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RELATIONSHIP BETWEEN CYTOTOXICITY AND CONVERSION OF THIOSANGIVAMYCIN ANALOGS TO TOYOCAMYCIN ANALOGS IN CELL CULTURE MEDIUM

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Abstract—Non-nucleoside analogs of the pyrrolopyrimidine nucleosides toyocamycin, sangivamycin and thiosangivamycin have been synthesized and their cytotoxicity in mammalian cells determined. While studying the effects of 5-thioamide-substituted analogs on cell growth, we observed an interesting phenomenon in which cells recovered spontaneously from growth inhibition during extended incubations. HPLC studies demonstrated that the 5-thioamide moiety of several structurally dissimilar 7-substituted 4-aminopyrrolo[2,3-*d*]pyrimidines, including thiosangivamycin, is unstable in cell culture medium and is converted to the corresponding 5-nitrile with a half-life of approximately 48 hr. In contrast, different substituents at the 4-position of the heterocycle significantly affected the stability of the 5-thioamide moiety. Conversion of the thioamide to the nitrile was caused by components in the cell culture medium, not components of serum. The above observations demonstrate that caution should be exercised in interpreting biological data obtained *in vitro* for 5-thioamide pyrrolo[2,3-*d*]pyrimidines.

Key words: pyrrolopyrimidine; non-nucleoside; cytotoxicity; stability; HPLC

The pyrrolo[2,3-*d*]pyrimidine nucleoside antibiotics toyocamycin (**1a**, Table 1) and sangivamycin (**1b**) originally were isolated from different strains of streptomyces in 1956 and 1963, respectively [1, 2]. The total synthesis of these compounds was reported in 1969 by Townsend and coworkers [3]. A synthetic analog, thiosangivamycin (**1c**), was described in 1970 by this same research group [4]. All three compounds (**1a-c**) are highly toxic to mammalian cells and have been evaluated as potential anti-cancer agents [5, 6]. Sangivamycin was advanced to phase I clinical trials, and a report on its toxicity has been published [7].

As part of our ongoing research involving pyrrolo[2,3-*d*]pyrimidines as potential antiviral agents [8-14], we recently described the synthesis and activity against HCMV† of a number of non-nucleoside derivatives related to toyocamycin, sangivamycin and thiosangivamycin (**1a-c**) [15]. We have expanded our studies in this area and now have synthesized a number of closely related compounds (**2-6**). While studying the cytotoxicity profile of these new analogs, we observed an unusual phenomenon in which cells spontaneously recovered from growth inhibition during extended incubation times. An

investigation and an explanation of this phenomenon follows.

MATERIALS AND METHODS

Materials. Toyocamycin, sangivamycin and thiosangivamycin were obtained from the library of compounds in the laboratory of one of us (L.B.T.). Compounds **2** and **3** were synthesized in four or five steps from the known 2-amino-5-bromo-3,4-dicyanopyrrole [16] by a modification of the procedure described in Ref. 15. Compounds **4-6** were synthesized from the key intermediate 4-chloro-7 - [(2 - methoxyethoxy)methyl]pyrrolo[2,3 - *d*] - pyrimidine-5-carbonitrile generated by a modification of the procedure described previously [17]. No impurities were detected in the ¹H NMR of all compounds examined and C-H-N analysis of these compounds was within the acceptable ±0.4% of theoretical values. Details of the synthetic procedures, physical characteristics, and antiviral activity of compounds **2-6** are being published elsewhere.

Cell cultures. KB cells, an established human cell line derived from an epidermal oral carcinoma, were grown in MEM supplemented with 10% calf serum and were subcultured by conventional methods by using 0.05% trypsin plus 0.02% EDTA [8]. The average PDT for KB cells in the absence of drugs was 21 hr. L1210 murine leukemic cells were grown in Fischer's medium supplemented with 10% heat-

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† Abbreviations: HCMV, human cytomegalovirus; MEM, minimal essential medium; PDT, population doubling time; and HBS, HEPES-buffered salt solution.

Table 1. Structure and cytotoxicity of 4-aminopyrrolo[2,3-*d*]pyrimidine analogs

Compound	R ₁	R ₂	Cytotoxicity, IC ₅₀ (μM)	
			KB cells	L1210 cells
1a	CN	β-D-Ribofuranose	0.03	0.004
b	CONH ₂	β-D-Ribofuranose	0.06	0.003
c	CSNH ₂	β-D-Ribofuranose	0.08	0.023
2a	CN	CH ₂ C ₆ H ₅	>100	>100
b	CONH ₂	CH ₂ C ₆ H ₅	84	16
c	CSNH ₂	CH ₂ C ₆ H ₅	18	70
3a	CN	CH ₂ O(CH ₂) ₂ OCH ₃	>100	>100
b	CONH ₂	CH ₂ O(CH ₂) ₂ OCH ₃	>100	>100
c	CSNH ₂	CH ₂ O(CH ₂) ₂ OCH ₃	>100	>100

Cytotoxicity was determined by measuring the effects of compounds on the growth of KB and L1210 cells. Data are the averages of two or more experiments. The greater than symbol indicates that the IC₅₀ was not reached at the highest concentration tested. Compounds **1a-c** are toyocamycin, sangivamycin and thiosangivamycin, respectively.

inactivated (56°, 30 min) horse serum and were subcultured by serial dilution. All cell lines were screened periodically for Mycoplasma contamination and were negative.

Cytotoxicity studies. The cytotoxicity of compounds **1a-c** to **3a-c** was determined initially in KB cells by a staining method previously described [18] and in an L1210 cell growth assay [19]. In more detailed studies, the inhibitory effect of compound **2c** was evaluated and PDTs were determined in a KB cell growth assay [20]. PDTs are significantly longer in cells whose growth is inhibited, or whose absolute numbers decrease as a result of toxicity affecting their growth. PDTs were calculated by means of a least squares program fitting the exponential portion of a growth curve. For growth studies with KB and L1210 cells, growth rates were calculated from determinations of the number of cells at 0, 24, 48, 72 and 96 hr in the presence of selected concentrations of the test compound. Cells were enumerated using a Coulter Counter (Coulter Electronics, Hialeah, FL). Growth inhibition was calculated as the slope of a semi-logarithmic plot of cell number against time for the treated culture as a percent of the control. In both the staining and cell growth studies, the IC₅₀ was defined as the concentration required to decrease the growth rate to 50% of the control rate.

Compound stability assays. The stability of compounds in cell culture medium was investigated by incubating 100 μM concentrations of compounds **1-6** at 37° in tightly capped tubes. The compounds were incubated in either Fischer's medium, MEM, or HBS [21] (10 mM HEPES, 137 mM NaCl, 5.4 mM KCl, 0.7 mM Na₂HPO₄, 5.6 mM dextrose, pH 6.5). In all cases, the incubation media contained neither

Table 2. Retention times and UV absorbance data of pyrrolo[2,3-*d*]pyrimidine analogs

Compound	RT (min)	λ _{max} (nm)
1a	12	274
b	10.5	278
c	11	292
2a	45	277
b	46	278
c	50	294
3a	22	273
b	25	275
c	28	292
4a	27	281
b	33	286
5a	33	290
b	26	289
6a	28	271
b	24	266

Retention times (RT) were determined by reverse-phase HPLC as described in the text. The λ_{max} (nm) of each compound was determined by UV spectroscopy.

cells nor serum. The pH of the medium was monitored and was found to be consistently between 7.5 and 7.8. At selected times during periods extending up to approximately 3 weeks, 100-μL aliquots of medium or buffer were removed and immediately analyzed by reverse-phase HPLC.

Analytical procedures. HPLC of compounds **1-6** was performed with a Beckman Ultrasphere ODS reverse-phase column (5 μm, 4.6 mm × 15 cm). HPLC analysis of **1a-c** was adapted from a literature

procedure [22]. Compounds **1a–c** were eluted in 30 min using a linear gradient starting with 0.02 M KH_2PO_4 (pH 5.6): 60% MeOH/40% H_2O solution (87:13, v/v) and ending with 100% of the 60% MeOH/40% H_2O solution. Elution of compounds **2–6** was accomplished in 70 min using a linear gradient starting with 0.1% trifluoroacetic acid in H_2O and ending with 50% MeOH. UV absorbance of the compounds was monitored simultaneously at the λ_{max} for the thioamide and nitrile substituted derivatives (Table 2). Retention times (RT) for compounds **1–6** are shown in Table 2. Peak areas were determined using System Gold Software (Beckman Instruments, Fullerton, CA). The purity of each compound was determined by HPLC before the start of each experiment.

RESULTS

7-Benzyl-substituted 4-aminopyrrolo[2,3-*d*]pyrimidine derivatives (**2a–c**, Table 1), related to toyocamycin (**1a**), sangivamycin (**1b**) and thiosangivamycin (**1c**), were synthesized and examined for activity against HCMV as potential antiviral agents. Results, described elsewhere [23], showed that similar to our previous report on 7-alkyl 4-aminopyrrolo[2,3-*d*]pyrimidines [15], only the thioamide-substituted derivative (**2c**) exhibited activity against HCMV [23]. We have expanded the cytotoxicity portion of those studies and now report that the thioamide (**2c**) was more toxic than the carboxamide (**2b**) in KB cells, but in L1210 cells **2b** was more toxic than **2c** (Table 1). In both cell lines, the nitrile derivative (**2a**) was the least toxic. Nonetheless, compounds **2a–c** were substantially less toxic in both KB and L1210 cells when compared with the parent nucleoside analogs **1a–c**. Compounds **3a–c** containing a 2-methoxyethoxy methyl side chain at N-7, were the least toxic compounds of those studied.

More detailed studies on the effect of compound **2c** on the growth of uninfected KB cells demonstrated that it suppressed growth at 10 and 32 μM (Fig. 1). Inhibition observed at 32 μM was reversible upon removal of the drug following a 24-hr incubation period (Fig. 1), suggesting that it acts as a cytostatic rather than a cytotoxic agent. Surprisingly, growth resumed even in cultures from which drug was not removed (Fig. 1). This observation was supported by data from a separate experiment in KB cells where PDTs were calculated at 24-hr intervals up to 72 hr after incubation of 32 μM **2c**. Over this 72-hr period, the PDT increased from 21 hr in the control to 56 hr in drug-treated cultures during the first 24-hr incubation with **2c**. In contrast, the PDT of the drug-treated cultures returned to control values (20 hr compared with 21 hr in the control) during the last 24-hr interval. Similar results were obtained in L1210 cells. Together these results suggested that **2c** was being converted to a less toxic compound *in vitro*. Various possibilities were considered for the basis for this recovery: (i) inactivation of the compound by cellular enzymes, (ii) inactivation by enzymes in the serum, or (iii) degradation in cell culture medium. Cell growth recovery could not be totally attributed to cellular or serum proteins

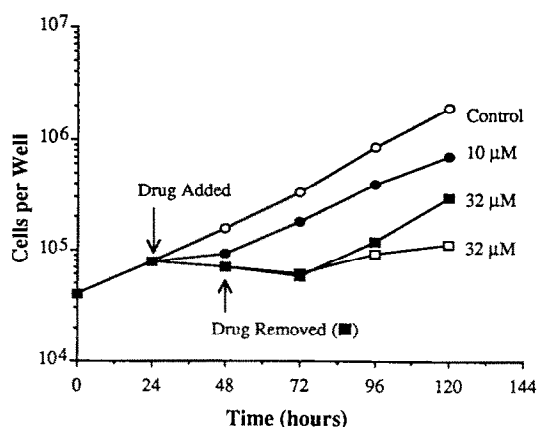


Fig. 1. Effect of compound **2c** on the growth of KB cells. Cells were seeded at 40,000 cells/well and incubated for 24 hr at which time 10 or 32 μM **2c** was added to three sets of cultures (●, □, ■) and incubated for an additional 24 hr. At that time, the medium was removed from one set of cultures containing 32 μM drug (■), rinsed with HBS, and the incubation continued with drug-free medium. Cells were harvested at the time periods indicated and enumerated with a Coulter Counter.

inactivating the compound since preincubation of 100 μM compound **2c** in medium without serum for 2 days before the addition of cells eliminated inhibition of L1210 cell growth. In this experiment, L1210 cell growth in drug-treated cultures was comparable to the control values even though the concentration of **2c** used was approximately 1.5 times the IC_{50} . Therefore, we concluded that compound **2c** was unstable in cell culture medium alone (without serum) and was spontaneously converting to a relatively non-toxic compound.

To investigate this phenomenon, 100 μM compound **2c** was incubated in serum-free Fischer's cell culture medium. HPLC analysis was performed on samples of cell culture medium removed at 24-hr intervals. Within the first 48 hr of incubation, more than half of **2c** was converted to a new product (Fig. 2). By 144 hr, nearly all of **2c** was converted to this new product. HPLC analysis using both **2a** and **2b** as markers showed that this new peak co-eluted with the corresponding nitrile (**2a**). From this it was calculated that the thioamide (**2c**) was converted to **2a** with a $T_{1/2}$ of 41 hr. Upon close examination of the HPLC chromatograms, a small amount of the nitrile, **2a**, and a peak which co-eluted with the carboxamide (**2b**) were observed at $t = 0$. These trace amounts most likely were impurities in **2c** that were not detected by either ^1H NMR or C-H-N analysis used to characterize **2c**. It should be noted that the peak **2b** did not change over the 144-hr incubation period (Fig. 2), indicating that the carboxamide was not an intermediate in the conversion of **2c** to **2a**. Further HPLC analysis demonstrated that incubation of the nitrile (**2a**) and the carboxamide (**2b**) for 4 days under the same conditions as those used for **2c** resulted in no change from the $t = 0$ elution pattern. This indicated that

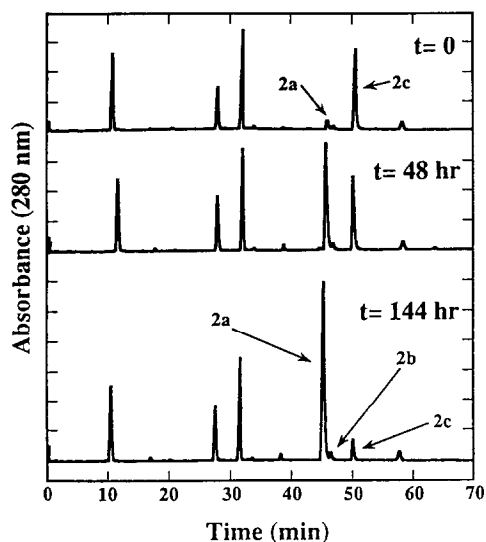


Fig. 2. Chromatograms of Fischer's medium without serum incubated for 0, 48 and 144 hr at 37° with 100 μ M **2c**. Absorbance of the effluent was monitored simultaneously at 280 and 294 nm; absorbance at 280 nm is presented. Peaks other than those labeled **2a**, **2b** or **2c** are from components of Fischer's medium. These peaks were unchanged during the experiment and were observed in a sample not incubated with **2c**. Key: (**2a**) 4-amino-7-(benzyl)pyrrolo[2,3-*d*]pyrimidine-5-carbonitrile; and (**2b** and **2c**) the corresponding 5-carboxamide and 5-thio-carboxamide, respectively.

these two compounds were stable under the culture conditions. These results further demonstrate that the carboxamide cannot be an intermediate in the conversion of the thioamide to the nitrile. Additional studies in which **2c** was incubated in medium containing serum and with cultured KB cells gave results similar to those obtained for **2c** in cell and serum-free medium.

These results prompted us to examine whether the conversion in cell culture medium of a thioamide moiety to a nitrile was a general phenomenon with 7-substituted 4-aminopyrrolo[2,3-*d*]pyrimidine-5-thioamides. To test this question, we investigated the stability of a related series of compounds, **3a-c** (Table 1). We chose this series for two reasons: (i) the ether substituent at N-7 of **3a-c** was structurally different than the benzyl group of **2a-c** and (ii) in contrast to **2a-c**, **3a-c** were non-toxic at 100 μ M in both KB and L1210 cells (Table 1). Compound **3c** was incubated for 5 days under the conditions described previously. HPLC analysis demonstrated that **3c** was converted to a single product that co-eluted with the nitrile, **3a**, with a $T_{1/2}$ of 50 hr. In contrast to the studies with **2c**, a peak that co-eluted with the carboxamide, **3b**, was not detected from $t = 0$ through the 5-day incubation period. As with compound **2a**, compound **3a** was stable under the incubation conditions as monitored by HPLC.

Based upon these findings, the stability of the parent nucleosides toyocamycin (**1a**), sangivamycin

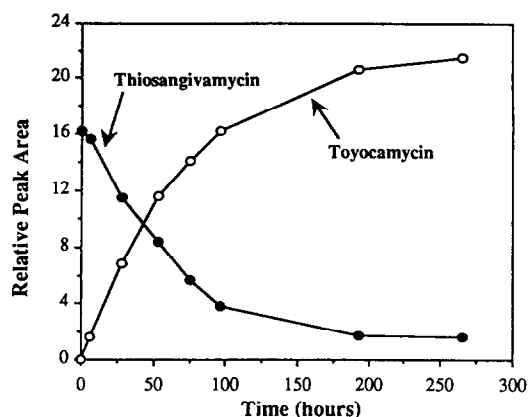


Fig. 3. Stability of the pyrrolopyrimidine nucleoside thiosangivamycin in cell culture medium. Thiosangivamycin was incubated in Fischer's medium without serum for 11 days. Aliquots were removed at the times indicated and analyzed by HPLC. The peak areas presented in the figure are those determined at the λ_{\max} for each nucleoside (274 nm for toyocamycin and 292 nm for thiosangivamycin). The peak area at later time points for toyocamycin is greater than the peak area for thiosangivamycin at $t = 0$ because the molar absorptivity (ϵ) for toyocamycin at 274 nm is greater than the ϵ for thiosangivamycin at 292 nm [3, 4].

(**1b**) and thiosangivamycin (**1c**) was also studied. Like studies with **2c** and **3c**, the stability of thiosangivamycin was examined by incubating it in Fischer's medium without serum for 11 days and analyzing samples by HPLC at selected times during this period. The results showed (Fig. 3) that like **2c** and **3c**, thiosangivamycin was converted to a product that co-eluted with toyocamycin (**1a**) with a $T_{1/2} = 55$ hr. Similar results were obtained by incubating the compound in MEM without serum ($T_{1/2} = 41$ hr). Like the studies with **3c**, a peak that co-eluted with sangivamycin (**1b**) was not detected from $t = 0$ through the 11-day incubation period. As with **2a-b** and **3a**, toyocamycin and sangivamycin were stable under the incubation conditions.

To examine the possible role that components of cell culture medium had on the conversion of the thioamide to the nitrile, a number of studies using buffered solutions or modified cell culture medium were undertaken. The results showed that contrary to those obtained in Fischer's medium and MEM, incubation of **1c** in HBS resulted in only a very slow conversion to the nitrile ($T_{1/2} > 504$ hr). This conversion of thiosangivamycin (**1c**) to the corresponding nitrile in HBS was >10 times slower than in cell culture medium, strongly suggesting that components contained in culture medium but not in HBS catalyzed the conversion. To investigate the role in the conversion of certain components in the medium, the pH of the HBS solution containing **1c** was increased from 6.5 to 7.6, but this did not affect the stability of the compound. Additionally, the benzyl derivative **2c** was incubated with MEM, which did not contain phenol red and glutamine. In this

Table 3. Structure, cytotoxicity and stability of several 4-substituted pyrrolo[2,3-*d*]pyrimidine analogs of compound **3c**

Compound	R ₁	R ₂	Cytotoxicity, IC ₅₀ (μM)		Half-life (hr)
			KB	L1210	
4a	HNMe	CN	>100	>100	NA*
b	HNMe	CSNH ₂	>100	>100	20
5a	N(Me) ₂	CN	>100	>100	NA
b	N(Me) ₂	CSNH ₂	>100	>100	150
6a	H	CN	>100	>100	NA
b	H	CSNH ₂	>100	>100	175

Cytotoxicity was determined by measuring the effects of compounds on the growth of KB and L1210 cells. Data are the averages of two or more experiments. The greater than symbol indicates that the IC₅₀ was not reached at the highest concentration tested.

* Not applicable.

medium, the $T_{1/2}$ was extended from 41 hr to approximately 65 hr demonstrating that although these two components affected the stability of the thioamide moiety, they could not completely account for the conversion.

To investigate whether changes on the heterocycle at sites other than N-7 could affect the stability of the thioamide group, we examined the effect of modifications at the 4-position of **3c** on the half-life. In this study, the 4-amino group of compound **3** was replaced with a proton (H), a methylamino group (NHCH₃) or a dimethylamino group (N(CH₃)₂)-compounds **4-6** (Table 3). Like compounds **3a-c**, compounds **4-6** were essentially non-toxic in both KB and L1210 cells with IC₅₀ values >100 μM (Table 3). Like compounds **1-3**, compounds **4-6** were incubated in Fischer's medium, and their stability was monitored by HPLC. The results demonstrated that modifications at the 4-position have a definite effect on the stability of the 5-thioamide moiety, whereas changes at N-7 tend not to affect the stability of the 5-thioamide moiety (Table 3). For example, replacement of the 4-amino group of **3c** with a methylamino group (**4b**) decreased the stability of the thioamide ($T_{1/2}$ = 20 hr). In contrast, replacement with a dimethylamino group (**5b**) or a hydrogen (**6b**) extended the $T_{1/2}$ from 50 hr for **3c** to 150 hr for **5b** and 175 hr for **6b**. In the case of **5b**, we also observed an unidentified peak by HPLC in addition to **5a** and **5b**. This possible intermediate appeared 24 hr into the incubation and rapidly decreased thereafter. This was the only compound during this entire study where we observed an HPLC peak other than the thioamide or nitrile analogs.

DISCUSSION

The present study has established that the

pyrrolopyrimidine nucleoside thiosangivamycin and several related analogs are converted to the corresponding nitrile analogs in cell culture medium. The conversion of thiosangivamycin to toyocamycin in cell culture medium has not, to the best of our knowledge, been observed previously. Even if the conversion had occurred in previous studies, there would have been no obvious reason to suspect this since thiosangivamycin and toyocamycin are equally toxic to mammalian cells. The toxicity of thiosangivamycin and toyocamycin arises because both compounds are phosphorylated by adenosine kinase [24] to afford the 5'-monophosphate derivatives that ultimately are incorporated into DNA or RNA [5, 6]. Whether the 5'-monophosphate of thiosangivamycin is converted to phosphorylated toyocamycin is unknown and was not included in the present study. However, our results, *vide supra*, suggest that this would occur since a variation of the substituent at N-7 does not seem to have a major effect on the conversion.

It is interesting that changes at the 4-position can have a major effect on the stability of the 5-thiocarboxamide moiety. This would suggest that the 4-position may play a key role in stabilizing or destabilizing the 5-thioamide group. Results obtained for the 4-substituted amino derivatives (**4b** and **5b**) revealed that increasing the basicity of the amino group at C-4 does not necessarily correlate with an increase or decrease in the stability of the thioamide. Instead, the stability of a substituted thioamide compound seems directly related to the presence or absence of an amino group with a free proton at C-4. Whether modifications at other sites on the heterocycle could affect the stability of the thioamide group is unknown.

The rapid conversion of **2c** to its corresponding

nitrile (**2a**) in cell culture medium raises an interesting issue regarding the antiviral activity. We have demonstrated [23] that **2a** is inactive against HCMV at concentrations up to 100 μM , whereas **2c** has potent activity against HCMV with an $\text{IC}_{50} = 0.4 \mu\text{M}$. Our present results suggest that the action of **2c** against HCMV must occur rather rapidly since by 48 hr essentially 50% of the active compound is converted to an inactive one. We provided evidence for this hypothesis while studying the temporal effect of **2c** on DNA and RNA synthesis [23]. We showed that in uninfected cells, various concentrations of **2c** dramatically affect both DNA and RNA syntheses approximately 1–2 hr following its addition to cultured CEM-SS cells. These results, however, were obtained in uninfected cells and whether the potent early inhibition of RNA and DNA syntheses by **2c** in uninfected cells correlates with the potent activity against HCMV has yet to be established and is currently under investigation.

Beside the nitrile derivatives, the identity of any other product involved in the conversion is unknown. Since **1c**, **2c**, **3c** and **4b** exhibit similar stability patterns but completely different cytotoxicity profiles, it is highly unlikely that the release of H_2S as a by-product of the conversion caused cell toxicity. Regarding the causes of this phenomenon, the identification of a single component has proved to be difficult. Our results suggest that there may be a combination of components, such as phenol red and glutamine, that are involved in catalyzing the observed conversion. Chemically, thioamides can be converted to nitriles by treatment with mercuric chloride and methylamine [25, 26]. In the current study, these reagents were not present during the incubation of any of the thioamide-substituted derivatives. There have been reports, however, of more mild reaction conditions furnishing a nitrile from a thioamide [4, 27]. Specifically, alkylation of thiosangivamycin (**1c**) with methyl iodide afforded a 5-methylthioformimidate intermediate that rapidly eliminated methanethiol to give toyocamycin (**1a**). Similar results were obtained upon treatment of 5-amino-4-thiocarboxamide-1- β -D-ribofuranosylimidazole (thio-AICA-riboside) with methyl iodide. From this information it may not be entirely unreasonable to propose the formation of an alkylated thio-intermediate that can rapidly eliminate to give the nitrile. The possible appearance of an intermediate in the conversion of 4-dimethylamino-7-[(2-methoxyethoxy)methyl]pyrrolo[2,3-*d*]pyrimidine-5-thiocarboxamide (**5b**) to the corresponding nitrile (**5a**) may support this argument; however, the exact mechanism of how the thioamide is converted to the nitrile in cell culture medium is unknown.

In summary, our results have demonstrated that the 5-thioamide moiety of many 7-substituted 4-aminopyrrolo[2,3-*d*]pyrimidines is unstable in cell culture medium, and is converted to the corresponding 5-nitrile regardless of the substituent at N-7. In contrast, substituents at the 4-position of the heterocycle significantly affect the stability of the 5-thioamide moiety. Investigators using these or structurally similar compounds should be aware of these results and appreciate that biological data

generated in cell culture may be due to a mixture of the thioamide and by-product(s) of the conversion.

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