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## Supporting Information

Electrochemical Dimerization of Phenylpropenoids and the Surprising Antioxidant Activity of the Resultant Quinone Methide Dimers<br>Kevin J. Romero, Matthew S. Galliher, Mark A. R. Raycroft, Jean-Philippe R. Chauvin, Irene Bosque, Derek A. Pratt,* and Corey R. J. Stephenson*

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1. General Procedures ..... S3
2. Reaction Materials ..... S3
3. Product Analysis ..... S3
4. General Stilbene Preparation ..... S4
a. Stilbene $\mathbf{S 0}$ ..... S5
b. Stilbene $\mathbf{S 1}$ ..... S5
c. Stilbene $\mathbf{S 2}$ ..... S5
d. Stilbene 21 ..... S5
e. Stilbene $\mathbf{2 2}$ ..... S5
f. Stilbene $\mathbf{~ S 3}$ ..... S6
g. Stilbene $\mathbf{S 4}$ ..... S8
h. Stilbene $\mathbf{S 5}$ ..... S10
i. Stilbene $\mathbf{S 6}$ ..... S12
j. Stilbene $\mathbf{S} 7$ ..... S14
k. Stilbene S8 ..... S16
5. Stilbene $\mathbf{S} 9$ ..... S18
m. Stilbene S10 ..... S20
n. Stilbene S11 ..... S22
o. Stilbene S12 ..... S24
p. Stilbene S13 ..... S26
q. Stilbene S14 ..... S28
r. Stilbene S15 ..... S30
s. Stilbene $\mathbf{2 5}$ ..... S32
t. Stilbene 26 ..... S34
6. Cyclic Voltammetry Experimental Procedure ..... S37
7. Cyclic Voltammetry Data ..... S37
a. Stilbene $\mathbf{S 0}$ ..... S37
b. Stilbene $\mathbf{S 1}$ ..... S37
c. Stilbene $\mathbf{S 3}$ ..... S38
d. Stilbene $\mathbf{S 4}$ ..... S38
e. Stilbene $\mathbf{S 5}$ ..... S38
f. $\quad$ Stilbene $\mathbf{S 7}$ ..... S39
g. Stilbene $\mathbf{S 8}$ ..... S39
h. Stilbene $\mathbf{S 9}$ ..... S39
i. Stilbene S10 ..... S40
j. Stilbene S11 ..... S40
k. Stilbene S12 ..... S40
8. Stilbene S13 ..... S41
m. Stilbene S14 ..... S41
9. General Dimerization Procedure ..... S43
10. Optimization and Scalability of Dimerization ..... S44
a. Dimer 5 ..... S45
b. Dimer 6 ..... S46
c. Dimer 7 ..... S47
d. Dimer 10 ..... S49
e. Dimer 11 ..... S51
f. Dimer 12 ..... S53
g. Dimer 13 ..... S55
h. Dimer $\mathbf{1 4}$ ..... S57
i. Dimer 15 ..... S59
j. Dimer 16 ..... S61
k. Dimer 17 ..... S63
11. Dimer 18 ..... S65
m. Dimer 19 ..... S67
n. Dimer 20 ..... S69
o. Dimer 8 ..... S71
p. Dimer 9 ..... S72
q. Dimer 23 ..... S74
r. Dimer 24 ..... S75
s. Dimer 27 ..... S76
t. Dimer 28 ..... S78
12. Synthesis of $( \pm)$-hierochin D ..... S80
a. Compound S16 ..... S80
b. Coniferyl Alcohol (35) ..... S81
c. ( $\pm$ )-hierochin D (3) ..... S83
13. Attempted Direct Cyclization of QMD products: ..... S85
14. Quadrangularin A Analog Synthesis ..... S86
a. Tautomerization Procedure ..... S86
i. Tautomer S17 ..... S87
ii. Tautomer S18 ..... S89
iii. Tautomer S19 ..... S91
iv. Tautomer S20 ..... S93
b. Cyclization Procedure ..... S95
i. Analog 30 ..... S96
ii. Analog 31 ..... S97
iii. Analog 32 ..... S99
iv. Analog 33 ..... S101
15. Analysis of RTA Activity ..... S103
16. References ..... S112

## General Procedures:

Unless specifically noted otherwise, all glassware was flame-dried under vacuum ( $\sim 0.5$ Torr) and cooled under inert atmosphere ( $\mathrm{N}_{2}$ or Ar) prior to use. Each reaction container was charged with a Teflon/PTFE-coated magnetic stir bar and sealed with a rubber septum to maintain a positive pressure of inert atmosphere ( $\mathrm{N}_{2}$ or Ar). Reagents sensitive to the atmosphere were transferred via syringe or cannula as necessary. Reaction conversion was evaluated using analytical thinlayer chromatography (TLC) using Merck silica gel 60 F254 TLC plates. TLC plates were visualized under a dual short wave/long wave UV lamp and/or stained using solutions of $p$-anisaldehyde or potassium permanganate or ceric ammonium molybdate. Stained plates were developed over a heat gun as needed. Reactions were purified via flash column chromatography either with RediSep $® R_{\mathrm{F}}$ Gold silica columns using a Teledyne Isco CombiFlash $\mathrm{R}_{\mathrm{F}}$ automated purification system or manually using 230-400 mesh silica gel. Either sodium sulfate or magnesium sulfate were utilized to exclude water from worked up reactions, and the solvent was removed on Büchi rotary evaporators and/or a Welch vacuum pump. All electrochemical experiments were acquired using either a CH1620E electrochemical analyzer (from CH Instruments) or a uSTAT4000 4-Channel Potentiostat/Galvanostat (from Metrohm USA). Cyclic voltammetry measurements were performed in five-neck cells ( 3 mL ) using a three-electrode set-up in which the working electrode was glassy carbon ( 3 mm diameter), the counter/auxiliary electrode was a platinum wire, and the reference electrode was $\mathrm{Ag} / \mathrm{AgCl}(3 \mathrm{M} \mathrm{KCl}$, from CHInstruments). Bulk electrolysis experiments were performed on discovery scale in open $10-\mathrm{mL}$ vials and in a beaker of the appropriate size ( $15-40 \mathrm{~mL}$ ) for the subsequent scale-up experiments. These reactions used RVC panels (reticulated vitreous carbon, $100 \mathrm{ppi}, 0.25$ inch thickness, $3 \%$ relative density, from McMaster Carr) as the working or counter/auxiliary electrodes and a $\mathrm{Ag} / \mathrm{AgCl}(3 \mathrm{M} \mathrm{KCl})$ reference electrode.

## Reaction Materials:

Commercially available reagents were used without further purification unless specified. Organic solvents (acetonitrile, dichloromethane, diethyl ether, dimethylformamide, dimethyl sulfoxide, methanol, tetrahydrofuran, and toluene) and amine bases (triethylamine, pyridine, $\mathrm{N}, \mathrm{N}$-diisopropylethylamine, and diisopropylamine) were purified prior to use with a Phoenix Solvent Drying System from JC-Meyer Solvent Systems and PureSolv Micro amine drying columns from Innovative Technology, respectively, and kept under a pressure of argon. Solutions of organolithium reagents and Grignard reagents were purchased from Acros Organics and titrated prior to use.

## Product Analysis:

Product names were obtained using ChemDraw Professional 16.0 from Perkin Elmer. For racemic compounds, the name corresponds to the depicted structure. Nuclear magnetic resonance (NMR) spectra were obtained using an internal deuterium lock on Varian Inova 500 or Varian VNMR 500 and 700 spectrometers. For ${ }^{1} \mathrm{H}$ spectra, chemical shifts were referenced to the center line of the residual solvent signal $\left(\mathrm{CDCl}_{3}: \delta 7.26\right.$; acetone- $\left.d_{6}: \delta 2.05\right)$ and are reported in parts per million (ppm). Signal multiplicity is reported as follows: ( $\mathrm{br}=$ broad, $\mathrm{s}=\operatorname{singlet}, \mathrm{d}=$ doublet, $\mathrm{t}=$ triplet, $\mathrm{dd}=$ doublet of doublets, $\mathrm{ddd}=$ doublet of doublet of doublets, $\mathrm{m}=$ multiplet), and the associated coupling constants are given in Hertz. For ${ }^{13} \mathrm{C}$ spectra, experiments were completely heterodecoupled (broadband) and chemical shifts are reported as ppm using the center line of the solvent signal as reference $\left(\mathrm{CDCl}_{3}: \delta 77.16\right.$; acetone- $d_{6}$ : $\delta 29.92$ ). The following resveratrol numbering scheme was used for the assignment of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR signals. High-resolution mass spectra (HRMS) were acquired using a Micromass AutoSpec Ultima Magnetic Sector mass spectrometer using electrospray ionization (ESI), positive ion mode. We thank James Windak and Paul
 Lennon at the University of Michigan, Department of Chemistry Instrumentation Facility for conducting the HRMS experiments. Infrared spectra were acquired using a Perkin-Elmer Spectrum BX FT-IR spectrophotometer using an ATR mount with a ZnSe crystal.

## General Stilbene preparation:

Unless otherwise specified, stilbene substrates were prepared from the benzyl alcohol or benzyl bromide via a Wittig olefination with one of the following aldehydes:


Aldehyde A


Aldehyde B


Aldehyde C

3,5-di-tert-butyl-4-hydroxybenzaldehyde (aldehyde A), commercially available;
4-((trimethylsilyl)oxy)benzaldehyde (aldehyde B), from silyl protection of 4-hydroxybenzaldehyde;
3-methoxy-4-((trimethylsilyl)oxy)benzaldehyde (aldehyde C), from silyl protection of vanillin.

## Starting from the benzyl alcohol:

A solution of triphenylphosphine $(0.97 \mathrm{~g}, 3.71 \mathrm{mmol})$ in dry THF $(2.5 \mathrm{~mL})$ was sparged with $\mathrm{N}_{2}(18 \mathrm{G}$ needle, 5 min$)$ and was added dropwise to a solution of (benzyl) alcohol ( 2.97 mmol ) and carbon tetrabromide ( 1.313 .71 mmol ) in dry THF ( 4.5 mL ) that had been previously sparged ( 18 G needle, 5 min .) in a 50 mL flame-dried round bottom flask and chilled to $0{ }^{\circ} \mathrm{C}$. The mixture was stirred at room temperature for 16 h followed by dropwise addition of methanol ( 1 mL ). The mixture was diluted with EtOAc $(100 \mathrm{~mL})$ and added to a separatory funnel and the organic phase was washed with a 1:1:1 mixture of $10 \%$ bicarbonate solution, saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$, and DI water solution ( 100 mL ). The organic phase was washed with brine and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The organic phase was then concentrated under reduced pressure and purified by silica gel (pre-neutralized) column chromatography and eluted with 10:1 (hexane/ ethyl acetate) to afford the brominated product.

## Starting from the benzyl bromide:

The benzyl bromide was added to a flame-dried round bottom flask charged with a stir bar and fitted with a reflux condenser and dissolved in toluene ( $0.15-0.20 \mathrm{M}$ ). To the stirring solution was added triphenylphosphine ( 1.5 equiv.), and the reaction was heated to $100{ }^{\circ} \mathrm{C}$ for 12 hours. After cooling the reaction mixture to room temperature, the white phosphonium salt was collected via vacuum filtration, and any excess triphenylphosphine was rinsed away with hexanes. The phosphonium salt was dried under vacuum for $>24$ hours prior to use in the Wittig olefination to ensure full removal of residual solvent and water.

The phosphonium salt was added to a flame-dried, 3-neck, round bottom flask charged with stir bar and fitted with a reflux condenser. The salt was suspended in solvent (toluene or THF, 0.1 M ), and to the stirring mixture was added $n \mathrm{BuLi}$ ( 1.6 or $2.5 \mathrm{M}, 1.00$ equiv.). After 30 minutes, the reaction mixture had become a brilliant red, clear solution, at which point the aldehyde was added under a stream of nitrogen, and the reaction was heated at reflux for 12 hours. After cooling the reaction mixture to room temperature, the reaction was quenched with saturated aqueous ammonium chloride, diluted with ethyl acetate, and added to a separatory funnel containing additional aqueous ammonium chloride. The layers were separated, and the aqueous layer was extracted with additional ethyl acetate. The combined organic layers were washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure. The crude stilbene products were purified via flash column chromatography (see characterization data for specific chromatography conditions).

The following substrates were acquired from previous investigations reported by our group. ${ }^{1,2}$

(S0) (E)-4-(3,5-bis(benzyloxy)styryl)-2,6-di-tert-butylphenol

(S1) (E)-2,6-di-tert-butyl-4-(3,5-dimethoxystyryl)phenol

(S2) (E)-2,6-di-tert-butyl-4-(3,5-dimethoxystyryl)phenol

(21) 4-((E)-2-((2S,3S)-6-(benzyloxy)-2-(4-(benzyloxy)phenyl)-3-(3,5-bis(benzyloxy)phenyl)-2,3-dihydrobenzofuran-4-yl)vinyl)-2,6-di-tert-butylphenol

(22) 4-((E)-2-((2S,3S)-6-(benzyloxy)-2-(4-(benzyloxy)phenyl)-3-(3,5-bis(benzyloxy)phenyl)-2,3-dihydrobenzofuran-4-yl)vinyl)-2,6-bis(trimethylsilyl)phenol


## (S3) (E)-4-(3,5-bis((4-methoxybenzyl)oxy)styryl)-2,6-di-tert-butylphenol

Commercially available 3,5 -dihydroxybenzyl alcohol ( $2.0 \mathrm{~g}, 14.3 \mathrm{mmol}$ ) was added to flask charged with potassium carbonate ( 2.25 equiv., 4.44 g ) and tetrabutylammonium iodide ( 0.2 equiv., 1.05 g ), and the solids were dissolved/suspended in acetone ( 42 mL ). To the stirring reaction mixture was added 4-methoxybenzyl chloride ( 2.20 equiv., 4.26 mL ), and the reaction was allowed to stir at room temperature for 16 hours. Upon completion, the reaction was diluted with ethyl acetate and poured into a separatory funnel containing deionized water. The layers were separated, and the aqueous layer was extracted with additional portions of ethyl acetate. The combined organic layer was washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure to afford the crude product, which was carried forward without further purification. The PMB-protected material ( $2.18 \mathrm{~g}, 5.72 \mathrm{mmol}$ ) was subjected to the general procedure using toluene as the solvent for the olefination with aldehyde A. The product was purified by column chromatography ( $4 \%$ to $28 \%$ ethyl acetate in hexanes) to afford (E)-4-(3,5-bis((4-methoxybenzyl)oxy)styryl)-2,6-di-tert-butylphenol ( $2.06 \mathrm{~g}, 62 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.22$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.37$ (d, $J=8.6 \mathrm{~Hz}, 4 \mathrm{H}$ ), 7.33 ( $\mathrm{s}, 2 \mathrm{H}$ ), $7.04(\mathrm{~d}, J=16.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.93$ (d, $J=8.6 \mathrm{~Hz}$, $4 \mathrm{H}), 6.87(\mathrm{~d}, J=16.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.76(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.51(\mathrm{t}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.29(\mathrm{~s}, 1 \mathrm{H}), 5.00(\mathrm{~s}, 4 \mathrm{H}), 3.83(\mathrm{~s}, 6 \mathrm{H})$, 1.48 (s, 18H).
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta 160.32$, 159.61, 154.08, 140.13, 136.29, 130.26, 129.44, 129.17, 128.56, 125.92, 123.66, 114.15, 105.56, 101.24, 70.04, 55.47, 34.53, 30.43.

IR (Neat): 3606, 2965, 1577, 1515, 1439, 1245, 1147, 1046, 1026, 967, $858 \mathrm{~cm}^{-1}$;
HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calculated for $\mathrm{C}_{38} \mathrm{H}_{45} \mathrm{O}_{5}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right) 581.3262$, found 581.3267.

${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Stilbene $\mathbf{S 3}$




## (S4) (E)-2,6-di-tert-butyl-4-styrylphenol

Commercially available benzyl bromide ((bromomethyl)benzene, $1.1 \mathrm{~g}, 6.4 \mathrm{mmol}$ ) was subjected to the general procedure using toluene as the solvent for the olefination with aldehyde A. The product was purified by column chromatography ( $2 \%$ to $12 \%$ ethyl acetate in hexanes) to afford ( $E$ )-2,6-di-tert-butyl-4-styrylphenol ( $1.64 \mathrm{~g}, 83 \%$ yield). The acquired ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were consistent with those reported in the literature.
$\mathrm{R}_{\mathrm{f}}=0.31$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.50(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{~s}, 2 \mathrm{H}), 7.34(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.23(\mathrm{t}, J=7.3 \mathrm{~Hz}$, $1 \mathrm{H}), 7.08(\mathrm{~d}, J=16.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.28(\mathrm{~s}, 1 \mathrm{H}), 1.49(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (126 MHz, Chloroform- $d$ ) $\delta 153.99$, 138.09, 136.28, 129.70, 128.73, 127.11, 126.33, 126.00, 123.57, 34.53, 30.44 .

IR (Neat): 3616, 2953, 1470, 1235, 1137, 1118, $957 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{22} \mathrm{H}_{29} \mathrm{O}^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$309.2213, found 309.2206.


${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Stilbene $\mathbf{S} 4$





## (S5) (E)-4-(2-(benzo[d][1,3]dioxol-5-yl)vinyl)-2,6-di-tert-butylphenol

Commercially available benzyl alcohol (benzo $[d][1,3]$ dioxol-5-ylmethanol, $1.00 \mathrm{~g}, 6.6 \mathrm{mmol}$ ) was subjected to the general procedure using toluene as the solvent for the olefination with aldehyde A. The product was purified by column chromatography ( $2 \%$ to $12 \%$ ethyl acetate in hexanes) to afford ( $E$ )-4-(2-(benzo[d][1,3]dioxol-5-yl)vinyl)-2,6-di-tertbutylphenol ( $1.72 \mathrm{~g}, 74 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.26$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.31$ (s, 2H), 7.06 (s, 1H), $6.92(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.91 (d, J = $16.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.86 (d, $J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.97$ (s, 2H), 5.26 (s, 1H), 1.48 (s, 18H).
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta 153.79,148.22$, 146.94, 136.28, 132.69, 128.82, 128.10, 125.71, 123.36, 121.03, 108.50, 105.52, 101.16, 34.53, 30.44.

IR (Neat): 3625, 2955, 1486, 1435, 1249, 1135, 1036, $950 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{23} \mathrm{H}_{29} \mathrm{O}_{3}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$353.2111, found 353.2107.


${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Stilbene $\mathbf{S 5}$



## (S6) (E)-4-(3,4-bis(benzyloxy)styryl)-2,6-di-tert-butylphenol

Freshly distilled diisopropylamine ( $1.04 \mathrm{mmol}, 146 \mu \mathrm{~L}$ ) was added to a flame-dried heart-shaped flask, dissolved in freshly distilled THF ( 2 mL ), and cooled to $-78^{\circ} \mathrm{C}$. To the stirring solution was added $n \mathrm{BuLi}(1.00 \mathrm{mmol}, 400 \mu \mathrm{~L}, 2.5 \mathrm{M}$ ), and the solution was allowed to stir at the same temperature for 30 min . Meanwhile, in a 3-neck round bottom flask, the phosphonium salt A ( $1.00 \mathrm{mmol}, 634 \mathrm{mg}$, available in 3 steps $^{2}$ ) was suspended in freshly distilled THF ( 10 mL ) and cooled to $-78{ }^{\circ} \mathrm{C}$. The freshly prepared LDA solution was added to the phosphonium salt suspension via cannula, and the ylid was allowed to form at the same temperature for 30 min , turning the solution deep red. To a flame-dried heart-shaped flask was added 3,4-benzyloxybenzaldehyde ( $0.80 \mathrm{mmol}, 255 \mathrm{mg}$, available in 3 steps via alkylation, reduction, and oxidation), and the solid was dissolved in THF ( 5 mL ). The aldehyde solution was added to the ylid via cannula, and the reaction was allowed to warm to room temperature overnight ( $\sim 15$ hours). The reaction was subsequently cooled to $0{ }^{\circ} \mathrm{C}$, and a solution of TBAF ( $1.00 \mathrm{mmol}, 1.00 \mathrm{~mL}, 1.0 \mathrm{M}$ ) was added. The desilylation was allowed to occur for 30 min , at which point the reaction was quenched via the addition of saturated ammonium chloride, diluted with EtOAc, and added to a separatory funnel containing deionized water. The layers were separated, and the aqueous layer was extracted with additional portions of EtOAc. The combined organic layers were washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure. The crude product was purified by flash column chromatography ( $10 \%$ to $60 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in Hexanes) to afford the stilbene product ( $360 \mathrm{mg}, 86 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.30\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} /\right.$ hexanes $\left.1: 1 ; \mathrm{UV}\right)$
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.49$ (d, $J=7.5 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.45 (d, $J=7.5 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.40-7.37$ (m, 2H), $7.37-7.35$ $(\mathrm{m}, 2 \mathrm{H}), 7.34-7.30(\mathrm{~m}, 2 \mathrm{H}), 7.30(\mathrm{~s}, 2 \mathrm{H}), 7.14(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.01(\mathrm{dd}, J=8.3,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.91(\mathrm{~d}, J=8.3 \mathrm{~Hz}$, $1 \mathrm{H}), 6.89(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.83(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.25(\mathrm{~s}, 1 \mathrm{H}), 5.21(\mathrm{~s}, 2 \mathrm{H}), 5.17(\mathrm{~s}, 2 \mathrm{H}), 1.48(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta$ 153.77, 149.34, 148.56, 137.52, 137.49, 136.26, 132.10, 128.90, 128.63, 128.61, $128.28,127.95,127.91,127.58,127.45,125.64,123.36,120.14,115.39,112.86,71.62,71.55,34.53,30.45$.

IR (Neat): 3617, 2951, 1680, 1595, 1504, 1255, 1133, 1016, 729, $695 \mathrm{~cm}^{-1}$;
HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calculated for $\mathrm{NaC}_{36} \mathrm{H}_{40} \mathrm{O}_{3}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$543.2870, found 543.2873 .
${ }^{1} \mathrm{H}$ NMR, 700 MHz , Chloroform- $d$, Stilbene S6

${ }^{13}$ C NMR, 176 MHz , Chloroform- $d$, Stilbene S6



## (S7) (E)-2,6-di-tert-butyl-4-(4-(methylthio)styryl)phenol

Commercially available benzyl bromide ((4-(bromomethyl)phenyl)(methyl)sulfane, $1.0 \mathrm{~g}, 4.6 \mathrm{mmol}$ ) was subjected to the general procedure using toluene as the solvent for the olefination with aldehyde A . The product was purified by column chromatography ( $2 \%$ to $12 \%$ ethyl acetate in hexanes) to afford ( $E$ )-2,6-di-tert-butyl-4-(4-(methylthio)styryl)phenol ( 1.35 g, $82 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.34$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.42(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.33(\mathrm{~s}, 2 \mathrm{H}), 7.24(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.04(\mathrm{~d}, J=16.2 \mathrm{~Hz}$, $1 \mathrm{H}), 6.89(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.28(\mathrm{~s}, 1 \mathrm{H}), 2.50(\mathrm{~s}, 3 \mathrm{H}), 1.48(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta 153.96$, 137.00, 136.30, 135.20, 129.18, 128.73, 127.04, 126.72, 125.31, 123.50, 34.53, 30.44, 16.18.

IR (Neat): 3616, 2955, 1435, 1249, 1184, 962, $800 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{23} \mathrm{H}_{31} \mathrm{OS}^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right) 355.2090$, found 355.2087.

${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Stilbene $\mathbf{S 7}$




## (S8) (E)-4-(4-(benzyloxy)styryl)-2,6-di-tert-butylphenol

Commercially available benzyl alcohol ((4-(benzyloxy)phenyl)methanol, $1.00 \mathrm{~g}, 4.7 \mathrm{mmol}$ ) was subjected to the general procedure using toluene as the solvent for the olefination with aldehyde A. The product was purified by column chromatography ( $2 \%$ to $12 \%$ ethyl acetate in hexanes) to afford ( $E$ )-4-(4-(benzyloxy)styryl)-2,6-di-tert-butylphenol ( 1.52 g , $78 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.28$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.44(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.44(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.40(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.33(\mathrm{t}, J$ $=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.32(\mathrm{~s}, 2 \mathrm{H}), 6.96(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.95(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.89(\mathrm{~d}, J=16.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.25(\mathrm{~s}, 1 \mathrm{H}), 5.09$ ( $\mathrm{s}, 2 \mathrm{H}$ ), 1.48 ( $\mathrm{s}, 18 \mathrm{H}$ ).
${ }^{13} \mathrm{C}$ NMR (126 MHz, Chloroform-d) $\delta 158.20$, 153.69, 137.16, 136.24, 131.21, 129.05, 128.73, 128.11, 127.83, 127.63, $127.48,125.53,123.30,115.17,70.22,34.53,30.45$.

IR (Neat): 3628, 2953, 1606, 1510, 1465, 1233, 1175, 1010, 954, $745 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{29} \mathrm{H}_{35} \mathrm{O}_{2}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$415.2632, found 415.2632.

${ }^{13}$ C NMR, 126 MHz , Chloroform- $d$, Stilbene $\mathbf{S 8}$



## (S9) tert-butyl (E)-(4-(3,5-di-tert-butyl-4-hydroxystyryl)phenyl)carbamate

Freshly distilled diisopropylamine ( $3.41 \mathrm{mmol}, 478 \mu \mathrm{~L}$ ) was added to a flame-dried heart-shaped flask, dissolved in freshly distilled THF ( 10 mL ), and cooled to $-78^{\circ} \mathrm{C}$. To the stirring solution was added $n \mathrm{BuLi}(3.28 \mathrm{mmol}, 1.31 \mathrm{~mL}, 2.5 \mathrm{M}$ ), and the solution was allowed to stir at the same temperature for 30 min . Meanwhile, in a 3 -neck round bottom flask, the phosphonium salt A ( $3.28 \mathrm{mmol}, 2.08 \mathrm{~g}$ ) was suspended in freshly distilled THF ( 25 mL ) and cooled to $-78^{\circ} \mathrm{C}$. The freshly prepared LDA solution was added to the phosphonium salt suspension via cannula, and the ylid was allowed to form at the same temperature for 30 min , turning the solution deep red. To a flame-dried heart-shaped flask was added 4 -NHBocbenzaldehyde ( $2.62 \mathrm{mmol}, 580 \mathrm{mg}$, available in 2 steps via Boc protection and oxidation), and the solid was dissolved in THF ( 10 mL ). The aldehyde solution was added to the ylid via cannula, and the reaction was allowed to warm to room temperature overnight ( $\sim 15$ hours). The reaction was subsequently cooled to $0^{\circ} \mathrm{C}$, and a solution of TBAF ( $3.28 \mathrm{mmol}, 3.28$ $\mathrm{mL}, 1.0 \mathrm{M}$ ) was added. The desilylation was allowed to occur for 30 min , at which point the reaction was quenched via the addition of saturated ammonium chloride, diluted with EtOAc, and added to a separatory funnel containing deionized water. The layers were separated, and the aqueous layer was extracted with additional portions of EtOAc. The combined organic layers were washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure. The crude product was purified by flash column chromatography ( $20 \%$ to $100 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in Hexanes) to afford the stilbene product ( 1.00 g , $90 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.30\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} /\right.$ hexanes 3:1; UV $)$
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.43(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.32(\mathrm{~s}, 2 \mathrm{H}), 6.98(\mathrm{~d}, J=16.2 \mathrm{~Hz}$, $1 \mathrm{H}), 6.89$ (d, $J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.47$ (s, 1H), 5.26 (s, 1H), 1.53 (s, 9H), 1.47 (s, 18H).
${ }^{13}$ C NMR ( 176 MHz , Chloroform- $d$ ) $\delta 153.79,152.77,137.31,136.22,133.06,128.87,128.36,126.93,125.42,123.39$, 118.69, 80.71, 34.52, 30.43, 28.50.

IR (Neat): 3623, 3325, 2957, 1709, 1600, 1521, 1234, 1153, 1051, $740 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{NaC}_{27} \mathrm{H}_{37} \mathrm{NO}_{3}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right) 446.2666$, found 446.2660 .
${ }^{1} \mathrm{H}$ NMR, 700 MHz , Chloroform- $d$, Stilbene $\mathbf{S 9}$


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${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Stilbene $\mathbf{S 9}$



## (S10) (E)-2,6-di-tert-butyl-4-(2-nitrostyryl)phenol

Commercially available benzyl bromide (1-(bromomethyl)-2-nitrobenzene, $1.0 \mathrm{~g}, 4.6 \mathrm{mmol}$ ) was subjected to the general procedure using toluene as the solvent for the olefination. The product was purified by column chromatography ( $2 \%$ to $12 \%$ ethyl acetate in hexanes) to afford ( $E$ )-2,6-di-tert-butyl-4-(2-nitrostyryl)phenol ( $927 \mathrm{mg}, 57 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.38$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.94(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.76(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.57(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.44(\mathrm{~d}, J$ $=16.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.37(\mathrm{~s}, 3 \mathrm{H}), 7.36(\mathrm{t}, J=7.7 \mathrm{~Hz}, 3 \mathrm{H}), 7.07(\mathrm{~d}, J=16.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.38(\mathrm{~s}, 1 \mathrm{H}), 1.48(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta 154.83,147.95,136.44,135.03,133.80,133.07,128.09,128.05,127.40,124.91$, 124.29, 120.65, 34.52, 30.38.

IR (Neat): 3626, 2961, 1626, 1603, 1515, 1348, 1233, 1150, 965, $742 \mathrm{~cm}^{-1}$;
HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calculated for $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{NO}_{3}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$354.2064, found 354.2064.

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${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Stilbene $\mathbf{S 1 0}$



## (S11) (E)-2,6-di-tert-butyl-4-(2-(trifluoromethyl)styryl)phenol

Commercially available benzyl bromide (1-(bromomethyl)-2-(trifluoromethyl)benzene, $0.79 \mathrm{~g}, 3.29 \mathrm{mmol}$ ) was subjected to the general procedure using THF as the solvent for the olefination. The product was purified by column chromatography ( $2 \%$ to $12 \%$ ethyl acetate in hexanes) to afford (E)-2,6-di-tert-butyl-4-(2-(trifluoromethyl)styryl)phenol ( $989 \mathrm{mg}, 73 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.40$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.76(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.66-7.63(\mathrm{~m}, 1 \mathrm{H}), 7.51(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.37(\mathrm{~s}, 2 \mathrm{H})$, $7.33-7.28(\mathrm{~m}, 2 \mathrm{H}), 7.04(\mathrm{~d}, J=16.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.34(\mathrm{~s}, 1 \mathrm{H}), 1.48(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (176 MHz, Chloroform-d) $\delta 154.48,137.16,136.34,133.57,131.92,128.37,127.32\left(\mathrm{q}, J_{\mathrm{C}-\mathrm{F}}=29 \mathrm{~Hz}\right), 126.90$, $126.71,125.99\left(\mathrm{q}, J_{\mathrm{C}-\mathrm{F}}=6.2 \mathrm{~Hz}\right), 124.66\left(\mathrm{q}, J_{\mathrm{C}-\mathrm{F}}=275 \mathrm{~Hz}\right), 124.03,121.71,34.50,30.36$.

IR (Neat): 3636, 2956, 1435, 1422, 1308, 1122, 1035, 960, $753 \mathrm{~cm}^{-1}$;
HRMS (ESI) m/z calculated for $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{~F}_{3} \mathrm{O}^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$377.2087, found 377.2084.

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Stilbene $\mathbf{S 1 1}$





## (S12) (E)-2,6-di-tert-butyl-4-(3-(trifluoromethyl)styryl)phenol

Commercially available benzyl bromide (1-(bromomethyl)-3-(trifluoromethyl)benzene, $0.78 \mathrm{~g}, 3.25 \mathrm{mmol}$ ) was subjected to the general procedure using THF as the solvent for the olefination. The product was purified by column chromatography ( $2 \%$ to $12 \%$ ethyl acetate in hexanes) to afford ( $E$ )-2,6-di-tert-butyl-4-(3-(trifluoromethyl)styryl)phenol ( $1.06 \mathrm{~g}, 87 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.40$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR (700 MHz, Chloroform- $d$ ) $\delta 7.73(\mathrm{~s}, 1 \mathrm{H}), 7.66(\mathrm{dd}, J=7.0,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.48-7.42(\mathrm{~m}, 2 \mathrm{H}), 7.36(\mathrm{~s}, 2 \mathrm{H}), 7.14$ $(\mathrm{d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.34(\mathrm{~s}, 1 \mathrm{H}), 1.48(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta 154.42,138.85,136.38,131.48,131.13\left(\mathrm{q}, J_{\mathrm{C}-\mathrm{F}}=32 \mathrm{~Hz}\right), 129.33,129.14,128.11$, $124.37\left(\mathrm{q}, J_{\mathrm{C}-\mathrm{F}}=273 \mathrm{~Hz}\right), 124.33,123.82,123.51\left(\mathrm{q}, J_{\mathrm{C}-\mathrm{F}}=3.5 \mathrm{~Hz}\right), 122.88\left(\mathrm{q}, J_{\mathrm{C}-\mathrm{F}}=3.5 \mathrm{~Hz}\right), 34.54,30.41$.

IR (Neat): 3639, 2957, 1655, 1437, 1328, 1163, 1121, $1073 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{~F}_{3} \mathrm{O}^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$377.2087, found 377.2093.

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Stilbene $\mathbf{S 1 2}$



## (S13) (E)-2,6-di-tert-butyl-4-(4-nitrostyryl)phenol

Commercially available benzyl bromide (1-(bromomethyl)-4-nitrobenzene, $1.0 \mathrm{~g}, 4.6 \mathrm{mmol}$ ) was subjected to the general procedure using toluene as the solvent for the olefination with aldehyde A. The product was purified by column chromatography ( $2 \%$ to $12 \%$ ethyl acetate in hexanes) to afford (E)-2,6-di-tert-butyl-4-(4-nitrostyryl)phenol ( $990 \mathrm{mg}, 61 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.38$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR ( 500 MHz, Chloroform- $d$ ) $\delta 8.20(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.60(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.38(\mathrm{~s}, 2 \mathrm{H}), 7.24(\mathrm{~d}, J=16.1 \mathrm{~Hz}$, $2 \mathrm{H}), 6.97(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.41(\mathrm{~s}, 1 \mathrm{H}), 1.48(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (126 MHz, Chloroform- $d$ ) $\delta 155.07$, 146.40, 144.77, 136.56, 134.49, 127.69, 126.53, 124.29, 123.44, 34.56, 30.39.

IR (Neat) $3618,2955,1629,1595,1505,1428,1332,1117,973,861,748 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{NO}_{3}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right) 354.2064$, found 354.2068.
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${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Stilbene $\mathbf{S 1 3}$



## (S14) (E)-2,6-di-tert-butyl-4-(4-(trifluoromethyl)styryl)phenol

Commercially available benzyl bromide (1-(bromomethyl)-4-(trifluoromethyl)benzene, $1.0 \mathrm{~g}, 4.18 \mathrm{mmol}$ ) was subjected to the general procedure using THF as the solvent for the olefination. The product was purified by column chromatography ( $2 \%$ to $12 \%$ ethyl acetate in hexanes) to afford ( $E$ )-2,6-di-tert-butyl-4-(4-(trifluoromethyl)styryl)phenol ( $1.15 \mathrm{~g}, 77 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.40$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.58$ ( $\mathrm{s}, 4 \mathrm{H}$ ), 7.36 (s, 2H), 7.16 (d, $J=16.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.95 (d, $J=16.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.35 (s, 1H), 1.49 (s, 18H).
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta$ 154.56, 141.63, 136.43, 132.26, 128.71 (q, $J=32 \mathrm{~Hz}$ ), 128.11, 126.33, 125.68 (q, $J$ $=3.5 \mathrm{~Hz}), 124.52(\mathrm{q}, J=271 \mathrm{~Hz}), 124.32$, 123.93, 34.54, 30.41 .

IR (Neat): 3636, 2955, 1435, 1422, 1308, 1122, 1035, 960, $753 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{~F}_{3} \mathrm{O}^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$377.2087, found 377.2084.

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Stilbene $\mathbf{S 1 4}$



## (S15) (E)-4-(3,5-bis(benzyloxy)styryl)-2,6-bis(triethylsilyl)phenol

Commercially available 3,5 -dihydroxybenzyl alcohol ( $2.0 \mathrm{~g}, 14.3 \mathrm{mmol}$ ) was added to flask charged with potassium carbonate ( 2.25 equiv., 4.44 g ) and tetrabutylammonium iodide ( 0.2 equiv., 1.05 g ), and the solids were dissolved/suspended in acetone ( 42 mL ). To the stirring reaction mixture was added 4-methoxybenzyl chloride ( 2.20 equiv., 3.74 mL ), and the reaction was allowed to stir at room temperature for 16 hours. Upon completion, the reaction was diluted with ethyl acetate and poured into a separatory funnel containing deionized water. The layers were separated, and the aqueous layer was extracted with additional portions of ethyl acetate. The combined organic layer was washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure to afford the crude product, which was carried forward without further purification. The benzyl-protected material ( $2.22 \mathrm{~g}, 5.79 \mathrm{mmol}$ ) was subjected to the general procedure using THF as the solvent for the olefination with aldehyde D , which is available from silyl protection of commercially available 3,5-dibromo-4-hydroxybenzaldehyde. The product was purified by column chromatography ( $2 \%$ to $12 \%$ ethyl acetate in hexanes, $2 \%$ increments, 2 column volumes each, then 2 column volumes at both $16 \%$ and $20 \%$ ) to afford both olefin isomers in a $\sim 1: 1$ ratio ( $2.26 \mathrm{~g}, 69 \%$ yield combined). The E-isomer ( $900 \mathrm{mg}, 1.59 \mathrm{mmol}$ ) was carried forward and dissolved in THF, and the reaction mixture was cooled to $-78{ }^{\circ} \mathrm{C} . n \mathrm{BuLi}(1.0$ equiv, 1.6 M in hexanes, $993 \mu \mathrm{~L}$ ) was added dropwise to the stirring solution, and the reaction was held at temperature for 15 min prior to being allowed to warm to room temperature. The retro-Brook rearrangement was quenched and worked up following the general procedure. The resulting product was silyl protected following standard conditions, then subjected to the same retro-Brook reaction to afford the product ( $E$ )-4-(3,5-bis(benzyloxy)styryl)-2,6-bis(triethylsilyl)phenol ( $861 \mathrm{mg}, 85 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.37$ (ethyl acetate/hexanes 1:9; UV)
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.48(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 4 \mathrm{H}), 7.46(\mathrm{~s}, 2 \mathrm{H}), 7.41(\mathrm{t}, J=7.5 \mathrm{~Hz}, 4 \mathrm{H}), 7.35(\mathrm{t}, J=7.2 \mathrm{~Hz}$, $2 \mathrm{H}), 7.07(\mathrm{~d}, J=16.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.88(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.80(\mathrm{t}, J=1.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.55(\mathrm{q}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{~s}, 4 \mathrm{H})$, $5.06(\mathrm{~s}, 1 \mathrm{H}), 1.02(\mathrm{t}, J=8.0 \mathrm{~Hz}, 18 \mathrm{H}), 0.91(\mathrm{q}, J=7.7 \mathrm{~Hz}, 12 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta 166.04$, 160.27, 140.16, 137.07, 136.01, 129.77, 129.09, 128.72, 128.12, 127.72, $125.88,121.44,105.62,101.18,70.25,7.70,3.98$.

IR (Neat): 3592, 2951, 2870, 1591, 1453, 1397, 1161, 1144, 1060, 1004, 953, 723, $692 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{40} \mathrm{H}_{53} \mathrm{O}_{3} \mathrm{Si}_{2}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right) 637.3528$, found 637.3535 .
${ }^{1} \mathrm{H}$ NMR, 700 MHz , Chloroform- $d$, Stilbene $\mathbf{S 1 5}$

\begin{abstract}
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| T | T | 1 | 1 | 1 | 1 | , | , | , | . | , | 1 | 1 | . 5 | 1. | 1 | 1 |
| 8.0 | 7.5 | 7.0 | 6.5 | 6.0 | 5.5 | 5.0 | 4.5 | $\begin{gathered} 4.0 \\ f 1(\mathrm{ppm}) \end{gathered}$ | 3.5 | 3.0 | 2.5 | 2.0 | 1.5 | 1.0 | 0.5 | 0.1 |

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Stilbene $\mathbf{S 1 5}$





## (25) (E)-4-(3,5-bis((4-methoxybenzyl)oxy)styryl)phenol

Commercially available 3,5 -dihydroxybenzyl alcohol ( $2.0 \mathrm{~g}, 14.3 \mathrm{mmol}$ ) was added to flask charged with potassium carbonate ( 2.25 equiv., 4.44 g ) and tetrabutylammonium iodide ( 0.2 equiv., 1.05 g ), and the solids were dissolved/suspended in acetone ( 42 mL ). To the stirring reaction mixture was added 4-methoxybenzyl chloride ( 2.20 equiv., 4.26 mL ), and the reaction was allowed to stir at room temperature for 16 hours. Upon completion, the reaction was diluted with ethyl acetate and poured into a separatory funnel containing deionized water. The layers were separated, and the aqueous layer was extracted with additional portions of ethyl acetate. The combined organic layer was washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure to afford the crude product, which was carried forward without further purification. A portion of the PMB-protected benzyl alcohol ( $783 \mathrm{mg}, 2.06 \mathrm{mmol}$ ) was subjected to the general procedure using toluene as the solvent for the olefination with aldehyde B. Upon completion of the olefination, the reaction was cooled to $0^{\circ} \mathrm{C}$, and TBAF ( 1.0 equiv., 1.0 M in THF, 2.06 mL ) was added dropwise. The reaction was allowed to stir for 30 min , at which point it was quenched and worked up following the general procedure. The product was purified by column chromatography ( $8 \%$ to $40 \%$ ethyl acetate in hexanes) to afford the desired product ( $478 \mathrm{mg}, 49 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.30$ (ethyl acetate/hexanes 3:7; UV)
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.38$ (d, $J=8.3 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.37 (d, $\left.J=8.5 \mathrm{~Hz}, 4 \mathrm{H}\right) 7.01(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.93$ (d, $J=8.6 \mathrm{~Hz}, 4 \mathrm{H}), 6.88(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.74(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.53(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $5.23(\mathrm{~s}, 1 \mathrm{H}), 5.00(\mathrm{~s}, 4 \mathrm{H}), 3.83(\mathrm{~s}, 6 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta 160.14,159.45,155.45,139.65,130.02,129.33,128.95,128.73,127.99,126.52,115.63$, 114.03, 105.57, 101.24, 69.92, 55.32.

IR (Neat): 3608, 2953, 1611, 1580, 1514, 1436, 1245, 1144, $1030 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{30} \mathrm{H}_{29} \mathrm{O}_{5}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$469.2010, found 469.2007.
${ }^{1} \mathrm{H}$ NMR, 500 MHz , Chloroform- $d$, Stilbene 25


${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Stilbene $\mathbf{2 5}$



## (26) (E)-4-(3,5-bis((4-methoxybenzyl)oxy)styryl)-2-methoxyphenol

Commercially available 3,5-dihydroxybenzyl alcohol ( $2.0 \mathrm{~g}, 14.3 \mathrm{mmol}$ ) was added to flask charged with potassium carbonate ( 2.25 equiv., 4.44 g ) and tetrabutylammonium iodide ( 0.2 equiv., 1.05 g ), and the solids were dissolved/suspended in acetone ( 42 mL ). To the stirring reaction mixture was added 4-methoxybenzyl chloride ( 2.20 equiv., 4.26 mL ), and the reaction was allowed to stir at room temperature for 16 hours. Upon completion, the reaction was diluted with ethyl acetate and poured into a separatory funnel containing deionized water. The layers were separated, and the aqueous layer was extracted with additional portions of ethyl acetate. The combined organic layer was washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure to afford the crude product, which was carried forward without further purification. A portion of the PMB-protected material ( $734 \mathrm{mg}, 1.93 \mathrm{mmol}$ ) was subjected to the general procedure using toluene as the solvent for the olefination with aldehyde C. Upon completion of the olefination, the reaction was cooled to $0{ }^{\circ} \mathrm{C}$, and TBAF ( 1.0 equiv., 1.0 M in THF, 1.93 mL ) was added dropwise. The reaction was allowed to stir for 30 min , at which point it was quenched and worked up following the general procedure. The product was purified by column chromatography ( $8 \%$ to $40 \%$ ethyl acetate in hexanes) to afford the desired product as a $\sim 3.6: 1 \mathrm{E}: \mathrm{Z}$ mixture ( $762 \mathrm{mg}, 79 \%$ yield).
$\mathrm{R}_{\mathrm{f}}=0.32$ (ethyl acetate/hexanes 3:7; UV)
E-isomer:
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.37(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H}$ ), $7.03(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.01(\mathrm{dd}, J=8.2,1.8 \mathrm{~Hz}, 1 \mathrm{H})$, $7.00(\mathrm{~d}, J=15.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H}), 6.91(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{~d}, J=15.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.52(\mathrm{t}, J=2.5 \mathrm{~Hz}$, 1 H ), 5.67 ( $\mathrm{s}, 1 \mathrm{H}$ ), $5.00(\mathrm{~s}, 4 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.83(\mathrm{~s}, 6 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Acetone- $d_{6}$ ) $\delta 160.30,159.58$, 146.83, 145.82, 139.70, 129.90, 129.40, 129.28, 129.06, 126.55, 120.71, 114.69, 114.13, 108.42, 105.63, 101.38, 70.01, 56.03, 55.42.

Z-isomer:
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.28(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 4 \mathrm{H}), 6.88(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 4 \mathrm{H}), 6.80(\mathrm{dd}, J=7.6,1.8 \mathrm{~Hz}, 1 \mathrm{H})$, $6.79(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.77(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.48(\mathrm{~d}, J=12.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{t}, J=2.3 \mathrm{~Hz}$, $1 \mathrm{H}), 6.44(\mathrm{~d}, J=12.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.57(\mathrm{~s}, 1 \mathrm{H}), 4.84(\mathrm{~s}, 4 \mathrm{H}), 3.81(\mathrm{~s}, 6 \mathrm{H}), 3.62(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Acetone- $d_{6}$ ) $\delta 159.95,159.51,146.08,145.08,139.74,130.53,129.90,128.61,124.41,122.87$, 121.77, 114.07, 112.30, 111.42, 107.93, 101.44, 69.88, 55.75, 55.42.

Mixture of isomers:
IR (Neat): 3522, 2954, 2869, 1592, 1454, 1353, 1287, 1212, 1194, 1157, 1149, 1056, 1037, 963, 828, 732, 695, $634 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{31} \mathrm{H}_{31} \mathrm{O}_{6}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right) 499.2115$, found 499.2111 .
${ }^{1} \mathrm{H}$ NMR, 700 MHz , Chloroform- $d$, Stilbene 26 ( $\sim 3.6: 1$ mixture of isomers)

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${ }^{1} \mathrm{H}$ NMR, 700 MHz , Chloroform- $d$, Stilbene 26 (Z isomer)

${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Stilbene 26 ( $\sim 3.6: 1$ mixture of isomers)



[^0]
## Cyclic Voltammetry Experimental Procedure:

For each stilbene substrate, two cyclic voltammetry experiments were conducted. The first was to measure oxidation of the substrate (green or blue curve), while the second was to measure oxidation in the presence of 2,6 -lutidine (orange curve). The stilbene substrate $(0.03 \mathrm{mmol})$ and the electrolyte ( $\mathrm{Bu}_{4} \mathrm{NPF}_{6}, 0.3 \mathrm{mmol}, 116 \mathrm{mg}$ ) were dissolved in acetonitrile ( 3 mL ). For the experiments with base, 2,6 -lutidine $(0.03 \mathrm{mmol}, 3.5 \mu \mathrm{~L})$ was added. The solution was transferred to a 5 -neck electrochemical cell, which was outfitted with a working electrode (glassy carbon, 3 mm diameter, surface area $=0.0707$ $\mathrm{cm}^{2}$ ), reference electrode ( $\mathrm{Ag} / \mathrm{AgCl}, 3 \mathrm{M} \mathrm{aq} . \mathrm{KCl}$ ), and counter/auxiliary electrode (platinum wire). The electrochemical cell was connected to the CH1620E electrochemical analyzer, and the potential was swept from 0.0 V to +1.0 V in two sweep segments at a scan rate of $100 \mathrm{mV} / \mathrm{s}$ to afford the observed cyclic voltammograms. It can be seen in the data below that direct oxidation of the electron rich substrates occurs between +0.8 to +1.0 V , however direct oxidation of electron deficient substrates occurs beyond +1.0 V . In the presence of 2,6 -lutidine, oxidation occurs below +0.6 V in all cases, suggesting that +0.6 V is a sufficient potential to attain the desired reactivity.

## Cyclic Voltammetry Data:



## Stilbene S3



## Stilbene S4




## Stilbene S7


——4-MeS-stilbene
_w/ 2,6-lutidine

Stilbene $\mathbf{S 8}$


- 4-OBn-stilbene
_- $\mathrm{w} / 2,6$-lutidine

Stilbene S9




Stilbene S10


Stilbene 11


Stilbene 12


## Stilbene S13



Stilbene 14


## General Dimerization Procedure ${ }^{3}$ :

The starting phenol ( 0.1 mmol ) was added to a reaction vial with $\mathrm{KPF}_{6}(74 \mathrm{mg}, 0.4 \mathrm{mmol})$ and $2,6-1 \mathrm{lutidine}(2.3 \mu \mathrm{~L}$, 0.02 mmol ) and dissolved in acetonitrile ( 8 mL ). Two pieces of $0.25 \times 2$-inch RVC panel ( 0.25 inch thickness) were cut. To each, a hole was made near one end, and copper wire was placed through the hole and wrapped around the top of each electrode. One end of the wire was left free in to connect to the alligator clips. These electrodes were carefully placed into the reaction vial along with the reference electrode ( $\mathrm{Ag} / \mathrm{AgCl}$ in 3 M KCl ) and a divider (see image). The alligator clips were connected such that the reference and working electrodes were adjacent to each other, while the counter electrode was opposite the divider. Care was taken to ensure the copper wire was not submerged in solvent, nor the active components of the alligator clips touching each other. The reaction was stirred at 750 rpm for $1-2 \mathrm{~h}$ at a constant voltage of 0.6 V . A chronoampergram was recorded to follow the course of the reaction. Upon completion of the reaction, the electrodes were removed and rinsed into a collection flask with DCM $(\sim 40 \mathrm{~mL})$. The contents of the reaction vial were also rinsed into the collection flask. The solvent was removed on the rotovap, the crude material was resuspended in DCM, and the electrolyte was filtered away with a plug of Celite. The filtrate was then concentrated to afford the product, which did not require further purification. The diastereomeric ratios were determined by integration of the aryl protons on the quinone methide. When these were overlapping with other aryl signals, the $\delta$-protons of the quinone methide were integrated to determine dr.


Prior to electrolysis


After electrolysis

Monitoring Current vs. Time


Sample chronoamperogram


Reaction components

## Optimization and Scalability of Dimerization

## Table S1. Optimization of Electrochemical Dimerization



| Entry | Scale (mmol) | Additive (equiv.) | 2,6-lut. (equiv.) | Solvent (ratio) | Time (h) | Yield (\%) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 0.1 | Fc (0.1) | 2.0 | A | 0.5 | 94 |
| 2 | 0.1 | - | 2.0 | A | 0.5 | 99 |
| 3 | 0.1 | - | 0.2 | A | 0.5 | 99 |
| 4 | 0.1 | - | 0 | A | 0.5 | 0 |
| 5 | 0.3 | - | 0.2 | A | 1 | 90 |
| 6 | 0.3 | - | 0.2 | $\mathrm{~A}: \mathrm{B}(2: 1)$ | 1 | 95 |
| 7 | 0.9 | - | 0.2 | $\mathrm{~A}: \mathrm{B}(2: 1)$ | 5 | 99 |
| 8 | 3.6 | - | 0.2 | $\mathrm{~A}: \mathrm{B} \mathrm{(2:1)}$ | 12 | 95 |

${ }^{\text {a }}$ Isolated yields. $\mathrm{Fc}=$ ferrocene. $\mathrm{A}=\mathrm{MeCN} . \mathrm{B}=\mathrm{CH}_{2} \mathrm{Cl}_{2}$.
The investigation of this electrochemical dimerization strategy began with protected resveratrol analogue $\mathbf{1 0}$, mimicking the key transformation in our prior syntheses of pallidol and quadrangularin A. ${ }^{1}$ Initial attempts sought to recapitulate the ferrocenium-mediated conditions, positing that the well-known, low and reversible oxidation potential of ferrocene (E1/2 $(\mathrm{Fc}+/ \mathrm{Fc})=0.40 \mathrm{~V}$ vs SCE$)$ would allow for the use of sub-stoichiometric equivalents of the oxidant. ${ }^{4}$ Gratifyingly, an excellent yield was achieved using only $10 \mathrm{~mol} \%$ of ferrocene and 2 equivalents of 2,6-lutidine (Table S1, Entry 1). It was speculated that the oxidative dimerization of $\mathbf{1 0}$ could occur without the need for a metal catalyst, given the apparent selectivity window that existed between the oxidation potential of the starting material ( $\mathrm{Ep} / 2=+0.81 \mathrm{~V} \mathrm{vs} \mathrm{Ag} / \mathrm{AgCl})$ and the desired product $11(\mathrm{Ep} / 2=+1.23 \mathrm{vs} \mathrm{Ag} / \mathrm{AgCl}$, see page S 37 for voltammetry data). This selectivity could be further enhanced by lowering the oxidation potential through the addition of base, an effect demonstrated by Corduneanu et al. in their investigation of the effect of pH on the oxidation potential of resveratrol. ${ }^{5}$ Indeed, we observed a dramatic decrease in the oxidation potential of $\mathbf{1 0}$ upon adding 2,6-lutidine $(\mathrm{Ep} / 2=+0.44 \mathrm{~V}(\mathrm{vs} \mathrm{Ag} / \mathrm{AgCl})$ ), while the bis-quinone methide $(\mathrm{BQM})$ product 11 was unaffected. Employing these metal-free conditions for the bulk electro-chemical processing of monomer $\mathbf{1 0}$ proved highly effective (Table S1, entry 2), generating the desired product in near quantitative yield after just 30 min at +0.6 V . Furthermore, it was observed that only a sub-stoichiometric amount of base was needed (Table S1, Entry 3), and the reaction did not occur at the same set potential when base was excluded (Table S1, Entry 4). Upon increasing the scale of the reaction (from 0.1 mmol to 0.3 mmol ), deposition of insoluble dimeric products on the surface of the electrodes was found to inhibit the reaction (Table S1, Entry 5), an issue that was ameliorated through the addition of dichloromethane as a co-solvent. Importantly, these optimized conditions proved readily-scalable (Table S1, Entries 6-8), including operation on multi-gram scale while still maintaining high yields ( $\geq 95 \%$ ). In addition to the efficiency, the operational simplicity of this method is viewed as a definitive benefit, as it is carried out on the benchtop, open to atmosphere, and only requires the addition of an electrolyte and a single, readily-available reagent. Furthermore, column chromatography is generally not required, increasing the attractiveness of this method for multi-step synthetic routes.

(5) 4,4'-((2R,3R)-2,3-bis(3,5-bis(benzyloxy)phenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1one)

Stilbene $\mathbf{S 0}$ ( $0.1 \mathrm{mmol}, 52.1 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide 5 $\left(51.7 \mathrm{mg}, 99 \%\right.$ yield, $4: 3 \mathrm{dr}$ ). The ${ }^{1} \mathrm{H}$ NMR spectrum was identical to the previous report for this compound. ${ }^{1}$
${ }^{1} \mathrm{H}$ NMR $(\mathrm{CDCl} 3,500 \mathrm{MHz}): \delta 7.40-7.30(\mathrm{~m}, 20 \mathrm{H}), \beta-\mathrm{H}$ 's of quinone methides: 7.12 (major diastereomer, $\mathrm{d}, J=2.2 \mathrm{~Hz}$, 2 H ), 7.02 (minor diastereomer, $\mathrm{d}, J=2.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.82 (minor diastereomer, $\mathrm{d}, J=2.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.72 (major diastereomer, $\mathrm{d}, J=2.0 \mathrm{~Hz}, 2 \mathrm{H}$ ); $\delta$-H's of quinone methides: $6.41-6.37$ (minor diastereomer, m, 2H), 6.33-6.29 (major diastereomer, $\mathrm{m}, 2 \mathrm{H}$ ); 6.48 (major diastereomer, $\mathrm{t}, J=2.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.47 (minor diastereomer, $\mathrm{t}, J=2.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.45 (major diastereomer, $\mathrm{d}, J=2.1 \mathrm{~Hz}, 4 \mathrm{H}$ ), 6.38 (minor diastereomer, d, $J=2.2 \mathrm{~Hz}, 4 \mathrm{H}$ ), 4.96 (major diastereomer, d, $J=11.5 \mathrm{~Hz}, 4 \mathrm{H}$ ), 4.94 (major diastereomer, $\mathrm{d}, ~ J=11.5 \mathrm{~Hz}, 4 \mathrm{H}$ ), 4.91 (minor diastereomer, d, $J=11.5 \mathrm{~Hz}, 4 \mathrm{H}$ ), 4.89 (minor diastereomer, d, $J=11.5 \mathrm{~Hz}$, 4 H ), 4.28 ( m , overlap, $\mathrm{sp}^{3}$ methines of both diastereomers, 4 H ), 1.26 (minor diastereomer $t \mathrm{Bu}$ 's, $\mathrm{s}, 18 \mathrm{H}$ ), 1.24 (major diastereomer $t \mathrm{Bu}$ 's, s, 36H), 1.23 (minor diastereomer $t \mathrm{Bu}$ 's, s, 18H).



## (6) 4,4'-((2R,3R)-2,3-bis(3,5-dimethoxyphenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1-one)

Stilbene $\mathbf{S 1}(0.1 \mathrm{mmol}, 36.9 \mathrm{mg})$ was subjected to the general dimerization procedure, affording bis-quinone methide $\mathbf{6}$ ( $36.5 \mathrm{mg}, 99 \%$ yield, $3: 2 \mathrm{dr}$ ). The ${ }^{1} \mathrm{H}$ NMR spectrum was identical to the previous report for this compound. ${ }^{1}$
${ }^{1} \mathrm{H}$ NMR (CDCl3, 700 MHz ): $\delta \beta$-H's of quinone methides: 7.13 (major diastereomer, d, $J=1.9 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.09 (minor diastereomer, d, $J=2.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.82 (minor diastereomer, d, $J=2.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.71 (major diastereomer, d, $J=2.2 \mathrm{~Hz}$, 2 H ); $\delta$-H's of quinone methides: 6.43 (minor diastereomer, m, 2H), 6.33 (major diastereomer, m, 2H); 6.35 (Ar-H major diastereomer, d, $J=2.1 \mathrm{~Hz}, 4 \mathrm{H}$ ), 6.31 (Ar-H major diastereomer, $\mathrm{t}, J=2.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.29-6.27 (Ar-H's minor diastereomer, overlap, 6 H ), $4.34-4.30$ (minor diastereomer sp ${ }^{3}$ methines, $\mathrm{m}, 2 \mathrm{H}$ ), $4.30-4.26$ (major diastereomer sp ${ }^{3}$ methines, $\mathrm{m}, 2 \mathrm{H}$ ), 3.74 (major diastereomer-OMe's, s, 12H), 3.70 (minor diastereomer-OMe's, s, 12H), 1.25 (minor diastereomer $t \mathrm{Bu}$ 's, s , 18 H ), 1.24 (major diastereomer $t \mathrm{Bu}$ 's, $\mathrm{s}, 18 \mathrm{H}$ ), 1.23 (minor diastereomer $t \mathrm{Bu}$ 's, s, 18H), 1.22 (major diastereomer $t \mathrm{Bu}$ 's, s , 18 H ).


(7) 4,4'-((2R,3R)-2,3-bis(3,5-bis((4-methoxybenzyl)oxy)phenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1-one)

Stilbene $\mathbf{S 3}$ ( $0.1 \mathrm{mmol}, 58.1 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide $\mathbf{7}$ ( $57.8 \mathrm{mg}, 99 \%$ yield, 5:4 dr).
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.30(\mathrm{~m}, 16 \mathrm{H}), 7.13(\beta-\mathrm{H}$, major diastereomer, $\mathrm{d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.03(\beta-\mathrm{H}$, minor diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{~m}, 16 \mathrm{H}), 6.83(\beta-\mathrm{H}$, minor diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.74(\beta-\mathrm{H}$, major diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), $6.46(\mathrm{~m}, 8 \mathrm{H}), 6.43(\delta-\mathrm{H}$, minor diastereomer, d, $J=8.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.39 (minor diastereomer, d, $J=2.2 \mathrm{~Hz}, 4 \mathrm{H}$ ), $6.35(\delta-\mathrm{H}$, major diastereomer, dd, $J=7.2,2.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.87(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 8 \mathrm{H}), 4.83(\mathrm{q}, J=10.6 \mathrm{~Hz}$, 8 H ), $4.30\left(\mathrm{~m}, \mathrm{sp}^{3}\right.$ methines of both diastereomers, 4 H ), $3.81(\mathrm{~s}, 12 \mathrm{H}), 3.80(\mathrm{~s}, 12 \mathrm{H}),{ }^{\mathrm{t}} \mathrm{Bu}$ signals: $1.26(\mathrm{~s}, 18 \mathrm{H}), 1.25(\mathrm{~s}$, 18 H ), 1.24 ( $\mathrm{s}, 18 \mathrm{H}$ ), 1.23 ( $\mathrm{s}, 18 \mathrm{H}$ ).
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta$ 186.66, 186.61, 160.44, 160.31, 159.72, 159.71, 149.02, 148.93, 147.58, 147.27, $145.28,143.91,143.12,142.79,134.86,134.71,133.08,132.03,129.55,129.53,128.59,128.54,126.13,125.97,114.17$, $107.95,107.75,100.33,100.31,70.15,70.13,55.44,55.42,51.78,51.14,35.49,35.47,35.03,34.97,29.62,29.56,29.54$.

IR (Neat): 2955, 2929, 1611, 1577, 1515, 1439, 1245, 1147, 1046, 1026, 967, $858 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{76} \mathrm{H}_{87} \mathrm{O}_{10}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$1159.6294, found 1159.6278 .


S47
${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Dimer 7





## (10) 4,4'-((2R,3R)-2,3-diphenylbutane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1-one)

Stilbene $\mathbf{S 4}$ ( $0.1 \mathrm{mmol}, 33 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide $\mathbf{1 0}$ ( $32.6 \mathrm{mg}, 99 \%$ yield, $4: 3 \mathrm{dr}$ ).
${ }^{1} \mathrm{H}$ NMR $(500 \mathrm{MHz}$, Chloroform- $d$ ) $\delta 7.31(\mathrm{t}, J=7.5 \mathrm{~Hz}, 4 \mathrm{H}), 7.25-7.08(\mathrm{~m}, 20 \mathrm{H}), \beta-\mathrm{H}$ 's of quinone methides: 6.82 (minor diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.70 (major diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), $\delta$-H's of quinone methides: 6.52 6.47 (minor diastereomer, $\mathrm{m}, 2 \mathrm{H}$ ), $6.42-6.35$ (major diastereomers, $\mathrm{m}, 2 \mathrm{H}$ ), $4.42\left(\mathrm{~m}, \mathrm{sp}^{3}\right.$ methines of both diastereomers, 4 H ), ${ }^{\text {'Bu signals: }} 1.25$ (s, 18 H ), 1.25 ( $\mathrm{s}, 18 \mathrm{H}$ ), 1.23 ( $\mathrm{s}, 18 \mathrm{H}$ ), 1.22 ( $\left.\mathrm{s}, 18 \mathrm{H}\right)$.
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta 186.59$, 186.55, 149.11, 148.91, 147.61, 147.32, 145.47, 144.98, 141.14, 140.63, $134.67,132.76,132.02,129.17,128.97,128.47,128.27,127.55,127.23,126.07,125.94,51.68,51.31,35.52,35.49,35.03$, 34.95, 29.60, 29.56, 29.52.

IR (Neat): 2951, 2915, 1614, 1577, 1453, 1358, 1257, 917, 883, 759, $697 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{NaC}_{44} \mathrm{H}_{54} \mathrm{O}_{2}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$637.4016, found 637.4012.

${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Dimer 10


(11) 4,4'-((2R,3R)-2,3-bis(benzo[d][1,3]dioxol-5-yl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1one)

Stilbene $\mathbf{S 5}$ ( $0.1 \mathrm{mmol}, 35.2 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide $\mathbf{1 1}$ ( 34.7 mg , $99 \%$ yield, 3:2 dr).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.16$ (minor diastereomer, $\mathrm{d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.10 (major diastereomer, d, $J=2.4$ $\mathrm{Hz}, 2 \mathrm{H}$ ), 6.79 (minor diastereomer, d, $J=2.3 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.73 (minor diastereomer, d, $J=7.9 \mathrm{~Hz}, 2 \mathrm{H}$ ), $6.69(\mathrm{~d}, J=1.6 \mathrm{~Hz}$, $4 \mathrm{H}), 6.68-6.64(\mathrm{~m}, 4 \mathrm{H}), 6.62(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.55(\mathrm{dd}, J=8.0,1.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.36(\delta-\mathrm{H}$, minor diastereomer, dt, $J=$ $8.7,4.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.32-6.26(\delta-\mathrm{H}$, minor diastereomer, $\mathrm{m}, 2 \mathrm{H}), 5.93(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 4 \mathrm{H}), 5.91(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 4 \mathrm{H}), 4.30-$ 4.21 ( $\mathrm{sp}^{3}$ methines of both diastereomers, m, 4H), 'Bu signals: $1.26(\mathrm{~s}, 18 \mathrm{H}), 1.25(\mathrm{~s}, 36 \mathrm{H}), 1.23(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta 186.58$, 186.56, 149.21, 148.97, 148.36, 148.15, 147.67, 147.36, 146.94, 146.67, $145.39,144.85,134.91,134.68,134.63,134.52$, 132.60, 131.90, 125.96, 125.93, 121.74, 121.61, 108.80, 108.70, 108.32, 108.27, 101.37, 101.29, 51.25, 50.95, 35.55, 35.50, 35.04, 34.98, 29.63, 29.58, 29.55, 29.53.

IR (Neat): 2954, 2914, 2361, 2336, 1616, 1539, 1362, 1243, 1034, 924, $808 \mathrm{~cm}^{-1}$;
HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calculated for $\mathrm{NaC}_{46} \mathrm{H}_{54} \mathrm{O}_{6}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right) 725.3813$, found 725.3817 .

${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Dimer 11


(12) 4,4'-((2R,3R)-2,3-bis(3,4-bis(benzyloxy)phenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1one)

Stilbene $\mathbf{S 6}(0.1 \mathrm{mmol}, 52.1 \mathrm{mg})$ was subjected to the general dimerization procedure, affording bis-quinone methide $\mathbf{1 2}$ (48.9 mg, 94\% yield, 5:4 dr).
${ }^{1} \mathrm{H}$ NMR $(500 \mathrm{MHz}$, Chloroform- $d$ ) $\delta 7.40(\mathrm{t}, J=6.5 \mathrm{~Hz}, 4 \mathrm{H}), 7.37-7.27(\mathrm{~m}, 16 \mathrm{H}), 7.10$ (major diastereomer, d, $J=2.3$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 7.01 (minor diastereomer, d, $J=2.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.86 (major diastereomer, d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.77 (s, 1 H ), 6.76 (d, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.68(\mathrm{~m}, 2 \mathrm{H}), 6.67-6.63(\mathrm{~m}, 2 \mathrm{H}), 6.56$ (minor diastereomer, $\mathrm{d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.49$ (minor diastereomer, dd, $J=8.3,2.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.29 (minor diastereomer, d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.22 (major diastereomer, d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.11 (major diastereomer, s, 2H), 5.08 (minor diastereomer, s, 2 H ), 5.04 (major diastereomer, d, $J=11.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.00 (major diastereomer, d, $J=11.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.98 (minor diastereomer, s, 2 H ), 4.23 (major diastereomer, d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.12 (minor diastereomer, $\mathrm{d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.26(\mathrm{~s}, 9 \mathrm{H}), 1.25(\mathrm{~s}, 9 \mathrm{H}), 1.23(\mathrm{~s}, 9 \mathrm{H}), 1.22(\mathrm{~s}, 9 \mathrm{H})$.
${ }^{13}$ C NMR (126 MHz, Chloroform-d) $\delta 186.53,186.49,149.08,149.00,148.99,148.83,148.49,148.26,147.54,147.39$, $145.58,144.90,137.23,137.17,137.12,137.06,136.94,134.75,134.66,133.98,133.33,132.52,131.72,128.63,128.61$, $128.03,128.01,128.00,127.98,127.47,127.41,127.35$ (2C), 126.10, 125.86, 121.69, 121.55, 120.41, 115.98, 115.69, $115.06,114.95,71.81,71.75,71.31,71.27,51.17,50.76,35.50,35.47,35.02,34.99,29.61$ (2C), 29.56, 29.54.

IR (Neat): 2960, 2867, 1616, 1510, 1453, 1358, 1262, 1136, 1021, 734, $697 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{NaC}_{72} \mathrm{H}_{78} \mathrm{O}_{6}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$1061.5691, found 1061.5672.

${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Dimer 12

$$
\begin{aligned}
& \checkmark \underbrace{V}_{\text {- }}
\end{aligned}
$$




## (13) 4,4'-((2R,3R)-2,3-bis(4-(methylthio)phenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1-one)

Stilbene $\mathbf{S 7}$ ( $0.1 \mathrm{mmol}, 35.5 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide $\mathbf{1 3}$ ( $32.9 \mathrm{mg}, 93 \%$ yield, 4:3 dr).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.19$ (major diastereomer, d, $J=8.0 \mathrm{~Hz}, 4 \mathrm{H}$ ), 7.12 (m, 12H), 7.02 (minor diastereomer, $\mathrm{d}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}$ ), $\beta$-H's of quinone methides: 6.80 (minor diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.70 (major diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ) , $\delta$-H's of quinone methides: 6.42 (minor diastereomer, dd, $J=7.2,2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), $6.36-6.29$ (major diastereomer, $\mathrm{m}, 2 \mathrm{H}$ ), $4.43-4.30\left(\mathrm{sp}^{3}\right.$ methines of both diastereomers, $\left.\mathrm{m}, 4 \mathrm{H}\right), 2.44(\mathrm{~s}, 3 \mathrm{H}), 2.42(\mathrm{~s}, 3 \mathrm{H}),{ }^{t} \mathrm{Bu}$ signals: 1.25 ( $\mathrm{s}, 18 \mathrm{H}$ ), $1.25(\mathrm{~s}, 18 \mathrm{H}), 1.24(\mathrm{~s}, 18 \mathrm{H}), 1.22(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta$ 186.56, 186.51, 149.25, 149.13, 147.71, 147.47, 145.03, 144.63, 137.93, 137.68, $137.59,137.21,134.62,132.65,132.02,128.88,128.70,127.11,126.98,125.92,125.79,50.91,50.64,35.54,35.53,35.05$, 34.99, 29.60, 29.58, 29.55, 29.52, 15.83, 15.80.

IR (Neat): 2960, 2906, 1698, 1591, 1437, 1358, 1262, 1096, 1023, 846, 815, $740 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{NaC}_{46} \mathrm{H}_{58} \mathrm{O}_{2} \mathrm{~S}_{2}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right) 729.3770$, found 729.3773 .

${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Dimer 13


(14) 4,4'-((2R,3R)-2,3-bis(4-(benzyloxy)phenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1-one)

Stilbene $\mathbf{S 8}(0.1 \mathrm{mmol}, 41.5 \mathrm{mg})$ was subjected to the general dimerization procedure, affording bis-quinone methide $\mathbf{1 4}$ ( $39.3 \mathrm{mg}, 95 \%$ yield, $3: 2 \mathrm{dr}$ ).
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.43-7.35(\mathrm{~m}, 16 \mathrm{H}), 7.34-7.31(\mathrm{~m}, 4 \mathrm{H}), 7.14$ ( $\beta-\mathrm{H}$, minor diastereomer, d, $J=2.4$ $\mathrm{Hz}, 2 \mathrm{H}$ ), 7.12 ( $\beta-\mathrm{H}$, major diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.09 (major diastereomer, d, $J=8.7 \mathrm{~Hz}, 4 \mathrm{H}$ ), 7.02 (minor diastereomer, d, $J=8.7 \mathrm{~Hz}, 4 \mathrm{H}$ ), 6.91 (major diastereomer, d, $J=8.7 \mathrm{~Hz}, 4 \mathrm{H}$ ), 6.85 (minor diastereomer, d, $J=8.7 \mathrm{~Hz}$, $4 \mathrm{H}), 6.80(\beta-\mathrm{H}$, minor diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.70(\beta-\mathrm{H}$, major diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.44(\delta-\mathrm{H}$, minor diastereomer, d, $J=8.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), $6.35(\delta-\mathrm{H}$, major diastereomer, d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 5.01 (major diastereomer, s, 4H), 4.99 (minor diastereomer, $\mathrm{s}, 4 \mathrm{H}$ ), 4.31 ( $\mathrm{sp}^{3}$ methines of both diastereomers, $\mathrm{m}, 4 \mathrm{H}$ ), ${ }^{\dagger} \mathrm{Bu}$ signals: $1.25(\mathrm{~s}, 36 \mathrm{H}), 1.24(\mathrm{~s}, 18 \mathrm{H})$, 1.23 (s, 18H).
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta$ 186.62, 186.58, 158.08, 157.85, 149.02, 148.88, 147.52, 147.25, 146.02, 145.41, $136.92,136.87,134.80,134.75,133.43,132.95,132.40,131.70,129.49,129.31,128.75,128.74,128.21,128.18,127.64$, $127.48,126.11,126.03,123.30,115.44,115.30,70.22,70.16,50.87,50.57,35.51,35.02,34.97,34.53,30.45,29.86,29.62$, 29.60, 29.56, 29.54.

IR (Neat): 2954, 2918, 1608, 1569, 1510, 1358, 1245, 1018, 824, 740, $695 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{58} \mathrm{H}_{67} \mathrm{O}_{4}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right) 827.5034$, found 827.5040.


S57
${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Dimer 14



## (15) Di-tert-butyl (((2R,3R)-1,4-bis(3,5-di-tert-butyl-4-oxocyclohexa-2,5-dien-1-ylidene)butane-2,3-diyl)bis(4,1phenylene))dicarbamate

Stilbene $\mathbf{S 9}$ ( $0.1 \mathrm{mmol}, 42.4 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide $\mathbf{1 5}$ ( $40.7 \mathrm{mg}, 95 \%$ yield, $3: 2 \mathrm{dr}$ ).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.31$ (major diastereomer, d, $J=8.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.23 (minor diastereomer, d, $J=8.1$ $\mathrm{Hz}, 2 \mathrm{H}$ ), 7.16 (minor diastereomer, d, $J=2.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.13 (major diastereomer, d, $J=2.3 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.09 (major diastereomer, d, $J=8.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.00 (minor diastereomer, d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.79 (minor diastereomer, d, $J=2.3 \mathrm{~Hz}$, 1 H ), 6.69 (major diastereomer, d, $J=2.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.45 (major diastereomer, s, 1 H ), 6.42 (minor diastereomer, d, $J=9.2$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 6.40 (minor diastereomer, s, 1H), 6.34 (major diastereomer, d, $J=8.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.35 (major diastereomer, dd, $J=$ $7.3,2.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.33 (minor diastereomer, dd, $J=7.4,2.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 1.50 (major diastereomer, $\mathrm{s}, 9 \mathrm{H}$ ), 1.49 (minor diastereomer, $\mathrm{s}, 9 \mathrm{H}$ ), 1.25 (minor diastereomer, $\mathrm{s}, 9 \mathrm{H}$ ), 1.24 (minor diastereomer, $\mathrm{s}, 9 \mathrm{H}$ ), 1.24 (major diastereomer, $\mathrm{s}, 9 \mathrm{H}$ ), 1.22 (major diastereomer, $\mathrm{s}, 9 \mathrm{H}$ ).
${ }^{13}$ C NMR (126 MHz, Chloroform- $d$ ) $\delta$ 186.60, 186.59, 152.74, 152.69, 149.10, 148.99, 147.56, 147.29, 145.64, 145.21, $137.68,137.43,135.41,134.98,134.78,134.71,132.46,131.81,128.99,128.86,126.04,125.94,118.95,118.79,80.88$, $80.80,50.86,50.59,35.51,35.50,35.02,34.97,29.60,29.58,29.55,29.52,28.45$ (2C).

IR (Neat): 3333, 2954, 2864, 1715, 1611, 1521, 1361, 1231, 1153, 1049, $818 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{NaC}_{54} \mathrm{H}_{72} \mathrm{~N}_{2} \mathrm{O}_{6}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$867.5283, found 867.5276.

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Dimer 15


(16) 4,4'-((2R,3R)-2,3-bis(2-nitrophenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1-one)

Stilbene $\mathbf{S 1 0}(0.1 \mathrm{mmol}, 35.3 \mathrm{mg})$ was subjected to the general dimerization procedure, affording bis-quinone methide $\mathbf{1 6}$ ( $32.8 \mathrm{mg}, 93 \%$ yield, 2:1 dr).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform-d) $\delta 7.80(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.64(\mathrm{~d}, J=4.3 \mathrm{~Hz}, 4 \mathrm{H}), 7.61-7.57(\mathrm{~m}, 2 \mathrm{H}), 7.56$ (dd, $J=$ $7.5,1.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.50(\mathrm{dd}, J=8.0,1.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.44(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.38(\mathrm{dt}, J=8.4,4.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.35-7.30(\mathrm{~m}, 2 \mathrm{H})$, $7.00(\beta-\mathrm{H}$, major diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.82(\beta-\mathrm{H}$, major diastereomer, d, $J=2.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.62(\beta-\mathrm{H}$, major diastereomer, d, $J=2.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.36(\delta-\mathrm{H}$, minor diastereomer, dd, $J=6.5,2.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.25(\delta-\mathrm{H}$, major diastereomer, dd, $J=7.1,2.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.49-5.41(\mathrm{~m}, 4 \mathrm{H})$, ${ }^{\text {'Bu }}$ signals: 1.27 ( $\left.\mathrm{s}, 18 \mathrm{H}\right), 1.25(\mathrm{~s}, 18 \mathrm{H}), 1.20(\mathrm{~s}, 18 \mathrm{H}), 1.20(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta$ 186.56, 186.47, 150.22, 149.82, 149.71, 148.86, 148.37, 148.00, 141.92, 140.99, $135.23,135.07,134.52,134.07,133.99,133.95,133.91,133.89,130.09,129.76,128.54,128.45,126.22,125.35,125.16$, $125.05,44.58,44.36,35.76,35.53,35.12,35.01,29.66,29.55,29.50,29.48$.

IR (Neat): 2956, 2917, 1616, 1520, 1345, 1255, 918, 886, 822, 782, $729 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{44} \mathrm{H}_{53} \mathrm{~N}_{2} \mathrm{O}_{6}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right) 705.3898$, found 705.3893.




(17) 4,4'-((2R,3R)-2,3-bis(2-(trifluoromethyl)phenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1one)

Stilbene $\mathbf{S 1 1}$ ( $0.1 \mathrm{mmol}, 37.6 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide $\mathbf{1 7}$ ( $36.8 \mathrm{mg}, 98 \%$ yield, 5:1 dr).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.73$ (major diastereomer, d, $J=7.9 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.62 (major diastereomer, d, $J=8.0 \mathrm{~Hz}$, 2 H ), 7.59 (major diastereomer, $\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.53 (minor diastereomer, d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.42 (minor diastereomer, $\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.32 (major diastereomer, $\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.24 (minor diastereomer, $\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}$ ), $\beta-\mathrm{H}$ 's of quinone methides: 7.17 (minor diastereomer, d, $J=2.3 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.12 (major diastereomer, d, $J=2.5 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.70 (minor diastereomer, d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.58 (major diastereomer, d, $J=2.3 \mathrm{~Hz}, 2 \mathrm{H}$ ), $\delta$-H's of quinone methides: 6.25 (major diastereomer, d, $J=8.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.20 (minor diastereomer, d, $J=8.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 5.03 ( $\mathrm{sp}^{3}$ methines of both diastereomers, $\mathrm{m}, 4 \mathrm{H}$ ), ${ }^{\dagger} \mathrm{Bu}$ signals: 1.27 (minor diastereomer, $\mathrm{s}, 18 \mathrm{H}$ ), 1.25 (major diastereomer, $\mathrm{s}, 18 \mathrm{H}$ ), 1.20 (minor diastereomer, s, 18H), 1.17 (major diastereomer, s, 18H),
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta 186.54$, 186.41, 149.07, 148.80, 148.03, 147.56, 143.23, 143.05, 138.97, 138.65, $134.40,134.06,132.74,132.56,132.40,132.15,129.51,129.21,128.50$ (major diastereomer, $\mathrm{q}, \mathrm{J}=28 \mathrm{~Hz}$ ), 128.44 (minor diastereomer, q, $J=28 \mathrm{~Hz}$ ), 127.69, 127.29, 126.80 (major diastereomer, q, $J=5.3 \mathrm{~Hz}$ ), 126.62 (minor diastereomer, $\mathrm{q}, ~ J$ $=5.3 \mathrm{~Hz}$ ), 126.17, 125.81, 124.50 (major diastereomer, $\mathrm{q}, J=273 \mathrm{~Hz}$ ), 124.43 (minor diastereomer, q, $J=273 \mathrm{~Hz}$ ), 46.40, 45.18, 35.48, 35.00, 34.91, 29.70, 29.51, 29.46.

IR (Neat): 3005, 2957, 2918, 1614, 1569, 1453, 1363, 1310, 1155, 1111, 1032, 931, $765 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{NaC}_{46} \mathrm{H}_{52} \mathrm{~F}_{6} \mathrm{O}_{2}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right) 773.3764$, found 773.3772 .


S63
${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Dimer 17


(18) 4,4'-((2R,3R)-2,3-bis(3-(trifluoromethyl)phenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1one)

Stilbene $\mathbf{S 1 2}$ ( $0.1 \mathrm{mmol}, 37.6 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide 18 ( $36.4 \mathrm{mg}, 97 \%$ yield, $4: 3 \mathrm{dr}$ ).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.52$ (major diastereomer, $\mathrm{d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.46(\mathrm{~m}, 4 \mathrm{H}), 7.43-7.35(\mathrm{~m}, 6 \mathrm{H}), 7.32$ (minor diastereomer, d, $J=7.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.26 (minor diastereomer, $\mathrm{s}, 2 \mathrm{H}$ ), $\beta-\mathrm{H}$ 's of quinone methides: 7.01 (major diastereomer, $\mathrm{d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.98 (minor diastereomer, $\mathrm{d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.85 (minor diastereomer , $\mathrm{d}, J=2.4 \mathrm{~Hz}$, 2 H ), 6.72 (major diastereomer, $\mathrm{d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), $\delta$-H's of quinone methides: 6.45 (minor diastereomer, d, $J=9.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.30 (major diastereomer, $\mathrm{d}, J=9.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.46\left(\mathrm{sp}^{3}\right.$ methine of major diastereomer, $\left.\mathrm{d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}\right), 4.42\left(\mathrm{sp}^{3}\right.$ methine of minor diastereomer, $\mathrm{d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), ${ }^{t} \mathrm{Bu}$ signals: 1.26 (minor diastereomer, $\mathrm{s}, 18 \mathrm{H}$ ), 1.22 (major diastereomer, $\mathrm{s}, 18 \mathrm{H}$ ), 1.22 (major diastereomer, $\mathrm{s}, 18 \mathrm{H}$ ), 1.21 (minor diastereomer, $\mathrm{s}, 18 \mathrm{H}$ ),
${ }^{13}$ C NMR ( 126 MHz , Chloroform- $d$ ) $\delta 186.41,186.34,149.80,148.26,148.07,142.29,141.68,141.64,141.04,134.15$, $134.12,133.92,132.96,132.29,132.21,131.59,131.52,131.51(\mathrm{q}, J=32 \mathrm{~Hz}), 131.39(\mathrm{q}, J=32 \mathrm{~Hz}), 129.81,129.65$, $128.70,128.61,125.33,125.31(\mathrm{q}, J=3.8 \mathrm{~Hz}), 125.09(\mathrm{q}, J=3.8 \mathrm{~Hz}), 125.01,124.65(\mathrm{q}, J=3.8 \mathrm{~Hz}), 124.34(\mathrm{q}, J=3.8$ $\mathrm{Hz}), 123.90(\mathrm{q}, J=271 \mathrm{~Hz}), 123.83(\mathrm{q}, J=271 \mathrm{~Hz}), 51.60,51.02,35.54,35.12,35.04,29.55,29.52$, 29.48.

IR (Neat): 2959, 1617, 1571, 1448, 1390, 1324, 1252, 1162, 1124, 1069, 883, $813 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{NaC}_{46} \mathrm{H}_{52} \mathrm{~F}_{6} \mathrm{O}_{2}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right) 773.3764$, found 773.3762.


S65
${ }^{13}$ C NMR, 126 MHz , Chloroform- $d$, Dimer 18

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(19) 4,4'-((2R,3R)-2,3-bis(4-nitrophenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1-one)

Stilbene $\mathbf{S 1 3}$ ( $0.1 \mathrm{mmol}, 35.3 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide 19 ( $32.5 \mathrm{mg}, 92 \%$ yield, $1: 1 \mathrm{dr}$ ).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 8.22(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H}), 8.12(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H}), 7.39(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 4 \mathrm{H}), 7.27(\mathrm{~d}, J$ $=8.4 \mathrm{~Hz}, 4 \mathrm{H}), \beta-\mathrm{H}$ 's of quinone methides: $7.06(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.00(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.81(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.68$ (d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), $\delta$-H's of quinone methides: $6.36(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.24(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, sp ${ }^{3}$ methines: $4.60-4.53$ $(\mathrm{m}, 2 \mathrm{H}), 4.53-4.46(\mathrm{~m}, 2 \mathrm{H}), 1.25(\mathrm{~s}, 18 \mathrm{H}), 1.24(\mathrm{~s}, 18 \mathrm{H}), 1.22(\mathrm{~s}, 18 \mathrm{H}), 1.22(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (126 MHz, Chloroform- $d$ ) $\delta 186.24$, 186.15, 150.31, 148.64, 148.54, 147.96, 147.45, 147.33, 147.19, 140.74, $140.57,134.16,133.86,133.74,133.46,129.26,128.99,124.98,124.63,124.58,124.51,51.16,50.57,35.67,35.64,35.17$, 35.11, 29.59, 29.55, 29.52, 29.48.

IR (Neat): 2953, 1613, 1518, 1344, 1254, 913, 857, $706 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{44} \mathrm{H}_{52} \mathrm{~N}_{2} \mathrm{O}_{6}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right) 727.3718$, found 727.3723.

${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Dimer 19


(20) 4,4'-((2R,3R)-2,3-bis(4-(trifluoromethyl)phenyl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1one)

Stilbene $\mathbf{S 1 4}$ ( $0.1 \mathrm{mmol}, 37.6 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide 20 ( $33.8 \mathrm{mg}, 90 \%$ yield, $1: 1 \mathrm{dr}$ ).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz, Chloroform- $d$ ) $\delta 7.60(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.51(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.31(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.22(\mathrm{~d}$, $J=8.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.07(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.03(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.70(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.46$ $-6.39(\mathrm{~m}, 2 \mathrm{H}), 6.33-6.26(\mathrm{~m}, 2 \mathrm{H}), 4.50(\mathrm{dd}, J=7.4,2.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.46(\mathrm{dt}, J=8.7,6.3 \mathrm{~Hz}, 2 \mathrm{H}), 1.25(\mathrm{~s}, 18 \mathrm{H}), 1.23(\mathrm{~s}$, $18 \mathrm{H}), 1.22(\mathrm{~s}, 36 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta 186.40,186.30,149.82,149.76,148.20,148.04,144.87,144.25,142.48,142.30$, $134.20,134.13,133.56,132.87,130.04(\mathrm{q}, ~ J=33 \mathrm{~Hz}), 129.80(\mathrm{q}, J=32 \mathrm{~Hz}), 128.85,128.54,126.21(\mathrm{q}, J=3.5 \mathrm{~Hz})$, $126.12(\mathrm{q}, J=3.5 \mathrm{~Hz}), 125.39,125.13,123.96(\mathrm{q}, J=273 \mathrm{~Hz}), 123.92(\mathrm{q}, J=273 \mathrm{~Hz}), 51.21,50.74,35.58,35.56,35.10$, 35.04, 29.54, 29.52, 29.51, 29.48.

IR (Neat): $2955,1612,1571,1361,1322,1165,1105,1067,883,834 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{NaC}_{46} \mathrm{H}_{52} \mathrm{~F}_{6} \mathrm{O}_{2}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right) 773.3764$, found 773.3769 .


S69
${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Dimer 20





## (8) 4,4'-((2R,3R)-2,3-bis(3,5-bis(benzyloxy)phenyl)butane-1,4-diylidene)bis(2,6-bis(trimethylsilyl)cyclohexa-2,5-dien-1-one)

Stilbene $\mathbf{S 2}$ ( $0.1 \mathrm{mmol}, 55 \mathrm{mg}$ ) was subjected to the general dimerization procedure, and the crude reaction mixture was purified by flash column chromatography ( 0 to $15 \%$ EtOAc in Hexanes) to afford bis-quinone methide $\mathbf{8}$ ( $35.8 \mathrm{mg}, 65 \%$ yield, 2:1 dr). The ${ }^{1} \mathrm{H}$ NMR spectrum was identical to the previous report for this compound. ${ }^{2}$
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl} 3,25^{\circ} \mathrm{C}$ ) $\delta: 7.47(\mathrm{~d}, \mathrm{~J}=2.5 \mathrm{~Hz}, 2 \mathrm{H}$, major), $7.45-7.28(\mathrm{~m}, 32 \mathrm{H}), 7.15(\mathrm{~d}, \mathrm{~J}=2.5 \mathrm{~Hz}, 2 \mathrm{H}$, minor), 7.04 (d, J = $2.5 \mathrm{~Hz}, 2 \mathrm{H}$, major), $6.50-6.47$ ( m , overlap, 4 H ), $6.45(\mathrm{~d}, \mathrm{~J}=2.2 \mathrm{~Hz}, 4 \mathrm{H}$, major), $6.38(\mathrm{~m}, 4 \mathrm{H}), 6.37(\mathrm{~d}, \mathrm{~J}=2.2$ $\mathrm{Hz}, 4 \mathrm{H}$, minor), 5.07 ( $\mathrm{d}, \mathrm{J}=9.3 \mathrm{~Hz}, 2 \mathrm{H}$, minor), 4.99 ( $\mathrm{d}, \mathrm{J}=9.3 \mathrm{~Hz}, 2 \mathrm{H}$, minor), 4.96 ( $\mathrm{s}, 8 \mathrm{H}$, major/minor overlap), 4.91 ( d , $\mathrm{J}=11.2 \mathrm{~Hz}, 2 \mathrm{H}$, major), 4.89 (d, J = $11.2 \mathrm{~Hz}, 2 \mathrm{H}$, major), $4.36-4.28$ ( m , overlap, 4H, major/minor), 0.20 ( $\mathrm{s}, 18 \mathrm{H}$, minor), 0.19 (s, 18H, minor), 0.18 ( $\mathrm{s}, 36 \mathrm{H}$, major)
${ }^{1} \mathrm{H}$ NMR, 500 MHz , Chloroform- $d$, Dimer $\mathbf{8}$

${ }^{1} \mathrm{H}$ NMR, 500 MHz , Chloroform- $d$, Compound $\mathbf{4 b}^{2}$



## (9) 4,4'-((2R,3R)-2,3-bis(3,5-bis(benzyloxy)phenyl)butane-1,4-diylidene)bis(2,6-bis(triethylsilyl)cyclohexa-2,5-dien-1-one)

Stilbene S15 ( $0.1 \mathrm{mmol}, 63.7 \mathrm{mg}$ ) was subjected to the general dimerization procedure, and the crude reaction mixture was purified by flash column chromatography ( 0 to $15 \%$ EtOAc in Hexanes) to afford bis-quinone methide 9 ( 43.4 mg , $68 \%$ yield, 2:1 dr).
${ }^{1} \mathrm{H}$ NMR $(700 \mathrm{MHz}$, Chloroform- $d) \delta 7.47(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.45(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 4 \mathrm{H}), 7.44(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.42-$ $7.28(\mathrm{~m}, 30 \mathrm{H}), 7.11(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.04(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.50-6.47(\mathrm{~m}, 4 \mathrm{H}), 6.47(\mathrm{~s}, 4 \mathrm{H}), 6.44(\mathrm{~d}, J=6.8 \mathrm{~Hz}$, $1 \mathrm{H}), 6.39(\mathrm{~s}, 3 \mathrm{H}), 6.38(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 3 \mathrm{H}), 5.08(\mathrm{~d}, J=3.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.00(\mathrm{~s}, 1 \mathrm{H}), 4.95(\mathrm{~s}, 6 \mathrm{H}), 4.92(\mathrm{~s}, 4 \mathrm{H}), 4.32(\mathrm{td}, J=$ $8.5,7.1,4.5 \mathrm{~Hz}, 2 \mathrm{H}), 0.88(\mathrm{q}, J=8.1 \mathrm{~Hz}, 46 \mathrm{H}), 0.76-0.70(\mathrm{~m}, 30 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta 193.51,150.13,149.94,145.83,144.64,142.70,142.56,141.40,141.15,141.12$, $141.10,139.41,139.01,136.56,136.49,132.42,131.56,128.79,128.77,128.33,128.32,127.73,127.69,107.93,107.86$, $100.47,70.38,70.37,51.55,51.04,7.74,7.71,7.70,7.66,3.22,3.17,3.15,3.11$.

IR (Neat): 2957, 2898, 1588, 1456, 1248, 1158, 1051, 841, 734, 695, $619 \mathrm{~cm}^{-1}$;

HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{80} \mathrm{H}_{103} \mathrm{O}_{6} \mathrm{Si}_{4}{ }^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$1271.6826, found 1271.6816.

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Dimer 9




## (23) 4,4'-((2S,3S)-2,3-bis((2S,3S)-6-(benzyloxy)-2-(4-(benzyloxy)phenyl)-3-(3,5-bis(benzyloxy)phenyl)-2,3-dihydrobenzofuran-4-yl)butane-1,4-diylidene)bis(2,6-di-tert-butylcyclohexa-2,5-dien-1-one)

Stilbene 21 ( $0.05 \mathrm{mmol}, 45 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide 23 (40 $\mathrm{mg}, 89 \%$ yield, $>19: 1 \mathrm{dr}$ ). The ${ }^{1} \mathrm{H}$ NMR spectrum was identical to the previous report for this compound. ${ }^{2}$
${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl3}, 50^{\circ} \mathrm{C}$ ) $\delta: 7.52-7.12(\mathrm{~m}, 40 \mathrm{H},-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5), 7.10(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 4 \mathrm{H}, \mathrm{C} 2 \mathrm{a}-\mathrm{H}), 7.01(\mathrm{~d}, J=$ $8.5 \mathrm{~Hz}, 4 \mathrm{H}, \mathrm{C} 3 \mathrm{a}-\mathrm{H}), 6.70(\mathrm{~s}, 2 \mathrm{H}, \mathrm{C} 14 \mathrm{~b}-\mathrm{H}), 6.42(\mathrm{~s}, 2 \mathrm{H}, \mathrm{C} 2 / 6 \mathrm{~b}-\mathrm{H}), 6.39$ (s, 2H, C12b-H), 6.37 (s, 2H, C12a-H), 6.23 (s, $2 \mathrm{H}, \mathrm{C} 2 / 6 \mathrm{~b}-\mathrm{H}), 6.16$ (s, 4H, C10a-H), 6.11 (dd, br, $J=7.3,9.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 7 \mathrm{~b}-\mathrm{H}), 5.11$ (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 7 \mathrm{a}-\mathrm{H}), 5.04$ (d, $J=12.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 4 \mathrm{a}-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5$ ), 5.02 (d, $J=12.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 4 \mathrm{a}-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5$ ), 4.83 (d, $J=10.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 13 \mathrm{~b}-$ OCH2C6H5), 4.77 (d, $J=10.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 13 \mathrm{~b}-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5$ ), 4.62 (d, $J=11.1 \mathrm{~Hz}, 4 \mathrm{H}, \mathrm{C} 11 \mathrm{a}-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5$ ), 4.56 (d, $J=$ $11.1 \mathrm{~Hz}, 4 \mathrm{H}, \mathrm{C} 11 \mathrm{a}-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5$ ), 4.06 (dd, br, $J=7.9,9.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 8 \mathrm{~b}-\mathrm{H}$ ), 3.87 (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 8 \mathrm{a}-\mathrm{H}$ ), 1.17 (s, 18H, $\mathrm{C} 3 / 5 \mathrm{~b}-\mathrm{C}(\mathrm{CH} 3) 3), 1.00(\mathrm{~s}, 18 \mathrm{H}, \mathrm{C} 3 / 5 \mathrm{~b}-\mathrm{C}(\mathrm{CH} 3) 3)$.
${ }^{1}$ H NMR, 400 MHz , Chloroform- $d$, Dimer 23

${ }^{1} \mathrm{H}$ NMR, 500 MHz , Chloroform- $d$, Compound ( $\pm$ )-5a ${ }^{2}$


S74

(24) 4,4'-((2S,3S)-2,3-bis((2S,3S)-6-(benzyloxy)-2-(4-(benzyloxy)phenyl)-3-(3,5-bis(benzyloxy)phenyl)-2,3-dihydrobenzofuran-4-yl)butane-1,4-diylidene)bis(2,6-bis(trimethylsilyl)cyclohexa-2,5-dien-1-one)

Stilbene 22 ( $0.039 \mathrm{mmol}, 37 \mathrm{mg}$ ) was subjected to the general dimerization procedure, affording bis-quinone methide 24 ( $21 \mathrm{mg}, 57 \%$ yield, $>19: 1$ ). The ${ }^{1} \mathrm{H}$ NMR spectrum was identical to the previous report for this compound. ${ }^{2}$
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl3}, 25^{\circ} \mathrm{C}$ ) $\delta: 7.44-7.10(\mathrm{~m}, 40 \mathrm{H},-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5), 7.11(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 4 \mathrm{H}, \mathrm{C} 2 \mathrm{a}-\mathrm{H}), 7.03(\mathrm{~d}, J=$ $8.6 \mathrm{~Hz}, 4 \mathrm{H}, \mathrm{C} 3 \mathrm{a}-\mathrm{H}), 6.80(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 2 / 6 \mathrm{~b}-\mathrm{H}), 6.69(\mathrm{~s}, \mathrm{br}, 2 \mathrm{H}, \mathrm{C} 14 \mathrm{~b}-\mathrm{H}), 6.51(\mathrm{~d}, \mathrm{br}, J=2.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 2 / 6 \mathrm{~b}-\mathrm{H})$, $6.44(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 12 \mathrm{~b}-\mathrm{H}), 6.34(\mathrm{t}, J=2.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 12 \mathrm{a}-\mathrm{H}), 6.21(\mathrm{dd}, J=7.7,9.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 7 \mathrm{~b}-\mathrm{H}), 6.15(\mathrm{~d}, J=2.0$ $\mathrm{Hz}, 4 \mathrm{H}, \mathrm{C} 10 \mathrm{a}-\mathrm{H}), 5.14(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 7 \mathrm{a}-\mathrm{H}), 5.01(\mathrm{~s}, 4 \mathrm{H}, \mathrm{C} 4 \mathrm{a}-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5), 4.85$ (d, $J=11.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 13 \mathrm{~b}-$ OCH2C6H5), 4.77 (d, $J=11.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 13 \mathrm{~b}-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5$ ), 4.56 (d, $J=11.0 \mathrm{~Hz}, 4 \mathrm{H}, \mathrm{C} 11 \mathrm{a}-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5), 4.51$ (d, $J=$ $11.0 \mathrm{~Hz}, 4 \mathrm{H}, \mathrm{C} 11 \mathrm{a}-\mathrm{OCH} 2 \mathrm{C} 6 \mathrm{H} 5), 4.11(\mathrm{dd}, J=7.7,9.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 8 \mathrm{~b}-\mathrm{H}), 3.99(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C} 8 \mathrm{a}-\mathrm{H}), 0.14(\mathrm{~s}, 18 \mathrm{H}$, C3/5b-Si(CH3)3), -0.02 (s, 18H, C3/5b-Si(CH3)3).
${ }^{1} \mathrm{H}$ NMR, 500 MHz , Chloroform- $d$, Dimer 24

${ }^{1} \mathrm{H}$ NMR, 500 MHz , Chloroform- $d$, Compound ( $\pm$ )-5b ${ }^{2}$


(27) 4-((2R,3R)-3-(3,5-bis((4-methoxybenzyl)oxy)phenyl)-5-((E)-3,5-bis((4-methoxybenzyl)oxy)styryl)-2,3-dihydrobenzofuran-2-yl)phenol
Stilbene 25 ( $0.1 \mathrm{mmol}, 46.9 \mathrm{mg}$ ) was subjected to the general dimerization procedure. The crude product was purified via flash column chromatography ( $1 \%$ to $5 \%$ Acetone in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to afford the dihydrobenzofuran 27 ( $27.5 \mathrm{mg}, 59 \%$ yield, $>19: 1 \mathrm{dr}$ ).
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.36$ (d, $J=8.4 \mathrm{~Hz}, 4 \mathrm{H}$ ), 7.31 (d, $J=8.6 \mathrm{~Hz}, 4 \mathrm{H}$ ), 7.19 (d, $J=8.5 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.17 (d, $J$ $=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.00(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{~d}, J=8.3,4 \mathrm{H}), 6.91(\mathrm{~d}, J=8.5,2 \mathrm{H}), 6.89(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 4 \mathrm{H}), 6.83(\mathrm{~d}, J=$ $16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.80(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.71(\mathrm{dd}, J=2.3,1.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.55(\mathrm{td}, J=2.3,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.50(\mathrm{td}, J=2.2,1.2$ $\mathrm{Hz}, 1 \mathrm{H}), 6.42(\mathrm{t}, J=1.9 \mathrm{~Hz}, 2 \mathrm{H}), 5.49(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.98(\mathrm{~s}, 4 \mathrm{H}), 4.93-4.88(\mathrm{~m}, 4 \mathrm{H}), 4.46(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.82$ (s, 6H), 3.79 (s, 6H).
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta 160.38$, 160.26, 159.80, 159.59, 159.57, 155.78, 144.03, 139.80, 132.88, 130.93, 130.77, 129.52, 129.46, 129.44, 129.17, 129.07, 128.81, 128.15, 127.67, 126.38, 123.30, 115.63, 114.14, 109.84, 107.62, 105.53, 101.38, 100.89, 93.11, 69.99, 57.92, 55.45, 55.44.

IR (Neat): 3391, 2932, 2835, 1585, 1512, 1440, 1301, 1244, 1146, 1030, $818 \mathrm{~cm}^{-1}$;
HRMS (ESI) m/z calculated for $\mathrm{C}_{60} \mathrm{H}_{55} \mathrm{O}_{10}{ }^{+}([\mathrm{M}+\mathrm{H}]+$ ) 935.3790, found 935.3800.

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Dimer 27


(28) 4-((2R,3R)-3-(3,5-bis((4-methoxybenzyl)oxy)phenyl)-5-((E)-3,5-bis((4-methoxybenzyl)oxy)styryl)-7-methoxy-2,3-dihydrobenzofuran-2-yl)-2-methoxyphenol
Stilbene $\mathbf{2 6}(0.1 \mathrm{mmol}, 49.9 \mathrm{mg})$ was subjected to the general dimerization procedure. The crude product was purified via flash column chromatography ( $1 \%$ to $5 \%$ Acetone in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to afford the dihydrobenzofuran 28 ( $28.4 \mathrm{mg}, 57 \%$ yield, $>19: 1 \mathrm{dr})$.
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.37(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.31(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 4 \mathrm{H}), 7.00(\mathrm{~d}, J=15.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.93$ (d, $J=8.2 \mathrm{~Hz}, 5 \mathrm{H}), 6.90(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 6 \mathrm{H}), 6.88-6.84(\mathrm{~m}, 3 \mathrm{H}), 6.81(\mathrm{~d}, J=4.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.74-6.71(\mathrm{~m}, 2 \mathrm{H}), 6.55(\mathrm{~s}, 1 \mathrm{H})$, $6.52(\mathrm{~s}, 1 \mathrm{H}), 6.45-6.42(\mathrm{~m}, 2 \mathrm{H}), 5.67(\mathrm{~s}, 1 \mathrm{H}), 5.54(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.99(\mathrm{~s}, 4 \mathrm{H}), 4.91(\mathrm{~d}, J=11.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.89(\mathrm{~d}, J$ $=11.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.54(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.99(\mathrm{~s}, 3 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}), 3.82(\mathrm{~s}, 6 \mathrm{H}), 3.79(\mathrm{~s}, 6 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta 160.35,160.27$, 159.60, 159.57, 148.34, 146.72, 145.86, 144.53, 143.70, 139.66, 132.07, 131.73, 131.51, 129.50, 129.43, 129.41, 129.36, 129.03, 128.77, 126.68, 119.79, 116.08, 114.34, 114.12, 114.11, $114.10,110.18,108.80,107.66,105.61,105.53,101.47,100.92,94.19,69.98,58.28,56.18,56.10,55.43,55.40$.

IR (Neat): 3395, 2999, 2934, 2833, 1585, 1512, 1439, 1302, 1245, 1150, 1032, 824, $734 \mathrm{~cm}^{-1}$;
HRMS (ESI) m/z calculated for $\mathrm{C}_{62} \mathrm{H}_{59} \mathrm{O}_{12}{ }^{+}([\mathrm{M}+\mathrm{H}]+)$ 995.4001, found 995.3979.

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Dimer 28

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## (S16) methyl ( $\boldsymbol{E}$ )-3-(4-hydroxy-3-methoxyphenyl)acrylate

Commercially available methyl bromoacetate ( $1.42 \mathrm{~mL}, 15 \mathrm{mmol}$ ) was dissolved in toluene ( 60 mL ) in a flame-dried round bottom flask. To the stirring solution was added triphenylphosphine ( $18 \mathrm{mmol}, 4.72 \mathrm{~g}$ ), and the reaction was heated to $80^{\circ} \mathrm{C}$ for 4 hours. Upon cooling the reaction mixture to room temperature, the white solid product was collected via vacuum filtration, and any excess triphenylphosphine was rinsed away with hexanes. The product was dried under vacuum for $>24$ hours to ensure full removal of solvent and water prior to use in the subsequent olefination.

To a flame-dried, three-neck, round bottom flask was added methyl acetophosphonium bromide ( $5.35 \mathrm{mmol}, 2.22 \mathrm{~g}$ ), which was subsequently suspended in THF ( 15 mL ) and cooled on ice to $0^{\circ} \mathrm{C}$. To the stirring suspension was added $n \mathrm{BuLi}$ ( $5.35 \mathrm{mmol}, 2.14 \mathrm{~mL}, 2.5 \mathrm{M}$ in hexanes), and the reaction mixture was allowed to stir at temperature for 30 minutes to form the ylid. In a separate flame-dried, heart-shaped flask, TMS-vanillin (aldehyde C, $5.35 \mathrm{mmol}, 1.20 \mathrm{~g}$ ) was dissolved in THF $(20 \mathrm{~mL})$. The aldehyde was added to the ylide solution via cannula, and the reaction was allowed to warm to room temperature over 12 hours. The reaction was subsequently cooled on ice to $0^{\circ} \mathrm{C}$, and TBAF ( $5.35 \mathrm{mmol}, 5.35 \mathrm{~mL}, 1.0 \mathrm{M}$ in THF) was added. The reaction was allowed to warm to room temperature over 1 hour, at which point it was quenched by the addition of saturated ammonium chloride, diluted with EtOAc, then transferred to a separatory funnel containing additional saturated ammonium chloride. The layers were separated, and the aqueous layer was extracted with additional EtOAc. The combined organic layers were washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure. The crude product was purified via flash column chromatography ( $6-42 \% \mathrm{EtOAc}$ in Hexanes, 7 -step gradient, 2 column volumes per step) to afford the product $\mathbf{S 1 6}$ as a clear, colorless, oil ( $857 \mathrm{mg}, 77 \%$ yield). The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of the product was consistent with the data previously reported in the literature. ${ }^{6}$
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 7.62(\mathrm{~d}, J=15.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.07(\mathrm{dd}, J=8.3,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H})$, $6.92(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.29(\mathrm{~d}, J=15.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.93(\mathrm{~s}, 1 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (176 MHz, Chloroform- $d$ ) $\delta 167.87$, 148.09, 146.88, 145.09, 127.05, 123.16, 115.26, 114.85, 109.46, 56.05, 51.76.
${ }^{1} \mathrm{H}$ NMR, 700 MHz , Chloroform- $d$, S16

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, S16



## coniferyl alcohol (29) ((E)-4-(3-hydroxyprop-1-en-1-yl)-2-methoxyphenol)

Compound $\mathbf{S 1 6}$ ( $1.12 \mathrm{mmol}, 233 \mathrm{mg}$ ) was dissolved in DCM under inert atmosphere, and the stirring solution was cooled to $0^{\circ} \mathrm{C}$ on ice. A solution of diisobutylaluminum hydride $(3.36 \mathrm{mmol}, 3.36 \mathrm{~mL}, 1.0 \mathrm{M} \mathrm{in} \mathrm{DCM})$ was added dropwise to the stirring solution, causing the reaction to change from colorless to slightly yellow. The reaction mixture was held at $0{ }^{\circ} \mathrm{C}$ for 10 minutes, at which point it was allowed to warm to room temperature over 12 hours. The reaction was quenched slowly with 4 mL of Rochelle's salt, turning the mixture cloudy. The mixture was let to stir at room temperature for $>6$ hours, at which point it was diluted with DI water and transferred to a separatory funnel. The layers were separated, and the aqueous layer was washed with multiple portions of DCM. The combined organic layers were washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure. The crude product was purified via flash column chromatography ( $28-98 \%$ EtOAc in Hexanes, 5 step gradient, 2 column volumes per step) to afford the product 35 as a white solid ( $167 \mathrm{mg}, 83 \%$ yield). The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of the product was consistent with the data previously reported in the literature. ${ }^{7}$
${ }^{1} \mathrm{H}$ NMR $(500 \mathrm{MHz}$, Chloroform- $d) \delta 6.92(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.90(\mathrm{dd}, J=8.1,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H})$, $6.54(\mathrm{dt}, J=15.8,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.23(\mathrm{dt}, J=15.8,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.63(\mathrm{~d}, J=2.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.30(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.91(\mathrm{~s}$, $3 \mathrm{H}), 1.41(\mathrm{~s}, 1 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (126 MHz, Chloroform- $d$ ) $\delta 146.76,145.74,131.55,129.38,126.30,120.47,114.58,108.44,64.04,56.03$.

${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, 29
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| 230 | 220 | 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -10 |


( $\pm$ )-hierochin D (3) (4-((2S,3R)-3-(hydroxymethyl)-5-((E)-3-hydroxyprop-1-en-1-yl)-7-methoxy-2,3-dihydrobenzofuran-2-yl)-2-methoxyphenol)

Coniferyl alcohol ( $\mathbf{2 9}, 0.1 \mathrm{mmol}, 18 \mathrm{mg}$ ) was subjected to the general dimerization conditions (see page S42), with the only variation being the amount of 2,6 -lutidine ( $5.8 \mu \mathrm{~L}, 0.05 \mathrm{mmol}$ ). The crude reaction material was purified via flash column chromatography to afford ( $\pm$ )-hierochin $\mathrm{D}(\mathbf{3})$ as a colorless oil ( $9.4 \mathrm{mg}, 53 \%$ yield). The ${ }^{13} \mathrm{C}$ NMR was consistent with the previous report for this compound. ${ }^{8}$
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Acetone- $d_{6}$ ) $\delta 148.97$, 148.38, 147.29, 145.18, 134.40, 131.93, 130.54, 130.42, 128.38, 119.60, 116.09, $115.68,111.70,110.47,88.54,64.63,63.43,56.39,56.28,54.79$.
${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d, 3$


## Attempted Direct Cyclization of QMD products:



Figure S1. Attempted direct cyclization of QMDs containing electron-donating substituents at C12 results in disproportionation back to the stilbene precursor. Note that $46 \%$ yield is relative to the mass of $\mathbf{1 3}$, therefore the fragmentation has a theoretical yield for $\mathbf{S} 7$ of $50 \%$.

## Natural Product Analog Synthesis:

Tautomerization of BQM dimer to MQM dimer - General Procedure:
The starting BQM dimer was added to a reaction vial charged with a stir bar. The atmosphere was evacuated and replaced with nitrogen, and the starting material was dissolved in THF. The reaction solution was cooled in an ice bath to $0^{\circ} \mathrm{C}$, and potassium bis(trimethylsilyl)amide (KHMDS) was added as a solution ( 1.25 equiv., 0.7 M in toluene). The reaction was allowed to stir until the starting material consumed based on TLC analysis. The reaction was quenched by the addition of aqueous saturated ammonium chloride, diluted with EtOAc, and transferred to a separatory funnel containing additional aq. sat. $\mathrm{NH}_{4} \mathrm{Cl}$. The layers were separated, and the aqueous layer was extracted with additional portions of ethyl acetate. The combined organic layers were washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure. The crude product was further purified by flash column chromatography.

(S17) (S,Z)-2,6-di-tert-butyl-4-(4-(3,5-di-tert-butyl-4-hydroxyphenyl)-2,3-diphenylbut-3-en-1-ylidene)cyclohexa-2,5-dien-1-one

Dimer 10 ( $0.026 \mathrm{mmol}, 16 \mathrm{mg}$ ) was subjected to the tautomerization conditions, and the crude product was purified by flash column chromatography ( $10-50 \%$ DCM/Hexanes) to afford MQM S17 ( $10 \mathrm{mg}, 62 \%$ yield).
${ }^{1} \mathrm{H}$ NMR $(500 \mathrm{MHz}$, Chloroform-d) $\delta 7.34(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.25-7.14(\mathrm{~m}, 7 \mathrm{H}), 7.12(\mathrm{t}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~s}, 2 \mathrm{H})$, $6.74(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.27(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.83(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{dd}, J=9.5,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.01(\mathrm{~s}, 1 \mathrm{H}), 3.67$ $(\mathrm{d}, J=9.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.38(\mathrm{~s}, 18 \mathrm{H}), 1.30(\mathrm{~s}, 9 \mathrm{H}), 0.92(\mathrm{~s}, 9 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta 186.41,152.63,147.21,146.57,145.86,144.46,143.44,141.60,135.28,135.20$, $130.05,128.51,128.24,128.18,127.76,127.47,126.68,126.39,123.82,65.46,56.53,54.16,34.78,34.58,34.23,30.28$, 29.77, 29.06.

IR (Neat): $3623,2954,2912,1653,1437,1366,1192,1116,886,734,700 \mathrm{~cm}^{-1}$;
HRMS (ESI) m/z calculated for $\mathrm{C}_{44} \mathrm{H}_{55} \mathrm{O}_{2}{ }^{+}([\mathrm{M}+\mathrm{H}]+)$ 615.4197, found 615.4194.

${ }^{13}$ C NMR, 126 MHz , Chloroform- $d$, Tautomer $\mathbf{S 1 7}$


(S18) (S,Z)-2,6-di-tert-butyl-4-(4-(3,5-di-tert-butyl-4-hydroxyphenyl)-2,3-bis(4-(methylthio)phenyl)but-3-en-1-ylidene)cyclohexa-2,5-dien-1-one

Dimer 13 ( $0.021 \mathrm{mmol}, 15 \mathrm{mg}$ ) was subjected to the tautomerization conditions, and the crude product was purified by flash column chromatography ( $10-50 \%$ DCM/Hexanes) to afford MQM S18 ( $14.8 \mathrm{mg}, 99 \%$ yield).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform-d) $\delta 7.24(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.13(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.10(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.09$ (d, $J$ $=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.91(\mathrm{~s}, 2 \mathrm{H}), 6.71(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.24(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.79(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.02(\mathrm{dd}, J=9.2$, $2.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.01(\mathrm{~s}, 1 \mathrm{H}), 3.60(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.44(\mathrm{~s}, 3 \mathrm{H}), 2.41(\mathrm{~s}, 3 \mathrm{H}), 1.38(\mathrm{~s}, 18 \mathrm{H}), 1.30(\mathrm{~s}, 9 \mathrm{H}), 0.91(\mathrm{~s}, 9 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta 186.52$, 152.83, 147.44, 146.09, 145.94, 144.52, 141.58, 140.43, 138.06, 136.31, $135.41,132.09,129.66,128.82,127.70,127.25,126.90,126.22,123.97,65.73,56.60,53.73,34.95,34.75,34.40,30.45$, 29.94, 29.22, 15.85, 15.72.

IR (Neat): 3645, 2960, 1659, 1496, 1439, 1363, 1316, 1237, 1150, 1094, 891, 814, $728 \mathrm{~cm}^{-1}$;
HRMS (ESI) m/z calculated for $\mathrm{C}_{46} \mathrm{H}_{59} \mathrm{O}_{2} \mathrm{~S}_{2}{ }^{+}([\mathrm{M}+\mathrm{H}]+) 707.3951$, found 707.3940 .


${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Tautomer $\mathbf{S 1 8}$




(S19) (S,Z)-4-(2,3-bis(benzo[d][1,3]dioxol-5-yl)-4-(3,5-di-tert-butyl-4-hydroxyphenyl)but-3-en-1-ylidene)-2,6-di-tert-butylcyclohexa-2,5-dien-1-one
Dimer $11(0.021 \mathrm{mmol}, 14.7 \mathrm{mg})$ was subjected to the tautomerization conditions, and the crude product was purified by flash column chromatography ( $10-50 \%$ DCM/Hexanes) to afford MQM S19 ( $9 \mathrm{mg}, 61 \%$ yield).
${ }^{1} \mathrm{H}$ NMR ( 700 MHz , Chloroform- $d$ ) $\delta 6.90$ ( $\mathrm{s}, 2 \mathrm{H}$ ), 6.88 (d, $J=1.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.81 (dd, $J=8.1,1.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), $6.70-6.66$ (m, $4 \mathrm{H}), 6.65(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.21(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.93-5.89(\mathrm{~m}, 2 \mathrm{H}), 5.89-5.84(\mathrm{~m}, 2 \mathrm{H}), 5.65(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H})$, $5.01(\mathrm{~s}, 1 \mathrm{H}), 4.94(\mathrm{dd}, J=9.5,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.59(\mathrm{~d}, J=9.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.37(\mathrm{~s}, 18 \mathrm{H}), 1.29(\mathrm{~s}, 9 \mathrm{H}), 0.90(\mathrm{~s}, 9 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta$ 186.54, 152.80, 147.88, 147.74, 147.31, 147.16, 146.24, 146.23, 145.91, 144.70, 141.82, 137.52, 135.37, 129.74, 128.76, 127.84, 123.90, 121.69, 120.77, 108.43, 108.40, 108.19, 107.19, 101.17, 100.99, 65.65, 56.40, 54.12, 34.93, 34.72, 34.39, 30.44, 29.94, 29.20.

IR (Neat): $3628,2955,2905,1653,1635,1503,1436,1363,1235,1038,933,809,729 \mathrm{~cm}^{-1}$;
HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calculated for $\mathrm{C}_{46} \mathrm{H}_{55} \mathrm{O}_{6}{ }^{+}([\mathrm{M}+\mathrm{H}]+) 703.3993$, found 703.3985.

${ }^{13}$ C NMR, 176 MHz, Chloroform- $d$, Tautomer S19


(S20) (S,Z)-4-(2,3-bis(4-(benzyloxy)phenyl)-4-(3,5-di-tert-butyl-4-hydroxyphenyl)but-3-en-1-ylidene)-2,6-di-tert-butylcyclohexa-2,5-dien-1-one

Dimer 14 ( $0.087 \mathrm{mmol}, 72 \mathrm{mg}$ ) was subjected to the tautomerization conditions, and the crude product was purified by flash column chromatography (5-35\% DCM/Hexanes) to afford MQM S20 ( $50 \mathrm{mg}, 69 \%$ yield).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.38(\mathrm{~m}, 8 \mathrm{H}), 7.32(\mathrm{~m}, 2 \mathrm{H}), 7.28(\mathrm{~m}, 2 \mathrm{H}), 7.14(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.92(\mathrm{~s}, 2 \mathrm{H})$, $6.83(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 4 \mathrm{H}), 6.72(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.24(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.69(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.01(\mathrm{~m}, 4 \mathrm{H}), 4.96(\mathrm{~s}$, $2 \mathrm{H}), 3.61(\mathrm{~d}, J=9.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.37(\mathrm{~s}, 18 \mathrm{H}), 1.29(\mathrm{~s}, 9 \mathrm{H}), 0.90(\mathrm{~s}, 9 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (126 MHz, Chloroform-d) $\delta 186.64,158.28,157.48,152.73,147.14,146.16,145.73,145.03,142.15,137.16$, $136.96,136.14,135.30,135.27,129.35,128.73,128.69,128.38,128.36,128.15,128.09,127.92,127.67,127.59,123.96$, $114.97,114.64,70.06$ (2C), 65.65, 56.53, 53.57, 34.92, 34.71, 34.39, 30.44, 29.95, 29.22.

IR (Neat): $3635,2959,2869,1656,1507,1454,1433,1364,1228,1178,1120,1027,886,732,695 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{58} \mathrm{H}_{67} \mathrm{O}_{4}^{+}\left([\mathrm{M}+\mathrm{H}]^{+}\right) 827.5034$, found 827.5010.

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Tautomer $\mathbf{S 2 0}$


## Cyclization of MQM dimer to quadrangularin A core:

The starting material was dried down into a flame-dried round bottom flask charged with stir bar. The atmosphere was evacuated and replaced with $\mathrm{N}_{2}$, and the starting material was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.01 \mathrm{M}$ reaction concentration). The solution was cooled to the reaction temperature, and $\mathrm{BF}_{3} \bullet \mathrm{OEt}_{2}$ ( 2 equiv.) was added dropwise. The reaction was allowed to stir for 3 hours, at which point it was raised from the ice bath and quenched via the addition of saturated $\mathrm{NaHCO}_{3}$. Once the reaction had thawed, it was poured into a separatory funnel, and the layers were separated. The aqueous layer was extracted with additional portions of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and the combined organic layers were washed with brine, dried over magnesium sulfate, and concentrated under reduced pressure. The crude product was purified by column chromatography.

(30) 4-((1S,2S)-5,7-bis(benzyloxy)-2-(3,5-bis(benzyloxy)phenyl)-3-((E)-3,5-di-tert-butyl-4-hydroxybenzylidene)-2,3-dihydro-1H-inden-1-yl)-2,6-di-tert-butylphenol

BQM 5 ( $500 \mathrm{mg}, 0.481 \mathrm{mmol}$ ) was subjected to the standard tautomerization conditions, and the crude product was subjected to the cyclization conditions at $-78{ }^{\circ} \mathrm{C}$ with $\mathrm{BF}_{3} \bullet \mathrm{OEt}_{2}$ ( $128 \mu \mathrm{~L}, 0.962 \mathrm{mmol}, 2.0$ equiv.). The product was purified by column chromatography ( $0 \%$ to $20 \%$ EtOAc in Hexanes) to afford compound 37 ( $452 \mathrm{mg}, 90 \%$ yield). The ${ }^{1} \mathrm{H}$ NMR spectrum was consistent with the prior report for this compound. ${ }^{1}$
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.55-7.49(\mathrm{~m}, 2 \mathrm{H}), 7.45-7.33(\mathrm{~m}, 13 \mathrm{H}), 7.33-7.28(\mathrm{~m}, 3 \mathrm{H}), 7.24(\mathrm{t}, J=2.8 \mathrm{~Hz}$, $3 \mathrm{H}), 7.19-7.16(\mathrm{~m}, 1 \mathrm{H}), 7.08(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.05(\mathrm{dd}, J=7.2,2.7 \mathrm{~Hz}, 4 \mathrm{H}), 6.98(\mathrm{t}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.62(\mathrm{~d}, J=2.1$ $\mathrm{Hz}, 2 \mathrm{H}), 6.45(\mathrm{dt}, J=5.4,2.0 \mathrm{~Hz}, 2 \mathrm{H}), 5.19(\mathrm{~s}, 1 \mathrm{H}), 5.18(\mathrm{~s}, 2 \mathrm{H}), 5.02(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.97(\mathrm{~s}, 4 \mathrm{H}), 4.93(\mathrm{~d}, J=12.4$ $\mathrm{Hz}, 1 \mathrm{H}), 4.86(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.43(\mathrm{~s}, 1 \mathrm{H}), 4.35(\mathrm{~s}, 1 \mathrm{H}), 1.34(\mathrm{~s}, 18 \mathrm{H}), 1.32(\mathrm{~s}, 18 \mathrm{H})$.


(31) 2,6-di-tert-butyl-4-((1S,2S)-3-((E)-3,5-di-tert-butyl-4-hydroxybenzylidene)-2-phenyl-2,3-dihydro-1H-inden-1yl)phenol
Compound $\mathbf{S 1 7}$ ( $330 \mathrm{mg}, 0.537 \mathrm{mmol}$ ) was subjected to the standard cyclization conditions at $0{ }^{\circ} \mathrm{C}$ with $\mathrm{BF}_{3} \bullet \mathrm{OEt}_{2}(0.132$ $\mathrm{mL}, 1.07 \mathrm{mmol}$ ), and the crude product was purified by column chromatography ( $3 \%, 6 \%, 9 \%, 12 \%, 17 \%, 26 \%, 37 \%, 51 \%$, DCM/Hexanes, 2 CV per step) to afford compound $\mathbf{3 1}$ ( $154 \mathrm{mg}, 47 \%$ yield).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.75(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.31(\mathrm{~m}, 3 \mathrm{H}), 7.24(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{~d}, J=7.9 \mathrm{~Hz}$, $1 \mathrm{H}), 7.15(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.10(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.03(\mathrm{~s}, 2 \mathrm{H}), 6.94(\mathrm{~s}, 2 \mathrm{H}), 5.16(\mathrm{~s}, 1 \mathrm{H}), 5.04(\mathrm{~s}, 1 \mathrm{H}), 4.47(\mathrm{~s}, 1 \mathrm{H})$, $4.28(\mathrm{~s}, 1 \mathrm{H}), 1.35(\mathrm{~s}, 18 \mathrm{H}), 1.25(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13}$ C NMR ( 126 MHz , Chloroform- $d$ ) $\delta$ 153.00, 152.30, 146.24, 145.16, 143.23, 141.47, 136.98, 135.75, 135.66, 128.82, $128.78,128.30,127.69,127.39,126.33,126.26,125.84,123.88,123.83,119.87,61.23,58.11,34.45,34.32,30.38,30.23$.

IR (Neat): 3616, 2953, 1470, 1311, 1235, 1137, 957, 767, $694 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{NaC}_{44} \mathrm{H}_{54} \mathrm{O}_{2}{ }^{+}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$637.3965, found 637.3965.

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${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Analog Core 31


(32) 2,6-di-tert-butyl-4-((1S,2S)-3-((E)-3,5-di-tert-butyl-4-hydroxybenzylidene)-6-(methylthio)-2-(4-(methylthio)phenyl)-2,3-dihydro-1H-inden-1-yl)phenol
Compound $\mathbf{S 1 8}(96 \mathrm{mg}, 0.136 \mathrm{mmol})$ was subjected to the standard cyclization conditions at $0{ }^{\circ} \mathrm{C}^{2}$ with $\mathrm{BF}_{3} \bullet \mathrm{OEt}_{2}(0.034$ $\mathrm{mL}, 0.27 \mathrm{mmol}$ ), and the crude product was purified by column chromatography ( $10 \%$ to $50 \% \mathrm{DCM} / \mathrm{Hexanes}$ ) to afford compound 32 ( $59 \mathrm{mg}, 62 \%$ yield).
${ }^{1} \mathrm{H}$ NMR $(500 \mathrm{MHz}$, Chloroform- $d) \delta 7.37(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.16(\mathrm{~m}, 3 \mathrm{H}), 7.11(\mathrm{~s}, 2 \mathrm{H}), 7.07(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.05(\mathrm{~s}$, $1 \mathrm{H}), 7.01(\mathrm{~s}, 1 \mathrm{H}), 6.98(\mathrm{~s}, 2 \mathrm{H}), 5.01(\mathrm{~s}, 1 \mathrm{H}), 4.98(\mathrm{~s}, 1 \mathrm{H}), 4.15(\mathrm{~s}, 1 \mathrm{H}), 4.06(\mathrm{~s}, 1 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}), 2.42(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 18 \mathrm{H})$, 1.29 (s, 18H).
${ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta 152.49,152.40,138.90,138.15,137.34,137.20,136.43,135.73,135.65,133.50$, $131.60,127.92,126.99,126.51,126.47,125.25,124.10,123.97,123.88,53.25,50.27,34.52,34.47,30.43,30.41,16.23$, 15.99 .

IR (Neat): 3614, 2957, 2920, 1597, 1434, 1240, 1150, 1119, 883, 821, 790, $765 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{46} \mathrm{H}_{58} \mathrm{O}_{2} \mathrm{~S}_{2}{ }^{+}\left([\mathrm{M}]^{+}\right) 706.3873$, found 706.3872 .

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$$
\begin{array}{llllllllllllllllllllllll}
7.55 & 7.50 & 7.45 & 7.40 & 7.35 & 7.30 & 7.25 & 7.20 & 7.15 & 7.10 & 7.05 & 7.00 & 6.95 & 6.90 & 6.85 \\
& & & & & (\mathrm{ppm})
\end{array}
$$



S99
${ }^{13} \mathrm{C}$ NMR, 126 MHz , Chloroform- $d$, Analog Core 32



No

$\begin{array}{llllllllllllllllllllllllllllll}141 & 140 & 139 & 138 & 137 & 136 & 135 & 134 & 133 & 132 & 131 & 130 & 129 & 128 & 127 & 126 & 125 & 124 & 123 & 122\end{array}$ f1 (ppm)

[^1]
(33) 4-((1S,2S)-6-(benzyloxy)-2-(4-(benzyloxy)phenyl)-3-((E)-3,5-di-tert-butyl-4-hydroxybenzylidene)-2,3-dihydro-1H-inden-1-yl)-2,6-di-tert-butylphenol
Compound $\mathbf{S 1 9}$ ( $102 \mathrm{mg}, 0.145 \mathrm{mmol}$ ) was subjected to the standard cyclization conditions at $0{ }^{\circ} \mathrm{C}$ with $\mathrm{BF}_{3} \bullet \mathrm{OEt}_{2}(0.036$ $\mathrm{mL}, 0.29 \mathrm{mmol}$ ), and the crude product was purified by column chromatography ( $10 \%$ to $50 \% \mathrm{DCM} / \mathrm{Hexanes}$ ) to afford compound 33 ( $65 \mathrm{mg}, 64 \%$ yield).
${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform- $d$ ) $\delta 7.14(\mathrm{~s}, 1 \mathrm{H}), 7.01(\mathrm{~s}, 2 \mathrm{H}), 6.97(\mathrm{~d}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.91(\mathrm{~s}, 2 \mathrm{H}), 6.76(\mathrm{~s}, 1 \mathrm{H}), 6.75$ $(\mathrm{d}, J=4.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.52(\mathrm{~s}, 1 \mathrm{H}), 5.95(\mathrm{~s}, 2 \mathrm{H}), 5.88(\mathrm{dd}, J=5.2,3.7 \mathrm{~Hz}, 2 \mathrm{H}), 5.13(\mathrm{~s}, 1 \mathrm{H}), 5.03$ $(\mathrm{s}, 1 \mathrm{H}), 4.34(\mathrm{~s}, 1 \mathrm{H}), 4.07(\mathrm{~s}, 1 \mathrm{H}), 1.36(\mathrm{~s}, 18 \mathrm{H}), 1.28(\mathrm{~s}, 18 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( 176 MHz , Chloroform- $d$ ) $\delta 152.81,152.38,148.46,147.95,147.79,145.86,141.64,140.57,139.39$, 136.91, $136.82,135.87,135.68,128.94,125.60,123.72,121.85,120.70,108.46,108.28,106.21,101.35,100.86,99.99,61.17$, 58.31, 34.47, 34.37, 30.41, 30.29 .

IR (Neat): 3634, 2957, 2912, 1476, 1431, 1231, 1150, 1035, 936, $737 \mathrm{~cm}^{-1}$;
HRMS (ESI) $m / z$ calculated for $\mathrm{C}_{46} \mathrm{H}_{54} \mathrm{O}_{6}{ }^{+}\left([\mathrm{M}]^{+}\right) 702.3915$, found 702.3900.

${ }^{13} \mathrm{C}$ NMR, 176 MHz , Chloroform- $d$, Analog Core 33



## Analysis of Radical-Trapping Antioxidant Activity:

## Kinetic Data

General Procedure for Inhibited Autoxidations
The inhibited autoxidations were carried out following our reported method. ${ }^{9}$ All autoxidations of 1-hexadecene ( 2.9 M ) and PBD-BODIPY $(10 \mu \mathrm{M})$ were initiated by AIBN $(6 \mathrm{mM})$ in chlorobenzene at $37^{\circ} \mathrm{C}$. A 3.5-mL quartz cuvette was charged with $\mathrm{PhCl}(0.44 \mathrm{~mL})$, 1-hexadecene $(2.00 \mathrm{~mL})$. The cuvette was preheated to $37{ }^{\circ} \mathrm{C}$ in a thermostatted sample holder of a UV-vis spectrophotometer and allowed to equilibrate for approximately 15 min . To the cuvette was added PBDBODIPY ( $12.5 \mu \mathrm{~L}$ of a 2.00 mM stock solution in 1,2,4-trichlorobenzene) and AIBN ( $50 \mu \mathrm{~L}$ of a 300 mM stock solution in chlorobenzene). The solution was thoroughly mixed prior to monitoring the uninhibited co-autoxidation via the disappearance of the PBD-BODIPY probe at 588 nm for 10 min to ensure the reaction was proceeding at a constant rate. Finally, the antioxidant under investigation was added ( $5.0 \mu \mathrm{~L}$ of a 2.5 mM solution in chlorobenzene), the solution was mixed thoroughly, and the absorbance readings were resumed. The resulting Abs vs time data were processed as previously reported. ${ }^{9}$ The rate of initiation $\left(R_{\mathrm{i}}=1.3 \times 10^{-9} \mathrm{Ms}^{-1}\right)$ and second order rate constant for propagation for the dye ( $k_{\text {PBD-BODIPY }}$ $=3792 \mathrm{M}^{-1} \mathrm{~s}^{-1}$ ) necessary to compute stoichiometric data ( $n$ ) and inhibition rate constants $\left(k_{\mathrm{inh}}\right)$ were determined using PMC as a standard, which has an established stoichiometry of $2 .{ }^{10,11}$

Table S2. Comparison of inhibition rate constants ( $k_{\mathrm{inh}}$ ) and stoichiometry ( $n$ ) during inhibited co-autoxidations of 1hexadecene $(2.9 \mathrm{M})$ and PBD-BODIPY $(10 \mu \mathrm{M})$ initiated by AIBN $(6 \mathrm{mM})$ in chlorobenzene at $37^{\circ} \mathrm{C}$.

| X |  | Quadrangularin A analogues |  |  | Quinone methide dimers |  |  | Parent phenols |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | $\begin{gathered} k_{\text {inh }} \\ \left(10^{5} M^{-1} \mathbf{s}^{-1}\right) \end{gathered}$ | $n$ |  | $\begin{gathered} k_{\text {inh }} \\ \left(10^{5} M^{-1} s^{-1}\right) \end{gathered}$ | $n$ |  | $\begin{gathered} k_{\text {inh }} \\ \left(10^{5} M^{-1} s^{-1}\right) \end{gathered}$ | $n$ |
| $4-\mathrm{CF}_{3}$ |  | N/A | N/A | 20 | $24 \pm 1$ | $1.4 \pm 0.1$ | S14 | $1.3 \pm 0.1$ | $2.3 \pm 0.1$ |
| 3,5-diOBn | 30 | $1.1 \pm 0.1$ | $3.8 \pm 0.3$ | 5 | $19 \pm 5$ | $1.7 \pm 0.1$ | 10 | $1.4 \pm 0.1$ | $2.3 \pm 0.1$ |
| H | 31 | $1.0 \pm 0.1$ | $4.1 \pm 0.3$ | 10 | $20 \pm 1$ | $1.4 \pm 0.1$ | S4 | $1.6 \pm 0.1$ | $2.0 \pm 0.1$ |
| 4-SMe | 32 | $0.8 \pm 0.1$ | $3.8 \pm 0.1$ | 13 | $29 \pm 1$ | $1.3 \pm 0.1$ | S7 | $1.7 \pm 0.1$ | $2.2 \pm 0.1$ |
| 3,4-dioxyl | 33 | $1.3 \pm 0.2$ | $4.1 \pm 0.1$ | 11 | $31 \pm 4$ | $1.3 \pm 0.1$ | S5 | $1.9 \pm 0.3$ | $2.1 \pm 0.1$ |

## Computational Data ${ }^{12}$

Table S3. Summary of enthalpies $\left(\Delta H, \Delta H^{\dagger}\right)$, free energies ( $\Delta \mathrm{G}, \Delta \mathrm{G}^{\dagger}$ ), and corresponding computed second order rate constants ( $k$ ) for the addition of methyl peroxyl radical to or hydrogen atom transfer (HAT) from a truncated analogue of 14 in the gas phase at $37^{\circ} \mathrm{C}$.

| Mechanism | Position | $\boldsymbol{\Delta H}$, <br> $\left(\boldsymbol{k c a l ~ m o l}^{-1}\right)$ | $\boldsymbol{\Delta G}$, <br> $\left(\boldsymbol{k c a l ~ m o l}^{-1}\right)$ | $\boldsymbol{\Delta} \boldsymbol{H}^{t}$, <br> $\left(\boldsymbol{k c a l ~ m o l}^{-1}\right)$ | $\boldsymbol{\Delta} \boldsymbol{G}^{\boldsymbol{t}}$, <br> $\left(\boldsymbol{k c a l ~ m o l}^{-1}\right)$ | $\boldsymbol{k}^{\mathrm{a}}$ <br> $\left(\mathbf{M}^{-1} \mathbf{s}^{-1}\right)$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| HAT | $\mathrm{C}_{\beta}-\mathrm{H}$ | -15.7 | -15.3 | 7.5 | 19.0 | 6 |
| Addition | $\mathrm{C}_{\alpha}$ | -31.5 | -19.4 | -1.7 | 10.8 | $4 \times 10^{6}$ |

${ }^{\text {a }}$ Computed using $k=\frac{R T}{P} \cdot \frac{k_{B} T}{h} e^{\frac{-\Delta G^{T S}}{R T}}$ where $\mathrm{T}=310.15 \mathrm{~K}$.

Optimized Gaussian Structures and CBS-QB3 ${ }^{13}$ or DFT Energies (Hartree)
Methyl peroxyl radical

## $\mathrm{H}_{3} \mathrm{COO}^{\bullet}$

CBS-QB3 Enthalpy = -189.954731
CBS-QB3 Free Energy = -189.985243

02

| C | 1.09605900 | -0.18318300 | 0.00000000 |
| :--- | ---: | ---: | ---: |
| H | 1.87467700 | 0.57860700 | -0.00001500 |
| H | 1.14885000 | -0.80070400 | 0.89699800 |
| H | 1.14883700 | -0.80072700 | -0.89698300 |
| O | -0.15733600 | 0.54388600 | 0.00000000 |
| O | -1.18625300 | -0.27864500 | 0.00000000 |

Methyl hydroperoxide

## $\mathrm{H}_{3} \mathrm{COOH}$

CBS-QB3 Enthalpy $=-190.589542 \quad$ CBS-QB3 Free Energy $=\quad-190.620447$
01

| C | 1.12961700 | -0.22363900 | 0.02672600 |
| :--- | ---: | ---: | ---: |
| H | 1.97292100 | 0.47126800 | 0.02466400 |
| H | 1.14400500 | -0.82582200 | 0.94203100 |
| H | 1.18954700 | -0.87767200 | -0.84877800 |
| O | -0.01619300 | 0.60684200 | -0.03138300 |
| O | -1.16412600 | -0.28550800 | -0.09072600 |
| H | -1.64161900 | 0.00339000 | 0.69859800 |

Truncated dimer


| CBS-QB3 Enthalpy $=-654.002976$ | CBS-QB3 Free |  |  |
| :--- | ---: | ---: | ---: |
| 01 |  |  |  |
| C | -2.28497800 | 0.40886900 | -1.38146900 |
| C | -3.55270400 | -0.03922500 | -1.32815500 |
| C | -4.13529300 | -0.56092000 | -0.07890600 |
| C | -3.23865900 | -0.55847700 | 1.09408900 |
| C | -1.97179800 | -0.10764000 | 1.02679200 |
| H | -1.87418700 | 0.78842400 | -2.31241000 |
| H | -4.20081500 | -0.03800300 | -2.19697400 |
| H | -3.66131800 | -0.94226000 | 2.01572400 |
| H | -1.34982200 | -0.12439600 | 1.91411400 |
| C | -1.40780200 | 0.40947300 | -0.21456200 |
| C | -0.14133100 | 0.88393400 | -0.33638800 |
| H | 0.15649000 | 1.24391200 | -1.31988600 |
| O | -5.28793700 | -0.97120500 | -0.01361300 |
| C | 0.93072100 | 0.98559600 | 0.71399400 |
| C | 2.15417900 | 0.17948500 | 0.27509100 |
| C | 2.43091400 | -1.05007100 | 0.87987500 |
| C | 3.00520900 | 0.62664700 | -0.74224100 |
| C | 3.53098300 | -1.81010100 | 0.48913600 |
| H | 1.77888600 | -1.41666800 | 1.66618900 |
| C | 4.10408800 | -0.13185400 | -1.13742500 |
| H | 2.81702700 | 1.57620900 | -1.23132600 |
| C | 4.37188000 | -1.35298100 | -0.52222200 |
| H | 3.72971200 | -2.75869500 | 0.97523300 |
| H | 4.75316900 | 0.23292500 | -1.92573000 |
| H | 5.22845900 | -1.94234200 | -0.82872500 |
| C | 1.25966000 | 2.46665200 | 1.00604200 |
| H | 1.57784500 | 2.99577200 | 0.10458500 |
| H | 0.38007500 | 2.98079300 | 1.40032500 |
| H | 2.06455800 | 2.53733300 | 1.74154700 |
| H | 0.57301500 | 0.54030300 | 1.64432900 |

C-H HAT radical product (truncated dimer)


| CBS-QB3 Enthalpy $=$ | -653.393246 | CBS-QB3 Free Energy $=$ | -653.450284 |
| :--- | ---: | ---: | ---: |
| 02 |  |  |  |
| C | 1.52663700 | -0.17878300 | -0.06031300 |
| C | 2.26612700 | 1.04646900 | -0.05163000 |
| C | 3.63125800 | 1.05609200 | -0.05797800 |
| C | 4.41372500 | -0.17389500 | -0.07235200 |
| C | 3.64436100 | -1.41088100 | -0.11084300 |
| C | 2.28208000 | -1.39605400 | -0.11637600 |
| H | 1.74048500 | 1.98910300 | -0.07297000 |
| H | 4.18983800 | 1.98508300 | -0.06360700 |
| H | 4.20631700 | -2.33724000 | -0.13798000 |
| H | 1.73406700 | -2.33262400 | -0.14873300 |
| C | 0.09559100 | -0.30235800 | -0.05597100 |
| H | -0.23773000 | -1.30730800 | -0.29735500 |
| C | -0.90681300 | 0.59889900 | 0.18954500 |
| C | -2.31412300 | 0.15072700 | 0.07608700 |
| C | -3.30656300 | 1.03105100 | -0.39201800 |
| C | -2.71066100 | -1.15268400 | 0.42751200 |
| C | -4.62699500 | 0.61798900 | -0.53178300 |
| H | -3.03832200 | 2.04156900 | -0.67632200 |
| C | -4.03301100 | -1.56037100 | 0.29989500 |
| H | -1.98216700 | -1.84097800 | 0.83868200 |
| C | -4.99806500 | -0.67954400 | -0.18603900 |
| H | -5.36792600 | 1.31206600 | -0.91227100 |
| H | -4.31363000 | -2.56608300 | 0.59198900 |
| H | -6.02906600 | -0.99866000 | -0.28584100 |
| O | 5.65560200 | -0.16513300 | -0.06495700 |
| C | -0.67848700 | 2.04011600 | 0.58017800 |
| H | 0.15428900 | 2.12915000 | 1.27928800 |
| H | -0.44885500 | 2.67090900 | -0.28670000 |
| H | -1.56247000 | 2.45544600 | 1.06398500 |

## Addition radical product (truncated dimer)



| CBS-QB3 Enthalpy $=$ | -844.008115 | CBS-QB3 Free |  |
| :--- | ---: | ---: | :---: |
| 02 |  |  |  |
| C | 1.68586100 | -0.73545200 | -0.94678200 |
| C | 2.03864400 | -2.05690200 | -1.04386900 |
| C | 1.65692100 | -3.01393600 | -0.01965400 |
| C | 0.89517400 | -2.49267000 | 1.10074700 |
| C | 0.5565100 | -1.16384100 | 1.17166500 |
| H | 1.98969400 | -0.02773600 | -1.70968300 |
| H | 2.61367700 | -2.43783300 | -1.87997300 |
| H | 0.61296200 | -3.19839200 | 1.87314800 |
| H | -0.01169700 | -0.79243000 | 2.01765600 |
| C | 0.93550600 | -0.26276200 | 0.15473600 |
| O | 1.97130500 | -4.22160000 | -0.09771200 |
| O | 1.58857600 | 2.04790200 | -0.20443700 |
| O | 2.61138400 | 1.96629700 | 0.84596800 |
| C | 3.78537300 | 2.50236300 | 0.26894200 |
| H | 4.52260200 | 2.46789200 | 1.07383500 |
| H | 4.12979300 | 1.89716600 | -0.57569200 |
| H | 3.63454500 | 3.53954000 | -0.04808300 |
| C | 0.53023100 | 1.19587100 | 0.22243000 |
| C | -0.67504300 | 1.54242800 | -0.70021500 |
| H | 0.27533000 | 1.45818500 | 1.25591200 |
| C | -1.89282900 | 0.70535500 | -0.33377400 |
| C | -2.36613100 | -0.27753900 | -1.20731200 |
| C | -2.57489100 | 0.90298300 | 0.87332900 |
| C | -3.48544600 | -1.04383900 | -0.88862100 |
| H | -1.85000100 | -0.44796500 | -2.14632700 |
| C | -3.69261200 | 0.13833500 | 1.19677100 |
| H | -2.23685900 | 1.66374000 | 1.56895000 |
| C | -4.15235100 | -0.83895000 | 0.31597800 |
| H | -3.83382500 | -1.80139600 | -1.58165800 |
| H | -4.20687700 | 0.30800700 | 2.13631100 |
| H | -5.02313000 | -1.43390300 | 0.56681800 |
| C | -0.98133300 | 3.05080000 | -0.68657800 |
| H | -1.22038800 | 3.40135600 | 0.32101200 |
| H | -0.12719900 | 3.62451100 | -1.04555200 |
| H | -1.84128200 | 3.26025700 | -1.32687200 |
| H | -0.37898200 | 1.26472500 | -1.71711200 |
|  |  |  |  |

C-H HAT TS (truncated dimer)


TS frequency: $-1557.85 \mathrm{~cm}^{-1}$

| CBS-QB3 Enthalpy $=$ | -843.945793 | CBS-QB3 Free Energy $=$ | -844.015840 |
| :--- | ---: | ---: | ---: |
| 0 2 |  |  |  |
| C | 2.25564500 | -0.81459400 | -1.31202000 |
| C | 3.59164400 | -1.00867700 | -1.27621500 |
| C | 4.34095000 | -0.94392700 | -0.01493100 |
| C | 3.53661400 | -0.67398800 | 1.18686700 |
| C | 2.19873300 | -0.47972300 | 1.13376700 |
| H | 1.72389600 | -0.86184400 | -2.25735800 |
| H | 1.65699900 | -0.30200500 | 2.05014700 |
| C | 1.47427800 | -0.52309700 | -0.12151100 |
| C | 0.11527300 | -0.32322000 | -0.28906400 |
| H | -0.22338500 | -0.43776100 | -1.31608500 |
| O | 5.55794900 | -1.10835800 | 0.03910700 |
| C | -0.95338200 | 0.11339200 | 0.59345900 |
| H | -0.98829400 | 1.32936100 | 0.26126100 |
| O | -0.99624200 | 2.63019100 | -0.02807000 |
| O | -0.04681900 | 2.76958300 | -1.00881500 |
| C | 1.15191800 | 3.32650200 | -0.45026100 |
| H | 1.58793500 | 2.64444600 | 0.28311000 |
| H | 1.82813900 | 3.45106200 | -1.29669400 |
| H | 0.93127600 | 4.29146200 | 0.01023200 |
| C | -0.78138600 | 0.19480600 | 2.11318300 |
| H | -0.60022900 | -0.79575900 | 2.54330900 |
| H | 0.03185600 | 0.86149900 | 2.40191900 |
| H | -1.69744200 | 0.58282100 | 2.56083200 |
| C | -2.32405400 | -0.39026600 | 0.18827000 |
| C | -3.44985900 | 0.43555600 | 0.30836200 |
| C | -2.50351700 | -1.70092100 | -0.27188600 |
| C | -4.71588900 | -0.03568000 | -0.02300300 |
| H | -3.32776700 | 1.46076400 | 0.63874700 |
| C | -3.77214400 | -2.17439200 | -0.60011000 |
| H | -1.64639200 | -2.36002900 | -0.35910000 |
| C | -4.88216500 | -1.34322800 | -0.47742900 |
| H | -5.57354200 | 0.62163600 | 0.06581100 |
| H | -3.89129700 | -3.19433200 | -0.94836100 |
| H | -5.86931600 | -1.70967500 | -0.73514600 |
| H | 4.16560400 | -1.21728600 | -2.17165600 |
|  | 4.07690800 | -0.64693800 | 2.12640000 |

Addition TS (truncated dimer)


TS frequency: $-343.37 \mathrm{~cm}^{-1}$
CBS-QB3 Enthalpy $=-843.960367 \quad$ CBS-QB3 Free Energy $=\quad-844.028986$

| 02 |  |  |  |
| :---: | :---: | :---: | :---: |
| C | 0.96314500 | -0.76857400 | -0.75517100 |
| C | 1.83634300 | -1.75262400 | -1.08048400 |
| C | 2.84270300 | -2.24399300 | -0.12820700 |
| C | 2.84016200 | -1.60372700 | 1.19251200 |
| C | 1.96163900 | -0.62042300 | 1.49770000 |
| H | 0.22717400 | -0.43962100 | -1.47830900 |
| H | 1.82743000 | -2.22622800 | -2.05555400 |
| H | 3.57672000 | -1.95858700 | 1.90398000 |
| H | 1.98057100 | -0.15155200 | 2.47654100 |
| C | 0.97599200 | -0.15653500 | 0.55037100 |
| O | 3.63885800 | -3.13567600 | -0.42560300 |
| O | 1.35833300 | 2.52336000 | 0.56707000 |
| O | 1.45649800 | 2.81734500 | -0.75543500 |
| C | 2.78161800 | 2.51858200 | -1.22898300 |
| H | 2.79283600 | 2.84364900 | -2.26985300 |
| H | 3.51406800 | 3.07645700 | -0.64293800 |
| H | 2.97189500 | 1.44691100 | -1.15775400 |
| C | 0.11111800 | 0.87785900 | 0.91263100 |
| C | -1.10441000 | 1.33029600 | 0.13304000 |
| H | -0.76303100 | 1.59726800 | -0.87257200 |
| H | 0.14440000 | 1.17923200 | 1.95527200 |
| C | -1.73255800 | 2.58771200 | 0.76472600 |
| H | -2.57633300 | 2.93237100 | 0.16356300 |
| H | -2.10800700 | 2.36997500 | 1.76906300 |
| H | -0.99437700 | 3.38714100 | 0.83252300 |
| C | -2.14961600 | 0.22465600 | -0.01552300 |
| C | -2.75441500 | -0.00316100 | -1.25515700 |
| C | -2.56682600 | -0.54188100 | 1.07849300 |
| C | -3.74636600 | -0.97085000 | -1.40242800 |
| H | -2.44737600 | 0.58550300 | -2.11425100 |
| C | -3.55574100 | -1.51108400 | 0.93481700 |
| H | -2.11460600 | -0.38633700 | 2.05227600 |
| C | -4.14944200 | -1.72962700 | -0.30691000 |
| H | -4.20104800 | -1.13240100 | $-2.37351500$ |
| H | -3.86193900 | -2.09794800 | 1.79366400 |
| H | -4.91795600 | -2.48583000 | -0.41895600 |

Addition TS (dimer)


TS frequency: $-398.43 \mathrm{~cm}^{-1}$
DFT Enthalpy $=\quad-1420.677955$
DFT Free Energy $=\quad-1420.774397$

| 02 |  |  |  |
| :--- | ---: | ---: | ---: |
| C | 0.60031500 | 2.05824600 | -0.77530600 |
| C | 0.75794700 | 2.68832800 | -2.01173100 |
| C | 1.55020200 | 3.82910900 | -2.13686000 |
| C | 2.19360300 | 4.35635900 | -1.02139000 |
| C | 2.03749900 | 3.73949600 | 0.21969400 |
| C | 1.24612200 | 2.60228100 | 0.34237400 |
| H | 0.25358100 | 2.28765000 | -2.88553500 |
| H | 1.65988900 | 4.30507700 | -3.10482300 |
| H | 2.81086000 | 5.24239300 | -1.11549500 |
| H | 2.53569300 | 4.14404000 | 1.09344200 |
| H | 1.12928000 | 2.13182200 | 1.31290800 |
| C | -0.26601800 | 0.80745700 | -0.65062800 |
| H | -0.65354100 | 0.58365300 | -1.64854100 |
| C | 0.58847600 | -0.45477700 | -0.21551700 |
| H | 0.70380600 | -0.41268100 | 0.86323600 |
| C | -0.09534900 | -1.77858300 | -0.56242100 |
| C | -0.57209500 | -2.61394400 | 0.45107800 |
| C | -0.23990800 | -2.19606200 | -1.89276900 |
| C | -1.18412900 | -3.82841500 | 0.14669800 |
| H | -0.48070200 | -2.29802400 | 1.48299800 |
| C | -0.85186000 | -3.40854500 | -2.19920000 |
| H | 0.12826900 | -1.57519900 | -2.70275000 |
| C | -1.32742200 | -4.23009800 | -1.17855700 |
| H | -1.55098700 | -4.46020200 | 0.94790300 |
| H | -0.95236300 | -3.71394000 | -3.23473300 |
| H | -1.80262400 | -5.17505700 | -1.41541800 |
| C | 1.94713100 | -0.42463400 | -0.86423100 |
| H | 1.95496400 | -0.26059900 | -1.93908900 |
| C | 3.15304500 | -0.61833900 | -0.27149600 |
| C | 3.33403600 | -0.85067100 | 1.15626800 |
| C | 4.36065200 | -0.59801200 | -1.09043500 |
| C | 4.55484800 | -1.04107200 | 1.69126700 |
| H | 2.45971200 | -0.87193200 | 1.79590100 |
| C | 5.58667100 | -0.78537200 | -0.56832200 |
| H | 4.23704300 | -0.42258000 | -2.15498300 |
| C | 5.78329600 | -1.02375900 | 0.87262700 |
| H | 4.69372400 | -1.21472500 | 2.75235800 |
| H | 6.48169800 | -0.76914800 | -1.17951300 |
| C | -2.79696000 | 0.78924200 | -0.00077400 |
|  |  |  |  |


| C | -3.20596700 | -0.26628500 | -0.89096000 |
| :--- | ---: | ---: | :---: |
| C | -3.83191900 | 1.52426900 | 0.68723200 |
| C | -4.51622300 | -0.55791500 | -1.08332400 |
| H | -2.44820900 | -0.84131600 | -1.40644800 |
| C | -5.14645400 | 1.24916000 | 0.50544900 |
| H | -3.52542700 | 2.32092700 | 1.35695700 |
| C | -5.58433400 | 0.17857900 | -0.39767200 |
| H | -4.83088200 | -1.34823500 | -1.75490300 |
| H | -5.92639000 | 1.80772700 | 1.00998700 |
| O | 6.89284500 | -1.19451600 | 1.36391300 |
| O | -6.77616500 | -0.08491000 | -0.57011800 |
| C | -1.46158100 | 1.14673700 | 0.23837500 |
| H | -1.33283400 | 2.05465700 | 0.81129200 |
| O | -1.02977200 | 0.19349100 | 2.02623500 |
| O | -1.67266300 | 0.82133800 | 3.05310100 |
| C | -2.64791700 | -0.06212400 | 3.63054900 |
| H | -2.15070500 | -0.94196400 | 4.04459000 |
| H | -3.11966800 | 0.51642200 | 4.42559700 |
| H | -3.38238700 | -0.35591500 | 2.87988800 |

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