

Heterogeneous & Homogeneous & Bio- & Nano-

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CATALYSIS

Supporting Information

Mesoporous High-Surface-Area Copper–Tin Mixed-Oxide Nanorods: Remarkable for Carbon Monoxide Oxidation

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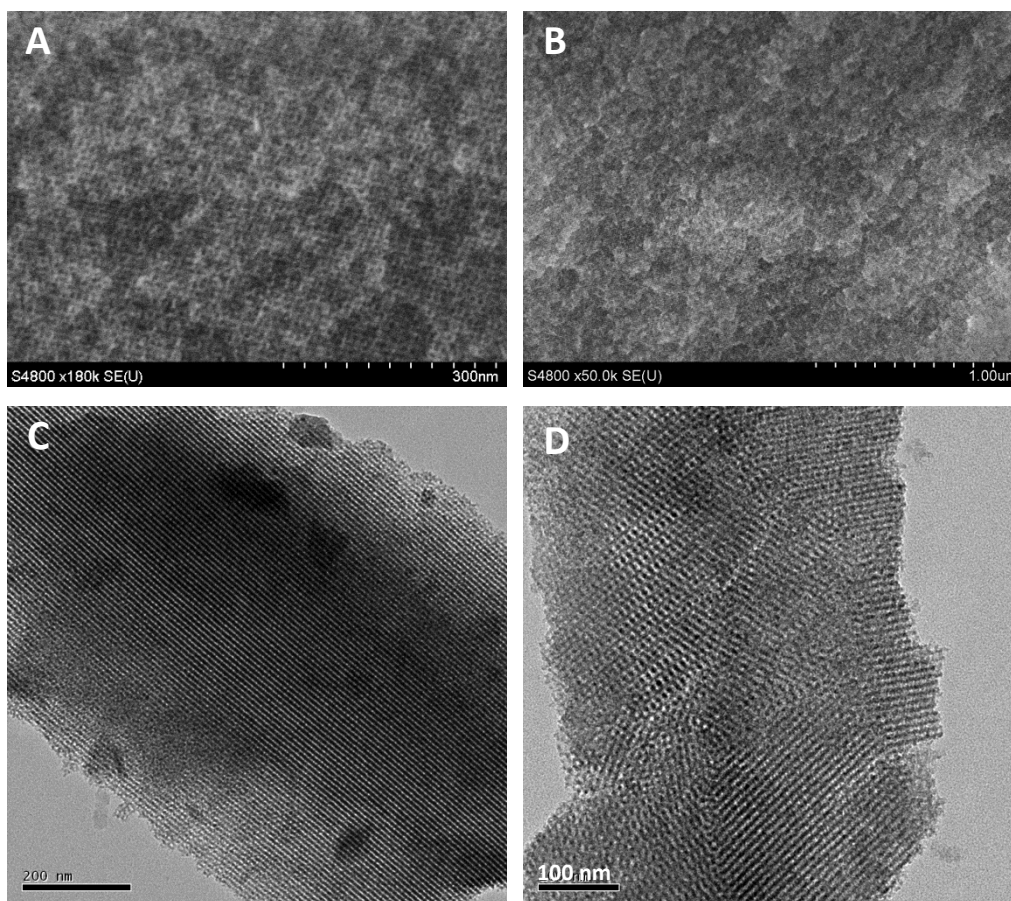


Figure. S1 SEM (A, B) and TEM (C, D) images of KIT-6.

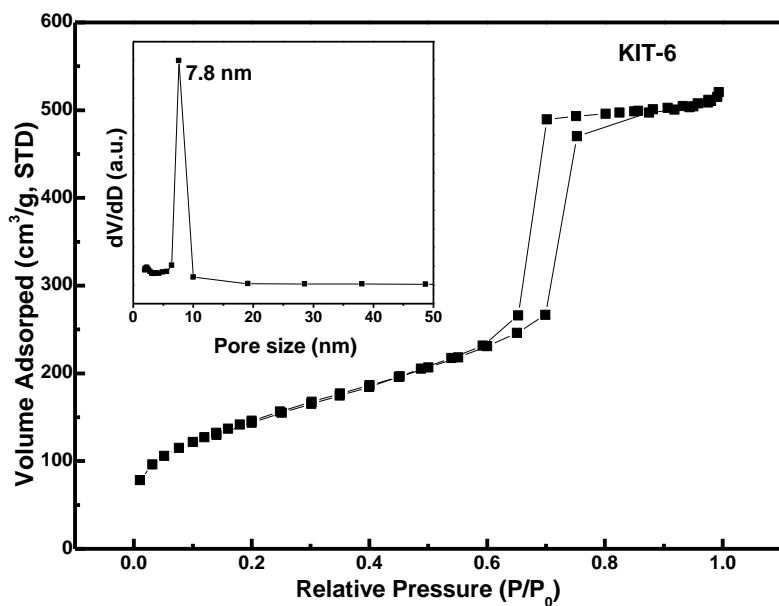


Figure. S2 N₂ adsorption-desorption isotherm and pore size distribution curve of KIT-6.

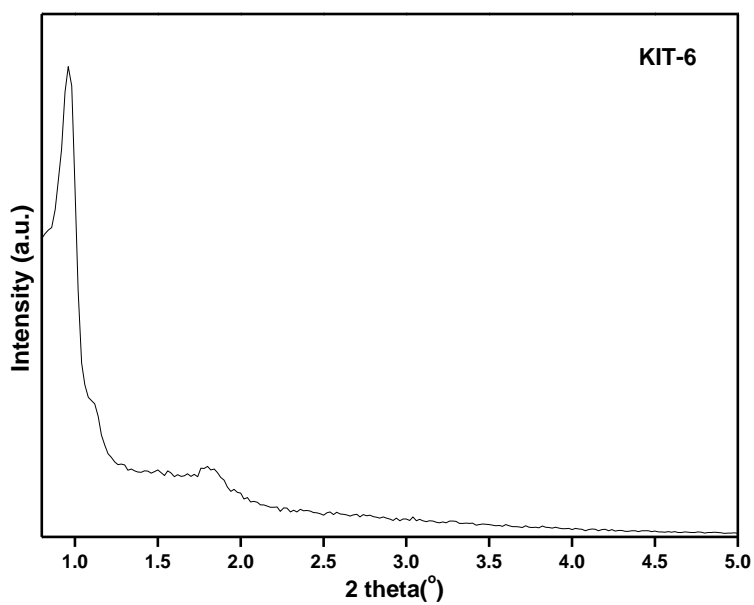


Figure. S3 Small angle XRD patterns of KIT-6.

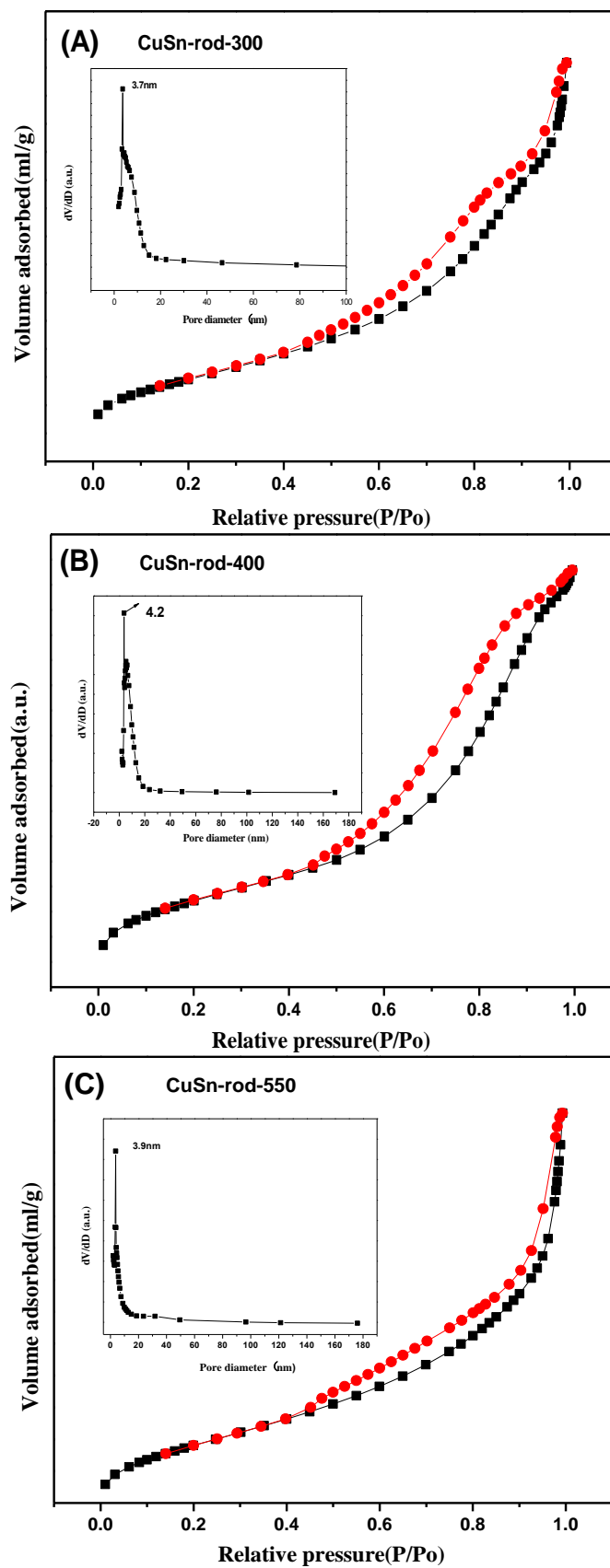


Figure. S4 N₂ adsorption-desorption isotherms and pore size distribution curves of (A)

CuSn-rod-300, (B) CuSn-rod-400 and (D) CuSn-rod-550.

Table S1 Physicochemical properties of mesoporous Cu-Sn nano-rods (molar ratio of Cu: Sn=1: 1) calcined at different temperatures

Samples	Cu/Sn Molar ratio ^a	Cu/Sn Molar ratio ^b	S _{BET} (m ² g ⁻¹)	Pore size (nm) ^c	Average pore Volume (cm ³ g ⁻¹)
KIT-6	-	-	796	7.8	1.10
CuSn-rod-300	0.81	0.76	263	3.7	0.56
CuSn-rod-400	1.12	1.23	157	4.2	0.28
CuSn-rod-550	1.15	1.09	112	3.9	0.28

[a] Measured by ICP. [b] Obtained from XPS.

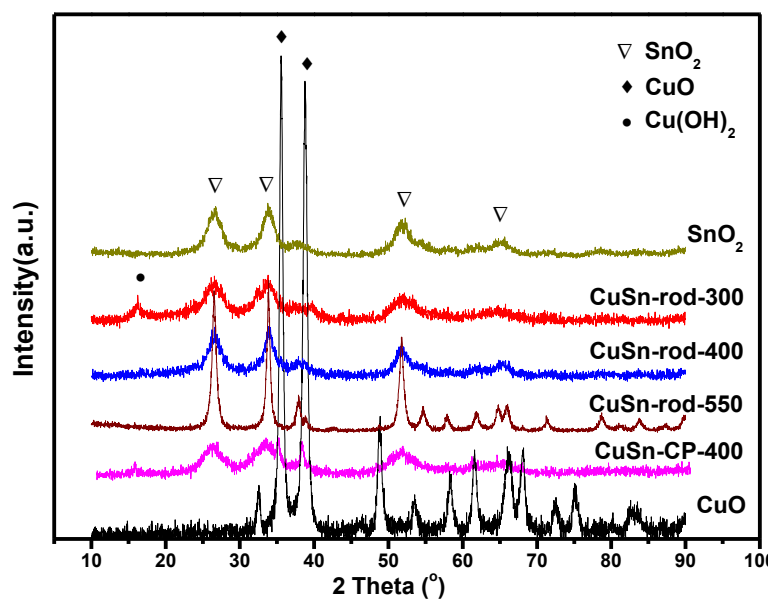


Figure. S5 XRD patterns of mesoporous Cu-Sn nano-rods (molar ratio of Cu: Sn=1: 1) calcined at different temperatures

Table S2 Quantitative XRD results mesoporous Cu-Sn nano-rods (molar ratio of Cu: Sn=1: 1) calcined at different temperatures

Catalysts	SnO ₂ mean crystallite size[nm]	SnO ₂ (110)		SnO ₂ (101)	
		2 θ (°)	d(Å)	2 θ (°)	d(Å)
SnO ₂	3.9	26.86	3.32	33.74	2.65
CuSn-rod-300	3.3	26.76	3.33	33.69	2.65
CuSn-rod-400	3.6	26.72	3.36	33.67	2.66
CuSn-rod-550	6.6	26.74	3.35	33.67	2.66

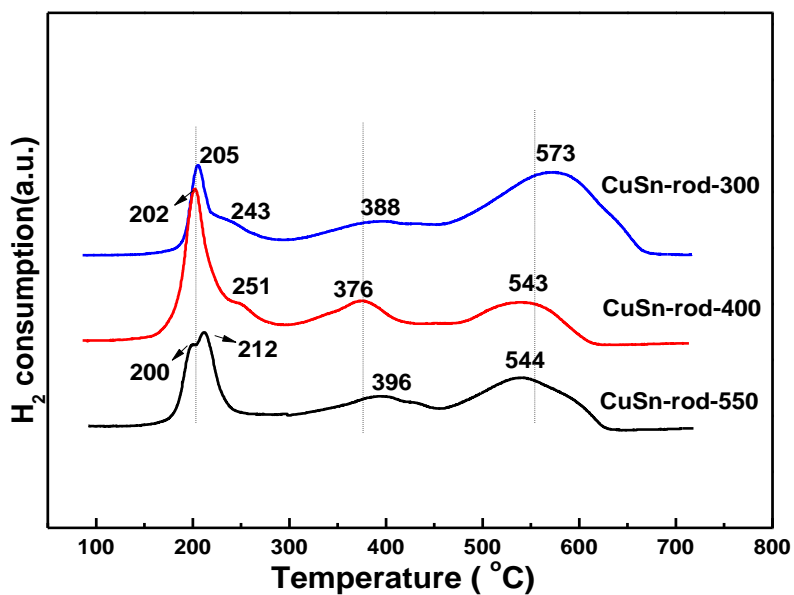


Figure. S6 H₂-TPR profiles of mesoporous Cu-Sn nano-rods (molar ratio of Cu: Sn=1: 1) calcined at different temperatures

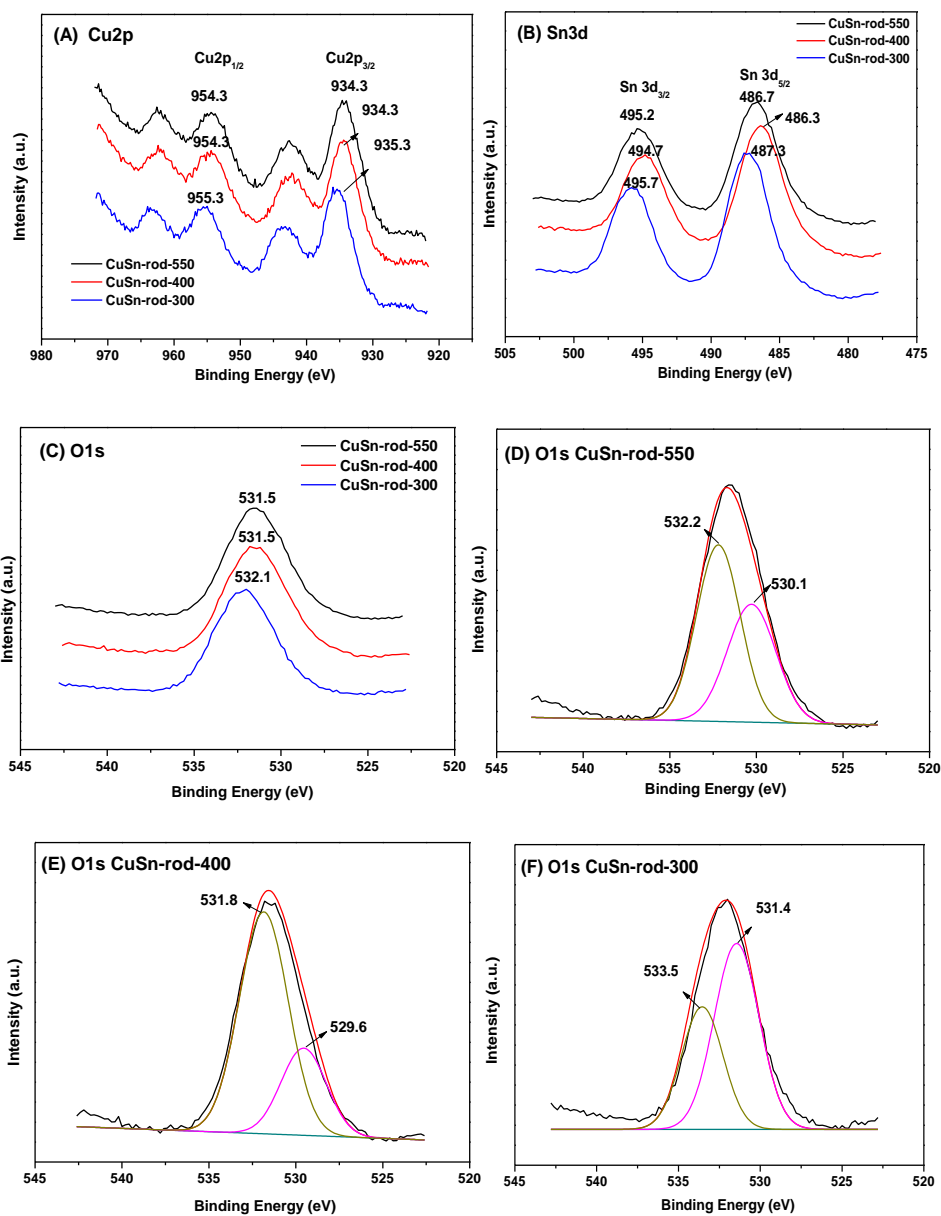


Figure. S7 XPS profiles of mesoporous Cu-Sn nanorods (molar ratio of Cu: Sn=1:1).