

**Efforts toward a stereocontrolled synthesis of
CYP3A4 inhibitor GF-I-1 and stereoisomers**

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Abstract

Several furocoumarin compounds that show significant inhibition of the main human metabolic enzyme CYP3A4 have been isolated from grapefruit juice. The most potent of these is GF-I-1, an asymmetrical homodimer whose stereochemistry remains unknown. We sought to synthesize GF-I-1 and its three stereoisomers in a stereocontrolled fashion to unambiguously determine the absolute stereochemistry of natural GF-I-1. We were also interested in comparing the biological activities of the four synthetic isomers, following work that showed that CYP3A4 is not sensitive to the absolute configuration of a simpler grapefruit furocoumarin.

We have achieved the synthesis of a key dimeric intermediate in racemic form on the route towards GF-I-1. In doing so, we assessed the efficiencies of BiCl_3 and $\text{Cu}^{\text{II}}(\text{BF}_4)_2$, two newly-discovered Lewis acid catalysts, for promoting the condensation of precious alcohols and epoxides. $\text{Cu}^{\text{II}}(\text{BF}_4)_2$ appears to be a promising catalyst for these transformations, for which well-established synthetic protocols have been notably absent.

We also progressed along the stereocontrolled route towards GF-I-1 and its stereoisomers, with the stereospecific synthesis of all required chiral

precursors in greater than 96% enantiomeric excess. While the total synthesis was not completed due to a transformation that could not be efficiently accomplished during the allotted time for the project, we have laid the groundwork for the rapid stereocontrolled synthesis of GF-I-1 in the future.

Chapter 1

Introduction and Background

In 1989, a drug interaction between grapefruit (*Citrus paradisi*) and felodipine, a dihydropyridine calcium channel antagonist, was serendipitously discovered by Bailey et al. They were investigating the interaction of felodipine and ethanol, using grapefruit juice as the delivery vehicle.¹ When felodipine was ingested with grapefruit juice, the plasma concentrations of the drug were more than five-fold greater than when taken with water. It was known that the bioavailability of felodipine typically averages only 15% due to exhaustive first-pass metabolism.² This, coupled with the observation that grapefruit juice has no effect on the elimination of plasma felodipine,³ suggested that regulation of the first-pass enzyme by compounds in the grapefruit juice may be responsible for the observed drug interaction. A search for that first-pass enzyme yielded cytochrome P450 3A4.⁴

Cytochrome P450 enzymes are a class of heme-containing proteins that generally catalyze the oxidation of their substrates. Of the many cytochrome P450 isoforms present, cytochrome P450 3A4 (CYP3A4) is the most abundant in the liver and epithelial tissue of the small intestine.⁵ It has been implicated in metabolizing approximately 50% of the clinical drugs on the market as well as a broad range of xenobiotics.⁶ For example, it is known

that CYP3A4 catalyzes the conversion of polycyclic aromatic hydrocarbons (PAHs) and heterocyclic amines (HAs) to reactive electrophiles. These metabolites from PAHs, often referred to as diol epoxides, react with DNA, and thus exert carcinogenic and mutagenic effects.⁷

Once the target metabolic enzyme CYP3A4 was identified, work ensued in identifying the compounds present in grapefruit that are responsible for altering its biological activity. A series of furocoumarins and furocoumarin dimers were isolated and their structures characterized from fractions of organic extracts of grapefruit juice that showed inhibition of CYP3A4.⁸ Of these, the most potent inhibitor was determined to be GF-I-1 (recently renamed: Paradisin A), a furocoumarin dimer that is a mechanism-based competitive inhibitor of CYP3A4, with an IC₅₀ on the order of 10 nM.⁹

It should be mentioned that neither do the grapefruit furocoumarins inhibit CYP3A4 exclusively nor are they the only known inhibitors of CYP3A4. The furocoumarins present in grapefruit have been shown to inhibit several other P450 isoforms.⁹ They recently have also been observed to interact with P-glycoprotein, a membrane-localized drug transporter, thus inhibiting the cellular efflux of its targets.¹⁰ CYP3A4 has long been known to be inhibited by Ketoconazole,¹¹ an anti-fungal drug, and recently a new class of bisalkaloids from white pepper that also possess moderate CYP3A4 inhibitory activity has been isolated.¹²

While the structures of the furocoumarins present in grapefruit have been identified, the stereochemical properties of these chiral molecules have not been unambiguously elucidated. Previously, through a total stereocontrolled synthesis, our laboratory has unambiguously determined the absolute stereochemistry of natural 6,7-dihydroxybergamottin (DHB), another CYP3A4 inhibitor from grapefruit that is somewhat less potent but more synthetically accessible than GF-I-1 (**Figure 1**).¹³ Furthermore, the CYP3A4 inhibition activity of the two stereoisomers of DHB were compared against each other as well as racemic DHB, revealing that all three activities are nearly equivalent. This suggested that the portion of DHB containing the stereocenter does not interact with the CYP3A4 active site.

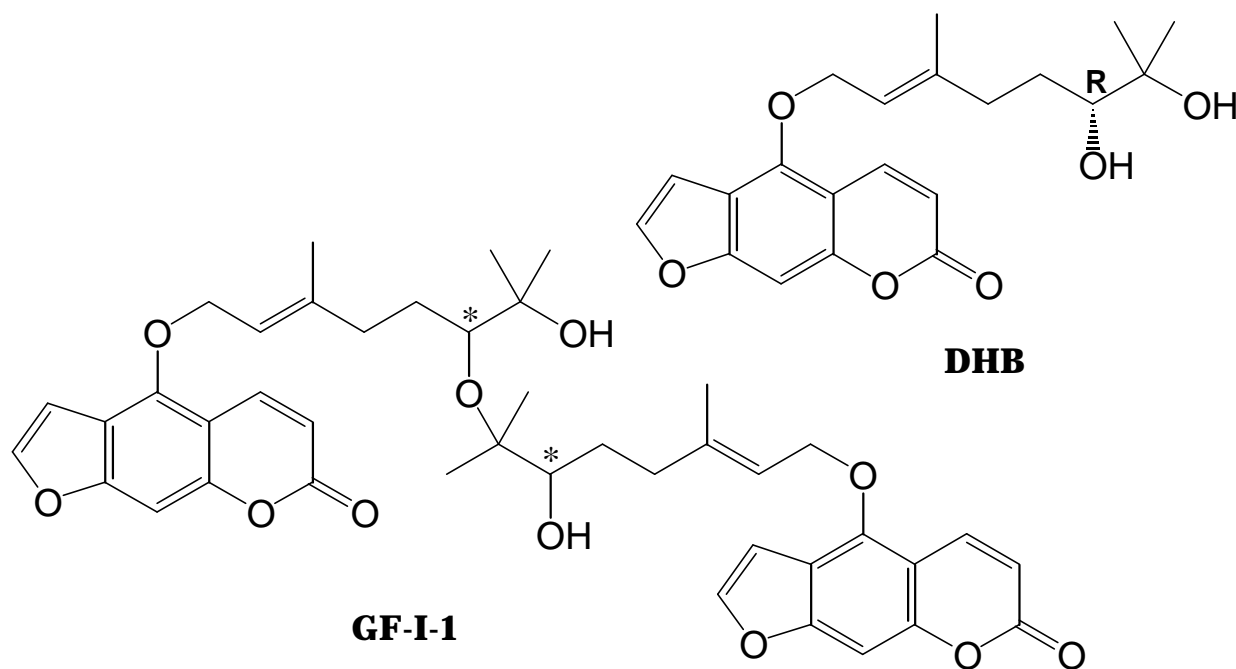


Figure 1. Synthetic target.

We now turned our attention towards the more potent GF-I-1 furocoumarin dimer. The absolute stereochemistry of this compound has been proposed to be (6R, 6'R) through ¹H NMR studies of the corresponding Mosher esters,¹⁴ however this method has been known to give erroneous results.^{15,16} We decided to unambiguously determine the absolute stereochemistry of GF-I-1 using a synthetic approach. We were also interested in comparing the inhibition activities of the four GF-I-1 diastereomers. In this dissertation, we describe our efforts toward the stereocontrolled total synthesis of GF-I-1 and its stereoisomers.

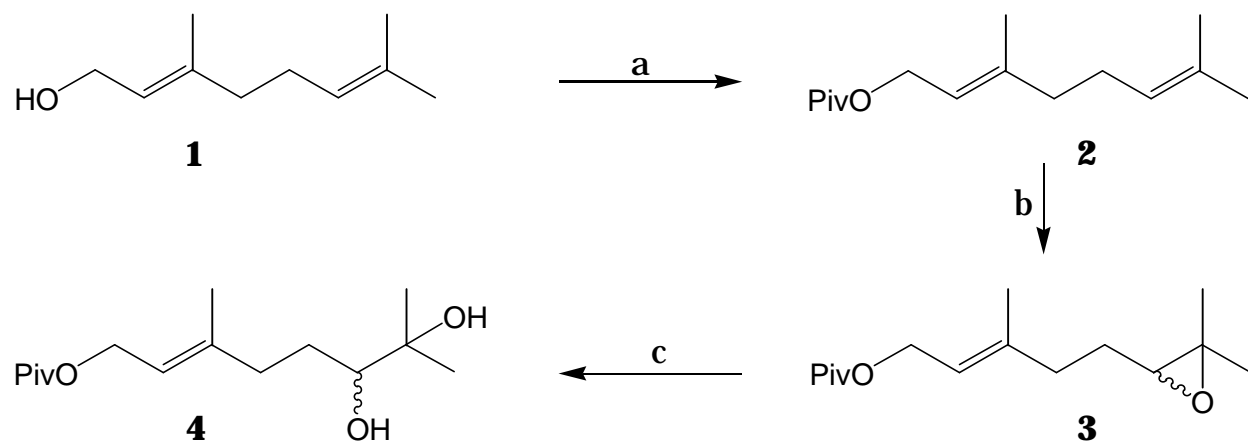
Chapter 2

Synthesis of Racemic Model System

The greatest challenge that we anticipated in the synthesis of GF-I-1 was the coupling of the two geranyl chains to form the ether-linked dimer. In fact, prior to the very recent work by Barluenga et al.,¹⁷ the literature was void of any systematic protocol for opening precious epoxides with precious alcohols. Cognizant of this difficulty, we synthesized a racemic model system for conducting epoxide-opening trials.

Starting with the commercially available geraniol (**1**), the racemic epoxide (**3**) and diol (**4**) were readily prepared (**Scheme 1**). The primary alcohol of geraniol (**1**) was protected with the pivaloate ester to prevent competition with the secondary alcohol of the diol (**4**) in opening the epoxide (**3**). The geranyl pivaloate (**2**) was oxidized with *m*CPBA to form the epoxide (**3**), which was hydrolyzed with strong aqueous acid to make the diol (**4**).

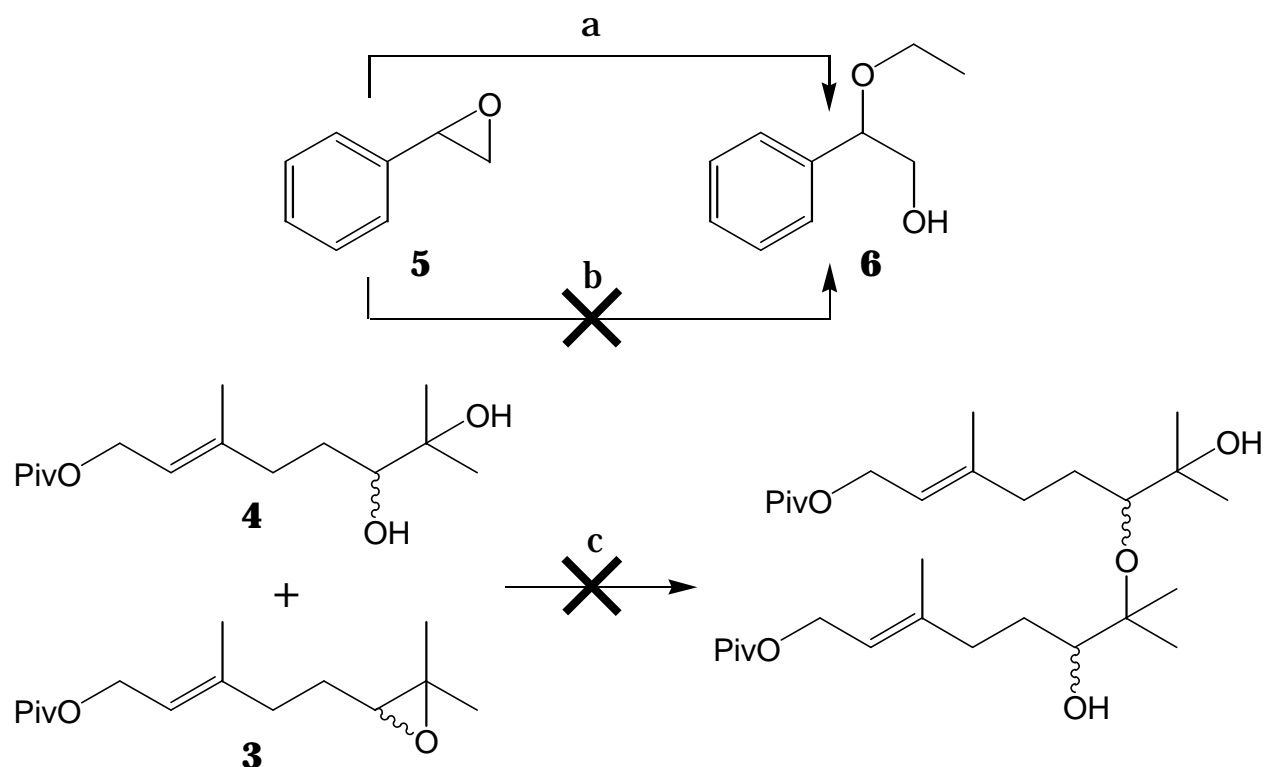
With a generous supply of racemic epoxide (**3**) and diol (**4**) in hand, we began a series of experiments to identify a suitable Lewis acid catalyst for their complexation to form the ether-linked dimer. We first tried BiCl₃, a Lewis acid catalyst which had been reported to promote efficient epoxide opening under mild conditions (**Scheme 2**).¹⁸ Although high yields of cleaved

Scheme 1^a

^a Reagents and Conditions: (a) PivCl, py, CH₂Cl₂, 0 °C, 3 h, 88%; (b) *m*CPBA, CH₂Cl₂, 0 °C, 2 h, 65%; (c) H₂SO₄, THF, rt, 45 m, 62%.

epoxide had been reported using BiCl₃, when we tried the reaction between styrene oxide (**5**) and isopropanol, we observed only partial formation of the resulting α -alkoxy alcohol (**6**). Furthermore, the trial was conducted using isopropanol as both a reagent and the solvent simultaneously, a luxury we could not afford with the precious diol (**4**). When this experiment was repeated without using a large excess of the alcohol (8 mol eq), no product (**6**) was observed, even when the reaction mixture was heated to reflux and the amount of catalyst increased five-fold. To conclusively verify that these results apply to our system, racemic epoxide (**3**) and diol (**4**) were subjected to the BiCl₃ catalyst and no product was observed. Thus, BiCl₃ is not a suitable catalyst for promoting the condensation of precious alcohols and epoxides.

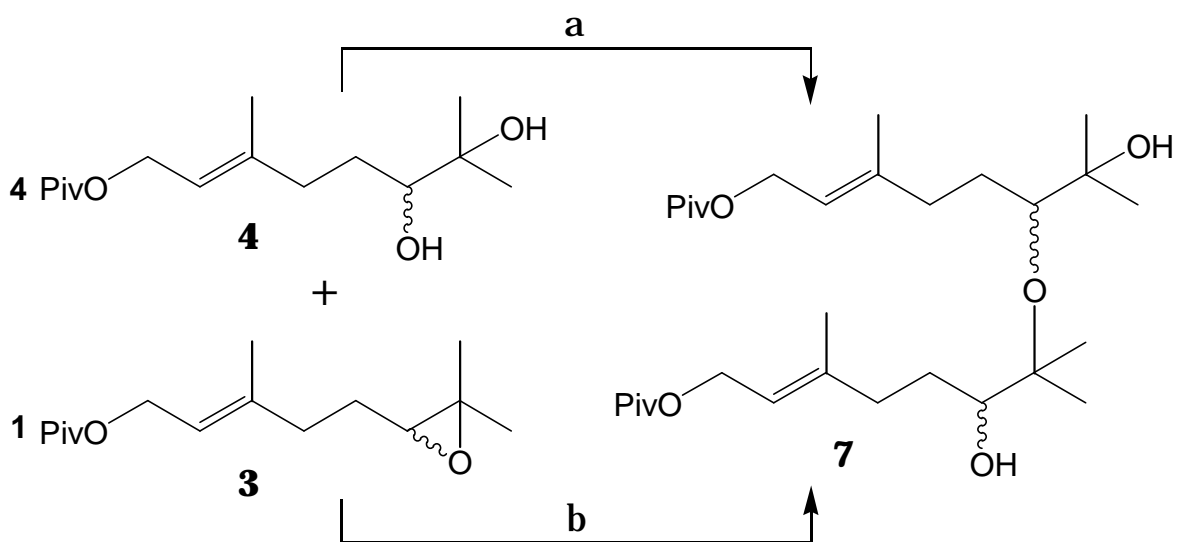
Scheme 2^a



^a Reagents and Conditions: (a) BiCl₃ (0.1 mol eq), isopropanol, rt, 2 h, 39%; (b) BiCl₃ (0.5 mol eq), isopropanol (8 mol eq), benzene, reflux, 4 h; (c) BiCl₃ (0.25 mol eq), hexafluoroisopropanol, rt, 48 h.

We next turned our attention towards the newly reported Cu^{II}(BF₄)₂ Lewis acid catalyst.¹⁷ Based on a reported epoxide opening using only 4 mole equivalents of methanol to afford the opened epoxide product in 99% yield, it looked to be very promising for our purposes. The first trial of Cu^{II}(BF₄)₂ with our racemic system (**Scheme 3**) afforded the desired product, which was confirmed to be the sought-after dimer (**7**) by ¹H and ¹³C NMR spectroscopy and electrospray mass spectrometry. Further work showed that the yield of

Scheme 3^a



^a Reagents and Conditions: (a) $\text{Cu}^{\text{II}}(\text{BF}_4)_2$ (0.1 mol eq), CH_2Cl_2 , rt, 20 h, 9%; (b) $\text{Cu}^{\text{II}}(\text{BF}_4)_2$ (0.3 mol eq), CH_2Cl_2 , reflux, 24 h, 28%.

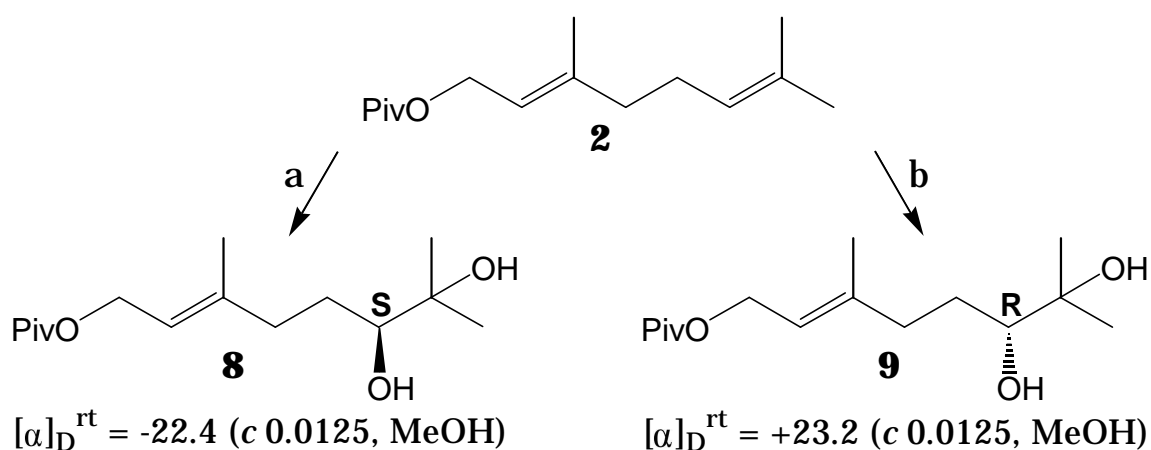
product (**7**) could be increased by raising the temperature of the reaction and increasing the mole equivalent of $\text{Cu}^{\text{II}}(\text{BF}_4)_2$ catalyst used. Thus, we had identified $\text{Cu}^{\text{II}}(\text{BF}_4)_2$ to be a suitable Lewis acid catalyst for promoting the formation of the ether-linked dimer (**7**) from the epoxide (**3**) and the diol (**4**).

Chapter 3

Stereocontrolled Synthesis

Having identified a reliable method for synthesizing the ether-linked dimer with the racemic model system, we concentrated our efforts on a stereocontrolled synthesis. The protected geranyl pivaloate (**2**) already on hand was oxidized to the corresponding (S)-diol (**8**) and (R)-diol (**9**) using the Sharpless method of asymmetric dihydroxylation (**Scheme 4**).¹⁹ It is noteworthy that both in these oxidations as well as in the oxidation of **2** to **3**, the reactions are cleanly regioselective towards the terminal alkene. The other alkene is thought to avoid oxidation due to the proximity of the electronegative ester, which reduces the nucleophilicity of that alkene.²⁰

Scheme 4^a



^a Reagents and Conditions: (a) AD-mix- α , MeSO_2NH_2 , 1:1 (v/v) H_2O :*t*-BuOH, 0 °C, overnight, 95%; (b) AD-mix- β , MeSO_2NH_2 , 1:1 (v/v) H_2O :*t*-BuOH, 0 °C, overnight, 62%.

The optical rotations of diols **8** and **9** were found to be opposite of each other and of approximately the same magnitude as the optical rotations of the corresponding diols of geranyl acetate,¹³ suggesting the possibility of optical purity. However, optical rotation data can be unreliable, and we were interested in quantitatively determining the enantiomeric excesses of the desired diol isomers.

The first strategy for determining the enantiomeric excess was to transform the secondary alcohol of the diol into the corresponding Mosher ester and analyze the ratio of diastereomers thus formed with NMR spectroscopy.²¹ However, this method proved to be unsuccessful. Due to the small reaction scales and losses during purification, ¹H NMR could not detect the minor stereoisomers. ¹⁹F NMR seemed promising due to its greater sensitivity in this case, however the signal peaks were transient and thus could not be correlated to specific stereoisomers. Furthermore, ¹⁹F NMR analysis of Mosher esters has been demonstrated to be unreliable.¹⁶

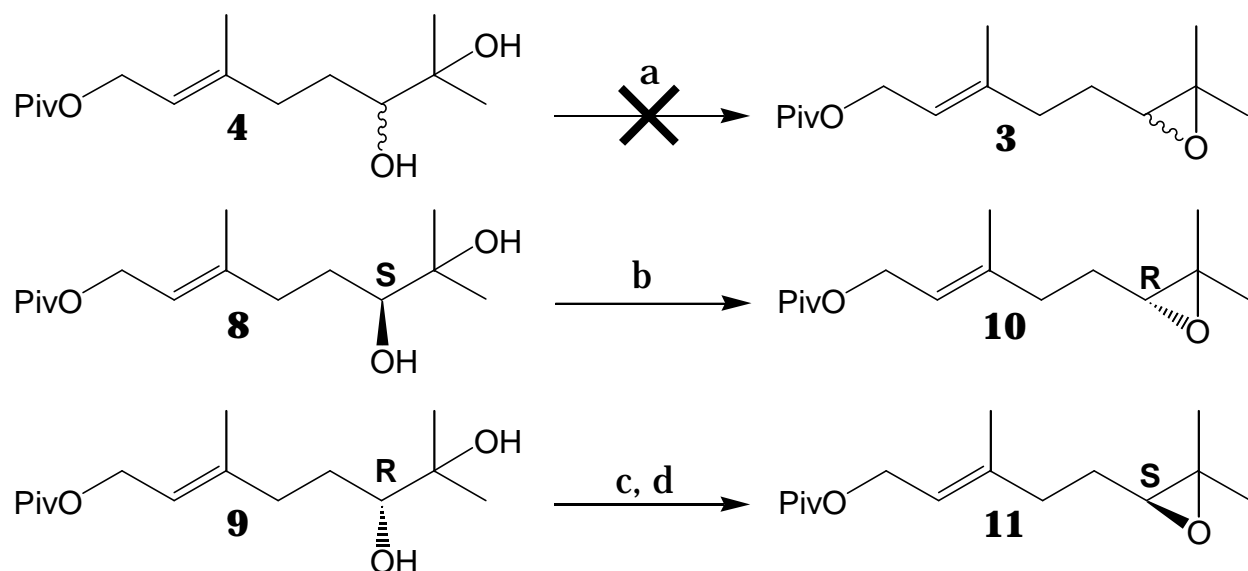
The next strategy was to allow the diols to complex to the chiral lanthanide NMR shift reagent Eu(hfc)₃ and then analyze these formed complexes by ¹H NMR.²² Diols **8** and **9** were combined in an NMR tube in mole ratios of 1:2 and 1:1 with 0.8 mole equivalents of Eu(hfc)₃ and analyzed by ¹H NMR. Two sets of signals with integration ratios very similar to the known mole ratios of diols **8** and **9** were readily identified in each spectrum

recorded. Then, diols **8** and **9** were individually analyzed by ^1H NMR in the presence of 0.8 mole equivalents of $\text{Eu}(\text{hfc})_3$ and the integrations of the major and minor stereoisomer signals were used to determine the enantiomeric excess. For both **8** and **9**, the enantiomeric excess of the major stereoisomer was found to be greater than 96%.

Having established the high enantiomeric purity of the first stereospecific synthetic intermediates **8** and **9**, we attempted to convert these diols to their corresponding epoxides. This seemingly facile transformation presented a synthetic challenge which we had not anticipated. All reactions we tried shared the common strategy of making the secondary alcohol a better leaving group and then promoting $\text{S}_{\text{N}}2$ attack with the tertiary alcohol, thus displacing the derivatized secondary alcohol and forming the epoxide ring. However, significant steric bulk around both alcohols prevented an efficient reaction.

Once the supply of racemic diol (**4**) was exhausted, stereopure diols **8** and **9** were alternated in these epoxidation trials (**Scheme 5**). Racemic diol (**4**) was converted into the secondary tosylate and treated with the mild base TEA. No epoxide product was observed from this reaction despite letting it run overnight. (S)-diol (**8**) was also converted to the corresponding tosylate; when treated with the strong base NaH, a small amount of epoxide (**10**) was observed to form. Furthermore, when (R)-diol (**9**) was converted to the

Scheme 5^a



^a Reagents and Conditions: (a) TsCl, TEA, CH₂Cl₂, rt, overnight; (b) TsCl, NaH, benzene, rt, overnight, 22%; (c) MsCl, py, THF, 0 °C, overnight; (d) NaH, rt, 5 h, 26%.

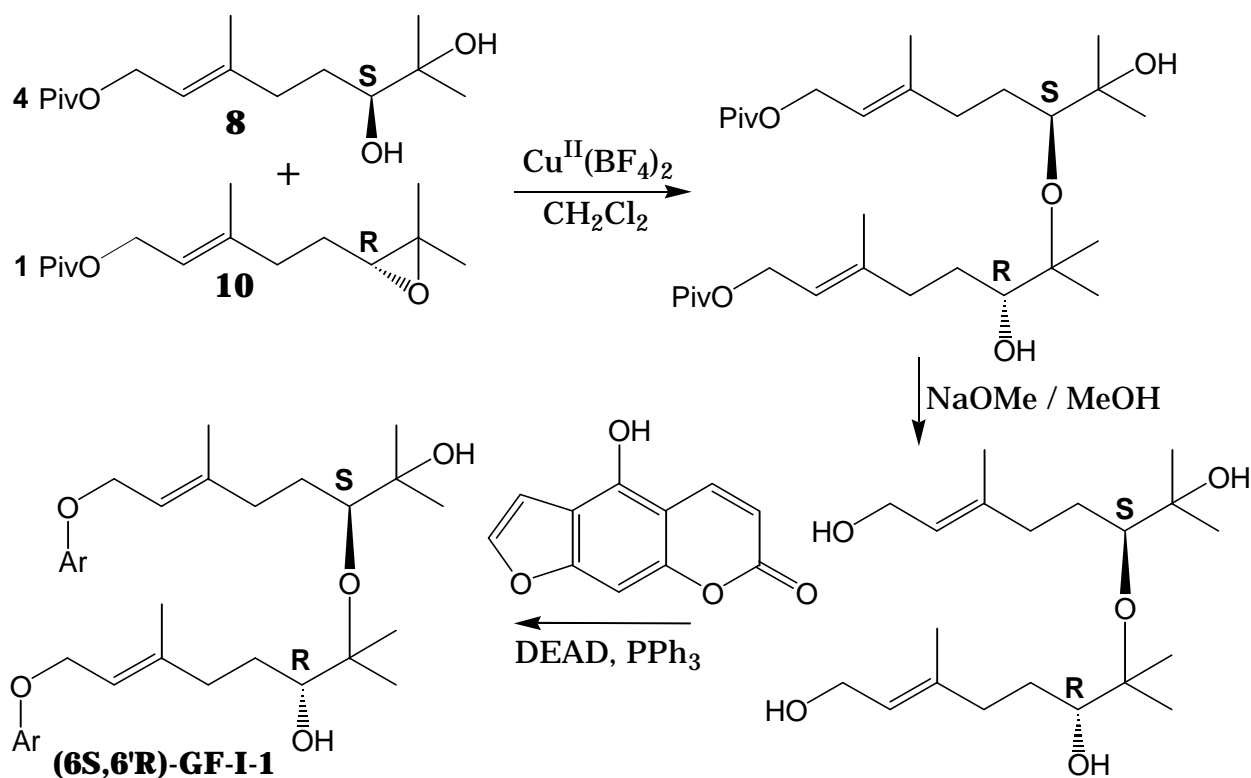
corresponding mesylate and NaH was introduced as a second step, the yield of epoxide (**11**) was observed to increase. Even so, the largest yield obtained was only 26%, which did not provide enough material to allow making the stereospecific ether-linked dimers. Time did not permit us to discover an efficient method of transforming the diol into the epoxide, though if greater amounts of starting material were used, this challenge could potentially be disregarded.

Chapter 4

Future Directions

We have determined a suitable method for achieving the key dimer intermediate **7** and progressed along the stereocontrolled synthetic route toward GF-I-1. Once optimal reaction conditions are worked out for the formation of epoxides **10** and **11**, they can be easily opened by diols **8** and **9** (**Scheme 6**, shown for the 6S,6'R diastereomer of GF-I-1), as has already been done with the racemic system. The coupling of the two stereopure diols with the two stereopure epoxides will lead to the formation of all four possible

Scheme 6



diastereomers of the ether-linked dimer. Each of these four diastereomeric dimers can then be deprotected with mild base to return the primary alcohols, followed by Mitsunobu coupling²³ to the psoralen heterocycle to form GF-I-1.

Comparisons of optical rotation data and ¹H NMR spectroscopic data of the four diastereomers with those of the natural GF-I-1 should result in the unequivocal assignment of the absolute stereochemistry of natural GF-I-1. The four diastereomers can then be bioassayed on their inhibition of CYP3A4 to determine the sensitivity of the enzyme to the different stereoisomers of GF-I-1.

Experimental Section

General Methods. All moisture and air sensitive reactions were performed in oven dried glassware equipped with rubber septa under a positive pressure of nitrogen or argon. When necessary, solvents were distilled prior to use. Reaction mixtures were magnetically stirred. Thin layer chromatography was performed on Merck precoated silica gel 5534-3 plates (0.2 mm). Concentration in vacuo was generally performed using a Büchi rotary evaporator and a high vacuum oil pump. Flash column chromatography was performed on Merck 70–230 mesh silica gel under a positive pressure of nitrogen. Optical rotations were determined using a Perkin-Elmer 241 polarimeter. Nuclear magnetic resonance spectra were recorded with Varian model Mercury 300 or UnityINOVA 400 instruments. ^1H NMR spectra were obtained with CDCl_3 as solvent using TMS as an internal standard (0 ppm). ^{13}C NMR spectra were recorded in ppm relative to the solvent signal: CDCl_3 (77 ppm). Electrospray mass spectra were recorded on a Micromass model LCT spectrometer and are reported in units of mass to charge (m/z).

Geranyl pivaloate (2). To a stirred solution of **1** (6.47 g, 42 mmol) and pyridine (6.4 mL, 84 mmol) in CH_2Cl_2 (40 mL) at 0 °C was added pivaloyl chloride (7.8 mL, 63 mmol). The solution was stirred for 3 h at that

temperature. The reaction was quenched with sat. NH_4Cl (80 mL) and extracted with CH_2Cl_2 (40 mL). The organic layer was washed with sat. NH_4Cl (2 x 80 mL) and sat. CuSO_4 successively. The combined organic layers were dried with Na_2SO_3 , filtered, and the solvent was removed in vacuo. Column chromatography (50:1 hexane:ethyl acetate) afforded **2** in 88% yield (8.77 g). $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ 5.32 (tq, $J = 7.0, 1.5$ Hz, 1H), 5.08 (tt, $J = 6.9, 1.5$ Hz, 1H), 4.57 (d, $J = 7.0$ Hz, 2H), 2.06 (m, 4H), 1.68 (d, $J = 7.5$ Hz, 6H), 1.60 (s, 3H), 1.20 (s, 9H).

Racemic 6,7-epoxygeranyl pivaloate (3). To a stirred solution of **2** (4.66 g, 19.6 mmol) in CH_2Cl_2 (117 mL) at 0 °C was added *m*CPBA (5.83 g). The solution was stirred for 2 h at that temperature. The reaction mixture was filtered, washed with 10% aq. Na_2SO_3 (100 mL) and 10% aq. Na_2CO_3 (100 mL), dried with MgSO_4 , and filtered. The solvent was removed in vacuo and column chromatography (100:6 hexane:ethyl acetate) afforded **3** in 65% yield (3.24 g). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 5.37 (tq, $J = 7.0, 1.1$ Hz, 1H), 4.58 (d, $J = 6.9$ Hz, 2H), 2.70 (t, $J = 6.2$ Hz, 1H), 2.19 (m, 2H), 1.73 (s, 3H), 1.66 (m, 2H), 1.31, (s, 3H), 1.27 (s, 3H), 1.19 (s, 9H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 178.50, 140.63, 119.36, 63.84, 61.17, 58.30, 38.70, 36.13, 27.17, 24.81, 22.61, 18.72, 16.43, 14.07.

Racemic 6,7-dihydroxygeranyl pivaloate (4). To a stirred solution of **3** (2.5 g, 9.84 mmol) in THF (180 mL) was added 5M H_2SO_4 dropwise (8.4 mL).

The solution was stirred for 45 m at room temperature. The reaction mixture was poured onto brine (200 mL) and extracted with diethyl ether (3 x 100 mL). The combined organic layers were dried with Na₂SO₄, filtered, and the solvent was removed in vacuo. Column chromatography (15:1 hexane: ethyl acetate) afforded **4** in 62% yield (1.57 g). ¹H NMR (400 MHz, CDCl₃) δ 5.38 (tq, *J* = 7.0, 1.1 Hz, 1H), 4.58 (d, *J* = 6.6 Hz, 2H), 3.33 (dd, *J* = 10.4, 1.8 Hz, 1H), 2.48 (br, 1H), 2.32 (m, 1H), 2.14 (m, 1H), 1.72 (s, 3H), 1.61 (m, 1H), 1.46 (m, 1H), 1.21 (s, 3H), 1.19 (s, 9H), 1.17 (s, 3H).

1-phenyl-1-isopropoxyethanol (6). To a stirred solution of **5** (360 mg, 3 mmol) in isopropanol (15 mL) at reflux was added BiCl₃ (95 mg, 0.1 mol eq). The solution was stirred for 2 h at reflux. The solvent was removed in vacuo and the crude reaction mixture was dissolved in diethyl ether (15 mL) and poured onto H₂O (15 mL). The organic phase was separated, and the aqueous phase was extracted with diethyl ether (2 x 15 mL). The combined organic layers were dried with MgSO₄, filtered, and the solvent was removed in vacuo. Column chromatography (1:1 hexane:ethyl acetate) afforded **6** in 39% yield (209 mg). ¹H NMR (400 MHz, CDCl₃) δ 7.40 (m, 5H), 4.53 (dd, *J* = 8.0, 4.7 Hz, 1H), 3.60 (m, 3H), 2.33 (br, 1H), 1.19 (d, *J* = 6.2 Hz, 3H), 1.13 (d, *J* = 6.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 128.41, 127.89, 126.79, 79.92, 69.54, 67.49, 23.45.

Racemic dimer (7). To a stirred solution of **3** (60 mg, 0.236 mmol) and **4** (244 mg, .897 mmol) in CH₂Cl₂ (1 mL) was added Cu^{II}(BF₄)₂ catalyst (15 mg, .30 mol eq). The solution was stirred for 24 h at reflux. The reaction mixture was quenched with H₂O (20 mL), and extracted with CH₂Cl₂ (3 x 20 mL). The combined organic layers were washed with brine (60 mL), dried with Na₂SO₄, filtered, and the solvent was removed in vacuo. Column chromatography (15:1 hexane:ethyl acetate) afforded **7** in 28% yield (40 mg). ¹H NMR (400 MHz, CDCl₃) δ 5.38 (td, *J* = 6.9, 1.2 Hz, 2H), 4.58 (d, *J* = 6.9 Hz, 4H), 3.53 (tq, *J* = 11.2, 2.2 Hz, 2H), 2.34 (m, 2H), 2.14 (m, 2H), 2.07 (m, 2H), 1.72 (s, 6H), 1.66 (s, 6H), 1.56 (m, 2H), 1.45 (m, 2H), 1.37 (s, 6H), 1.31 (s, 6H), 1.19 (s, 18H); ¹³C NMR (100 MHz, CDCl₃) δ 178.59, 141.17, 119.33, 98.74, 97.10, 76.26, 76.04, 61.21, 38.72, 36.07, 29.14, 29.09, 27.18, 23.65, 23.40, 21.37, 21.14, 16.45; ES MS found: *m/z* 531.4, calcd for [C₃₀H₅₄O₇ + Na⁺ - H₂O]: *m/z* 531.

(S)-6,7-dihydroxygeranyl pivaloate (8). To a stirred solution of H₂O (80 mL) and *t*-BuOH (80 mL) at 0 °C were added the AD-mix- α very slowly (20.6 g), MeSO₂NH₂ (1.40 g), and **2** (3.97 mL, 14.7 mmol), successively. The solution was stirred overnight at that temperature. The reaction mixture was quenched with Na₂SO₃ (22.0 g). The organic layers were extracted with ethyl acetate (3 x 80 mL), washed with 2M KOH (2 x 200 mL), dried with MgSO₄, and filtered. The solvent was removed in vacuo and column

chromatography (20:1 CH₂Cl₂:MeOH) afforded **8** in 95% yield (3.81 g). $[\alpha]_D^{rt}$ -22.4 (c 0.0125 in MeOH); ¹H NMR (400 MHz, CDCl₃) δ 5.39 (tq, *J* = 6.7, 1.1 Hz, 1H), 4.58 (dd, *J* = 7.0, 0.7 Hz, 2H), 3.35 (dd, *J* = 10.5, 1.8 Hz, 1H), 2.31 (m, 2H), 2.13 (m, 2H), 1.72 (s, 3H), 1.66 (m, 3H), 1.46 (m, 1H), 1.21 (s, 3H), 1.20 (s, 9H), 1.17 (s, 3H).

(R)-6,7-dihydroxygeranyl pivaloate (9). To a stirred solution of H₂O (80 mL) and *t*-BuOH (80 mL) at 0 °C were added the AD-mix- β very slowly (20.6 g), MeSO₂NH₂ (1.40 g), and **2** (2.50 mL, 9.26 mmol), successively. The solution was stirred overnight at that temperature. The reaction mixture was quenched with Na₂SO₃ (22.0 g). The organic layers were extracted with ethyl acetate (3 x 80 mL), washed with 2M KOH (2 x 200 mL), dried with MgSO₄, and filtered. The solvent was removed in vacuo and column chromatography (20:1 CH₂Cl₂:MeOH) afforded **9** in 62% yield (1.56 g). $[\alpha]_D^{rt}$ +23.2 (c 0.0125 in MeOH); ¹H NMR (400 MHz, CDCl₃) δ 5.37 (tq, *J* = 6.8, 1.1 Hz, 1H), 4.58 (d, *J* = 6.9 Hz, 2H), 3.34 (d, *J* = 10.4 Hz, 1H), 2.46 (br, 1H), 2.32 (m, 1H), 2.21 (br, 1H), 2.12 (m, 1H), 1.72 (s, 3H), 1.61 (m, 1H), 1.46 (m, 1H), 1.21 (s, 3H), 1.19 (s, 9H), 1.17 (s, 3H).

(R)-6,7-epoxygeranyl pivaloate (10). To a stirred solution of **8** (200 mg, 0.736 mmol) and tosyl chloride (74 mg, 0.772 mmol) in benzene (16 mL) was added NaH slowly (58 mg). The solution was stirred overnight at room temperature. The reaction mixture was quenched with H₂O (32 mL) and

extracted with ethyl acetate (16 mL). The organic layer was washed with H₂O (2 x 32 mL), dried with MgSO₄, filtered, and the solvent was removed in vacuo. Column chromatography (10:1 hexane:ethyl acetate) afforded **10** in 22% yield (42 mg). ¹H NMR (400 MHz, CDCl₃) δ 5.37 (tq, *J* = 7.0, 1.5 Hz, 1H), 4.58 (d, *J* = 6.9 Hz, 2H), 2.71 (t, *J* = 6.2 Hz, 1H), 2.17 (m, 2H), 1.73 (s, 3H), 1.68 (m, 2H), 1.31 (s, 3H), 1.27 (s, 3H), 1.19 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 178.49, 140.61, 119.44, 119.18, 63.99, 63.68, 61.41, 61.14, 60.86, 60.31, 58.31, 38.66, 36.09, 27.25, 27.01, 24.90, 24.67, 20.99, 20.95, 18.80, 18.56, 16.49, 16.32, 14.24, 14.01.

(S)-6,7-epoxygeranyl pivaloate (11). To a stirred solution of **9** (200 mg, 0.736 mmol) and pyridine (1 mL) in THF (5 mL) at 0 °C was added the mesyl chloride dropwise (91 μL). The solution was stirred overnight at that temperature. NaH (210 mg) and additional THF (5 mL) were added to the reaction mixture, and it was allowed to stir for an additional 5 h. The reaction was quenched with H₂O (15 mL). The organic layer was extracted with ethyl acetate (15 mL) and washed with sat. CuSO₄ (2 x 20 mL) and brine (20 mL). The combined organic layers were dried with Na₂SO₄, filtered, and the solvent was removed in vacuo. Column chromatography (10:1 hexane:ethyl acetate) afforded **11** in 26% yield (49 mg). ¹H NMR (400 MHz, CDCl₃) δ 5.37 (tq, *J* = 6.9, 1.5 Hz, 1H), 4.58 (d, *J* = 6.8 Hz, 2H), 2.71 (t, *J* = 6.2 Hz, 1H), 2.17 (m, 2H), 1.73 (s, 3H), 1.70 (m, 2H), 1.31 (s, 3H), 1.27 (s, 3H), 1.19 (s, 9H).

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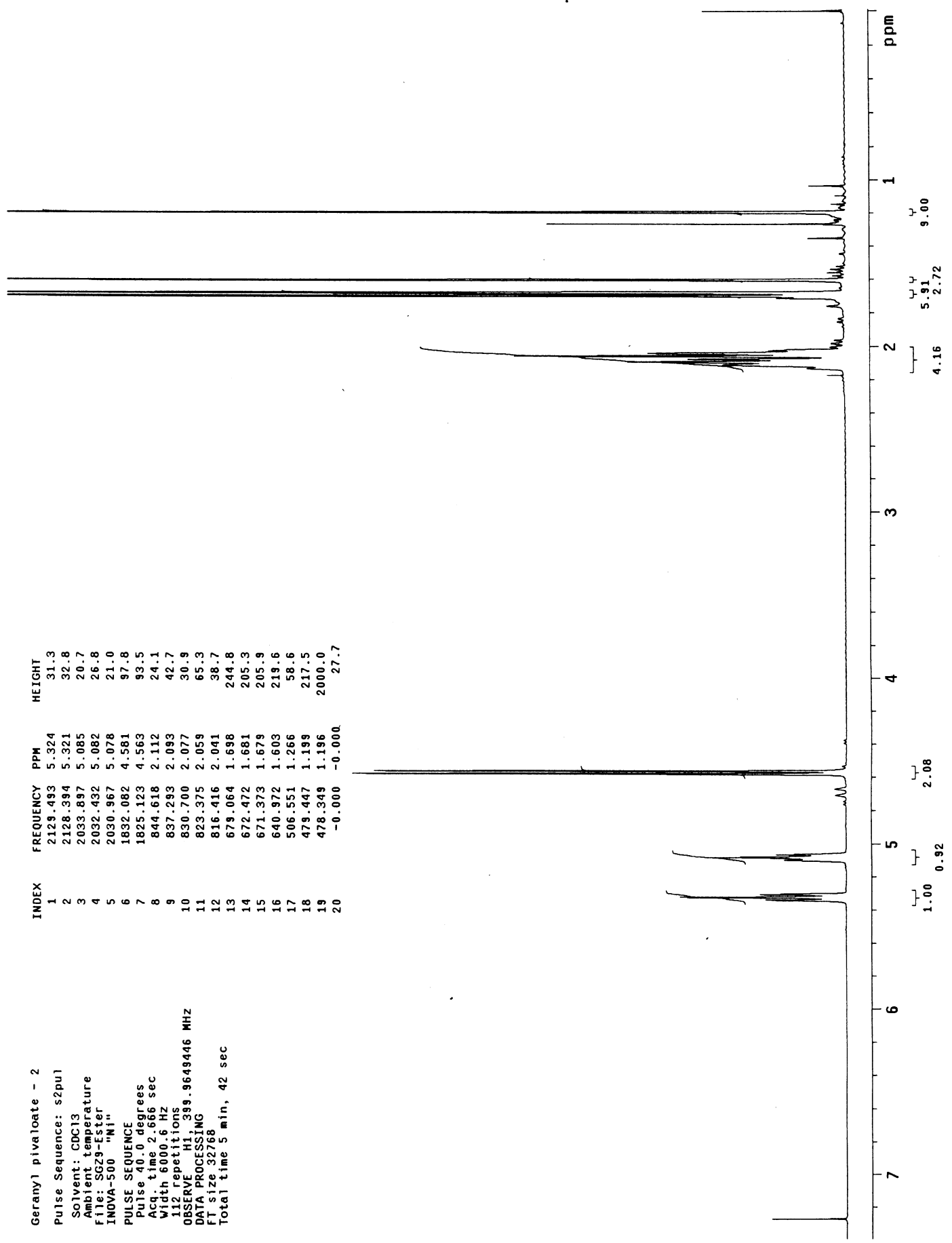
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Geranyl pivaloate - 2

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 Ambient temperature
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 Acq. time 2.666 sec
 Width 6000.6 Hz
 112 repetitions
 OBSERVE H1, 399.9649446 MHz
 DATA PROCESSING
 FT size 32768
 Total time 5 min, 42 sec

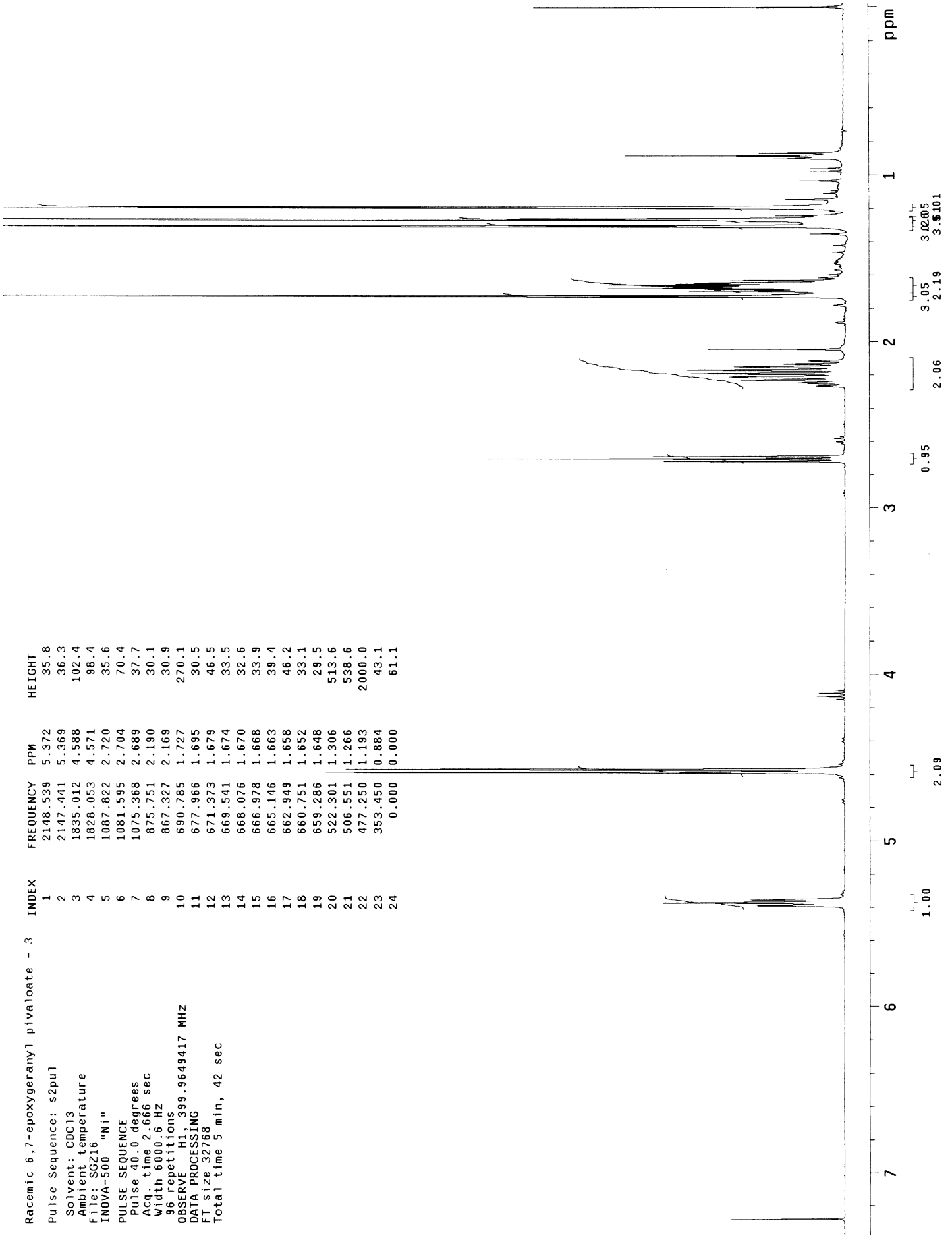
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4	2032.432	5.082
5	2030.967	5.078
6	1832.082	4.581
7	1825.123	4.563
8	844.618	2.112
9	837.293	2.093
10	830.700	2.077
11	823.375	2.059
12	816.416	2.041
13	679.064	1.698
14	672.472	1.681
15	671.373	1.679
16	640.972	1.603
17	506.551	1.266
18	479.447	1.199
19	478.349	1.186
20	-0.000	-0.000



Racemic 6,7-epoxygeranyl pivaloate - 3

Pulse Sequence: s2pu1
 Solvent: CDCl3
 Ambient temperature
 File: SGZ16
 INOVA-500 "N1"
 PULSE SEQUENCE
 Pulse 40.0 degrees
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 FT size 32768
 Total time 5 min, 42 sec

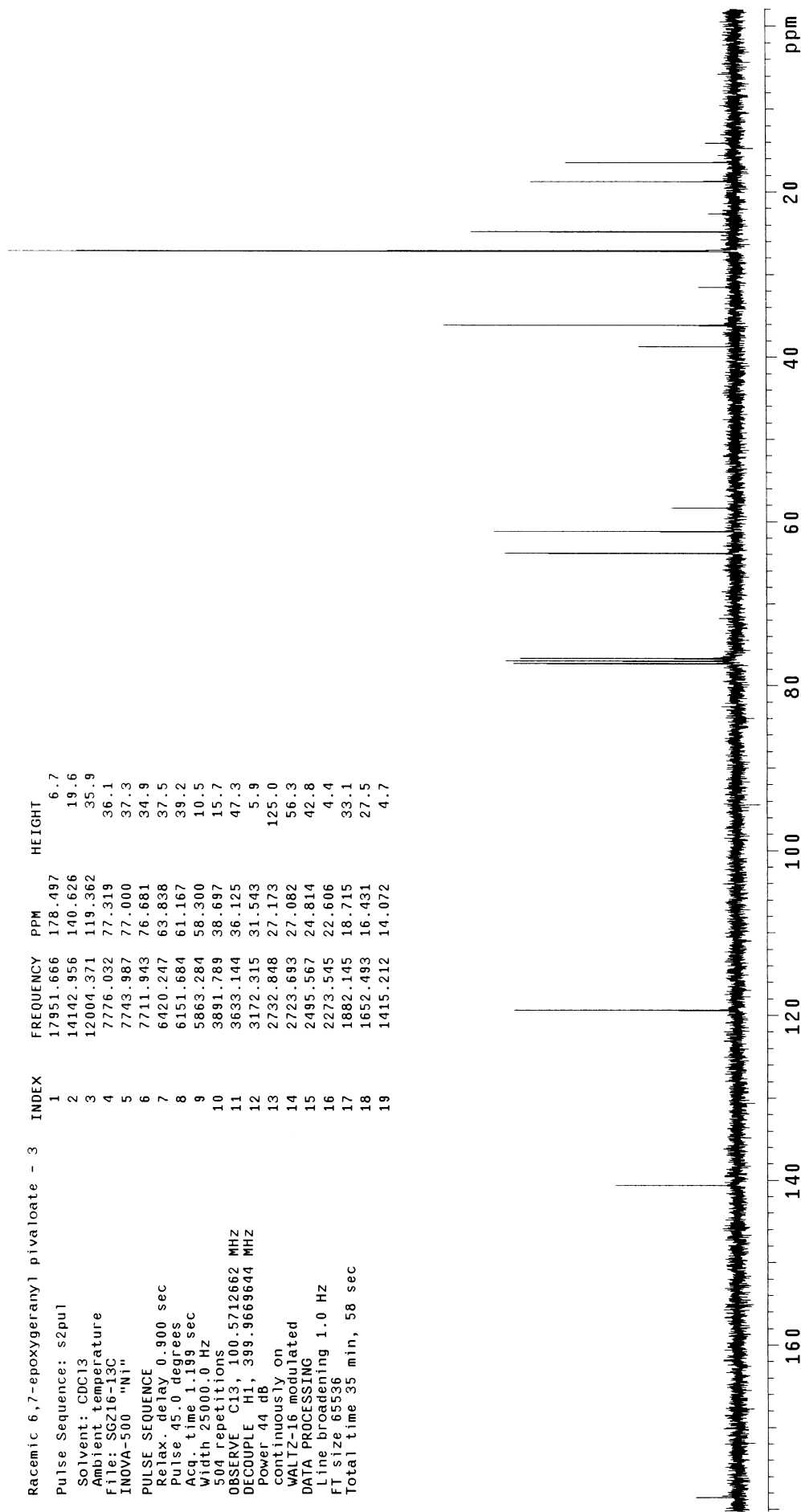
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4	1828.053	4.571
5	1087.822	2.720
6	1081.595	2.704
7	1075.368	2.689
8	875.751	2.190
9	867.327	2.169
10	690.785	1.727
11	677.966	1.695
12	671.373	1.679
13	669.541	1.674
14	668.076	1.670
15	666.978	1.668
16	665.146	1.663
17	662.949	1.658
18	660.751	1.652
19	659.286	1.648
20	522.301	1.306
21	506.551	1.266
22	477.250	1.193
23	353.450	0.884
24	0.000	0.000



Racemic 6,7-epoxygeranyl pivaloate - 3

Pulse Sequence: s2pul
 Solvent: CDCl3
 Ambient temperature
 File: SG216-13C
 INOVA-500 "N1"
 PULSE SEQUENCE
 Relax. delay 0.900 sec
 Pulse 45.0 degrees
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 DECOUPLE H1, 399.9669644 MHZ
 Power 44 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 35 min, 58 sec

INDEX	FREQUENCY PPM	HEIGHT
1	17951.666	178.497
2	14142.956	140.626
3	12004.371	119.362
4	7776.032	77.319
5	7743.987	77.000
6	7711.943	76.681
7	6420.247	63.838
8	6151.684	61.167
9	5863.284	58.300
10	3891.789	38.697
11	3633.144	36.125
12	3172.315	31.543
13	2732.848	27.173
14	2723.693	27.082
15	2495.567	24.814
16	2273.545	22.606
17	1882.145	18.715
18	1652.493	16.431
19	1415.212	14.072



Racemic 6,7-dihydroxygeranyl pivaloate - 4

Pulse Sequence: s2pul

Solvent: CDC13

Ambient temperature

INDVA-400 "Zr"

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Acq. time 2.666 sec

Width 6000.6 Hz

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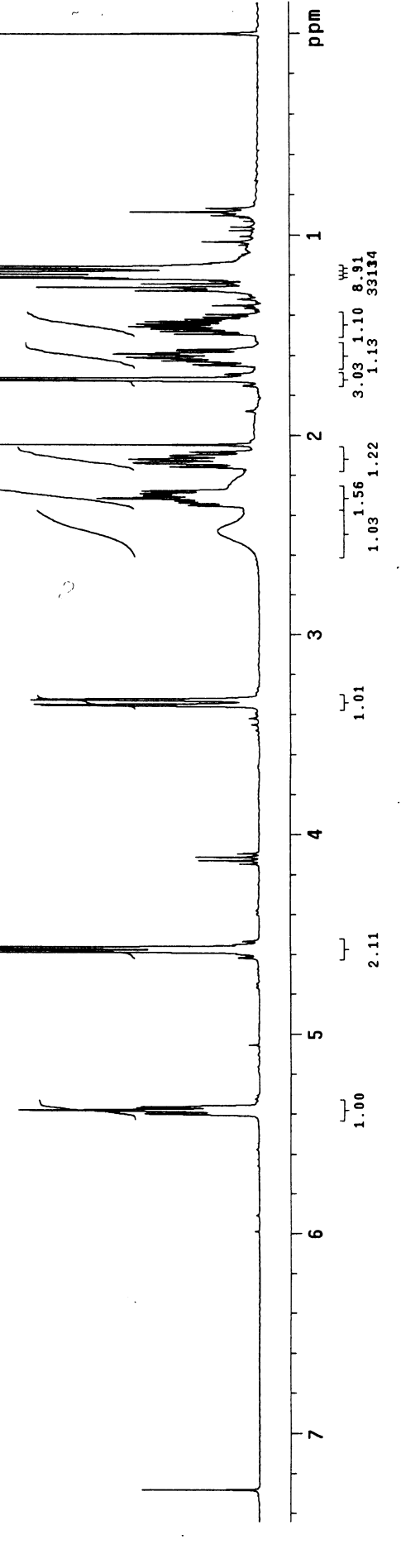
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DATA PROCESSING

FT size 32768

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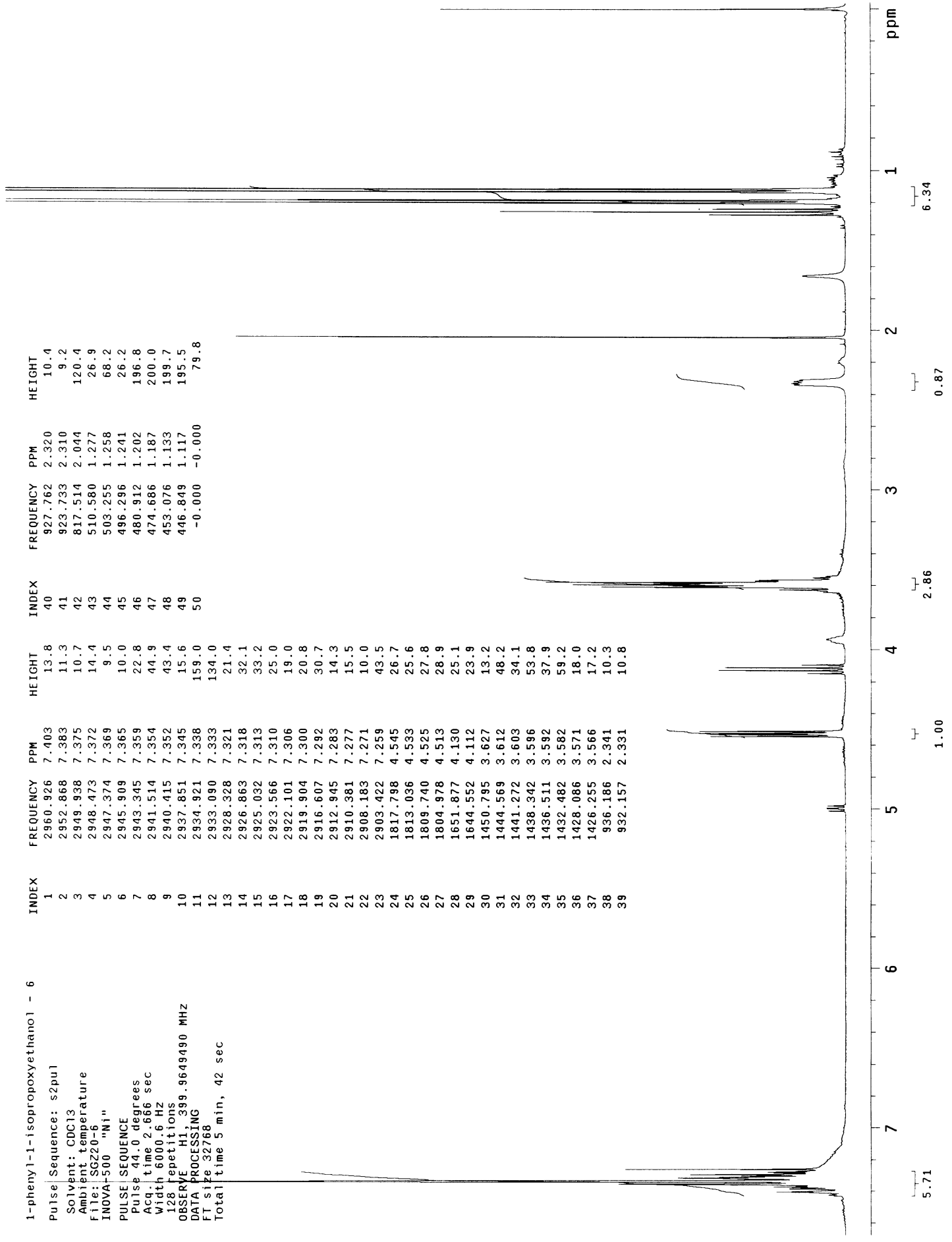
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2	2159.894	5.400	41	653.059	1.633	80	468.895	1.165	443.1		
3	2158.795	5.397	42	650.862	1.627	81	458.006	1.140	8.1		
4	2154.400	5.386	43	648.298	1.621	82	418.152	1.053	9.3		
5	2153.301	5.384	44	646.100	1.615	83	388.711	0.974	4.9		
6	2151.836	5.380	45	643.903	1.610	84	368.118	0.958	4.6		
7	2150.737	5.377	46	641.705	1.604	85	368.043	0.900	7.7		
8	2147.441	5.369	47	639.141	1.598	86	358.084	0.883	20.7		
9	2146.342	5.366	48	637.310	1.593	87	348.125	0.865	8.6		
10	2144.877	5.363	49	634.746	1.587	88	348.000	0.000	62.8		
11	2143.778	5.360	50	629.984	1.575						
12	1833.547	4.584	51	627.787	1.570						
13	1832.815	4.582	52	597.386	1.494						
14	1826.588	4.567	53	591.892	1.480						
15	1820.361	4.551	54	588.229	1.471						
16	1652.244	4.131	55	586.764	1.467						
17	1644.918	4.113	56	582.735	1.457						
18	1341.647	3.354	57	581.637	1.454						
19	1339.815	3.350	58	577.974	1.445						
20	1331.025	3.328	59	574.311	1.436						
21	1329.560	3.324	60	572.480	1.431						
22	992.591	2.482	61	568.817	1.422						
23	940.215	2.351	62	567.718	1.419						
24	935.087	2.338	63	563.689	1.409						
25	931.058	2.328	64	558.562	1.397						
26	925.930	2.315	65	540.248	1.351						
27	920.803	2.302	66	511.313	1.278						
28	917.140	2.293	67	510.214	1.276						
29	911.646	2.279	68	508.383	1.271						
30	863.298	2.158	69	505.819	1.265						
31	854.874	2.137	70	503.987	1.260						
32	847.915	2.120	71	499.592	1.249						
33	840.589	2.102	72	497.028	1.243						
34	833.264	2.083	73	496.296	1.241						
35	821.177	2.053	74	491.900	1.230						
36	818.980	2.048	75	490.802	1.227						
37	688.587	1.722	76	487.139	1.218						
38	677.599	1.694	77	483.476	1.209						
39	659.652	1.649	78	479.447	1.199						



1-phenyl-1-isopropoxyethanol - 6

Pulse Sequence: s2pul
 Solvent: CDC13
 Ambient temperature
 File: SGZ20-6
 INOVA-500 "N1"
 PULSE SEQUENCE
 Pulse 44.0 degrees
 Acq. time 2.666 sec
 Width 6000.6 Hz
 128 repetitions
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 DATA PROCESSING
 FT size 32768
 Total time 5 min, 42 sec

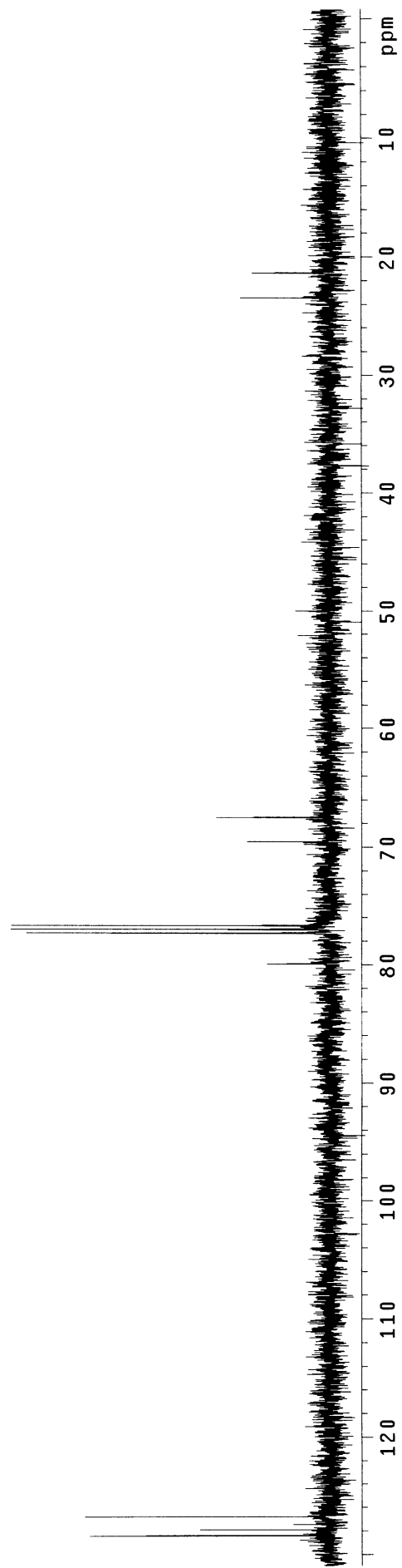
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2	2952.868	7.383	41	923.733	2.310
3	2949.938	7.375	42	817.514	2.044
4	2948.473	7.372	43	510.580	1.277
5	2947.374	7.369	44	503.255	1.258
6	2945.909	7.365	45	496.296	1.241
7	2943.345	7.359	46	480.912	1.202
8	2941.514	7.354	47	474.686	1.187
9	2940.415	7.352	48	453.076	1.133
10	2937.851	7.345	49	446.849	1.117
11	2934.921	7.338	50	-0.000	-0.000
12	2933.090	7.333			
13	2928.328	7.321			
14	2926.863	7.318			
15	2925.032	7.313			
16	2923.566	7.310			
17	2922.101	7.306			
18	2919.904	7.300			
19	2916.607	7.292			
20	2912.945	7.283			
21	2910.381	7.277			
22	2908.183	7.271			
23	2903.422	7.259			
24	1817.798	4.545			
25	1813.036	4.533			
26	1809.740	4.525			
27	1804.978	4.513			
28	1651.877	4.130			
29	1644.552	4.112			
30	1450.795	3.627			
31	1444.569	3.612			
32	1441.272	3.603			
33	1438.342	3.596			
34	1436.511	3.592			
35	1432.482	3.582			
36	1428.086	3.571			
37	1426.255	3.566			
38	936.186	2.341			
39	932.157	2.331			



1-phenyl-1-isopropoxyethanol - 6

Pulse Sequence: s2pul
 Solvent: CDCl3
 Ambient temperature
 File: SG220-6C
 INOVA-500 "ni"
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 Pulse 45.0 degrees
 Acq. time 1.189 sec
 Width 25000.0 Hz
 152 repetitions
 OBSERVE C13, 100.5712670 MHZ
 DECOUPLE H1, 399.9669644 MHZ
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 WALTZ-16 modulated
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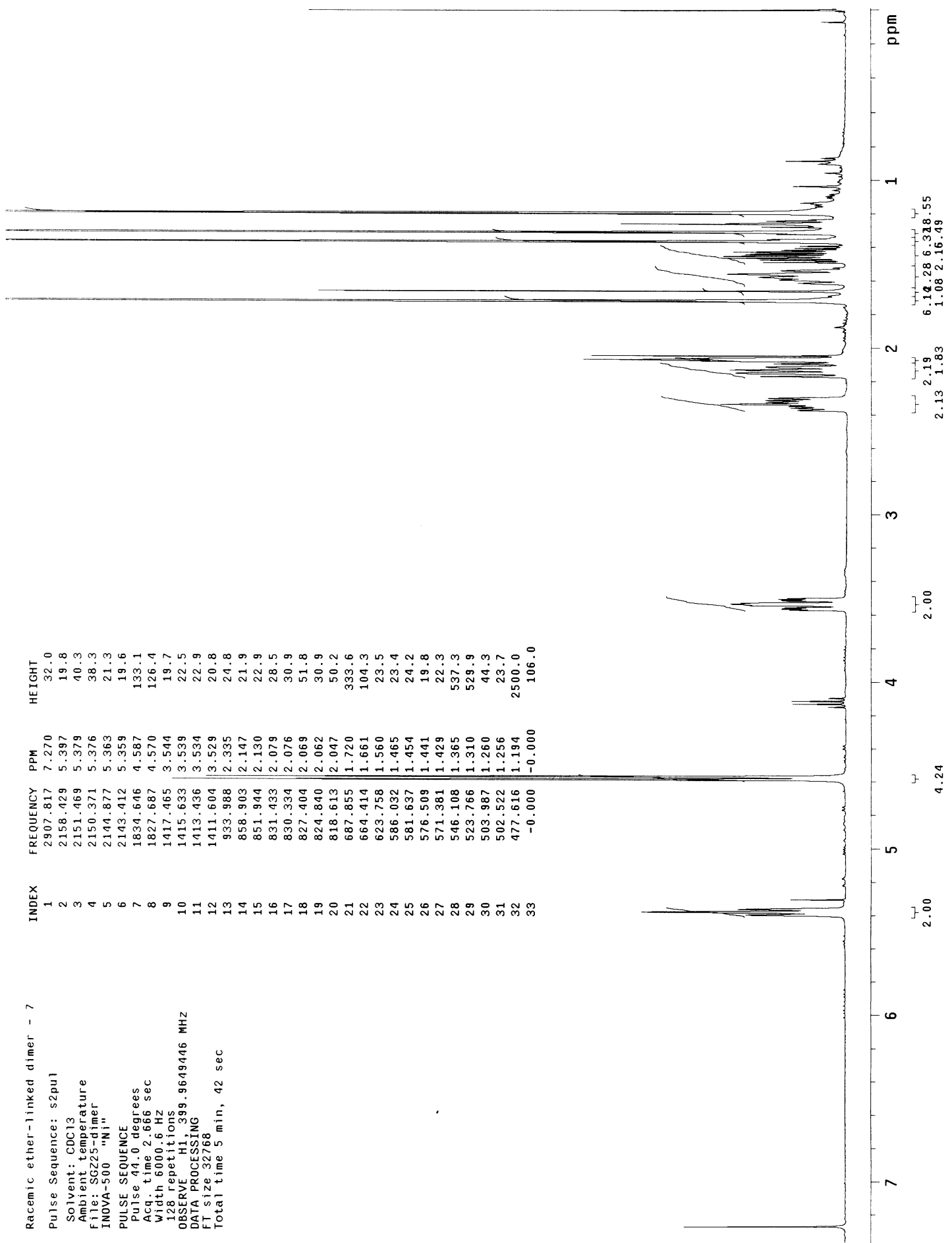
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4	8037.728	79.921	9.8
5	7776.032	77.319	47.6
6	7743.988	77.000	50.0
7	7711.943	76.681	49.9
8	6993.232	69.535	13.0
9	6787.995	67.494	17.8
10	2358.234	23.448	14.0
11	2145.367	21.332	12.1



Racemic ether-linked dimer - 7

Pulse Sequence: s2pul
 Solvent: CDCl3
 Ambient temperature
 File: SG225-dimer
 INOVA-500 "N1"
 PULSE SEQUENCE
 Pulse 44.0 degrees
 Acq. time 2.666 sec
 Width 6000.6 Hz
 128 repetitions
 OBSERVE H1, 399.9649446 MHZ
 DATA PROCESSING
 FT size 32768
 Total time 5 min, 42 sec

INDEX	FREQUENCY	PPM	HEIGHT
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3	2151.469	5.379	40.3
4	2150.371	5.376	38.3
5	2144.877	5.363	21.3
6	2143.412	5.359	19.6
7	1834.646	4.587	133.1
8	1827.687	4.570	126.4
9	1417.465	3.544	19.7
10	1415.633	3.539	22.5
11	1413.436	3.534	22.9
12	1411.604	3.529	20.8
13	933.988	2.335	24.8
14	858.903	2.147	21.9
15	851.944	2.130	22.9
16	831.433	2.079	28.5
17	830.334	2.076	30.9
18	827.404	2.069	51.8
19	824.840	2.062	30.9
20	818.613	2.047	50.2
21	687.855	1.720	333.6
22	664.414	1.661	104.3
23	623.758	1.560	23.5
24	586.032	1.465	23.4
25	581.637	1.454	24.2
26	576.509	1.441	19.8
27	571.381	1.429	22.3
28	546.108	1.365	537.3
29	523.766	1.310	529.9
30	503.987	1.260	44.3
31	502.522	1.256	23.7
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33	-0.000	-0.000	106.0



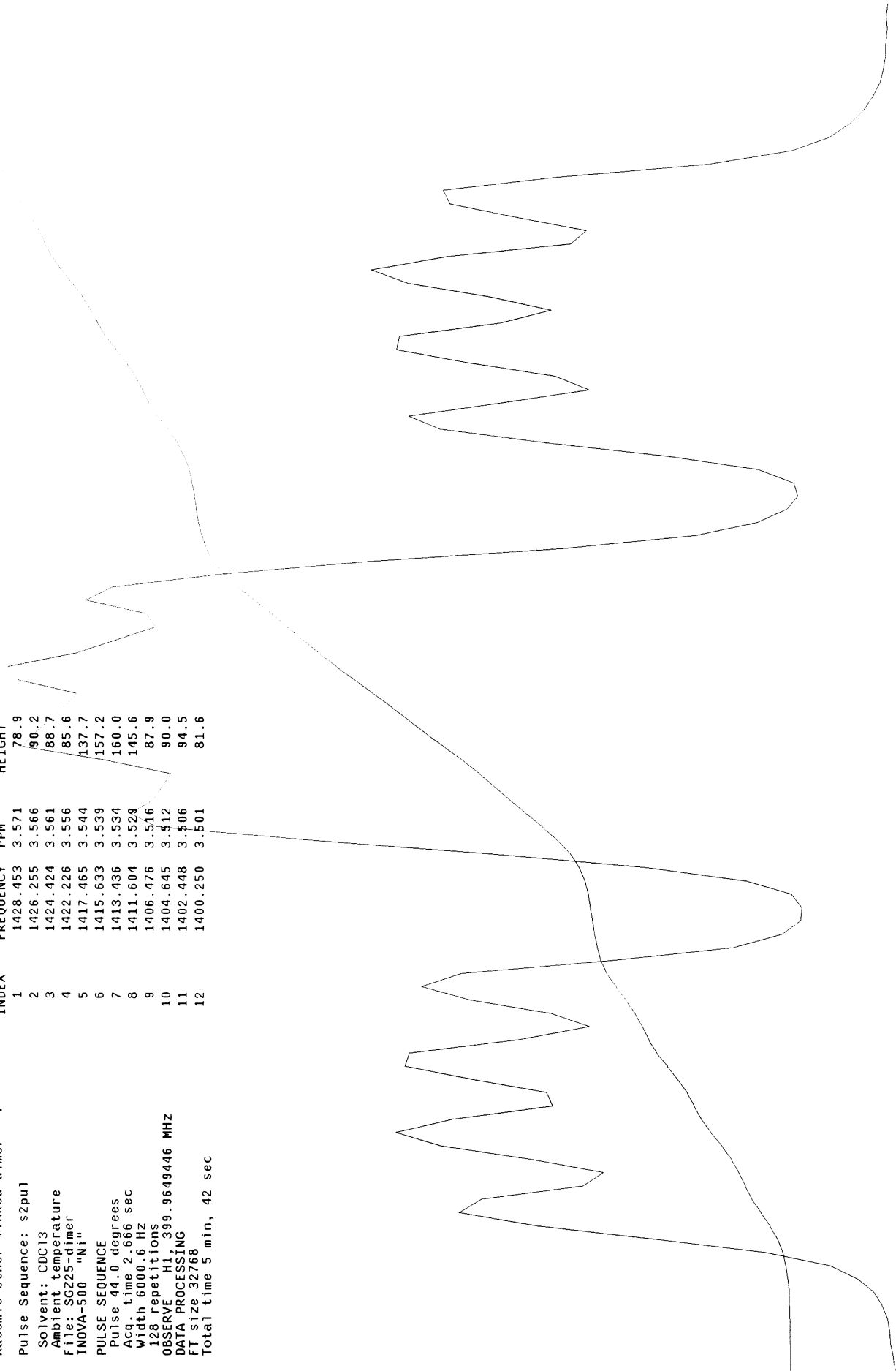
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2.00 4.24 2.00 2.00 2.13 1.83 2.19 6.14 2.28 6.32 8.55 1.08 2.16 4.9

Racemic ether-linked dimer - 7

Pulse Sequence: s2pul
Solvent: CDCl3
Ambient temperature
File: SGZ25-dimer
INOVA-500 "N1"
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Pulse 44.0 degrees
Acq. time 2.666 sec
Width 6000.6 Hz
128 repetitions
OBSERVE H1, 399.9649446 MHZ
DATA PROCESSING
FT size 32768
Total time 5 min, 42 sec

INDEX	FREQUENCY	PPM	HEIGHT
1	1428.453	3.571	78.9
2	1426.255	3.566	90.2
3	1424.424	3.561	88.7
4	1422.226	3.556	85.6
5	1417.465	3.544	137.7
6	1415.633	3.539	157.2
7	1413.436	3.534	160.0
8	1411.604	3.529	145.6
9	1406.476	3.516	87.9
10	1404.645	3.512	90.0
11	1402.448	3.506	94.5
12	1400.250	3.501	81.6

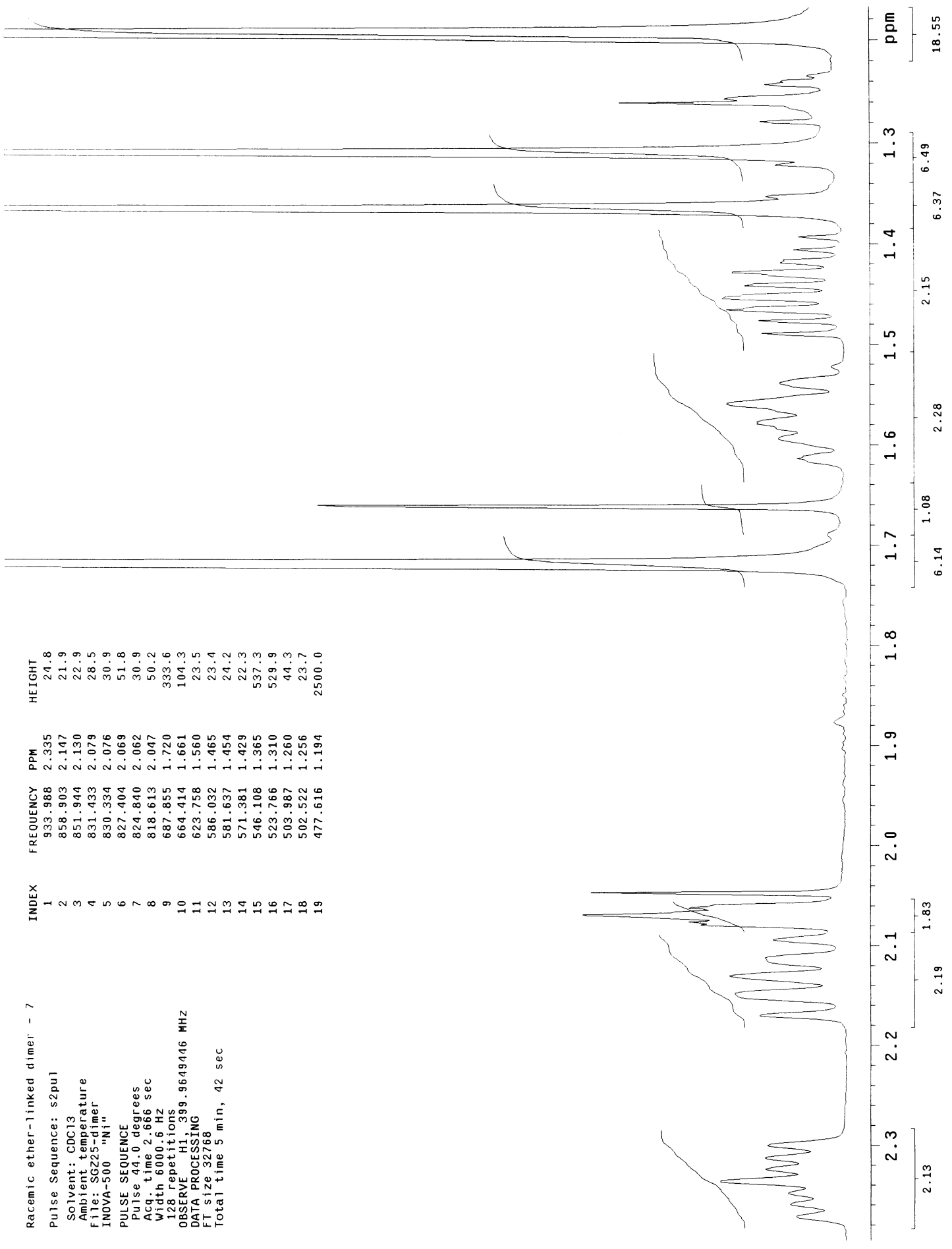


3.58 3.57 3.56 3.55 3.54 3.53 3.52 3.51 3.50 ppm

Racemic ether-linked dimer - 7

Pulse Sequence: s2pul
 Solvent: CDCl3
 Ambient temperature
 File: SG225-dimer
 INOVA-500 "N1"
 PULSE SEQUENCE
 Pulse 44.0 degrees
 Acq. time 2.666 sec
 Width 6000.6 Hz
 128 repetitions
 OBSERVE F1, 399.9649446 MHZ
 DATA PROCESSING
 FT size 32768
 Total time 5 min, 42 sec

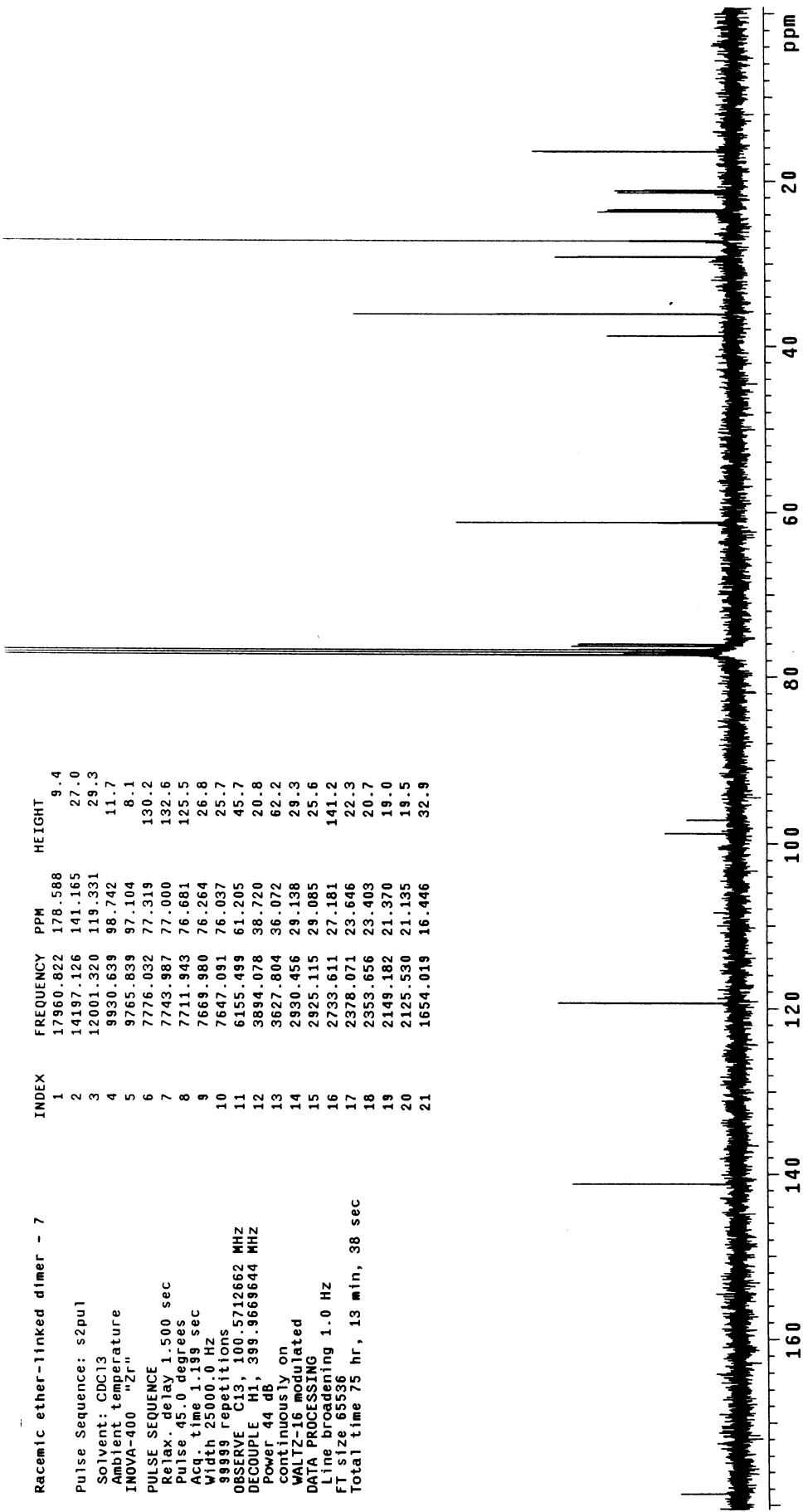
INDEX	FREQUENCY	PPM	HEIGHT
1	933.988	2.335	24.8
2	858.903	2.147	21.9
3	851.944	2.130	22.9
4	831.433	2.079	28.5
5	830.334	2.076	30.9
6	827.404	2.069	51.8
7	824.840	2.062	30.9
8	818.613	2.047	50.2
9	687.855	1.720	333.6
10	664.414	1.661	104.3
11	623.758	1.560	23.5
12	586.032	1.465	23.4
13	581.637	1.454	24.2
14	571.381	1.429	22.3
15	546.108	1.365	537.3
16	523.766	1.310	529.9
17	503.987	1.260	44.3
18	502.522	1.256	23.7
19	477.616	1.194	2500.0



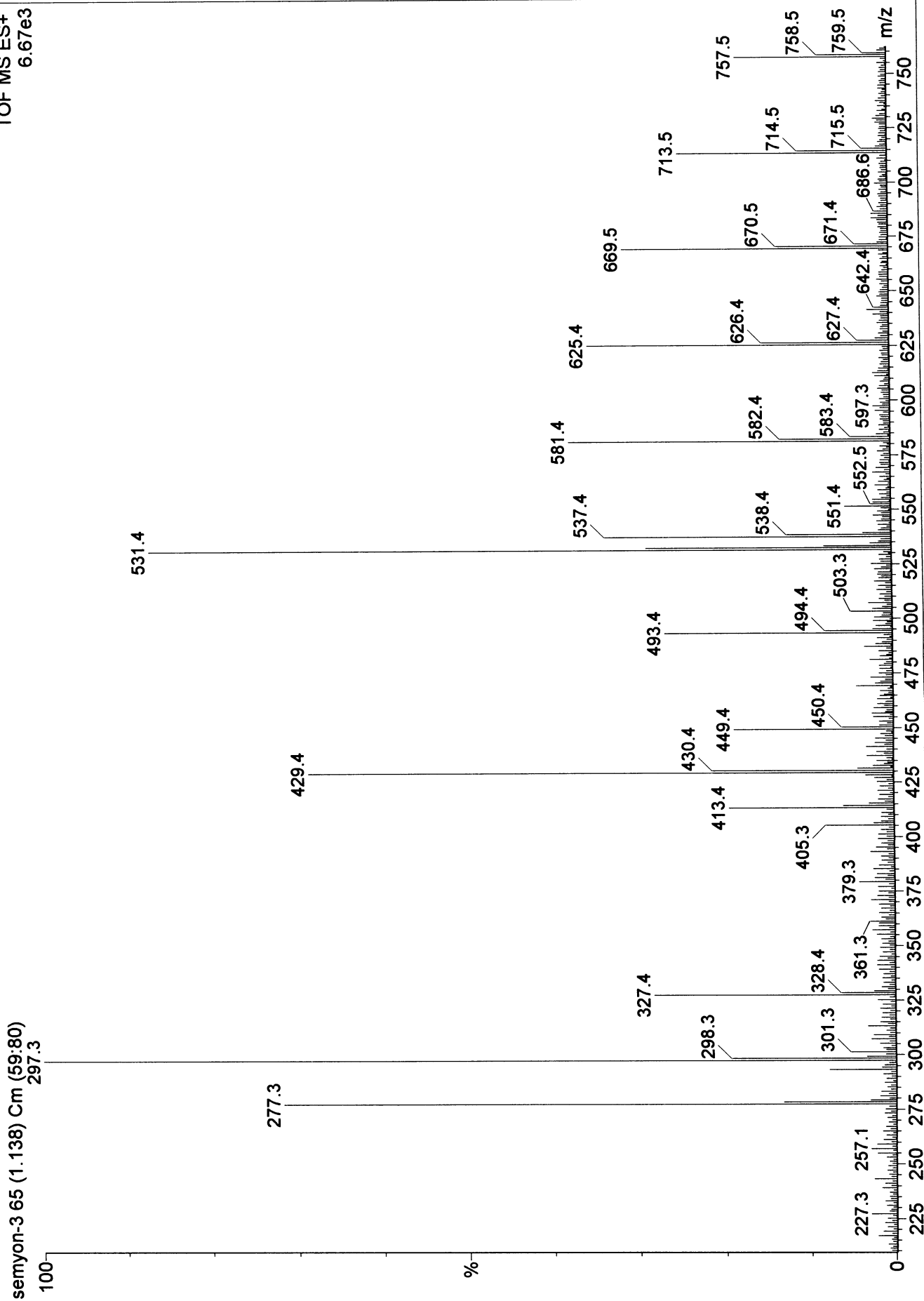
Racemic ether-linked dimer - 7

Pulse Sequence: s2pu1
 Solvent: CDC13
 Ambient temperature
 INOVA-400 "Zr"
 PULSE SEQUENCE
 Relax. delay 1.500 sec
 Pulse 45.0 degrees
 Acq. time 1.199 sec
 Width 25000.0 Hz
 9999 repetitions
 OBSERVE C13, 100.5712662 MHz
 DECOUPLE H1, 399.9669644 MHz
 Power 44 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 75 hr, 13 min, 38 sec

INDEX	FREQUENCY	PPM	HEIGHT
1	17960.822	178.588	9.4
2	14197.126	141.165	27.0
3	12001.320	119.331	29.3
4	9930.639	98.742	11.7
5	9765.839	97.104	8.1
6	7776.032	77.319	130.2
7	7743.987	77.000	132.6
8	7711.943	76.681	125.5
9	7669.980	76.264	26.8
10	7647.091	76.037	25.7
11	6155.499	61.205	45.7
12	3894.078	38.720	20.8
13	3627.804	36.072	62.2
14	2930.456	29.138	29.3
15	2925.115	29.085	25.6
16	2733.611	27.181	141.2
17	2378.071	23.646	22.3
18	2353.656	23.403	20.7
19	2149.182	21.370	19.0
20	2125.530	21.135	19.5
21	1654.019	16.446	32.9



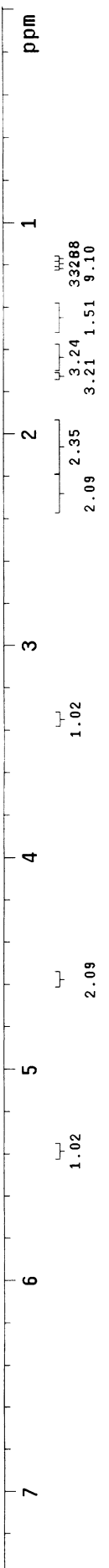
26-Aug-2002 13:12:55
TOF MS ES+
6.67e3



(S)-6,7-dihydroxygeranyl pivaloate - 8

Pulse Sequence: s2pu1
Solvent: CDCl3
Ambient temperature
File: SG22-Cp1
INOVA-500 "N1"
PULSE SEQUENCE
Pulse 40.0 degrees
Acq. time 2.666 sec
Width 6000.6 Hz
72 repetitions
OBSERVE H1, 399.9649457 MHz
DATA PROCESSING
FT size 32768
Total time 5 min, 42 sec

INDEX	FREQUENCY PPM	HEIGHT	INDEX	FREQUENCY PPM	HEIGHT
1	2906.718	7.267	40	477.982	2000.0
2	2155.498	5.389	41	475.418	105.4
3	2154.033	5.386	42	466.994	454.7
4	2152.935	5.383	43	457.837	32.7
5	2121.802	5.305	44	0.000	185.7
6	1834.280	4.586			
7	1833.547	4.584			
8	1827.321	4.569			
9	1826.588	4.567			
10	1396.221	3.491			
11	1344.211	3.361			
12	1342.379	3.356			
13	1333.589	3.334			
14	1331.757	3.330			
15	924.465	2.311			
16	857.804	2.145			
17	850.845	2.127			
18	689.320	1.723			
19	668.809	1.672			
20	665.879	1.665			
21	661.850	1.655			
22	660.018	1.650			
23	654.891	1.637			
24	653.059	1.633			
25	650.495	1.626			
26	647.931	1.620			
27	646.100	1.615			
28	641.339	1.603			
29	638.775	1.597			
30	636.577	1.592			
31	589.328	1.473			
32	583.834	1.460			
33	582.369	1.456			
34	578.706	1.447			
35	573.579	1.434			
36	454.464	1.236			
37	451.168	1.228			
38	488.604	23.7			
39	484.941	468.2			



(R)-6,7-dihydroxygeranyl pivaloate - 9

Pulse Sequence: s2pul

Solvent: CDCl3

Ambient temperature

File: SQ25-cp2main

INOVA-500 "Ni"

PULSE SEQUENCE

Pulse 45.0 degrees

Acq. time 2.666 sec

Width 6000.6 Hz

128 repetitions

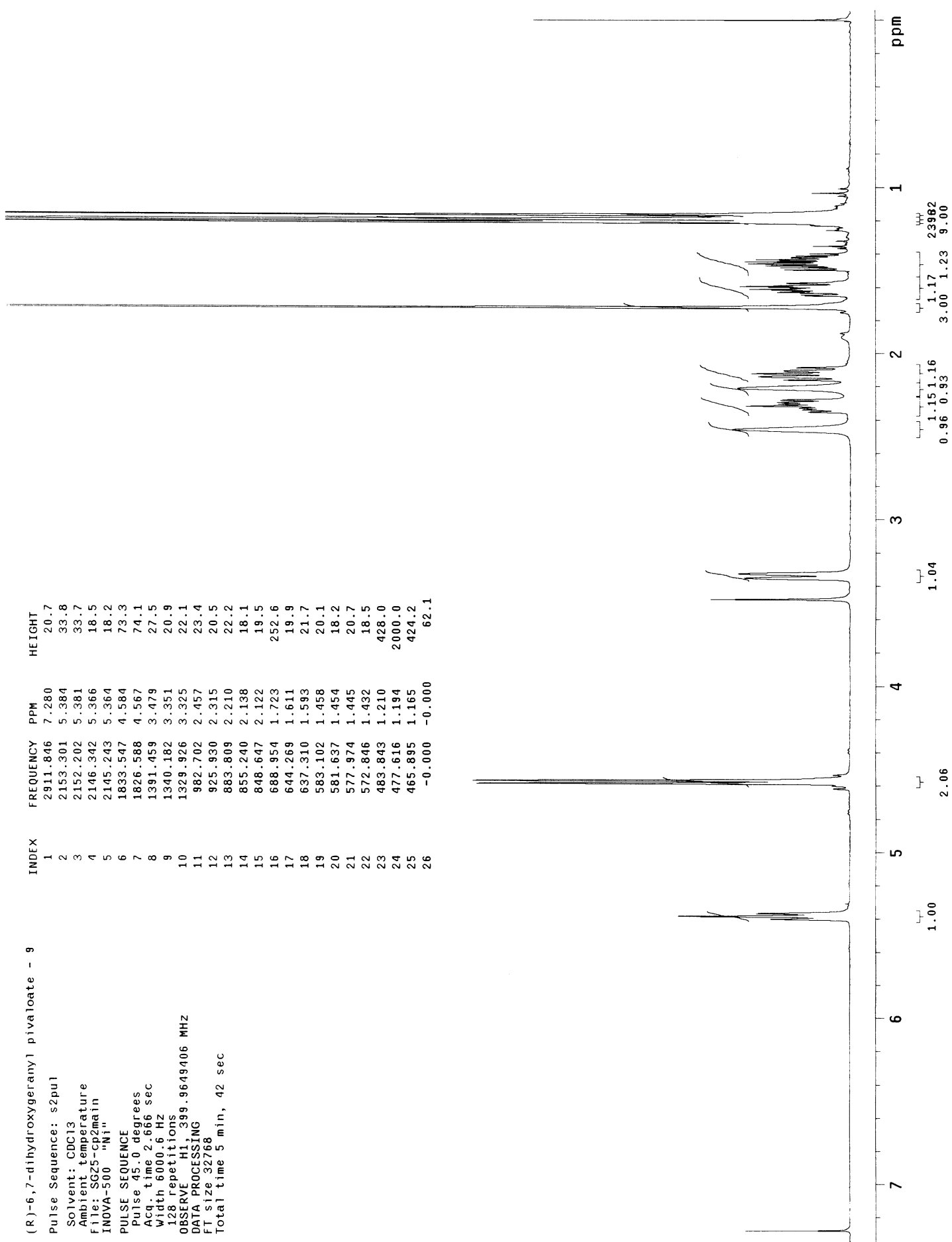
OBSERVE H1, 399.9649406 MHZ

DATA PROCESSING

FT size 32768

Total time 5 min, 42 sec

INDEX	FREQUENCY PPM	HEIGHT
1	2911.846	20.7
2	2153.301	33.8
3	2152.202	33.7
4	2146.342	18.5
5	2145.243	18.2
6	1833.547	73.3
7	1826.588	74.1
8	1391.459	27.5
9	1340.182	20.9
10	1329.926	22.1
11	982.702	23.4
12	925.930	20.5
13	883.809	22.2
14	855.240	18.1
15	848.647	19.5
16	688.954	252.6
17	644.269	19.9
18	637.310	21.7
19	583.102	20.1
20	581.637	18.2
21	577.974	20.7
22	572.846	18.5
23	483.843	428.0
24	477.616	2000.0
25	465.895	424.2
26	-0.000	62.1



DioIs 8 and 9 with 0.8 mol eq Eu(hfc)3, 1:1

Pulse Sequence s2pul

Solvent: CDCl3

Ambient temperature

File: diols-1-11.8

INOVA-500 "N1"

PULSE SEQUENCE

Pulse 35.0 degrees

Acq. time 3.833 sec

Width 4500.51 Hz

32 repetitions

OBSERVE H1, 300.0732252 MHZ

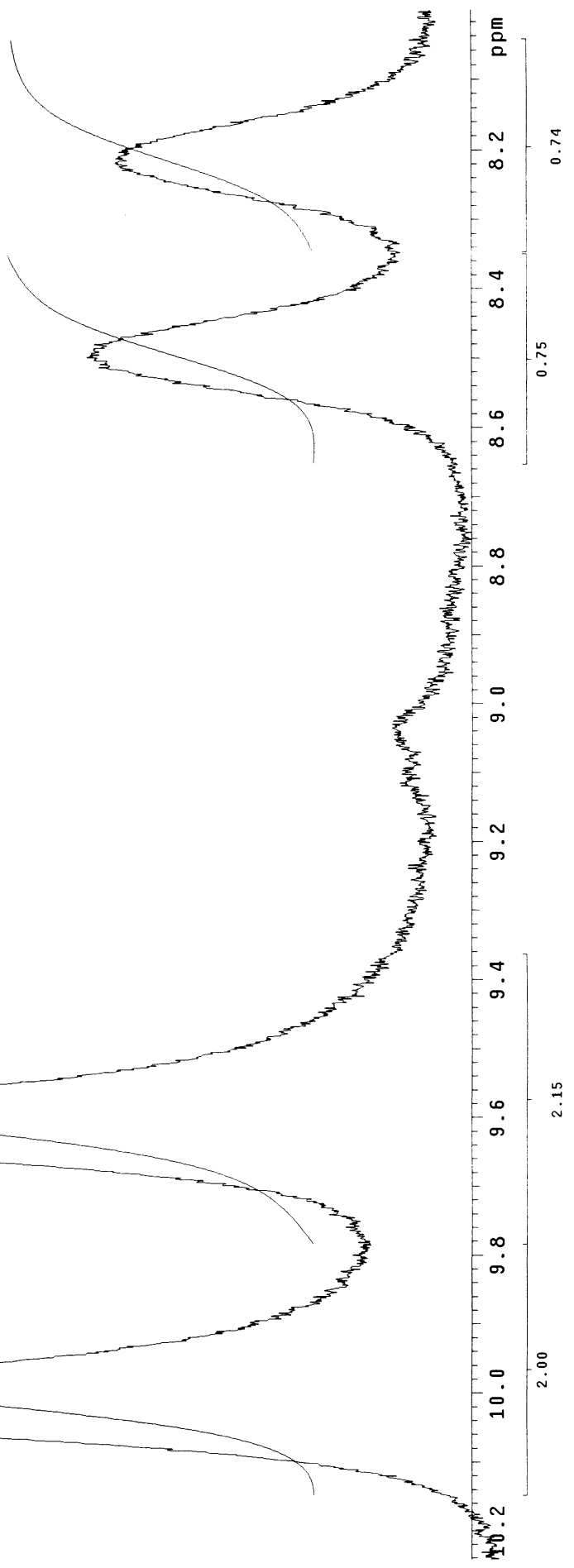
DATA PROCESSING

Line broadening 0.2 Hz

FT size 32768

Total time 7 min, 7 sec

INDEX	FREQUENCY	PPM	HEIGHT
1	3006.343	10.019	157.5
2	2884.650	9.613	149.4
3	2550.612	8.500	55.5
4	2466.004	8.218	51.0



DioIs 8 and 9 with 0.8 mol eq Eu(hfc)3, 1:2

Pulse Sequence: s2pul

Solvent: CDCl3

Ambient temperature

File: diols-2-1-18

INOVA-500 "Ni"

PULSE SEQUENCE

Pulse 35.0 degrees

Acq. time 3.333 sec

Width 4500.5 Hz

96 repetitions

OBSERVE H1, 300.0732213 MHZ

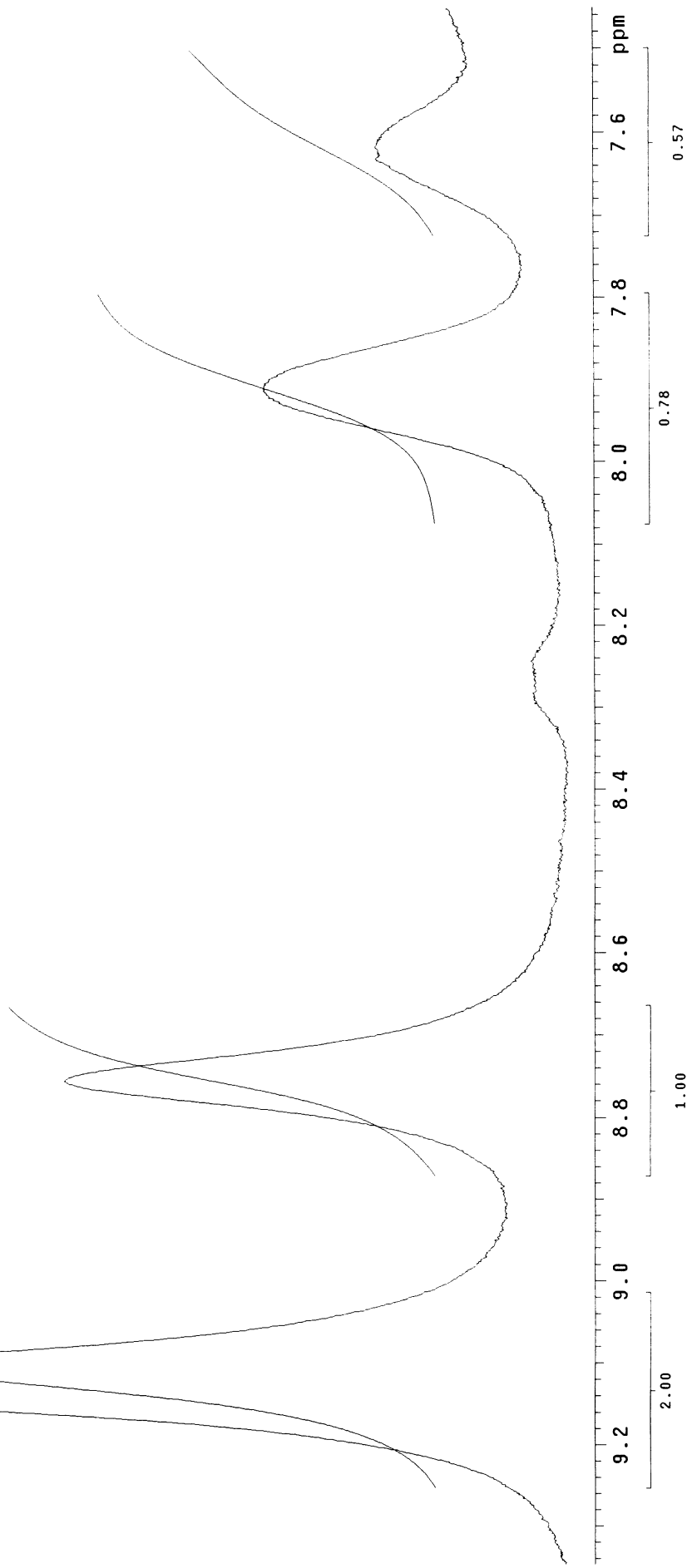
DATA PROCESSING

Line broadening 0.2 Hz

FT size 32768

Total time 7 min, 7 sec

INDEX	FREQUENCY PPM	HEIGHT
1	2737.684	157.1
2	2627.254	77.9
3	2374.803	46.7



Diol 8 with 0.8 mol eq Eu(hfc)3

Pulse Sequence: s2pu1

Solvent: CDCl3

Ambient temperature

File: diol-3eealtint

INOVA-500 "N1"

PULSE SEQUENCE

Pulse 35.0 degrees

Acq. time 3.333 sec

Width 4500.5 Hz

32 repetitions

OBSERVE H1, 300.0732271 MHz

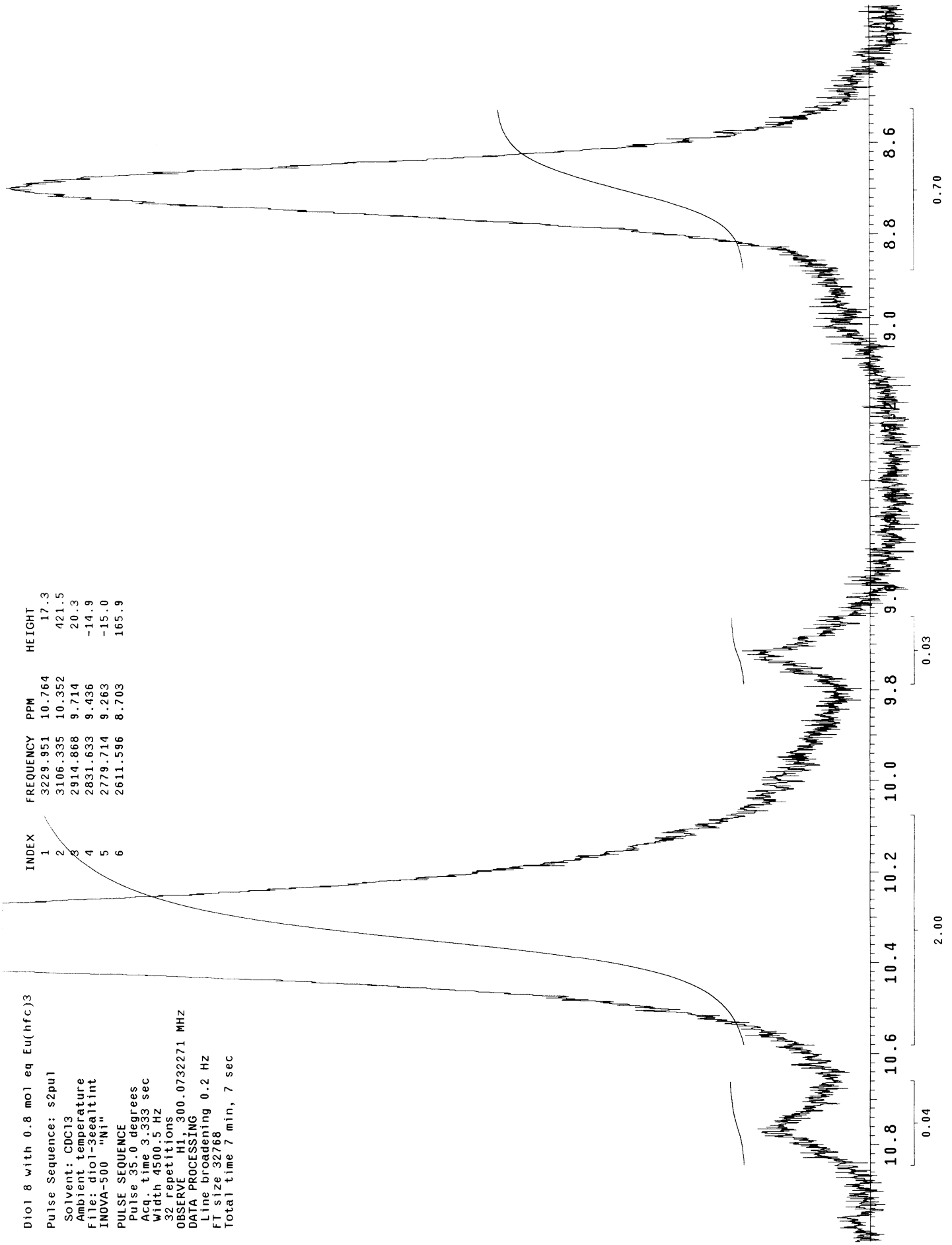
DATA PROCESSING

Line broadening 0.2 Hz

FT size 32768

Total time 7 min, 7 sec

INDEX	FREQUENCY	PPM	HEIGHT
1	3229.951	10.764	17.3
2	3106.335	10.352	421.5
3	2914.868	9.714	20.3
4	2831.633	9.436	-14.9
5	2779.714	9.263	-15.0
6	2611.596	8.703	165.9



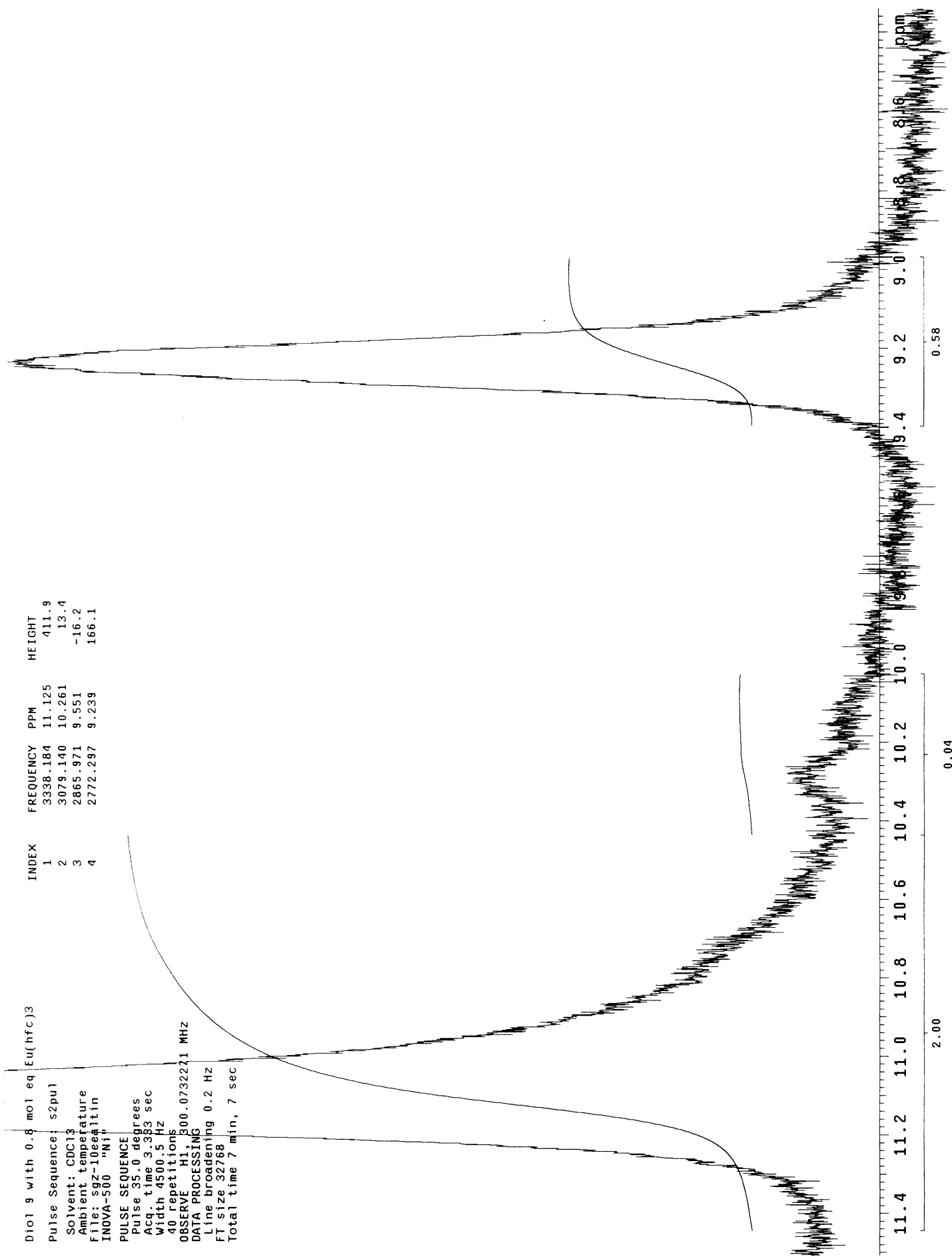
Diol 9 with 0.8 mol eq Eu(hfc)3

Pulse Sequence: s2pul
Solvent: CDCl3
Ambient temperature
File: sgz-10eeatlin
INOVA-500 "N1"

PULSE SEQUENCE

Pulse 35.0 degrees
Acq. time 3.383 sec
Width 4500.5 Hz
40 repetitions
OBSERVE H1 800.0732271 MHZ
DATA PROCESSING
Line broadening 0.2 Hz
FT size 32768
Total time 7 min, 7 sec

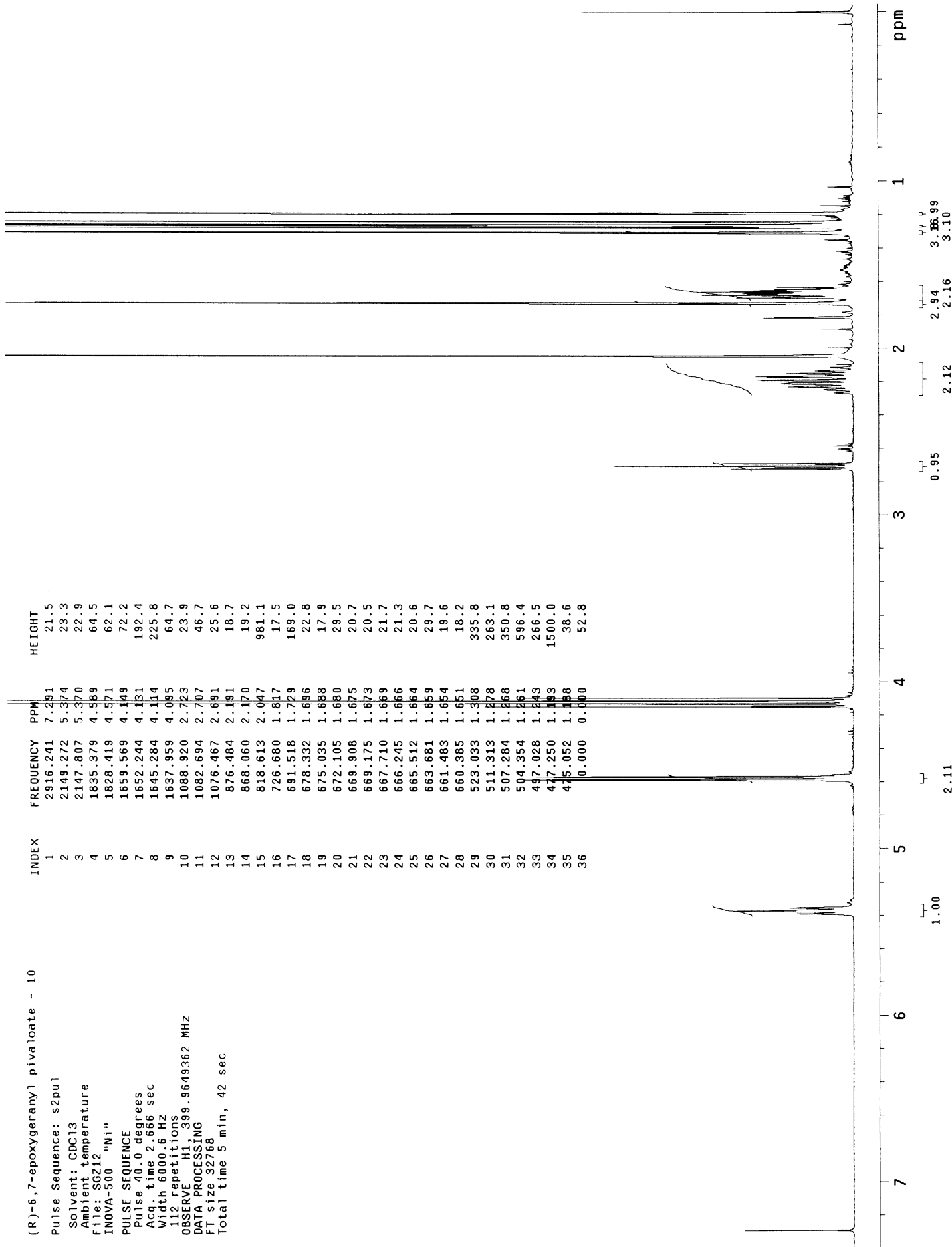
INDEX	FREQUENCY PPM	HEIGHT
1	3338.184	411.9
2	3079.140	13.4
3	2865.971	-16.2
4	2772.297	166.1



(R)-6,7-epoxygeranyl pivaloate - 10

Pulse Sequence: s2pul
 Solvent: CDC13
 Ambient temperature
 File: SG212
 INOVA-500 "N1"
 PULSE SEQUENCE
 Pulse 40.0 degrees
 Acq. time 2.666 sec
 Width 6000.6 Hz
 112 repetitions
 OBSERVE H1, 399.9649362 MHZ
 DATA PROCESSING
 FT size 32768
 Total time 5 min, 42 sec

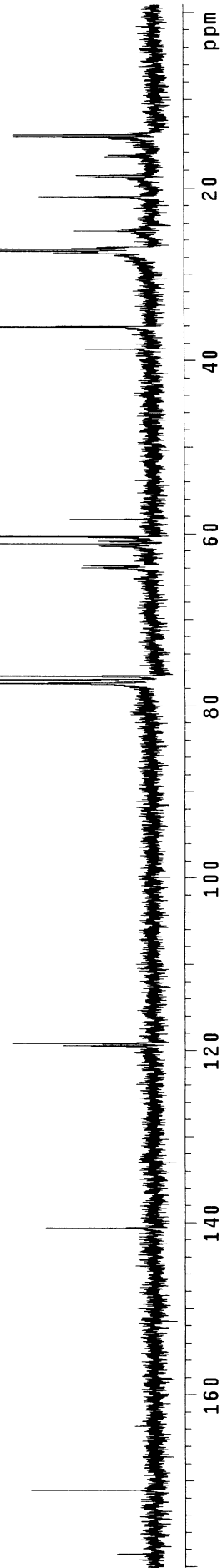
INDEX	FREQUENCY PPM	HEIGHT
1	2916.241	7.291
2	2149.272	5.374
3	2147.807	5.370
4	1835.379	4.589
5	1828.419	4.571
6	1659.569	4.149
7	1652.244	4.131
8	1645.284	4.114
9	1637.959	4.095
10	1088.920	2.723
11	1082.694	2.707
12	1076.467	2.691
13	876.484	2.191
14	868.060	2.170
15	818.613	2.047
16	726.680	1.817
17	691.518	1.729
18	678.332	1.696
19	675.035	1.688
20	672.105	1.680
21	669.908	1.675
22	669.175	1.673
23	667.710	1.669
24	666.245	1.666
25	665.512	1.664
26	663.681	1.659
27	661.483	1.654
28	660.385	1.651
29	523.033	1.308
30	511.313	1.278
31	507.284	1.268
32	504.354	1.261
33	497.028	1.243
34	477.250	1.193
35	475.052	1.188
36	0.000	0.000



(R)-6,7-epoxygeranyl pivaloate - 10

Pulse Sequence: s2pul
Solvent: CDCl3
Ambient temperature
File: SGZ12-13C
INOVA-500 "N1"
PULSE SEQUENCE
Relax. delay 1.000 sec
Pulse 40.0 degrees
Acq. time 1.706 sec
Width 18761.7 Hz
95Z repetitions
OBSERVE C13, 75.4534754 MHZ
DECOUPLE H1, 300.0746975 MHZ
Power 40 dB
continuously on
WALTZ-16 modulated
DATA PROCESSING
Line broadening 1.0 Hz
FT size 65536
Total time 3 hr, 5 min, 18 sec

INDEX	FREQUENCY	PPM	HEIGHT
1	13467.601	178.489	5.7
2	12907.618	171.067	19.1
3	10609.282	140.607	16.8
4	9011.784	119.435	14.1
5	8992.317	119.177	22.0
6	5841.982	77.425	97.8
7	5809.918	77.000	99.6
8	5777.853	76.575	96.2
9	4827.943	63.986	11.1
10	4805.040	63.682	10.8
11	4633.838	61.413	8.3
12	4613.226	61.140	41.1
13	4592.040	60.859	8.5
14	4550.814	60.313	40.0
15	4399.653	58.309	12.9
16	2917.244	38.663	10.4
17	2723.139	36.090	34.9
18	2073.834	27.485	15.4
19	2056.084	27.250	43.1
20	2038.906	27.022	51.9
21	2020.584	26.779	8.5
22	1879.156	24.905	12.1
23	1861.407	24.670	12.9
24	1583.705	20.989	17.5
25	1581.415	20.959	16.2
26	1418.230	18.796	10.0
27	1400.480	18.561	11.8
28	1244.165	16.489	7.3
29	1231.569	16.322	6.8
30	1091.859	14.471	6.6
31	1074.682	14.243	21.6
32	1056.932	14.008	21.6



(S)-6,7-epoxygeranyl pivaloate - 11

Pulse Sequence: s2pul

Solvent: CDCl3

Ambient temperature

File: SG214-1

INOVA-500 "ni"

PULSE SEQUENCE

Pulse 40.0 degrees

Acq. time 2.666 sec

Width 6000.6 Hz

80 repetitions

OBSERVE HI, 399.9649413 MHZ

DATA PROCESSING

FT size 32768

Total time 5 min, 42 sec

INDEX	FREQUENCY	PPM	HEIGHT
1	2148.539	5.372	31.6
2	2147.074	5.368	32.5
3	1834.646	4.587	94.2
4	1827.687	4.570	90.5
5	1088.920	2.723	34.9
6	1082.694	2.707	67.9
7	1076.467	2.691	37.1
8	875.751	2.190	26.1
9	867.327	2.169	27.5
10	818.613	2.047	27.9
11	690.053	1.725	249.6
12	677.966	1.695	32.5
13	674.303	1.686	25.7
14	671.739	1.679	42.3
15	669.908	1.675	29.2
16	668.443	1.671	28.5
17	667.344	1.669	30.0
18	665.512	1.664	39.6
19	663.315	1.658	43.0
20	661.117	1.653	28.6
21	659.652	1.649	26.3
22	522.667	1.307	467.0
23	506.918	1.267	498.8
24	504.354	1.261	35.9
25	477.250	1.193	2000.0
26	-0.000	-0.000	59.7

