

Diethanolamine-Induced Alteration of Hepatic Mitochondrial Function and Structure¹

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Diethanolamine-Induced Alteration of Hepatic Mitochondrial Function and Structure. BARBEE, S. J., AND HARTUNG, R. (1979). *Toxicol. Appl. Pharmacol.* 47, 431-440. Diethanolamine (DEA) has been shown to interfere with phospholipid metabolism. It was hypothesized that DEA may alter membrane-bound biochemical processes which are dependent upon phospholipids. DEA was investigated for its effect on hepatic mitochondrial function and structure in the male Sprague-Dawley rat. DEA did not produce mitochondrial alterations in *in vitro* preparations from liver tissue. Similarly, acute treatment *in vivo* was without effect. However, a loss of mitochondrial integrity developed after subacute administration of DEA. Rats were given 0.25 (42 mg/kg/day), 1.0 (160 mg/kg/day), or 3.0 (490 mg/kg/day) mg/ml in their drinking water for varying periods up to 5 weeks. The effects on mitochondria were dose- and time dependent. Significant alterations appeared after 3 days at 3.0 mg/ml and after 1 week at 1.0 mg/ml. Mitochondrial State 4 activity was significantly elevated at all three dose levels following 2 weeks of DEA treatment. Concomitantly, an increase was noted in the Mg²⁺-dependent ATPase activity from rats treated with DEA at 2.0 mg/ml for 3 weeks. Electron micrographs indicated that treated animals had swollen hepatic mitochondria. The changes in the organelle may be related to alteration of phospholipid metabolism by DEA.

Diethanolamine (DEA), a member of a class of compounds known as the alkanolamines, is used in the chemical industry as an intermediate in the production of detergents, solubilizers, and textile finishing agents, and as an absorbent for acidic gases (Sutton, 1963). It is used in the pharmaceutical industry as a buffer and stabilizer for certain drugs (Soreat, 1973). Many biologically

exogenous alkanolamine compounds have been shown to interact with phospholipid metabolism (Welch and Landau, 1942; Wells and Remy, 1961; Chaplin and Mulford, 1961; Chojnacki and Korzybski, 1963; Morin, 1969). DEA can be incorporated *in vivo* into hepatic phospholipids (Artom *et al.*, 1958; Barbee and Hartung, 1979) and can inhibit the *in vitro* and *in vivo* synthesis of phospholipid derivatives of choline and ethanolamine (Barbee and Hartung, 1979). Choline and ethanolamine are essential in lipid metabolism (Fleischer and Rouser, 1965; Fleischer *et al.*, 1967) and membrane-bound enzymatic processes (Green, 1959; Green and Fleischer, 1963; Lenaz, 1973). The ability of DEA to compete with choline

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and ethanolamine in phospholipid metabolism raises the question of whether it exerts an effect on membrane-bound enzymatic processes. Foster (1971) noted that repeated oral administration of DEA reduced the activity of hepatic microsomal drug-metabolizing enzymes from the rat. This information suggests that DEA may exert an effect on other membrane-bound cellular processes.

The purpose of this study was to investigate whether DEA alters the function and structure of the mitochondrion from hepatic tissue of the rat. This organelle was chosen for study because it is dependent upon phospholipids for proper function (Green and Fleischer, 1963) and it is intimately involved in cell metabolism (Chance and Hollunger, 1961; Chance, 1961).

METHODS

Male albino rats of the Sprague-Dawley strain were used throughout this investigation at an initial weight of 150–200 g. They were maintained in air-conditioned quarters and supplied with food and water *ad libitum*. All animals were allowed to acclimate for at least 3 days before they were used in experimentation. Sacrifice was accomplished by exsanguination via open chest heart puncture after the animal was anesthetized with ether.

Diethanolamine (DEA) was purchased from the J. T. Baker Chemical Company, and it was used without further purification.

DEA was obtained as the free base, and as such was not suitable for direct administration because of its high pH in water. A desired amount of DEA was placed in a beaker with an equal volume of water. The beaker was placed in a water ice bath and neutralized to pH 7.4 with HCl. Throughout the neutralization process, the temperature was kept below 30°C to minimize alteration of the compound.

The *in vitro* and acute *in vivo* effect of DEA on hepatic mitochondria was assayed at 5 mM and 490 mg/kg, respectively. To assess the subacute *in vivo* effect of this substance on hepatic mitochondrial function DEA was administered in the drinking water according to the following dosing regimen: 0.25 mg/ml (42 mg/kg/day) for 2 and 5 weeks, 1.0 mg/ml (160 mg/kg/day) for 1, 3, and 5 weeks, and 3.0 mg/ml (490 mg/kg/day) for 1 day, 3 days, and 1, 2, 3, and 5 weeks. Water consumption was monitored to approximate the dose consumed in mg/kg/day. The dosages were calculated in terms of the free base.

Hepatic mitochondria were isolated according to the method of Brabec *et al.* (1974). The respiratory rate, acceptor control ratio, and ADP/O for hepatic mitochondria were determined using a Model 53 oxygen electrode system from Yellow Springs Instrument Company as described by Estabrook (1967) in 0.15 M sucrose, 20 mM potassium chloride, 20 mM Tris (pH 7.0 at 30°C), 10 mM potassium phosphate (monobasic), 5 mM magnesium chloride, and 100 mM succinate from the method of Brabec *et al.* (1974).

The medium was added in a volume of 2.4 ml combined with 0.1 ml of succinate. The succinate was prepared by placing the desired amount in a volumetric container and adjusting the pH to 7.0 with potassium hydroxide. The system was maintained at 30°C in a circulating water bath. The oxygen electrode was set in place and the system was allowed to equilibrate at 100% of oxygen saturation. The nanomoles of oxygen contained in the medium were calculated based on volume and temperature (Strickland and Parsons, 1968). The assay was initiated by the addition of 50 μ l of mitochondrial suspension and 200 nmol ADP. The ADP causes an immediate increase in the rate of oxygen utilization; this rate (State 3) is maintained until the added ADP is phosphorylated. At this time, ADP becomes limiting and the rate of oxygen consumption decreases (State 4). State 3 and State 4 activity may be determined by calculating the nanomoles of oxygen consumed per minute and relating this to milligrams of mitochondrial protein added. The acceptor control ratio is determined by dividing State 3 activity by that of State 4. The ADP/O ratio is calculated by dividing the nanomoles of ADP phosphorylated by nanoatoms of oxygen consumed during each interval at State 3.

The study to assess the subacute effects of DEA on hepatic mitochondrial membrane permeability was divided into two parts, evaluating the inner and outer mitochondrial membranes. The method and duration of administration and dose of DEA were kept constant for both investigations. Each experimental and control group consisted of four animals each. The experimental group was given 3.0 mg/ml DEA in their drinking water for 2 weeks.

Outer mitochondrial membrane permeability was assayed according to a modification of the method of Wattiaux-DeConnick and Wattiaux (1971) by following the spectrophotometric reduction of cytochrome *c* by sulfite-cytochrome *c* reductase at 550 nm. Mitochondria were prepared as previously described. The reagents consisted of a buffer, prepared by placing 1.0 ml of 0.2 M Tris and 0.1 mM EDTA (pH 8.5) in 9 ml of isolation medium, 0.01 M sodium sulfite, 0.25 mM cytochrome *c*, 13.5 mM potassium cyanide, buffered to pH 8.0 with the Tris-EDTA solution described above, and mitochondria diluted

1:10 with 0.25 M sucrose just prior to assay to achieve 2–3 mg protein/ml. Two milliliters of buffer, 0.1 ml potassium cyanide, 0.1 ml sodium sulfite, and 50 μ l of mitochondrial suspension were introduced into 3-ml cuvettes. In order to establish a recorded baseline, the system was allowed to equilibrate 4 min before addition of substrate. The reaction was started by adding 0.2 ml cytochrome *c*. The progress of the reaction was monitored for approximately 20 min at 30°C in a Gilford recording spectrophotometer.

Cytochrome *c* does not freely permeate the outer membrane. However, when the outer membrane is sufficiently altered by detergents or toxic substances, the cytochrome *c* is free to migrate through the membrane where it is reduced, then back out into the medium where it is measured. Potassium cyanide is added to the medium to prevent the reoxidation of cytochrome *c* by cytochrome oxidase. The activity of sulfite-cytochrome *c* reductase is expressed as Δ OD/min/mg protein.

The permeability of the inner membrane was determined following the method of Byard *et al.* (1975). Mitochondria were isolated and assayed for oxygen consumption as previously described using NADH instead of succinate as the substrate. The results were expressed as the difference in respiration between an excess of NADH and ADP compared to no added NADH and an excess of ADP with units given as nmol O₂/min/mg protein. According to Van den Bergh and Slater (1962), extramitochondrial NADH itself cannot penetrate the mitochondrial

membrane. If the mitochondria are sufficiently damaged, NADH can enter the inner compartment and be oxidized at the expense of oxygen.

Mitochondrial ATPase was measured following the method of Pederson and Morris (1974). Treated animals received 3.0 mg/ml DEA in their drinking water for 3 weeks.

Hepatic tissue was taken for ultrastructural examination as described by Sibrack *et al.* (1974). Animals in the treated group received drinking water containing 3.0 mg/ml DEA for 2 weeks.

Protein concentration was determined by the method of Lowry *et al.* (1951).

Statistical significance of the data was determined by analysis of variance and Student's *t* test.

RESULTS

DEA was ineffective in producing hepatic mitochondrial alteration *in vitro*; and similarly, acute *in vivo* treatment for 24 hr was without effect. These data are presented in Table 1.

When DEA was administered subacutely, it produced significant alteration of hepatic mitochondrial integrity. No significant effect was observed in renal mitochondria exposed *in vitro* and *in vivo* to DEA.

The results of the subacute effect of DEA

TABLE 1
LACK OF EFFECT OF *in Vitro* AND ACUTE *in Vivo* EXPOSURE OF DEA
ON HEPATIC MITOCHONDRIA^a

Treatment	Mitochondrial variable			
	State 3 activity ^b	State 4 activity ^b	Acceptor control ratio ^c	ADP/O ^d
Control	134.9 ± 1.9	25.2 ± 0.5	5.37 ± 0.03	1.46 ± 0.02
<i>In vitro</i>	140.4 ± 2.2	25.9 ± 0.6	5.42 ± 0.04	1.45 ± 0.01
<i>In vivo</i>	135.8 ± 3.0	25.7 ± 0.7	5.29 ± 0.03	1.45 ± 0.02

^a Animals administered the acute dose received 3 mg/ml (490 mg/kg/day) DEA in their drinking water for 24 hr. The *in vitro* effect was determined by isolating control mitochondria and assaying them in the presence of 5 mM DEA. Each value represents the mean ± SE of data from at least four animals.

^b Expressed as nmol O₂/mg protein/min.

^c Expressed as the ratio of State 3/State 4.

^d Expressed as the ratio of nmol ADP phosphorylated to nanomoles O₂ consumed during State 3 respiration.

on hepatic mitochondrial State 4 activity and acceptor control ratio are presented in Fig. 1. The data indicate a continued increase in State 4 for up to 3 weeks of DEA administration, after which it remains at more or less constant levels compared to those attained after 3 weeks of dosing. The increases in State 4 are dependent both on the dose and

the duration of administration. DEA given at 1.0 mg/ml for 1 week produces no increase compared to controls, whereas 3.0 mg/ml for the same time produces a 9% increase. When 1.0 mg/ml is given for 3 weeks, a 45% increase is seen. The dose at 0.25 mg/ml increases State 4 activity to only 12% above controls after 2 weeks, and this level is main-

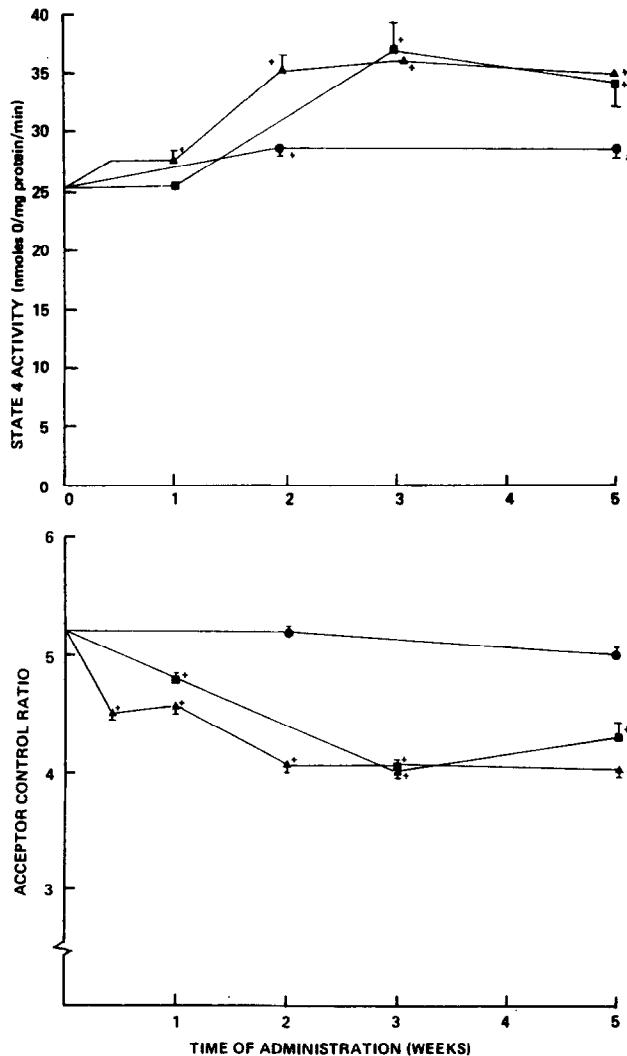


FIG. 1. The upper figure represents hepatic mitochondrial State 4 activity after subacute oral administration of DEA. The lower figure represents hepatic mitochondrial acceptor control ratio after subacute oral administration of DEA. DEA was administered orally in drinking water at 0.25 mg/ml (42 mg/kg/day) ●—●, 1.0 mg/ml (160 mg/kg/day) ■—■, and 3.0 mg/ml (490 mg/kg/day) ▲—▲. Each point represents the mean \pm SE of data from at least four animals. A plus sign (+) indicates values which are statistically significant ($p < 0.05$).

tained throughout 5 weeks of administration. Significant differences were obtained for all values except after 1.0 mg/ml DEA for 1 week.

The decline in the acceptor control ratio (ACR) in hepatic mitochondria is the result of an increase in State 4 since no change was observed in State 3 activity. As with State 4 activity, the effect on the ACR is dose and time dependent. The only dose which did not show a significant difference from control values is that at 0.25 mg/ml. The ACR calculated for animals receiving 1.0 and 3.0 mg/ml declined to 75% of control after 2 and 3 weeks of administration. This value is maintained for both dose levels through 5 weeks of testing.

To provide an explanation for the increase in State 4, Mg^{2+} -dependent ATPase was determined in animals given 3.0 mg/ml DEA for 2 weeks (Table 2). A significant increase in the activity of this enzyme occurred. The rise was comparable to that observed for State 4 activity. Since State 4 activity is defined as respiration in the absence of ADP, any endogenous process providing a source of ADP would stimulate electron transport, and hence oxygen consumption by mitochondria.

The ADP/O ratio did not exhibit significant differences at any time or dose level. This may appear anomalous, but two factors must be considered in connection with these

findings. First, ADP/O ratios are determined when the mitochondrion is in State 3 respiration, and very little change was observed from controls during this phase of respiration. Second, mitochondria which are tightly coupled always have high ADP/O ratios, and mitochondria which are loosely coupled may or may not have low ADP/O ratios. It is of interest that the acceptor control ratio is more indicative of mitochondrial integrity than is the ADP/O ratio (Chance and Williams, 1955).

The permeability of the mitochondrial membranes was determined to try to explain partially the increase in State 4 activity. Although experimental animals treated with 3.0 mg/ml DEA for 2 weeks showed no evidence of differences in permeability from controls, the possibility still exists that DEA treatment could result in alterations in permeability. The evaluation of inner and outer membrane permeability, in this study, is based on the passage of relatively large molecules through the membrane, NADH and cytochrome *c* with molecular weights of 709 and 12,270, respectively. A point to remember regarding NADH is that it is actively excluded from mitochondria, thus the cytoplasmic and intramitochondrial pools of this molecule are effectively segregated (Van den Bergh and Slater, 1962). These substances are considerably larger than ions, whose entry into the cell and its organelles also is selectively controlled. The possibility cannot be overlooked that DEA-treated mitochondria possess altered permeability toward small ions without any observable change in the permeability to much larger molecules. Data consistent with this statement are illustrated in Fig. 2. These are transmission electron micrographs of sections from hepatic tissue. Figure 2a represents untreated tissue, while Fig. 2b illustrates a section from an animal treated with 3.0 mg/ml DEA for 2 weeks. The mitochondria from DEA-treated animals are consistently spherical, and also appear larger than mitochondria from control animals.

TABLE 2

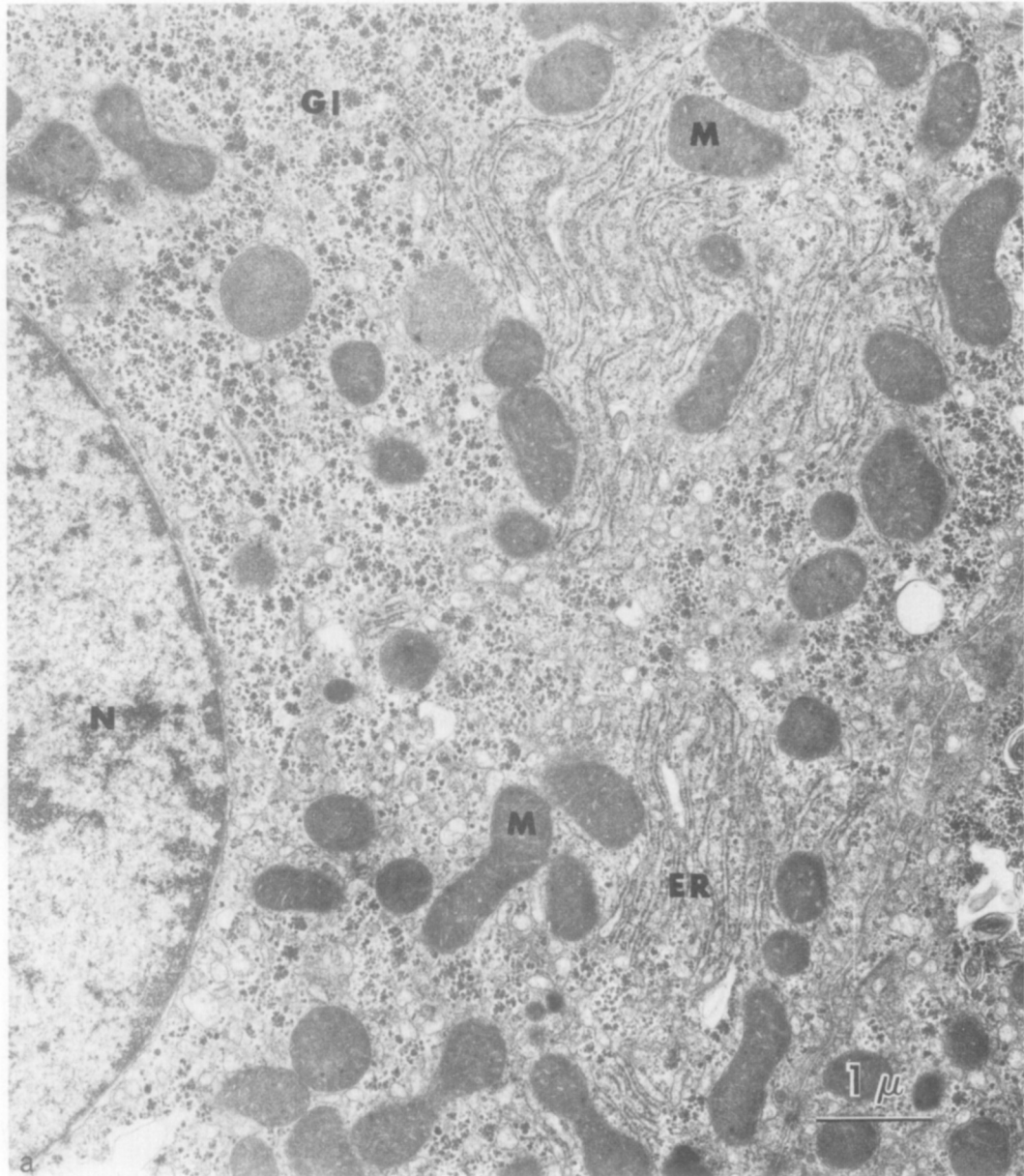
COMPARISON OF Mg^{2+} -DEPENDENT ATPase ACTIVITY IN HEPATIC MITOCHONDRIA ISOLATED FROM CONTROL AND DEA-TREATED ANIMALS^a

Animal group	ATPase activity ^b
Control	42.3 + 4.1
DEA treated	69.7 + 4.8 ^c

^a Each value represents the mean \pm SE of data from four animals. Treated animals were given 3.0 mg/ml (490 mg/kg/day) DEA via the drinking water for 2 weeks.

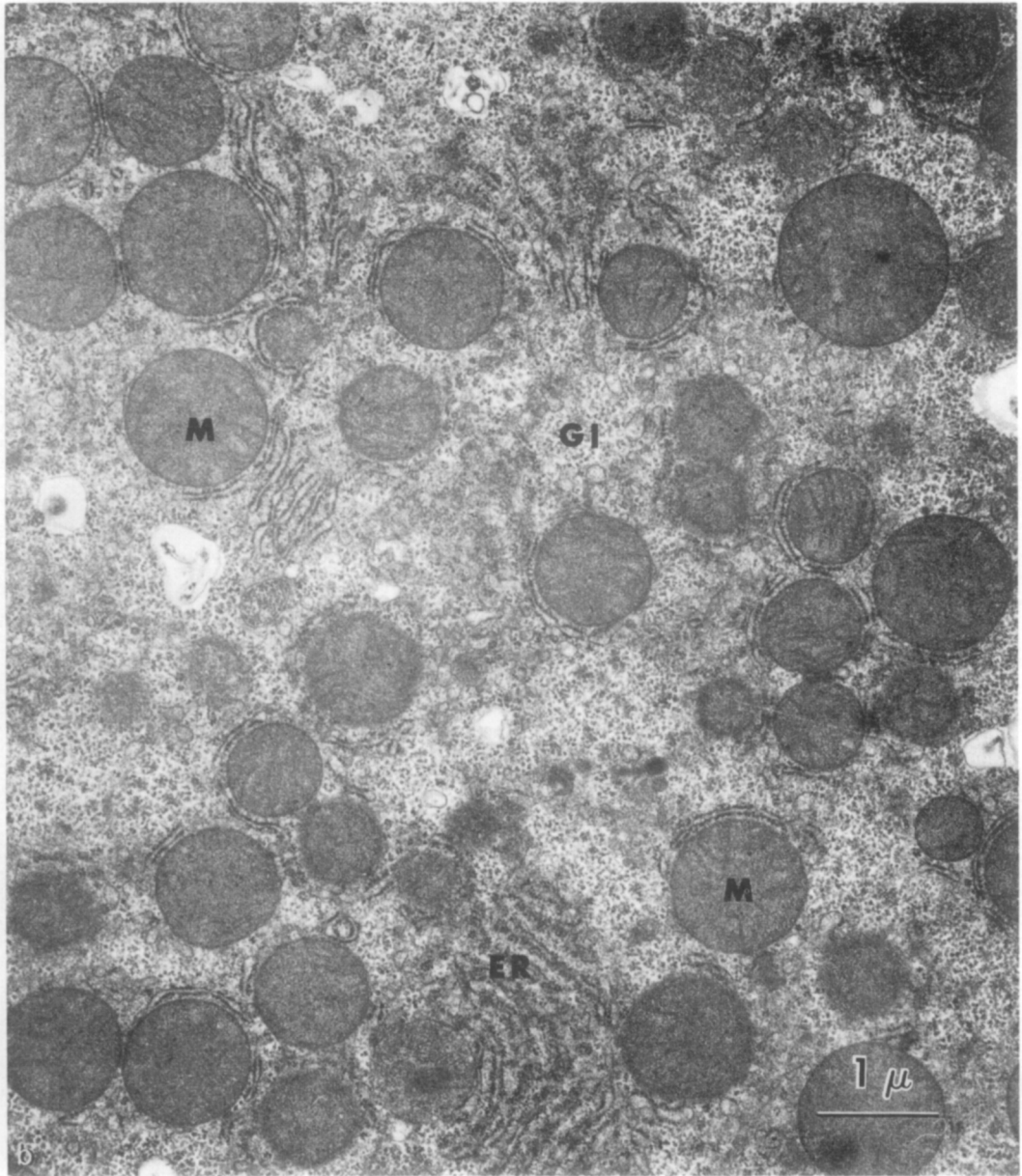
^b Expressed as nmol P_i released/mg protein/min.

^c $p < 0.05$.



(a)

FIG. 2. Transmission electron micrographs of hepatic tissue from the rat. (a) Tissue from a control animal; (b) tissue from a rat fed diethanolamine at a concentration of 3 mg/ml in drinking water for 2 weeks. The following structures may be identified: mitochondria (M), endoplasmic reticulum (ER), cell nucleus (N), and glycogen deposits (Gl). (a) $\times 15,750$; (b) $\times 16,500$.



(b)

DISCUSSION

Repeated administration of DEA to rats produces alteration of function in hepatic mitochondria. No observable change is noted in this organelle upon *in vitro* and acute *in vivo* exposure to DEA. The functional alteration seen in hepatic mitochondria is due to an increase in State 4 activity caused in part, if not completely, by elevation in the Mg^{2+} -dependent ATPase activity.

The ATPase activity in mitochondria is normally low, but it is greatly stimulated by *in vitro* treatment with agents that alter membrane permeability (Mitchell, 1961). Interestingly, DEA exerts no effect on State 4 activity *in vitro*. Barbee and Hartung (1979) noted that subacute administration of DEA inhibits the *in vivo* synthesis of phosphoglycerides of choline and ethanolamine. These investigators also found that DEA can serve as a substrate in phospholipid metabolism *in vitro*, and confirmed that it can be incorporated into hepatic phospholipids *in vivo*. The interaction of DEA in phospholipid metabolism may result in the incorporation of DEA into mitochondrial membrane structure as a phospholipid derivative.

Since mitochondrial function is highly dependent upon phospholipids, any alteration in phospholipid metabolism could easily result in functional change in this organelle. Atypical constituents may produce alterations in the membrane, thereby increasing mitochondrial membrane permeability. Even though no change was observed in permeability to cytochrome *c* and NADH, the possibility still exists that alterations have occurred to allow lower molecular-weight substances easier access through the mitochondrial membrane, e.g., hydrogen ion (H^+) and alkali metal ions.

Lipids are impermeable to protons and thereby act as an insulator to inhibit H^+ transport (Scarpa and DeGier, 1971). Cereijo-Santalo (1967) noted that intact mitochondria are impermeable to H^+ and he concluded that the latency of ATPase activity is due to

accumulation of H^+ ions in the membrane during that portion of respiration in which ATP is not being actively synthesized (State 4). The removal of these protons due to increased permeability or by exchange with other cations may act to unmask the latent ATPase and thus increase State 4 activity.

Influx of alkali-metal ions can increase ATPase activity when the mitochondrial membrane is made more permeable to these ions through treatment with the detergent Triton X-100 (Cereijo-Santalo, 1968). The membrane appears to be susceptible to differential structural alteration, since the author found that as the detergent concentration rose, the alkali-metal ions became increasingly more permeable in the following order: Li^+ , Na^+ , K^+ , Rb^+ , Cs^+ . This sequence, in order of their decreasing hydrated ionic radii, may indicate the formation of flaws or pores in the membrane which increase in diameter with increasing concentration of detergent. Atypical membrane constituents may produce similar flaws or pores thereby altering the permeability to selected elemental ions.

Cockrell *et al.* (1966) found that mitochondria in a medium of K^+ , Mg^{2+} , P_i , and an oxidizable substrate exhibited a stimulation of oxygen uptake and swelling when made more permeable to K^+ . They attribute the swelling to permeable anions which accompany the cations. Chappell and Crofts (1965) found that the influx of K^+ ions into mitochondria was followed by H^+ ejection in order to maintain ion balance, and mitochondrial swelling due to P_i entry. It is possible that small increases in the permeability of the mitochondrial membrane toward certain cations will produce swelling and elevations in oxygen consumption without affecting the ability to synthesize ATP. This may offer an explanation for the swollen appearance of hepatic mitochondria from rats treated with DEA.

It is not possible, at this time, to define the mechanism by which DEA produces alteration of hepatic mitochondrial function and

structure, although it presumably is related to the involvement of DEA with phospholipid metabolism. DEA does undergo incorporation to a phospholipid derivative *in vivo*, and, as such, is probably incorporated into mitochondrial membrane structure. It is tempting to speculate that prolonged exposure to DEA could alter lipid environments of other hepatic cellular constituents, or perhaps in other tissues, creating changes in other physiologic processes.

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