, which enjoy the special combination of high surface productivity, magnified organic matter export, an intensified oxygen minimum zone, elevated sinking rates, and rapid sedimentation, the survival of organic matter is improved, and as much as 10% of the initially produced material can survive sinking and become preserved in the seafloor [e.g., Emerson and Hedges, 1988]. Organic carbon concentrations commonly range between several percent to several tens of a percent in the sediments under these high-productivity systems, yet these special areas constitute only about 1% of the ocean surface. As a consequence, nearly all modern marine sediments contain less than a few tens of a percent of organic carbon. Because most ancient marine sediments similarly contain very little organic carbon, conditions of generally low productivity and poor preservation of organic matter similar to those of today evidently dominated much of the history of the oceans.

Against this backdrop of scant accumulation of organic matter in marine sediments, the episodic occurrence of organic-carbon-rich “black shales” throughout the marine geologic record is especially interesting (Figure 1). These curious deposits contain between a few percent to several tens of a percent organic carbon, and they are best known from the Mesozoic [Dean et al., 1986; Emeis and Weissert, 2009; Negri et al., 2006; Jenkyns, 2010] although well-studied Paleozoic examples exist (Table 1). More ancient examples are also known (e.g., Fig Tree, Onverwacht, and Nonesuch shales of the Proterozoic), but thermal alteration and the passage of time have degraded their organic matter and imposed limits on its characterization. Virtually, no black shales were deposited during the Cenozoic (Figure 1), with the important exception of the Pliocene-Pleistocene sapropels of the Mediterranean Sea, which are considered to be near-modern analogs of the

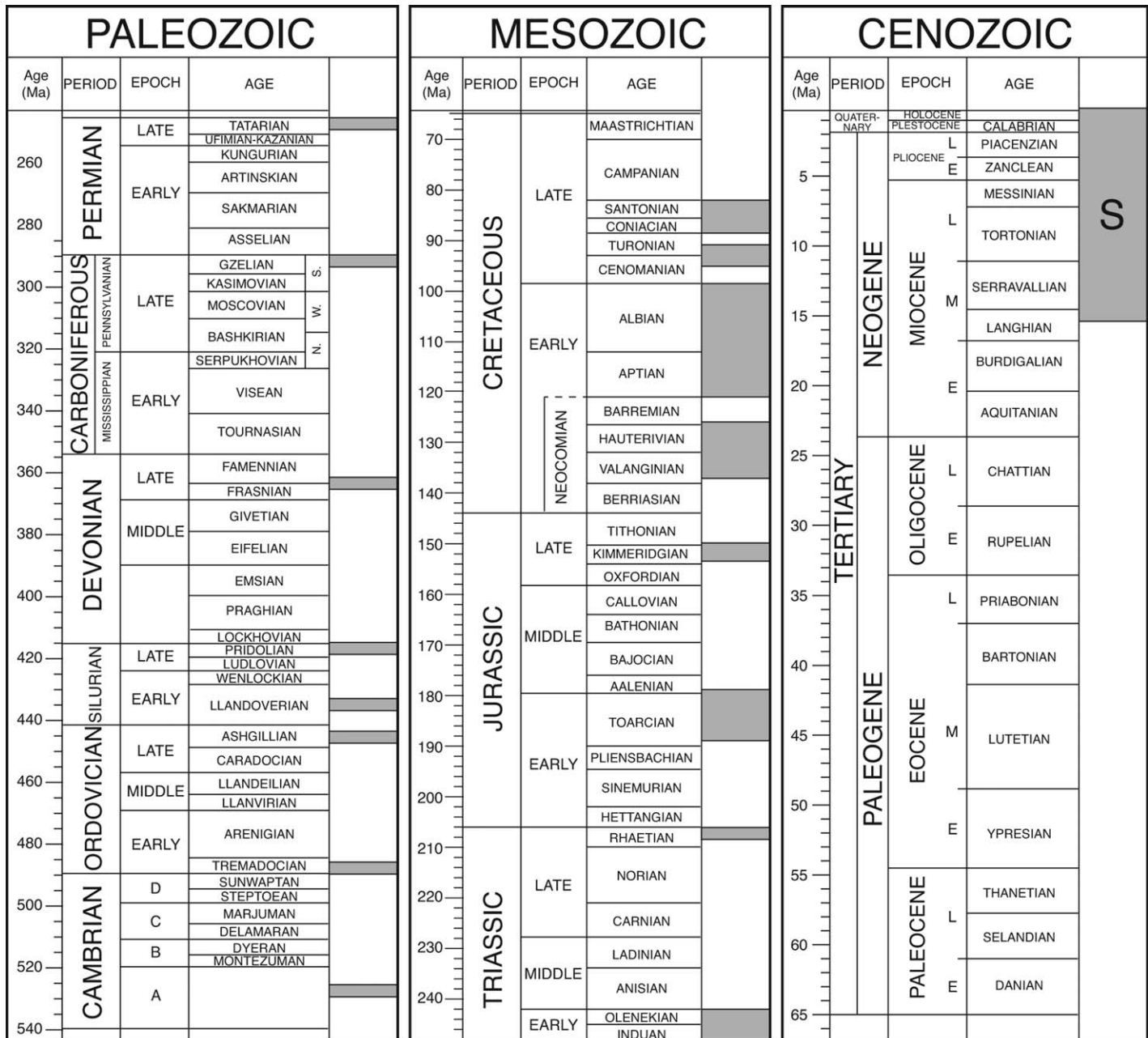


Figure 1. Phanerozoic occurrences of black shale deposition, including the sapropels of the Mediterranean Sea (S), which are considered near-modern analogs of the Paleozoic and Mesozoic black shales. Modified from the compilation of Negri et al. [2006].

ancient black shales [e.g., Erbacher et al., 2001; Meyers, 2006; Negri et al., 2006; Emeis and Weissert, 2009]. The widespread distributions of some of the mid-Cretaceous black shale occurrences in sediments deposited concurrently in former epicontinental seas and in deep sea areas challenge the possibility that they record former upwelling systems and indicate that conditions quite unlike those of the modern ocean have recurred extensively in the past.

In addition to having organic carbon concentrations that are unusually high for marine sediments, another notable feature of most black shales is carbon isotopic compositions that are unusually depleted in ¹³C relative to modern marine organic matter (Table 1). This feature has been recognized since the advent of carbon isotope measurements [e.g., Craig, 1953]. Proterozoic, Paleozoic, and Mesozoic black shales commonly

Table 1. Total Organic Carbon (TOC) Concentrations and Carbon Isotopic Compositions of Marine Organic Matter in Representative Phanerozoic Black Shales^a

Formation-Location	Age (Ma)	TOC Range (%)	$\delta^{13}\text{C}$ Range (‰)	Reference
<i>Paleozoic Black Shale Sequences</i>				
Xiaotan Section, China	Terreneuvian (~525)	1–4	–36 to –21	Cremonese et al. [2013]
Vinini Creek, NV	Katian (~450)	1–2	–32 to –31	LaPorte et al. [2009]
Decorah Fm, IA	Katian (~450)	1–20	–31 to –23	Pancost et al. [2013a]
Perdix, AL, Canada	Frasnian (~380 Ma)	0.2–2.5	–30 to –25	Śliwiński et al. [2011]
Woodford Shale, OK	Frasnian (~375 Ma)	4–8	–29 to –28	Quan et al. [2013]
New Albany Shale, IN	Frasnian (~375 Ma)	10–16	–29 to –27	de la Rue et al. [2007]
Ohio Shale, KY	Frasnian (~375 Ma)	n.a. ^b	–29.2	Maynard [1981]
Hanover Shale, NY	Frasnian (~375 Ma)	2.5–4.6	–29 to –28	Tuite and Macko [2013]
Catskill Shale, NY ^c	Frasnian (~375 Ma)	n.a.	–24.8	Maynard [1981]
<i>Mesozoic Black Shale Sequences</i>				
Perth Basin, Australia	Induan (~250)	0.2–1.5	–31 to –27	Grice et al. [2005]
Idrijca Valley, Slovenia	Scythian (~250)	0.1–0.2	–26 to –25	Schwab and Spangenberg [2004]
Eiberg Basin, Austria	Rhaetian (~201)	0.1–15	–33 to –25	Bonis et al. [2010]
St. Adrie's Bay, UK	Rhaetian (~201)	n.a.	–29 to –25	Hesselbo et al. [2002]
St. Adrie's Bay, UK	Hettangian (198)	n.a.	–30 to –26	Hesselbo et al. [2002]
Posidonia Shale, Germany	Toarcian OAE (~183)	1–16	–34 to –27	Röhl et al. [2001]
Yorkshire, UK	Pre-Toarcian OAE (~184)	2–3	–26 to –25	Kemp et al. [2011]
Yorkshire, UK	Toarcian OAE (~183)	4–19	–34 to –29	Kemp et al. [2011]
Yorkshire, UK	Post-Toarcian OAE (~181)	3–4	–28 to –27	Kemp et al. [2011]
Whitby Mudstone, UK	Toarcian OAE (~183)	12–16	–30.9	Saalen et al. [2000]
Oxford Clay, UK	Oxfordian OAE (~161)	4–16	–28.0	Kenig et al. [1994]
Kimmeridge Clay, UK	Kimmeridgian OAE (~153)	1–35	–29 to –21.5	Morgans-Bell et al. [2001]
Livello Selli, Roter Sattel, Switz	OAE 1a, Aptian (120)	0.5–5	–28.5 to –24.5	Menegatti et al. [1998]
Livello Selli, Cismon, Italy	OAE 1a, Aptian (120)	0.1–8	–26.3 to –21.8	Menegatti et al. [1998]
Santa Rosa Canyon, Mexico	OAE 1a, Aptian (120)	0.2–2.6	–25.5 to –23.4	Bralower et al. [1999]
Site 1207, Shatsky Rise, N. Pacific	OAE 1a, Aptian (120)	10–40	–27.0 to –25.4	Dumitrescu and Brassell [2006a]
Santa Rosa Canyon, Mexico	OAE 1b, Albian (~111)	0.1–1.2	–26.3 to –23.2	Bralower et al. [1999]
Site 1049, western N. Atlantic	OAE 1b, Albian (~111)	1–6	–24 to –17	Kuypers et al. [2002b]
Site 1276, Newfoundland Basin	Mid-Cenomanian Event (95)	1–3.3	~26.5	Arnaboldi and Meyers [2006a]
Livello Bonarelli, Gorgo Cerbara, Italy	Cenomanian (~94)	1–12	–27 to –24	Kuroda et al. [2007]
Site 1258, Demerara Rise, N. Atlantic	Cenomanian (~94)	5–15	~–28	Erbacher et al. [2005]
Livello Bonarelli, Gorgo Cerbara, Italy	OAE 2 (93.6)	1–25	–24 to –23	Kuroda et al. [2007]
Site 1258, Demerara Rise, N. Atlantic	OAE 2 (93.6)	5–29	~–22	Erbacher et al. [2005]
Site 1258, Demerara Rise, N. Atlantic	Turonian (~92)	3–13	~–27	Erbacher et al. [2005]
Site 1257, Demerara Rise, N. Atlantic	OAE 3 (86)	8–12	~–27	Meyers and Bernasconi [2006]
Monterey Formation, CA	Miocene (~12 Ma)	5–16	~–22.5	Schimmelmann et al. [2001]
<i>Quaternary Sequences</i>				
Benguela upwelling system, S. Atlantic	Pleistocene (~2)	2–12	~–21	Robinson and Meyers [2002]
Site 967, Mediterranean Sapropels	Pleistocene (1)	2–4	–23.4	Meyers and Arnaboldi [2008]
Algal organic matter	Modern		–22 to –18	Emerson and Hedges [1988]

^aCarbon isotopic composition of typical modern marine organic matter is included for comparison.

^bn.a.—not available.

^cnonmarine.

have $\delta^{13}\text{C}_{\text{org}}$ values between -32‰ and -26‰ [e.g., Towe, 1982; Arthur et al., 1985; Dean et al., 1986; Ho et al., 1990; Jenkyns, 2010], whereas modern marine organic matter has values between -22‰ and -18‰ [Emerson and Hedges, 1988; Meyers, 1994; Galimov, 2006]. A number of hypotheses have been presented and discussed to explain the difference between the isotopic compositions of black shales and modern organic matter [e.g., Dean et al., 1986; Saun et al., 2008; Jenkyns, 2010]. This contribution reconsiders these hypotheses in the light of new information and fresh insights that have been made available by advances in isotope analysis of organic matter in black shales. This revisit will necessarily be limited to black shales of the Phanerozoic eon because of the paucity of information about Proterozoic sequences.

2. Controls on Biological Carbon Isotopic Compositions

The isotopic composition of organic carbon is widely used as a general indicator of the biological source of organic matter in sedimentary sequences [e.g., Meyers, 1994]. The initial biosynthetic step—assimilation of CO_2 by photosynthetic autotrophs—discriminates against ^{13}C and thereby imparts a distinctive ^{13}C -depleted signature that distinguishes biotic and abiotic carbon. Nearly all plants have used the Calvin C_3 pathway since the advent of photosynthesis, which on average discriminates against ^{13}C by 20‰ during

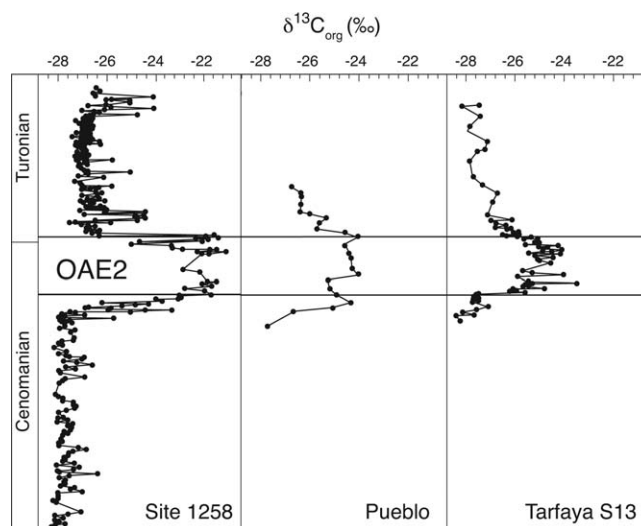


Figure 2. Three examples of the typical organic carbon isotope excursion to less negative values that are associated with high marine productivity during the Cenomanian-Turonian oceanic anoxic event (OAE 2). Ocean Drilling Program Site 1258 is on the Demerara Rise in the tropical western North Atlantic Ocean (data from *Erbacher et al.* [2005]). The paleoposition of Site 1258 during OAE 2 was close to the equator. The Pueblo (Colorado) sequence was deposited in the Western Interior Seaway of North America at a paleolatitude of about 45°N (data from *Pratt et al.* [1993]). The OAE 2 sequence in Core S13 from the Tarfaya Basin on the Moroccan margin was deposited at a paleolatitude of about 15°N (data from *Kolonik et al.* [2005]). At these three locations $\delta^{13}\text{C}$ values were $\sim -28\text{‰}$ prior to OAE 2 and $\sim -27\text{‰}$ after it, although the magnitudes of the isotopic excursions differed.

that can modify the isotopic composition of the bioavailable inorganic carbon, *Rau* [1978] notes that algal organic matter produced in a small montane lake is exceptionally light ($\delta^{13}\text{C} = -45\text{‰}$) because in-lake oxidation of organic matter and recycling of isotopically light biogenic carbon results in a ^{13}C -depleted dissolved CO_2 photosynthetic pool. As an example of how abundance of the carbon can be important, *Rau et al.* [1989] have documented that the $\delta^{13}\text{C}$ values of organic matter synthesized by algae in the frigid waters surrounding Antarctica can be as negative as -28‰ . These low $\delta^{13}\text{C}$ values exist because the cold waters contain a substantially larger concentration of dissolved CO_2 than can be accommodated in temperate or tropical ocean waters, thereby allowing greater photosynthetic discrimination against ^{13}C .

Rates of algal productivity also influence the carbon isotopic composition of the organic matter that becomes buried in sediments. Concentrations of the dissolved CO_2 available to algae are commonly drawn down during times of elevated productivity in both oceans and lakes. The lower concentrations lead to a decrease in isotopic discrimination against ^{13}C [e.g., *Fogel and Cifuentes*, 1993] and to less negative $\delta^{13}\text{C}$ values in aquatic organic matter. The shifts to less negative $\delta^{13}\text{C}$ values that have been documented in Cenomanian-Turonian black shale sequences (Figure 2) have been interpreted to indicate that greatly amplified marine productivity accompanied their deposition [*Erbacher et al.*, 1996; *Kuypers et al.*, 2002b; *Tsikos et al.*, 2004]. Such isotopic excursions are sometimes sufficiently well developed in black shale sequences that they can be used for chemochronological correlations [e.g., *Pratt et al.*, 1993; *Tsikos et al.*, 2004; *Erbacher et al.*, 2005; *Jenkyns*, 2010].

3. Carbon Isotopic Signatures of Phanerozoic Black Shale Sequences

Examples of the ranges of organic carbon concentrations and isotopic compositions in black shales dating from the Cambrian to the Pleistocene are summarized in Table 1. Because subduction of seafloor has erased virtually all of the marine geological record prior to the late Middle Jurassic (~ 165 Ma), older black shale sequences represent relatively shallow marine paleosettings on continental crust in epicontinental seas or on continental margins. In contrast, Cretaceous and younger sequences are known from both shallow to

biosynthesis of organic matter [cf. *O'Leary*, 1988; *Wolfe et al.*, 2001]. The $\delta^{13}\text{C}$ value of atmospheric CO_2 in the Holocene has been about -7‰ , which yields an average $\delta^{13}\text{C}$ value of -27‰ for land plant organic matter. In contrast, the $\delta^{13}\text{C}$ value of oceanic HCO_3^- is about 0‰ [*Degens*, 1969; *Galimov*, 2006], which leads to $\delta^{13}\text{C}$ values that average roughly -20‰ for marine algal organic matter.

The source and abundance of the carbon utilized during photosynthesis can yield isotopic differences in organic matter that are diagnostic of different environmental processes and conditions. The difference between the isotopic signatures of land-plant and marine organic matter is a clear and common example of the importance of the isotopic composition of the carbon that is assimilated by organisms to the $\delta^{13}\text{C}$ values of their resulting organic matter. As an example of a special circumstance

fairly deep (>2 km) paleodepths and hence give a more comprehensive geographic record of the paleoceanographic conditions that existed during their deposition.

3.1. Paleozoic Black Shale Sequences

Isotopically light black shales have been described from Cambrian, Ordovician, and Devonian epicontinental sea sedimentary sequences that exist in China, North America, and Europe. The Xiaotan Section of South China (~550 to ~520 Ma) was deposited on a carbonate bank that gradually shoaled over time to provide a continuous and unaltered marine record of the Precambrian-Cambrian boundary and the early Cambrian (Terreneuvian). Its organic $\delta^{13}\text{C}$ values are about -29‰ through most of the 600 m thick sequence. However, two black shale layers in the lower Terreneuvian have isotope values between -36‰ and -34‰ and are separated by a low-TOC layer that has values between -25‰ and -21‰ [Cremonese *et al.*, 2013]. Except for this one layer, the early Cambrian $\delta^{13}\text{C}$ values in the Xiaotan Section are consistently lighter than the range of -22‰ and -18‰ that is typical of modern marine organic matter (Table 1).

Black shale sequences were deposited at numerous locations now found in North America, northwestern Europe, and southern China during the Katian Age (456–445 Ma) of the Late Ordovician Epoch. The sequences were deposited in tropical settings that included carbonate banks and foreland basins [e.g., Young *et al.*, 2008; Pancost *et al.*, 2013a]. Their organic $\delta^{13}\text{C}$ values generally fall between -32‰ and -27‰ [Young *et al.*, 2008; LaPorte *et al.*, 2009; Pancost *et al.*, 2013a], and they commonly include a distinctive dramatic excursion to values as high as -23‰ [Pancost *et al.*, 2013a]. This excursion to less negative values is known as the Guttenberg Carbon Isotope Excursion (GICE) and is also expressed in carbonate $\delta^{13}\text{C}$ values [Young *et al.*, 2005]. It represents a smaller difference between carbonate and organic carbon isotopic compositions [Young *et al.*, 2008] that, Pancost *et al.* [2013a, 2013b] conclude, was at least partially caused by a change in marine algal communities at this time that is recorded by the different molecular biomarker compositions of the GICE sections.

Late Devonian black shale sequences were deposited at several times and places in North America. The Perdrix Formation of Alberta was deposited on a carbonate ramp during the Frasnian Age (385–375 Ma) in an equatorial setting. Its organic $\delta^{13}\text{C}$ values range between -30‰ and -27‰ [Śliwiński *et al.*, 2011]. The widespread latest Frasnian (~375 Ma) black shales of central and eastern North America are locally known as the Woodford, New Albany, Antrim, Chattanooga, Hanover, and Ohio shales, yet they constitute a generally contiguous accumulation that was deposited in the epicontinental seas that flooded foreland basins west of the Appalachian Mountains. Their organic $\delta^{13}\text{C}$ values typically fall between -29‰ and -28‰ . In comparison, $\delta^{13}\text{C}$ value of the same-age nonmarine Catskill Shale of New York State is -24.8‰ [Maynard, 1981], which is in the range of modern lake sediments [Meyers and Ishiwatari, 1993]. Even including the Catskill Shale, the $\delta^{13}\text{C}$ values of virtually all of the marine Paleozoic black shales are more negative than those of modern marine organic matter (Table 1).

3.2. Mesozoic Black Shale Sequences

Isotopically light organic carbon is also typical of most Mesozoic black shales, beginning in the Early Triassic and continuing into the middle Cretaceous (Table 1). The mid-Cretaceous sequences have been found at many locations and have been especially widely studied. These sequences are the basis of the term “Oceanic Anoxic Event” (OAE) that was introduced by Schlanger and Jenkyns [1976] to express how they may have originated. This term has since become widely used to help describe black shale sequences deposited at specific times in the Mesozoic, although its interpretation remains contentious.

The massive biotic extinction event that helps to define the Permian-Triassic boundary (~251 Ma) was accompanied by deposition of ^{13}C -depleted organic carbon enriched rocks at a number of now widespread locations. The Gujo-Hachiman Permian-Triassic boundary section of Honshu, Japan, contains a 65 cm thick black shale layer that contains over 2% TOC [Algeo *et al.*, 2010]. The Induan age petroleum source rocks of the Perth Basin of Australia contain up to 1.5% TOC and exhibit $\delta^{13}\text{C}$ values that fall between -31‰ and -27‰ [Grice *et al.*, 2005]. The Permian-Triassic boundary sequence in western Slovenia studied by Schwab and Spangenberg [2004] is similarly isotopically light. Its organic $\delta^{13}\text{C}$ values fall between -28.5‰ and -26.5‰ in upper Permian limestones and decrease to between -30‰ and -28.7‰ in lower Triassic laminated limestones, with an intervening less negative excursion to between -27‰ and -25.4‰ in the dark colored boundary packstones.

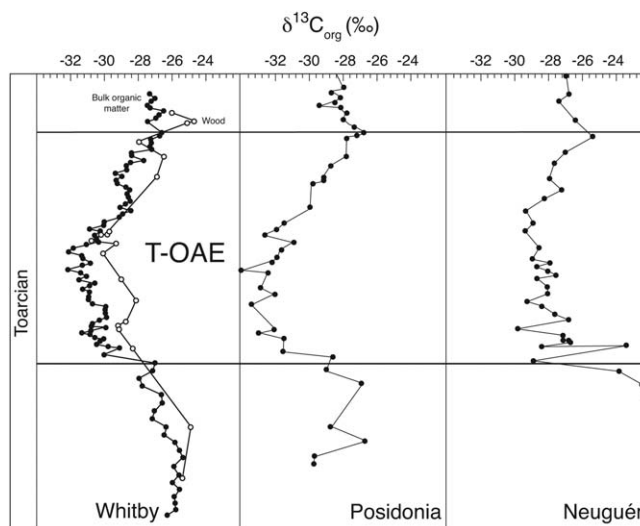


Figure 3. Three examples of typical organic carbon isotope excursions to more negative values associated with the Toarcian oceanic anoxic event (T-OAE). The Whitby Mudstone section in Yorkshire, England shows the excursions in both bulk organic matter and wood fragments (data from Hesselbo *et al.* [2000]). The Posidonia Shale section is in Bavaria, Germany (data from Röhl *et al.* [2001]). The Neuquén Basin is in Mendoza Province, Argentina (data from Mazzini *et al.* [2010]).

–2‰ in the upper Rhaetian section that has the least negative organic $\delta^{13}\text{C}$ values [Morante and Hallam, 1996]. The different organic and inorganic isotopic patterns in the upper Rhaetian sequence has been postulated by Morante and Hallam [1996] to result from diagenetic alteration of organic matter, yet a detailed analysis of the Triassic–Jurassic boundary sequences at St. Audrie’s Bay, UK, and in the Kap Stewart Formation of East Greenland yields the same organic carbon isotopic patterns that are found in the Austrian sites [Hesselbo *et al.*, 2002], which argues against localized diagenesis.

3.2.1. Jurassic Black Shales

Well-developed black shale sequences were deposited during three ages of the Jurassic period—the early Toarcian (~183 Ma), the early Oxfordian (~161 Ma), and in the Kimmeridgean (156–151 Ma). Of these three episodes, the one emplaced during the early Toarcian is particularly extensive and has been recognized globally [Jenkyns, 1988]. Black shale sequences of this age are especially well expressed in Europe, where they are the basis for the term “Toarcian Oceanic Anoxic Event” (T-OAE). However, true anoxia was evidently rare in the paleoseas under which the T-OAE was deposited. For example, the presence of benthic macrofauna in the Posidonia Shale of southern Germany mandates at least punctuated interruptions of oxygenation of bottom waters within a generally low-oxygen environment [Röhl *et al.*, 2001].

All of the Toarcian black shale sequences contain abundant marine organic matter that is isotopically light (Figure 3). For example, TOC concentrations in the Posidonia Shale vary between 1% and 16%, but generally are between 4% and 10% [Röhl *et al.*, 2001]. Organic $\delta^{13}\text{C}$ values in this sequence vary inversely with the TOC concentrations; they drop to as low as –34‰ in conjunction with peak carbon concentrations of the T-OAE from a background of about –27‰ [Röhl *et al.*, 2001]. Similar patterns exist in the T-OAE sequences at Sancerre in the Paris Basin of France and in Yorkshire, England. At Sancerre, TOC concentrations are <1% before the T-OAE and peak around 8–10% at it, and $\delta^{13}\text{C}$ values decrease from background values of about –25‰ before the T-OAE to as low as –33‰ at it before returning to values of about –26‰ above it [Hermoso *et al.*, 2012]. In Yorkshire, concentrations are ~2% before the event, rise to as high as 19% within it, and then relax to ~3% after it, and the $\delta^{13}\text{C}$ values drop to as low as –34‰ within the T-OAE [Kemp *et al.*, 2011]. The dramatic negative excursions that accompanied magnified organic matter accumulation have been important to seeking explanation of the conditions that existed during the deposition of the T-OAE. However, the isotopically light organic carbon isotopic compositions evident in sediments deposited before and after the T-OAE (Figure 3) are important geochemical features that are often overlooked.

Similar to the Permian–Triassic boundary, the massive extinction event that occurred at the Triassic–Jurassic boundary was accompanied by carbon isotopic perturbations. Deposition of thin TOC-rich black shale sequences occurred at a few boundary locations and is best documented in Austria, where TOC concentrations peaked from 4% to 14% at different Rhaetian sections and are accompanied by negative isotopic excursions to –33‰ to –31‰ [Rühl *et al.*, 2009]. The isotopic fluctuations across the Triassic–Jurassic boundary sections in Austria are complicated. Organic $\delta^{13}\text{C}$ values are about –28‰ before black shale deposition and about –25‰ after it before decreasing again to about –28‰ in Hettangian sections about the boundary. However, inorganic $\delta^{13}\text{C}$ values change from +2‰ below and above the boundary to

Table 2. Carbon Isotopic Compositions of Marine Organic Matter Components in Phanerozoic Black Shales and Adjacent Rocks

Sample Description	Age (Ma)	$\delta^{13}\text{C}_{\text{comp}}$ (‰)	$\delta^{13}\text{C}_{\text{tot}}$ (‰)	$\Delta\delta^{13}\text{C}$ (‰)	Reference
Phytane, Whitby Mudstone, UK	Toarcian (~181)	-29.7	-28	1.3	Saelen et al. [2000]
Phytane, Oxford Clay, UK	Oxfordian (~161)	-31.5	-28	3.5	Kenig et al. [1994]
Phytane, Kimmeridge Clay, UK	Kimmeridgean (~153)	-30.0	-22.7	7.3	Saelen et al. [2000]
Porphyryns, Piobiccio, Italy	OAE 1a (120)	-18.9	-21	-2.1	Kashiyama et al. [2008]
Alkenones, Shatsky Rise, N. Pacific	OAE 1a (120)	-31.6	-26.4	3.2	Dumitrescu and Brassell [2006a]
C ₂₇ Sterols, DSDP Site 545	OAE 1b (~111)	-33.0	-27	6	Wagner et al. [2008]
Porphyryns, Greenhorn Fm, KS	Cenomanian (~94)	-28	-27	1	Hayes et al. [1989]
Phytane, Tarfaya, Morocco	Cenomanian (~94)	-29.9	-27	2.9	Tsikos et al. [2004]
Phytane, DSDP Site 367	Cenomanian (~94)	-30.5	-28	2.5	Kuypers et al. [2002b]
Phytane, DSDP Site 530	Cenomanian (~94)	-31.4	-26.5	2.9	Forster et al. [2008]
Phytane, ODP Site 1260	Cenomanian (~94)	-31	-28	3	van Bentum et al. [2012]
Phytane, DSDP Site 367	OAE 2, (93.6)	-25	-23	2	Kuypers et al. [2002b]
Phytane, DSDP Site 530	OAE 2, (93.6)	-29	-25	4	Forster et al. [2008]
Phytane, ODP Site 1260	OAE 2, (93.6)	-24	-22.5	1.5	van Bentum et al. [2012]
Phytane, Tarfaya, Morocco	OAE 2, (93.6)	-25.6	-24.2	1.4	Tsikos et al. [2004]
Porphyryns, Greenhorn Fm, KS	OAE 2 (93.6)	-25	-23.5	1.5	Hayes et al. [1989]
Porphyryns, Piobiccio, Italy	OAE 2 (93.6)	-18.5	-23.5	-5.0	Kashiyama et al. [2008]
Phytane, Gubbio, Italy	OAE 2, (93.6)	-28	-25	3	Tsikos et al. [2004]
Phytane, DSDP Site 530	Turonian (~92)	-31	-27.2	3.8	Forster et al. [2008]
Phytane, ODP Site 1260	Turonian (~92)	-30	-27	3	van Bentum et al. [2012]
Phytane, Tarfaya, Morocco	Turonian (~92)	-28	-26.5	1.5	Tsikos et al. [2004]
Algal organic matter	Modern		-22 to -18		Emerson and Hedges [1988]

More examples of isotopically light organic carbon in Toarcian sequences exist. Saelen et al. [2000] report that the $\delta^{13}\text{C}$ values of total organic carbon in a section of the Whitby Mudstone Formation from eastern England range between -31.6‰ and -26.9‰ and average -28‰ , whereas those of phytane, which is a biomarker derived from the phytol sidechain of chlorophyll, average -29.7‰ (Table 2). The somewhat greater ^{13}C depletion of phytane relative to TOC is typical of long-carbon-chain compounds and is consistent with a common biosynthetic origin for both substances. In the Whitby Mudstone section from Yorkshire, UK, Cohen et al. [2004] describe a marked negative excursion to $\delta^{13}\text{C}$ values of -32‰ in the OAE from background values averaging -27.2‰ before and -25.5‰ after it.

Light isotopic compositions also appear in TOC-rich sequences deposited later in the Jurassic. For example, organic carbon concentrations in the Oxford Clay (~161 Ma) of central and southern England are typically between 2% and 10% and sometimes reach as high as 16% [Kenig et al., 1994]. The $\delta^{13}\text{C}$ values of total organic carbon range between -31.6‰ and -23.1‰ and average -28‰ , whereas those of phytane average -31.5‰ (Table 2) [Kenig et al., 1994]. On the premise that the isotopic composition of this saturated hydrocarbon would be better preserved than that of bulk organic matter and using the general principle that alkyl compounds are isotopically lighter than bulk organic matter, Kenig et al. [1994] estimated that the primary organic matter inputs to the Peterborough Member of the Oxford Clay Formation had a $\delta^{13}\text{C}$ value of -28.2‰ , which is virtually identical to the measured average in this unit (Table 1).

The general pattern of low organic $\delta^{13}\text{C}$ values in Jurassic black shales also exists in the Kimmeridge Clay Formation (156–151 Ma) on the south Dorset coast of England, which is the type section of the Kimmeridgian source rock of the North Sea oilfields. This formation is extraordinarily rich in organic matter and has TOC concentrations that average ~10% and can exceed 50% (P. Wignall, personal communication, 2014). Its $\delta^{13}\text{C}$ values range between -29 and -21.5‰ (Table 1) and are less negative in units with the higher TOC concentrations. Morgans-Bell et al. [2001] postulate that this pattern results from the preferential removal of ^{12}C from seawater during episodes of elevated marine productivity that is observed in other black shales [e.g., Erbacher et al., 2005; Pancost et al., 2013a]. Like other Mesozoic black shales, the $\delta^{13}\text{C}$ values of the biomarker phytane in this formation average -30‰ (Table 2), which suggests that the isotopic composition of bulk organic matter was light throughout deposition of the Kimmeridge Clay Formation [Saelen et al., 2000].

3.2.2. Cretaceous Black Shales

The record of sediment accumulation on the floors of the Cretaceous oceans is extensive and has been studied both as sequences of existing ocean floor and as sequences of former ocean floor exposed on continents. Multiple black shale sequences were deposited at intervals during ~50 Myr starting in the Early

Cretaceous Epoch (Figure 1). The oldest of these is the Valanginian Weissert event (~ 133 Ma), which was followed first by the latest Hauterivian Faroani event (~ 127 Ma) and then the Barremian event (~ 124 Ma) before the Selli event of the early Aptian (OAE 1a, ~ 120 Ma), which represents a particularly well-developed and widespread black shale occurrence. The OAE 1a was followed by the Aptian-Albian Paquier event (OAE 1b, 113–109 Ma), the Albian OAE 1c (~ 102 Ma), and the late Albian Breistroffer event (OAE 1d, ~ 100 Ma), but none of these events are well developed or widely expressed as the OAE 1a. Early in the Late Cretaceous Epoch, the mid-Cenomanian event (~ 95 Ma) was deposited at scattered locations and is considered to be a prelude to the Bonarelli event at the Cenomanian-Turonian boundary (OAE 2, ~ 94 Ma), which is the best preserved OAE in the geological record and appears to be the most strongly developed. The final Cretaceous black shale event was the Coniacian-Santonian OAE 3 (~ 86 Ma), which is known from only a few locations. Most of these Cretaceous black shales have been subjects of multiple studies, and many of these studies have included analyses of the carbon isotopic compositions of their bulk organic matter. The $\delta^{13}\text{C}$ values of the bulk organic matter in these sequences are consistently more negative than modern marine organic matter, with the important exception of a shift to less negative values that is common to the peak expression of the OAE 2 (Table 1 and Figure 2).

Both the late Aptian OAE 1a and Cenomanian-Turonian OAE 2 black shale events had essentially global distributions [e.g., *Jenkyns*, 2010], and the strong expression and abundant occurrence of these sequences have made them the most studied. *Menegatti et al.* [1998] compare the organic $\delta^{13}\text{C}$ profiles of OAE 1a sequences at Roter Sattel and Cison, which were deposited on the western and eastern continental margins of the Tethys Ocean and are now uplifted and exposed, respectively, in the Swiss Prealps and the Venetian Alps of northern Italy. In both sequences, the $\delta^{13}\text{C}$ values increase up-section, but they are offset by about 2‰ , the values changing from -28.5‰ to -24.5‰ at Roter Sattel and from -26.3‰ to -21.8‰ at Cison (Table 1). A similar up-section shift is described by *Bralower et al.* [1999] in the OAE 1a sequence at Santa Rosa Canyon in the Sierra Madre of northeastern Mexico. The values here change from -25.5‰ to -23.4‰ (Table 1). In contrast, the organic $\delta^{13}\text{C}$ values of the OAE 1a sequence in ODP Hole 1207B on the Shatsky Rise of the North Pacific fluctuate between -27.0‰ and -25.4‰ but remain close to -26.4‰ throughout the sequence, which contains up to 40% TOC [*Dumitrescu and Brassell*, 2006a]. Although these four late Aptian sequences have isotopic compositions that are consistently more depleted in ^{13}C than modern marine organic matter, the amounts of the ^{13}C depletion differ at each location, which suggests that local factors at least partially influenced their isotopic compositions.

Like the OAE 1a black shale sequences, the $\delta^{13}\text{C}$ values of OAE 2 sequences commonly differ at different locations. As one example, the excursions to less negative values that are distinguishing features of this OAE reach averages of -22‰ on the Demerara Rise, -24‰ at Pueblo, and -25‰ at Tarfaya (Figure 2). Because these excursions are generally interpreted as consequences of magnified organic matter production during the OAE that diminished the availability of dissolved inorganic carbon to the marine photoautotrophs and hence lessened their isotopic discrimination against ^{13}C [*Erbacher et al.*, 1996; *Kuypers et al.*, 2002b; *Tsikos et al.*, 2004], their differences imply regional influences in the magnified production. Although the OAE 2 $\delta^{13}\text{C}$ values of these three sequences differ, their background isotopic compositions are very similar. The background $\delta^{13}\text{C}$ values in the Cenomanian sediments before the OAE are about -28‰ and those that followed it in the Turonian are about -27‰ (Figure 2). These values are systematically and markedly lower than in modern marine organic matter and also lower than those reported in most Cenozoic marine sediments.

As found in Jurassic black shales, the $\delta^{13}\text{C}$ values of the biomarker molecule phytane are consistently lower than those of total organic carbon in Cretaceous black shales (Table 2). The difference is usually about 2‰ , which is similar to the Jurassic examples and is expected for this biomarker, and it indicates a more or less common source of the phytane and the bulk organic matter in these sequences. However, the difference expands to $\sim 4\text{‰}$ in the OAE 2 and the Turonian intervals of the Site 530 sequence (Table 2), which might indicate that the bulk organic matter originated from a different source than the chlorophyll-derived phytane. As postulated by *Kuypers et al.* [2004] and *Kashiyama et al.* [2008], the production of organic matter in parts of the Site 530 sequence might have been dominated by archaea instead of algae. Archaea employ pigments in addition to the chlorophyll that is the origin of phytane and that is the dominant photosynthetic pigment in algae. They also can use phosphoenolpyruvate carboxylase (PEPC) instead of ribulose-1,5-bisphosphate carboxylase oxygenase (RuBisCO) during photosynthetic uptake of carbon to yield less discrimination against ^{13}C .

3.3. Cenozoic Organic-Carbon-Rich Marine Sequences

Marine sediments rich in organic carbon have not been deposited widely during the Cenozoic Era, but they have accumulated in two very different kinds of special settings. One kind of special setting is where wind fields are suitably geographically positioned to blow surface waters aside so that underlying waters can upwell and deliver the nutrients necessary to sustain high rates of marine organic matter production. The circumstances necessary to support these upwelling systems are limited by the global pattern of atmospheric circulation and by the positioning of continental margins. These areas are found along the equator, along the eastern margins of the Atlantic and Pacific oceans, and in the northern Indian Ocean, which collectively represent only about 1% of the surface area of the modern ocean. This limited geographic occurrence is unlikely to have been larger in the past. In addition, it may have been further constrained by the availability of nutrient-rich water masses to be upwelled inasmuch as the best developed examples of upwelling-derived sedimentary sequences date from the last half of the Cenozoic Era. The mid-Miocene (~12 Ma) Monterey Formation of California is one of the older Cenozoic sequences that was deposited under an upwelling system. It is enriched in organic matter, containing 5–16% organic carbon that has a carbon isotopic value that averages -22.5‰ [Schimmelmann *et al.*, 2001], which is similar to the average of $\sim 21\text{‰}$ in early Pleistocene sediments deposited under the Benguela Current upwelling system in the southeastern Atlantic Ocean [Robinson and Meyers, 2002] and which is typical of modern marine organic matter (Table 1).

The Mediterranean Sea is the second special setting in which sediments unusually rich in organic carbon have accumulated during the Cenozoic (Figure 1). The sedimentary record of this sea contains multiple layers of “sapropels” that have been periodically deposited since the late Miocene and are particularly well developed since the late Pliocene. Their occurrence is especially notable because the region around the Mediterranean Sea has a semiarid climate, giving the sea a negative water balance. This hydrologic situation results in downwelling of surface waters and their nutrient depletion, which has led to a low production of marine organic matter during most of the paleoceanographic history of the Mediterranean. However, during times of wetter climate in the past, the productivity of the sea evidently was amplified and resulted in deposition of the sapropels. These times correspond to precessional (~ 20 kyr) orbital cycles that enhanced regional summer-winter seasonal differences. The sapropels exist as decimeter-scale layers of organic-carbon-rich sediment within a predominantly organic-carbon-poor matrix. Their organic carbon concentrations are typically between 2 and 4% and sometimes reach 30% [Arnaboldi and Meyers, 2006b]. Their carbon isotopic values commonly vary between -24 and -21‰ [Meyers and Arnaboldi, 2006b], which like the Monterey Formation is not very different from modern marine organic matter (Table 1). Although the oceanographic conditions in the Mediterranean Sea are unusual and are not duplicated elsewhere in the modern ocean, the sapropels are often considered near-modern analogs for the Paleozoic and Mesozoic black shales [e.g., Meyers, 2006; Negri *et al.*, 2006; Emeis and Weissert, 2009].

4. Hypotheses for Light Carbon Isotopic Signatures of Black Shales

A variety of hypotheses have been advanced to explain why the $\delta^{13}\text{C}$ values of Phanerozoic black shales are more negative than modern marine organic matter and are often even more negative than those of land-derived organic matter. Five of the major concepts that are the bases of these hypotheses are summarized in Table 3. An earlier hypothesis that the organic matter in black shales was indeed derived from land plants was extensively evaluated and rejected by Dean *et al.* [1986] on the basis of their compilation of Rock Eval pyrolysis data from a large number of Cretaceous black shales. The dominance of marine organic matter in almost all black shales has subsequently been verified in many investigations [e.g., Dumitrescu and Brassell, 2006b; Meyers, 2006]. Different aspects of the five existing hypotheses have been variously presented in detail by Dean *et al.* [1986], Röhl *et al.* [2001], Suan *et al.* [2008], and Jenkyns [2010]. Their principles are discussed and evaluated in the following sections.

4.1. Diagenetic Enrichment of ^{13}C -Depleted Organic Matter Components

The principle behind this hypothesis involves selective degradation of ^{13}C -richer forms of organic matter (carbohydrates, proteins) that amplifies the proportion of ^{13}C -depleted forms (lipids) in the surviving organic matter [e.g., Spiker and Hatcher, 1984]. As reported by Degens *et al.* [1968], the bulk organic matter $\delta^{13}\text{C}$ value of marine phytoplankton is about -20‰ and is the weighted average of proteins (-17‰),

Table 3. Hypotheses to Explain Light Isotopic Compositions of Marine Organic Matter in Phanerozoic Black Shales

Concept	Selected References
Enhanced preservation of ^{13}C -depleted lipid fraction of organic matter	<i>Kaplan and Nissenbaum</i> [1966]
Change in amount of photoautotrophic isotopic discrimination against ^{12}C	<i>Arthur et al.</i> [1985] and <i>Hayes et al.</i> [1999]
Massive release of ^{13}C -depleted carbon from marine methane hydrates	<i>Hesselbo et al.</i> [2000] and <i>Beerling et al.</i> [2002]
Massive delivery of mantle ^{13}C -depleted carbon by volcanic eruptions	<i>Larson</i> [1991] and <i>Kerr</i> [1998]
Greater availability of CO_2 during times of elevated atmospheric $p\text{CO}_2$	<i>Dean et al.</i> [1986] and <i>Freeman and Hayes</i> [1992]
Oxidation of organic matter and recycling of isotopically light carbon in photic zone	<i>Küspert et al.</i> [1982] and <i>Saalen et al.</i> [1998]

carbohydrates (-19‰), and lipids (-29‰). Proteins and carbohydrates contain nitrogen and oxygen functional groups that make them more liable to degradation and hence less likely to survive in sedimentary sequences. In contrast, lipids contain hydrocarbon-like carbon chains and fewer functional groups and hence have biochemical compositions that render them less sensitive to degradation.

In principle, preferential diagenetic loss of the protein and carbohydrate portions of marine organic matter should result in the remaining material having carbon isotopic compositions that approach those of the lipid components of the initial material. However, the potential effects of selective diagenesis appear to be no more than 2 to 3‰ [Dean et al., 1986; Prahel et al., 1997; Galimov, 2006] and not sufficient to explain the difference of 6–10‰ between modern marine organic matter and the organic matter in Phanerozoic black shales. Furthermore, the high organic carbon concentrations that are common to the black shales indicate unusually good preservation of their organic matter, a factor that would have limited the opportunities for diagenetic alteration of the carbon isotopic compositions of the black shales. These arguments have led most investigators to reject the diagenesis hypothesis [e.g., Dean et al., 1986; Saalen et al., 2000; Suan et al., 2008; LaPorte et al., 2009].

4.2. Massive Release of ^{13}C -Depleted Carbon From Marine Methane Hydrates

This hypothesis is based on the very low $\delta^{13}\text{C}$ value of the biogenic methane ($\sim -60\text{‰}$) that is sequestered in methane hydrates in the sea floor [Kvenvolden, 1993]. It calls for the massive dissociation of sea floor methane hydrates and release of their isotopically light C, thereby changing both the magnitude and isotopic compositions of the C reservoirs available for uptake by photoautotrophs [Hesselbo et al., 2000; Beerling et al., 2002; Kemp et al., 2005]. The stability range of methane hydrates in ocean sediments is defined by depth-related pressure and water temperature. Its upper limit can be as shallow as 250 m if the water temperature is very cold (-1.5°C), but this limit deepens to over 1400 m if the water temperature warms to 15°C [Dickens and Quinby-Hunt, 1994]. A possible trigger to destabilize methane hydrates would therefore be warmer bottom-water temperatures, as postulated by Hesselbo et al. [2000] to explain the -6‰ isotopic excursion they describe for the Toarcian Whitby Mudstone of Yorkshire, England. Warmer water temperatures are generally considered to have been typical of past periods of greenhouse climate [e.g., Bice et al., 2006; Royer et al., 2007; Meyer and Kump, 2008; Jenkyns, 2010] when most black shales were deposited.

If massive dissociations of seafloor methane hydrates were indeed responsible for the ^{13}C -depleted compositions of the marine organic matter in Phanerozoic black shales, the isotopically light C released from the hydrates would have been partitioned between seawater and the atmosphere, and the compositions of land plant organic matter that was formed at the same times as the black shales should also be depleted in ^{13}C . Evidence of such ^{13}C depletion actually exists in wood from Toarcian black shale sequences [Kemp et al., 2011]. Wood particles in the Toarcian OAE sections from both the Whitby Mudstone in England and from Peniche, Portugal, have $\delta^{13}\text{C}$ values of about -30‰ , which is about 5‰ more negative than in wood that was deposited before or after the OAE (Figure 3 and Table 4). However, $\delta^{13}\text{C}$ values of wood found in Devonian, Kimmeridgian, and Aptian OAEs and in charcoal from an Albian OAE 1d interval are not nearly so isotopically light and instead are more like the values of recent wood (Table 4). Moreover, the $\delta^{13}\text{C}$ values of plant-wax hydrocarbons extracted from Toarcian, Kimmeridgian, and Albian black shales are essentially the same as those from modern plant leaves (Table 4), which suggests that the atmospheric carbon pool was not notably ^{13}C -depleted for long periods of the Mesozoic, even though the bulk marine organic matter in most Mesozoic black shales is ^{13}C -depleted (Table 1). Similarly, the $\delta^{13}\text{C}$ value of the freshwater Catskill Shale that is 2–4‰ less negative than those of same-age marine black shales (Table 1) suggests that the Late Devonian atmospheric carbon reservoir was not ^{13}C -depleted.

Table 4. Carbon Isotopic Compositions of Land Plant Organic Matter Components in Phanerozoic Black Shales^a

Sample Description	Age (Ma)	$\delta^{13}\text{C}_{\text{comp}}$ (‰)	$\delta^{13}\text{C}_{\text{tot}}$ (‰)	$\Delta\delta^{13}\text{C}$ (‰)	Reference
Wood, Huron Shale, KY	Late Devonian (~375)	-26.8	n.a. ^b		Maynard [1981]
Wood, Oolite Shale, OH	Late Devonian (~375)	-25.6	n.a.		Maynard [1981]
Wood, East Greenland	Rhaetian (~201)	-26 to -23	n.a.		Hesselbo et al. [2002]
Wood, East Greenland	Hettangian (198)	-27 to -24	n.a.		Hesselbo et al. [2002]
Wood, Peniche, Portugal	Pre-Toarcian OAE (~184)	-24.5	n.a.		Hesselbo et al. [2007]
Wood, Whitby Mudstone, UK	Pre-Toarcian OAE (~184)	-25.5	-26	+0.5	Hesselbo et al. [2000]
Wood, Whitby Mudstone, UK	Toarcian OAE (~183)	-30	-32	+1.7	Hesselbo et al. [2000]
Wood, Peniche, Portugal	Toarcian OAE (~183)	-29.5	n.a.		Hesselbo et al. [2007]
C ₂₉ n-alkane, Whitby Mudstone, UK	Toarcian OAE (~183)	-29.2	-30.9	+1.7	Saalen et al. [2000]
Wood, Whitby Mudstone, UK	Post-Toarcian OAE (~181)	-26	-27	+1	Ando et al. [2002]
Wood, Peniche, Portugal	Post-Toarcian OAE (~180)	-23.0	-29.5	+6.5	Hesselbo et al. [2007]
Wood, Oxford Clay, UK	Oxfordian OAE (~161)	-23.8	n.a.		Kenig et al. [1994]
C ₂₉ n-alkane, Kimmeridge Clay, UK	Kimmeridgean OAE (~155)	-28.3	-21.7	-6.4	Saalen et al. [2000]
Wood, Hokkaido, Japan	OAE 1a (124)	-22.5	n.a.		Ando et al. [2002]
Wood, Hokkaido, Japan	Aptian (~120)	-24	n.a.		Ando et al. [2002]
C ₂₇₋₃₁ n-alkanes, Site 545	OAE 1b (~111)	-29.5	-27	-2.5	Wagner et al. [2008]
Charcoal, Dakota Fm, NE	Albian OAE1d (100)	-25.2	-26.2	+1	Gröcke et al. [2006]
C ₃₁ n-alkane, Site 367, C. Verde Basin	Cenomanian (~94)	-34.1	-27.5	-6.6	Sinninghe-Damsté et al. [2008]
C ₂₇ n-alkane, Site 367, C. Verde Basin	Cenomanian (~94)	-31.8	-27.5	-4.3	Sinninghe-Damsté et al. [2008]
C ₃₁ n-alkane, Site 367, C. Verde Basin	OAE 2 (93.6)	-26.5	-22.0	-4.5	Sinninghe-Damsté et al. [2008]
C ₂₇ n-alkane, Site 367, C. Verde Basin	OAE 2 (93.6)	-25.6	-22.0	-3.6	Sinninghe-Damsté et al. [2008]
C ₃₁ n-alkane, oak leaf, England	Modern	-35.7	-28.8	-6.9	Collister et al. [1994]
C ₂₇ n-alkane, oak leaf, England	Modern	-32.9	-28.8	-4.1	Collister et al. [1994]
Spruce wood, MI	Holocene	-23.6	n.a.		Meyers et al. [1995]
Spruce wood, MI	Modern	-23.1	n.a.		Meyers et al. [1995]
C ₃ land plant organic matter	Modern		-28 to -26		Degens [1969]

^a $\Delta\delta^{13}\text{C}$ represents the isotopic difference between the land plant component and the bulk organic matter of the black shale.

^bn.a.—not available.

Although the massive release of carbon from methane hydrates is an attractive hypothesis to explain the isotopically light compositions of Phanerozoic black shales, a number of arguments against this explanation exist. *Suan et al.* [2008] estimate that the duration of the Toarcian OAE to have been about 400 kyr, and they argue that the protracted release of methane necessary to maintain a ¹³C-depleted photosynthetic reservoir over this long period of time is unlikely. *Korte et al.* [2010] reach a similar conclusion for the 4–7‰ negative excursion in carbonates associated with three Permian-Triassic boundary sequences. Furthermore, the $\delta^{13}\text{C}$ values of sediments deposited before and after Phanerozoic black shales are commonly much more negative than modern marine organic matter (Figures 2 and 3 and Table 1), which indicates that the isotopically light carbon pools were maintained for multimillion year periods of time. Finally, questions remain about the size of the seafloor methane hydrate deposits that were available to provide the isotopically light carbon. Because the methane that is in hydrates is produced by microbial degradation of sediment organic matter, the size of sea floor hydrate deposits is limited by the availability of organic-carbon-rich sediments to fuel methanogenesis. With the important exception of the episodes of Phanerozoic black shale deposition, these kinds of sediments are not abundant in the paleoceanographic record. Earlier estimates of 8000–11,000 Gt of carbon [e.g., *Dickens et al.*, 1995] in the methane hydrate deposits of the modern ocean evidently exaggerated their magnitude. More recent estimates have decreased the magnitude to 500–2500 Gt of carbon [e.g., *Milkov*, 2004]. As calculated by *Röhl et al.* [2001], release of 700–1100 Gt of methane carbon would have been necessary to yield the low $\delta^{13}\text{C}$ values of Toarcian black shales, which would demand that at least a third and potentially all of the methane hydrates in the seafloor would need to have dissociated during this OAE. They conclude that such extensive and massive release was unlikely, and other explanations for the curiously low $\delta^{13}\text{C}$ values of black shales are necessary.

4.3. Massive Delivery of ¹³C-Depleted Magmatic Carbon From Earth's Mantle

This hypothesis is similar in principle to the methane hydrate hypothesis except that the source of the released large amount of isotopically light CO₂ that changed the C reservoirs available for uptake by photoautotrophs was magma eruptions from the Earth's mantle. Likely sources of the mantle CO₂ are seafloor spreading centers, which have released magma throughout the Phanerozoic, and large igneous provinces (LIPs), which have sporadically been emplaced through Earth history. The $\delta^{13}\text{C}$ values of CO₂ released from mantle-derived magma are mostly about -5‰, but values of as low as -25‰ are also common [*Deines*,

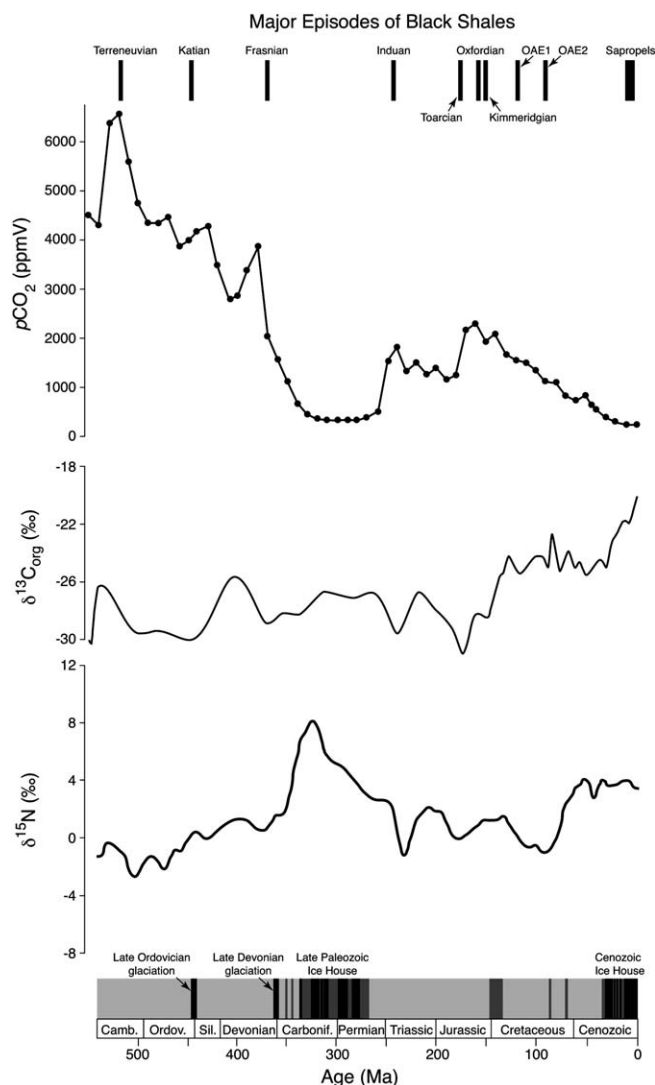


Figure 4. Relation of major episodes of Phanerozoic black shale deposition with variations in (top) atmospheric $p\text{CO}_2$, (middle) averaged sediment organic $\delta^{13}\text{C}_{\text{org}}$ values, and (bottom) averaged sediment $\delta^{15}\text{N}$ values from Royer et al. [2001], Hayes et al. [1999], and Algeo et al. [2014], respectively. Periods of cooler climate are indicated as shaded intervals within the prevailing warm climate of the past 550 Myr.

compositions coincident with the initiations of the two major Cretaceous OAEs. In the OAE 1a sequences at DSDP Site 463 in the Mid-Pacific Mountains and at Cison, Italy, concentrations of Os increase abruptly just before or at the onset of black shale deposition, and their $^{187}\text{Os}/^{188}\text{Os}$ ratios decrease abruptly to strongly magmatic values [Bottini et al., 2012]. In the OAE 2 sequences at ODP Site 1260 on the Demerara Rise and at Furlò, Italy, the concentrations of Os similarly increase and the Os isotopic ratios decrease abruptly at the start of black shale deposition [Turgeon and Creaser, 2008]. These changes imply that the emplacement of the Ontong-Java Plateau triggered the deposition of the OAE 1a black shales and that the formation of the Caribbean/Columbian Plateau and the Madagascar Rise similarly acted as the trigger for the deposition of the OAE 2 black shales.

Although virtually no seafloor older than Late Jurassic has survived subduction to provide evidence of oceanic magmatic events before the Cretaceous, continental preservation of a number of LIPs document that these phenomena occurred from time to time on land and could have contributed to the negative $\delta^{13}\text{C}$ values of pre-Cretaceous black shales. A prime example is the Karoo-Ferrar LIP that was emplaced on Gondwana about 183–180 Ma [Jourdan et al., 2008] and has since been rifted apart to exist largely on Antarctica

[2002]. Of particular significance to the deposition of mid-Cretaceous black shales, rates of seafloor spreading were 50%–100% higher from 125 to 80 Ma than before or after, with the consequence that production of oceanic crust greatly increased [Larson, 1991]. In addition, two major submarine LIPs, the Ontong-Java and Manihiki plateaus in the Pacific Ocean, were formed at about the Barremian-Aptian boundary (125 Ma). The Ontong-Java Plateau remains the largest existing oceanic plateau by a factor of at least two [Larson and Erba, 1999]. Two other major submarine LIPs, the Caribbean/Columbian Plateau and the Madagascar Rise, were formed at about the time of the Cenomanian-Turonian boundary (~94 Ma) and were accompanied by accumulation of additional magma on the Ontong-Java Plateau [Kerr, 1998]. The increased formation of new seafloor and the multiple emplacements of large oceanic plateaus were all likely accompanied by massive releases of isotopically light CO_2 from the mantle-derived basalt, which would have also contributed to elevating atmospheric $p\text{CO}_2$ (Figure 4) and thereby helped to maintain the greenhouse climate of the mid-Cretaceous. Particularly compelling evidence for a connection between the massive release of magmatic CO_2 and formation of OAEs exists in changes in osmium isotopic

and South Africa. Deposition of the Toarcian OAE coincides with the formation of this LIP. Of particular significance, wood fragments isolated from T-OAE sections in England and Portugal have $\delta^{13}\text{C}$ values about 5‰ more negative than wood fragments found either before or after the black shale event (Table 4). Because the trees that synthesized the wood at these two locations obtained their CO_2 from air, this excursion very probably documents a change in the isotopic composition of the atmosphere. Although direct injection of isotopically light CO_2 into the atmosphere from the magma likely occurred, *McElwain et al.* [2005] and *Svensen et al.* [2007] postulate that a greater part of the injected gas was derived from thermal decomposition of the organic-carbon-rich coals and shales in the Karoo Basin that were intruded by the magma. This gas, probably a mixture of CO_2 and CH_4 , would have been isotopically very light ($\sim -30\text{‰}$) and could well have been responsible for the excursion evident in the wood fragments. This excursion to lighter isotopic compositions in the wood is paralleled by an identical excursion in the marine organic matter of the Whitby Mudstone (Figure 3). Because the bulk organic matter $\delta^{13}\text{C}$ values before the T-OAE are $\sim -26\text{‰}$ and $\sim -27\text{‰}$ after it and hence considerably more negative than modern marine organic matter (-20‰), the isotopic composition of the dissolved inorganic carbon available to marine photoautotrophs must have been significantly lighter than in the modern ocean before and after the injection of light carbon associated with the Karoo-Ferrar LIP. Unlike the marine carbon isotopic composition, the $\delta^{13}\text{C}$ values of wood fragments from before and after the T-OAE are close to those of modern wood (Table 4). This contrast in the long-term impacts of the Karoo-Ferrar magmatism on marine and continental organic matter suggests that the ocean and atmosphere carbon cycles functioned differently.

Another continental LIP whose emplacement coincides with a negative carbon isotope excursion and deposition of black shales is the Central Atlantic magmatic province (CAMP) that formed during the breakup of Pangaea in the latest Triassic (~ 201 Ma). The CAMP is the most extensive of any LIP and exists as thick layers of basalt on four continents. Despite its great size, its emplacement occurred over the relatively short-time span of 600 kyr [*Blackburn et al.*, 2013; *Hesselbo et al.*, 2002]. This magmatic eruption has been implicated in increasing atmospheric $p\text{CO}_2$ and in causing a negative shift of about 4‰ in marine and land-derived organic matter [*Bachan et al.*, 2012] that *Hesselbo et al.* [2002] estimate persisted for perhaps 600 kyr.

As yet another likely continental example of magmatic injection of CO_2 associated with black shale deposition, the extensive end-Permian Siberian Traps have CO_2 $\delta^{13}\text{C}$ values of about -9‰ [*Sobolev et al.*, 2011], suggesting that this LIP could well have contributed to the global 4–7‰ negative excursion that is expressed in organic and inorganic carbon at the Permian-Triassic boundary [*Korte and Kozur*, 2010; *Shen et al.*, 2012] and that yielded organic matter $\delta^{13}\text{C}$ values ranging from -31 to -27‰ [*Grice et al.*, 2005; *Riccardi et al.*, 2007; *Schneebell-Hermann et al.*, 2013]. Atmospheric $p\text{CO}_2$ also peaked at this time (Figure 4), suggesting that the amount of gas that was released was large. However, *Korte et al.* [2010] conclude that the volume of isotopically light CO_2 released from the Siberian Traps was not adequate by itself to have been responsible for the $\delta^{13}\text{C}$ excursion. Instead, as similarly postulated for the Toarcian OAE [*Svensen et al.*, 2007], thermal degradation of locally extensive coal measures and other organic matter-rich sequences by the magma may have been the major source of the isotopically light CO_2 [*Svensen et al.*, 2009; *Korte et al.*, 2010]. In addition, large-scale seafloor magma production could have been another important source of CO_2 , but any evidence that existed for a seafloor contribution has been lost to subduction. Regardless of whether a continental or marine source contributed the isotopically light CO_2 , respective shifts of 5.5 and 2.7‰ to more negative $\delta^{13}\text{C}$ values in land plant cuticles and wood fragments from the Permian-Triassic boundary at Amb, Pakistan, document a change in the carbon isotopic composition of the atmosphere that is estimated to have lasted 200 kyr [*Schneebell-Hermann et al.*, 2013]. Bulk organic matter in this shallow marine paleosetting exhibits a shift from -24‰ to -30‰ across this boundary, and its $\delta^{13}\text{C}$ values remain at their lower levels above the boundary, unlike the land plant values [*Schneebell-Hermann et al.*, 2013]. Similarly, bulk organic matter $\delta^{13}\text{C}$ values in the deep water Xinman section of China shift from -24‰ in the upper Permian to -29‰ in the early Triassic [*Shen et al.*, 2012]. The difference in the continental and marine isotopic records again suggests that the long-term impacts of the magmatic CO_2 on the isotopic compositions of the atmosphere and the oceans were different. In addition, a marked long-term increase in atmospheric $p\text{CO}_2$ coincides with the Permian-Triassic boundary (Figure 4), which may indicate that more than magmatic gas evolution was involved with the difference in the continental and marine organic matter patterns.

Evidence of earlier extensive magmatism that may have been concurrent with the Late Devonian and Ordovician episodes of black shale deposition is lacking. Nonetheless, continental LIPs that may have been

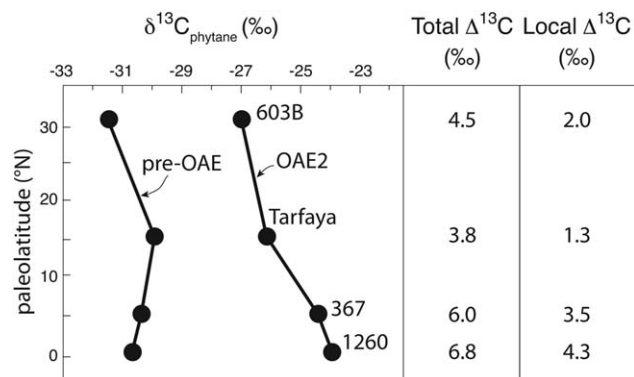


Figure 5. Latitude-related differences in excursions to less negative $\delta^{13}\text{C}$ values of four Atlantic Ocean OAE 2 sequences as preserved in sulfur-bound phytane. Data from *Kuypers et al.* [2002a], *Tsikos et al.* [2004], and *van Bentum et al.* [2012]. Both total excursions and excursions adjusted by 2.5‰ for drawdown of globally elevated $p\text{CO}_2$ by enhanced productivity and that reflect local effects are shown.

conditions active in producing the ^{13}C -depleted organic matter were longer-lived than the magmatic events. It is helpful to remember that the oceanic carbon reservoir is 65 times the size of the atmospheric carbon reservoir (39,000 Gt versus 600 Gt) [Killops and Killops, 2005] and that the carbon cycles of atmosphere and oceans function differently. Modern marine carbon isotopic compositions are controlled principally by production, sedimentation, and recycling of ^{13}C -depleted organic matter and are further influenced by multiple other factors, including atmospheric $p\text{CO}_2$, sea level fluctuations, and oceanic circulation [e.g., Hayes et al., 1999; Kump and Arthur, 1999; Galimov, 2006]. These modern processes have also likely been the major influences on isotopic compositions in the oceans and, indirectly, the atmosphere through much of geological time, despite the evidence of their perturbations from magmatic injections of CO_2 .

4.4. Enhanced Discrimination Against ^{13}C During Times of Greater Atmospheric $p\text{CO}_2$

This hypothesis is based on the idea that formerly higher $p\text{CO}_2$ levels permitted greater discrimination against ^{13}C during photosynthetic assimilation. As evident in Figure 4, episodes of black shale deposition generally have occurred during times of elevated $p\text{CO}_2$. Also evident is the rough correspondence between greater isotopic discrimination against ^{13}C and many of the black shales, particularly those earlier than the Cretaceous. This correspondence has been interpreted as indicating that modern marine photosynthetic fractionation of about -20‰ (Figure 4) is not typical of the greater fractionation of -25 to -28‰ (Figure 5) that existed through most of the Phanerozoic [Arthur et al., 1985; Hayes et al., 1999; Popp et al., 1989]. However, this interpretation is based largely on the $\delta^{13}\text{C}_{\text{org}}$ values that can be obtained from Phanerozoic rocks that contain enough organic matter to allow these determinations. These rocks are usually black shales, which are noted for their atypically light carbon isotopic compositions relative to modern marine organic matter.

Although the black-shale-based argument for greater isotopic fractionation in the past might seem circular, field studies and laboratory experiments verify that isotopic fractionation increases as concentrations of CO_2 increase. As noted in section 2, algal organic matter in the frigid waters surrounding Antarctica can be as negative as -28‰ because the cold waters can accommodate more dissolved CO_2 than warmer waters [Rau et al., 1989]. Freeman and Hayes [1992] compiled the results of this and other studies to show that the isotopic fractionation systematically decreases from -30‰ in water temperatures of -2°C to -20‰ at 30°C . At the time most of these studies were done, $p\text{CO}_2$ was about 325 ppm, and the calculated equilibrium solubility of CO_2 in sea water decreased from 23 to 9 μmol as temperature increased. The nearly three-fold higher concentration of dissolved CO_2 in the cold high-latitude waters evidently allows for greater discrimination against ^{13}C during photosynthetic uptake, although latitudinal differences in algal assemblages and productivity rates could be additional influences on the $\delta^{13}\text{C}$ values [e.g., Hayes et al., 1989].

During the period of Earth history when Cretaceous black shales were deposited, $p\text{CO}_2$ was 1000–2000 ppm (Figure 4) and a warm greenhouse climate existed. Cretaceous sea surface temperatures have been estimated to have been about 35°C in the equatorial Atlantic Ocean [Forster et al., 2007] and the marine equator to pole temperature gradient was smaller than today [Huber et al., 1995]. The warmer temperatures

emplaced during the Late Devonian or Ordovician could well have been eroded or buried and their evidence erased or hidden by the passage of time, and any submarine LIPs would have been destroyed by subduction. The concordance of magmatic events with the major OAEs of the Cretaceous, Jurassic, latest, Triassic, and the Permian-Triassic boundary makes the magmatic release of isotopically light CO_2 a good explanation for the relatively short-lived negative $\delta^{13}\text{C}$ excursions that exist in land plant organic matter associated with the OAEs. However, the presence of isotopically light compositions of marine organic matter before and after these excursions indicates that the marine

would have diminished solubility of CO₂ in sea water, but the higher atmospheric concentrations would have more than balanced the temperature effect. The result would have been higher concentrations of dissolved CO₂ and greater isotopic discrimination against ¹³C during algal uptake of dissolved CO₂, which is found in both the organic and inorganic carbon fractions of Cretaceous and earlier black shales [Kump and Arthur, 1999] and also in their marine biomarkers (Table 2). Like those of the Cretaceous, the earlier black shales are also from times of greenhouse climate, as implied by the high *p*CO₂ values that accompanied their deposition (Figure 4).

The isotopic compositions of land plants do not display the same sensitivity to CO₂ concentrations as found in marine algae [e.g., Popp *et al.*, 1989; Sinninghe-Damsté *et al.*, 2008]. Two factors are primarily responsible for the different sensitivities. The principal one is that algae take up CO₂ by diffusion across their cell walls, which is a concentration-dependent process, whereas land plants take up CO₂ through their stomatal apertures, which they can open or close depending on environmental conditions. The second factor is that environmental conditions undoubtedly were different during times of greenhouse climates than during those of “icehouse” climates. Greater evaporation rates likely accompanied the warmer temperatures of the greenhouse climates, which would have led land plants to constrict their stomata, restrict their intake of the more available CO₂, and dampen the potential isotopic effects of the higher atmospheric *p*CO₂ values. Additional factors, such as the different carbon assimilation pathways that exist in land plants, can complicate their responses to *p*CO₂ changes. For example, Sinninghe-Damsté *et al.* [2008] postulate that the shifts to less negative $\delta^{13}\text{C}$ values that they find in land plant wax components in the OAE 2 sequence from DSDP Site 367 in the Atlantic (Table 4) may record the earliest evidence for dry-climate C₄ plants on Africa.

The importance of water availability to isotopic fraction in land plants is demonstrated by experiments conducted by Schubert and Jahren [2012], who found that plants grown in chambers in which their water-use efficiencies are held constant increase their discrimination against ¹³C as *p*CO₂ increases. These conditions are rarely found under natural conditions, and therefore the excursions to more negative $\delta^{13}\text{C}$ values that are found in wood fragments from Toarcian black shales (Figure 3 and Table 4) and in plant cuticles from a Permian-Triassic black shale sequence [Schneebell-Hermann *et al.*, 2013] imply that a change to lighter isotopic compositions of atmospheric CO₂ accompanied the increases in *p*CO₂ values that occurred at these times (Figure 4). Massive releases of isotopically light CO₂ from the LIPs that were emplaced at those times are good candidates for these changes in atmospheric compositions. Because isotopically light marine organic matter was deposited both before and after the isotopic excursions found in the Toarcian, Rhaetian, and the Permian-Triassic land plant matter (Figure 3 and Tables 1 and 2), the changes in isotopic composition were probably superimposed on existing high concentrations of CO₂ in the ocean and atmosphere.

4.5. Oxidation and Recycling of ¹³C-Depleted Carbon in the Surface Ocean

The principle of this hypothesis is that increased density stratification of the ocean surface slowed sinking of organic matter, which allowed more of this material to be oxidized and more of its isotopically light carbon to be recycled in the near-surface ocean. Because this hypothesis was first proposed by Küspert [1982], it is commonly called the “Küspert Hypothesis.” The major episodes of black shale deposition coincided with periods of elevated atmospheric *p*CO₂ (Figure 4), which were times of warm greenhouse climate as evidenced by the absence of significant continental glaciation, expansion of warm-weather biota to high latitudes, and warmer temperatures in the surface and the deep ocean [e.g., Huber *et al.*, 2002; Bice *et al.*, 2006; Forster *et al.*, 2007]. The warmer conditions increased evaporation of water from the sea surface and created a globally wetter climate. As one example of the consequences of a wetter climate, Cohen *et al.* [2004] report an excursion to higher ¹⁸⁷Os/¹⁸⁸Os values that accompanied the excursion to more negative $\delta^{13}\text{C}$ values in the Yorkshire T-OAE sequence (Table 1) that they interpret to signify a four to eightfold increase in continental weathering rates during the estimated 150 kyr duration of this OAE [Hesselbo *et al.*, 2000]. Somewhat similar threefold increases in weathering rates for the mid-Cretaceous OAEs 1a and 2 have been estimated by Blättler *et al.* [2011] from decreases in $\delta^{44/42}\text{Ca}$ values at multiple locations. The combination of increased continental runoff associated with the greater weathering and the overall warmer greenhouse conditions evidently increased thermohaline stratification, particularly in epicontinental seas and in the parts of the oceans not far from continents [e.g., Erbacher *et al.*, 2001].

Although the Küspert hypothesis provides a rational explanation for the light isotopic compositions of black shales, the surface stratification on which it is founded would seemingly present an impossible impediment

Table 5. Averaged Carbon and Nitrogen Isotopic Compositions of Marine Organic Matter in Phanerozoic Black Shales and Near-Modern Analogs

Formation	Age (Ma)	TOC (%)	$\delta^{13}\text{C}$ (‰)	$\delta^{15}\text{N}$ (‰)	Reference
Xiaotan Section, China	Early Cambrian (~525)	1–4	–35	+0.5	Cremonese <i>et al.</i> [2013]
Vinini Creek, NV	Late Ordovician (~445)	2–25	–31.5	–0.5	LaPorte <i>et al.</i> [2009]
Woodford Shale, OK	Late Devonian (~375)	4–8	–28.7	+2.9	Quan <i>et al.</i> [2013]
New Albany Shale, IN	Late Devonian (~375)	10–16	–28.5	+0.2	de la Rue <i>et al.</i> [2007]
Hanover Shale, NY	Late Devonian (~375)	3–5	–28.4	–0.1	Tuite and Macko [2013]
Whitby Mudstone, UK	Early Jurassic (~181)	12–16	–30.9	+2.3	Saalen <i>et al.</i> [2000]
Kimmeridge Clay, UK	Late Jurassic (~155)	28–30	–21.7	+1.9	Saalen <i>et al.</i> [2000]
OAE 1a, Shatsky Rise, Pacific	Aptian (124)	2–40	–26.4	–2.9	Dumitrescu and Brassell [2006a]
OAE 1b, ODP Site 1049, Atlantic	Albian (111)	1–6	–20	–3	Kuypers <i>et al.</i> [2002b]
OAE 1c, ODP Site 1276, Atlantic	Albian (102)	1–2	–26	–2	Arnaboldi and Meyers [2006a]
OAE 1d, ODP Site 1276, Atlantic	Albian (100)	1–2	–26	+0.7	Arnaboldi and Meyers [2006a]
MCE ^a , ODP Site 1276, Atlantic	Cenomanian (95)	1–3	–26.6	–1.1	Arnaboldi and Meyers [2006a]
OAE 2, Demerara Rise, Atlantic	Cenomanian (93.6)	4–22	–28.5	–1.9	Meyers <i>et al.</i> [2009]
Monterey Formation, California	Miocene (~12)	5–16	–22.5	+5.8	Schimmelmann <i>et al.</i> [2001]
Mediterranean sapropels, Site 967	Pleistocene (~1)	2–4	–23.4	–1.5	Meyers and Arnaboldi [2008]
Benguela upwelling, Atlantic	Pleistocene (~1)	5–17	–20	+5	Robinson <i>et al.</i> [2002]
Typical marine sediment	Modern	0.1	–20	+5	Altabet and Francois [1994]

^aMid-Cenomanian event.

to the vertical mixing of nutrients that is considered essential to sustaining the high marine productivity implied by the high concentrations of organic carbon in black shales (Table 1). However, the increased continental runoff during the times of greenhouse-amplified hydrologic cycle would enhance delivery of land-derived nutrients to the ocean margins and thereby help to sustain marine productivity [e.g., van Helmond *et al.*, 2014]. Furthermore, if stratification were strong enough and organic matter oxidation were great enough, dissolved oxygen in areas of the near-surface ocean would become depleted and anoxic conditions would become established. Under these conditions, phosphorus regeneration [e.g., Mort *et al.*, 2007; Krall *et al.*, 2010] and nitrogen fixation [e.g., Meyers *et al.*, 2009] would occur, and nutrients would remain available for sustained production of organic matter. The common occurrence of derivatives of isorenieratene, a photosynthetic pigment distinctive of anaerobic green sulfur bacteria, in black shale sequences provides compelling evidence that anoxia existed within the photic zone during episodes of their formation [e.g., Sinninghe-Damsté and Köster, 1998; Kuypers *et al.*, 2002b]. The presence of the green sulfur bacteria mandates the reduction of sea water sulfate to sulfide, which these organisms require and implies the complete denitrification of nitrate, which encourages bacterial nitrogen fixation. Similar to the molecular evidence for the past presence of green sulfur bacteria, nitrogen fixation leaves a distinctive isotopic signature on sedimentary organic matter.

The $\delta^{15}\text{N}$ values of black shales compiled in Table 5 are consistently lower than the +5‰ that is typical of modern marine organic matter [Altabet and Francois, 1994] and that are found in Pleistocene sediments deposited under the Benguela Current Upwelling System [Robinson *et al.*, 2002] and in the Monterey Formation [Schimmelmann *et al.*, 2001], which was deposited under a Miocene upwelling system. They are instead generally within or close to the range of –4 to +1‰ of organic matter produced by cyanobacteria [Minagawa and Wada, 1986]. They are also similar to the average value of –1.5‰ of Mediterranean sapropels, which are often considered near-modern analogs of the black shales [e.g., Erbacher *et al.*, 2001; Meyers, 2006; Negri *et al.*, 2006; Emeis and Weissert, 2009]. The sapropels are generally believed to have been formed during precessional minima that created periods of several millennia of wetter climate in the Mediterranean region that led to dilution the surface of this sea and creation of strong surface stratification. Like the black shales, sapropels contain molecular evidence of green sulfur bacteria [e.g., Menzel *et al.*, 2002] that combined with the isotopic evidence of nitrogen fixation indicate that photic zone anoxia was important to their formation. Moreover, their relatively high TOC concentrations (1–10%) imply heightened marine productivity and improved organic matter preservation, much as found in Phanerozoic black shales. For both the sapropels and the black shales, the combination of enhanced delivery of continental nutrients during times of wetter climates, water column phosphorus regeneration, and photic zone nitrogen fixation evidently was sufficient to sustain high rates of marine productivity in the stratified oceans that existed during their formation.

Although sharing many similarities, the Mediterranean sapropels and Phanerozoic black shales also have some important differences. Foremost among these is that the $\delta^{13}\text{C}$ values of the sapropels are not notably more negative than those of modern organic matter (Table 5) and do not display excursions to more negative

values, as in the T-OAE, or to less negative values, as in the OAE 2. A second major difference is that the duration of sapropel deposition was generally only a few thousand years, in contrast to the ~ 500 kyr of the black shales. The smaller temporal scale of the sapropels may be part of the reason why they do not share the curiously light isotopic compositions of the black shales. Although marine biomarker $\delta^{13}\text{C}$ values suggest that a slightly negative inorganic carbon shift occurred in the upper water column [Menzel *et al.*, 2003], the surface stratification that led to their deposition was probably not long enough for the isotopic composition of inorganic carbon in the photic zone to have been significantly altered by near-surface recycling of organic matter. Another factor in the isotopic differences is that atmospheric $p\text{CO}_2$ during the Pliocene-Pleistocene deposition of the sapropels was not much different from that of today (Figure 4) and therefore did not allow for the greater isotopic discrimination that was possible during the Phanerozoic periods of greenhouse climate when black shales were deposited. Finally, deposition of the sapropels was a local response to precessional minima, a global forcing, that was regionally amplified by the particular sensitivity of Mediterranean circulation to periods of wetter climate and was not a global phenomenon like many of the OAEs.

A paleodepositional consideration that is inherent in the Küspert hypothesis is that the amount of continental runoff would be different at different locations, which would result in regional differences in delivery of land-derived nutrients and in surface stratification. These differences would likely be expressed as same-age black shale sequences that record different conditions of organic matter production and preservation. Evidence for such place-to-place variations in productivity and preservation is found in Mediterranean sapropels [Gallego-Torres *et al.*, 2011], and evidence for similar regional differences in black shale expression exists. Andrusevich *et al.* [2000] note that the carbon isotopic compositions of Upper Jurassic crude oils differ with latitude; low-latitude $\delta^{13}\text{C}$ values are $\sim -27\text{‰}$ whereas high-latitude values are $\sim -32\text{‰}$, but a few oils deviate from this pattern. The source rocks for these oils are Jurassic black shales. Although the authors interpret these results as reflecting differences in sea surface temperatures where the black shales were deposited, place-to-place differences in productivity could also be important.

A particularly apt example of regional differences in expression of same-age black shales is provided by the comparison by van Bentum *et al.* [2012] of latitudinal differences in the $\delta^{13}\text{C}$ values of the marine biomarker phytane in OAE 2 sequences from four locations in the North Atlantic Basin (Figure 5). They interpret differences in the carbon isotopic compositions to indicate that pre-OAE 2 paleoproductivity was higher at Tarfaya (Morocco) and DSDP Site 367 (Senegal) than at DSDP Site 603 (Hatteras Rise) and ODP Site 1260 (Demerara Rise), but it increased at all locations during the black shale event. Assuming a 4‰ offset between phytane and bulk marine organic matter [Schouten *et al.*, 1998], the phytane $\delta^{13}\text{C}$ values in Figure 5 indicate that the isotopic compositions of total organic matter were $4\text{--}5\text{‰}$ lighter than modern organic matter at the four locations. These values became less negative during OAE 2 as consequences of the magnified productivity associated with this event, but the extents of the isotopic excursions differed at the four locations. The differences in the excursions are interpreted by van Bentum *et al.* [2012] to reflect local differences in sea surface temperature, stratification, and delivery of land-derived nutrients that affected productivity. Moreover, they argue persuasively that magmatic activity, atmospheric $p\text{CO}_2$, sea surface stratification, and organic matter production and burial interacted during OAE 2 to create positive and negative feedbacks on the global carbon cycle. These interactions likely were important during other episodes of deposition of Phanerozoic black shales.

5. Summary and Conclusions

The five major hypotheses that have been proposed to explain why the marine organic matter in Phanerozoic black shales is generally about 7‰ lighter than modern marine organic matter are discussed and evaluated in light of the results of advances in organic geochemical isotopic analyses. Two of the hypotheses—enrichment of ^{13}C -depleted components in the residual organic matter by selective diagenetic removal of ^{12}C -depleted components and massive release of ^{13}C -depleted carbon from extensive dissociation of methane hydrate deposits in the sea floor—are rejected, largely on mass balance considerations.

Because the episodes of black shale deposition occurred during Phanerozoic periods of elevated $p\text{CO}_2$, the hypothesis that their light isotopic compositions are a result of enhanced discrimination against ^{12}C during these times of more abundant dissolved CO_2 seems valid. Moreover, these periods were times of warmer and wetter greenhouse climates that increased density stratification of the surface ocean, especially in

epicontinental and marginal seas that received more continental runoff. This situation of greater stratification favors the hypothesis that the light isotope compositions reflect greater near-surface oxidation of sinking organic matter and reutilization of its ^{12}C -depleted carbon, with productivity being enhanced by greater delivery of land-derived nutrients (the K uspert hypothesis). The presence of marine organic matter that is isotopically light in TOC-lean sediments deposited before and after most of the black shale events and of land-derived organic matter that is not isotopically light suggests that the high $p\text{CO}_2$ levels of the greenhouse climates that existed at these time somehow affected the isotopic compositions of only marine photoautotrophs. A reasonable possibility is that the oceans remained strongly surface stratified throughout the periods of elevated $p\text{CO}_2$ and that their photic zone inorganic carbon reservoirs remained isotopically light because of organic matter recycling.

The fifth and remaining hypothesis that massive release of isotopically light magmatic carbon was responsible for the low $\delta^{13}\text{C}$ values also appears to have had a role in affecting isotopic compositions. Emplacement of the Karoo-Ferrar LIP seems to have been an agent responsible for 3‰ shifts to lighter isotope values in both marine and land-derived organic matter in Toarcian OAEs. Emplacements of other LIPs are implicated in end-Permian and mid-Cretaceous black shale events. Although their impacts on carbon isotopic compositions are not evident, the gases released by these LIPs likely enhanced greenhouse warming. However, the effects of the magmatic releases of CO_2 appear to have been relatively short-lived and cannot have been responsible for the long periods of deposition of isotopically light marine organic matter that occurred during the times of greenhouse climate. Instead, the magmatic releases seem to have served as the triggers that initiated deposition of the black shales.

In conclusion, the isotopically light compositions that typify the marine organic matter of Phanerozoic black shales appear to result from a combination of high atmospheric $p\text{CO}_2$, strongly stratified surface ocean, and magmatic releases of CO_2 . The high $p\text{CO}_2$ levels encouraged greater discrimination against ^{12}C during photosynthetic assimilation of carbon and sustained long periods of warm and wet greenhouse climate. The greenhouse climate was accompanied by stronger density stratification of the surface ocean because of higher sea surface temperatures and more dilution from continental runoff. Sinking of organic matter from the photic zone was slowed by the strong stratification, and a greater proportion of the organic matter was recycled and its isotopically light carbon was reassimilated, producing new organic matter with more negative $\delta^{13}\text{C}$ values. Magmatic releases of isotopically light mantle CO_2 augmented the greenhouse climate conditions and also may have triggered the episodes of black shale deposition.

Acknowledgments

This overview is dedicated in memory of J. S. L. Casford, who was a principal in founding the Development of Isotopic Proxies for Paleoenvironmental Interpretation: A Carbon Perspective (DIPPI-C) initiative and whose insights into the deposition of Mediterranean sapropels were illuminating to me. I thank Alex Dickson, Christoph Korte, and Hayden Mort for stimulating discussions about black shales and their deposition. I am especially grateful to Anthony Cohen and Paul Wignall for their very helpful and constructive comments that helped to improve this contribution. This overview is an outgrowth of presentations given at a special session on *Modern Insights into the Paleo-Carbon Cycle: $\delta^{13}\text{C}$ and Biomarker Perspectives* at the 2011 AGU Annual Meeting in San Francisco, CA, and at a DIPPI-C workshop in Durham, UK, May 2012. I thank the organizers of these meetings for inviting my participation and for encouraging me to construct this overview.

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