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Implications for post-comminution processes in subglacial suspended sediment using coupled radiogenic strontium and neodymium isotopes

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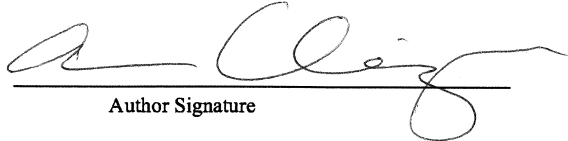
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1 **Abstract**

2 Enhanced physical weathering rates in subglacial systems promote high levels of
3 comminution, transport, and deposition of fine-grained sediment within the subglacial
4 drainage network. The impact of shifts in sediment loads due to variations in meltwater
5 flux, and their effects on downstream ecosystems, remains poorly quantified and places a
6 fundamental importance on our ability to characterize subglacial depositional
7 environments. Here, for the first time, we assess the seasonal evolution of the subglacial
8 suspended sediment using coupled radiogenic strontium ($^{87}\text{Sr}/^{86}\text{Sr}$) and neodymium
9 ($^{143}\text{Nd}/^{144}\text{Nd}$) isotopic ratios with elemental ratios, and in-situ measurements. Weathering
10 rates in fluvial and riverine systems have been traditionally assessed using radiogenic
11 isotopic tracers: $^{143}\text{Nd}/^{144}\text{Nd}$ ratios relate to the crustal age whereas $^{87}\text{Sr}/^{86}\text{Sr}$ ratios relates
12 to both age and preferential mineral dissolution. Relative shifts in these ratios allow us to
13 characterize distinct sediment transport networks. We apply this technique to the Lemon
14 Creek Glacier (LCG), Alaska, USA and the Athabasca Glacier (AG), Alberta, CA. At the
15 LCG, the $^{143}\text{Nd}/^{144}\text{Nd}$ values range from ϵ_{Nd} of -4.6 (0.9) to -8.7 (0.2), which suggests a
16 poorly mixed sediment flux. However, the greatest period of variability may correlate
17 with the drainage of a supraglacial lake and suggests caution should be exerted in time-
18 scale $^{143}\text{Nd}/^{144}\text{Nd}$ provenance studies that may be affected by climatic disturbances. In
19 contrast, limited variation is observed within the AG $^{143}\text{Nd}/^{144}\text{Nd}$ seasonal record. A
20 consistent, direct relation between the Rb/Sr elemental ratio and the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio enables
21 us to unravel incongruent weathering trends in the radiogenic Sr record. Correlation
22 between the $^{87}\text{Sr}/^{86}\text{Sr}$ and total discharge suggests the process is partially controlled by
23 mantling of the bedrock, which can be detected using post-comminution ages. While the
24 subglacial structure may be enabled by the subglacial till beneath the AG, our study
25 supports the use of Sr-Nd as a new proxy in the subglacial environment.

26
27 **Keywords:** Subglacial environment; radiogenic isotopes; suspended sediment;
28 comminution;

29
30

31

32 **1. Introduction**

33 The predicted escalation of glacial retreat by the end of the twentieth century
34 (IPCC, 2013) highlights the fundamental importance of quantifying how the environment
35 is impacted by an increase in glacio-fluvial sediment deposition. In the subglacial
36 hydrologic system, elevated pressures can promote the presence and motion of water at
37 the base of the glacier. In turn, this meltwater flux facilitates the production and transfer
38 of dissolved and suspended sediment (e.g. Collins, 1990), as well as significant physical
39 and chemical weathering between bedrock and basal ice. Shifts in meltwater
40 hydrochemistry and sediment load can have direct and profound influences on the nature
41 of downstream ecosystems (e.g. Jacobsen et al., 2012; Muhlfeld et al., 2011; Xu et al.,
42 2009). The impact of glacially derived sediment release is widely variable. For example,
43 high sediment loads can disrupt salmon spawning grounds, whereas reductions in
44 meltwater quality can destabilize benthic communities (Milner et al., 2009). Additionally,
45 sediment flux can influence the degree and stability of channel distribution in pro-glacial
46 riparian zones (Milner et al., 2009). Investigating the rate and scale at which sediment
47 deposition, associated with hydrologic change, is occurring may provide insight into the
48 sensitivity of downstream ecosystems to sediment flux.

49 While certain aspects of subglacial hydrology, such as water transit velocity,
50 water quality, and subglacial water pressures, have been well studied (e.g. Anderson et
51 al., 2004; Anderson, 2007; Brown, 2002; Hodge, 1976; Lamb et al., 1995; Moore et al.,
52 2013; Stenborg, 1969; Tranter, 2005), the spatial distribution of the subglacial
53 hydrological networks is less well constrained. In general, glaciers can be characterized
54 by the annual presence of basal water. Cold-based temperatures glaciers generally lie

55 below the pressure melting point throughout the entire year, whereas warm-based
56 temperatures glaciers typically reach pressure melting point throughout the entire year,
57 which promotes the presence of basal meltwater (e.g. Tranter, 2003). Poly-thermal
58 glaciers exhibit both conditions per annum (e.g. Wadham et al., 1998). In systems where
59 subglacial water is present, basal water is routed beneath glaciers and meltwater conduits
60 can carve the glacial bedrock. These subglacial drainage systems can be classified into
61 two categories: a distributed, slow-transit hydrologic system and a channelized, rapid-
62 transit system (Raymond et al., 1995). However, subglacial systems cannot be defined
63 seasonally or spatially by a single configuration, as the subglacial drainage network is
64 constantly evolving (Fountain and Walder, 1998). In particular, dye-trace studies have
65 revealed a hydrologic dispersion through the subglacial drainage network changes during
66 the melt season where fast, efficient channels tend to dominate slow, inefficient networks
67 as the melt season progresses and the subglacial drainage channels expand up-glacier
68 (Bingham et al., 2005; Nienow et al., 1998). Hydrologic variability and glacial
69 sedimentary output are directly linked; comminution of rock occurs as glacial ice
70 physically abrades and meltwater chemically weathers the bedrock. Here, the resulting
71 fresh and highly reactive sediment contributes to the dissolved and suspended sediment
72 loads.

73 The geochemical behavior of glacially derived sediment has been increasingly
74 studied over the past two decades (e.g. Brown et al., 1994; Hodgkins et al., 1995; Tranter
75 et al., 1993; Tranter et al., 1995; Tranter et al., 2005). Yet, these limited number of
76 studies, which characterize the subglacial depositional environment, have focused largely
77 on the deformation of the glacial bed and till (e.g. Boulton et al., 2001; Evans et al., 2006;

78 van der Meer et al., 2003). Even fewer studies have addressed the seasonal dynamics of
79 sediment entrainment, transport, and deposition in the subglacial environment (e.g. Alley
80 et al., 1997; Collins, 1988; Lawson, 1993; Swift et al., 2002) or explored the coupled
81 chemical-physical behavior of the subglacial sediment load (Brown et al., 1996).

82 If subglacial hydrological network processes mirror those of subaerial or riverine
83 environments, they may exhibit similar depositional characteristics such as mantling of
84 the bedrock or mineral sorting between grain sizes. Therefore the sediment depositional
85 environment should also vary, like riverine environments, with location and lithology.

86 But, glacial environments notably differ from riverine environments in that glacial
87 abrasion leads to relatively high physical and low chemical weathering rates (Raiswell et
88 al., 2006), a ratio that likely contributes defining characteristics to the sediment flux in
89 the subglacial environment. The production of sediment through comminution processes
90 leads to the exposure of fresh, reactive surfaces that can be deposited or excavated
91 throughout the glacial system. While subglacial sediment deposition may simply be a
92 function of the net sediment flux into a given subglacial area (Hart, 1995), entrainment
93 occurs when a critical shear stress at the bed is exceeded (Walder and Fowler, 1994).

94 Further, during physical and chemical weathering processes, minerals entrained
95 within the bedrock can be weathered both congruently and incongruently. Congruent
96 weathering relates to processes involving the complete dissolution of minerals, whereas
97 incongruent weathering relates to the conversion of an initial mineral into a secondary
98 mineral through precipitation and dissolution processes. However, discerning the
99 contribution of these two processes to the dissolved load and suspended sediments
100 remains a challenge (e.g. Hindshaw et al., 2014), but the relative contributions may pose

101 an interesting relationship if the sediment behaves in response to the high levels of
102 comminution.

103 As tracers of source and weathering processes in fluvial environments, radiogenic
104 isotopes have the potential for uncovering weathering and sediment transport processes
105 within the subglacial environment. Weathering rates in fluvial and riverine systems have
106 been assessed using radiogenic isotopic tracers (e.g. Derry and France-Lanord, 1996;
107 Edmond, 1992).

108 Strontium isotope ($^{87}\text{Sr}/^{86}\text{Sr}$) fluxes have typically been analyzed in stream and
109 river waters to trace mineral weathering reactions and rates and compare them to the
110 isotopic characteristics of the underlying and local bedrock (e.g. Arn et al., 2003; Blum
111 and Erel, 1995; Blum et al., 1993; Clow et al., 1997a; Stevenson et al., In review-b;
112 Taylor and Blum, 1995). Due to the radiogenic decay of ^{87}Rb to ^{87}Sr , the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio
113 can also be used to place constraints on lithographic age. Radiogenic Sr is preferentially,
114 incongruently, released from minerals such as biotite during periods of high weathering
115 (e.g. Peucker-Ehrenbrink and Blum, 1998) . Potassium and calcium ions in minerals can
116 readily substitute for Rb and Sr, respectively; therefore $^{87}\text{Sr}/^{86}\text{Sr}$ ratios in combination
117 with elemental data can help discern differences in bedrock lithology (e.g. Krishnaswami
118 et al., 1992). For example, carbonate-based catchments exhibit a relatively unradiogenic
119 signature in comparison to silicate-based catchments, ($^{87}\text{Sr}/^{86}\text{Sr} = 0.708$ versus 0.721 ,
120 respectively (Allègre et al., 2010)). The ratio may also de-convolve incongruent mineral
121 weathering rates rock (e.g. Bain and Bacon, 1994; Capo et al., 1998; Clow et al., 1997b).

122 Neodymium isotopic ratios ($^{143}\text{Nd}/^{144}\text{Nd}$) have proven themselves as powerful
123 tracers of dust transport, water mass, and sediment source (Aarons et al., 2013; Jiang et

124 al., 2013; Jones et al., 1994; Piepgras and Jacobsen, 1988). Natural variation in the
125 $^{143}\text{Nd}/^{144}\text{Nd}$ composition of rocks relates to the extent of mixing experienced by the
126 material derived from the mantle from differently aged sources, which is enabled by the
127 limited mobility of rare earth elements (REE), such as Sm and Nd, during sedimentary
128 processes (Öhlander et al., 2014; Taylor and McLennan, 1995). Studies utilizing Nd as a
129 tracer for intense chemical weathering are few, but pedogenic studies have shown intense
130 weathering preferentially removes ^{143}Nd from soil profiles (Ma et al., 2010). However,
131 Garçon et al. (2014) modeled the Nd composition of different minerals in sediment and
132 found minerals highly enriched in Nd, notably monazite and aluminates to dominate the Nd
133 isotopic budgets, despite the mineral proportion consistently falling below 0.5% of the
134 total rock weight. The study supports the general classification of Nd as a proxy for
135 congruent weathering. Whilst still in development, the strength of the Nd proxy as a
136 tracer of intensive chemical weathering may be supported by correlation with the more
137 tested radiogenic isotope proxies of radiogenic strontium ($^{87}\text{Sr}/^{86}\text{Sr}$). Therefore, the
138 relationship between radiogenic Nd and Sr systems has the potential to reveal dominant
139 weathering mechanisms in subglacial environments.

140 By correlating the radiogenic Nd and Sr isotopic ratios with local geology, daily
141 in-situ measurements and elemental data, we quantify the ability of this combined Sr-Nd
142 proxy to track channel evolution and model weathering processes in subglacial
143 environments. This study investigates for the first time the effectiveness of the
144 application of coupled Sr-Nd ratios in suspended sediments from two distinct subglacial
145 environments. Here we characterize the seasonal evolution of the subglacial meltwater

146 channels from the Lemon Creek glacier (LCG), Juneau Icefield, Alaska and the
147 Athabasca Glacier (AG), Columbia Icefield, Alberta.

148

149 **2. Site Descriptions**

150 *2.1 Lemon Creek Glacier*

151 The LCG is a warm-based valley glacier located in the Coast Mountain Range of
152 southeast Alaska (11.6 km², 58° 24.418' N, 134° 22.379' W). The LCG forms the
153 southernmost extension of the Juneau Icefield, which itself extends ~3,900 km². Annual
154 mass balance surveys indicate that the glacier is retreating by an average of 0.48 m yr⁻¹ of
155 water equivalents (Miller and Pelto, 1999). Geologically, the LCG is located on a mid-
156 Cretaceous metamorphic-pluton complex, with associated migmatites (Figure 1) (Brew
157 and Ford, 1985; Kistler et al., 1993). Young tonalite sills (62-69 Ma, with a
158 northeastward younging of the bedrock (Gehrels et al., 1984)) characterize the terrain to
159 the immediate west and high-grade metamorphosed sedimentary and volcanic rocks
160 characterize the surrounding area to the east. Late-Permian metamorphosed sedimentary
161 (Greenschist facies) rocks characterize the area further west and beneath the glacial head
162 (Kistler et al., 1993; Stevenson et al., In review-a). The local region maintains a maritime
163 climate with average annual precipitation of 1.4 m yr⁻¹ and mean winter and summer air
164 temperatures of -1°C and 16°C, respectively (Stevenson et al., In review-b).

165 In a corresponding study, Stevenson et al. (In review-a) analyzed the radiogenic
166 Sr isotopic compositions of the suspended sediment at the LCG. The time-series values
167 ranged from ⁸⁷Sr/⁸⁶Sr = 0.708487(6) to 0.710003(8), for Julian Day (JD) 226—252. The
168 study additionally includes measured radiogenic strontium values of four bedrock

169 samples: quartzite ($^{87}\text{Sr}/^{86}\text{Sr} = 0.72960(1)$), gneiss ($^{87}\text{Sr}/^{86}\text{Sr} = 0.70761(4)$), plutonic
170 igneous granodiorite ($^{87}\text{Sr}/^{86}\text{Sr} = 0.70710(4)$), and a metamorphosed crystalline carbonate
171 ($^{87}\text{Sr}/^{86}\text{Sr} = 0.70800(2)$), with values in parentheses representing two standard error. The
172 bedrock values align well with the radiogenic Sr ratios measured in the Juneau Gold Belt
173 (Kistler et al., 1993). However, no measurements have characterized the radiogenic Nd
174 compositions of the bedrock directly beneath the LCG. In a survey of the accretionary
175 terranes of the Alaskan portion of the Coast Mountains, Samson et al. (1991) reported
176 values from juvenile plutons (i.e. the Gravina belt and the Taku terrane) and metamorphic
177 assemblages (i.e. the Tracy Arm, Endicott assemblage, Port Houghton assemblage, and
178 the Ruth assemblage), but the wide range of $^{143}\text{Nd}/^{144}\text{Nd}$ values from 0.511290 ± 12 to
179 0.513030 ± 7 highlights need for direct bedrock measurements to characterize the local
180 geology.

181 *2.2 Athabasca Glacier*

182 The Athabasca Glacier is one of the eight primary glaciers extending from the
183 Columbia Icefield in the Canadian Rockies, Alberta (8.6 km^2 , $52^\circ 12.54' \text{ N}$, $117^\circ 14.29'$
184 W). It is smaller than the Juneau Icefield spanning approximately 325 km^2 . The icefield
185 accumulates the largest volumes of snow and ice south of the Arctic Circle in the
186 Northern Hemisphere and contributes freshwater to the Arctic, Pacific, and Atlantic
187 Oceans (Paterson, 1964). The glacier itself spans over three icefalls and into a valley, is
188 $\sim 1950 \text{ m a.s.l.}$, and extends $\sim 6 \text{ km}$ in length. Uplift and rotation define the regional
189 geology, which is primarily Middle Cambrian limestone and shale (Figure 2) and part of
190 the Pika Formation (Charlesworth and Erdmer, 1989). Beneath the ice, a thin ($0.05\text{-}0.30$
191 m) deformable layer of till exists which promotes till erosion, rotation, and detachment

192 (Hart, 2006). The Athabasca Glacier in particular has been the focus of many previous
193 studies (e.g. Arendt et al., 2015; Kite and Reid, 1977; Paterson and Savage, 1963;
194 Raymond, 1969; Xu et al., 2010). Like the LCG, the glacier has consistently decreased in
195 total mass over the past decades of observation (Kite and Reid, 1977). In the past decade,
196 the region has experienced 395 to 475 mm of annual precipitation and ranges of -15 to -
197 19°C and +16 to +20°C winter and summer temperatures, respectively (Archive, 2013;
198 Arendt et al., 2015; Shea and Marshall, 2007).

199 Similarly to the LCG, the bedrock provides a framework for isotopic analysis.
200 The shales and limestones of the Pika Formation have been well-characterized as highly
201 radiogenic by Boghossian et al. (1996), with ϵ_{Nd} of -27.1. However, the carbonate bedrock
202 obstructs the possibility of measuring such radiogenic values within the strontium system.
203 To a greater degree, the values from previous study (i.e. Millot et al., 2003) relate to
204 weathering and preferential dissolution of carbonate and, in Athabasca River sediments,
205 range from $^{87}\text{Sr}/^{86}\text{Sr} = 0.71285$ to 0.71612 .

206 **3. Sampling and Analysis**

207 *3.1 Sample collection*

208 Samples were collected daily from the main meltwater channel draining the LCG
209 from the 30th June 2012 to the 8th September 2012 (JD 182—252) and the AG from the
210 11th August 2014 to the 25th October 2014 (JD 223—298). The sampling site at the LCG
211 shifted mid-season due to accessibility. From JD 182 to 210, sampling occurred from a 3
212 m snow pit where flowing water was observed. Once the seasonal snowpack decreased
213 and the glacial toe was accessible, the sampling site was moved up-glacier 50 m closer to
214 the glacier toe. Samples were collected at the second location daily from JD 227 to 252.

215 Ten liters of subglacial water were filtered using a Masterflex modular peristaltic
216 pump and a Perfluoroether (PFA) 47mm diameter filtration unit (SavilleX). Hydrophilic
217 Polyvinylidene fluoride (PVDF) Millipore filter membranes (0.22 μm) were used to
218 separate the suspended sediment.

219 Daily in-situ field protocols are outlined in (Stevenson et al., In review-b). In
220 summary: Electrical conductivity, temperature, pH, dissolved oxygen (DO) and alkalinity
221 measurements were taken using a YSI Handheld Multiparameter Instrument (Pro Plus
222 Multiparameter). The electrical conductivity, temperature, and pH measurements were all
223 conducted on-site in the subglacial outlet channels. Approximately 100 mL of filtered
224 subglacial water was used for alkalinity measurements. For anticipated high alkalinities,
225 the 100 mL sample was mixed with 10 mL of Total Alkalinity Reagent (FisherScientific)
226 solution, shaken, and the pH measured. For anticipated low alkalinities, the 100 mL
227 sample was mixed with 1 mL of Total Alkalinity Reagent solution, shaken, and the pH
228 was measured. The pH was converted to the total alkalinity using a pH-total alkalinity
229 conversion (e.g. Fujita, 2008; Hedin et al., 1994)

230 Two diurnal cycles were tracked at the AG. In August (JD 236–237), samples
231 and in-situ measurements were collected every two hours. In October (JD 297–298)
232 samples were collected every three hours due harsher field conditions.

233

234 *3.2 Sample preparation*

235 *3.2.1 Lemon Creek suspended sediments*

236 The samples were prepared for isotopic analysis in a class 10 laminar flow hood
237 inside a 10,000 level clean room at the University of Michigan. Each filter membrane

238 was rinsed with 18.1 MΩ cm to collect the sediment, which was subsequently dried in
239 Teflon beakers. For the LCG samples, 10 mg of sediment were weighed and digested for
240 7 days in 2 mL concentrated nitric acid (HNO₃) with 0.5 mL concentrated hydrofluoric
241 acid (HF). Samples were dried down and further digested in aqua regia for 24 hours to
242 oxidize and remove any residual organic material. Each sample was digested in 1 mL
243 aqua regia for 24 hours, dried on a hotplate, and dissolved in 1 mL of 9 M hydrochloric
244 acid (HCl) in preparation for elemental separation.

245

246 *3.2.2 Athabasca suspended sediments*

247

248 Prior to dissolution procedures (above) the dry AG sediments were first filtered
249 through a 0.63 μm sieve to eliminate potential effects of grain size distribution during
250 fluvial transport. Approximately 10 mg of sediment were weighed and digested using
251 Parr bombs. The sample was loaded into 3 mL Savillex beakers with 2 mL of
252 concentrated HF. The beakers were placed in a 125 mL PTFE container and 6 mL of
253 concentrated HF with trace concentrated HNO₃ was added. The vessel was enclosed in
254 the Parr bomb and placed in a 220°C oven for 48 hours. The solutions were dried down
255 on a hot plate and the procedure was replicated with 6 M HCl in both the Savillex
256 beakers and PTFE container at 180°C for 12-16 hours. After the final dry-down, the
257 samples were digested in 1 mL of 9 M HCl in preparation for elemental separation.

258

259 *3.3 Elemental analysis of Athabasca Glacier samples*

260 Trace and major elemental concentrations were measured in triplicate on the
261 Thermo Scientific ELEMENT2 Inductively Coupled Plasma mass spectrometer at the
262 University of Michigan Keck Laboratory in pulse counting mode. The digested sediment
263 samples were acidified and diluted to 2 mL solutions. An acid blank and standard river
264 reference standards were run every five samples to assess long-term reproducibility and
265 accuracy. Repeat measurements of international standard NIST1640a are provided in
266 Aciego et al., (2015). Baseline detection measurements from the total procedural blank
267 indicate that analytical error was never greater than 10% the concentration even for the
268 smallest concentrations. Such analyses were not possible for the LCG samples due to
269 prior consumption of sample supply in earlier studies (Sheik et al., 2015).

270

271 *3.4 Neodymium isotope analysis*

272 Samples were aliquoted to obtain > 25 ng of Nd. Isolation of Nd was performed
273 through ion exchange column chemistry involving two columns. Each sample was first
274 loaded into a 700 μ L PFA column filled with 50-100 mesh TruSpec resin. Following the
275 procedures of Aciego et al. (2009), HCl was used to elute high field strength elements
276 (HFSE) and REEs. The eluted volumes were then loaded into a preconditioned 2 mL
277 PFA column filled with clean 50-100 μ m LnSpec resin. The subsequent volumes eluted
278 with HCl isolated the Nd from the REE fraction (Aciego et al. 2009). An acid blank and a
279 standard of known concentration (BCR-2) were processed using the same procedure to
280 ensure long-term reproducibility and assess error.

281 Nd was loaded onto outgassed rhenium double filaments after 1 μ L of 1 M HCl—
282 1 M HNO₃ was added to each dried sample. A current of 0.8 A was run through the

283 filament until the sample was dry. The current was slowly increased to 1.8 A and held
284 constant for 1 minute. The current was then flashed at 2.2 A and decreased to 0 A.

285 Isotopic ratios were determined using a Thermo-Finnigan Triton Thermal
286 Ionization Mass Spectrometer (TIMS) at the Glaciochemistry and Isotope Geochemistry
287 Laboratory in the Department of Earth and Environmental Science at the University of
288 Michigan (Aarons et al., 2013; Arendt et al., 2014). Instrumental mass bias was corrected
289 for by applying an exponential mass fractionation law with the $^{146}\text{Nd}/^{144}\text{Nd} = 0.7219$ and
290 mass 149 was monitored for Sm interference. Amplifier gains and baselines were run
291 prior to each set of analysis. The Nd isotopic standard JNdi-1 (10 ng) was measured as
292 $^{143}\text{Nd}/^{144}\text{Nd} = 0.512099 \pm 0.000016$ (2 s.d. n=8) which is in agreement with the accepted
293 JNdi-1 standard value of $^{143}\text{Nd}/^{144}\text{Nd} = 0.512115$ (Tanaka et al., 2000). The $^{143}\text{Nd}/^{144}\text{Nd}$ of
294 BCR-2 was 0.512643 ± 36 (2 s.d. in the last decimal place; n=5) and in agreement with
295 the literature (Li et al., 2007; Raczek et al., 2003; Weis et al., 2006).

296

297 *3.5 Strontium radiogenic isotope analysis*

298 Radiogenic strontium values for suspended sediment were obtained from
299 Stevenson et al., (In review-a). Here we expand their existing data set using the same
300 analytical preparation and methods. In brief, samples were partitioned to provide 2 μg Sr,
301 these aliquots were dried on a hot plate and dissolved in 7.5 M HNO_3 . Samples were
302 loaded in 500 μL 3 M HNO_3 onto Sr columns containing 150 μL Eichrom Strontium
303 specific resin bed in 500 μL 3 M HNO_3 .

304 The column was washed and eluted in several stages with HNO_3 following the
305 procedure outlined by (Aciego et al., 2009b). The procedural Sr blank was less than ~ 60

306 pg, constituting < 0.1% of the total Sr analyzed for a typical Sr analysis. Strontium
307 samples were loaded onto outgassed 99.98% Re filaments in 1.0 μL 7.5 N HNO_3 along
308 with 0.8 μL TaF_5 activator to enhance the ionization efficiency of Sr (Charlier et al.,
309 2006).

310 Strontium isotope measurements were performed on a Thermo-Finnigan Triton
311 Plus Thermal Ionization Mass Spectrometer (TIMS) at the University of Michigan using
312 the method outlined in Stevenson et al., (2015b). Fractionation caused during machine
313 analysis was corrected for using $^{86}\text{Sr}/^{88}\text{Sr} = 0.1194$. External precision on the standard
314 runs (NBS987) for $^{87}\text{Sr}/^{86}\text{Sr}$ was 0.710264 ± 0.000016 (2 s.d. n=50). A basalt rock (BCR-
315 2) was used to monitor precision of column chemistry and TIMS analysis. The $^{87}\text{Sr}/^{86}\text{Sr}$
316 of BCR-2 was within error of literature 0.70504 ± 5 (2 s.d. in the last decimal place, n=3),
317 respectively (e.g. Krabbenhöft et al., 2009; Ma et al., 2013; Moynier et al., 2010).

318

319 **4. Results**

320 *4.1 Physiochemical Properties*

321 The seasonal physiochemical dataset of the LCG and AG can be found in
322 Appendices A and B, respectively. Appendix B additionally contains the diurnal values
323 from August and October cycles at the AG. Greater discharge, alkalinity, and
324 conductivities characterize measurements from the AG in comparison to the LCG. In
325 general, the pH is higher at the AG (average pH = 8.89; 2 s.d. of 0.37) than the LCG
326 (average pH = 7.40; 2 s.d. of 0.40). It is important to highlight that field seasons at the
327 AG and LCG did not occur at the same time of year, with sample for the AG from JD
328 223—298 and for the LCG from JD 182—252. At the LCG the pH begins to increase

329 starting JD 238 and reached a maximum measured pH on JD 248 whereas sampling at the
 330 AG did not begin until JD 223. Daily precipitation trends, provided by the Agroclimatic
 331 Information Service (ACIS, 2015) and the NOAA Juneau Airport weather station
 332 (NOAA, 2014), are additionally reported in Figure 3 and Figure 4, respectively.

333

334 *4.2 Radiogenic isotope measurements*

335 Appendices C and D present the seasonal radiogenic Sr and Nd measurements at
 336 the LCG and AG, respectively. Errors are reported as two standard errors of the mean and
 337 are given in parentheses with variation in the last decimal place. Correlations between the
 338 $^{87}\text{Sr}/^{86}\text{Sr}$ and ϵ_{Nd} are presented in Figures 5 and 6 for the LCG and AG, respectively, and
 339 include both time-series and diurnal values. The plots contrast the wide ϵ_{Nd} and narrow
 340 $^{87}\text{Sr}/^{86}\text{Sr}_{\text{range}}$ at the LCG with the narrow ϵ_{Nd} range and wide $^{87}\text{Sr}/^{86}\text{Sr}_{\text{range}}$ at the AG.

341

342 *4.2.1 Radiogenic Neodymium*

343 The time-series Nd isotopic values are recorded in Appendices C and D and
 344 depicted in Figure 3 and 4 for the LCG and AG, respectively. The Nd isotopic
 345 compositions are represented in epsilon notation as ϵ_{Nd} (Eq. 1), which is defined as:

$$\epsilon_{\text{Nd}(t)} = \left[\frac{\left(\frac{^{143}\text{Nd}}{^{144}\text{Nd}} \right)_{\text{sample}(t)}}{\left(\frac{^{143}\text{Nd}}{^{144}\text{Nd}} \right)_{\text{CHUR}(t)}} - 1 \right] \times 10000$$

346

(Eq. 1)

347 where $(^{143}\text{Nd}/^{144}\text{Nd})_{\text{CHUR}}$ is the Nd isotopic composition of the Chondritic Uniform
 348 Reservoir (CHUR) which is $^{143}\text{Nd}/^{144}\text{Nd} = 0.512638$ (Jacobsen and Wasserburg, 1980).

349 The internal error of 2σ of the mean is parenthesized for each value and external error is
350 $\epsilon_{Nd} = 0.49$.

351 At the LCG, we observe an average radiogenic Nd value of $\epsilon_{Nd} = -6.2 \pm 1.0$ (2
352 s.d.). The greatest degree of spread is observed between JD 216 to 234. Values range
353 from ϵ_{Nd} of -4.6 (0.9) to -8.7 (0.2). The remaining subset from JD 230—251 exhibits an
354 average $\epsilon_{Nd} = -6.2 \pm 1.1$ (2 s.d.) and mirror early season values.

355 The time-series AG values are presented in Figure 4. The dataset exhibits little
356 seasonal variability with an average ϵ_{Nd} of -27.3 ± 0.6 (2 s.d.), which is only slightly
357 larger than the external reproducibility of $\epsilon_{Nd} = 0.49$. Similarly, the diurnal values
358 reported in Appendix D align with the time-series record. We report averages of $\epsilon_{Nd} =$
359 27.5 ± 0.7 and -27.3 ± 0.5 for the August and October cycles, respectively and with two
360 standard deviations.

361

362 *4.2.2 Radiogenic Strontium*

363 In a corresponding study, Stevenson et al. (In review-a) analyzed the radiogenic
364 Sr isotopic compositions of the suspended sediment from JD 226—252 at the LCG. The
365 LCG dataset exhibits a slightly parabolic seasonal record. The measurements trend
366 towards less radiogenic values from JD 182—213 ($R^2 = 0.93$) with a measured $^{87}\text{Sr}/^{86}\text{Sr}_{\text{range}}$
367 $= 0.709673(8)$ to $0.708562(8)$. Mid-season values from JD 214—235 exhibit poor
368 correlation ($R^2 = 0.06$), coinciding the mid-season spread of measured ϵ_{Nd} values. During
369 the time-period, we report an average value of 0.709054 ± 0.00352 (2 s.d.). At the end of
370 the season (JD 236—251), increases towards more radiogenic values mark seasonal
371 progression ($^{87}\text{Sr}/^{86}\text{Sr} = 0.709036(6)$ to $0.70975(1)$; $R^2 = 0.20$).

372 The time-series radiogenic Sr values from the AG (Figure 4) exhibit higher
373 isotopic ratios, which generally decrease over the course of the melt season ($^{87}\text{Sr}/^{86}\text{Sr}_{\text{range}}$
374 of 0.712424(7) to 0.71606(2); $R^2 = 0.34$); however, the correlation is stronger at the
375 beginning of the melt season (JD 233-267; $R^2 = 0.45$). A greater range of seasonal
376 variability is observed at the AG than the LCG ($^{87}\text{Sr}/^{86}\text{Sr}_{\text{range}} = 0.0036$ versus 0.0015,
377 respectively).

378 Total variation in $^{87}\text{Sr}/^{86}\text{Sr}_{\text{range}}$ over a diurnal cycle are much larger for August,
379 0.0014, compared to October 0.0004. While no discernable trend exists in the October
380 record, there is a shift towards less radiogenic values from 10:00 to 20:00, which is
381 followed by an increase in the isotopic ratio from 20:00 to 4:00. However, consistently
382 high discharge measurements occurred during the August diurnal sampling with all
383 values greater than $0.75 \text{ m}^3 \text{ s}^{-1}$. The observed trends from the diurnal and time-series
384 samples correlate well to in-situ discharge measurements (Figure 7). The $^{87}\text{Sr}/^{86}\text{Sr}$ ratio
385 increases with discharge ($R^2 = 0.52$) until a critical value of $0.9 \text{ m}^3 \text{ s}^{-1}$ was reached. High
386 variability in the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio is apparent at high discharge levels during both diurnal
387 and time-series measurements.

388

389 *4.3 Elemental Concentrations*

390 Elemental concentrations from the AG are reported in Appendix B. In general, Al,
391 Ca, K, Mg, Na, Rb, seasonal trends are highly variable, with peaks on JD 243, 244, and
392 290. Rain events did occur on each of these days; however, there are no consistent
393 seasonal trends between the daily elemental variation and the rain events or total
394 discharge. While individual concentrations indicate mineralogical variability of the

395 sediment, molar ratios track seasonal weathering trends. Here, we refer to the Rb/Sr,
396 Sm/Nd, and Ca/K values listed in Appendix B. A strong, positive correlation between the
397 $^{87}\text{Sr}/^{86}\text{Sr}$ and Rb/Sr ratios for the time-series dataset ($R^2 = 0.79$) (Figure 8) encapsulates
398 the traditional Rb-Sr isochron. Inclusion of the diurnal ratios weakens the trend ($R^2 =$
399 0.57), particularly as the October cycle exhibits relatively high Rb/Sr ratios and low
400 $^{87}\text{Sr}/^{86}\text{Sr}$ values. A moderate, inverse relationship exists between the $^{87}\text{Sr}/^{86}\text{Sr}$ and Ca/K
401 ratios (Figure 9, $R^2 = 0.47$ for the entire dataset). However, there is no correlation
402 between ϵ_{Nd} and the seasonal Ca/K, further emphasizing the lack of variability in the ϵ_{Nd}
403 record.

404 Due to the limited variability in the ϵ_{Nd} ratios, we examined variation in Nd
405 concentrations through the Sm/Nd ratio as a comparative plot to Figure 8. There is weak,
406 positive trend ($R^2 = 0.17$) between the Sm/Nd and total discharge over the entire dataset
407 (Figure 11, in the Supplementary Online Material). The August diurnal values exhibit a
408 distinctly wide range of Sm/Nd which correlate with high discharge volumes consistently
409 greater than $2.7 \text{ m}^3 \text{ s}^{-1}$ and captures peak melt conditions that occur at approximately
410 16:00. Still, exclusion of the diurnal cycles leads to a weak correlation ($R^2 = 0.09$).

411

412 **5. Discussion**

413 *5.1 Characterizing the subglacial environment utilizing combined Sr-Nd analysis*

414 Combining radiogenic Nd and Sr isotopic systems has greatly enhanced the
415 current understanding of how sediment is sourced (Goldstein and Jacobsen, 1988;
416 McCulloch et al., 2003; Yang et al., 2007), transported (Viers et al., 2008; Weldeab et al.,
417 2002), and weathered (Lupker et al., 2013; Négrel, 2006) in rivers and soils. Our case

418 studies present two contrasting scenarios regarding the processing and mixing of
419 sediment in the subglacial environment. Differences between the systems help us to
420 classify sediment transport mechanisms into two distinct categories: poorly-mixed and
421 well-mixed subglacial suspended sediment.

422

423 *5.1.1 Poorly mixed suspended sediment*

424 The LCG exhibits a wide range of ϵ_{Nd} values, ($\epsilon_{Nd (Range)} = 4.1$). In rare instances,
425 intense weathering of basalts has been attributed to ϵ_{Nd} drifts of ~ 2.5 (Ma et al., 2010),
426 which is still notably lower than the LCG variability. The few studies reporting annual
427 variability of the dissolved and suspended sediment loads in global rivers has been within
428 the range of $\epsilon_{Nd (Range)} = 1-2.5$ (e.g. Andersson et al., 2001; Viers et al., 2008). However, a
429 large variability has been reported in suspended sediments within the major channel and
430 tributaries along the Yangtze River ($\epsilon_{Nd (Range)} = 3.9$), which has been attributed to lack of
431 dominant source rocks and complicated underlain geology ranging from Archean to
432 Quaternary material (Yang et al., 2007).

433 While the geology beneath the LCG is primarily mid-Cretaceous, late-Permian
434 facies extend to the west and underlay part of the glacial head. Geologic units underlying
435 the LCG also include: biotite schist, biotite gneiss, marble and calc-silicate granofels,
436 hornblende gneiss, and granitoid rocks (Brew and Ford, 1985). In comparison, the AG is
437 predominately underlain by Middle Cambrian shales and limestones. These differences
438 may create a hydrologic environment where there exists a greater variation in sediment
439 age and composition at the LCG.

440 While the individual glaciers (LCG and AG) are relatively similar in length (Hart,
441 2006; Miller and Pelto, 1999), the differences in range of the isotopic values may be a
442 consequence of the rate of discharge levels or environmental forcings within the
443 subglacial system. At lower discharge rates, there may be a greater potential for
444 heterogeneities (e.g. previously isolated cavities containing sediment pockets) to impact
445 the bulk measured sediment ratio at the terminus of the glacier (sample site). This is
446 because the sample is only a snapshot of the inferred daily sediment load. As discharge
447 increases, greater flow levels could provide enough peak shear stress for more consistent
448 and well-mixed excavations. Still, the alternative explanative of an environmental forcing
449 to the system is more compelling.

450 At the LCG, the greatest spread in ϵ_{Nd} values occurs within JD 216–234, during
451 which the drainage network has likely reached its full extent. The subsequent range
452 narrows and aligns with early season values; the excluded subset exhibits an average ϵ_{Nd}
453 value of -6.2 ± 1.0 (2 s.d.). This period of high variability also coincides with the
454 drainage of periglacial Lake Linda, which is located at the head of the glacier. Each
455 summer, rapid drainage events cause the water to flow throughout the glacier. Complete
456 drainage has been observed to occur over a 48-hour period and the event may occur on
457 separate times during the summer. In 2012, a flyover on JD 213 affirmed the lake had not
458 yet drained. However, a sampling excursion on JD 230 revealed that drainage had
459 occurred. During the drainage, the abrupt flux of water may have overwhelmed the
460 system and mobilized the surface texture, which may have freed interlocked or underlain
461 grains. Such disturbances could also maximize the introduction of unexposed sediment
462 pockets. As the season progressed past JD 234, the transport network may then have re-

463 equilibrated. However, we see little evidence of a change in measured discharge during
464 this period that would have been expected with the abrupt Lake Linda drainage. This may
465 further emphasize the uncertainty that can be associated with subglacial dynamics.

466 Precipitation may play a qualifying role as well. There was a higher frequency of
467 precipitation across the sample collection period at the LCG than the AG (Figure 3 vs.
468 Figure 4). As water transported from the surface and through englacial conduits plays an
469 important role in the total water flux, consistent perturbations could cause the subglacial
470 regime to shift and evolve on a small scale. While no consistent trend between
471 precipitation and isotopic ratios exists, there may potentially be lead-lag relations. Hence,
472 when using fluvial sediment as a source provenance, it may be important to assess
473 potential disruptions to the hydrologic network as they may amplify heterogeneity and
474 obscure chemical weathering signals. Suspended sediment itself can be more
475 heterogeneous in mineralogy than dissolved sediment in nature. The composition of the
476 dissolved load relates more to which minerals are preferentially dissolved through
477 chemical alteration whereas the composition of the suspended load relates to the eroded
478 residue, secondary mineral phases, and the introduction of additional authigenic minerals
479 (Tricca et al., 1999). Further comparison between the subglacial environments from
480 different climatic regimes could elucidate our understanding of the impact of climatic
481 events on the partitioning of chemical signatures between the dissolved and suspended
482 sediment loads.

483

484 *5.1.2 Well-mixed suspended sediment*

485 In contrast to the LCG, limited variability is observed in the Sr-Nd record at the
486 AG. The consistency does not oversimplify the depositional processes within the
487 subglacial environment. Rather, it suggests a continuous, dynamic manner in which the
488 sediment is mixed. The result is a similarly aged product, as reflected in the ϵ_{Nd} values.
489 The overall greater flow-through discharge measurements may reflect greater
490 entrainment and transport of available sediment. Comparison between the LCG and AG
491 bedrock suggests shifts in the ϵ_{Nd} record would be observed at both sites. The rotation of
492 the bedrock during the genesis of the Canadian Rockies leads to a general increase in
493 bedrock age in the northeast direction, or down-glacier (Figure 2). One explanation is
494 also that the age-gradient could be insignificant on the scale of the AG. An undetermined
495 variable however is that samples from the AG were collected later in the melt season than
496 the LCG (JD 223-298 versus JD 189-252, respectively). Despite potential complications
497 of drainage evolution and progressive closure of basal network, the values remain
498 constant through the end of the season. This measurement consistency proves interesting
499 as it enables us to unravel incongruent weathering trends in the radiogenic Sr record.

500

501 *5.2 Strontium weathering signal and depositional implications*

502 The AG isotopic record requires additional attention due to its insignificant
503 variation in the ϵ_{Nd} ratios over the course of the melt season and large scale of variability
504 in the $^{87}Sr/^{86}Sr$ record (approximately twice the observed radiogenic Sr variability at the
505 LCG). The dual combination promotes observation of chemical weathering trends as
506 variation due to bedrock source changes has been eliminated. Négrel (2006) has related
507 similar trends to incongruent weathering mechanisms in soils and waters. Hence, it is

508 important to analyze the Sr systematics with their corresponding elemental data to
509 determine if the comparisons between traditional, well-studied weathering environments
510 and the subglacial environment can be systematic.

511 The $^{87}\text{Sr}/^{86}\text{Sr}$ against the Rb/Sr ratio is presented in Figure 8. While the measured
512 trends follow traditional isochron lines (Goldstein and Jacobsen, 1988), the AG plot does
513 not represent variation in geologic age and instead likely represents a mineralogical
514 control on the radiogenic Sr composition of the sediments (Colin et al., 2006; Singh and
515 France-Lanord, 2002). Small deviations create a slightly sub-linear array. Brass (1975)
516 attributed partial causation to the retention of Rb during weathering of sediments and
517 preferential leaching of Sr. Slight complications may also occur due to the presence of
518 un-metamorphosed sedimentary rocks (Eisenhauer et al., 1999), and studies generally
519 present greater scatter in suspended sediment than dissolved sediment trends (Goldstein
520 and Jacobsen, 1988).

521 To further support the weathering model, Figure 9 reveals an inverse relationship
522 between the $^{87}\text{Sr}/^{86}\text{Sr}$ ratios and the Ca/K ratio. Carbonate minerals within the bedrock are
523 preferentially dissolved during weathering in the subglacial environment (e.g. Fairchild et
524 al., 1994; Tranter et al., 1993) and the plot reinforces the release of low $^{87}\text{Sr}/^{86}\text{Sr}$ bearing
525 minerals during the initial stages of chemical weathering followed by the release of more
526 radiogenic $^{87}\text{Sr}/^{86}\text{Sr}$ K-bearing minerals during slower, silicate weathering. The
527 relationship combines with this pseudo-isochron to provide the conceptual framework for
528 detecting chemical weathering trends through correlation of radiogenic Sr record and
529 hydrologic parameters.

530 Comminution through physical weathering processes leads to the exposure of
531 fresh, reactive surfaces. While these reactive surfaces may be primed for chemical
532 alteration, glacierized environments have been generally reported to have relatively
533 higher physical and lower chemical weathering rates in comparison to non-glacierized
534 environment due to high levels of glacial erosion and abrasion (Raiswell et al., 2006).
535 The ratio of the two weathering rates may contribute unique characteristics to sediments
536 produced in the subglacial environment. In particular, the time elapsed since initial
537 comminution, and hence sediment formation, has taken place may help explain trends
538 between the $^{87}\text{Sr}/^{86}\text{Sr}$ ratios and hydro-physical properties such as bulk discharge (Figure
539 7). While the geologic age of the sediment presents as invariable, we suggest the concept
540 of a post-comminution age to describe the correlation and depositional processes.

541 Subglacial deposition may simply be a function of the net sediment flux into a
542 given subglacial area (Hart, 1995) whereas entrainment occurs when a critical shear
543 stress at the bed is exceeded (Walder and Fowler, 1994). The texture of the subglacial
544 environment is largely unconstrained; however, the AG does maintain a layer of
545 deformable subglacial till (Hart, 2006). Subglacial diamict is highly variable in grain size
546 (e.g. Drewry, 1986) and has been described to maintain a matrix rich in clay and silt-
547 sized grains, presumably derived from comminution processes (Walder and Fowler,
548 1994). Excavation of sediment is thought to differ between such soft beds and contrasting
549 rigid beds (longitudinal movement of sediment at the base of deforming layer *vs.* erosion
550 by plucking and abrasion, respectively) (Hart, 1995), and thus we would expect
551 excavation at the AG to be dependent on the depositional layering of sediment.

552 There is an increase in $^{87}\text{Sr}/^{86}\text{Sr}$ ratios until a seemingly critical discharge value of
553 $\sim 0.9 \text{ m}^3 \text{ s}^{-1}$ is reached at the AG. Critical discharges are likely unique to a given glacier
554 and may depend on the location, bed topography, and climatic factors. The relationship
555 suggests that, as flow rates increase, sediment with a younger post-comminution age will
556 become entrained. As the mantle is deepened, as shown in Figure 10, ‘older’ sediment is
557 exposed and primed for transport. However, once the critical discharge is reached,
558 excavation of sediment is likely non-discriminatory, which may be due shear stresses
559 exceeding conditions for equal mobility or the additional inclusion of previously isolated
560 sediment. All available sediment has the same potential to flush out of the system. The
561 consistency of this trend has implications for the production of sediment throughout the
562 melt season. Often, the greatest sediment loads are excavated at the beginning of the melt
563 season due to the build-up of sediment during the low flow and likely less channelized
564 drainage network during the winter months (e.g. Collins, 1990). Our field campaign
565 occurred during the final stage of the melt season (August—October), which suggests the
566 depositional environment continues to evolve on a daily scale throughout the entire melt
567 season.

568 The results pose the unanswered question of the impact of a soft, deformable
569 layer. If quarrying effects are only observed concurrently with similar soft-sediment
570 layers, they may be requisite. Unfortunately, comparative analysis from the LCG is
571 inconclusive due to the high variation of ϵ_{Nd} record.

572

573 *5.3 Diurnal Variability at the Athabasca Glacier*

574 The diurnal AG isotopic and elemental trends are relatively consistent with the
575 time-series values. The sampling in August indicates high discharge levels ($> 0.7 \text{ m}^3 \text{ s}^{-1}$)
576 throughout the day with peak discharge levels occurring in the afternoon. The resulting
577 high $^{87}\text{Sr}/^{86}\text{Sr}_{\text{range}}$ further supports the importance of flow conditions to the understanding
578 of sediment transport, entrainment, and erosion. While it is expected that heterogeneities
579 are inherent to chemical characterization of sediment, our study suggests the presence of
580 systemic depositional processes (e.g. mantling) exist on short temporal scales. The
581 October sampling aligns with a majority of the time-series trends (Figure 6, 7, and 9).
582 However, the $^{87}\text{Sr}/^{86}\text{Sr}$ vs. Rb/Sr molar ratio exhibits a distinctly low $^{87}\text{Sr}/^{86}\text{Sr}$ for given
583 Rb/Sr values (Figure 8). As the October sampling likely occurred during the progressive
584 closure of the drainage network, an increase of preferential leaching of mobile Sr may
585 help explain these observations.

586

587 **6. Conclusions**

588 High physical weathering rates within the subglacial drainage network could
589 largely impact how the suspended sediment is processed, deposited, and entrained. In an
590 attempt to reveal subglacial dynamics, our study presents the first application of Sr-Nd
591 systematics to the suspended sediment of the subglacial meltwater. The systematics
592 traditionally track variation of source and weathering in fluvial environments. The Sr-Nd
593 correlation reported in our study provides important implications for how well the
594 sediment is mixed as it is processed through the drainage network at two geologically
595 distinct glaciers.

596 Time-series values at the LCG that show high variability in the ϵ_{Nd} values suggest
597 a poorly mixed sediment flux. The variation may relate to a glacial flood event, which
598 suggests the sediment transport is sensitive to local climate disturbances. Such events
599 may lead to a period of disequilibrium within the depositional environment and the
600 implications suggest degree of caution should be taken when assessing weathering trends.
601 Conversely, the insights gained from records of low seasonal variability in the ϵ_{Nd} record
602 strengthens our ability to understand chemical and physical processes affecting the
603 radiogenic Sr record. At the AG, correlation between the $^{87}Sr/^{86}Sr$ ratios and total
604 discharge measurements suggest the occurrence of bedrock mantling. We track the
605 process through the post-comminution age of the suspended sediment for which we
606 capitalize on traditional radiogenic Sr systematics to construct a pseudo-isochron of the
607 suspended sediment. Such depositional characteristic may be unique to glaciers with soft,
608 deformable beds. Application to additional subglacial environments with diverse
609 geology, subglacial texture, and locations would provide useful comparisons for future
610 analysis. The introduction of more isotopic tracers such as lead or hafnium could
611 constrain our interpretation further. As we continue to geochemically characterize the
612 processes affecting the erosion, deposition, and transport of the subglacial sediment, a
613 greater quantification of how the subglacial sediment flux impacts the underlying
614 depositional environment and downstream ecosystems may be facilitated.

615

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628 **References**

629

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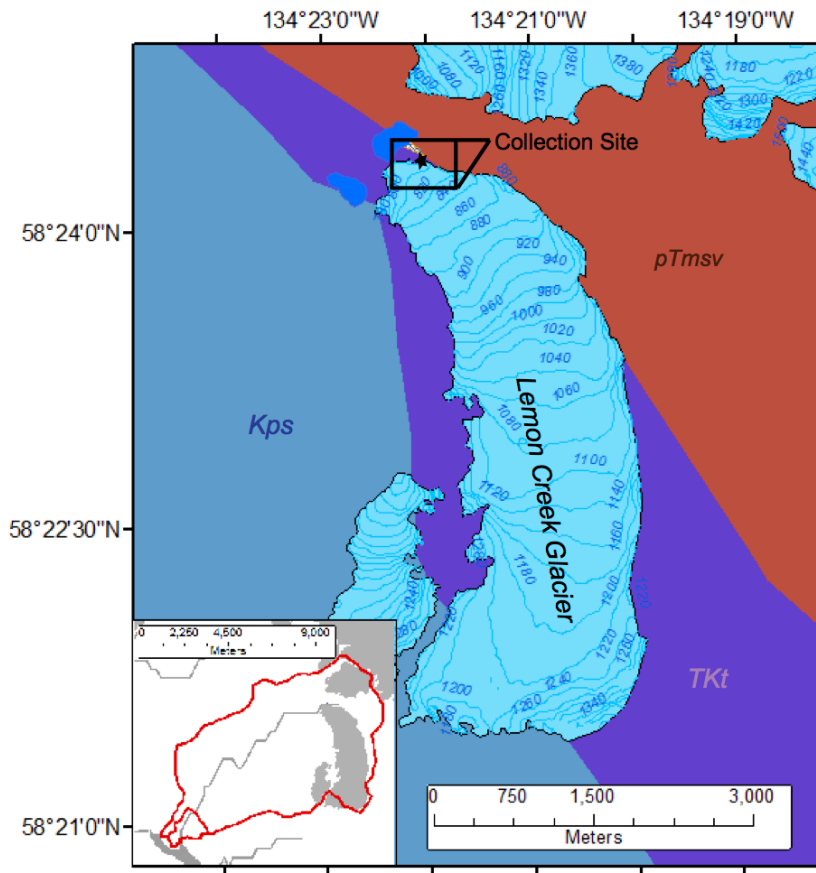
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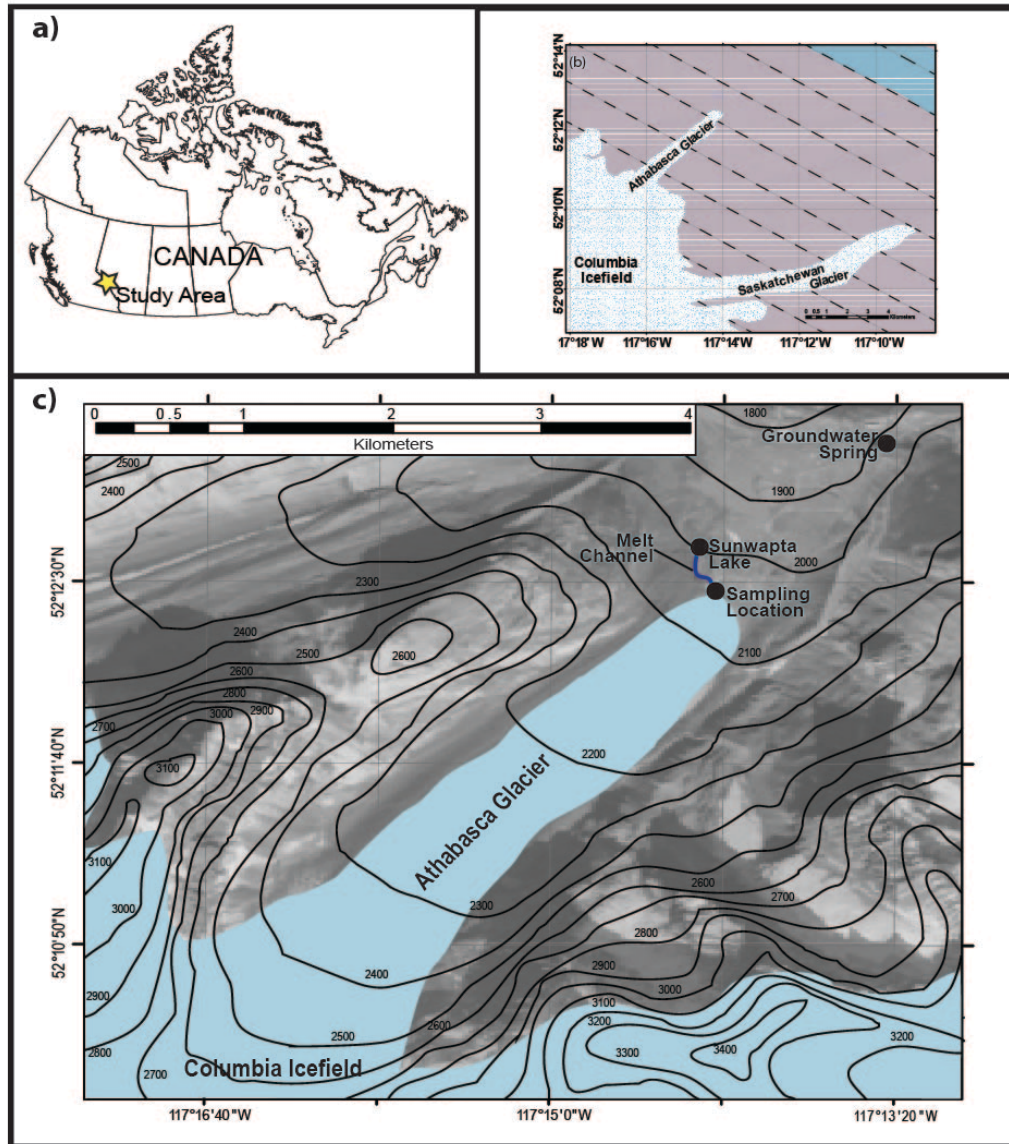
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947 Figures:



948
949 Figure 1: Modified from Stevenson et al. (In review) and provides geologic context of the
950 Lemon Creek Glacier (LCG). Kps = Taku Terrane, composed of Greenschist facies and
951 metamorphosed sedimentary rocks, Late Permian. TKt = Taku terrane, tonalite sills (62-
952 69 Ma). pTmsv = Yukon-Tanana Terrane, high grade metamorphosed sedimentary and
953 volcanic rocks, Carboniferous (Gehrels et al., 1984; Kistler et al., 1993; Samson et al.,
954 1990). The inset depicts the Lemon Creek watershed, which extends to the Gastineau
955 Channel and to the Gulf of Alaska. The watershed was delineated using USGS
956 HydroSHED digital elevation maps (Lehner et al., 2008). The topographic lines represent
957 10 m of elevation on the LCG.
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959
 960 Figure 2: Modified from Arendt et al. (2015). Frames A and B show the geographic
 961 context and lithography, respectively, of the Athabasca Glacier (AG). The AG is
 962 primarily underlain by Middle Cambrian limestone and shale with the oldest rocks in the
 963 southwest, or increasing distance from the toe of the AG, and are indicated by the purple
 964 section. The oldest rocks are indicated in blue and the dashed lines represent rotation of
 965 the AG bedrock during formation. Frame C provides an elevation profile and detailed
 966 plot of the sampling location.
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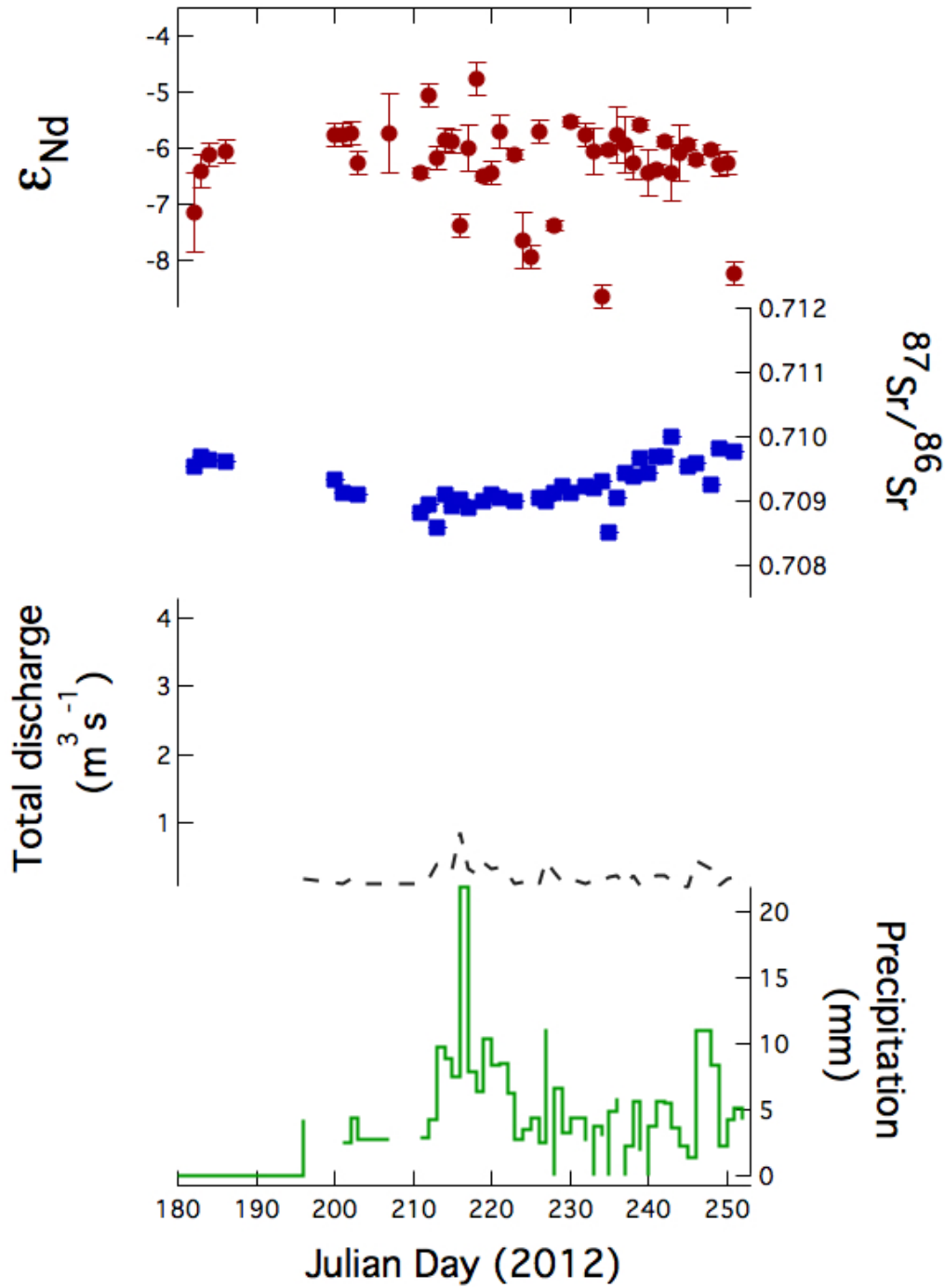
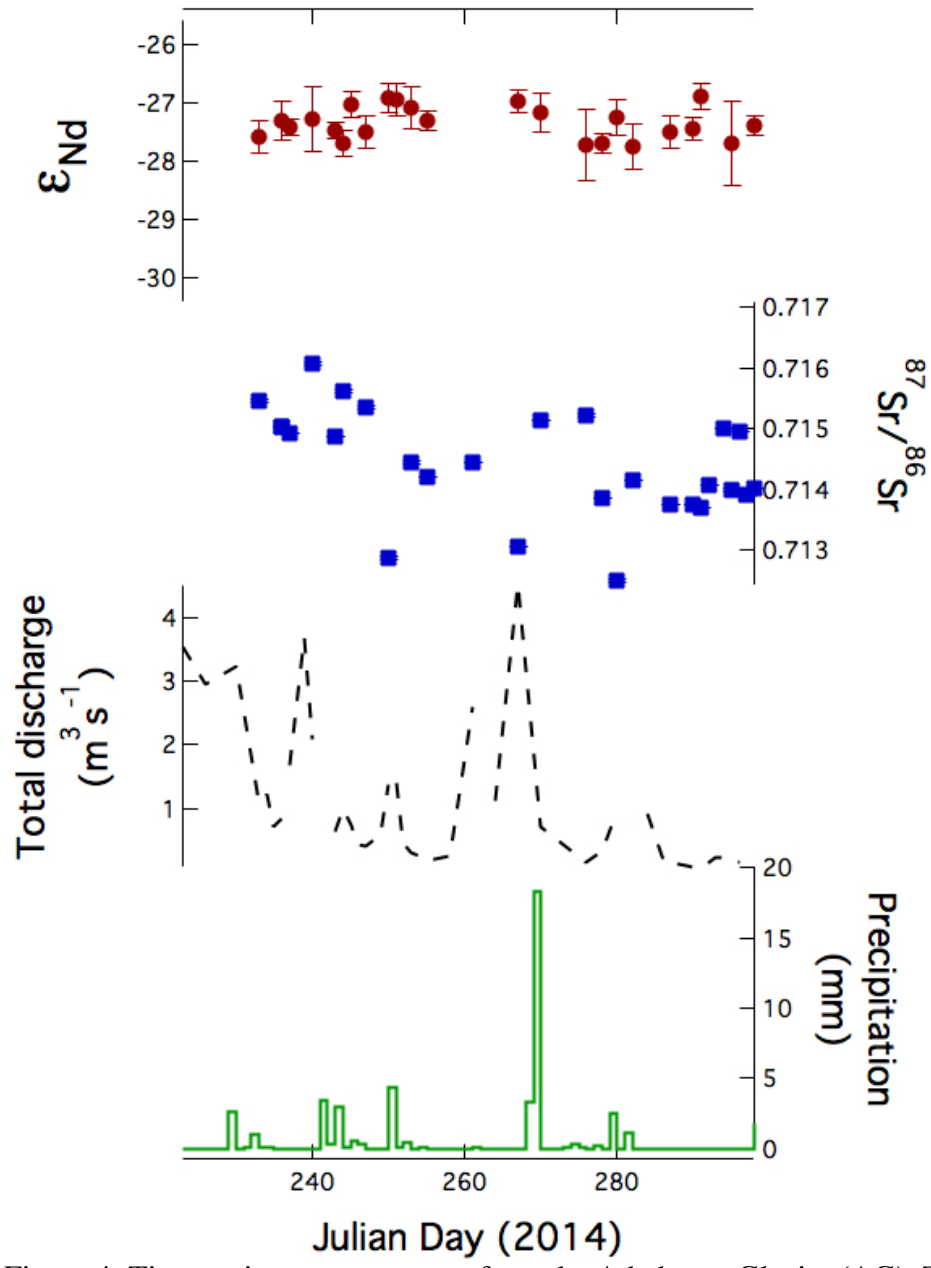
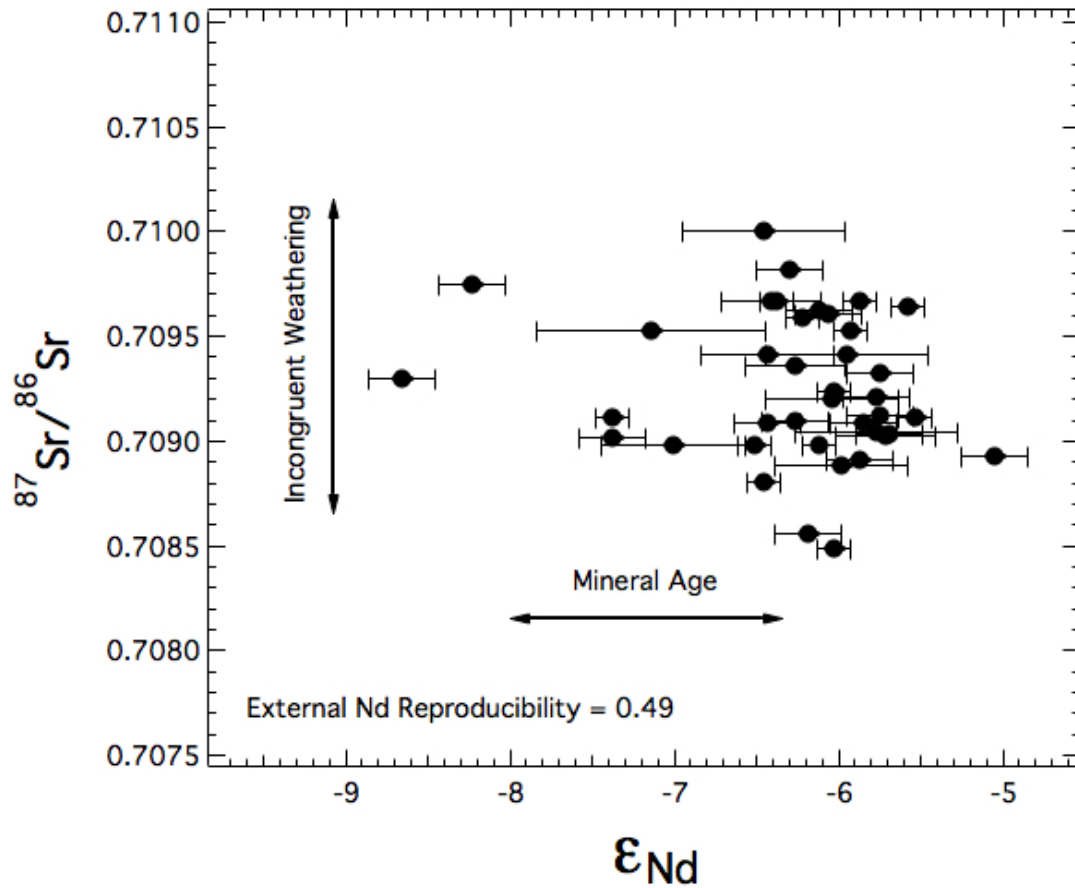


Figure 3:

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 969 Time-series measurements from the Lemon Creek Glacier (LCG). The upper panels
 970 represent radiogenic neodymium (written in ϵ_{Nd} ; red circles) and strontium (blue squares)
 971 values of the suspended sediment. Two standard errors are included. The lower panels
 972 depict seasonal trends of total discharge (black dashed line) and precipitation (solid green
 973 line).
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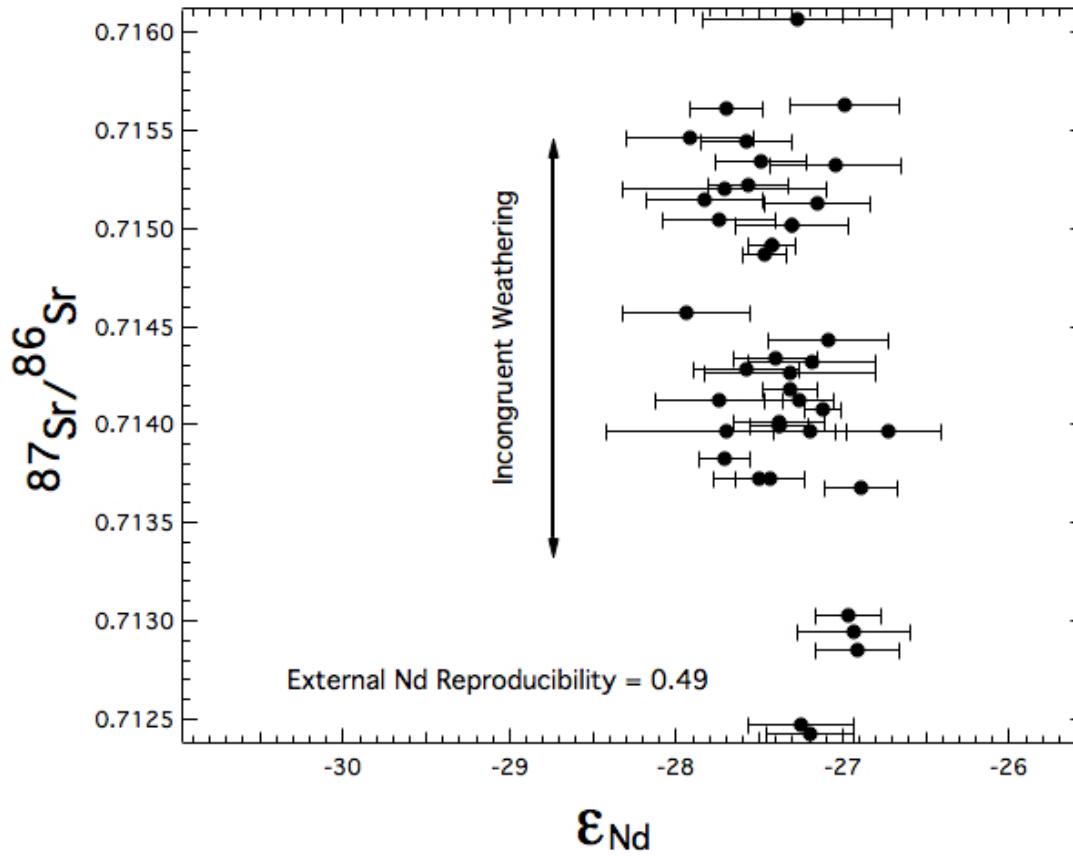


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 976 Figure 4: Time-series measurements from the Athabasca Glacier (AG). The upper panels
 977 represent radiogenic neodymium (written in ϵ_{Nd} ; red circles) and strontium (blue squares)
 978 values of the suspended sediment. Two standard errors are included. The lower panels
 979 depict seasonal trends of total discharge (black dashed line) and the daily precipitation
 980 (solid green line).
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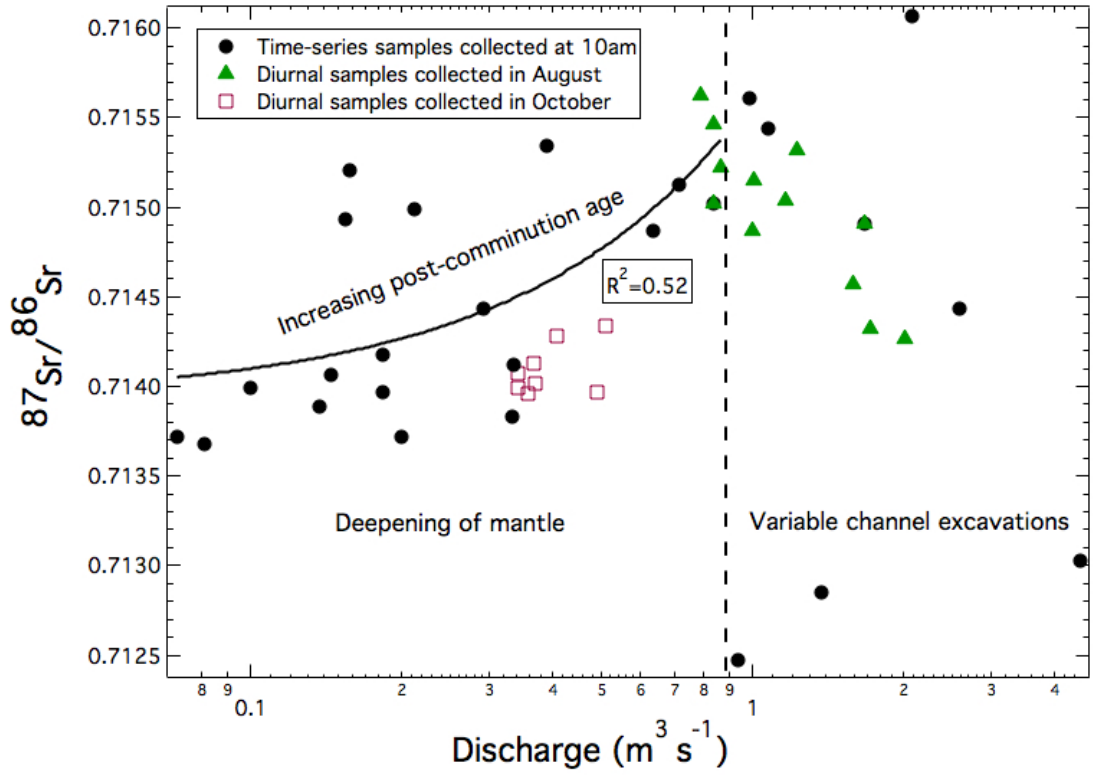
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Figure 5: Correlation between $^{87}\text{Sr}/^{86}\text{Sr}$ and ϵ_{Nd} in the suspended sediment at the Lemon Creek Glacier. The error bars represent two standard error. The large range of ϵ_{Nd} (ϵ_{Nd} (Range) ~ 4 units) at the LCG relates to a wide range of sediment mineral ages and suggests that the sediment is poorly mixed.



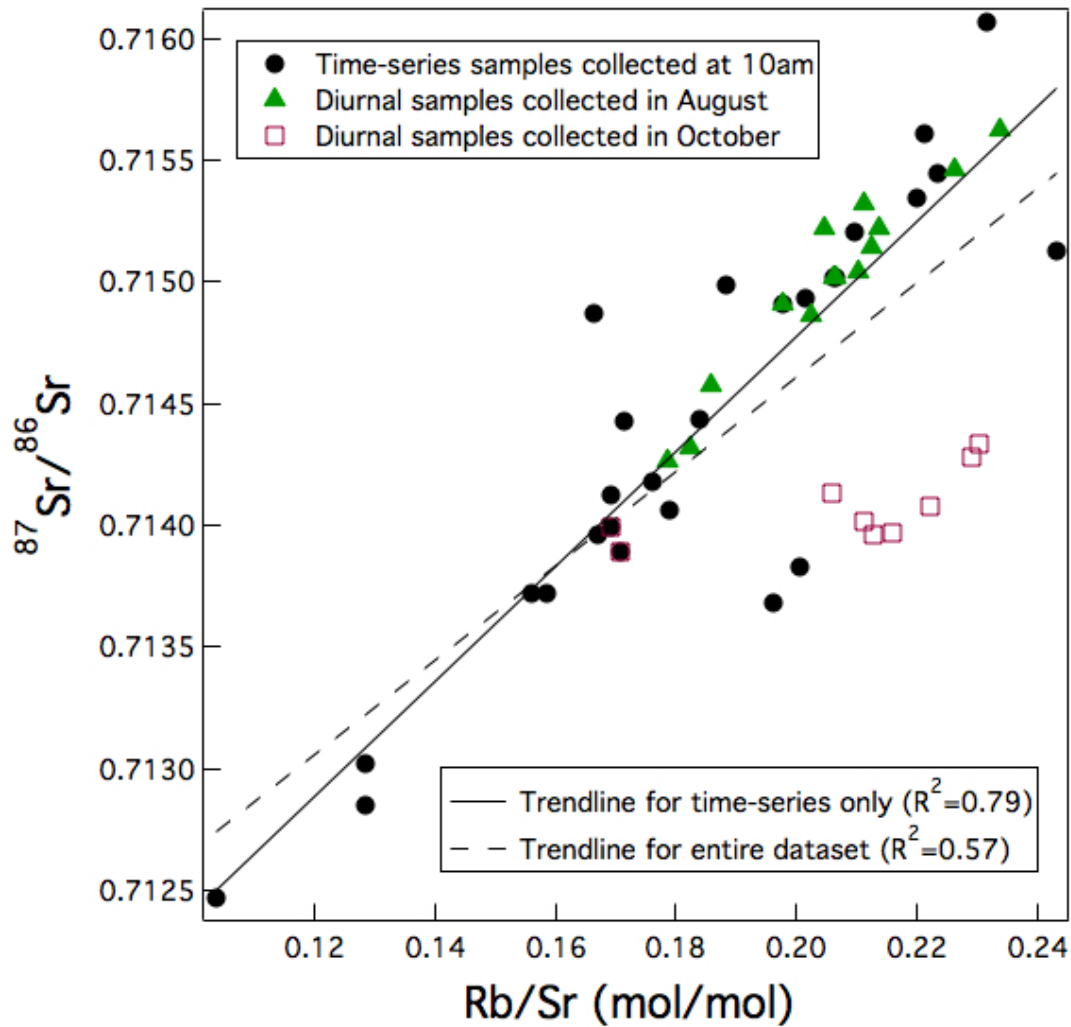
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Figure 6: Correlation between $^{87}\text{Sr}/^{86}\text{Sr}$ and ϵ_{Nd} in the suspended sediment at the AG. Low variability is observed in the ϵ_{Nd} , with the included external reproducibility of $\epsilon_{\text{Nd}} = 0.49$. The narrow range of ϵ_{Nd} ($\epsilon_{\text{Nd (Range)}} \sim 2$ units) at the AG suggests suspended sediment is well-mixed with regards to sediment age and variation is due to incongruent weathering processes.



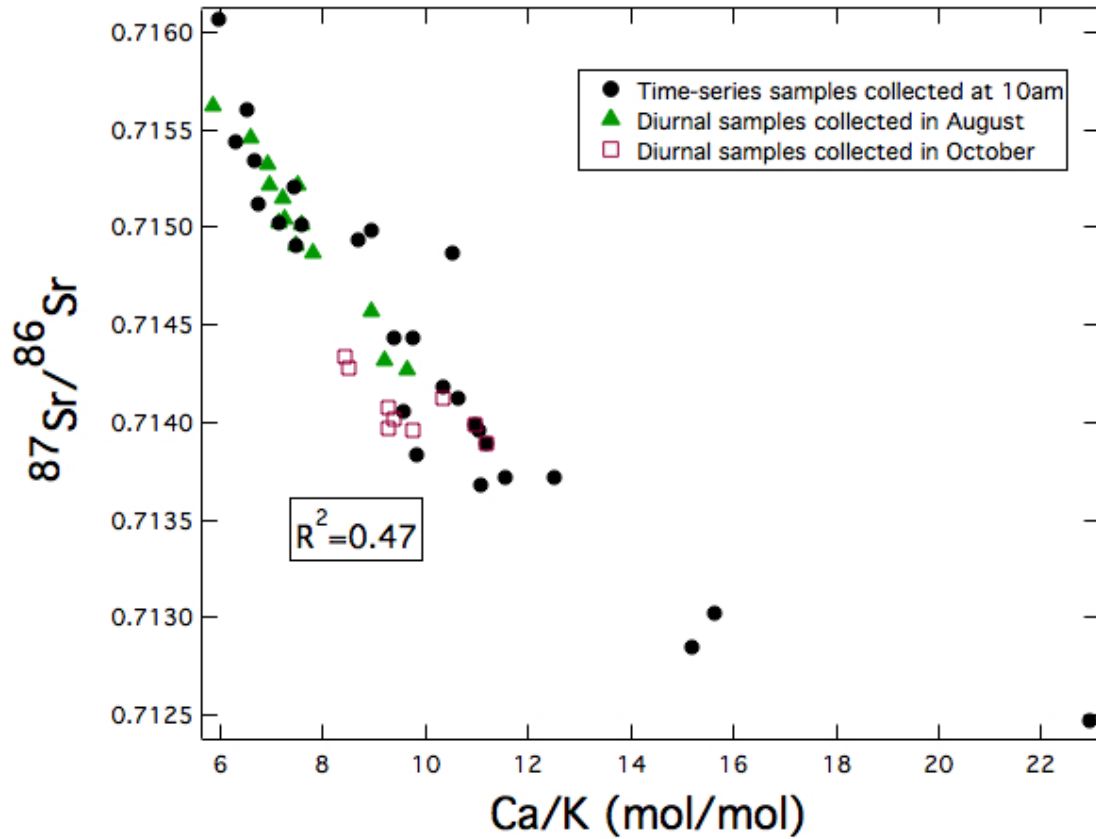
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Figure 7: Correlation between $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of the suspend sediment and total outflow discharge at the AG. Comminuted sediment mantles the bedrock with the greatest post-comminution age (i.e. greatest $^{87}\text{Sr}/^{86}\text{Sr}$ ratio) at the bedrock-sediment interface. At low discharge levels, the mantle is progressively deepened with increasing meltwater volumes. However, once a critical discharge level is reached ($\sim 0.9 \text{ m}^3 \text{s}^{-1}$), sediment excavation is non-discriminatory and the wide range of $^{87}\text{Sr}/^{86}\text{Sr}$ ratios can relate to processes such as exposure of previously isolated cavities.



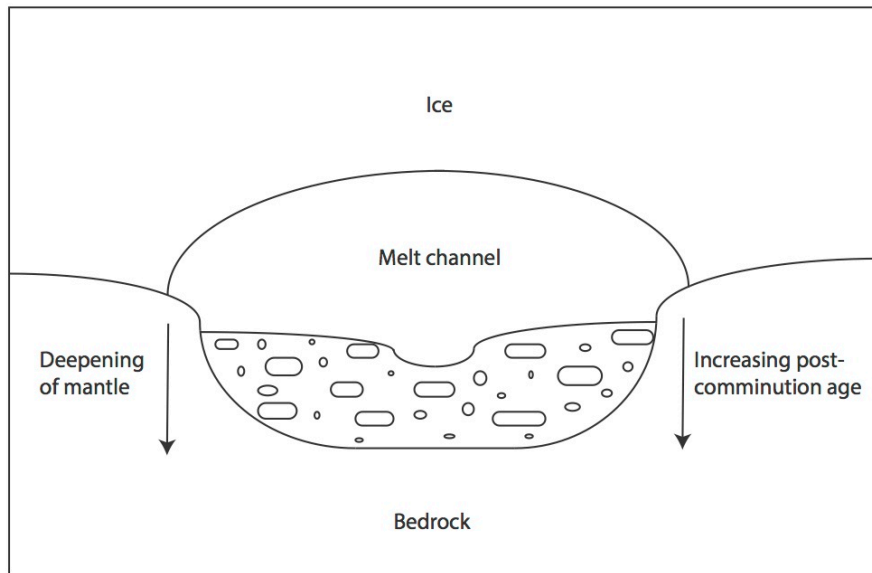
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Figure 8: Plot relating $^{87}\text{Sr}/^{86}\text{Sr}$ and Rb/Sr ratios. A stronger correlation exists between the time-series measurements ($R^2=0.79$) in comparison to the complete dataset which includes both diurnal sample sets ($R^2=0.57$).



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Figure 9: Relationship between $^{87}\text{Sr}/^{86}\text{Sr}$ and Ca/K in suspended sediment of the AG. Despite the predominately carbonate bedrock, the inverse relationship suggests preferential release of low $^{87}\text{Sr}/^{86}\text{Sr}$ bearing minerals during the initial stages of carbonate weathering followed by the release of higher $^{87}\text{Sr}/^{86}\text{Sr}$ bearing minerals during silicate weathering.



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 1017 Figure 10: Simplified diagram of the deepening of the subglacial mantle with increasing
 1018 meltwater excavating the comminuted sediment from the channel.
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1020 Supporting Online Material: Appendices 1 and 2 present the hydrophysical data and
 1021 Appendices 3 and 4 present the isotopic data of the Lemon Creek Glacier and Athabasca
 1022 Glacier, respectively. Additional figures provide support for the use of the coupled Nd-Sr
 1023 proxy. Figure 11 depicts the correlation between the Sm/Nd and discharge of the
 1024 suspended sediment at the AG and Figure 12 plots the ϵ_{Nd} and Ca/K of the suspended
 1025 sediment at the AG. External reproducibility of 0.49 is included. The lack of correlation
 1026 between K^+ -bearing minerals (e.g. biotite, clays) indicates there is no trend between the
 1027 age of the sediment and excavation processes, which further suggests the suspended
 1028 sediment was well-mixed prior to the deposition of comminuted sediment.
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