# **LETTER**

# REE/Fe variations in hydrothermal sediments: Implications for the REE content of seawater

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Abstract—Seafloor hydrothermal vent solutions exhibit rare earth element (REE) enrichments ranging between one to three orders of magnitude greater than average seawater. To assess the impact of these hydrothermal inputs on ocean chemistry, we have examined the behavior of REEs for hydrothermal sediments collected adjacent to two Pacific spreading ridge sites: the East Pacific Rise at 19°S, and the Southern Juan de Fuca Ridge at 45°N. In general, the REE/Fe ratios for both proximal and distal hydrothermal sediments are greater than vent solutions by a factor of 2 to 500, and these ratios increase with increasing distance away from the ridge axis. An evaluation of these results in the context of previous models of REE behavior indicates that, in fact, seawater experiences a net depletion in REEs as a result of hydrothermal activity. This is due primarily to the large scavenging capacity of iron oxyhydroxides which precipitate from these solutions. Such an interpretation explains why the REE content of seawater collected in the vicinity of hydrothermal vents is anomalously lower than "normal" seawater sampled from a comparable depth.

## INTRODUCTION

THE CHEMICAL BUDGET of the rare earth elements (REEs) in the marine environment is not well understood because our knowledge of the significant sources and sinks of these elements is incomplete. Much of this uncertainty concerns the relationship between seafloor hydrothermal activity and the REE mass balance in the oceans. For example, PIPER (1974) suggested REE inputs from seafloor hydrothermal activity may be partially responsible for the observation that, in the pelagic Pacific Ocean, REEs are accumulating at a rate of 11 to 18 times greater than their supply from rivers. Specific questions regarding the role of hydrothermal processes in the marine REE budget have been stated succinctly by FLEET (1984): "Overall, does seawater gain or lose REEs while undergoing hydrothermal circulation at mid-ocean ridges and. if it does gain REEs, does the gain have a significant effect on the composition of seawater, or are the REEs lost at or near ridges by incorporation in metalliferous sediments?"

Recent investigations suggest the answers to these questions lie in an assessment of the net effect of two competing processes. Analyses of hydrothermal vent waters (MICHARD et al., 1983; MICHARD and ALBAREDE, 1986; HINKLEY and TATSUMOTO, 1987) indicate these solutions are enriched in REEs by a factor of 10 to nearly 10<sup>4</sup> times that of ambient seawater. Too few analyses are now available to determine whether these REE levels are characteristic of "average" vent water; if they are, however, then hydrothermal REE inputs may represent a significant fraction of the total REE inputs to seawater. On the other hand, it is well documented that hydrothermal precipitates (especially Fe oxyhydroxides) that form in vent fluids can rapidly scavenge dissolved REEs (RUHLIN and OWEN, 1986a,b; OLIVAREZ and OWEN, 1987). Consequently, the enriched concentrations of REEs in vent

The present study examines these competing processes by developing a comparison between reported data on the REE chemistry of Pacific hydrothermal vent fluids and that of hydrothermal precipitates we have analyzed from the North and South Pacific Oceans. The net effect upon seawater is assessed by using the REE/Fe ratio of vent fluids and hydrothermal precipitates as an index of the fate of hydrothermal REE inputs.

## METHODS AND SAMPLE DESCRIPTION

The sediment chemical data discussed in this report were obtained from the analysis of sediment cores collected from two areas of known seafloor hydrothermal activity. The first of these was recovered approximately 1150 km west of the axis of the East Pacific Rise (EPR), at Deep Sea Drilling Project (DSDP) Site 598 (19°0.283'S, 124°40.606'W). The 43-meter column of sediments from this site represents a complete record of hydrothermal sedimentation along the western flank of the EPR during the past 16 million years and over paleodistances ranging from 9 to 1150 km from the ridge axis. A detailed description of the procedures used to analyze the 50 samples taken from this core, as well as the geochemical and mass accumulation rate data, is given in RUHLIN and OWEN (1986a,b). The second core was obtained during a 1985 cruise of the R/VSurveyor as part of the National Oceanic and Atmospheric Administration's (NOAA) VENTS Program. This was an 80 cm Kasten core, designated KC-1, recovered from a water depth of 2635 m approximately 15.6 km west (44°44.21'N, 130°33.38'W) of the axis of the Southern Juan de Fuca Ridge (SJFR). The whole spreading rate in this area is 6 cm/yr (ATWATER, 1970; ELVERS et al., 1974), and age dating of KC-1 sediments yields a linear sedimentation rate of 0.61 cm/ky (G. MASSOTH, pers. comm.). Assuming a constant linear sedimentation rate, we estimate the sediments in KC-1 include the upper 24% of the total sediment cover to basement and represent hydrothermal deposition during the past 130 thousand years over paleodistances ranging from approximately 12 to 16 km from the rise crest. A total of 20 samples collected at intervals of 1 to 5 cm were taken from this core, freezedried, and ground with mortar and pestle prior to analysis. Each sample was analyzed for the elements Al, Ba, Co, Fe, Mg, Mn, Ni, Si, Ti, and the rare earth elements La, Ce, Nd, Sm, Eu, Tb, Yb, and

fluids are not entirely transferred to seawater, but instead are diminished to some degree by the scavenging action of hydrothermal precipitates.

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Lu by instrumental neutron activation analysis (INAA) at the Phoenix Memorial Laboratory, The University of Michigan, using standard INAA procedures (GORDON et al., 1968; DAMS and ROBBINS, 1970). Calcium carbonate content was measured using a conventional pressure bomb technique. Statistical analyses of the chemical data were performed using the Q-Mode factor analysis procedure of LEINFN and PISIAS (1984), kindly provided to us by these authors, and using MIDAS, a package of computer programs provided by the Statistical Research Laboratory at The University of Michigan (FOX and GUIRE, 1976).

#### RESULTS AND DISCUSSION

## Composition of the hydrothermal component

Different approaches were required to characterize the composition of the hydrothermal component of the sediments at the two sampling locations. For Site 598, this process was straightforward because the sediments here are composed of a simple mixture of only two components: hydrothermal precipitates and biogenous carbonates (RUHLIN and OWEN, 1986a,b). Biogenous carbonates contain uniformly low concentrations of both Fe and the REEs (PALMER, 1985), and it can be assumed that the shell material acts as a simple diluent (i.e. it contributes negligibly to the total amount of Fe and REEs present, relative to the hydrothermal component). Consequently, the Fe and REE composition of the hydrothermal component of Site 598 sediments is represented by the bulk sediment concentrations of these elements corrected for the amount of carbonate material present (RUHLIN and OWEN, 1986a).

The sediments from SJFR are more complex because they contain detrital material in addition to biogenous and hydrothermal components. Because detrital material can contain significant amounts of both Fe and REEs, it was necessary to isolate and distinguish the composition of this phase relative to the hydrothermal materials present. The Q-mode factor analysis of LEINEN and PISIAS (1984), which is designed to identify the composition of geochemically significant endmembers in a bulk chemical data set, was used for this purpose. This analysis identified three compositional end-members (factors), which collectively account for 97% of the total variance in the chemical data set for the SJFR samples (Table 1).

Our interpretations of the geochemical significance of the three compositional end-members are as follows: Factor 1 = proximal hydrothermal component; Factor 2 = distal hydrothermal component; Factor 3 = detrital component. These interpretations were based on comparisons between key interelement ratios in each end-member and analogous ratios reported in the literature and considered to be characteristic for various marine sedimentary phases. For example, chemical fractionation of Fe from Mn occurs within a hydrothermal plume as it wafts from proximal to distal locations (MOTTL and McCONACHY, 1989). This change should also be reflected in the composition of hydrothermal precipitates from the plume. The fractionation of Fe and Mn has been reported by MASSOTH et al. (1984) for surficial sediments along the SJFR and both Factors 1 and 2 exhibit properties consistent with an evolving plume. The Mn/Fe ratio of Factor 1 is 0.07, which is in agreement with values (Mn/Fe = 0.06-0.07) reported for proximal metalliferous sediments in the North and South Pacific (PIPER, et al. 1975; BONATTI and

Table 1. End-member composition scores (in ppm, except as noted) for KC-1 samples resulting from oblique rotation of Varimax axes.

	Factor 1	Factor 2	Factor 3 #
Al%	4.82	2.75	7.35
Ba	5210	8900	5180
Co	44.9	101	25.1
Fe%	6.30	6.24	5.48
Mg	1020	3620	7830
Mn	4580	27000	4170
Ni	0.00	245	13.9
Si%	24.0	23.4	21.4
Ti	1960	220	4800
الما	44.0	60.2	23.9
Ce	48.4	32.3	38.7
Nd	54.3	78.0	56.6
Sm	8.41	10.1	8.09
Eu	2.10	2.34	1.93
Тъ	2.32	0.00	0.00
Yb	4.82	6.57	4.28
Lu	1.41	1.28	1.32
% Variance	42	30	25
% Total Variance	42	72	97

<sup>#</sup> Factor 1=proximal hydrothermal; Factor 2=distal hydrothermal; Factor 3=detrital component.

JOENSUU, 1966) and with hydrothermal plume particulates sampled in the proximal region ( $\sim$ 7 km from the axis) along the SJFR (G. MASSOTH, unpubl. data). The Mn/Fe ratio of Factor 2 is 0.43, which is close to those values reported for "typical" hydrothermal sediments (0.31-0.41) by GRAYBEAL and HEATH (1984). The value of this ratio also suggests that the sediments in core KC-1 have not undergone a significant degree of diagenetic alteration. For example, the Mn/Fe ratios for diagenetically altered ferromanganese deposits are usually much greater than 1.0 and typically range between 5 and 70 (BONATTI et al., 1972; DYMOND et al., 1984). Factor 3 is considered to represent a detrital component because several inter-element ratios (e.g. Al/Si, Mg/Al, Co/Al) fall within the range of values observed for lithogenous materials (GRAY-BEAL and HEATH, 1984; GROMET et al., 1984). (X-ray diffraction analyses of several samples from core KC-1 confirmed the presence of detrital minerals, including quartz, K-feldspar, albite, and clay minerals.)

Other aspects of the SJFR data support the interpretations described above. For example, the relative amounts of each compositional end-member present in each SJFR sample were calculated using a linear programming normative analysis model, in which it was assumed that the bulk composition of each sample can be represented by some linear combination of each of the three end-members. The results of these calculations are shown as down-core profiles of each endmember in Fig. 1. The proximal hydrothermal factor systematically increases in total relative abundance downcore (i.e. towards the ridge axis), while the distal hydrothermal factor increases in total relative abundance towards the top of the core. No systematic change is observed for the detrital factor. BONATTI (1981) has shown that ternary diagrams of Fe vs. Mn vs. Co + Ni + Cu can serve to distinguish hydrothermal and hydrogenous phases in ferromanganese deposits. Both the proximal and distal hydrothermal end-members interpreted here fall within the hydrothermal region of this type of plot (Fig. 2). Finally, we note that pure hydrothermal sed-

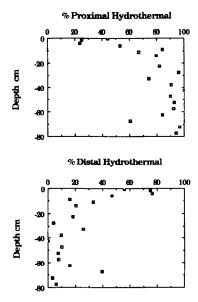


FIG. 1. Down-core profiles of the percentage fraction of each hydrothermal factor for SJFR core KC-1, reported on a detrital-free basis. Note that the proximal hydrothermal factor increases in abundance down-core (closer to the ridge axis) while the distal factor becomes more abundant at shallower depths.

iments have characteristic REE patterns which strongly resemble seawater patterns (BENDER et al., 1971; DYMOND et al., 1973). Previous studies have shown that the REE patterns of Site 598 (RUHLIN and OWEN, 1986a,b) and SJFR sediments (OLIVAREZ and OWEN, 1986) become more seawaterlike, and the Ce anomaly of Site 598 sediments approaches that of seawater with increasing distance from the rise crest (RUHLIN and OWEN, 1986a,b). The Ce anomaly (Ce\*) is defined as:  $Ce^* = \log (Ce/(\frac{1}{3}La + \frac{1}{3}Nd))$ , where Ce, La, and Nd refer to the shale-normalized values (ELDERFIELD and GREAVES, 1981); the Ce\* value for "average" seawater = -0.78. A linear regression of the Ce\* values for the SJFR sediments shows that they systematically decrease with increasing abundance of the distal hydrothermal factor and predicts a Ce\* value of -0.75 for a pure distal component, which is in good agreement with the "average" seawater value.

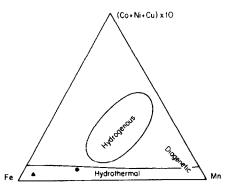


FIG. 2. Fe-Mn-(Co + Ni + Cu) ternary diagram (after BONATTI, 1981). The composition of both Factor I (proximal hydrothermal, a) and Factor 2 (distal hydrothermal, •) for KC-1 samples fall within the hydrothermal region. Cu values were estimated using known ratios of Cu/Ni and Cu/Co in hydrogenous phases (GRAYBEAL and HEATH, 1984).

## Fate of REE hydrothermal inputs

Chemical analyses of seafloor hydrothermal vent fluids (HINKLEY and TATSUMOTO, 1987; MICHARD and ALBAREDE, 1986; MICHARD et al., 1983, 1984; PHILPOTTS et al., 1987; VON DAMM and BISCHOFF, 1987) have consistently shown that these solutions are significantly enriched in REEs and Fe relative to ambient seawater. However, the vent-derived REEs are susceptible to removal from solution through the scavenging action of particulate matter, especially Fe oxyhydroxides (RUHLIN and OWEN, 1986a; OLIVAREZ and OWEN, 1987), which rapidly precipitate from the same vent fluids. The question of whether or not REEs in vent waters represent a true input to seawater thus depends upon the extent to which they can "escape" this scavenging process. In order to assess the fate of REEs derived from vent fluids, we have developed a comparison between the REE/Fe ratios which are displayed by hydrothermal fluids vs. analogous ratios determined for hydrothermal sediments (Table 2). Vent fluid data were obtained from the literature, whereas the REE/ Fe values for hydrothermal sediments were taken from the results of the end-member factor analysis (Table 1; SJFR) and from RUHLIN and OWEN (1986a, Site 598). For Site 598, the proximal end-member ratios represent the y-intercept (0 km) values derived from the linear regression analyses of the REE/Fe concentration data vs. the distance of deposition from the paleo-rise crest. This regression was performed for each of the eight REEs and is based on 34 samples which span a deposition range between 9 km and 474 km from the paleorise crest. In all cases the y-intercept value representing the proximal ratio is significant at the 0.01 level. The distal endmember values reported for Site 598 are from sediments deposited 802 km from the paleo-rise crest (RUHLIN and OWEN, 1986a, Table 2).

There are three possible outcomes to any comparison of REE/Fe ratios in vent fluids *versus* hydrothermal sediments. These outcomes, along with our interpretation of each, are summarized in the following cases:

Table 2. REE/Fe ratio (x10<sup>6</sup>) in Pacific hydrothermal phases.

	Solutions §		Sediments *				
Site:	EPR 13N 21N		SJFR	EPR Site 598		SJFR KC-1	
Dist. #				Prox.	Distal 802	Prox. ≤10	Distal ≥16
La/Fe				270	1220	700	960
Ce/Fe	24.0	13.3		33	218	760	510
Nd/Fe	14.0	3.33	2.99	180	<b>99</b> 1	860	1200
Sm/Fe	6.0	0.910	0.63	41	203	130	162
Eu/Fe	9.9	2.02		11	56	33	37
Tb/Fe				54	36	36	
Yb/Fe	0.59	0.29		22	132	76	100
Lu/Fe				5	28	22	21

- § Data sources: REE's: EPR: Michard and Albarede (1986); avg. of samples 14Ti2 & 14Ti4 (13 N), HG & OBS (21 N). SJFR: Hinkley and Tatsumoto (1987). Fe: 13 N: Michard et al. (1984); 21 N: Von Damm et al. (1985); SJFR: Philpotts et al. (1987).
- \* Site 598 data from Ruhlin and Owen (1986a, Table 2). Prox. values obtained by linear regression of REE/Fe conc. ratio vs. distance from palee-rise crest to 0 km. (n=34, above lysocline samples). SJFR data from Table 1, this text.
- # Orthogonal distance (km) from axis of paleo-rise crest.

Case 1: Vent fluid ratio = sediment ratio. This suggests there is a quantitative removal of vent-derived REEs by hydrothermal precipitates; i.e., REEs from hydrothermal sources have no net effect upon the composition of seawater.

Case 2: Vent fluid ratio > sediment ratio. This result implies that some portion of the vent-derived REEs escaped the scavenging process and thus represent a net input to seawater.

Case 3: Vent fluid ratio < sediment ratio. This outcome suggests that the scavenging capacity of hydrothermal precipitates is sufficiently large to incorporate all vent-derived REEs as well as additional REEs from the ambient seawater that mixes with the vent fluids.

All comparisons (Fig. 3a,b) of the REE/Fe ratio of vent fluids vs. hydrothermal sediments at both sampling sites are consistent with Case 3; i.e., the influx of hydrothermal fluids results in a net removal of REEs from seawater. Moreover, all but one of these comparisons exhibit the following trend for the REE/Fe ratios: vent fluids < proximal sediments < distal sediments, which suggests a continuous uptake of REEs away from the ridge axis. The distance which separates the proximal and distal hydrothermal sediments at SJFR is on the order of ~4 km, while at Site 598 this distance (802 km) is more than two orders of magnitude greater. Inasmuch as they have travelled a greater distance from the injection site, distal hydrothermal sediments at Site 598 have also been exposed to seawater longer than distal sediments at SJFR. Note that the percent increase in the REE/Fe ratio between proximal and distal components is greater at Site 598 relative to SJFR (Table 2). We believe that continuous REE scavenging over a greater distance accounts for the differences observed between these two sites.

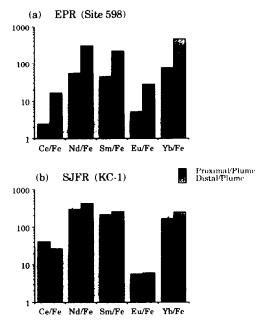


FIG. 3. Bar graph of the ratio of REE/Fe values in hydrothermal sediments from (a) Site 598, EPR at 19°S, and (b) Core KC-1, SJFR, divided by the REE/Fe value in hydrothermal plumes (Table 2). For nearly all cases the ratios are much greater than one (implying a net REE removal from seawater), and distal ratios are generally greater than proximal ratios (implying continuous REE scavenging by distal hydrothermal precipitates).

The fact that the sediment REE/Fe ratio for the proximal factor is significantly greater than the vent fluid ratio suggests that removal of REEs from the plume is both rapid and extensive. Rapid removal is evident in analyses of nearly pure vent fluids from the SJFR which indicate that both Fe and the REEs behave nonconservatively upon mixing with seawater (HINKLEY and TATSUMOTO, 1987; PHILPOTTS et al., 1987); i.e., their concentrations are markedly less than what would be expected based upon simple mixing between the hydrothermal fluid and seawater end-members. DYMOND (1981) has shown that hydrothermal precipitates, most likely transported as colloidal-sized particles, are dispersed thousands of kilometers away from the vent site. These particles apparently continue to scavenge REEs during transport, as indicated by measurements of REEs in seawater collected down-plume from active hydrothermal vent fields (KLINK-HAMMER et al., 1983). These samples are depleted in REEs relative to normal seawater at a comparable depth (Table 3).

Analyses of vent fluids at 13°N, 21°N and along the SJFR indicate Fe is enriched 10<sup>6</sup> to 10<sup>7</sup> times its average seawater concentration (MICHARD et al., 1984; VON DAMM et al., 1985; PHILPOTTS et al., 1987), while the corresponding REE enrichments in these fluids (MICHARD and ALBAREDE, 1986; HINKLEY and TATSUMOTO, 1987) are in the range of 10 to 104. Assuming these data are representative, we conclude the fate of hydrothermal REE inputs is determined by the overwhelming amount of Fe in vent fluids and that the scavenging capacity of the Fe-rich hydrothermal precipitates is sufficient to remove additional REEs from the ambient seawater that mixes with the plume. As a first approximation, our calculations assume that all hydrothermal Fe inputs will precipitate as oxides/hydroxides upon mixing with seawater and thus be in a form capable of scavenging REEs. We recognize that this assumption is not always valid; however, MOTTL and McConachy (1989) have studied the evolution of hydrothermal solutions near 21°N and concluded that  $50 \pm 30\%$ of the Fe will precipitate as oxides even in the presence of excess H<sub>2</sub>S. Taking these results into account, our estimates of the REE/Fe ratios for hydrothermal plumes may be too low by a factor of 1.25 to 5.0. This consideration would have no effect on the sense of our results in the average case (only 50% of the hydrothermal Fe precipitates as oxides): hydrothermal Fe-oxide precipitates would still produce a net removal of REEs from seawater. In the extreme case (only 20% of the hydrothermal Fe precipitates as oxides) some ventderived Ce and Eu would initially be added to seawater, but these additions would either be roughly balanced (for Eu) or more than offset (for Ce) by the scavenging action of distal hydrothermal precipitates. Furthermore, in each of these cases it is likely that REE removal from seawater would be augmented by scavenging by other solid phases, such as Mn oxyhydroxides (e.g. KLINKHAMMER et al., 1983).

## SUMMARY AND CONCLUSIONS

The role that seafloor hydrothermal activity plays in determining the REE composition of seawater depends upon the interplay between two competing processes. Seawater-basalt reactions during hydrothermal circulation produce significant enrichments, relative to normal seawater concentrations, of both Fe and REEs in hydrothermal vent fluids.

Table 3. Actual and normalized REE concentrations (ng/kg) from seawater near hydrothermal vents.\*

	NW Pacific (2500 m)		Normalized Concentration #
La	6.5	4.2	0.61
Ce	1.3	0.49	2.2
Nd	4.3	2.3	0.53
Sm	0.80	0.41	0.43
Eu	0.21	0.12	0.47
Gd	1.3	0.79	0.52
Yb	1.4	1.3	0.46

- \* From Klinkhammer et al. (1983); NW Pacific = Mariana Trough; SE Pacific = EPR at 19S.
- # Average value for both sites divided by seawater concentration (de Baar et al., 1985) at 2750 meters (after Owen and Olivarez, 1988).

Upon discharge into oxidized bottom waters, however, the dissolved Fe is rapidly precipitated as iron oxyhydroxides which, in turn, can serve as highly efficient scavengers of REEs. Thus, for seafloor hydrothermal activity to represent a source of REEs to seawater requires that some portion of the hydrothermal REE inputs escape this scavenging process. Our comparison of the REE/Fe ratios measured in hydrothermal vent fluids vs. those determined for proximal and distal hydrothermal sediments at two Pacific sites indicates the scavenging capacity of hydrothermal precipitates is not only sufficient to incorporate all vent-derived REEs but also causes significant REE removal from the ambient seawater that mixes with the vent fluids.

This finding may in part explain certain puzzling observations concerning the marine geochemistry of the REEs. For example, PIPER (1974) identified an imbalance in the REE budget of Pacific pelagic sediments in which REE removal rates were several times greater than their river input rates. Although active seafloor venting had not yet been discovered, he suggested that the input and subsequent removal of REEs from hydrothermal sources might account for this discrepancy. Subsequent documentation of REE enrichments in vent fluids, along with evidence presented here of both a rapid removal of vent-derived REE inputs by hydrothermal precipitates and of their augmentation with REEs from surrounding seawater, is consistent with this explanation. A more recent problem is that, contrary to expectations based upon documented REE enrichments in vent fluids, REE concentrations in seawter collected near hydrothermal vents (KLINKHAMMER et al., 1983) are depleted relative to normal seawater at a comparable depth (OWEN and OLIVAREZ, 1988). Our results suggest that this anomaly is the result of intense REE scavenging by hydrothermal precipitates derived from the nearby vents.

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