

Large-eddy simulation of atmospheric chemistry during the DISCOVER-AQ 2011 campaign

Yang Li¹, Mary C. Barth², Gao Chen³, Edward G. Patton², Si-Wan Kim^{4,5}, Armin Wisthaler^{6,8},
Tomas Mikoviny^{7,8}, Alan Fried⁹, Richard Clark¹⁰, Allison L. Steiner¹

¹Climate and Space Sciences and Engineering, University of Michigan, Ann Arbor, Michigan, USA, ²National Center for Atmospheric Research, Boulder, Colorado, USA, ³NASA Langley Research Center, Hampton, Virginia, USA, ⁴Chemical Sciences Division, NOAA Earth System Research Laboratory, Boulder, Colorado, USA, ⁵Cooperative Institute for Research in Environmental Sciences, University of Colorado, Boulder, Colorado, USA, ⁶Institute of Ion Physics and Applied Physics, University of Innsbruck, Austria, ⁷Oak Ridge Associated Universities (ORAU), Oak Ridge, Tennessee, USA, ⁸Department of Chemistry, University of Oslo, Norway, ⁹Institute of Arctic and Alpine Research, University of Colorado, Boulder, Colorado, USA, ¹⁰Department of Earth Sciences, Millersville University of Pennsylvania, Millersville, Pennsylvania, USA

Contents of this file

Figures S1 to S3

Tables S1 to S3

Introduction

This supporting information provides figures and tables that introduce meteorological (Figure S1, Table S1) and chemical conditions (Table S2) used in the simulations. Vertical cross sections of chemical species and vertical velocity are shown to better understand segregation of isoprene and OH (Figure S2). The bar chart (Figure S3) and the photolysis lifetime table (Table S3) are used as a supplement to show the reactivity of the BVOC species.

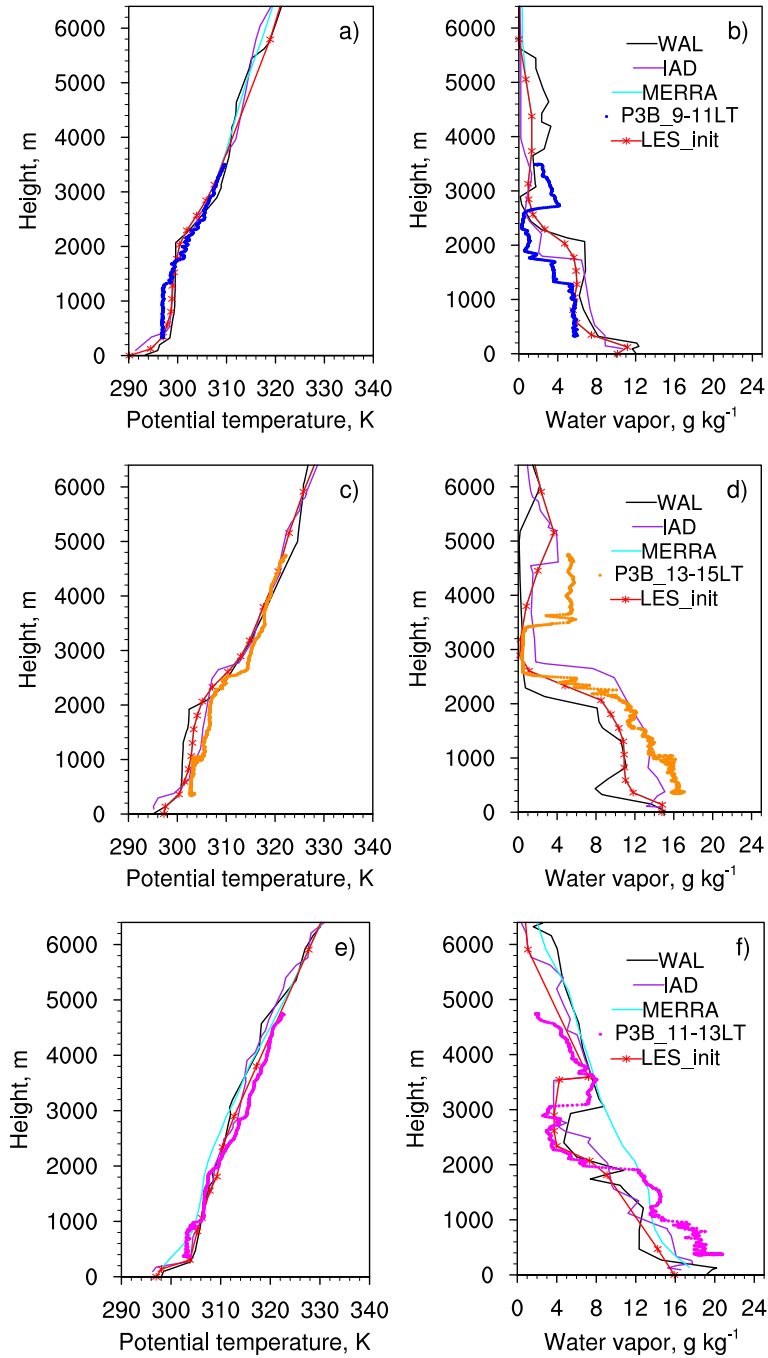


Figure S1. Comparison of potential temperature (K) (a, c, e) and water vapor mixing ratio (g kg^{-1}) (b, d, f) of the initial profiles at 0530 LT used in LES (red), interpolated 0530 LT profiles derived from 6-hourly finer resolution ($0.5^\circ \times 0.67^\circ$) MERRA data (cyan), sounding data at 0800 LT from WAL (black) and IAD (purple) sites, and the earliest P-3B observed profiles (0900-1100 LT for Case 1, 1300-1500 LT for Case 2 and 1100-1300 LT for Case 3) for the three cases (Case 1: a, b; Case 2: c, d; Case 3: e, f).

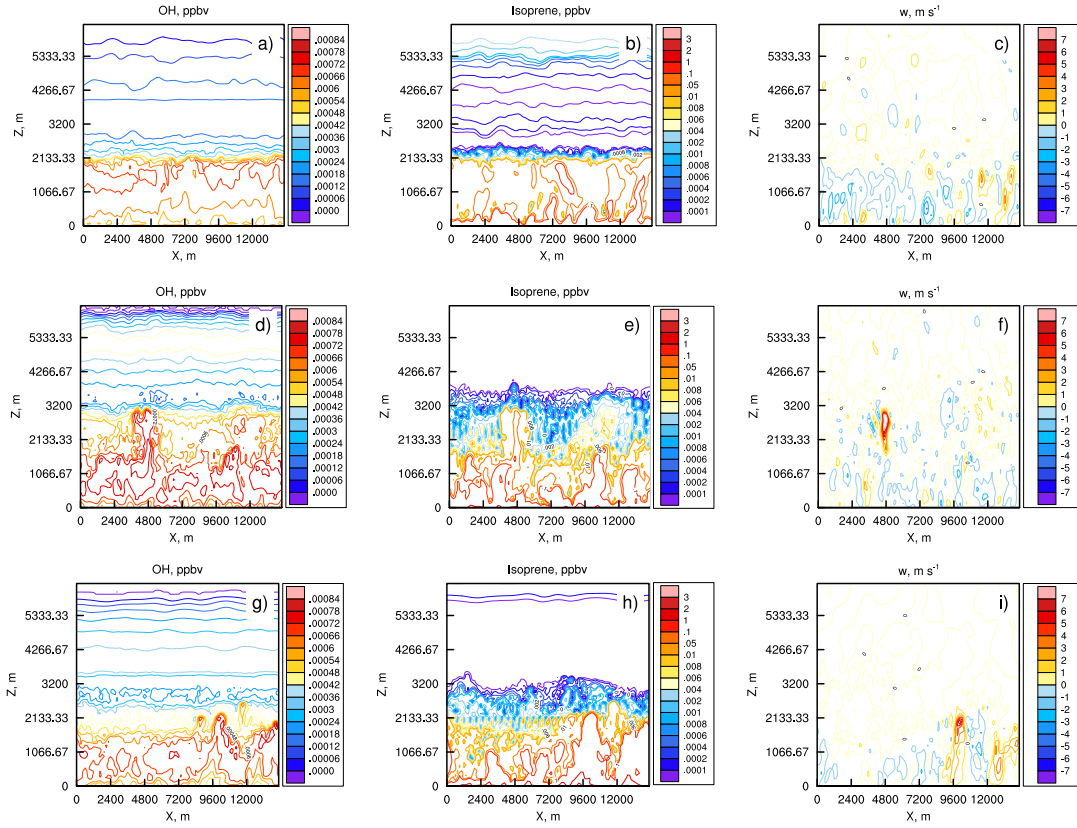


Figure S2. LES simulated instantaneous vertical cross section of OH (a, d, g), isoprene (b, e, h) and vertical velocity w (c, f, i) in the middle of the domain at 1400 LT for the three cases (Case 1: a-c; Case 2: d-f; Case 3: g-i).

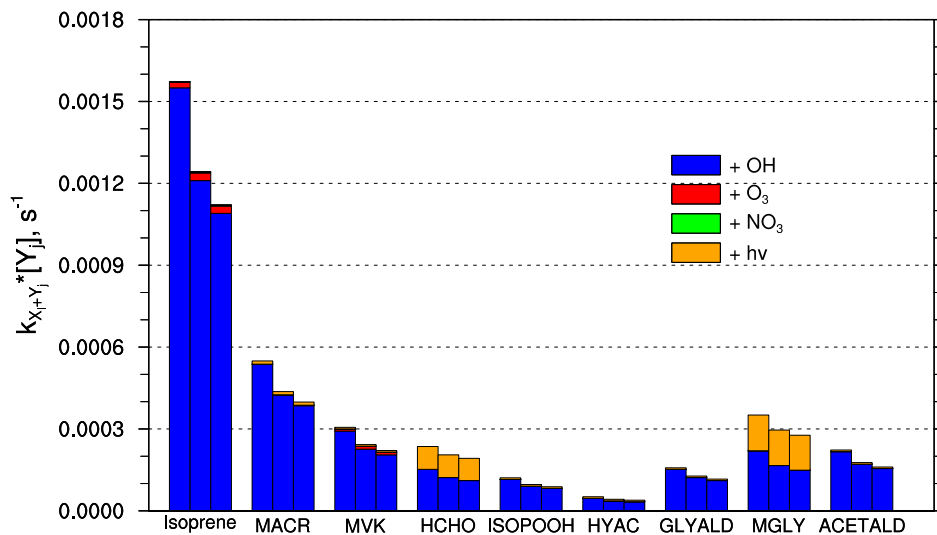


Figure S3. Photochemical box model simulated $k_{X_i+Y_j} * [Y_j]$ (s^{-1}) during 1100-1300 LT (with three vertical bars for each BVOC species representing the values for the three cases), where $k_{X_i+Y_j}$ is the reaction rate coefficient for each reaction of BVOC species X_i with Y_j and $[Y_j]$ is the ambient concentration of OH, O₃, or NO₃, $k_{X_i+Y_j}$ is the photolysis rate and $[Y_j]$ is 1 for the photolysis pathway.

| Case 1 | | Potential temperature tendency (K h ⁻¹) | | | Water vapor tendency (g kg ⁻¹ h ⁻¹) | | | Radiative tendency (K h ⁻¹) | | |
|--------|---------|---|--------------|--------|--|--------------|--------|---|--------------|--------|
| Time | | 0-1 km | 1-2 km | > 2 km | 0-1 km | 1-2 km | > 2 km | 0-1 km | 1-2 km | > 2 km |
| LT, hr | UTC, hr | | | | | | | | | |
| 3.5 | 7.5 | -0.190 | 0.005 | -0.143 | -0.124 | -0.236 | -0.129 | -0.193 | -0.091 | -0.078 |
| 6.5 | 10.5 | -0.164 | -0.016 | 0.014 | -0.362 | -0.222 | -0.124 | -0.112 | -0.071 | -0.062 |
| 9.5 | 13.5 | -0.125 | -0.019 | 0.077 | -0.273 | -0.325 | -0.038 | 0.051 | -0.008 | -0.018 |
| 12.5 | 16.5 | -0.139 | -0.043 | 0.117 | -0.046 | -0.233 | 0.053 | 0.097 | 0.030 | 0.000 |
| 15.5 | 19.5 | -0.154 | -0.097 | 0.206 | 0.066 | -0.056 | 0.044 | 0.055 | 0.030 | -0.002 |
| 18.5 | 22.5 | -0.068 | -0.031 | 0.125 | 0.086 | -0.008 | 0.034 | -0.061 | -0.025 | -0.032 |
| Case 2 | | Potential temperature tendency (K h ⁻¹) | | | Water vapor tendency (g kg ⁻¹ h ⁻¹) | | | Radiative tendency (K h ⁻¹) | | |
| Time | | 0-1 km | 1-2 km | > 2 km | 0-1 km | 1-2 km | > 2 km | 0-1 km | 1-2 km | > 2 km |
| LT, hr | UTC, hr | | | | | | | | | |
| 3.5 | 7.5 | 0.142 | Interpolated | 0.000 | 0.184 | Interpolated | 0.000 | -0.172 | Interpolated | 0.000 |
| 6.5 | 10.5 | 0.125 | Interpolated | 0.000 | 0.231 | Interpolated | 0.000 | -0.117 | Interpolated | 0.000 |
| 9.5 | 13.5 | 0.005 | Interpolated | 0.000 | 0.204 | Interpolated | 0.000 | 0.038 | Interpolated | 0.000 |
| 12.5 | 16.5 | -0.003 | Interpolated | 0.000 | 0.156 | Interpolated | 0.000 | 0.086 | Interpolated | 0.000 |
| 15.5 | 19.5 | -0.051 | Interpolated | 0.000 | 0.154 | Interpolated | 0.000 | 0.008 | Interpolated | 0.000 |
| 18.5 | 22.5 | 0.001 | Interpolated | 0.000 | 0.086 | Interpolated | 0.000 | -0.121 | Interpolated | 0.000 |
| Case 3 | | Potential temperature tendency (K h ⁻¹) | | | Water vapor tendency (g kg ⁻¹ h ⁻¹) | | | Radiative tendency (K h ⁻¹) | | |
| Time | | 0-1 km | 1-2 km | > 2 km | 0-1 km | 1-2 km | > 2 km | 0-1 km | 1-2 km | > 2 km |
| LT, hr | UTC, hr | | | | | | | | | |
| 3.5 | 7.5 | 0.239 | 0.065 | -0.096 | 0.310 | 0.081 | -0.079 | -0.042 | -0.013 | -0.088 |
| 6.5 | 10.5 | 0.156 | 0.053 | 0.125 | 0.239 | 0.043 | -0.166 | -0.040 | -0.033 | -0.077 |
| 9.5 | 13.5 | 0.120 | 0.078 | 0.028 | 0.060 | 0.014 | -0.102 | 0.054 | -0.018 | -0.016 |
| 12.5 | 16.5 | 0.126 | 0.088 | -0.188 | 0.039 | -0.205 | -0.312 | 0.063 | 0.010 | 0.014 |
| 15.5 | 19.5 | 0.342 | 0.350 | -0.146 | -0.111 | 0.097 | -0.172 | -0.008 | -0.006 | 0.010 |
| 18.5 | 22.5 | 0.420 | 0.339 | -0.189 | -0.159 | 0.521 | 0.240 | -0.142 | -0.063 | -0.036 |

Table S1. Meteorological boundary conditions for the three LES cases derived from 3-hourly 1.25° × 1.25° MERRA reanalysis data at the Fair Hill site (39.71°N, 75.87°W).

| Chemical species | Case 1 | | Case 2 | | Case 3 | |
|------------------|-----------------------|------------------------|-----------------------|------------------------|-----------------------|------------------------|
| | Boundary layer (ppbv) | Free atmosphere (ppbv) | Boundary layer (ppbv) | Free atmosphere (ppbv) | Boundary layer (ppbv) | Free atmosphere (ppbv) |
| O3 | 59.96 | 60.44 | 74.57 | 73.47 | 69.48 | 65.45 |
| NO2_NCAR | 1.49 | 0.02 | 1.87 | 0.39 | 1.14 | 0.05 |
| NO | 0.45 | 0.02 | 0.21 | 0.03 | 0.20 | 0.02 |
| HNO3_TD-LIF | 0.83 | 0.08 | 1.39 | 0.50 | 1.69 | 0.56 |
| CH2O_DFGAS | 2.70 | 0.39 | 7.29 | 0.54 | 6.75 | 0.51 |
| Isoprene | 0.27 | 0.03 | 0.93 | 0.02 | 0.62 | 0.01 |
| Monoterpenes | 0.06 | 0.01 | 0.11 | 0.01 | 0.09 | 0.00 |
| MVK+MAC+ISOPOOH | 0.47 | 0.01 | 1.79 | 0.02 | 1.23 | 0.01 |
| Acetaldehyde | 0.46 | 0.05 | 1.23 | 0.28 | 1.03 | 0.22 |
| Methanol | 4.65 | 3.23 | 6.53 | 2.45 | 6.92 | 1.64 |
| Acetone | 2.16 | 1.91 | 4.95 | 2.39 | 4.37 | 1.98 |

Table S2. Initial concentrations (ppbv) for the PBL (below 1 km) and free atmosphere (above 3 km) of P-3B measured chemical species for the three LES cases. Data derived from P-3B measurements from the Fair Hill spiral.

| | Photolysis lifetime (τ_{CH} , min) | | |
|------|--|-------------------|--------------------|
| | Case 1 | Case 2 | Case 3 |
| HCHO | 201.79 \pm 8.91 | 203.38 \pm 9.39 | 207.92 \pm 10.46 |
| MGLY | 128.94 \pm 4.15 | 129.67 \pm 4.35 | 131.70 \pm 4.85 |

Table S3. Modeled midday (1100-1300 LT) photolysis lifetimes of HCHO and MGLY for the three cases. Values shown are temporal (1100-1300 LT) averages, with plus and minus one standard deviation.