

THE STRUCTURE OF CRYSTALLINE AND MEMBRANE-BOUND GRAMICIDIN A  
BY VIBRATIONAL ANALYSIS

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Received October 15, 1984

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The results of normal mode calculations on the  $\beta^{4.4}$ ,  $\beta^{6.3}$ ,  $\uparrow\uparrow\beta^{5.6}$ , and  $\uparrow\uparrow\beta^{7.2}$  structures of gramicidin A are compared with infrared and Raman spectra of crystalline native, crystalline  $\text{Cs}^+$ -bound, and vesicle-bound gramicidin A. The observed frequencies and frequency splittings are in good agreement with an assignment of  $\uparrow\uparrow\beta^{5.6}$ ,  $\uparrow\uparrow\beta^{7.2}$ , and  $\beta^{6.3}$  structures, respectively, to the gramicidin A molecules in the above three systems. © 1984 Academic Press, Inc.

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Gramicidin A (GA), an antibiotic polypeptide isolated from Bacillus brevis, is a linear alternating L,D pentadecapeptide which facilitates the passive transmembrane transport of ions in biological membranes and synthetic lipid bilayers (1,2). This transmembrane channel has been shown to consist of two molecules of GA (3-8). The detailed structure of the dimer has been under investigation for a number of years, with two main models having been proposed: a formyl end-to-formyl end hydrogen-bonded association of single  $\beta^{6.3}$  helices (9-11), and a double-stranded  $\beta$ -helix structure with a most probable antiparallel alignment of hydrogen-bonded GA molecules (12-14). Conformational energy calculations on single- and double-stranded  $\beta$ -helical poly(L,D-alanine) (15-17) have yielded almost identical conformational energies for these two kinds of helices, thus providing no basis for the preference of one model over the other. In a recent calculation on the head-to-head dimer with side chains explicitly included, it was shown that left-handed  $\beta^{6.3}$  helices are of lower energy than right-handed (18).

Different experimental studies have indicated that GA can adopt single-stranded or double-stranded structures depending on the environment. Diffraction studies on GA crystallized in its native and various cation-bound forms

0006-291X/84 \$1.50

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(19,20), and low-resolution Fourier maps (21), are consistent with a helical molecule, but no detailed crystal structure has yet been published. However, the x-ray studies do show that the length of the dimer decreases from 32 to 26 Å, and the diameter of the cylinder formed by the peptide main chain atoms increases from 5 to 6.8 Å, on binding Cs<sup>+</sup> ions (20). From <sup>13</sup>C and <sup>23</sup>Na NMR studies and CD studies on GA and synthetic selectively <sup>13</sup>C-enriched GA (22-25), a head-to-head dimer structure has been inferred for the active transmembrane channel. A recent CD study (26) has emphasized the differences in structure between GA in membranes (a head-to-head dimer) and in organic solvents (a double-stranded helix), and a two-dimensional NMR study (27) has shown that a double-stranded  $\uparrow\uparrow\beta^{5.6}$  helix exists in dioxane.

There have been a number of infrared (IR) (12,28-35) and Raman (26,28, 35,36) studies on GA and its analogs. In some of the IR work (31,33,34), D<sub>2</sub>O was used in order to avoid the water absorption in the amide I region (1600-1700 cm<sup>-1</sup>), and neglect of the effect of deuteration of the peptide group on this frequency (see below) leaves these interpretations questionable. In one of the IR studies (34) the measurements were made on dried membrane preparations, leaving open the question of whether the structure in a dried specimen is the same as in an aqueous membrane environment, even assuming that the GA is still within the membrane. In all cases, except for our initial report (35), interpretations were based on qualitative correlations or approximate coupling calculations (30).

#### METHODS

In order to provide a reliable interpretation of the vibrational spectra of GA in terms of structure, we have used the powerful approach of normal coordinate analysis (37). We have done normal mode calculations (38) on all structures that are consistent with the observed dimer length and channel radii, viz., single-helical  $\beta^{4.4}$  and  $\beta^{6.3}$  and double-helical  $\uparrow\uparrow\beta^{5.6}$  and  $\uparrow\uparrow\beta^{7.2}$ . The normal modes were computed with the side chain approximated by a point mass equal to CH<sub>3</sub>, using a force field developed for  $\beta$ -poly(L-alanine) in this approximation (39). Transition dipole coupling (40,41) was included for amide I and amide II modes, using calculated, not assumed (30), eigenvectors. The calculated frequencies for amide I modes, expected to be accurate to about  $\pm 5$  cm<sup>-1</sup> (42), are given in Table 1, and are compared to IR and Raman data in the crystalline state (35) and in vesicles.

Table 1. Observed and Calculated Amide I Frequencies (in  $\text{cm}^{-1}$ ) for Single- and Double-Stranded Gramicidin A Structures

Structure	Calculated <sup>a</sup>			Observed			State
	$\nu(\text{IR})^b$	$\Delta\nu^c$	$\nu(\text{R})^d$	$\nu(\text{IR})$	$\Delta\nu$	$\nu(\text{R})$	
$\beta^{4.4}$	1631	13	1648-53				
$\beta^{6.3}$	1643	11	1652-54	-1640 <sup>e</sup>	-	1654 <sup>f</sup>	GA-lipid
$\uparrow\uparrow\beta^{5.6}$	1636	39	1656-75	1638 <sup>g</sup>	42 <sup>g</sup>	1666 <sup>g</sup>	GA (cryst)
$\uparrow\uparrow\beta^{7.2}$	1632	45	1651-86	1630 <sup>g</sup>	55 <sup>g</sup>	1668 <sup>g</sup>	GA-Cs <sup>+</sup> (cryst)

<sup>a</sup>Ref. 38. <sup>b</sup>Strong parallel-polarized IR mode (38). <sup>c</sup>Difference between strong parallel and strongest (though weaker) perpendicular IR modes (38). <sup>d</sup>Range of IR-weak frequencies; Raman mode in crystalline forms observed at mid-point of range. <sup>e</sup>Estimated from observations on vesicles (31,34) (see text). <sup>f</sup>This work. <sup>g</sup>Ref. 35.

### RESULTS AND DISCUSSION

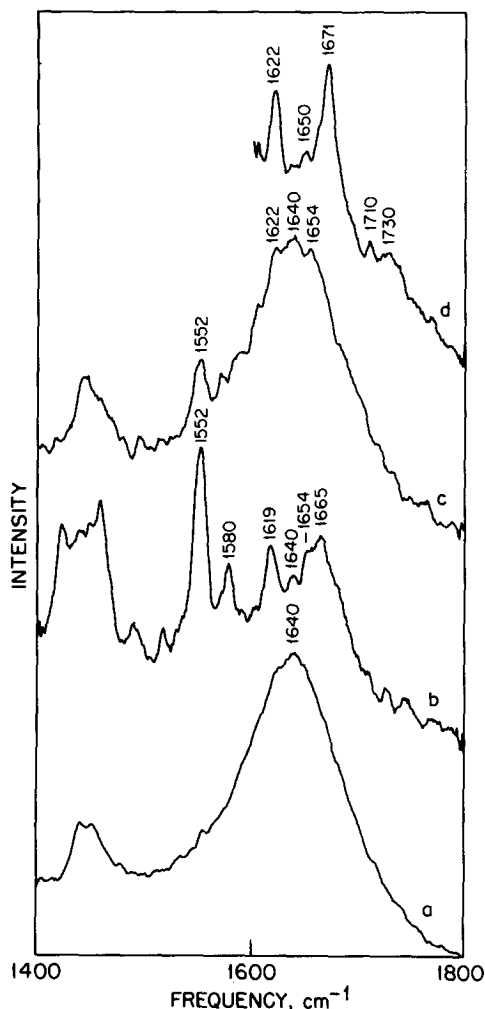
In the IR spectrum of native GA crystals there is a strong amide I band at  $1638 \text{ cm}^{-1}$  with a shoulder at  $\sim 1680 \text{ cm}^{-1}$ , corresponding to a splitting,  $\Delta\nu$ , of  $\sim 42 \text{ cm}^{-1}$  (35). On binding  $\text{Cs}^+$  or  $\text{K}^+$  the strong band shifts down to  $1632 \text{ cm}^{-1}$  and the higher frequency band becomes a distinct peak at  $1685 \text{ cm}^{-1}$ , giving  $\Delta\nu = 55 \text{ cm}^{-1}$  (35). The calculated frequencies of the strong (parallel polarized) bands, their shifts on ion binding, and the magnitudes and changes in  $\Delta\nu$  (the difference between the weak high frequency perpendicular mode and the strong low frequency parallel mode) are in excellent agreement with an assignment of the  $\uparrow\uparrow\beta^{5.6}$  structure to crystalline native GA and the  $\uparrow\uparrow\beta^{7.2}$  structure to crystalline ion-bound GA. An assignment to single-stranded helices is precluded primarily by the much smaller  $\Delta\nu$  calculated for these structures, but also by the poorer agreement with the observed strong IR and Raman modes (note that the latter are observed at about the mid-point of the range of calculated IR-weak frequencies). It should be mentioned that the  $\uparrow\uparrow\beta^{5.6}$  structure, with a channel radius for unhindered ion passage of  $0.9 \text{ \AA}$ , would be unlikely to hold a  $\text{Cs}^+$  ion, whose ionic radius is  $1.48 \text{ \AA}$ , whereas the  $\uparrow\uparrow\beta^{7.2}$  structure, with its comparable channel radius of  $2.0 \text{ \AA}$ , could easily do so. We also wish to note that a previous calculation of  $\Delta\nu$  (30) gave values ( $64 \text{ cm}^{-1}$ ) for the double-stranded helices much higher than ours, a result of

assuming that the eigenvectors for the amide I modes correspond to those of an infinite  $\beta$ -sheet structure rather than to those of the actual helical structure, which are obviously different and which we obtain from our normal mode calculations.

The situation for GA in lipid vesicles is more complicated because of the assignment and interpretation of the IR amide I mode. The measurements on a liposome suspension (31) and those on a film dried from vesicles (34), which gave amide I bands at 1634 and  $1633 \pm 1 \text{ cm}^{-1}$ , respectively, were both done after  $\text{D}_2\text{O}$  treatment. It is well known that deuteration of the peptide group causes the amide I frequency of  $\beta$  structures to decrease (43,44), and these observations indicate a likely decrease of about  $6 \text{ cm}^{-1}$  in the present case. Thus, the unshifted amide I mode for GA in vesicles should be at  $\sim 1640 \text{ cm}^{-1}$ . As is indicated by the calculations, the distinction between  $\beta^{6.3}$  and  $+\beta^{5.6}$  is therefore difficult to make on the basis of the strong IR mode alone, although the absence of a higher frequency component in the liposome system (31), which CD spectra (31) show to be in the conducting state (34), favors the  $\beta^{6.3}$  structure.

Since the calculations indicate that the Raman mode should be at a significantly lower frequency in single-stranded than in double-stranded structures, we have examined the Raman spectra of GA in vesicles. We have prepared such vesicles in  $\text{H}_2\text{O}$  by the method of Urry et al. (34), and the Raman spectra are shown in Fig. 1. The spectrum of vesicles alone is shown in Fig. 1a, the  $1640 \text{ cm}^{-1}$  band being due to  $\text{H}_2\text{O}$ . In the early stages of incorporation (Fig. 1b), a band is seen at  $1665 \text{ cm}^{-1}$ , accompanied by a weak band at  $1654 \text{ cm}^{-1}$ . In the conducting state (Fig. 1c), the former band is absent and only the  $1654 \text{ cm}^{-1}$  band is present. When the vesicles are dried (Fig. 1d), a strong band returns at  $1671 \text{ cm}^{-1}$ , with a weak band remaining at  $\sim 1650 \text{ cm}^{-1}$ .

These results can be interpreted as follows. The initial structure of GA is  $+\beta^{5.6}$  (possibly with some  $+\beta^{7.2}$  if complexing with  $\text{Na}^+$  occurs), having a characteristic Raman band near  $1667 \text{ cm}^{-1}$ . As incorporation proceeds, the structure converts to  $\beta^{6.3}$  with a characteristic Raman band near  $1654 \text{ cm}^{-1}$



**Figure 1.** Raman spectra (Spex 1403) of (a) vesicles in H<sub>2</sub>O, (b) gramicidin A in vesicles after 10 min. of incubation at 45°C, (c) gramicidin A in vesicles after 8-15 hr. of incubation at 60°C, (d) dried gramicidin A - vesicles. ( $\lambda_{exc} = 5145 \text{ \AA}$ , laser power  $\sim 100 \text{ mW}$ , resolution  $\sim 2 \text{ cm}^{-1}$ , co-addition of at least 20 scans per spectrum.)

(this is preferable to an assignment to  $\beta^{4.4}$ , not only because of the somewhat better agreement with the calculated Raman mode but also because of the much better agreement with the strong IR mode). This structure has also been proposed from NMR studies (22,25). On drying, the GA converts back substantially to the  $++\beta^{5.6}$  (or possibly  $++\beta^{7.2}$ ) structure, although it is not known whether or not the GA is still incorporated in the membrane. This accounts for the complex band structure in the dried film (34), whereas a simple IR band is seen for the liposome system (31).

Normal mode calculations thus provide a powerful base for analyzing the structure of gramicidin A from infrared and Raman spectra.

#### REFERENCES AND NOTES

1. Haydon, D.A., and Hladky, S. B. (1972) *Quart. Rev. Biophys.* 5, 187-282.
2. Krasne, S., Eisenman, G., and Szabo, G. (1972) *Science* 174, 412-415.
3. Urry, D.W., Goodall, M.C., Glickson, J.D., and Mayers, D.F. (1971) *Proc. Natl. Acad. Sci. USA* 68, 1907-1911.
4. Bamberg, E., and Lauger, P. (1973) *J. Membr. Biol.* 11, 177-194.
5. Kolb, H.A., Lauger, P., and Bamberg, E. (1975) *J. Membr. Biol.* 20, 133-154.
6. Veatch, W.R., Mathies, R., Eisenberg, M., and Stryer, L. (1975) *J. Mol. Biol.* 99, 75-92.
7. Apell, H.J., Bamberg, E., and Alpes, H. (1977) *J. Membr. Biol.* 31, 171-188.
8. Veatch, W., and Stryer, L. (1977) *J. Mol. Biol.* 113, 89-102.
9. Urry, D.W. (1971) *Proc. Natl. Acad. Sci. USA* 68, 672-676.
10. Urry, D.W. (1972) *Proc. Natl. Acad. Sci. USA* 69, 1610-1614.
11. Ramachandran, G.N., and Chandrasekaran, R. (1972) *Ind. J. Biochem. Biophys.* 9, 1-11.
12. Veatch, W.R., Fossel, E.T., and Blout, E.R. (1974) *Biochemistry* 13, 5249-5256.
13. Veatch, W.R., and Blout, E.R. (1974) *Biochemistry* 13, 5257-5264.
14. Fossel, E.T., Veatch, W.R., Ovchinnikov, Y.A., and Blout, E.R. (1974) *Biochemistry* 13, 5264-5275.
15. Lotz, B., Colonna-Cesari, F., Heitz, F., and Spach, G. (1976) *J. Mol. Biol.* 106, 915-942.
16. Venkataram Prasad, B.V., and Chandrasekaran, R. (1977) *Int. J. Peptide Protein Res.* 10, 129-138.
17. Popov, E.M., and Lipkind, G.M. (1979) *Mol. Biol. (USSR)* 13, 363-376.
18. Venkatachalam, C.M., and Urry, D.W. (1983) *J. Comp. Chem.* 4, 461-469.
19. Koeppe II, R.E., Hodgson, K.O., and Stryer, L. (1978) *J. Mol. Biol.* 121, 41-54.
20. Koeppe II, R.E., Berg, J.M., Hodgson, K.O., and Stryer, L. (1979) *Nature (London)* 279, 723-725.
21. Koeppe II, R.E., and Schoenborn, B.P. (1984) *Biophys. J.* 45, 503-507.
22. Weinstein, S., Wallace, B.A., Blout, E.R., Morrow, J., and Veatch, W. (1979) *Proc. Natl. Acad. Sci. USA*, 76, 4230-4234.
23. Weinstein, S., Wallace, B.A., Morrow, J., and Veatch, W.R. (1980) *J. Mol. Biol.* 143, 1-19.
24. Wallace, B.A., Veatch, W.R., and Blout, E.R. (1981) *Biochemistry* 20, 5754-5760.
25. Urry, D.W., Trapane, T.L., and Prasad, K.U. (1983) *Science* 221, 1064-1067.
26. Wallace, B.A. (1983) *Biopolymers* 22, 397-402.
27. Arseniev, A.S., Bystrov, V.F., Ivanov, V.T., and Ovchinnikov, Y.A. (1984) *FEBS Lett.* 165, 51-56.
28. Iqbal, Z., and Weidekamm, E. (1979) *Infrared Phys.* 19, 475-479.
29. Iqbal, Z., and Weidekamm, E. (1980) *Arch. Biochem. Biophys.* 202, 639-649.
30. Sychev, S.V., Nevskaya, N.A., Jordanov, S.T., Shepel, E.N., Miroshnikov, A.I., and Ivanov, V.T. (1980) *Bioorg. Chem.* 9, 121-151.
31. Sychev, S.V., and Ivanov, V.T. in *Membrane Transport* (1982) ed. A.N. Martonosi, Vol. 2, pp 301-307, Plenum, New York.
32. Nabadryk, E., Gingold, M.P., and Breton, J. (1982) *Biophysical J.* 38, 243-249.
33. Ovchinnikov, Yu. A., and Ivanov, V.T. in *Conformation in Biology* (1983) eds. R. Srinivasan and R.H. Sarma, pp 155-174, Adenine Press, New York.

34. Urry, D.W., Shaw, R.G., Trapane, T.L., and Prasad, K.U. (1983) *Biochem. Biophys. Res. Comm.* 114, 373-379.
35. Naik, V.M., and Krimm, S. (1984) *Biophys. J.* 45, 109-112.
36. Rothschild, K.J., and Stanley, H.E. (1974) *Science* 185, 616-618.
37. Naik, V.M., Krimm, S., Denton, J.B., Nemethy, G., and Scheraga, H.A., *Int. J. Peptide Protein Res.*, in press.
38. Naik, V.M., and Krimm, S., to be published.
39. Dwivedi, A.M., and Krimm, S. (1984) *J. Phys. Chem.* 88, 620-627.
40. Krimm, S., and Abe, Y. (1972) *Proc. Natl. Acad. Sci. USA* 69, 2788-2792.
41. Moore, W.H., and Krimm, S. (1975) *Proc. Natl. Acad. Sci. USA* 72, 4933-4935.
42. Krimm, S. (1983) *Biopolymers* 22, 217-225.
43. Dwivedi, A.M., and Krimm, S. (1982) *Macromolecules* 15, 177-185.
44. Dwivedi, A.M., and Krimm, S. (1982) *Macromolecules* 15, 186-193; (1983) 16, 340.
45. This is paper 28 in a series on "Vibrational Analysis of Peptides, Polypeptides, and Proteins," of which (37) is paper 27. This research was supported by a fellowship from the Macromolecular Research Center at The University of Michigan (to V.M.N.) and by NSF grants PCM-8214064 and DMR-8303610.