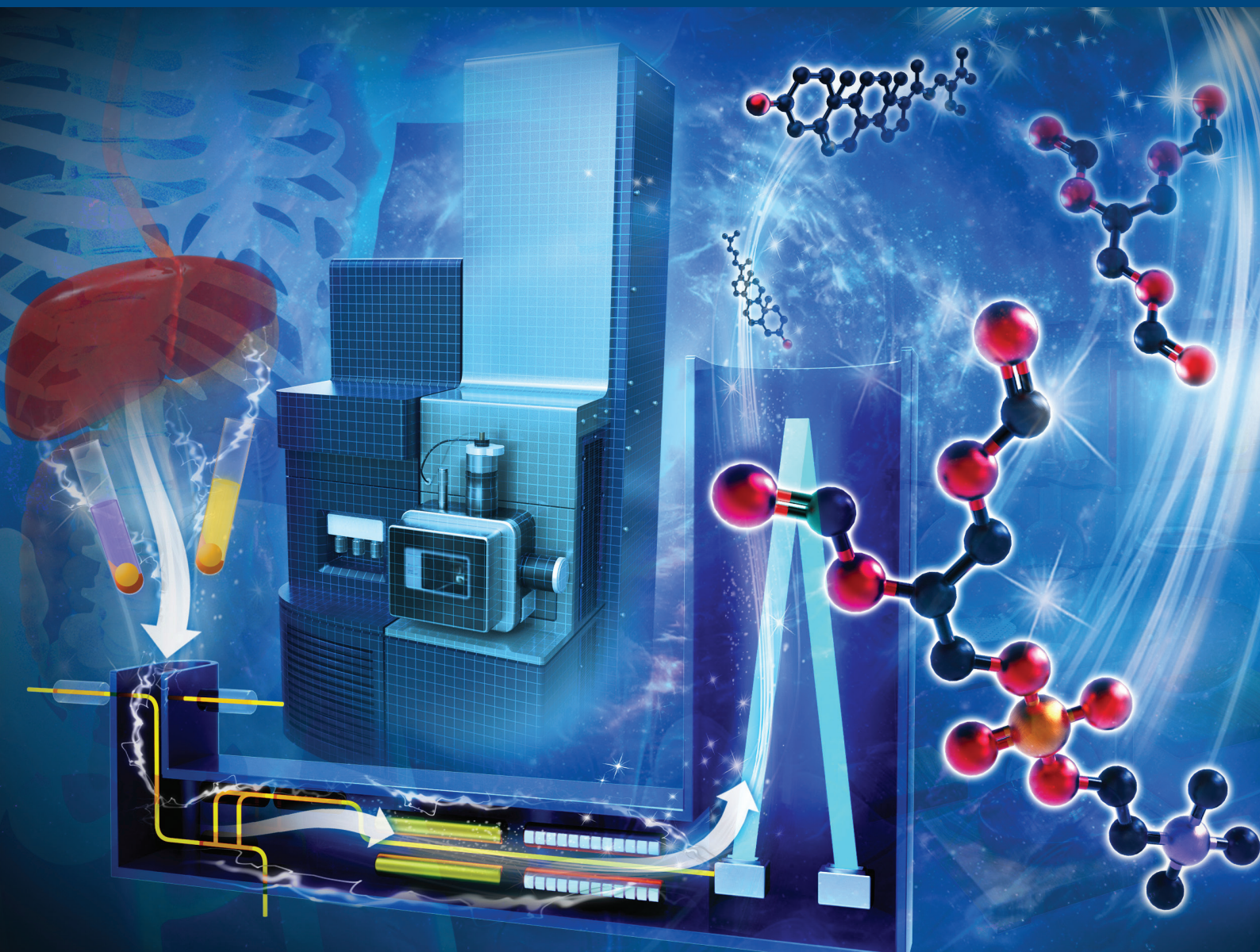


JOURNAL OF SEPARATION SCIENCE

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RESEARCH ARTICLE

Quantitative analysis of γ -glutamylisoleucine, γ -glutamylthreonine, and γ -glutamylvaline in HeLa cells using UHPLC-MS/MS

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Funding information

National Institutes of Health, Grant/Award Numbers: P30DK089503, P30DK081943, T32DK101357, U2C ES026553; Juvenile Diabetes Research Foundation United States of America, Grant/Award Number: 5-COE-2019-861-S-B

γ -Glutamylpeptides have been identified as potential biomarkers for a number of diseases including cancer, diabetes, and liver disease. In this study, we developed and validated a novel quantitative analytical strategy for measuring γ -glutamylisoleucine, γ -glutamylthreonine, and γ -glutamylvaline, all of which have been previously reported as potential biomarkers for prostate cancer in HeLa cells using ultra-high-performance liquid chromatography-tandem mass spectrometry. A BEH C18 column was used as the stationary phase. Mobile phase A was 99:1 water:formic acid and mobile phase B was acetonitrile. Chemical isotope labeling using benzoyl chloride was used as the internal standardization strategy. Sample preparation consisted of the addition of water to a frozen cell pellet, sonication, derivatization, centrifugation, and subsequent addition of an internal standard solution. The method was validated for selectivity, accuracy, precision, linearity, and stability. The determined concentrations of γ -glutamylisoleucine, γ -glutamylthreonine, and γ -glutamylvaline in HeLa cells were 1.92 ± 0.06 , 10.8 ± 0.4 , and 1.96 ± 0.04 pmol/mg protein, respectively. In addition, the qualitative analysis of these analytes in human serum was achieved using a modified sample preparation strategy. To the best of our knowledge, this is the first report of the use of benzoyl chloride for chemical isotope labeling for metabolite quantitation in cells.

KEYWORDS

chemical isotope labeling, derivatization, internal standard, quantification, glutamylpeptides

1 | INTRODUCTION

Glutathione is a tripeptide present in all eukaryotic cells that serves several critical functions including the regula-

tion of the intracellular redox state, the detoxification of reactive oxygen species, and the metabolism of xenobiotic compounds [1,2]. Glutathione is the most abundant thiol species in cells, with intracellular concentrations typically ranging from 0.5 to 10 mM [2]. Glutathione deficiency can severely impair an organism's ability to respond to oxidative stress, which may result in cellular damage and the eventual death of the organism [1]. Glutathione is not efficiently transported across cell membranes and, thus, must be synthesized intracellularly from its constituent amino acids [3].

Article Related Abbreviations: ACN, acetonitrile; BzCl, benzoyl chloride; CIL, chemical isotope labeling; FA, formic acid; LLOQ, lower limit of quantification; LQC, low quality control; MRM, multiple reaction monitoring; PBS, phosphate-buffered saline; γ -Glu-Ile, γ -glutamylisoleucine; γ -Glu-Thr, γ -glutamylthreonine; γ -Glu-Val, γ -glutamylvaline

The metabolites and enzymes involved in the extracellular degradation and intracellular synthesis of glutathione comprise the γ -glutamyl cycle [3]. The first step in the extracellular degradation of glutathione is the transfer of its γ -glutamyl moiety to an acceptor molecule, such as an amino acid, catalyzed by γ -glutamyltransferase [3]. The products of this reaction are cysteinylglycine and, usually, a γ -glutamylpeptide. γ -Glutamyltransferase activity has been reported to correlate positively with alcohol consumption [4], as well as with the incidences of diabetes [5,6], hypertension [7], liver disease [8], cardiovascular disease [9,10], cancer [11], and mortality [12,13]. γ -Glutamyltransferase activity is also affected by age, race, sex, diet, body mass index, and the use of tobacco and oral contraceptives [14].

The influence of patient demographics, lifestyle choices, and various disease states on γ -glutamyltransferase activity limits its usefulness as a single biomarker to diagnose or prognosticate any particular disease [14]. The analysis of its products, γ -glutamylpeptides, is an attractive alternative that provides many potential biomarkers. γ -Glutamylpeptides are also produced via γ -glutamylcysteine synthetase, another enzyme involved in the γ -glutamyl cycle [3]. γ -Glutamylpeptides have been identified as potential biomarkers for aging [15], amyotrophic lateral sclerosis [16], cancer [17–22], diabetes [23], liver disease [24,25], metabolic syndrome [26], obesity [27], and dietary intake [28]. Additionally, the analysis of γ -glutamylpeptides as a biomarker panel may improve the sensitivity and specificity of a clinical test for a particular disease or prognosis. Indeed, Soga et al. were able to discriminate between different forms of liver disease using the serum concentrations of several γ -glutamylpeptides [25]. Interestingly, several studies have shown positive correlations of the concentrations of some γ -glutamylpeptide species with disease states and negative correlations of the concentrations of other species with the same disease state [16,17].

To date, few studies have been reported on the quantitative analysis of γ -glutamylpeptides in biological samples. Soga et al. quantitated γ -glutamylpeptides in human serum from patients with different types of liver diseases as well as healthy controls using LC-MS/MS [25]. The same group later reported a quantitative method for analyzing γ -glutamylpeptides in serum and liver samples using CE-MS/MS [29]. Kobayashi et al. quantitated 21 γ -glutamylpeptides and 45 amino acids in mice plasma and liver samples using LC-MS/MS [30]. *N*-ethylmaleimide was used to derivatize thiol-containing analytes to prevent their oxidation. Saoi et al. used multisegment injection-CE-MS/MS to quantitate 16 γ -glutamylpeptides in serum from nonalcoholic steatohepatitis patients [31]. Interestingly, γ -glutamylpeptides have also been analyzed in foods.

Desfontaine et al. analyzed 27 glutamylpeptides, including 15 γ -glutamylpeptides, in food ingredients using LC-MS/MS [32]. To the best of our knowledge, no method thus far has been reported that provides each γ -glutamylpeptide species with its own internal standard, the absence of which risks inaccuracy due to the susceptibility of electrospray ionization to matrix effects [33,34].

Given the numerous important health implications that have been discovered for γ -glutamylpeptides, there is a critical need for the development of validated methods for their quantitative analysis in biological samples. The aim of this study was to develop a UHPLC-MS/MS method to determine the concentrations of γ -glutamylisoleucine (γ -Glu-Ile), γ -glutamylthreonine (γ -Glu-Thr), and γ -glutamylvaline (γ -Glu-Val) in HeLa cells. All three analytes have been reported as potential biomarkers for fatal prostate cancer [20,22], and their structures are displayed in Figure 1A. Chemical isotope labeling (CIL) using benzoyl chloride (BzCl) was used as the internal standardization strategy, and the reaction scheme is displayed in Figure 1B. To the best of our knowledge, this is the first report of the use of BzCl for CIL for quantitative analysis in cells.

2 | MATERIALS AND METHODS

2.1 | Chemicals and reagents

γ -Glu-Ile ($\geq 95\%$) was purchased from Santa Cruz Biotechnology (Dallas, TX, USA). γ -Glu-Thr ($>95\%$) was synthesized by the Proteomics and Peptide Synthesis Core at the University of Michigan. γ -Glu-Val was purchased from Bachem Americas (Torrance, CA, USA). Acetonitrile (ACN; LC-MS grade), fetal bovine serum, methanol (LC-MS grade), minimum essential medium, penicillin-streptomycin (10 000 U/mL), phosphate-buffered saline (PBS, pH 7.4), trypsin-EDTA (0.25%), and water (LC-MS grade) were purchased from Thermo Fisher Scientific (Waltham, MA, USA). Acetic acid (LC-MS grade) and formic acid (FA; LC-MS grade) were purchased from EMD Millipore Corp. (Burlington, MA, USA). $^{12}\text{C}_6$ -BzCl ($\geq 99\%$), $^{13}\text{C}_6$ -BzCl (99%), and sodium carbonate ($\geq 99.5\%$) were purchased from Sigma-Aldrich (St. Louis, MO, USA). Human serum (pooled) was purchased from Innovative Research (Novi, MI, USA).

2.2 | Preparation of standard solutions

Stock solutions of γ -Glu-Ile, γ -Glu-Thr, and γ -Glu-Val were prepared in water. Derivatized γ -glutamylpeptide standard solutions were each prepared at 500 μM by diluting the standards from the stock solutions to 1 mM at a final

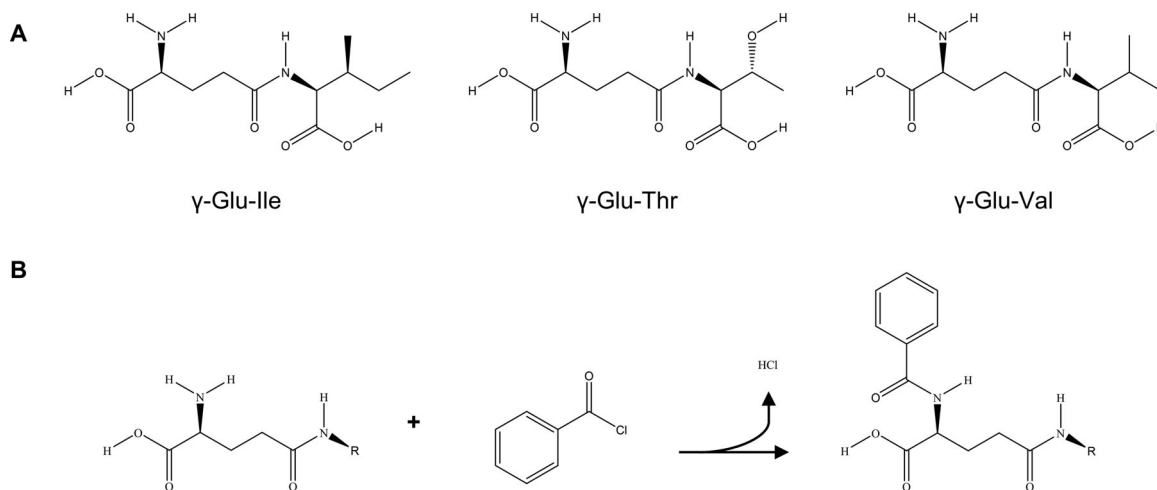


FIGURE 1 (A) Chemical structures of γ -glutamylisoleucine, γ -glutamylthreonine, and γ -glutamylvaline. (B) Reaction of a γ -glutamylpeptide with benzoyl chloride

volume of 75 μ L with water followed by the additions of 37.5 μ L 200 mM sodium carbonate in water and 37.5 μ L 1% v/v $^{12}\text{C}_6$ - or $^{13}\text{C}_6$ -BzCl in ACN. The benzoylation reactions were then allowed to proceed at room temperature for 5 min. The internal standard solution consisted of each $^{13}\text{C}_6$ -BzO- γ -Glu-Ile, $^{13}\text{C}_6$ -BzO- γ -Glu-Thr, and $^{13}\text{C}_6$ -BzO- γ -Glu-Val at 1 μ M in 85:15:1 v/v/v water:ACN:FA. Calibration standard solutions were prepared at 40 μ L, containing 4 μ L of the internal standard solution as well as $^{12}\text{C}_6$ -BzO- γ -Glu-Ile, $^{12}\text{C}_6$ -BzO- γ -Glu-Thr, and $^{12}\text{C}_6$ -BzO- γ -Glu-Val at concentrations ranging from 0.5 to 1000 nM. All solutions were stored at -20°C when not in use.

2.3 | UHPLC-MS/MS analysis

All UHPLC-MS/MS experiments were performed using an Agilent 6495 triple quadrupole mass spectrometer (Agilent Technologies, Santa Clara, CA, USA) equipped with a binary pump (Agilent; 1290 Infinity series), autosampler (Agilent; 1290 Infinity series), and a thermostatted column compartment (Agilent; 1260 Infinity series). An Acquity UPLC BEH C18 column (2.1 \times 100 mm, 1.7 μ m; Waters Corp., Milford, MA, USA) equipped with an Acquity UPLC BEH C18 VanGuard Pre-column (2.1 \times 5 mm, 1.7 μ m; Waters) was used as the LC column. Mobile phase A consisted of 99:1 v/v water:FA, and mobile phase B consisted of ACN. A solution of 80:20 v/v water:ACN was used as the needle wash solvent. The UHPLC gradient program is displayed in Table 1. The following LC parameters were used: injection volume, 1 μ L; autosampler temperature, 4 $^\circ\text{C}$; and column compartment temperature, 15 $^\circ\text{C}$.

MS/MS was performed in the positive ionization multiple reaction monitoring (MRM) mode with the following parameters: collision gas, nitrogen; drying gas temper-

TABLE 1 UHPLC gradient program

Time (min)	%B	Flow rate (mL/min)
0.00	15	0.200
8.00	45	0.200
8.01	45	0.400
8.50	99	0.400
9.50	99	0.400
9.51	15	0.400
10.50	15	0.400
10.51	15	0.200
11.50	15	0.200

ature, 290 $^\circ\text{C}$; sheath gas temperature, 400 $^\circ\text{C}$; drying gas flow rate, 12 L/min; sheath gas flow rate, 12 L/min; nebulizer pressure, 30 psi; capillary voltage, 3500 V; nozzle voltage, 500 V; and dwell time, 50 ms. Both quantitative and qualitative MRM transitions were monitored for each $^{12}\text{C}_6$ -/ $^{13}\text{C}_6$ -benzoylated analyte/internal standard. The $[\text{M}+\text{H}]^+$ specie was monitored as the precursor ion for each MRM transition. MRM transitions for each $^{12}\text{C}_6$ -/ $^{13}\text{C}_6$ -benzoylated analyte/internal standard were monitored within their own time segment. MRM parameters are displayed in Table 2.

2.4 | Cell culture

HeLa cells were cultured in minimum essential medium supplemented with 10% w/v fetal BS and 100 U/mL of each penicillin and streptomycin at 37 $^\circ\text{C}$ in 5% CO_2 to a confluence of 70%. Plates were then washed twice with PBS at 37 $^\circ\text{C}$, followed by the addition of trypsin-EDTA

TABLE 2 Multiple reaction monitoring (MRM) parameters for the $^{12}\text{C}_6$ -/ $^{13}\text{C}_6$ -benzoylated analytes/ internal standards

Time segment (min)	Compound	Retention time (min)	Quant. MRM	Qual. MRM
0.00→4.49	$^{12}\text{C}_6$ -BzO- γ -Glu-Thr	3.26	353.0→186.1; 16 V	353.0→140.0; 30 V
0.00→4.49	$^{13}\text{C}_6$ -BzO- γ -Glu-Thr	3.26	359.1→111.0; 35 V	359.1→341.0; 8 V
4.50→5.99	$^{12}\text{C}_6$ -BzO- γ -Glu-Val	5.53	351.0→156.2; 22 V	351.0→229.1; 18 V
4.50→5.99	$^{13}\text{C}_6$ -BzO- γ -Glu-Val	5.53	357.2→311.1; 12 V	357.2→111.1; 35 V
6.00→7.99	$^{12}\text{C}_6$ -BzO- γ -Glu-Ile	6.65	365.1→86.0; 16 V	365.1→170.1; 25 V
6.00→7.99	$^{13}\text{C}_6$ -BzO- γ -Glu-Ile	6.65	371.2→111.0; 35 V	371.2→325.2; 12 V

Quant., quantitative; Qual., qualitative.

at 37°C until the cells were detached. Ice-cold serum-free minimum essential medium equal to the volume of trypsin-EDTA was then added, and the cells were gently aspirated to separate from each other and the plate. Cells were then counted via hemocytometry, washed twice in ice-cold PBS, and resuspended in PBS to a concentration of 2×10^6 cells/mL. Aliquots of 1 mL were then collected into cryovials, centrifuged to remove the PBS, flash-frozen in liquid nitrogen, and stored at -80°C .

2.5 | Sample preparation

HeLa cells were taken out of storage at -80°C immediately followed by the addition of 55 μL of ice-cold water and bath sonication for 30 s. Samples were then placed on ice, and a 5 μL aliquot to assay protein content (DC Protein Assay; Bio-Rad, Hercules, CA, USA) was taken and stored at -20°C . Ice-cold 200 mM sodium carbonate in water (25 μL) was then added followed by 25 μL 1% v/v $^{12}\text{C}_6$ -BzCl in ACN. The sample was then inverted and vortexed for 10 s and then held at room temperature for 5 min to allow for the reaction to complete. The sample was then centrifuged for 10 min at $17\,100 \times g$ and 4°C , and 18 μL of the supernatant was then mixed with 2 μL of the internal standard solution followed by centrifugation for 10 min at $17\,100 \times g$ and 4°C . Finally, the supernatant was transferred to an autosampler vial and analyzed by UHPLC-MS/MS. Four HeLa cell samples were analyzed, subjected to ten injections each, on consecutive days. $^{12}\text{C}_6$ -BzCl and sodium carbonate solutions were prepared fresh daily.

Human serum was also prepared and analyzed to assess the methodology's applicability to other biological matrices. A 50 μL aliquot of thawed serum was added to 200 μL ice-cold 0.1% v/v acetic acid in methanol for protein precipitation. The sample was then vortexed and incubated on ice for 5 min followed by centrifugation for 5 min at $17\,100 \times g$ and 4°C . Note that 50 μL of the supernatant was then mixed with 25 μL 200 mM sodium carbonate in water followed by the addition of 25 μL 1% v/v $^{12}\text{C}_6$ -BzCl in ACN. The sample was then inverted, vortexed, and held at room

temperature for 5 min. Finally, 90 μL of the sample was added to 10 μL of the internal standard solution, and the sample was centrifuged for 5 min at $17\,100 \times g$ and 4°C prior to analysis by UHPLC-MS/MS.

2.6 | Data analysis

Peak areas, retention times, and S/N ratios were obtained using MassHunter Quantitative Analysis software (B.07.00; Agilent) and were all reviewed manually. MS response henceforth refers to the ratio of the peak areas for the quantitative MRM transition of the $^{12}\text{C}_6$ -benzoylated analyte to that of its $^{13}\text{C}_6$ -benzoylated internal standard. The concentrations used in the calibration standard solutions were: 0, 0.5, 1, 2.5, 5, 10, 25, 50, 100, 250, 500, and 1000 nM. For validation experiments, the concentrations used in the calibration curves were 0 and 0.5–1000 nM for $^{12}\text{C}_6$ -BzO- γ -Glu-Ile, and 0 and 1–1000 nM for $^{12}\text{C}_6$ -BzO- γ -Glu-Thr and $^{12}\text{C}_6$ -BzO- γ -Glu-Val. For the determination of the analyte concentration in samples, seven-point calibration curves were constructed from calibration standard solutions and spanned the concentration ranges of 1–100, 5–500, and 2.5–250 nM for $^{12}\text{C}_6$ -BzO- γ -Glu-Ile, $^{12}\text{C}_6$ -BzO- γ -Glu-Thr, and $^{12}\text{C}_6$ -BzO- γ -Glu-Val, respectively. Each calibrant was analyzed in triplicate. Weighted least-squares regression ($1/x^2$) was used to obtain the calibration equations. Recoveries were taken into account when determining analyte concentrations in samples. Analyte concentrations in samples were normalized to their protein content obtained from the DC Protein Assay.

2.7 | Method validation

Method validation was conducted as much as possible in accordance with the US Food and Drug Administration (FDA) guidelines for bioanalytical method validation for industry [35]. However, the lack of a commercially available matrix-free of the endogenous analytes prevented the ability to follow FDA guidelines requiring an analyte-free

blank of the same biological matrix. As recommended by the European Commission, selectivity was assessed by verifying that the retention times and the relative peak areas of the quantitative and qualitative MRM transitions for each $^{12}\text{C}_6$ -benzoylated analyte measured in HeLa cells were within 2.5 and 20% of those values measured in 25 nM calibration standard solutions, respectively [36]. In addition, blanks consisting of water prepared in the same manner as samples were also analyzed to ensure that the reagents and internal standard solution did not contribute to the MS response of samples.

For accuracy and precision experiments, quality control standard solutions were separately prepared from the calibration standard solutions and analyzed five times each. QC standards at concentrations of 0.5, 1, 2.5, and 5 nM were used to determine the lower LOQs (LLOQs) and low QC (LQC) concentrations for each analyte. LLOQs were determined as the lowest analyte concentration at which their accuracies based on their experimentally determined concentrations as well as their RSDs were 100 ± 20 and $<20\%$, respectively, and whose next highest concentration, the LQC concentration, had values of $100 \pm 15\%$ and $<15\%$, respectively. Medium QC and high QC concentrations were 500 and 800 nM, respectively, for all analytes and were subjected to the same criteria as the LQC. These experiments were performed on three separate days with freshly prepared QC and calibration standards. The intraday accuracy and precision values were obtained from the experiments on the first day, and the interday values were obtained throughout the 3 days.

Autosampler stability was assessed by analyzing three HeLa samples immediately after preparation followed by reanalysis of the same samples 12 and 24 h later. Freeze-thaw stability was assessed by subjecting three samples to three freeze-thaw cycles with at least 12 h between cycles prior to processing and analysis. For long-term stability, three samples were kept in storage at -80°C and processed and analyzed 2 weeks after the autosampler stability experiment. All samples were analyzed five times for each treatment.

Spike and recovery was used to determine the recoveries of the analytes in HeLa cells. Water (55 μL) was added to two HeLa samples followed by bath sonication for 30 s. The samples were then pooled and split into eight 12 μL subsamples. Four of these subsamples, the prederivatization spiked subsamples, were spiked with 2 μL underivatized analyte solution at concentrations of 0, 1.75, 8.75, and 17.5 μM such that their final concentrations, assuming 100% recoveries, were 0, 100, 500, and 1000 nM. Water (2 μL) was added to the other four subsamples, the postderivatization spiked subsamples. To each subsample, 7 μL 200 mM sodium carbonate was added, followed by 7 μL 1% v/v $^{12}\text{C}_6$ -BzCl in ACN. The samples were vortexed,

allowed 5 min to react, and centrifuged. The supernatant (16 μL) was then taken from each subsample, and 2 μL water was added to the prederivatization subsamples. To the postderivatization spiked subsamples, 2 μL of $^{12}\text{C}_6$ -benzoylated analytes at concentrations of 0, 1, 5, and 10 μM were added, such that their final concentrations were the same as the prederivatization spiked subsamples. Finally, 2 μL of the internal standard solution was added prior to analysis. The recoveries were determined by dividing the slope of the MS response versus spiked concentration of the prederivatization spiked subsamples by those of the postderivatization spiked subsamples. Matrix effects were assessed by comparing the slopes of the calibration curve to that of the postderivatization spiked subsamples.

Accuracy was also assessed via standard addition where a HeLa sample was split into three equal volumes and spiked prior to derivatization with underivatized analytes such that their final concentrations were 0, 25, and 50 nM for γ -Glu-Ile and γ -Glu-Val, and 0, 250, and 500 nM for γ -Glu-Thr. The concentrations determined by standard addition were assumed as the true concentrations, and this was compared to the concentration obtained from the calibration curves, corrected for recovery, for the sample spiked at 0 nM.

We systematically assessed whether the reactions of BzCl and the γ -glutamylpeptide analytes were quantitative. Briefly, 55 μL 10 μM underivatized analyte solution was added to three HeLa samples followed by bath sonication. Each sample was then split into two 20 μL subsamples followed by the addition of 10 μL 200 mM sodium carbonate. For subsample A, 10 μL 1% v/v $^{13}\text{C}_6$ -BzCl in ACN was added, and for subsample B, 10 μL ACN was added. The subsamples were then vortexed and the reactions were allowed to proceed for 5 min. After centrifugation, supernatant from each subsample (5 μL) was taken and diluted with 245 μL water followed by the additions of 125 μL 200 mM sodium carbonate and 125 μL 1% v/v $^{12}\text{C}_6$ -BzCl in ACN. The reactions were allowed to proceed for 5 min prior to analysis.

3 | RESULTS AND DISCUSSION

3.1 | Method development

One of the most formidable challenges in quantitative LC-MS/MS method development for biological samples is correcting for matrix effects. This is most often achieved through the use of isotopically labeled internal standards. These internal standards were not commercially available for the investigated analytes, but their possession of nucleophilic primary amine functional groups allowed for the use of CIL as an internal standardization strategy.

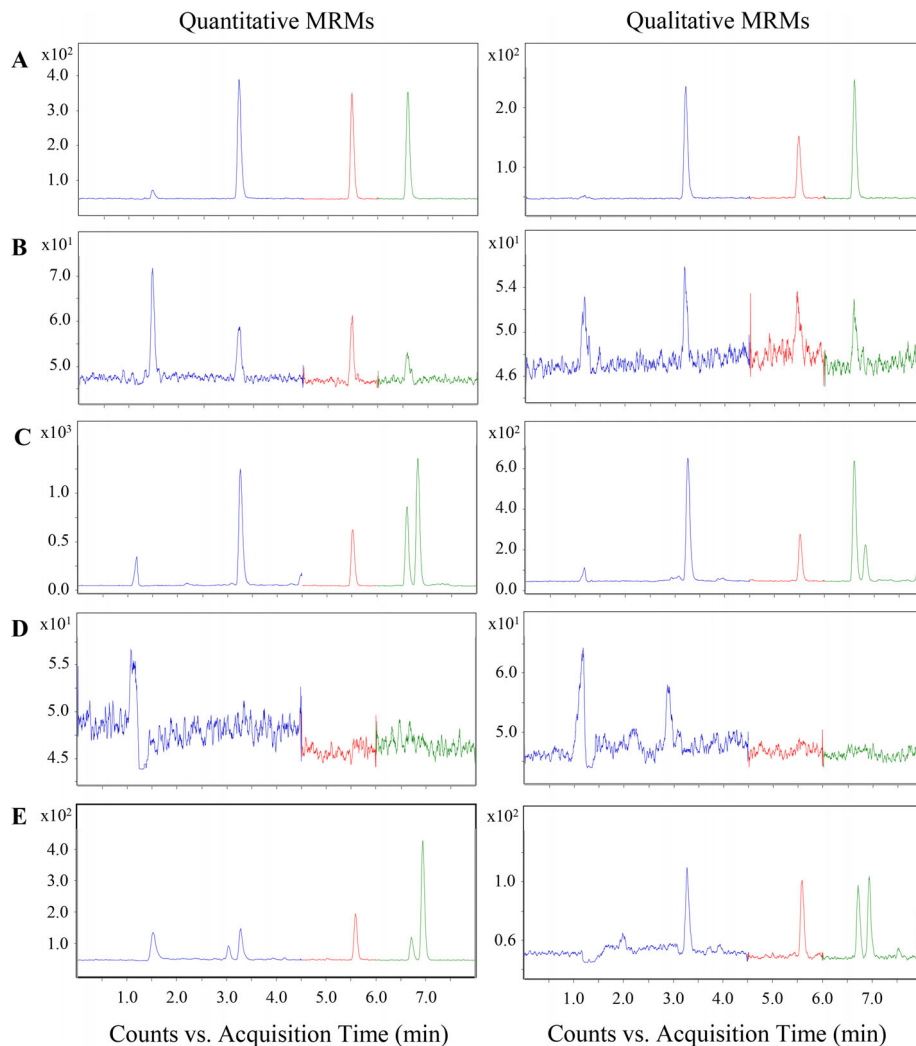


FIGURE 2 UHPLC-MS/MS chromatograms of (A) 25 nM standard solution, (B) standard solution at the lower LOQs (LLOQs), (C) HeLa sample, (D) HeLa sample without the addition of benzoyl chloride or internal standard solution, and (E) human serum sample

Additional benefits of derivatizing hydrophilic analytes with hydrophobic moieties, such as benzoyl or dansyl groups, include greater RPLC retention and improved ESI-MS sensitivity [37–40]. The use of BzCl as a derivatizing agent for CIL had previously been effectively demonstrated for analytes possessing a variety of nucleophilic functional groups including primary and secondary amines, phenols, thiols, and some alcohols [37,41,42]. Unlike other commonly used derivatizing reagents for CIL, such as dansyl chloride, BzCl has the advantages of commercial availability both as isotopically labeled and unlabeled, and its reaction with nucleophiles proceeds rapidly at room temperature [37]. For these reasons, BzCl was chosen as the derivatizing reagent.

Several RPLC columns were screened during method development. The Acquity UPLC BEH C18 column resulted in narrow, symmetrical chromatographic peaks and was ultimately chosen. Mobile phase flow rates were

increased after the elution of analytes during the washing and re-equilibration steps in order to shorten the analysis time. Based on previous experience with UHPLC columns, the last minute of re-equilibration was set at the initial flow rate to reduce the elevated column temperature caused by frictional heating at higher flow rates, which may have otherwise had a detrimental impact on efficiency and reproducibility [43]. Chromatograms of a standard solution at 25 nM, a standard solution at the LLOQs, a HeLa sample, a HeLa sample without the addition of BzCl or an internal standard, and a human serum sample are displayed in Figure 2. All targeted analytes were baseline separated from closely eluting interferences in the HeLa cell sample as well as the serum sample. Interestingly, an interferent that eluted immediately after $^{12}\text{C}_6$ -BzO- γ -Glu-Ile was present in the quantitative and qualitative MRM chromatograms for both the HeLa cell and serum sample. No attempt was made to identify this interferent, but given

TABLE 3 Results from accuracy and precision experiments

Analyte	Concentration (nM)	Intraday		Interday	
		Accuracy (%)	RSD (%)	Accuracy (%)	RSD (%)
BzO- γ -Glu-Ile	0.5	101.6	10.6	105.0	7.4
	1	101.6	5.9	103.1	5.8
	500	102.9	1.8	100.1	8.1
	800	108.3	1.4	101.0	7.3
BzO- γ -Glu-Thr	1	95.2	5.4	103.7	13.2
	2.5	106.8	6.3	104.0	10.4
	500	100.5	1.1	98.5	9.0
	800	104.2	0.8	99.9	5.7
BzO- γ -Glu-Val	1	97.6	6.4	101.1	5.6
	2.5	96.8	5.1	100.7	9.6
	500	102.6	2.4	99.3	7.9
	800	108.7	2.2	101.2	6.9

its similar retention time to and response to both MRM channels of $^{12}\text{C}_6$ -BzO- γ -Glu-Ile, it is possibly the benzylation product of γ -glutamylleucine. For this analytical challenge, the high efficiency of UHPLC, imparted by its sub-2 μm particles, may have been necessary to baseline separate the quantitative MRM chromatographic peaks of $^{12}\text{C}_6$ - γ -Glu-Ile from the closely eluting interferent, whereas HPLC, with larger sized particles, would likely not have provided the sufficient number of theoretical plates for these separations [44].

3.2 | Method validation

The retention times and the relative ion intensities of the quantitative and qualitative MRM transitions for each $^{12}\text{C}_6$ -benzoylated analyte measured in HeLa samples differed from those measured in 25 nM calibration standard solutions <1 and 15%, respectively, demonstrating acceptable selectivity. In addition, the ion intensities of the quantitative MRM transition in blanks were <1% of those of HeLa samples, indicating that the contributions of the reagents and internal standard to the sample measurement are negligible.

Results from the accuracy and precision experiments are displayed in Table 3. The LLOQ for γ -Glu-Ile was 0.5 nM and those for γ -Glu-Thr and γ -Glu-Val were 1 nM. All QC concentrations for all analytes for both the intra- and interday datasets had an accuracy of $100 \pm 10\%$ and an RSD of <15%. The linearities of the calibration curves throughout the study were excellent, with $r^2 > 0.99$. Although 1000 nM was chosen as the highest calibrant concentration, the calibration curves remained linear ($r^2 > 0.99$) up to 100 μM , the highest standard solution concentra-

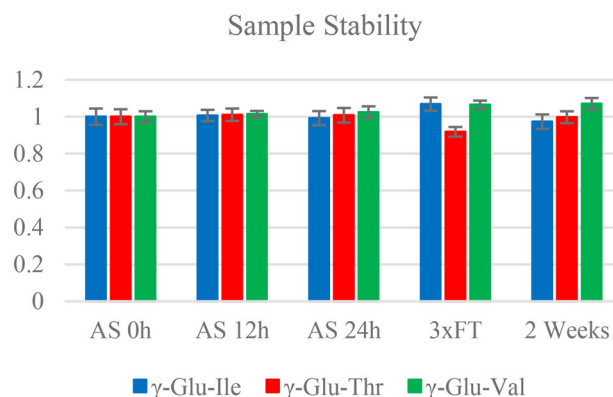


FIGURE 3 Relative analyte concentrations in HeLa cells: processed and stored in the autosampler (AS; 4°C) for 0, 12, and 24 h; subjected to three freeze-thaw cycles (3 \times FT) prior to sample preparation, and processed after storage of the pellet at -80°C for 2 weeks

tion analyzed. Accuracies determined via standard addition were 91.8, 82.6, and 106.0 for γ -Glu-Ile, γ -Glu-Thr, and γ -Glu-Val, respectively. The linearities in the standard addition experiments were excellent, with $r^2 > 0.999$.

Figure 3 displays the results from the sample stability experiments. Sample concentrations after 12 and 24 h in an autosampler (4°C), three freeze-thaw cycles, and storage at -80°C for 2 weeks were all within 10% of the concentration of freshly processed samples, demonstrating excellent autosampler, freeze-thaw, and long-term stabilities.

Recoveries for γ -Glu-Ile, γ -Glu-Thr, and γ -Glu-Val were 82.0, 90.8, and 87.5%, respectively. Linearities for the pre- and postderivatization spiked sample subsets ($n = 4$ each) were excellent, with $r^2 > 0.999$. This is indicative that the benzylation reaction is highly reproducible. The % recoveries decreased with increasing retention time.

TABLE 4 Analyte concentrations in HeLa cells

Compound	Concentration (pmol/ mg protein) \pm SD				
	HeLa 1	HeLa 2	HeLa 3	HeLa 4	Average
γ -Glu-Ile	1.949 \pm 0.074	1.913 \pm 0.076	1.838 \pm 0.091	1.970 \pm 0.050	1.917 \pm 0.058
γ -Glu-Thr	10.58 \pm 0.67	10.52 \pm 0.22	10.51 \pm 0.32	11.41 \pm 0.60	10.76 \pm 0.44
γ -Glu-Val	2.01 \pm 0.13	1.962 \pm 0.062	1.92 \pm 0.11	1.92 \pm 0.10	1.955 \pm 0.042

SD, standard deviation.

This relationship may be due to the increase in analyte hydrophobicity with increased retention time [45], which may also result in increased adsorption to precipitates formed during sample preparation and other insoluble material. If this is the case, analyte recovery may be dependent upon a sample's concentration of cells during sample preparation, so it would be critical for all samples to have the same concentration of cells like those in the recovery experiments. The slope of the calibration curve differed by <10% to that of the postderivatization spiked subsamples for all analytes, indicating a minimal influence of matrix effects on the MS response in HeLa samples.

The benzylation reaction of the analytes in the HeLa cell matrix was found to be nearly quantitative. For the A subsamples (see Section 2.7 for details), which were first reacted with $^{13}\text{C}_6$ -BzCl prior to dilution and reaction with $^{12}\text{C}_6$ -BzCl, the peak areas for the $^{12}\text{C}_6$ -benzoylated analytes were <2% of those of the B subsamples, which were not reacted with $^{13}\text{C}_6$ -BzCl prior to their dilution and reaction with $^{12}\text{C}_6$ -BzCl.

3.3 | Analysis of HeLa cells

The aforementioned validated UHPLC-MS/MS method was applied to four HeLa samples measured ten times each on consecutive days. Measured analyte concentrations in HeLa cells are presented in Table 4. Average concentrations of γ -Glu-Ile, γ -Glu-Thr, and γ -Glu-Val in HeLa cells were 1.92, 10.8, and 1.96 pmol/mg protein, respectively. RSDs for all analyte concentrations in all samples were <10%, and the RSDs for concentration among the four analyzed HeLa samples were <5% for all analytes. To the best of our knowledge, this is the first reported validated method for the quantitation of any metabolites in cells using BzCl for CIL.

3.4 | Future perspective

Most metabolomics studies to date have been semiquantitative, which severely limits their usefulness when compared to one another. The development and adoption

of more quantitative metabolomics methods would allow for meta-analyses of studies using different sample preparation techniques, instrumentation, and instrumental parameters, which could be immensely beneficial for biomarker discovery and the elucidation of molecular mechanisms of pathogenesis. Correcting for matrix effects continues to be the greatest obstacle for quantitative metabolomics using LC-MS/MS and is most commonly addressed through the use of isotopically labeled internal standards. CIL is an extremely promising internal standardization strategy, allowing for the generation of labeled internal standards from relatively inexpensive labeled reagents and unlabeled standards. While BzCl has been demonstrated to be a convenient CIL reagent for analytes possessing nucleophilic functional groups, there is a need for the development of new methods using reagents that react quantitatively, rapidly, predictably, and under mild conditions with other functional groups.

Although the goal of the present study was to develop, validate, and apply a quantitative LC-MS/MS method for the analysis of γ -Glu-Ile, γ -Glu-Thr, and γ -Glu-Val in HeLa cells, this methodology is likely applicable to other nucleophilic compounds and other biological matrices. The broadness of the methodology's applicability has been supported by its successful application to serum, with a modified sample preparation strategy, as well as the apparent detection of γ -glutamylleucine in both HeLa cells and serum. Application to other biological matrices will require sample preparation optimization and method validation. Other γ -glutamylpeptides could likely be added as targeted analytes, though some amino acids require special considerations. The side chains of cysteine, lysine, and tyrosine contain a thiol, primary amine, and phenol functional group, respectively, which are all reactive to BzCl [41]; the benzylation product of a γ -glutamylpeptide will contain an additional benzoyl group for each of these amino acids. In addition, cysteine is particularly susceptible to oxidation, and consequently, a carefully considered and optimized sample preparation strategy and thorough method validation are critical. Obviously, this strategy requires the purchase or synthesis of unlabeled targeted analytes, but these are far less expensive than their labeled analogs.

4 | CONCLUDING REMARKS

In this report, we describe the development and validation of a method for the quantitative analysis of γ -Glu-Ile, γ -Glu-Thr, and γ -Glu-Val using UHPLC-MS/MS in HeLa cells. The method used CIL using BzCl as its internal standardization strategy, and sample preparation consisted of the addition of water, sonication, derivatization, centrifugation, and the addition of internal standard solution. This methodology would also likely work well for other peptide species as well as reducing carbohydrates, polyamines, and other metabolites possessing nucleophilic functional groups. In addition, we have demonstrated the applicability of the method to serum with a modified sample preparation strategy. In the future, we would like to apply this method to human cells from clinical samples and include additional γ -glutamyl peptides in the assay.

ACKNOWLEDGMENTS

This work was supported by the National Institutes of Health grants P30DK089503, P30DK081943, U2C ES026553, and T32DK101357 (to Chenchen He) and the Juvenile Diabetes Research Foundation United States of America grant 5-COE-2019-861-S-B (to Subramaniam Pennathur). The authors would also like to thank Nancy Roeser for her help with experimental procedures.

CONFLICT OF INTEREST

The authors have declared no conflict of interest.

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How to cite this article: Thacker JB, He C, Pennathur S. Quantitative analysis of γ -glutamylisoleucine, γ -glutamylthreonine, and γ -glutamylvaline in HeLa cells using UHPLC-MS/MS. *J Sep Sci*. 2021;44:2898–2907. <https://doi.org/10.1002/jssc.202001266>